

Electronic Supplementary Information

Stabilization of the base-off forms of vitamin B₁₂ and coenzyme B₁₂ by complexation of the α -axial 5,6-dimethylbenzimidazole ligand with cucurbit[7]uril

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Materials and Methods

The vitamin B12, vitamin B12a, coenzyme B12, and 5,6-dimethylbenzimidazole were used as received (Sigma-Aldrich). The 1,5,6-trimethylbenzimidazole was prepared by reacting 5,6-dimethylbenzimidazole with methyl iodide.^{S1} The cucurbit[7]uril was prepared and characterized by the method of Day *et al.*^{S2} The UV-visible spectra were recorded on a Hewlett-Packard 8452A diode-array spectrometer in 1.00 cm quartz cells. The ¹H NMR spectra were obtained on Bruker Avance 400 and 500 instruments in D₂O. The host-guest stability constants were determined using ¹H NMR competitive binding measurements with 3-trimethylsilyl-propionic-2,2,3,3-*d*₄ acid ($K_{CB[7]} = (1.82 \pm 0.22) \times 10^7 \text{ dm}^3 \text{ mol}^{-1}$) at pD = 4.75 (0.05 mol dm⁻³ sodium acetate/acetic acid).^{S3} The electrospray ionization mass spectra were obtained in aqueous solution employing Applied Biosystems/MDS Sciex QSTAR XL QqTOF and Waters ZQ Single Quad ESI-MS mass spectrometers.

References

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- S2. A. Day, A.P. Arnold, R. J. Blanch and B. Snushall, *B. J. Org. Chem.* 2001, **66**, 8094.
- S3. S. Liu, C. Ruspic, P. Mukhopadhyay, S. Chakrabarti and P. Y. Zavalij, L. Isaacs, *J. Am. Chem. Soc.* 2005, **127**, 15959.

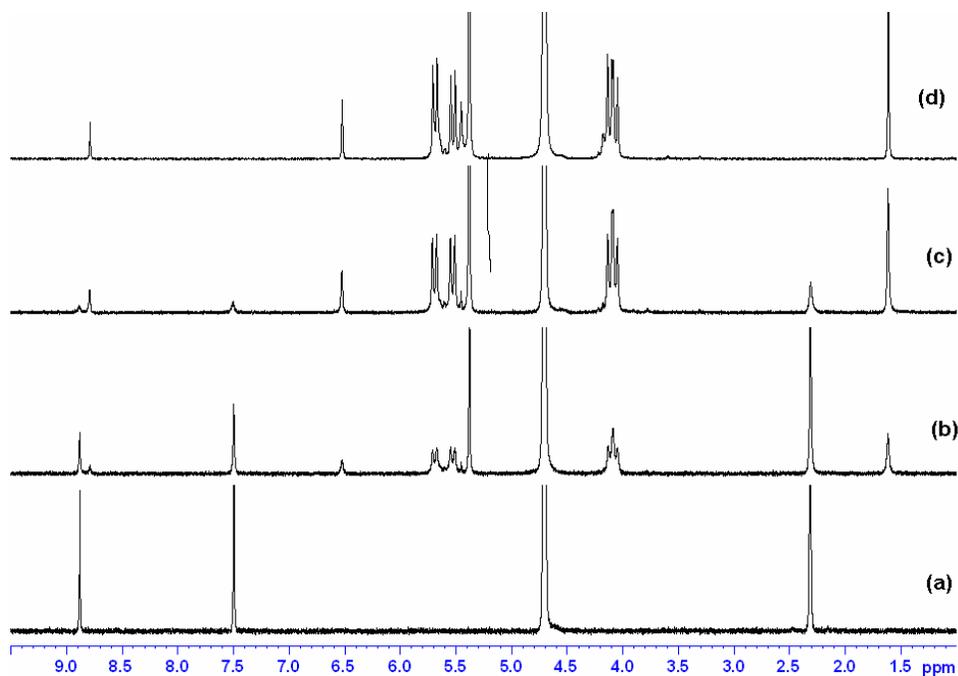


Figure S1. ^1H NMR spectra (400 MHz, D_2O) of protonated 5,6-dimethylbenzimidazole (HDMB^+) in the (a) absence and in the presence of (b) 0.3, (c) 0.8 and (d) 1.3 equivalents of CB[7].

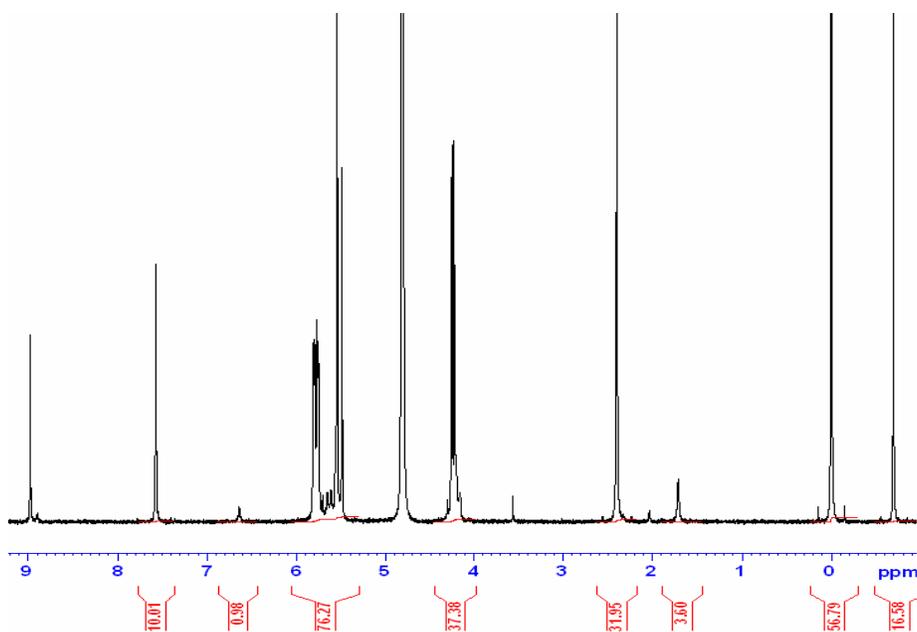


Figure S2. The ^1H NMR spectrum (400 MHz, D_2O , acetate buffer) of the competitive binding of CB[7] by protonated 5,6-dimethylbenzimidazole (HDMB^+) and TSP. ($K_{\text{CB}[7]} = 6.65 \times 10^6 \text{ dm}^3 \text{ mol}^{-1}$)

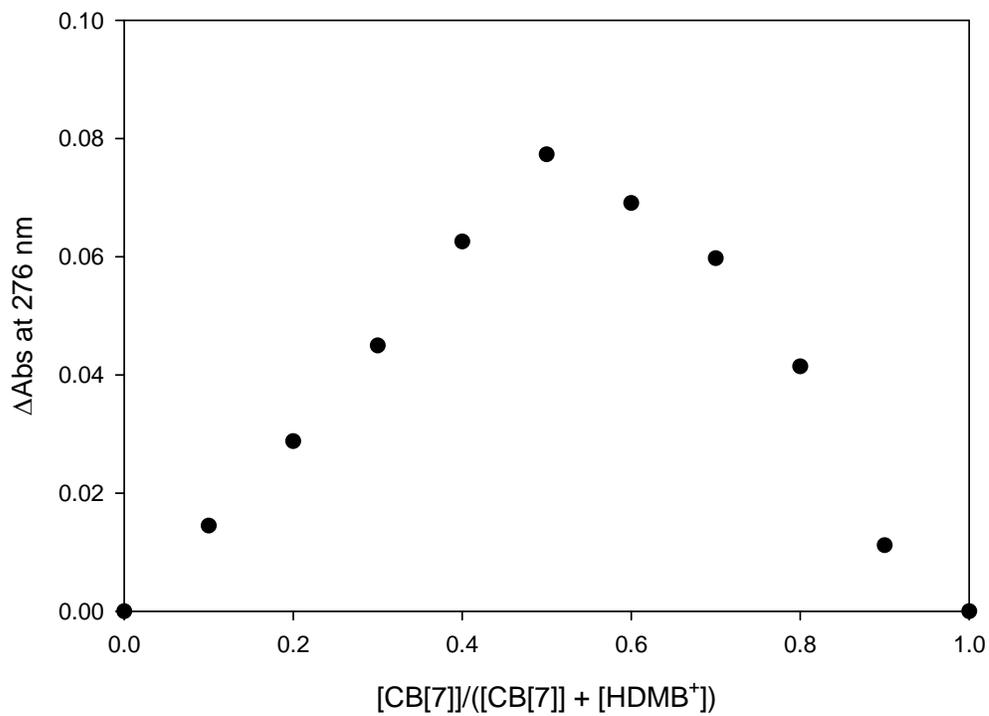


Figure S3. The Job's plot of HDMB⁺ and CB[7], with the total concentration maintained at 0.10 mmol dm⁻³.

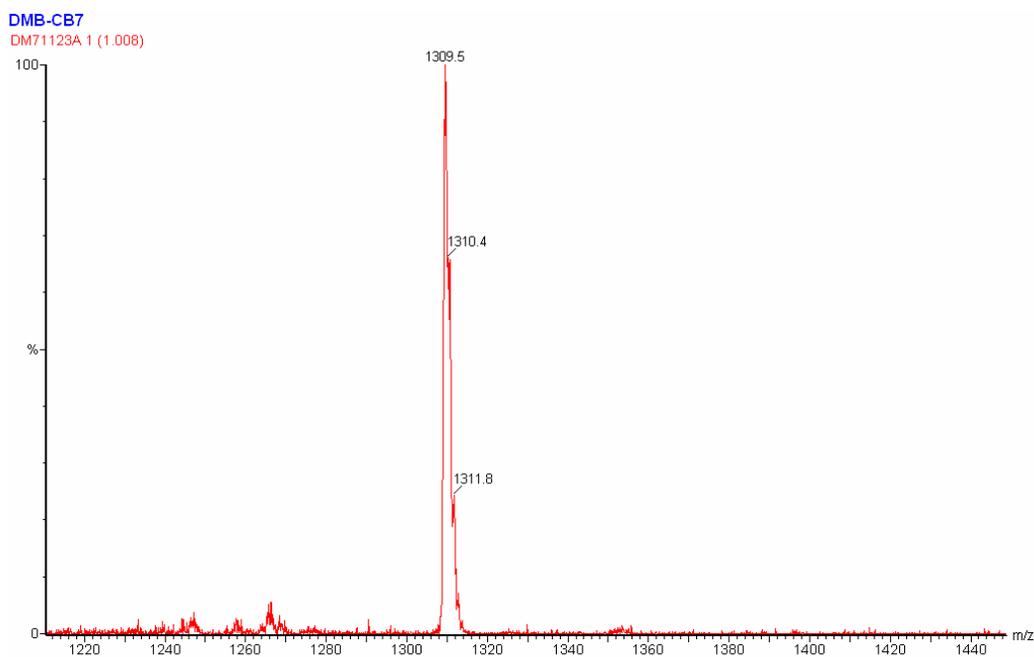


Figure S4. The electrospray mass spectrum of $\{\text{HDMB}\cdot\text{CB}[7]\}^+$

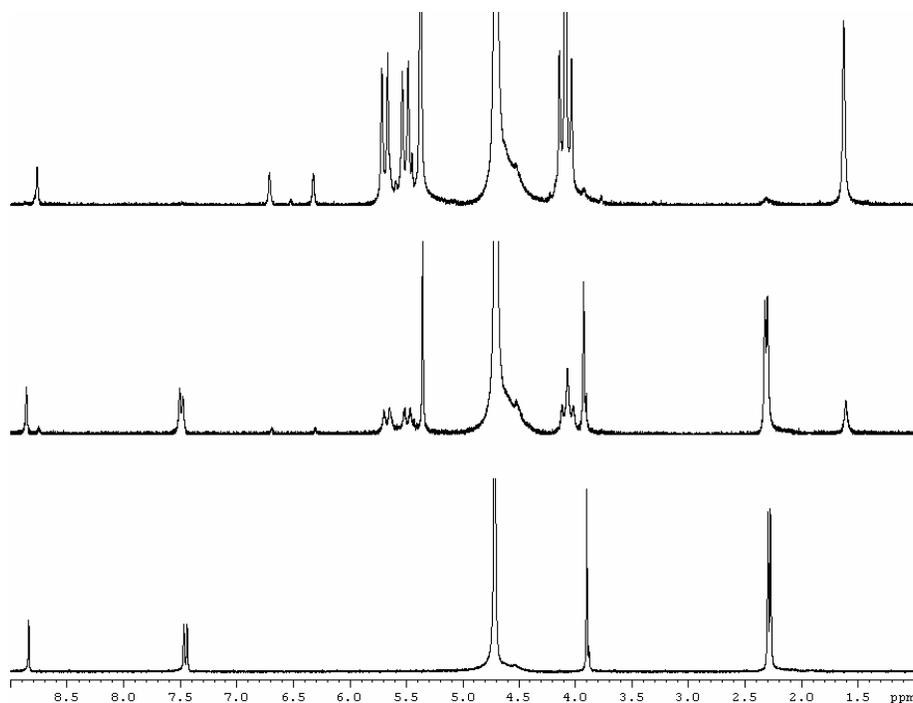


Figure S5. ^1H NMR spectra (400 MHz, D_2O) of protonated 1,5,6-trimethylbenzimidazole (HTMB^+) in the absence (bottom) and in the presence of 0.2 (middle) and 1.3 equivalents (top) of CB[7].

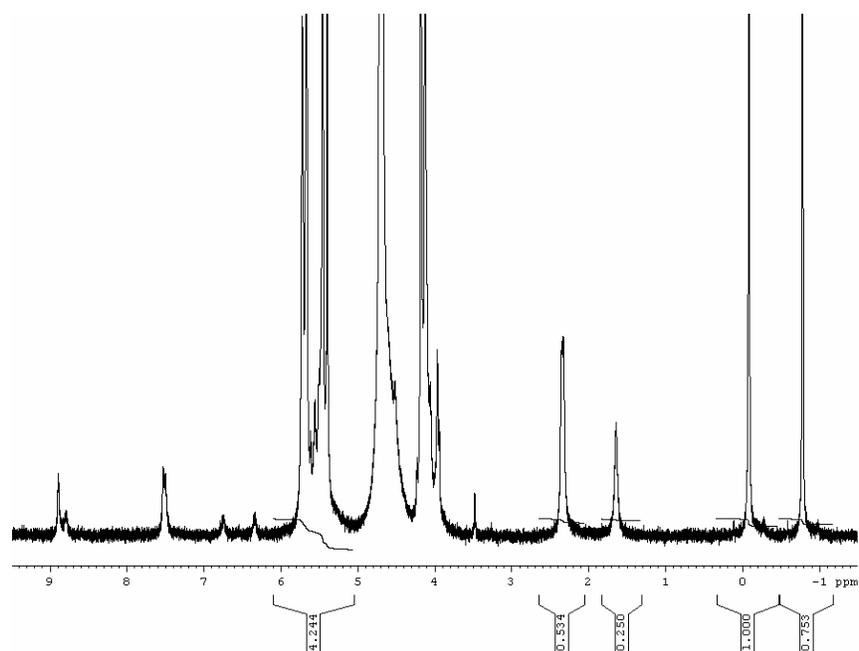


Figure S6. The ^1H NMR spectrum (400 MHz, D_2O , 0.050 M $\text{CH}_3\text{COONa/H}$) of competitive binding of CB[7] between 1,5,6-trimethylbenzimidazole and TSP ($K_{\text{CB}[7]} = (1.13 \pm 0.82) \times 10^7 \text{ dm}^3 \text{ mol}^{-1}$).

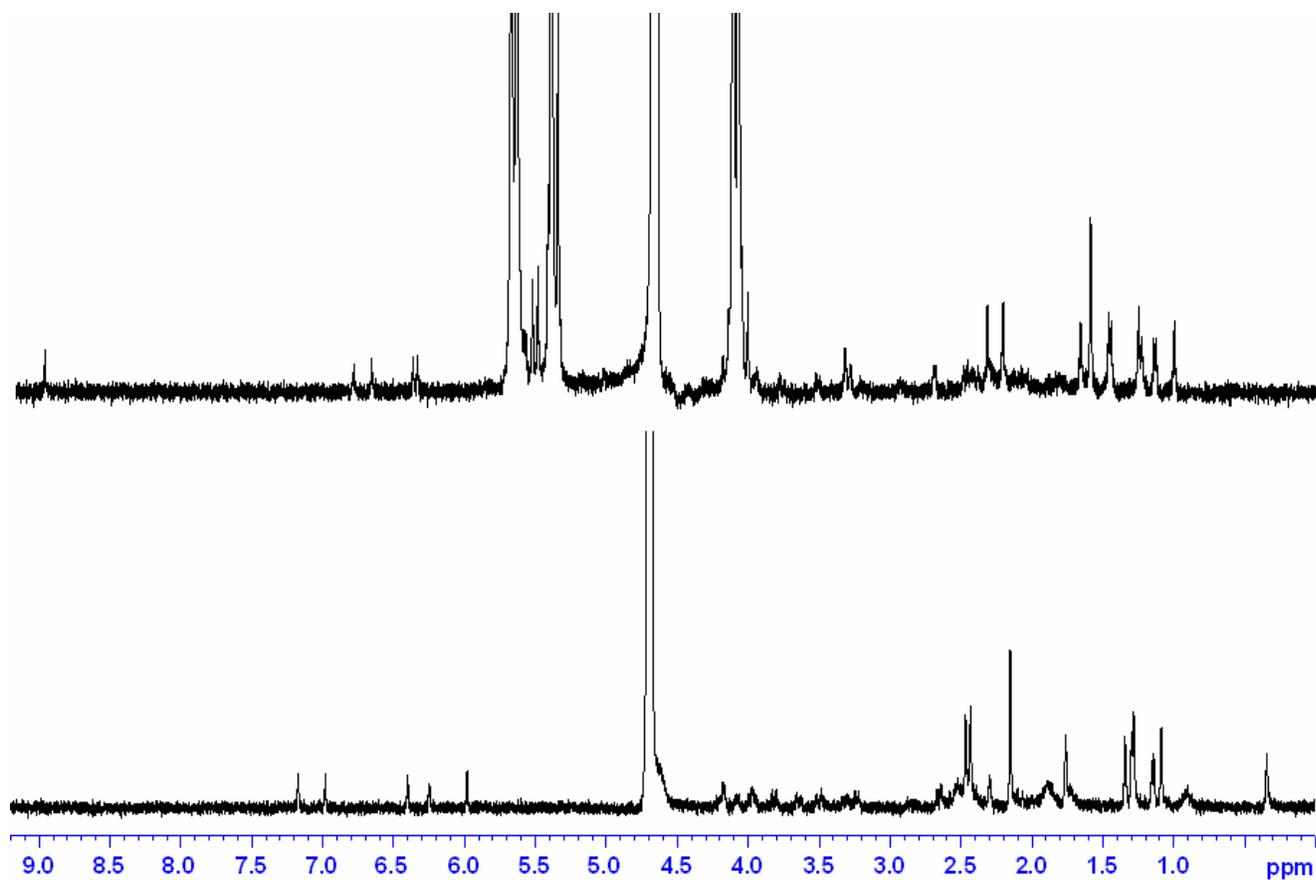


Figure S7. ¹H NMR spectra of CNCbl (bottom) and {CNCbl•CB[7]}⁺ (2 equivalents of CB[7], top) in D₂O at pD 2.

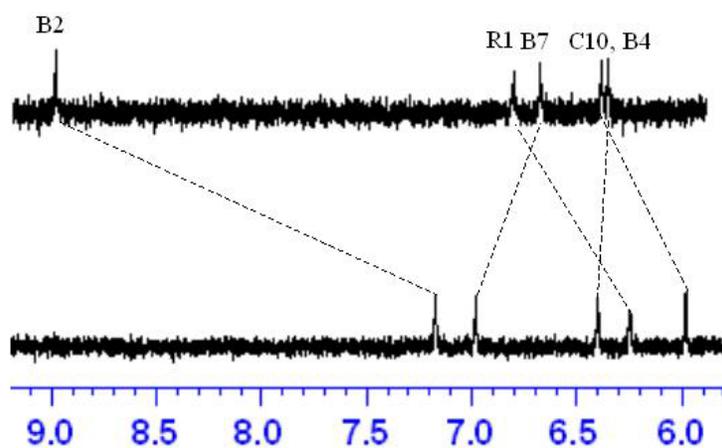


Figure S8. The aromatic portion of the ¹H NMR spectra (400 MHz, D₂O) of CNCbl (pD = 2) in the absence (lower spectrum) and in the presence of 2 equivalents of CB[7] (upper spectrum).

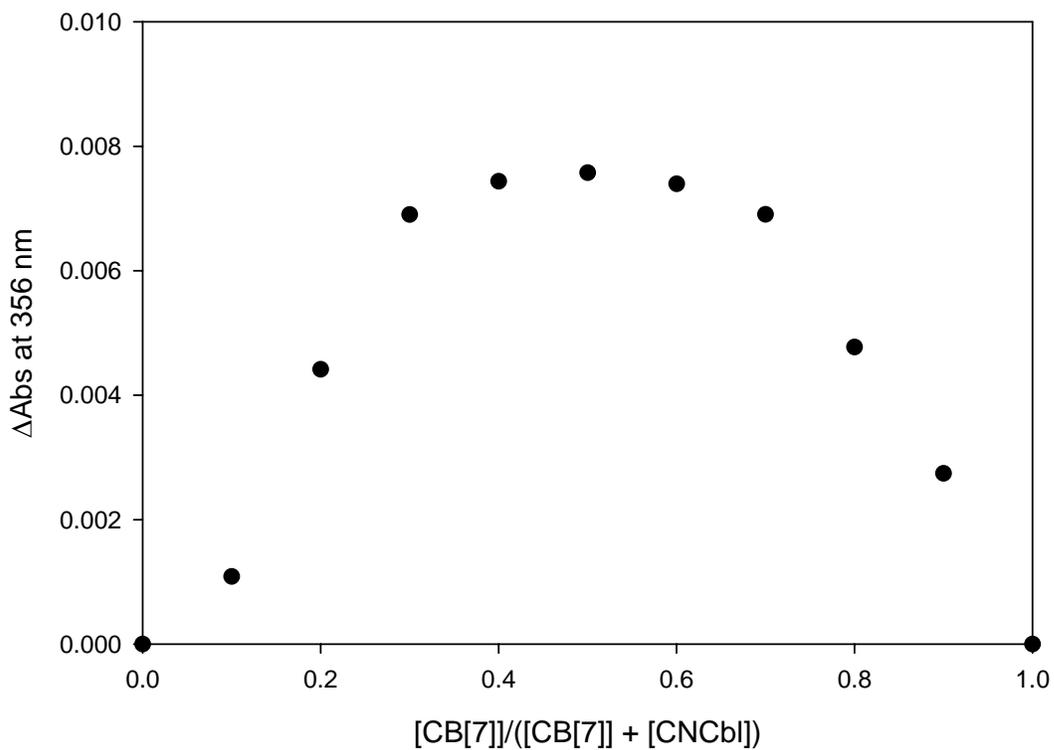


Figure S9. The Job's plot of CNCbl and CB[7], with the total concentration maintained at $18 \mu\text{mol dm}^{-3}$.

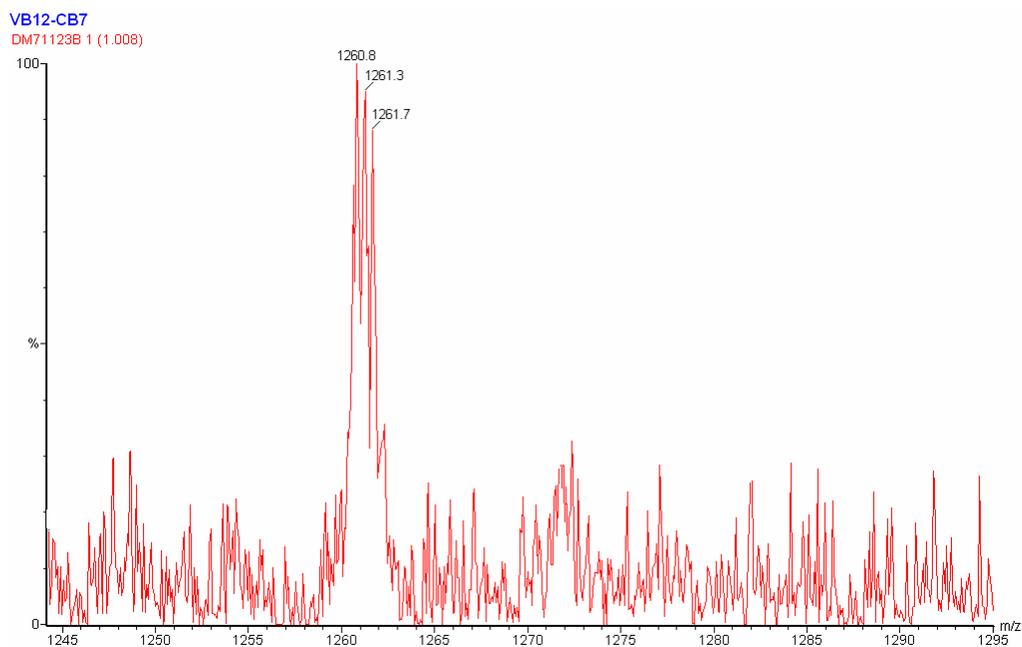


Figure S10. The electrospray mass spectrum of $\{\text{CNCbl}\cdot\text{CB}[7]\}^{2+}$

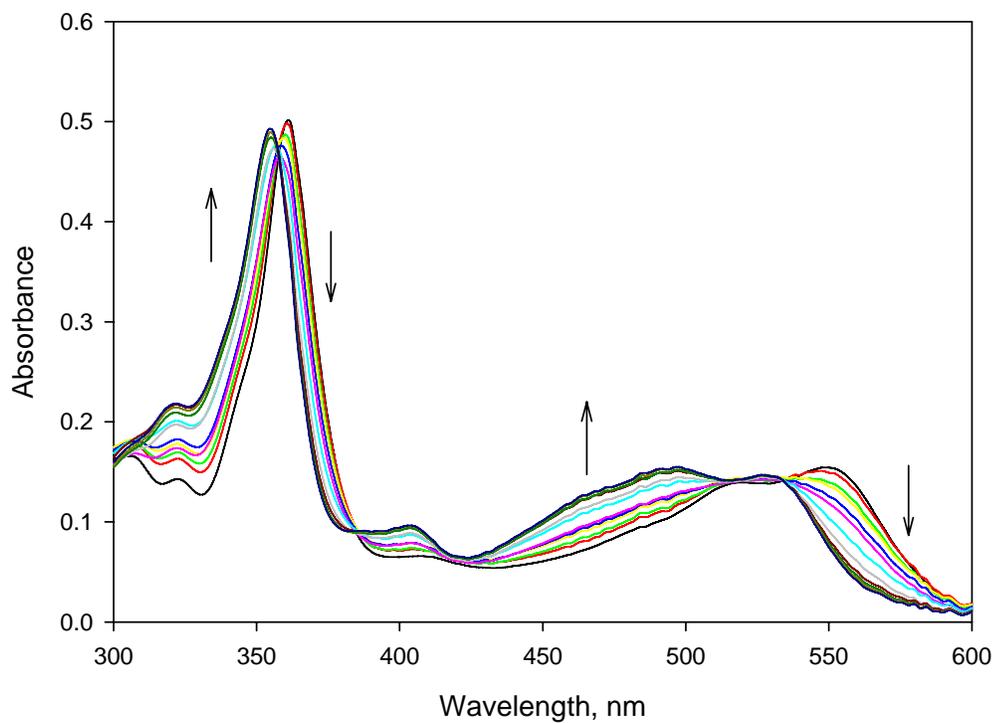


Figure S11. The UV-visible titration of CNCbl (20 μM) at pH=2 (HClO_4) with CB[7] (up to 2.5 equivalents).

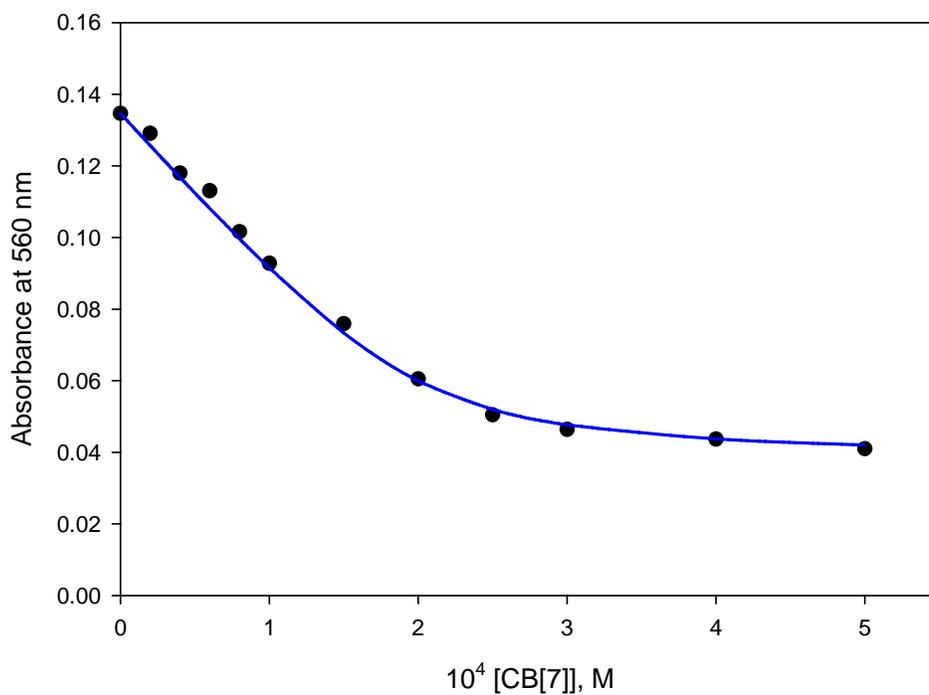


Figure S12. The titration of CNCbl with CB[7] (pH = 2), monitored at 560 nm ($K_{\text{CB}[7]} = 7.5 \times 10^4 \text{ dm}^3 \text{ mol}^{-1}$).

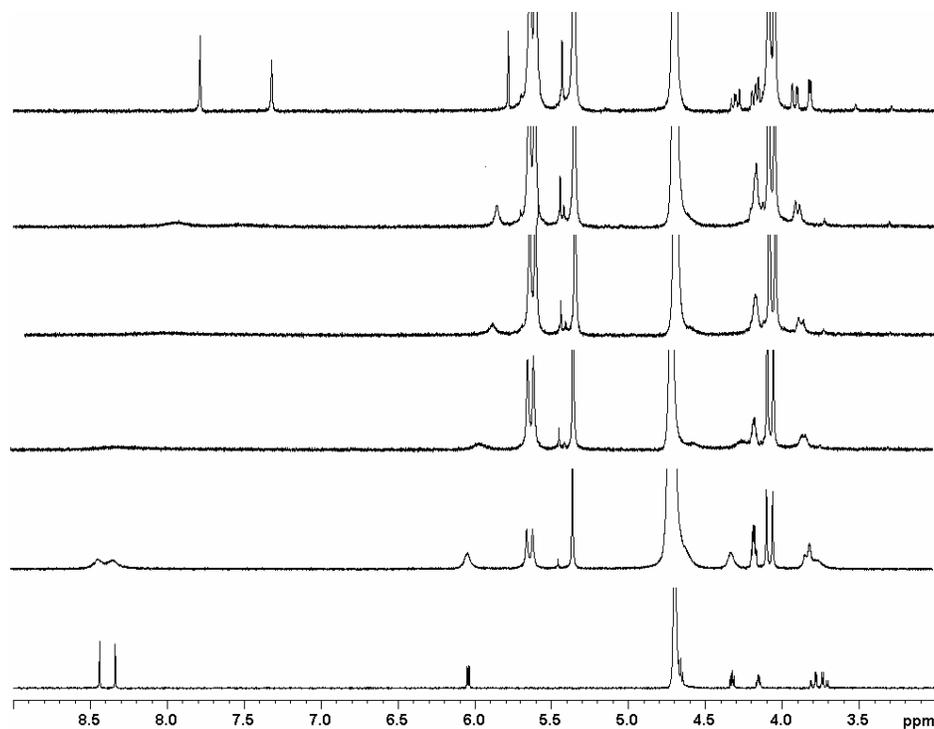


Figure S13. The ^1H NMR spectra (400 MHz, D_2O), bottom to top, of adenosine ($\text{pD} = 2$) in the absence and in the presence of 0.2, 0.7, 1.0, 1.2 and 1.4 equivalents of CB[7].

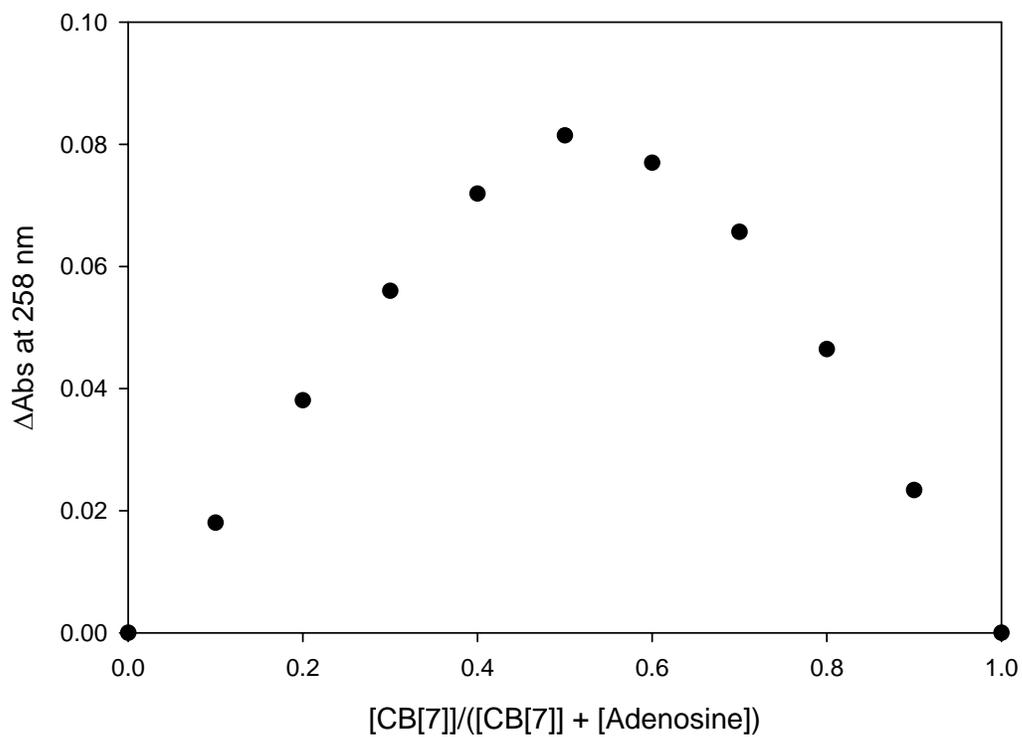


Figure S14. The Job's plot of protonated adenosine and CB[7], keeping the total concentration at $50 \mu\text{mol dm}^{-3}$.

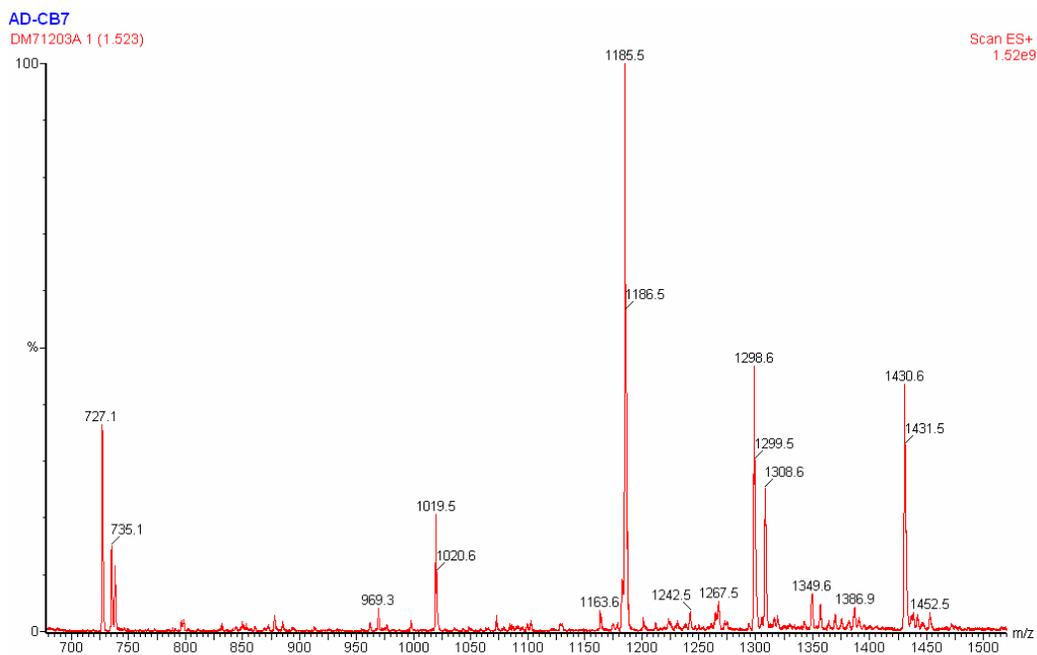


Figure S15. The electrospray mass spectrum of adenosine with CB[7].

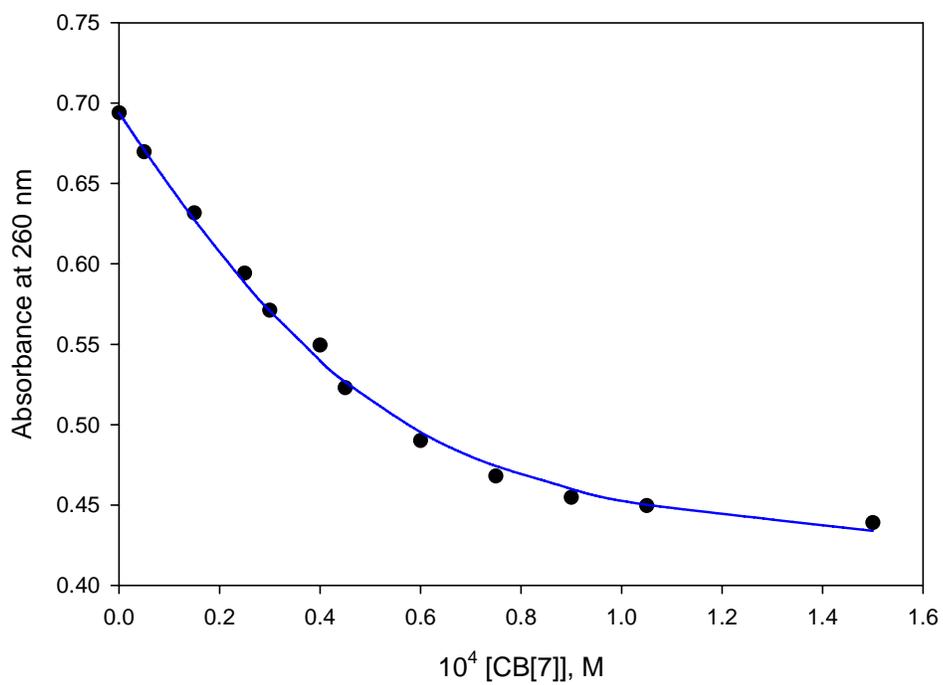


Figure S16. The titration of protonated adenosine (pH = 2) with CB[7]. $K_{\text{CB}[7]} = 8.6 \times 10^4 \text{ M}^{-1}$.

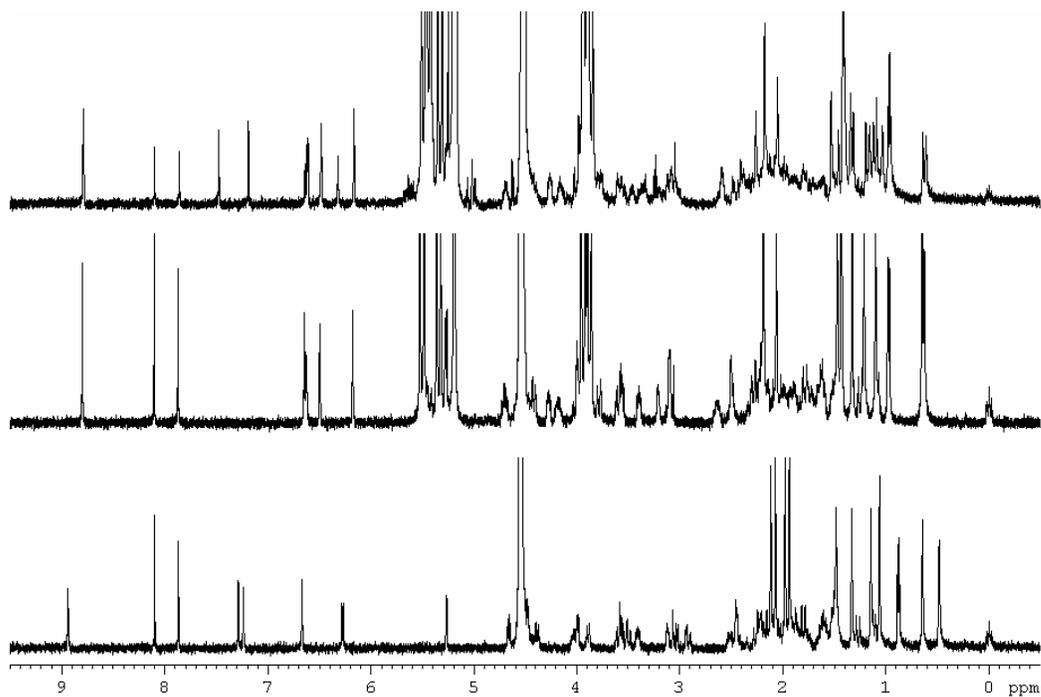


Figure S17. The ^1H NMR spectra of AdoCbl in the absence (bottom) and presence of 1.2 equivalents (middle) and 4.2 equivalents (top) of CB[7] in D_2O .

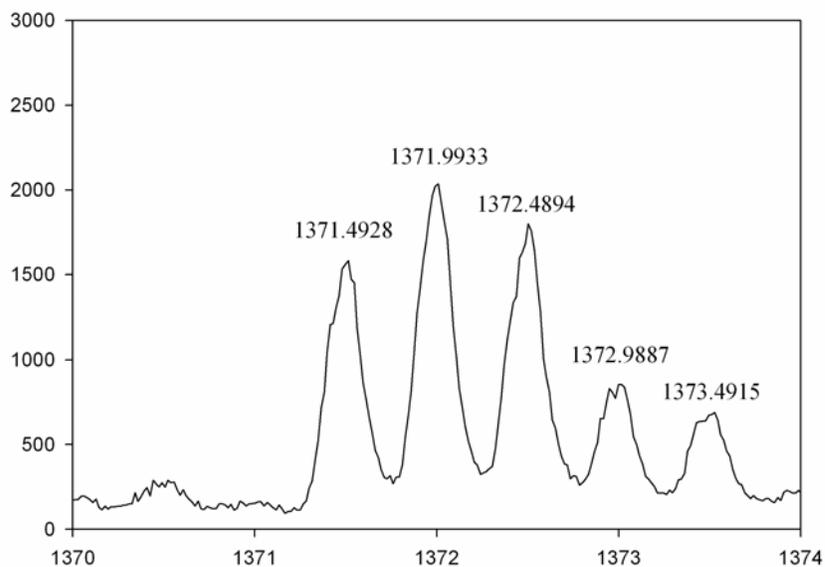


Figure S18. High resolution electrospray TOF mass spectrum of the $\{\text{Ado}\cdot\text{CB}[7] + 2\text{H}\}^{2+}$ species.

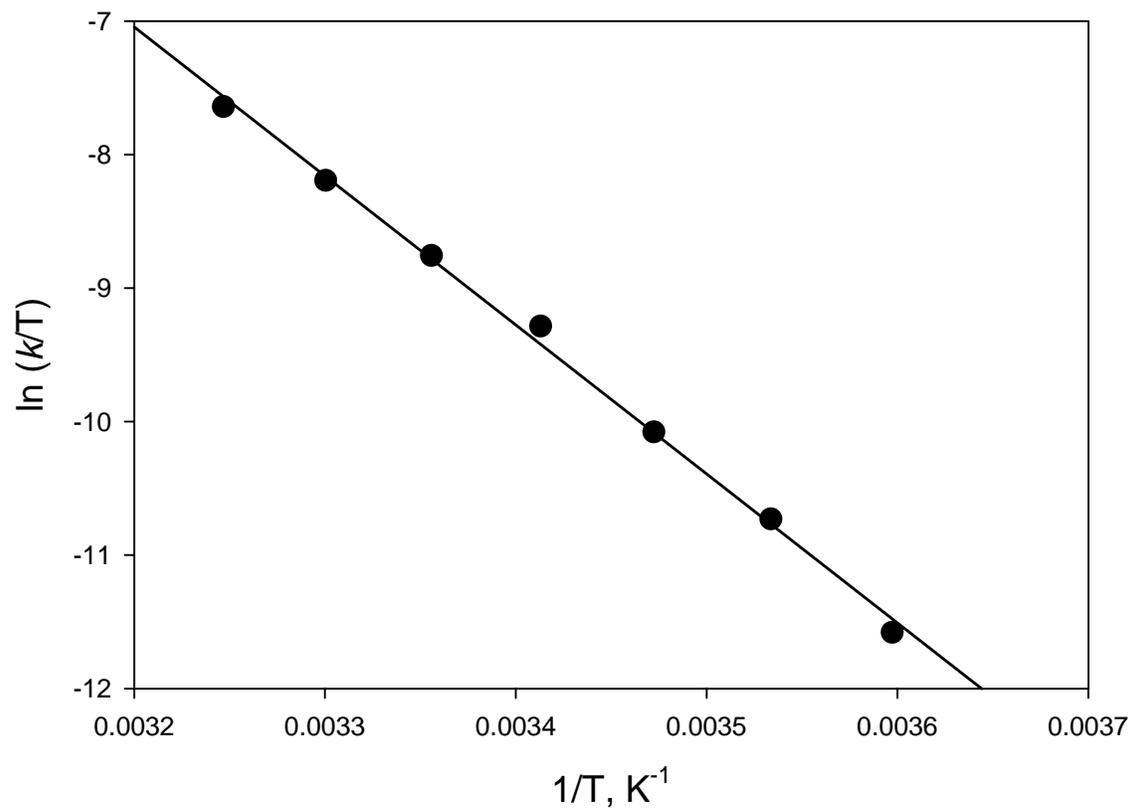


Figure S19. Eyring plot for the reaction of CNCbl with CB[7] at pH 2.

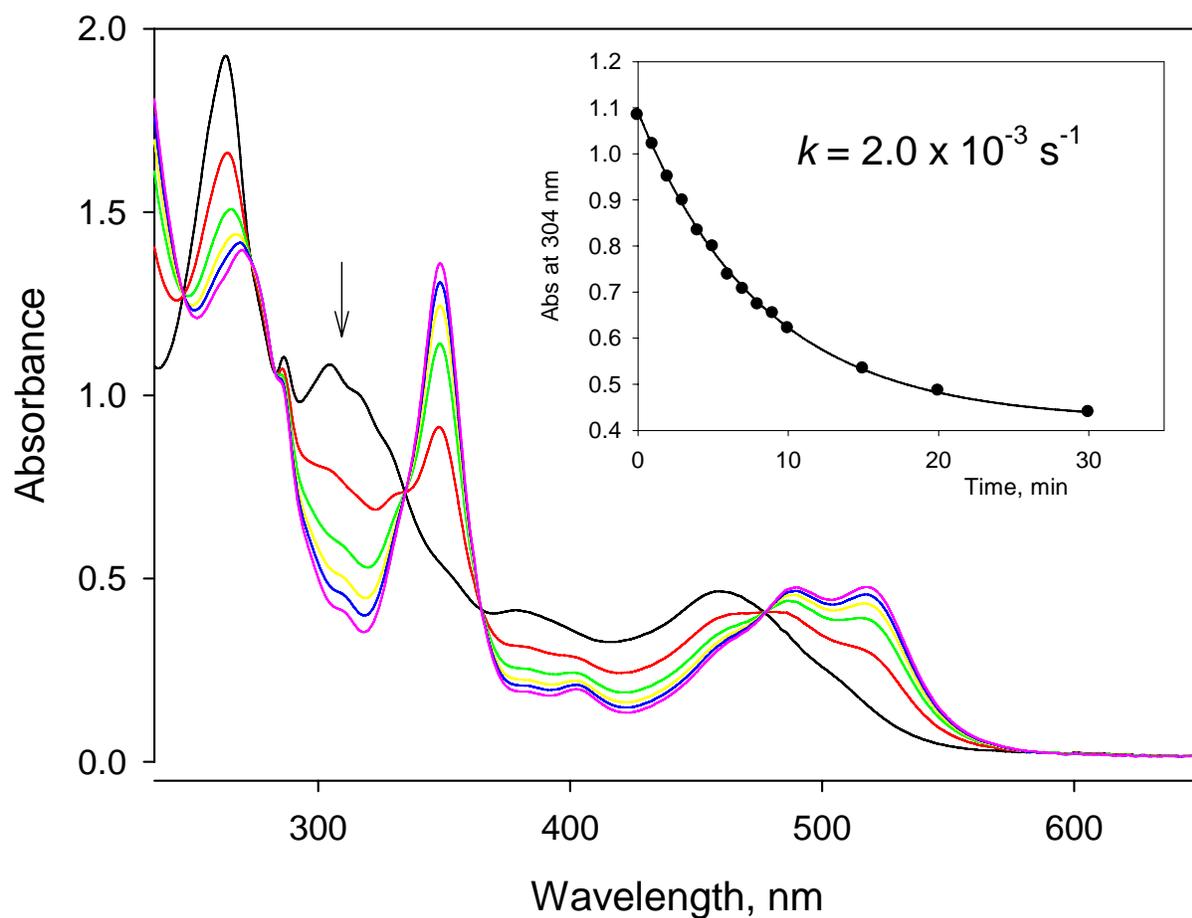


Figure S20. UV-visible spectra of the heterolytic photochemical cleavage of the Co-C bond of AdoCbl ($5.0 \times 10^{-5} \text{ mol dm}^{-3}$) in the presence of $7.5 \times 10^{-5} \text{ mol dm}^{-3}$ CB[7]. Inset: plot of the absorbance at 304 nm against time, fit with $k = 2.0 \times 10^{-3} \text{ s}^{-1}$.