

Electronic Supplementary Information (ESI) for

**Optically Active Uniform Potassium and Lithium Rare Earth Fluoride Nanocrystals Derived from Metal Trifluoroacetate Precursors**

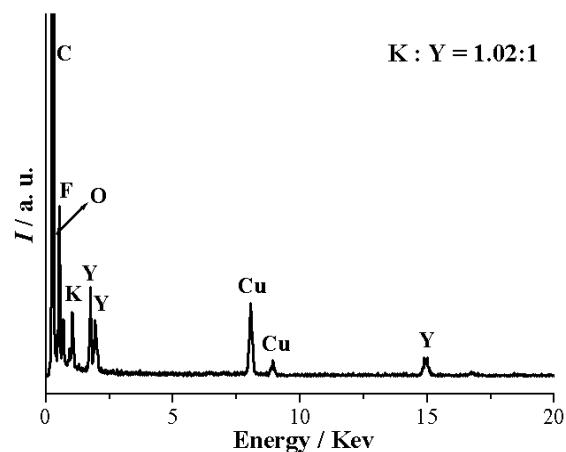
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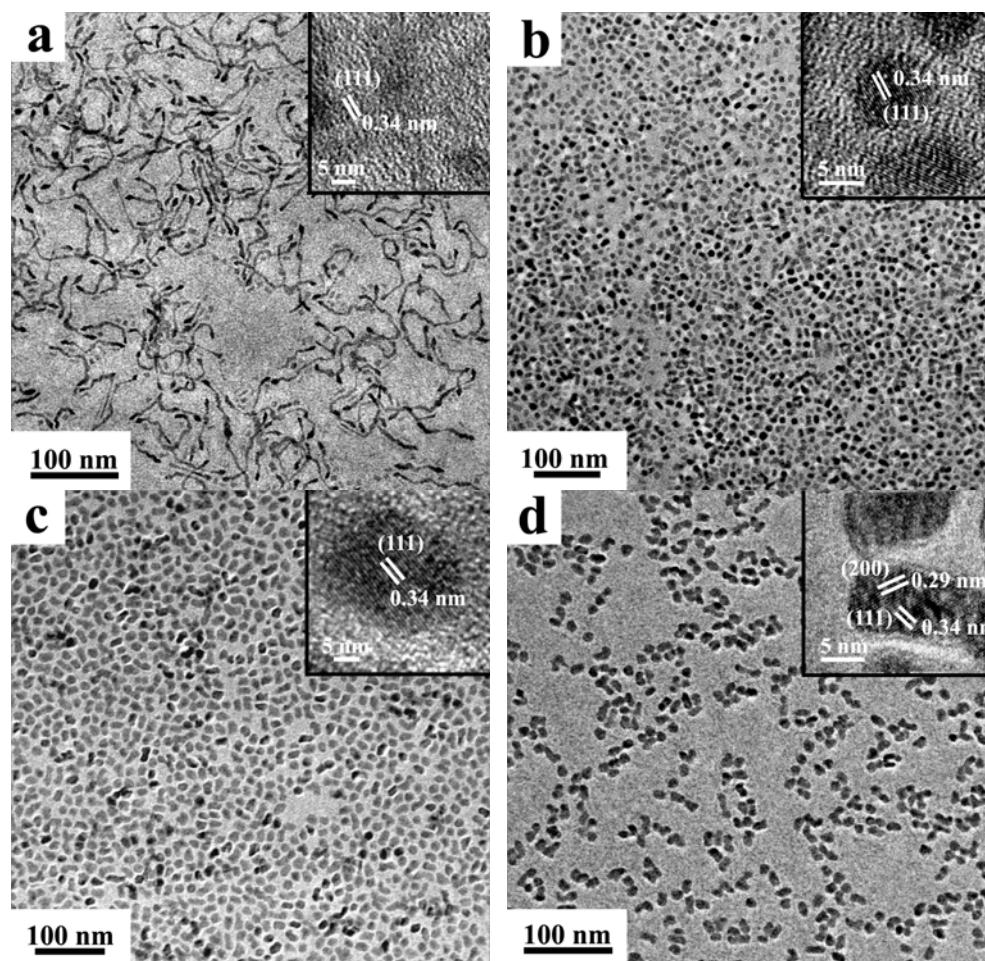
**Table S1.** Crystal Structure, Morphology, and Size of the As-Synthesized KREF<sub>4</sub> (RE = La to Gd, Y) and LiREF<sub>4</sub> (RE = Tb to Lu, Y) Nanocrystals *via* the Co-thermolysis of 1 mmol K(CF<sub>3</sub>COO) or Li(CF<sub>3</sub>COO) and 1 mmol RE(CF<sub>3</sub>COO)<sub>3</sub> in Oleic Acid (OA)/Oleylamine (OM)/1-Octadecene (ODE) (40 mmol).

	OA:OM:ODE	T [°C]	t [min]	structure	morphology	size [nm] <sup>[a]</sup>
KLaF <sub>4</sub>	1:1:2	280	60	cubic	wormlike wire	(8.2±0.5) × (23–100)
	1:1:2	300	60	cubic	cube	(9.4±0.9)
	1:1:2	300	60	cubic	polygon	(9.5±0.5)
KCeF <sub>4</sub>	1:1:2	260	60	cubic	wormlike wire	(7.8±0.4) × (30–120)
	1:1:2	280	60	cubic	cube	(10.4±1.1)
	1:1:2	300	60	cubic	cube (predominate) and rod	(15.4±1.3) (12.4±0.8) × (33.1±1.8)
KPrF <sub>4</sub>	1:1:2	320	60	cubic	polygon	(13.5±0.5)
	1:1:2	280	60	cubic	polyhedron	(18.2±3.6)
	1:1:2	280	60	cubic	polyhedron	(16.4±4.2)
KSmF <sub>4</sub>	1:1:2	280	60	cubic	polyhedron	(18.9±3.8)
KEuF <sub>4</sub>	1:1:2	280	60	cubic	polyhedron	(19.5±4.1)
KGdF <sub>4</sub>	1:1:2	300	60	cubic	polyhedron	(21.2±6.1)
KYF <sub>4</sub>	1:1:2	300	30	cubic	polyhedron	(13.2–25.9)
LiTbF <sub>4</sub>	1:0:1	330	30	tetragonal	rhombic plate	(27.1±2.4) × (25.8±4.3)
LiDyF <sub>4</sub>	1:0:1	330	30	tetragonal	rhombic plate	(55.1±2.1) × (49.9±3.7)
LiHoF <sub>4</sub>	1:0:1	280	60	tetragonal	rhombic plate	(53.1±3.4) × (49.8±2.3)
LiErF <sub>4</sub>	1:0:1	320	60	tetragonal	rhombic plate	(51.1±2.3) × (50.9±3.0)
LiErF <sub>4</sub>	1:0:1	300	60	cubic (ErOF)	polyhedron	(23.2±0.4)
LiTmF <sub>4</sub>	1:0:1	320	60	tetragonal	rhombic plate	(56.1±2.3) × (54.9±3.0)
LiYbF <sub>4</sub>	1:0:1	320	60	tetragonal	rhombic plate	(40.3±5.9) × (36.9±4.3)
LiLuF <sub>4</sub>	1:0:1	280	60	tetragonal	rhombic plate	(91.5±2.3) × (89.3±3.2)
LiLuF <sub>4</sub>	1:0:1	300	60	tetragonal	rhombic plate	(41.6±11.8) × (39.8±9.9)
LiYF <sub>4</sub>	1:0:1	330	8	tetragonal	rhombic plate	(29.2±3.9) × (23.2±3.0)

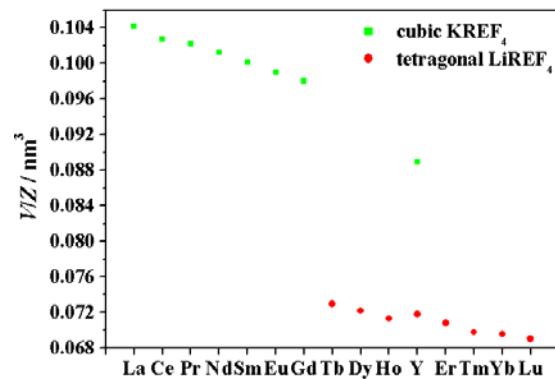
<sup>[a]</sup> The statistic standard deviation from at least 50 particles.



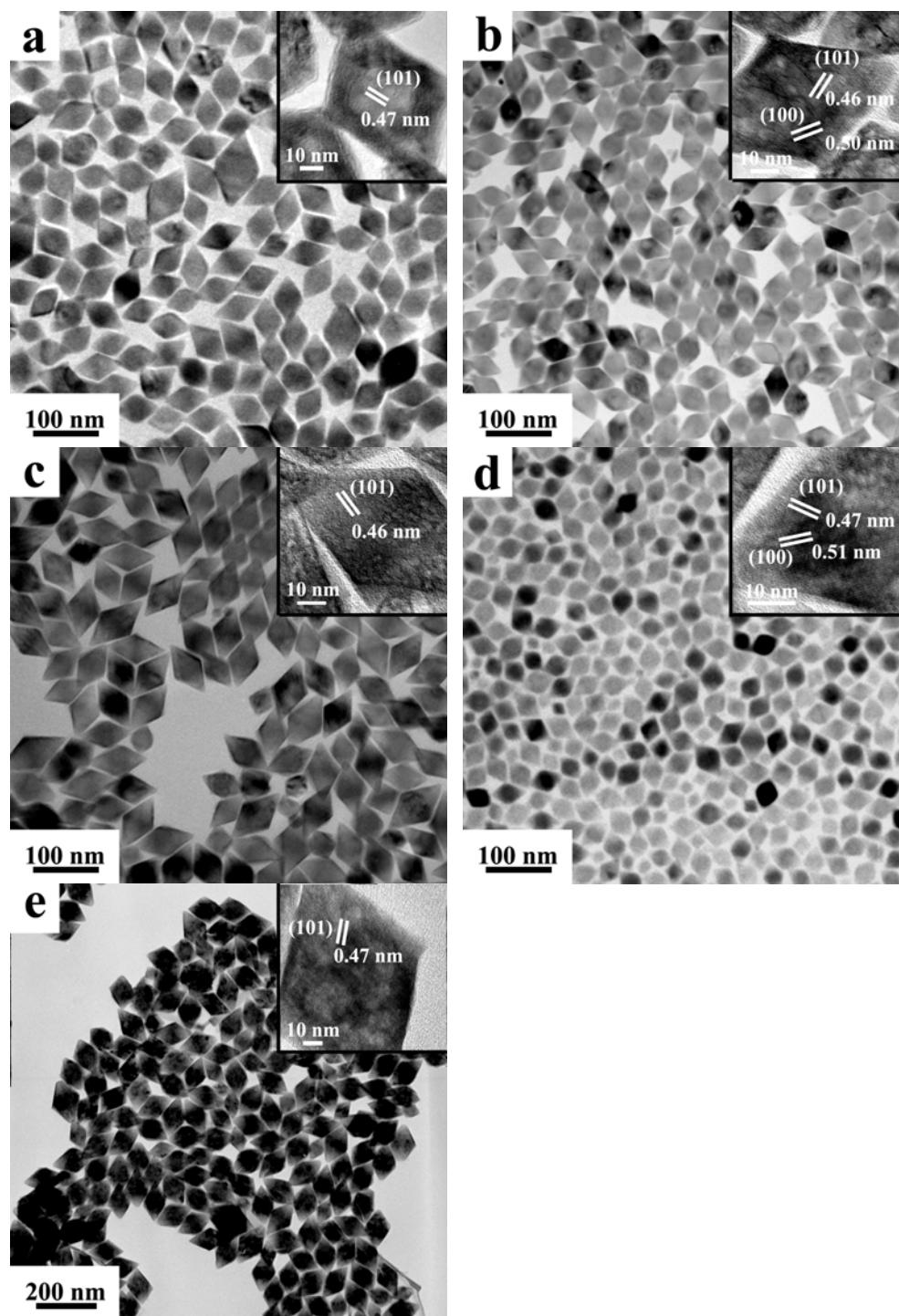
**Fig. S1.** Energy-dispersive X-ray analysis (EDAX) spectrum of  $\text{KYF}_4$  nanopolyhedra, from which the atomic ratio of K:Y was determined as 1.02:1.



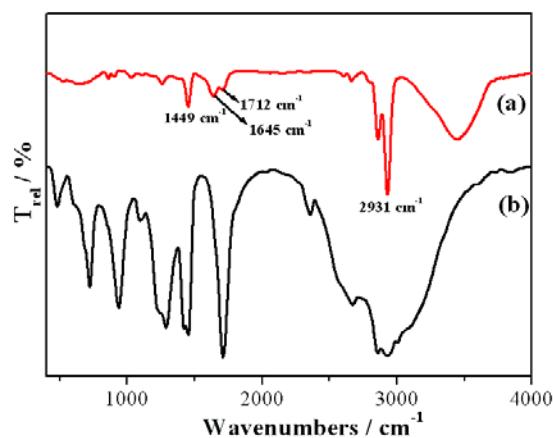
**Fig. S2.** TEM images of (a)  $\text{KCeF}_4$  wormlike nanowires, (b)  $\text{KNdF}_4$  (c)  $\text{KSmF}_4$ , and (d)  $\text{KGdF}_4$  nanopolyhedra.



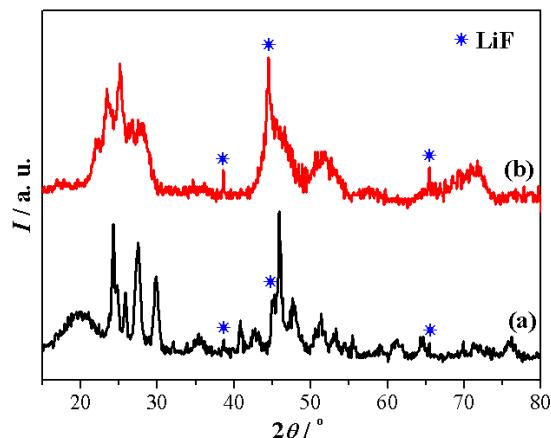
**Fig. S3.** Plots of the measured lattice volume over the number of KREF<sub>4</sub> and LiREF<sub>4</sub> molecules (Z) in one unit cell against the rare-earth series.



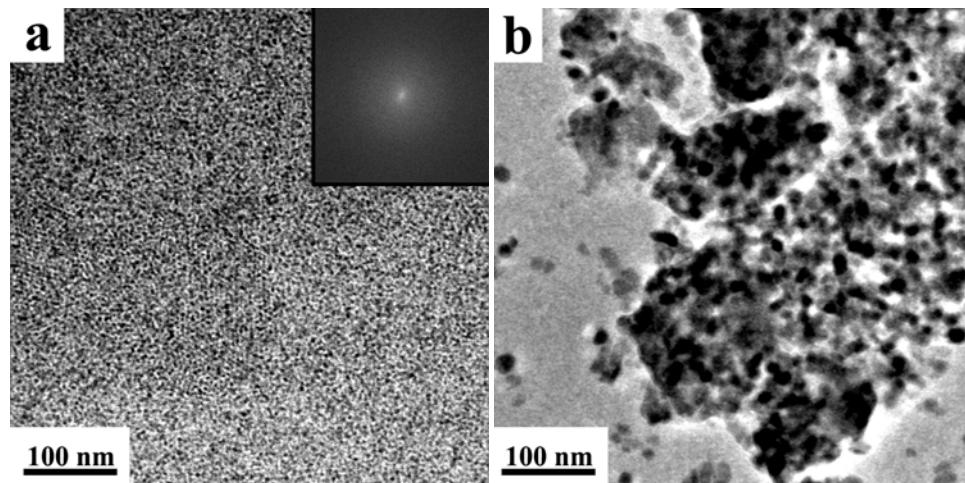
**Fig. S4.** TEM images of (a) LiDyF<sub>4</sub>, (b) LiHoF<sub>4</sub>, (c) LiTmF<sub>4</sub>, (d) LiYbF<sub>4</sub>, and (e) LiLuF<sub>4</sub> nanoplates.



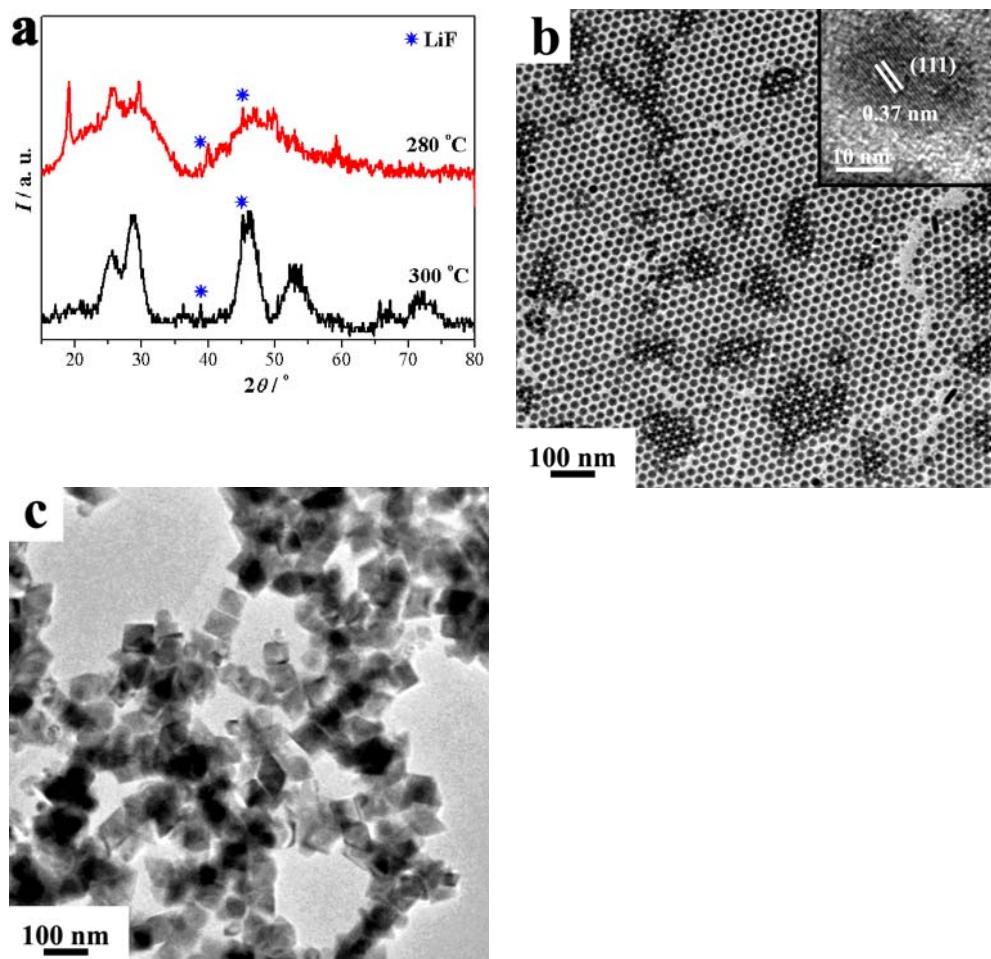
**Fig. S5.** FTIR spectra of (a) surfactants bound to LiYF<sub>4</sub> nanocrystals and (b) oleic acid.



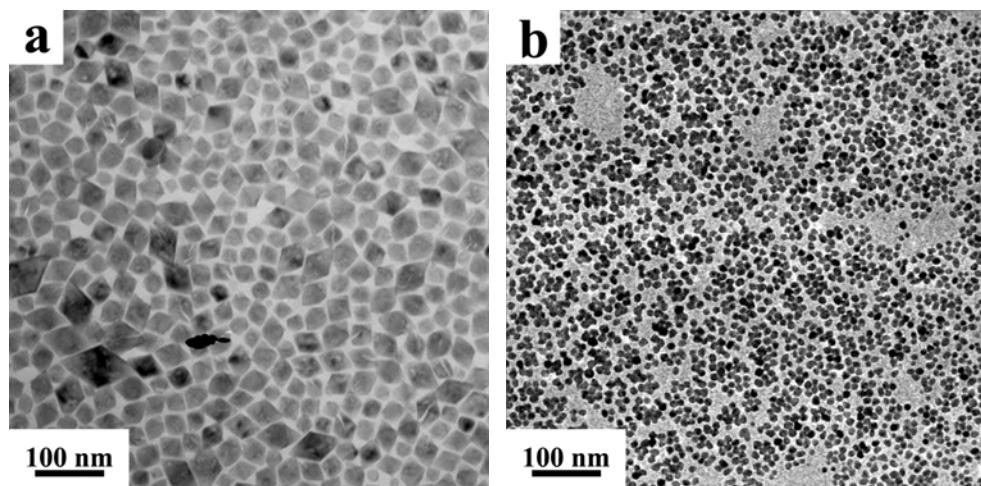
**Fig. S6.** XRD patterns of (a) LiLuF<sub>4</sub> and (b) LiYF<sub>4</sub> nanocrystals synthesized from co-thermolysis of 1 mmol Li(CF<sub>3</sub>COO)<sub>2</sub> and 1 mmol Lu(CF<sub>3</sub>COO)<sub>3</sub> or Y(CF<sub>3</sub>COO)<sub>3</sub> under OA/OM/ODE (4:1:5) at 280 °C for 60 min.



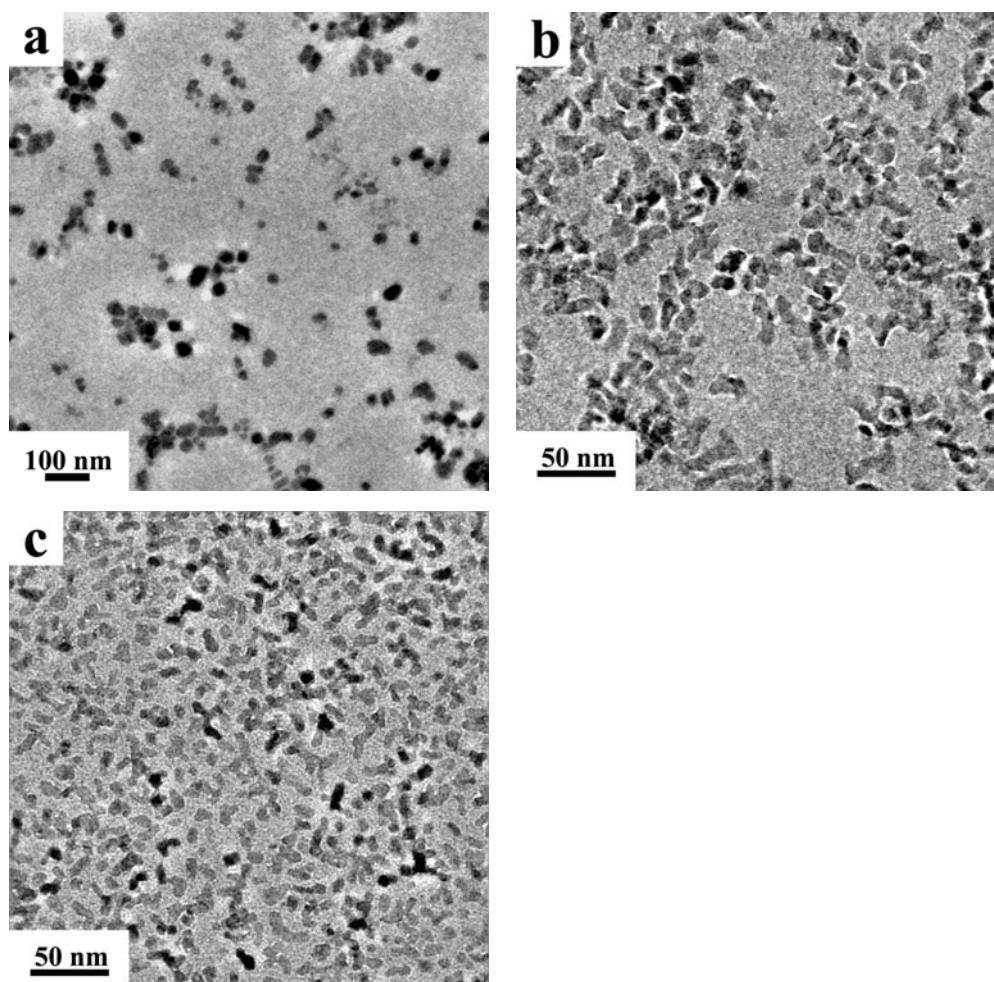
**Fig. S7.** TEM images of KYF<sub>4</sub> nanocrystals obtained from the co-thermolysis of 1 mmol K(CF<sub>3</sub>COO) and 1 mmol Y(CF<sub>3</sub>COO)<sub>3</sub> in OA/OM/ODE (1:1:2) for 30 min at different reaction temperatures: (a) 300 °C and (b) 330 °C.



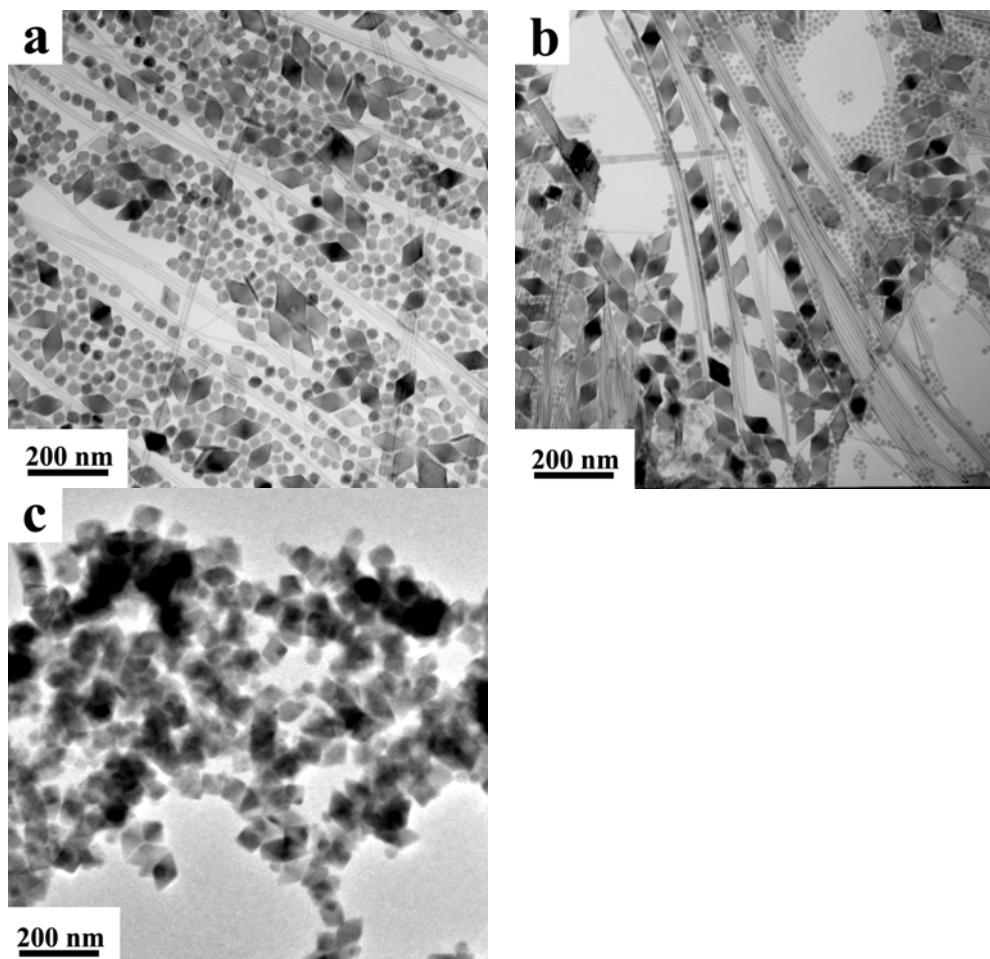
**Fig. S8.** XRD patterns of LiErF<sub>4</sub> nanocrystals obtained from the thermolysis of 1 mmol Li(CF<sub>3</sub>COO) and 1 mmol Er(CF<sub>3</sub>COO)<sub>3</sub> in OA/ODE (1:1) for 60 min at different reaction temperatures. TEM images of LiErF<sub>4</sub> nanocrystals obtained from the thermolysis of 1 mmol Li(CF<sub>3</sub>COO) and 1 mmol Er(CF<sub>3</sub>COO)<sub>3</sub> in OA/ODE (1:1) for 60 min at different reaction temperatures: (b) 300 °C and (c) 330 °C .



**Fig. S9.** TEM images of  $\text{LiLuF}_4$  nanocrystals obtained from the co-thermolysis of 1 mmol  $\text{Li}(\text{CF}_3\text{COO})$  and 1 mmol  $\text{Lu}(\text{CF}_3\text{COO})_3$  in OA/ODE (1:1) for 60 min at different reaction temperatures: (a) 300 °C and (b) 320 °C.



**Fig. S10.** TEM images of KYF<sub>4</sub> nanocrystals obtained from the thermolysis of 1 mmol K(CF<sub>3</sub>COO) and 1 mmol Y(CF<sub>3</sub>COO)<sub>3</sub> in OA/OM/ODE (1:1:2) at 300 °C for different reaction times: (a) 15 min, (b) 45 min, and (c) 60 min.



**Fig. S11.** TEM images of  $\text{LiErF}_4$  nanocrystals obtained from the thermolysis of 1 mmol  $\text{Li}(\text{CF}_3\text{COO})$  and 1 mmol  $\text{Er}(\text{CF}_3\text{COO})_3$  in OA/ODE (1:1) at 330 °C for different reaction times: (a) 15 min, (b) 30 min, and (c) 120 min.