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Supplementary Materials for **B910133K**



Fig. S1 Crystal structure of PLN (left) and space filling drawing of PLN (right).



Fig. S2a. Space filling drawing of complex 1.



Fig. S2b. Two dimensional network of complex 1 *via* nonclassical hydrogen bonds of C–H···O, view from *a* direction.



Fig. S3 1D chain structure of complex 2 formed by intermolecular hydrogen bonding.

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Fig.S4 *UV–vis* spectra of PLN (6.0×10^{-4} M) in the absence (----) and presence (—) of increasing amounts of DNA (2.1×10^{-3} M, 10µl DNA per scan).



Figure S5. UV-Vis absorption of complex **2** by titration of SDS with increasing concentration. (The concentration of complex **2** is 2×10^{-5} mol/L, while the stock solution of SDS for titration was prepared as 2×10^{-2} mol/L. The dashed line and the solid lines represent the absorption band of complex **2** in the absence and presence of SDS, respectively.)



Fig. S6 Emission spectra of PLN(1.0×10^{-4} M, excited at 282 nm) in the absence (----) and presence (----) of increasing amounts of DNA(2.1×10^{-3} M, 50µl per scan).

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Fig. S7 Emission spectra of DNA–EB system (3 ml solution, [DNA]/[EB] ratio of 6:1, excited at 350 nm) in the absence (----) and presence (—) of increasing amounts of PLN. $PLN(6.0 \times 10^{-3} \text{ M})$.