

Supplementary material

Bis(2-pyridyl)ditelluride, bis(3-methyl-2-pyridyl)ditelluride and their tellurolate complexes of zinc, cadmium, mercury: Synthesis, characterization and their conversion to metal telluride nanoparticles

G. Kedarnath, Vimal K. Jain,* Amey Wadawale and Gautam K. Dey

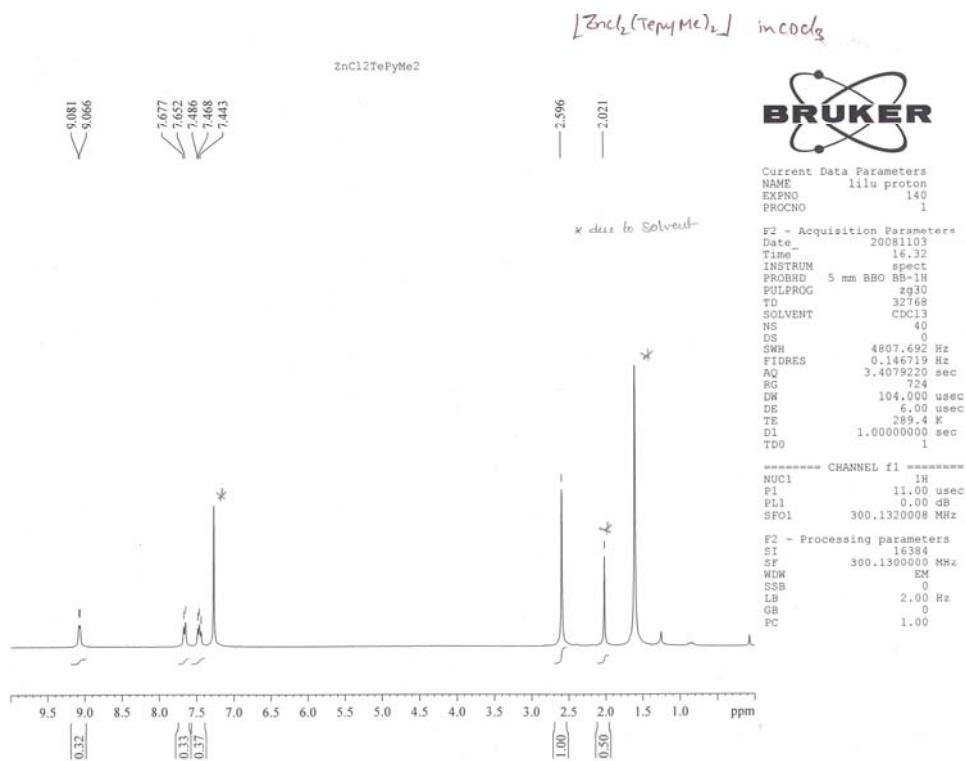


Fig. 1 ^1H NMR spectrum of $[\text{ZnCl}_2\{\text{Te}_2(\text{pyMe})_2\}]$.

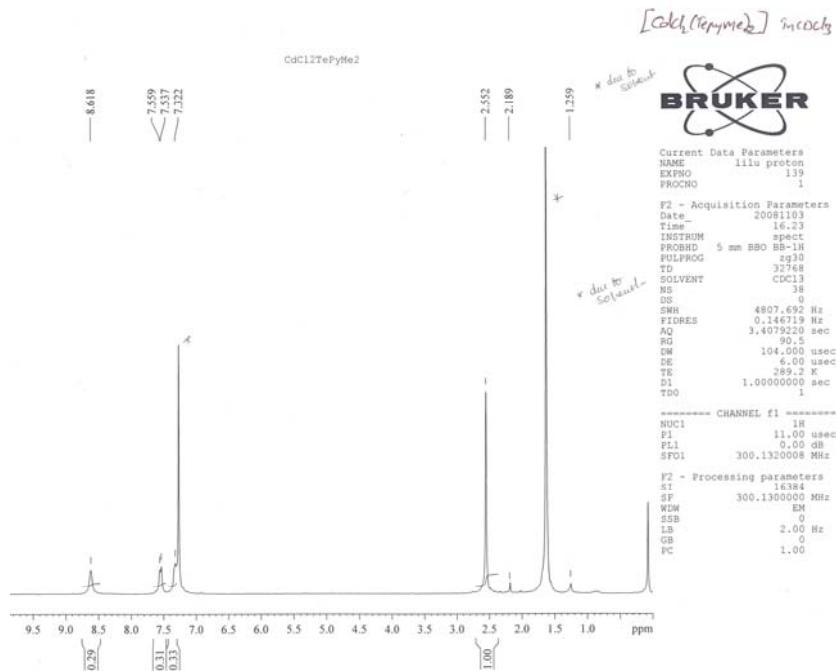


Fig. 2 ¹H NMR spectrum of [CdCl₂{Te₂(pyMe)₂}].

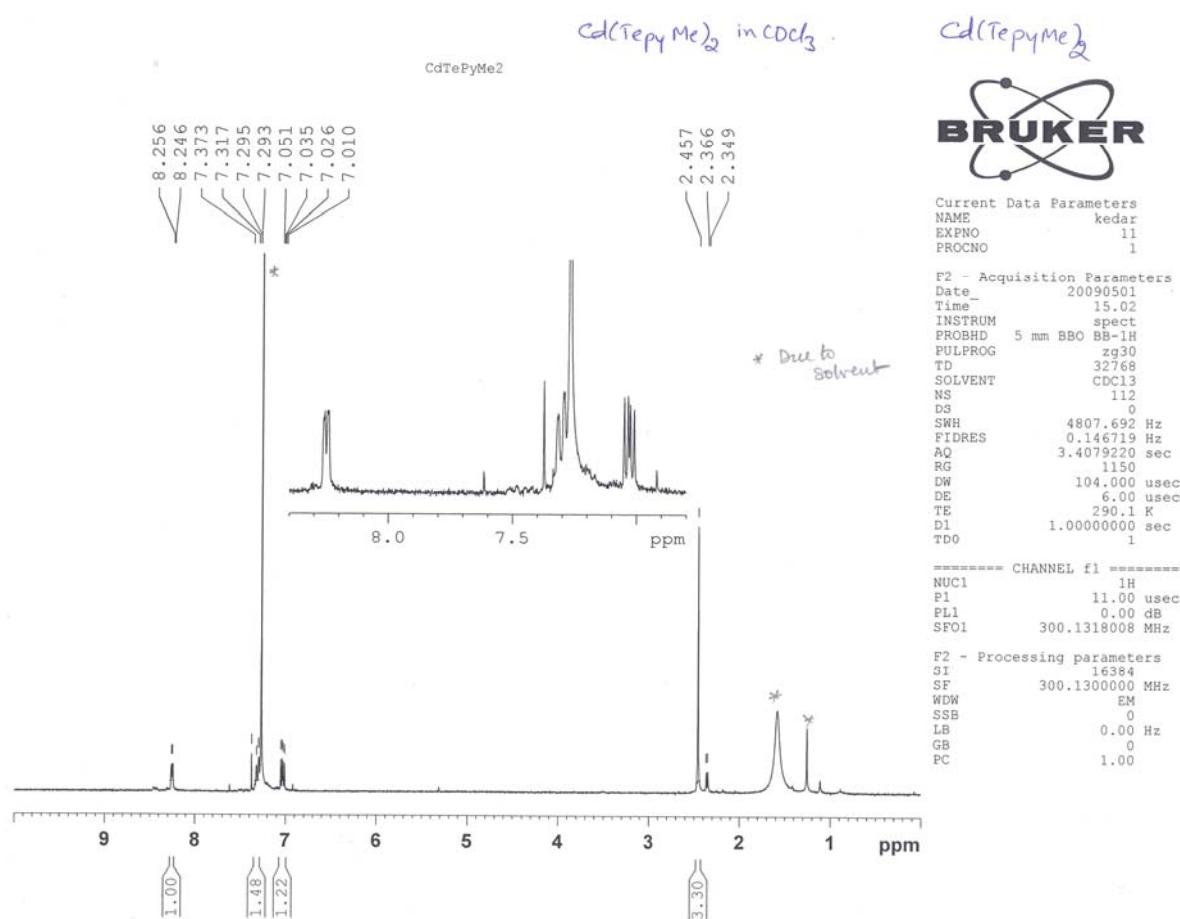


Fig. 3 ^1H NMR spectrum of $[\text{Cd}(\text{TepyMe})_2]$.

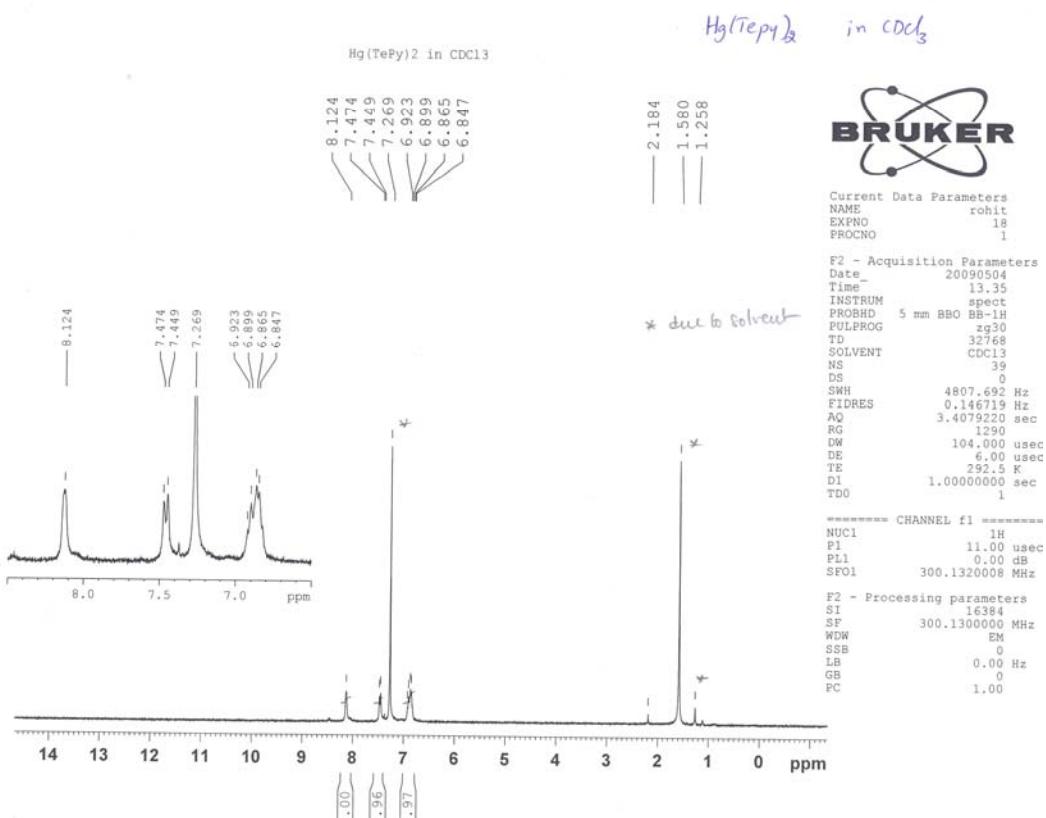


Fig. 4 ¹H NMR spectrum of [Hg(Tepy)₂].

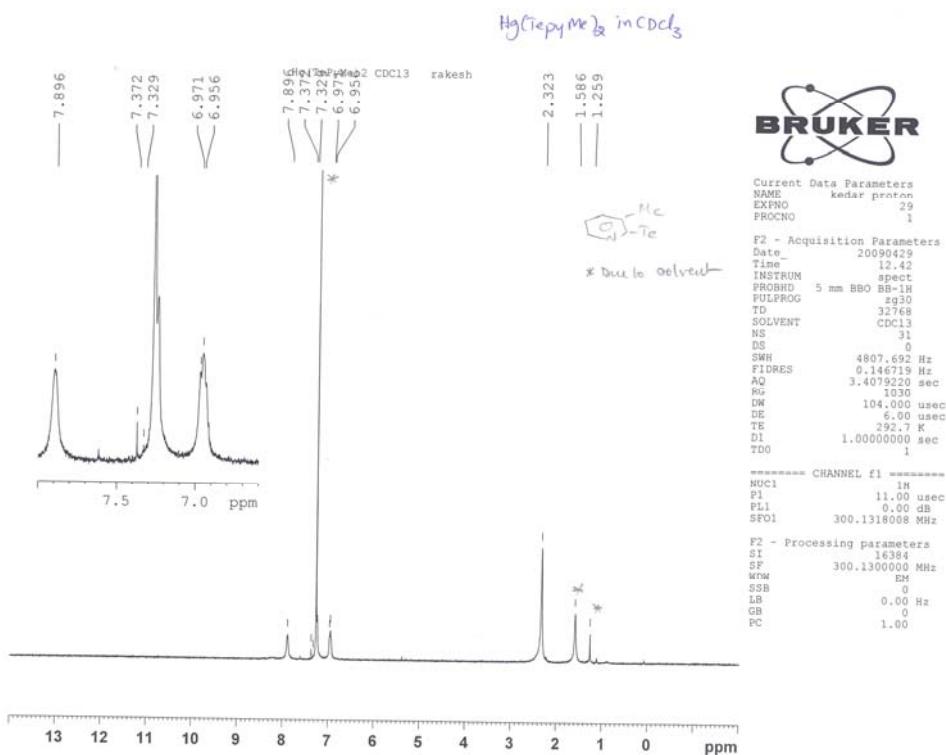


Fig. 5 ¹H NMR spectrum of [Hg(TepyMe)₂].

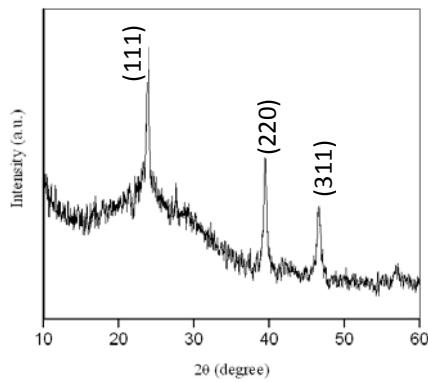


Fig. 6 CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ at 350°C in a furnace for 1h.

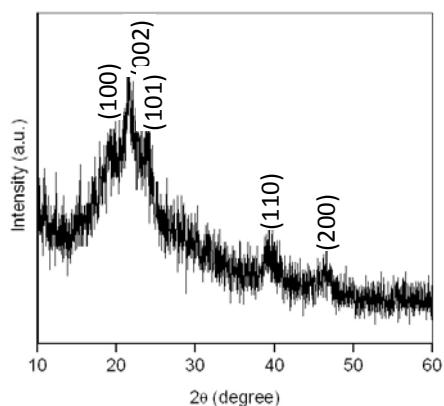


Fig. 7 CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ in HDA at $160\text{ }^\circ\text{C}$ for 20 min.

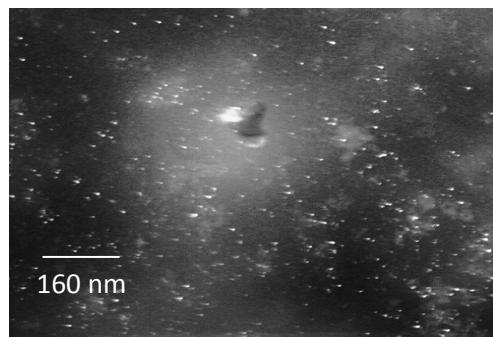


Fig. 8 TEM image of CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ in HDA at 160 °C for 20 min.

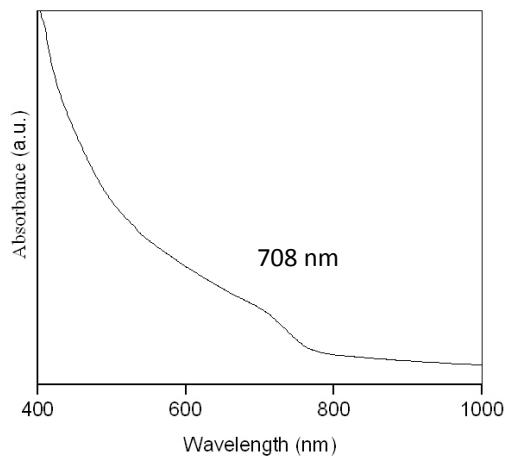


Fig. 9 Absorption spectrum of CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ in HDA at 160 °C for 20 min.

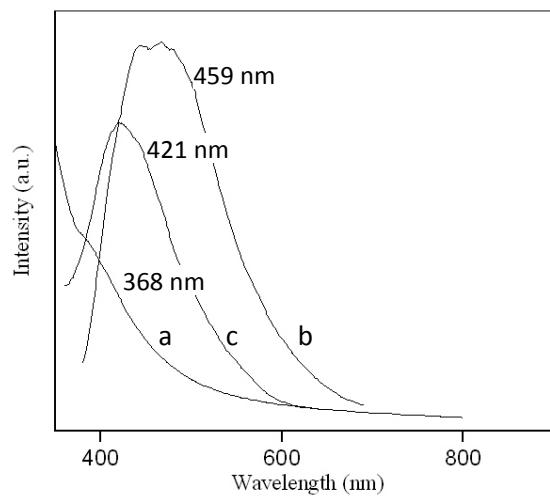


Fig. 10 a) Absorption, b) emission spectra of CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ in TOPO (2 g) at 160 °C for 20 min (**experiment 2**) and c) emission spectrum of CdTe nanoparticles recapped with TGA.

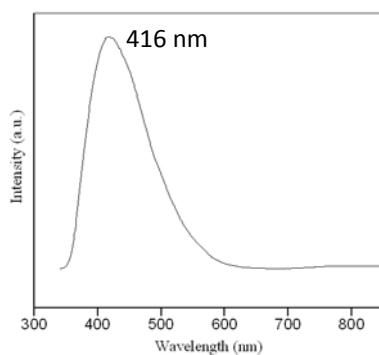


Fig. 11 Emission spectrum of CdTe nanoparticles obtained by the pyrolysis of $[\text{Cd}(\text{Tepy})_2]$ in HDA/TOPO at 160 °C for 20 min (**experiment 4**).

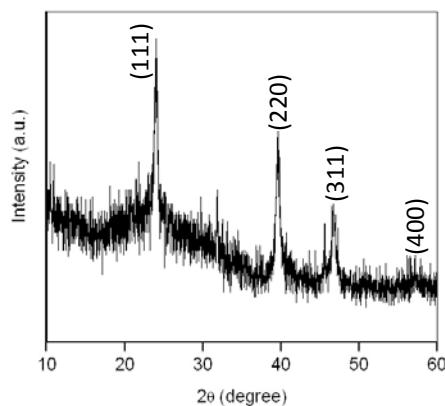


Fig. 12 XRD pattern of HgTe obtained by the pyrolysis of $[\text{Hg}(\text{pyTe})_2]$ in a furnace at 175°C for 1 h.

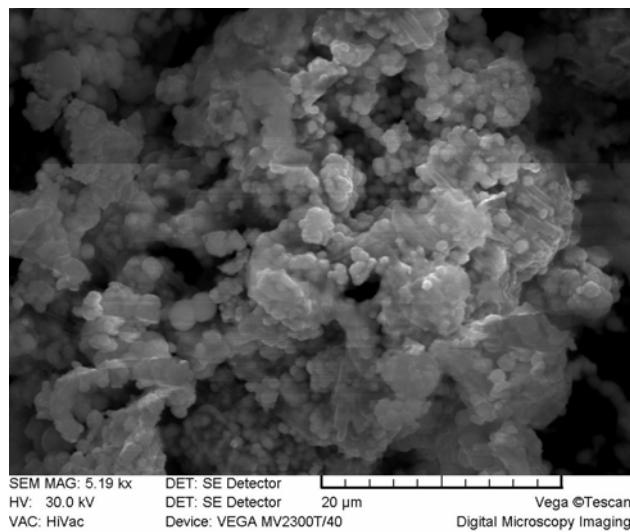


Fig. 13 SEM picture of HgTe obtained by the pyrolysis of $[\text{Hg}(\text{pyTe})_2]$ in a furnace at 175°C for 1 h.

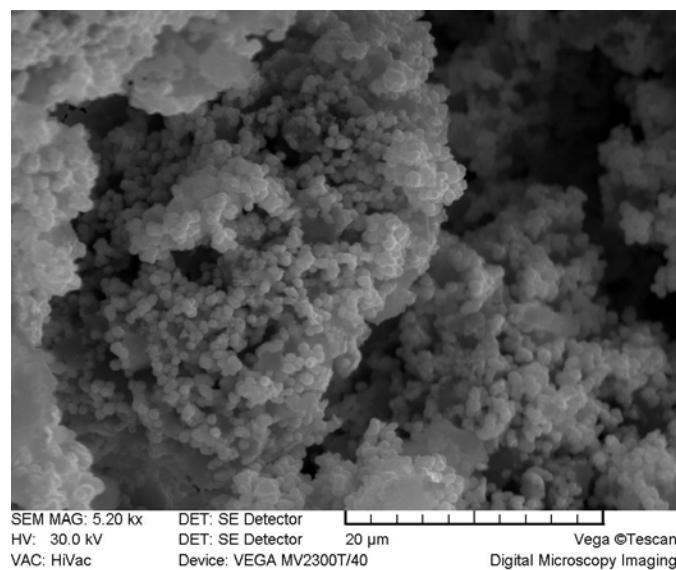


Fig. 14 SEM picture of HgTe obtained by the pyrolysis of [Hg(TepyMe)₂] in a furnace at 250 °C for 1 h.

Crystal information (.cif) for [ZnCl₂(TepyMe)₂].CH₃CN (1b)

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$F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

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N2 C10 H11 117.4 . . ?
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C9 C8 C7 C12 178(2) . . . . ?
N2 Zn1 N1 C1 29.9(19) . . . . ?
Cl1 Zn1 N1 C1 149.0(15) . . . . ?
Cl2 Zn1 N1 C1 -88.4(16) . . . . ?
N2 Zn1 N1 C5 -156.3(14) . . . . ?
Cl1 Zn1 N1 C5 -37.2(15) . . . . ?
Cl2 Zn1 N1 C5 85.3(14) . . . . ?
C1 N1 C5 C4 4(3) . . . . ?
Zn1 N1 C5 C4 -170(2) . . . . ?
N1 C5 C4 C3 -5(4) . . . . ?
C2 C3 C4 C5 3(4) . . . . ?
C4 C3 C2 C1 0(4) . . . . ?
C4 C3 C2 Cl1 -173(3) . . . . ?
C6 Te2 Te1 C1 109.4(8) . . . . ?
Cl1 C2 C1 N1 172(2) . . . . ?
C3 C2 C1 N1 -1(3) . . . . ?
Cl1 C2 C1 Te1 -6(3) . . . . ?
C3 C2 C1 Te1 -179.0(18) . . . . ?
C5 N1 C1 C2 -1(3) . . . . ?
Zn1 N1 C1 C2 173.1(16) . . . . ?
C5 N1 C1 Te1 177.4(14) . . . . ?
Zn1 N1 C1 Te1 -9(2) . . . . ?
Te2 Te1 C1 C2 122.2(18) . . . . ?
Te2 Te1 C1 N1 -56.0(15) . . . . ?
N1 Zn1 N2 C10 -159.6(13) . . . . ?
Cl1 Zn1 N2 C10 81.2(14) . . . . ?
Cl2 Zn1 N2 C10 -42.8(14) . . . . ?
N1 Zn1 N2 C6 25(2) . . . . ?
Cl1 Zn1 N2 C6 -94.0(18) . . . . ?
Cl2 Zn1 N2 C6 142.0(18) . . . . ?
C10 N2 C6 C7 1(3) . . . . ?
Zn1 N2 C6 C7 175.9(19) . . . . ?
C10 N2 C6 Te2 -179.7(15) . . . . ?
Zn1 N2 C6 Te2 -4(3) . . . . ?
C8 C7 C6 N2 -1(4) . . . . ?
C12 C7 C6 N2 -180(2) . . . . ?
C8 C7 C6 Te2 179.5(17) . . . . ?
C12 C7 C6 Te2 1(3) . . . . ?
Te1 Te2 C6 N2 -58.0(17) . . . . ?
Te1 Te2 C6 C7 122(2) . . . . ?
C7 C8 C9 C10 2(3) . . . . ?
C6 N2 C10 C9 1(3) . . . . ?
Zn1 N2 C10 C9 -174.8(15) . . . . ?
C8 C9 C10 N2 -3(3) . . . . ?

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_refine_diff_density_rms 0.247

Crystal information (.cif) for [Cd(Tepy)₂(tmada)] (3)

data_gk-11

_audit_creation_method SHELXL-97
_chemical_name_systematic
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_chemical_name_common ?
_chemical_melting_point ?
_chemical_formula_moiety ?
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'C16 H24 Cd N4 Te2'
_chemical_formula_weight 639.99

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_atom_type_description
_atom_type_scat_dispersion_real
_atom_type_scat_dispersion_imag
_atom_type_scat_source
'C' 'C' 0.0033 0.0016
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'H' 'H' 0.0000 0.0000
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'N' 'N' 0.0061 0.0033
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Te' 'Te' -0.5308 1.6751
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Cd' 'Cd' -0.8075 1.2024
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'

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_symmetry_space_group_name_Hall '-C 2yc'
_symmetry_Int_Tables_number 15

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'-x, y, -z+1/2'
'x+1/2, y+1/2, z'
'-x+1/2, y+1/2, -z+1/2'
'-x, -y, -z'
'x, -y, z-1/2'
'-x+1/2, -y+1/2, -z'
'x+1/2, -y+1/2, z-1/2'

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_cell_length_b 7.3572(11)
_cell_length_c 14.250(3)
_cell_angle_alpha 90.00
_cell_angle_beta 105.382(16)
_cell_angle_gamma 90.00

_cell_volume	2112.6 (7)
_cell_formula_units_Z	4
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_exptl_crystal_density_diffrn	2.012
_exptl_crystal_density_method	'not measured'
_exptl_crystal_F_000	1200
_exptl_absorpt_coefficient_mu	3.745
_exptl_absorpt_correction_type	psi-scan
_exptl_absorpt_process_details	'(North, Phillips & Mathews, 1968)'
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_diffrn_radiation_monochromator	graphite
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_diffrn_measurement_method	\w
_diffrn_detector_area_resol_mean	?
_diffrn_standsards_number	3
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_diffrn_standsards_decay_%	-5.93
_diffrn_reflns_number	3266
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_diffrn_reflns_theta_min	2.95
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_reflns_number_gt	1920
_reflns_threshold_expression	>2sigma(I)
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_computing_cell_refinement	'WINAFC'
computing data reduction	'CRYSTAL STRUCTURE'

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_computing_molecular_graphics     'ORTEP-3 FOR WINDOWS'
_computing_publication_material   'WINGX 1.70.01'

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    Refinement of F^2^ against ALL reflections. The weighted R-factor wR
and
    goodness of fit S are based on F^2^, conventional R-factors R are
based
    on F, with F set to zero for negative F^2^. The threshold expression
of
    F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc.
and is
    not relevant to the choice of reflections for refinement. R-factors
based
    on F^2^ are statistically about twice as large as those based on F,
and R-
    factors based on ALL data will be even larger.
;

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_refine_ls_weighting_details
    'calc w=1/[s^2^(Fo^2^)+(0.0675P)^2^+5.1667P] where
P=(Fo^2^+2Fc^2^)/3'
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_atom_sites_solution_secondary    difmap
_atom_sites_solution_hydrogens    geom
_refine_ls_hydrogen_treatment     mixed
_refine_ls_extinction_method      SHELXL
_refine_ls_extinction_coef        0.0022(2)
_refine_ls_extinction_expression
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_refine_ls_number_restraints      0
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_refine_ls_wR_factor_ref          0.1162
_refine_ls_wR_factor_gt            0.1070
_refine_ls_goodness_of_fit_ref    1.027
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_refine_ls_shift/su_max           0.000
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Cd1 Cd 0.0000 0.20270(7) 0.2500 0.0415(2) Uani 1 2 d S . .
N1 N 0.1290(3) 0.0419(7) 0.3348(4) 0.0539(12) Uani 1 1 d . .
N2 N 0.0614(3) 0.4632(7) 0.3348(4) 0.0578(14) Uani 1 1 d . .
C1 C 0.1435(3) 0.0097(8) 0.2512(4) 0.0445(12) Uani 1 1 d . .
C2 C 0.2046(4) -0.0617(11) 0.2492(6) 0.067(2) Uani 1 1 d . .
C5 C 0.1749(4) 0.0070(13) 0.4178(6) 0.074(2) Uani 1 1 d . .
C4 C 0.2374(4) -0.0573(13) 0.4205(7) 0.082(3) Uani 1 1 d . .
C7 C 0.1242(3) 0.4943(11) 0.3083(6) 0.0688(19) Uani 1 1 d . .
H7A H 0.1150 0.5107 0.2391 0.103 Uiso 1 1 calc R . .
H7B H 0.1528 0.3913 0.3276 0.103 Uiso 1 1 calc R . .
H7C H 0.1455 0.6011 0.3408 0.103 Uiso 1 1 calc R . .
C3 C 0.2518(4) -0.0932(14) 0.3351(8) 0.085(3) Uani 1 1 d . .
C8 C 0.0166(4) 0.6193(11) 0.3029(8) 0.094(3) Uani 1 1 d . .
C6 C 0.0743(5) 0.4382(13) 0.4396(7) 0.089(3) Uani 1 1 d . .
H6A H 0.1035 0.3363 0.4596 0.133 Uiso 1 1 calc R . .
H6B H 0.0333 0.4160 0.4558 0.133 Uiso 1 1 calc R . .
H6C H 0.0948 0.5457 0.4725 0.133 Uiso 1 1 calc R . .
H1 H 0.211(4) -0.079(11) 0.197(6) 0.07(3) Uiso 1 1 d . .
H3 H 0.263(5) -0.067(12) 0.479(8) 0.11(4) Uiso 1 1 d . .
H8B H -0.020(4) 0.597(9) 0.337(5) 0.06(2) Uiso 1 1 d . .
H4 H 0.163(4) 0.045(10) 0.480(6) 0.08(2) Uiso 1 1 d . .
H8A H 0.044(4) 0.745(12) 0.326(6) 0.09(2) Uiso 1 1 d . .
H2 H 0.286(5) -0.143(14) 0.334(7) 0.10(3) Uiso 1 1 d . .

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_atom_site_aniso_U_13
_atom_site_aniso_U_12
Te1 0.0441(3) 0.0541(3) 0.0495(3) -0.00150(15) 0.01645(17) 0.00397(16)
Cd1 0.0394(3) 0.0350(3) 0.0544(3) 0.000 0.0198(2) 0.000
N1 0.052(3) 0.060(3) 0.053(3) 0.003(2) 0.021(2) 0.003(3)
N2 0.045(3) 0.048(3) 0.084(4) -0.023(3) 0.023(3) -0.008(2)
C1 0.039(3) 0.039(3) 0.059(3) 0.005(2) 0.019(2) -0.003(2)
C2 0.047(4) 0.082(5) 0.077(5) 0.013(4) 0.025(3) 0.018(3)
C5 0.078(5) 0.083(5) 0.060(4) 0.009(4) 0.017(4) -0.001(5)
C4 0.061(5) 0.088(6) 0.083(6) 0.032(5) -0.006(4) -0.005(4)
C7 0.050(4) 0.051(4) 0.108(6) -0.012(4) 0.025(4) -0.015(3)
C3 0.038(4) 0.092(6) 0.124(8) 0.037(5) 0.019(4) 0.016(4)
C8 0.061(5) 0.041(4) 0.179(10) -0.034(5) 0.027(6) 0.003(3)
C6 0.078(6) 0.099(7) 0.094(6) -0.050(5) 0.030(5) -0.017(5)

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All esds (except the esd in the dihedral angle between two l.s.
planes)
are estimated using the full covariance matrix. The cell esds are
taken
into account individually in the estimation of esds in distances,
angles
```

and torsion angles; correlations between esds in cell parameters are only

used when they are defined by crystal symmetry. An approximate (isotropic)

treatment of cell esds is used for estimating esds involving l.s. planes.

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Te1 Cd1 2.7739(6) . ?
Cd1 N2 2.441(5) . ?
Cd1 N2 2.441(5) 2 ?
Cd1 Te1 2.7739(6) 2 ?
N1 C1 1.326(7) . ?
N1 C5 1.336(10) . ?
N2 C6 1.457(11) . ?
N2 C8 1.475(10) . ?
N2 C7 1.477(8) . ?
C1 C2 1.389(8) . ?
C2 C3 1.372(12) . ?
C2 H1 0.80(7) . ?
C5 C4 1.379(12) . ?
C5 H4 1.02(8) . ?
C4 C3 1.355(14) . ?
C4 H3 0.86(11) . ?
C7 H7A 0.9600 . ?
C7 H7B 0.9600 . ?
C7 H7C 0.9600 . ?
C3 H2 0.81(10) . ?
C8 C8 1.48(2) 2 ?
C8 H8B 1.02(6) . ?
C8 H8A 1.09(9) . ?
C6 H6A 0.9600 . ?
C6 H6B 0.9600 . ?
C6 H6C 0.9600 . ?
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N2 Cd1 N2 76.5(3) . 2 ?
N2 Cd1 Te1 104.67(13) . 2 ?
N2 Cd1 Te1 108.04(12) 2 2 ?
N2 Cd1 Te1 108.04(12) . . ?
N2 Cd1 Te1 104.67(13) 2 . ?
Te1 Cd1 Te1 138.00(3) 2 . ?
C1 N1 C5 118.7(6) . . ?
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C6 N2 C8 110.1(7) . . ?
C6 N2 C7 110.3(6) . . ?
C8 N2 C7 109.6(7) . . ?
C6 N2 Cd1 109.8(4) . . ?
C8 N2 Cd1 104.6(5) . . ?
C7 N2 Cd1 112.4(4) . . ?
N1 C1 C2 121.1(6) . . ?
N1 C1 Te1 116.6(4) . . ?
C2 C1 Te1 122.2(5) . . ?
C3 C2 C1 119.5(8) . . ?
C3 C2 H1 123(6) . . ?
C1 C2 H1 118(6) . . ?
N1 C5 C4 122.8(8) . . ?
N1 C5 H4 115(5) . . ?
C4 C5 H4 121(5) . . ?
C3 C4 C5 118.5(8) . . ?
C3 C4 H3 129(7) . . ?
C5 C4 H3 113(7) . . ?
N2 C7 H7A 109.5 . . ?
N2 C7 H7B 109.5 . . ?
H7A C7 H7B 109.5 . . ?
N2 C7 H7C 109.5 . . ?
H7A C7 H7C 109.5 . . ?
H7B C7 H7C 109.5 . . ?
C4 C3 C2 119.3(8) . . ?
C4 C3 H2 121(7) . . ?
C2 C3 H2 119(7) . . ?
N2 C8 C8 113.6(7) . 2 ?
N2 C8 H8B 103(4) . . ?
C8 C8 H8B 106(4) 2 . ?
N2 C8 H8A 109(4) . . ?
C8 C8 H8A 112(4) 2 . ?
H8B C8 H8A 113(6) . . ?
N2 C6 H6A 109.5 . . ?
N2 C6 H6B 109.5 . . ?
H6A C6 H6B 109.5 . . ?
N2 C6 H6C 109.5 . . ?
H6A C6 H6C 109.5 . . ?
H6B C6 H6C 109.5 . . ?

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C1 Te1 Cd1 N2 64.2(2) ?
C1 Te1 Cd1 N2 144.5(2) . . . 2 ?
C1 Te1 Cd1 Te1 -75.18(15) . . . 2 ?
N2 Cd1 N2 C6 132.3(6) 2 . . . ?
Te1 Cd1 N2 C6 26.9(5) 2 . . . ?
Te1 Cd1 N2 C6 -126.4(5) ?
N2 Cd1 N2 C8 14.2(5) 2 . . . ?

Te1 Cd1 N2 C8 -91.2(6) 2 . . . ?
Te1 Cd1 N2 C8 115.5(5) ?
N2 Cd1 N2 C7 -104.6(6) 2 ?
Te1 Cd1 N2 C7 149.9(5) 2 ?
Te1 Cd1 N2 C7 -3.3(5) ?
C5 N1 C1 C2 1.0(10) ?
C5 N1 C1 Te1 179.9(6) ?
Cd1 Te1 C1 N1 4.9(4) ?
Cd1 Te1 C1 C2 -176.2(6) ?
N1 C1 C2 C3 -2.4(11) ?
Te1 C1 C2 C3 178.8(6) ?
C1 N1 C5 C4 1.5(12) ?
N1 C5 C4 C3 -2.6(14) ?
C5 C4 C3 C2 1.1(14) ?
C1 C2 C3 C4 1.3(13) ?
C6 N2 C8 C8 -161.1(9) . . . 2 ?
C7 N2 C8 C8 77.5(10) . . . 2 ?
Cd1 N2 C8 C8 -43.2(10) . . . 2 ?

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