## **Equations Supplementary Information**

 $[Pd(PP_{3})_{2}]Cl_{2} + AgNO_{3} \xrightarrow{DMSO-d_{6}+CD_{3}OD} [PdAg(NO_{3})(PP_{3})_{2}]Cl_{2} \quad (S1)$   $g \qquad I$   $3 [Pt(PP_{3})_{2}]Cl_{2} + 3 AgNO_{3} \xrightarrow{DMSO-d_{6}+CD_{3}OD} \rightarrow 2 [Pt(PP_{3})_{2}]Cl_{2} + [PtAg(NO_{3})(PP_{3})_{2}](NO_{3})_{2} + 2 AgCl_{4}$   $(S2) \qquad I2 \qquad I$ 

$$2 [M(PP_3)_2]Cl_2 + 4 AgNO_3 \xrightarrow{DMSO-d_6+CD_3OD} \\ 9/12 \longrightarrow [MAgCl(PP_3)_2]Cl_2 + [MAg_2(NO_3)_4(PP_3)_2] + AgCl \downarrow (S3) \\ I III$$

$$2[Pd(PP_{3})_{2}]Cl_{2} + 6AgNO_{3} \xrightarrow{DMSO-d_{6}+CD_{3}OD} \\ \rightarrow [PdAg_{2}(NO_{3})_{2}(PP_{3})_{2}]Cl_{2} + [PdAg_{2}(NO_{3})_{4}(PP_{3})_{2}] + 2 AgCl \downarrow$$
(S4)
$$II III$$

$$2[Pd(PP_{3})_{2}]Cl_{2} + 8AgNO_{3} \xrightarrow{DMSO-d_{6}+CD_{3}OD}$$
9
$$\rightarrow [PdAg_{2}(NO_{3})_{2}(PP_{3})_{2}](NO_{3})_{2}+[PdAg_{2}(NO_{3})_{4}(PP_{3})_{2}]+4AgCl\downarrow$$
(S5)
II III

$$[Pt(PP_3)_2]Cl_2 + 3 \text{ AgNO}_3 \xrightarrow{\text{DMSO-d}_6 + CD_3 \text{OD}} [PtAg_2Cl(NO_3)_3 (PP_3)_2] + AgCl\downarrow (S6)$$
12
III

$$[Pt(PP_3)_2]Cl_2 + 4 \text{ AgNO}_3 \xrightarrow{\text{DMSO-d}_6 + CD_3OD} [PtAg_2(NO_3)_4(PP_3)_2] + 2 \text{ AgCl} \downarrow (S7)$$
12
III

1

Reaction <sup>a</sup>	$\partial P^{A}$ , $\partial P^{B}$ , $\partial P^{C}$	<i>δ</i> P for 1/2/5	$\partial \mathbf{P}^{\mathrm{L}}/(\mathbf{P}^{\mathrm{T}},\mathbf{P}^{\mathrm{M}})$	${}^{1}J({}^{31}\mathrm{P},{}^{195}\mathrm{Pt})$	Solvent
1 + 1 eq NP <sub>3</sub>	17.6s, -20.7s	35.0m, 23.1s (1)	-19.5		$\begin{array}{c} CD_3OD\\ +CD_2Cl_2 \end{array}$
1 + 2 eq NP <sub>3</sub>	17.5s, -20.5s	36.0m, 23.5br (1)	-19.3s		
2 + 1 eq PP <sub>3</sub>	56.0s, 52.1br, -14.6br	134.0s, 29.4s ( <b>2</b> )	- 14.8d, -19.1q		CDCl <sub>3</sub>
2 + 2 eq PP <sub>3</sub>	56.1s, 52.2br, -14.6br	134.0s, 29.4s ( <b>2</b> )	- 14.8d, -19.1q		
5 + 1 eq PP <sub>3</sub>	47.9s <sup>b</sup> , - <u>12.0</u> br, - <u>12.0</u> br	118.3s <sup>c</sup> , 25.0s <sup>d</sup> ( <b>5</b> )	- 14.6d	2259 <sup>b</sup> ,2503 <sup>c</sup> , 2591 <sup>d</sup>	CDCl <sub>3</sub>
5 + 1.5 eq PP <sub>3</sub>	47.9s <sup>b</sup> , - <u>12.5</u> br, - <u>12.5</u> br		- 14.6d, -19.1q	2258 <sup>b</sup>	

Table S1.  ${}^{31}P{}^{1}H$  NMR data at room temperature for reactions of 1, 2 and 5 with ligand

 $^{\rm a}$  See Scheme 1 for labels in P atoms. The underlined broad signals include  $P^{\rm B}$  and  $P^{\rm c}$ 

Compound <sup>a</sup>	$\partial P^{A}, \partial P^{B}, \partial P^{c}$	∂PHomo <sup>b</sup>	<sup>1</sup> <i>J</i> ( <sup>31</sup> P- <sup>195</sup> Pt)	${}^{1}J({}^{31}P - {}^{107/109}Ag)$
15+1eqPP <sub>3</sub>	58.5br, 13.6 d <sup>e</sup> <u>8.9</u> br ( <b>15*</b> )			528°
15+2eqPP <sub>3</sub>	61.0br, 31.9br, -11.9br ( <b>17</b> )	16.2d <sup>e</sup> , 8.1br, 0.4d <sup>e</sup> ,-0.5br		376°,176°
15+3eqPP <sub>3</sub>	60.9s, 32.2br, -7.2br ( <b>17</b> )	16.1d <sup>e</sup> ,0.5d <sup>e</sup> , -4.3br		384°,202°
16+1eqPP <sub>3</sub>	56.6s <sup>c</sup> , <u>14.4</u> brd <sup>e</sup> , <u>14.4</u> br ( <b>16*)</b>		2280°	496 <sup>e</sup>
16+2eqPP <sub>3</sub>	56.8s <sup>c</sup> , <u>13.0</u> br, -7.1br ( <b>18</b> )	19.8d <sup>e</sup> , <u>13.0</u> d <sup>e</sup> 5.3d,1.0b	2284°	378°,369°

## Table S2 ${}^{31}P{}^{1}H$ NMR data at room temperature for reactions of 15 and 16 with ligand in CDCl<sub>3</sub>

<sup>a</sup> See Scheme 1 for labels in P atoms. The underlined broad peaks include two resonances. <sup>b</sup>  $\partial$ PHomo =  $\partial$ P for the homonuclear complex [Ag<sub>2</sub>( $\mu$ - PP<sub>3</sub>)<sub>2</sub>](NO<sub>3</sub>)<sub>2</sub>

Reaction <sup>a</sup>	$\partial \!\!\!\!/ P^{\rm D} / P^{\rm H} / P^{\rm P} / P^{\rm S}$	$\partial \!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\!\partial \!\!\!\!\!\!\!\!\!\!P^U/P^Q/P^I/P^E$	$\partial \!$	$\partial P^{G}/P^{N}/P^{R}$	$\partial P^{V}/P^{Y}/P^{X}/P^{Z}$	$^{1}J(^{31}P,^{195}Pt)$	$^{1}J(^{31}\mathrm{P},^{107/109}\mathrm{Ag})$	Solvent
9+1eqCuCl <b>→</b> I	138.0br	37.4br	-6.9br	14.5br				DMSO-d <sub>6</sub>
9+2eqCuCl <b>→</b> II	128.0br	50.0br	15.0br	-4.9br				
12+1eqCuCl→I 12+2eqCuCl→III	129.2s <sup>b</sup> 61.3m <sup>b</sup>	34.3s <sup>c</sup> 58.6m <sup>d,e</sup>	-6.0br	14.9br -5.3br		2487 <sup>b</sup> /2565 <sup>c</sup> 2169 <sup>b</sup> /2272 <sup>d</sup>		DMSO-d <sub>6</sub>
9+1eqAgCl→I+III 9+2eqAgCl→II	142.6br, <u>66.0</u> m 127.4br	<u>66.0</u> m,37.5br 65.0br	<u>3.3</u> br <sup>f</sup> 7.6brd <sup>g</sup>	10.0d <sup>g</sup> 10.4br			4 23 <sup>g</sup> 160 <sup>g</sup>	DMSO-d <sub>6</sub>
12+1eqAgCl→I+III 12+2eqAgCl→II	135.0 br,57.0m 84.0 br	$\frac{57.0}{57.8}$ m <sup>d</sup> ,34.3s <sup>c</sup> 57.8br <sup>h</sup>	$\frac{5.1}{10.3} \text{br}^{\text{f}}$	9.8d <sup>g</sup>		2573 <sup>c</sup> ,2315 <sup>d</sup> 2310 <sup>h</sup>	423 <sup>g</sup>	DMSO-d <sub>6</sub>
9+1eqAgNO <sub>3</sub> →I	143.0br	37.4s	$\underline{6.6}\mathrm{br}^{\mathrm{e,f,g}}$				335 <sup>g</sup>	DMSO-d <sub>6</sub>
9+2eqAgNO <sub>3</sub> →I+III	143.0br, <u>66.4</u> m	<u>66.4</u> m,37.6br	$\underline{5.4}brd^{f, g}$	10.9d <sup>g</sup>			334 <sup>g</sup> ,434 <sup>g</sup>	+CD <sub>3</sub> OD
9+3eqAgNO <sub>3</sub> →II+III	123.0br, <u>67.0</u> m	<u>67.0</u> m,52.3br	<u>7.1</u> br <sup>g, i</sup>	13.6br			218 <sup>g</sup>	
12+1eqAgNO <sub>3</sub> →I <sup>j</sup>	129.1s <sup>b</sup>	34.2s <sup>c</sup>	<u>5.1</u> br <sup>f</sup>			2398 <sup>b</sup> /2570 <sup>c</sup>		DMSO-d <sub>6</sub>
12+2eqAgNO <sub>3</sub> →I+III	128.0br,58.1s <sup>e</sup>	57.2s <sup>d</sup> ,34.1s <sup>c</sup>	5.1br <sup>f</sup>	10.6d <sup>g</sup>		2578°, 2255 <sup>d</sup>	428 <sup>g</sup>	+CD <sub>3</sub> OD
12+3eqAgNO <sub>3</sub> →III	58.2br <sup>b,e</sup>	57.4br		13.0br		2103 <sup>b</sup>		
12+4eqAgNO₃→III	58.4br <sup>b,e</sup>	57.4br		14.5d <sup>g</sup>		2266 <sup>b</sup>	482 <sup>g</sup>	
9+1eqAuCl(tdg)→ →II '+IV	123.0br, <u>78.0</u> br	<u>78.0</u> br,53.0br			$\frac{40.1}{39.1} br^k,$			DMSO-d <sub>6</sub> +CD <sub>3</sub> OD
9+2eqAuCl(tdg) <b>→</b> IV	90.2br	74.3s			$\frac{38.0}{36.0}$ m <sup>e, m</sup> ,			
12+1eqAuCl(tdg)→ → II <sup>^</sup> +IV	96.5s <sup>p</sup> ,62.9br <sup>q</sup>	51.0s <sup>r</sup> ,49.0s <sup>s</sup>			$\frac{39.0}{37.0}$ s <sup>1</sup> , k,	2963 <sup>p</sup> ,3389 <sup>q</sup> 3660 <sup>r</sup> , 2507 <sup>s</sup>		DMSO-d <sub>6</sub> +CD <sub>3</sub> OD
12+2eqAuCl(tdg)→IV	63.0br <sup>q</sup>	51.0s <sup>r</sup>			<u>39.0</u> s <sup>m</sup> <u>37.0</u> s <sup>n</sup>	3378 <sup>q</sup> , 3641 <sup>r</sup>		

**Table S3.**  ${}^{31}P{}^{1}H$  NMR data at room temperature for reactions of 9 and 12 with Cu(I), Ag(I) and Au(I).

<sup>a</sup> See Scheme 3 for structures **I** -**I**V. The underlined signals include 2 or more resonances. <sup>e</sup>  $J({}^{31}P, {}^{31}P)$  in Hz: 126 [**I** ( $P^{F}, P^{G}$ )], 281 [**III** ( $P^{P}, P^{Q}$ )], 157 [**III** ( $P^{P}, P^{R}$ )], 53 [**IV** ( $P^{X}, P^{Y}$ ). <sup>f</sup>  $P^{F}$  and  $P^{G}$ . <sup>i</sup>  $P^{J}$ ,  $P^{K}$  and  $P^{N}$ . <sup>j</sup> **I** is coexisting with unreacted complex **12** [ $\delta$  56.7(<sup>1</sup>J( ${}^{31}P, {}^{195}Pt$ ) = 2168 Hz), 37.2, 12.0]. <sup>k</sup>  $P^{N}, P^{V}$  and  $P^{Y}$ . <sup>1</sup>  $P^{J}, P^{K}, P^{X}$  and  $P^{Z}$ . <sup>m</sup>  $P^{V}$  and  $P^{Y}$ . <sup>n</sup>  $P^{X}$  and  $P^{Z}$ .

**Supplementary Information** 

**Equations S1-S7** 

Tables S1-S3

**Figure Captions** 

Figure S1. <sup>31</sup>P{<sup>1</sup>H}NMR spectra at room temperature for : (a) 1 in CD<sub>3</sub>OD; (b) or (c) 1+1 or 2 eq NP<sub>3</sub> in CD<sub>3</sub>OD+ CD<sub>2</sub>Cl<sub>2</sub> affording 8 in coexistence with 1 (•) and free PP<sub>3</sub>; (d) 5 + 1.5 eq PP<sub>3</sub> in CDCl<sub>3</sub> affording 12 in coexistence with free PP<sub>3</sub>.

Figure S2. (a) ORTEP diagram for 10; (b)View of the unit cell for 10. 4CHCl<sub>3</sub>. Phenyl rings omitted for clarity.

- Figure S3. (a) ORTEP diagram for 18; (b)View of the unit cell for 18. 2CHCl<sub>3</sub>. Phenyl rings omitted for clarity
- Figure S4. <sup>31</sup>P{<sup>1</sup>H}NMR spectra (r.t) for : (a) 12+1 eq AgCl in DMSO-d<sub>6</sub> affording I+III and (b) 9+1eq AgNO<sub>3</sub> in DMSO-d<sub>6</sub> + CD<sub>3</sub>OD affording I.



Figure S1



**(a)** 



Figure S2



Figure S3



Figure S4