

## Supporting information belonging to the paper

“A new hexakis(isocyanato)silicate(IV) and the first neutral Lewis-base adducts of silicontetraakis(isocyanato)”

### Content:

- 1) General experimental conditions
- 2) Syntheses and full spectroscopic data
- 3) Single crystal X-ray diffraction analyses
- 4) Crystallographic information files for compounds **1a**, **2a,b** and **4**
- 5) Variable temperature NMR spectra of compound **3**

### 1) General

KNCO and KNCS (Aldrich) were dried at 105°C and 130°C, respectively, in high vacuum. SiCl<sub>4</sub> was treated with K<sub>2</sub>CO<sub>3</sub>, trap-to-trap condensed and degassed prior to use. (PPN)NCO, m.p. 212–214°C and (PPN)NCS, m.p. 186–189°C were prepared according to published procedures.<sup>28</sup> 2,2'-bipyridine (bpy) and 1,10-phenanthroline (phen) were sublimed. Solvents were obtained from a Grubbs column, stirred over CaH<sub>2</sub> and trap-to-trap condensed prior to use. Schlenk tube and Glove box techniques were used under Argon throughout. Fourier transform infrared (FTIR) spectra were obtained from a Galaxy 2000 spectrometer at 4 cm<sup>-1</sup> resolution between NaCl (nujol suspension) or CaF<sub>2</sub> (solution, d = 0.2 mm) windows. NMR spectra were recorded at r.t. on Bruker spectrometers and are calibrated against the signal of the residual proton content (<sup>1</sup>H, 1.94 ppm and 5.32 ppm) or natural abundance (<sup>13</sup>C, 53.8 ppm) of the solvents CD<sub>3</sub>CN and CD<sub>2</sub>Cl<sub>2</sub>. Thermogravimetry (TGA) and differential scanning calorimetry (DSC) were performed under N<sub>2</sub> using the Perkin Elmer Pyris 1 apparatus at a heating rate of 10 K min<sup>-1</sup>. Elemental analyses were obtained from the Centre for chemical instrumentation and analytical services at Sheffield. Mass spectra were recorded on a VG Autospec, calibrated against PFK. The positive ion current was recorded between 50 < m/z < 700 at temperatures of ca. 200°C. Peaks below 10% of base intensity are omitted for m/z < 200. Fragmentation indicated as *bpy* and *phen* closely resembles published data on genuine samples of the diimines.

### 2) Syntheses and full spectroscopic data

Preparation of **1b**. KNCS (2.385 g, 24.54 mmol) was suspended in MeCN at 0°C. Under stirring SiCl<sub>4</sub> (0.20 ml, 1.7 mmol) were added using a syringe upon which the suspension turned cloudy immediately. The reaction mixture was stirred at r.t. for 4 d, during which the reaction solution, obtained upon sedimentation, was monitored by IR spectroscopy. The diluted, supernatant, clear and colourless solution revealed the presence of two bands in the NCS stretch region at 2101 and 2058 cm<sup>-1</sup> which arise from the presence of Si(NCS)<sub>6</sub><sup>2-</sup> and KNCS, respectively. Afterwards the reaction mixture was filtered (0.203 g residue). The filter solution collected in a new Schlenk tube and diminished three times to ca. one half of the respective original volume which caused precipitation each time, while the ratio of Si(NCS)<sub>6</sub><sup>2-</sup> over NCS<sup>-</sup> increased. A total of 1.38 g filter residues were obtained, IR spectra of which feature

mainly KNCS. Evaporation of the final filter solution resulted in oily slush (nujol: 2257 (MeCN), 2099 ( $\nu_{as}(NC)$ ), 910 ( $\nu(CS)$ ) $cm^{-1}$ ), which was dried until mass remained almost constant, resulting in 1.297 g of white powder that still contains MeCN.

IR (nujol,  $cm^{-1}$ ): 2297vw (MeCN), 2263vw (MeCN), 2130vs ( $\nu(CN)$ ), 2093vw, sh, 907w ( $\nu(CS)$ ).

IR (MeCN,  $cm^{-1}$ ): 2101,  $\nu(CN)$ .

IR (THF,  $cm^{-1}$ ): 2105,  $\nu(CN)$ .

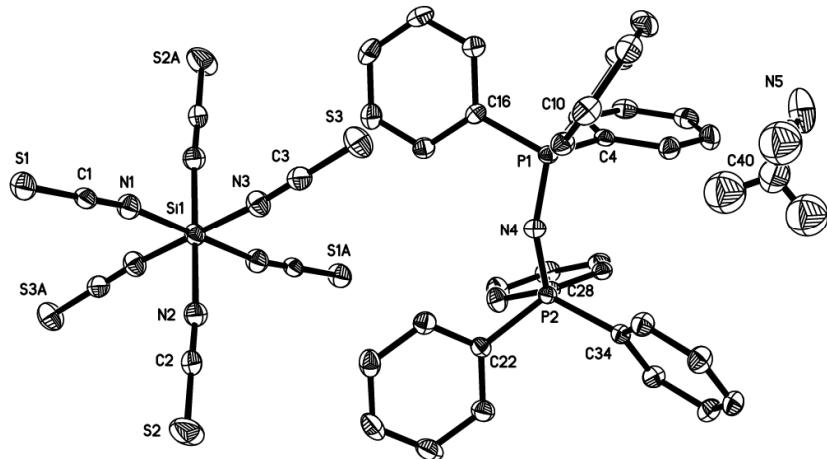
**Preparation of  $(PPN)_2[Si(NCS)_4] \cdot 2MeCN$ , **2b**·2MeCN.** A Schlenk tube was charged with **1b** (0.228 g) and (PPN)NCS (0.430 g, 0.721 mmol). MeCN (50 ml) was added at  $-20^{\circ}C$  and the resulting suspension was rapidly stirred and allowed to warm to r.t., after which the suspension was left to settle. An IR spectrum of the colourless supernatant solution in the region of the NCS stretch showed the presence of only two bands at  $2101\text{ cm}^{-1}$  (weak) and  $2059\text{ cm}^{-1}$  (very strong) due to the presence of  $[Si(NCS)_6]^{2-}$  and  $NCS^-$  ions, respectively. The solution, containing mainly (PPN)NCS and KNCS, was filtered off and discarded (0.290 g non-volatile content). The white filter residue was dried, resulting in a raw product (0.364 g) which was dissolved in MeCN (55 ml) at  $50^{\circ}C$  and filtered at this temperature, which afforded a clear solution. Cooling to  $-30^{\circ}C$  caused the formation of crystals, which were filtered off in the cold and dried *in vacuo*. Recrystallisation was repeated using slightly less solvent, which afforded 0.294 g of colourless, sparkling crystals (0.191 mol, 53% with respect to (PPN)NCS), mp.  $257 - 261^{\circ}C$ . Anal. calcd. for  $C_{82}H_{66}N_{10}P_4S_6Si$  (1535.87 g  $mol^{-1}$ ): C, 64.13; H, 4.33; N, 9.12%. Found: C, 63.97; H, 4.22; N, 9.00 %. IR (nujol,  $cm^{-1}$ )  $\nu = 3086, 3056, 2286$  and  $2249vw$ , MeCN, 2098vs,  $\nu(CN)$ , 1588, 1572, 1481, 1467, 1437, 1314, 1302, 1284, 1268, 1181, 1160, 1116, 1071, 1025, 998, 916, 797, 760, 745, 724, 691, 617, 549; THF: 2105;  $CH_2Cl_2$ : 2106; MeCN: 2101. DSC  $T_{on} = 83^{\circ}C$  (MeCN loss),  $256^{\circ}C$  (melting). TGA  $T_{on} = 82^{\circ}C$  (MeCN loss,  $\Delta m/m_0$  found 5.4%, calcd. 5.3% for loss of two equivalents of MeCN),  $223^{\circ}C$  (dec.).

**Si(NCO)<sub>4</sub>(phen)·MeCN (4·MeCN).** A Schlenk tube was charged with **1a** (0.200 g) and phen (0.100 g, 0.554 mmol). The tube was immersed in a cold bath ( $-40^{\circ}C$ ) and MeCN (ca. 35 ml) was added via cannula transfer. Upon rapid stirring a mixture of fully dissolved **1a** and crystalline phen resulted. The cold bath was removed and phen completely dissolved while the temperature of the reaction mixture approached ambient conditions. The resulting clear reaction solution turned cloudy within several minutes. Stirring was continued for 24 h after which the white suspension was allowed to settle. An IR spectrum of an evaporated sample of the reaction solution (nujol) showed the presence of **4** ( $2318\text{ cm}^{-1}$  m,  $2260\text{ cm}^{-1}$  vs). The white precipitate (66 mg), consisting mainly of KNCO (IR:  $2161\text{ cm}^{-1}$ ), was filtered off and the filtrate diminished to ca. 3 ml causing crystallisation, which was completed at  $-28^{\circ}C$  for 2 h. The supernatant was filtered off and the remainder, 0.215 g, raw off-white **4**, recrystallised from a minimal volume of warm MeCN (ca.  $50^{\circ}C$ ) at  $-28^{\circ}C$  over-

night, and washed in the cold (3.5 ml MeCN). The residue was dried in vacuo at the same temperature, which afforded fine, colourless and electrostatic sheets (0.151 g, 66% with respect to phen) mp 224 °C (DSC). Anal. calcd. for C<sub>16</sub>H<sub>8</sub>N<sub>6</sub>O<sub>4</sub>Si (376.36 g mol<sup>-1</sup>): C, 51.06; H, 2.14; N, 22.33%. Found C, 50.59; H, 2.35; N, 22.82%. IR (nujol, cm<sup>-1</sup>)  $\nu$  = 3645vw, 2334, 2329, 2279, 2261, 2242, 1456, 1436, 1413, 1318, 1229, 1208, 1151, 1114, 1114, 1062, 1042, 881, 853, 752, 7, 660, 615; THF: 2311m, 2255vs; CH<sub>2</sub>Cl<sub>2</sub>: 2315m, 2283w, sh, 2256vs. <sup>1</sup>H NMR (250 MHz, CD<sub>3</sub>CN, ppm):  $d$  = 1.94 (s, CH<sub>3</sub>), 8.25 (dd, 2H), 8.37 (s, 2H), 9.01 (dd, 2H), 9.70 (dd, 2H); <sup>3</sup>J(H<sub>2</sub>,H<sub>3</sub>) = 5.4, <sup>4</sup>J(H<sub>2</sub>,H<sub>4</sub>) = 1.3, <sup>3</sup>J(H<sub>3</sub>,H<sub>4</sub>) = 8.3 Hz; CD<sub>2</sub>Cl<sub>2</sub> (400 MHz)  $d$  = 1.97 (s, CH<sub>3</sub>), 8.21 (dd, 2H), 8.28 (s, 2H), 8.86 (dd, 2H), 9.70 (dd, 2H); <sup>3</sup>J(H<sub>2</sub>,H<sub>3</sub>) = 5.3, <sup>4</sup>J(H<sub>2</sub>,H<sub>4</sub>) = 1.3, <sup>3</sup>J(H<sub>3</sub>,H<sub>4</sub>) = 8.3 Hz; <sup>13</sup>C{<sup>1</sup>H} (100 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $d$  = 126.8, 127.8, 129.5, 134.0, 141.4, 144.4 (resonances due to carbon in NCO could not be observed as a result of insufficient solubility). EI MS m/z = 196 (40) [M<sup>+</sup> – phen], 181 (27) [phen], 180 (100) [phen], 179 (37) [phen], 155 (14) [phen], 154 (73) [M<sup>+</sup> – phen – NCO], 153 (23) [phen], 152 (14) [phen], 127 (13) [phen], 112 (13) [M<sup>+</sup> – phen – 2NCO], 90 (29) [phen], 76 (20) [phen], 70 (56) [M<sup>+</sup> – phen – 3NCO], 63 (14) [phen].

### 3) Single crystal X-ray diffraction analyses of **1a**, **2a,b** and **4**

Single crystals were grown from acetonitrile solution of pure **1a**, **2a,b** and **4** upon cooling saturated solutions in MeCN to -28°C. Diffraction data were collected on a Bruker Smart CCD area detector<sup>28</sup> with an Oxford Cryosystems low temperature unit and Mo-K $\alpha$  radiation ( $\bar{\lambda}$  = 0.71073 Å). Crystallographic and experimental details are summarized in Table 3. All measured reflections were corrected for Lorentz and polarisation effects and for absorption by semi empirical methods based on symmetry-equivalent and repeated reflections. The structures were solved by direct methods and refined by full matrix least squares methods on  $F^2$ . H atoms were placed geometrically and refined with a riding model and with  $U_{\text{iso}}$  constrained to be 1.2 times (1.5 for CH<sub>3</sub> groups)  $U_{\text{eq}}$  of the carrier atom. Complex scattering factors were taken from the program package SHELXTL as implemented on a Viglen Pentium computer.<sup>29</sup> The weighting schemes  $w = 1/[S^2(F_o^2) + (a \cdot P)^2 + b \cdot P]$ ,  $P = (F_o^2 + 2F_c^2)/3$ ,  $a = 0.0485$ ,  $b = 3.0619$  (**1a**), 0.0498, 1.1541 (**2a**), 0.0440, 0.7175 (**2b**), 0.0499, 0.0291 (**4**), were used in the later stages of refinement. Mean and max.  $\delta / \sigma$  were 0.000 / 0.000 throughout. Plots were created with SHELX XP and are set at the 50% probability level. **2a** has a residual electron density peak of 1.317 e·Å<sup>-3</sup> at a distance of 0.71 Å from O<sub>2</sub>. All structures were investigated for κN / κO NCO bond isomerism. Alternative solutions were invariably associated with an increase of R<sub>1</sub> and wR<sub>2</sub> values.



Structure of the asymmetric unit in a single crystal of **2a·2MeCN**, Ph hydrogen atoms are omitted. Selected bond lengths [Å] and angles [°]: Si-N1 1.823(2), Si-N2 1.829(2), Si-N3 1.831(2), N1-C1 1.170(2), N2-C2 1.168(2), N3-C3 1.170(2), C1-S1 1.607(2), C2-S2 1.608(2), C3-S3 1.611(2), N1-Si-N2 90.41(6), N1-Si-N3 90.11(6), N2-Si-N3 90.06(7), P2-N4-P1 139.97(9), Si-N1-C1 168.20(14), Si-N2-C2 170.13(15), Si-N3-C3 175.12(15), N1-C1-S1 179.03(15), N2-C2-S2 178.44(17), N3-C3-S3 179.53(18).

**Table 1** A selection of recently published structural data on related silicon complexes in relation to **1a**, **2a,b** and **4**.

Complex	Si-N [Å]	N-C [Å]	C-E [Å]	Si-N-C [°]	N-C-E [°]	Ref.
[K(MeCN) <sub>2</sub> ] <sub>2</sub> [Si(NCO) <sub>6</sub> ] ( <b>1a</b> )	1.81-1.84	1.14-1.16	1.19-1.20	171-173	178-179	a
(PPN) <sub>2</sub> [Si(NCO) <sub>6</sub> ] ( <b>2a</b> )	1.83-1.86	1.16-1.17	1.19-1.20	155-162	177-178	a
[K(crown)] <sub>2</sub> [Si(NCO) <sub>6</sub> ]-2MeCN	1.82-1.83	1.15	1.18	157-165	177-179	b
[Si(NCO) <sub>4</sub> (phen)] ( <b>4</b> )	eq.	1.78-1.79	1.15-1.17	1.18-1.20	178-179	178-179
	ax.	1.82-1.83	1.18-1.19	1.19	177-178	177-178
(PPN) <sub>2</sub> [Si(NCS) <sub>6</sub> ]-2MeCN ( <b>2b</b> )	1.82-1.83	1.17	1.61	168-175	178-180	a
(NEt <sub>4</sub> ) <sub>2</sub> [Si(NCS) <sub>6</sub> ]	1.82-1.83	1.17	1.59-1.60	170-180	179-180	b
[K(crown)] <sub>2</sub> [Si(NCS) <sub>6</sub> ]-2MeCN	1.82-1.84	1.16-1.17	1.59-1.61	159-171	177-180	b

a) this paper, b) Seiler, O.; Burschka, C.; Goetz, K.; Kaupp, M.; Metz, S.; Tacke, R., *Z. Anorg. Allg. Chem.* **2007**, 633, 2667-2670.

#### 4a) Crystallographic information file for compound **1a** (data\_chppx55m)

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and is  
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O1 0.054(2) 0.019(2) 0.058(2) -0.0041(17) 0.030(2) -0.0060(16)  
N3 0.0230(18) 0.034(2) 0.0220(19) 0.0022(18) 0.0000(15) -0.0013(16)  
N1 0.032(2) 0.017(2) 0.036(2) 0.0002(18) 0.0069(16) 0.0019(16)  
C3 0.019(2) 0.028(3) 0.024(2) 0.001(2) 0.0106(19) 0.0016(19)  
N2 0.0235(18) 0.033(2) 0.030(2) 0.0027(18) 0.0097(16) -0.0027(17)  
N4 0.028(2) 0.038(3) 0.050(3) 0.010(2) -0.0005(19) -0.0086(19)  
C1 0.019(2) 0.026(3) 0.023(2) -0.006(2) 0.0097(17) -0.0032(18)  
C4 0.027(2) 0.025(3) 0.026(2) 0.004(2) -0.0011(19) 0.001(2)  
C2 0.023(2) 0.026(3) 0.0141(19) 0.0017(19) 0.0036(17) -0.0031(18)  
N5 0.072(3) 0.064(4) 0.037(3) 0.000(3) 0.008(2) 0.040(3)  
C6 0.030(2) 0.044(3) 0.022(2) 0.002(2) 0.003(2) 0.004(2)  
C5 0.027(2) 0.038(3) 0.046(3) 0.004(2) 0.011(2) -0.002(2)  
C7 0.024(2) 0.048(3) 0.038(3) -0.013(3) 0.004(2) 0.003(2)  
  
\_geom\_special\_details  
;  
  All esds (except the esd in the dihedral angle between two l.s.  
  planes)  
  are estimated using the full covariance matrix. The cell esds are  
  taken  
    into account individually in the estimation of esds in distances,  
    angles  
    and torsion angles; correlations between esds in cell parameters are  
    only  
    used when they are defined by crystal symmetry. An approximate  
    (isotropic)  
    treatment of cell esds is used for estimating esds involving l.s.

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planes.  
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K1 N5 2.758(5) . ?  
K1 O3 2.789(3) . ?  
K1 O1 2.793(3) . ?  
K1 N4 2.798(4) . ?  
K1 O1 2.829(3) 3_656 ?  
K1 O3 2.850(3) 3_656 ?  
Si1 N2 1.814(3) 1_556 ?  
Si1 N2 1.814(3) 3_756 ?  
Si1 N1 1.822(4) 4_656 ?  
Si1 N1 1.822(4) 2_656 ?  
Si1 N3 1.834(4) 3_757 ?  
Si1 N3 1.834(4) . ?  
O2 C2 1.201(5) . ?  
O3 C3 1.200(5) . ?  
O3 K1 2.850(3) 3_656 ?  
O1 C1 1.192(6) . ?  
O1 K1 2.829(3) 3_656 ?  
N3 C3 1.139(5) . ?  
N1 C1 1.155(6) . ?  
N1 Si1 1.822(4) 2_646 ?  
N2 C2 1.157(5) . ?  
N2 Si1 1.814(3) 1_554 ?  
N4 C4 1.133(6) . ?  
C4 C5 1.460(6) . ?  
N5 C6 1.128(6) . ?  
C6 C7 1.440(7) . ?  
C5 H5A 0.9800 . ?  
C5 H5B 0.9800 . ?  
C5 H5C 0.9800 . ?  
C7 H12A 0.9800 . ?  
C7 H12B 0.9800 . ?  
C7 H12C 0.9800 . ?  
  
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O2 K1 N5 77.67(12) . . ?  
O2 K1 O3 165.30(11) . . ?  
N5 K1 O3 94.85(12) . . ?  
O2 K1 O1 100.46(10) . . ?  
N5 K1 O1 86.74(15) . . ?  
O3 K1 O1 66.17(11) . . ?  
O2 K1 N4 78.23(12) . . ?  
N5 K1 N4 95.93(16) . . ?  
O3 K1 N4 115.46(12) . . ?  
O1 K1 N4 176.66(12) . . ?
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O2 K1 O1 124.93(11) . 3\_656 ?  
N5 K1 O1 154.00(13) . 3\_656 ?  
O3 K1 O1 65.78(11) . 3\_656 ?  
O1 K1 O1 99.81(10) . 3\_656 ?  
N4 K1 O1 78.64(12) . 3\_656 ?  
O2 K1 O3 78.45(9) . 3\_656 ?  
N5 K1 O3 138.91(15) . 3\_656 ?  
O3 K1 O3 100.10(8) . 3\_656 ?  
O1 K1 O3 65.44(11) . 3\_656 ?  
N4 K1 O3 111.23(12) . 3\_656 ?  
O1 K1 O3 64.90(11) 3\_656 3\_656 ?  
O2 K1 K1 125.97(8) . 3\_656 ?  
N5 K1 K1 131.17(12) . 3\_656 ?  
O3 K1 K1 50.79(7) . 3\_656 ?  
O1 K1 K1 50.34(7) . 3\_656 ?  
N4 K1 K1 128.04(10) . 3\_656 ?  
O1 K1 K1 49.47(7) 3\_656 3\_656 ?  
O3 K1 K1 49.31(7) 3\_656 3\_656 ?  
N2 Si1 N2 180.000(1) 1\_556 3\_756 ?  
N2 Si1 N1 90.18(17) 1\_556 4\_656 ?  
N2 Si1 N1 89.82(17) 3\_756 4\_656 ?  
N2 Si1 N1 89.82(17) 1\_556 2\_656 ?  
N2 Si1 N1 90.18(17) 3\_756 2\_656 ?  
N1 Si1 N1 180.0(3) 4\_656 2\_656 ?  
N2 Si1 N3 90.56(16) 1\_556 3\_757 ?  
N2 Si1 N3 89.44(16) 3\_756 3\_757 ?  
N1 Si1 N3 89.60(17) 4\_656 3\_757 ?  
N1 Si1 N3 90.40(17) 2\_656 3\_757 ?  
N2 Si1 N3 89.44(16) 1\_556 . ?  
N2 Si1 N3 90.56(16) 3\_756 . ?  
N1 Si1 N3 90.40(17) 4\_656 . ?  
N1 Si1 N3 89.60(17) 2\_656 . ?  
N3 Si1 N3 180.000(1) 3\_757 . ?  
C2 O2 K1 124.4(3) . . ?  
C3 O3 K1 143.9(3) . . ?  
C3 O3 K1 132.3(3) . 3\_656 ?  
K1 O3 K1 79.90(8) . 3\_656 ?  
C1 O1 K1 148.0(3) . . ?  
C1 O1 K1 127.6(3) . 3\_656 ?  
K1 O1 K1 80.19(10) . 3\_656 ?  
C3 N3 Si1 172.7(4) . . ?  
C1 N1 Si1 172.3(4) . 2\_646 ?  
N3 C3 O3 179.3(5) . . ?  
C2 N2 Si1 170.5(4) . 1\_554 ?  
C4 N4 K1 136.8(4) . . ?  
N1 C1 O1 179.4(4) . . ?  
N4 C4 C5 178.8(5) . . ?  
N2 C2 O2 178.1(4) . . ?  
C6 N5 K1 161.6(5) . . ?  
N5 C6 C7 178.9(5) . . ?  
C4 C5 H5A 109.5 . . ?  
C4 C5 H5B 109.5 . . ?  
H5A C5 H5B 109.5 . . ?  
C4 C5 H5C 109.5 . . ?  
H5A C5 H5C 109.5 . . ?  
H5B C5 H5C 109.5 . . ?  
C6 C7 H12A 109.5 . . ?  
C6 C7 H12B 109.5 . . ?  
H12A C7 H12B 109.5 . . ?  
C6 C7 H12C 109.5 . . ?  
H12A C7 H12C 109.5 . . ?

H12B C7 H12C 109.5 . . ?

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#### 4b) Crystallographic information file for compound 2a (data\_chppx69\_0m)

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_chemical_name_common ?
_chemical_melting_point ?
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'C78 H60 N8 O6 P4 Si'
_chemical_formula_weight 1357.31

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_atom_type_description
_atom_type_scat_dispersion_real
_atom_type_scat_dispersion_imag
_atom_type_scat_source
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'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'H' 'H' 0.0000 0.0000
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'N' 'N' 0.0061 0.0033
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'O' 'O' 0.0106 0.0060
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Si' 'Si' 0.0817 0.0704
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'P' 'P' 0.1023 0.0942
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'

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_symmetry_space_group_name_H-M P-1

loop_
_symmetry_equiv_pos_as_xyz
'x, y, z'
'-x, -y, -z'

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_cell_length_b 12.9974(5)
_cell_length_c 13.4729(6)
_cell_angle_alpha 63.885(2)
_cell_angle_beta 70.577(2)
_cell_angle_gamma 75.503(2)
_cell_volume 1707.67(12)
_cell_formula_units_Z 1
_cell_measurement_temperature 120(2)
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_exptl_crystal_colour            colourless
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_exptl_absorpt_process_details   sadabs

_exptl_special_details
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_diffrn_radiation_type             MoK\alpha
_diffrn_radiation_source           'fine-focus sealed tube'
_diffrn_radiation_monochromator    graphite
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_diffrn_reflns_av_sigmaI/netI     0.0282
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_diffrn_reflns_theta_max           27.50
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_computing_cell_refinement         'Bruker SMART'
_computing_data_reduction          'Bruker SAINT'
_computing_structure_solution      'Bruker SHELXTL'
_computing_structure_refinement    'SHELXL-97 (Sheldrick, 1997)'
_computing_molecular_graphics      'Bruker SHELXTL'
_computing_publication_material    'Bruker SHELXTL'

_refine_special_details
;
  Refinement of F^2 against ALL reflections. The weighted R-factor
wR and
```

goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\text{sigma}(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger. The residual electron density (1.317) was found to be 0.71 Angstroms from O2. DFIX Restraints were applied to atoms N2 C2 and C2 O2 and refinement converged successfully.  
;

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_refine_ls_matrix_type full  
_refine_ls_weighting_scheme calc  
_refine_ls_weighting_details  
'calc w=1/[s^2^(Fo^2^)+(0.0498P)^2^+1.1541P] where  
P=(Fo^2^+2Fc^2^)/3'  
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_atom_sites_solution_secondary difmap  
_atom_sites_solution_hydrogens geom  
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_refine_ls_extinction_method none  
_refine_ls_extinction_coef ?  
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_atom_site_fract_z  
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_atom_site_occupancy  
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Sil Si 0.0000 1.0000 0.0000 0.02593(16) Uani 1 2 d S . . .  
P1 P 0.19131(4) 0.30352(3) 0.26845(4) 0.01753(10) Uani 1 1 d . . .  
P2 P 0.36555(4) 0.40965(3) 0.30098(3) 0.01587(10) Uani 1 1 d . . .  
N1 N 0.04729(15) 0.84553(14) 0.07263(15) 0.0327(4) Uani 1 1 d . . .  
N2 N -0.02516(15) 1.02014(14) 0.13406(10) 0.0345(4) Uani 1 1 d D . . .  
N3 N 0.16025(14) 1.02722(15) -0.04307(16) 0.0351(4) Uani 1 1 d . . .

N4 N 0.24460(12) 0.39925(12) 0.27821(12) 0.0198(3) Uani 1 1 d . . .  
O1 O 0.04718(15) 0.64924(12) 0.19769(14) 0.0452(4) Uani 1 1 d . . .  
O2 O -0.0564(2) 0.96685(19) 0.33178(10) 0.0811(7) Uani 1 1 d D . . .  
O3 O 0.33921(13) 1.06520(13) -0.02469(15) 0.0443(4) Uani 1 1 d . . .  
C1 C 0.04622(16) 0.74906(16) 0.13546(17) 0.0276(4) Uani 1 1 d . . .  
C2 C -0.0428(2) 0.99279(17) 0.23222(10) 0.0389(5) Uani 1 1 d D . . .  
C3 C 0.24770(16) 1.04556(15) -0.03170(16) 0.0280(4) Uani 1 1 d . . .  
C4 C 0.17353(15) 0.35139(14) 0.12660(14) 0.0205(3) Uani 1 1 d . . .  
C5 C 0.19124(15) 0.46389(15) 0.04856(14) 0.0234(3) Uani 1 1 d . . .  
H5 H 0.2166 0.5145 0.0688 0.028 Uiso 1 1 calc R . . .  
C6 C 0.17154(16) 0.50194(17) -0.05949(15) 0.0290(4) Uani 1 1 d . . .  
H6 H 0.1833 0.5787 -0.1129 0.035 Uiso 1 1 calc R . . .  
C7 C 0.13483(17) 0.42783(18) -0.08911(17) 0.0329(4) Uani 1 1 d . . .  
H7 H 0.1223 0.4538 -0.1630 0.040 Uiso 1 1 calc R . . .  
C8 C 0.11635(18) 0.31594(18) -0.01121(18) 0.0344(4) Uani 1 1 d . . .  
H8 H 0.0909 0.2657 -0.0319 0.041 Uiso 1 1 calc R . . .  
C9 C 0.13492(17) 0.27725(16) 0.09653(17) 0.0293(4) Uani 1 1 d . . .  
H9 H 0.1216 0.2008 0.1500 0.035 Uiso 1 1 calc R . . .  
C10 C 0.03900(15) 0.28493(14) 0.36265(14) 0.0217(3) Uani 1 1 d . . .  
C11 C -0.01050(17) 0.18216(16) 0.40452(18) 0.0325(4) Uani 1 1 d . . .  
H11 H 0.0383 0.1181 0.3885 0.039 Uiso 1 1 calc R . . .  
C12 C -0.13183(18) 0.17362(19) 0.4700(2) 0.0415(5) Uani 1 1 d . . .  
H12 H -0.1656 0.1034 0.4994 0.050 Uiso 1 1 calc R . . .  
C13 C -0.20308(17) 0.26736(19) 0.49228(18) 0.0379(5) Uani 1 1 d . . .  
H13 H -0.2858 0.2612 0.5371 0.045 Uiso 1 1 calc R . . .  
C14 C -0.15474(17) 0.37023(18) 0.44972(17) 0.0349(4) Uani 1 1 d . . .  
H14 H -0.2047 0.4345 0.4648 0.042 Uiso 1 1 calc R . . .  
C15 C -0.03356(16) 0.37989(16) 0.38510(15) 0.0270(4) Uani 1 1 d . . .  
H15 H -0.0002 0.4503 0.3564 0.032 Uiso 1 1 calc R . . .  
C16 C 0.28510(14) 0.16518(13) 0.30269(14) 0.0197(3) Uani 1 1 d . . .  
C21 C 0.36531(17) 0.12816(16) 0.21757(16) 0.0276(4) Uani 1 1 d . . .  
H21 H 0.3617 0.1699 0.1402 0.033 Uiso 1 1 calc R . . .  
C20 C 0.45023(18) 0.03011(17) 0.24656(19) 0.0351(4) Uani 1 1 d . . .  
H20 H 0.5038 0.0044 0.1891 0.042 Uiso 1 1 calc R . . .  
C19 C 0.45672(18) -0.02997(16) 0.35905(19) 0.0354(5) Uani 1 1 d . . .  
H19 H 0.5161 -0.0958 0.3782 0.042 Uiso 1 1 calc R . . .  
C18 C 0.37686(18) 0.00552(16) 0.44393(18) 0.0313(4) Uani 1 1 d . . .  
H18 H 0.3814 -0.0364 0.5210 0.038 Uiso 1 1 calc R . . .  
C17 C 0.29037(16) 0.10209(14) 0.41650(15) 0.0240(4) Uani 1 1 d . . .  
H17 H 0.2348 0.1253 0.4750 0.029 Uiso 1 1 calc R . . .  
C22 C 0.35121(15) 0.36343(13) 0.45169(13) 0.0185(3) Uani 1 1 d . . .  
C23 C 0.24970(16) 0.30930(15) 0.53244(15) 0.0242(4) Uani 1 1 d . . .  
H23 H 0.1875 0.2985 0.5078 0.029 Uiso 1 1 calc R . . .  
C24 C 0.23941(18) 0.27119(17) 0.64867(15) 0.0305(4) Uani 1 1 d . . .  
H24 H 0.1709 0.2333 0.7032 0.037 Uiso 1 1 calc R . . .  
C25 C 0.32889(18) 0.28836(16) 0.68517(15) 0.0295(4) Uani 1 1 d . . .  
H25 H 0.3217 0.2619 0.7646 0.035 Uiso 1 1 calc R . . .  
C26 C 0.42918(17) 0.34424(15) 0.60584(15) 0.0252(4) Uani 1 1 d . . .  
H26 H 0.4898 0.3567 0.6312 0.030 Uiso 1 1 calc R . . .  
C27 C 0.44061(15) 0.38184(14) 0.48945(14) 0.0209(3) Uani 1 1 d . . .  
H27 H 0.5090 0.4201 0.4353 0.025 Uiso 1 1 calc R . . .  
C28 C 0.39065(15) 0.55890(13) 0.23677(13) 0.0179(3) Uani 1 1 d . . .  
C29 C 0.29125(15) 0.64149(15) 0.25535(15) 0.0228(3) Uani 1 1 d . . .  
H29 H 0.2125 0.6183 0.3006 0.027 Uiso 1 1 calc R . . .  
C30 C 0.30849(17) 0.75740(15) 0.20721(16) 0.0276(4) Uani 1 1 d . . .  
H30 H 0.2412 0.8134 0.2196 0.033 Uiso 1 1 calc R . . .  
C31 C 0.42378(17) 0.79198(15) 0.14097(15) 0.0269(4) Uani 1 1 d . . .  
H31 H 0.4349 0.8713 0.1079 0.032 Uiso 1 1 calc R . . .  
C32 C 0.52237(16) 0.71020(15) 0.12336(14) 0.0239(4) Uani 1 1 d . . .  
H32 H 0.6011 0.7337 0.0785 0.029 Uiso 1 1 calc R . . .  
C33 C 0.50633(15) 0.59379(14) 0.17120(14) 0.0208(3) Uani 1 1 d . . .

H33 H 0.5741 0.5381 0.1592 0.025 Uiso 1 1 calc R . . .  
C34 C 0.50304(14) 0.33274(13) 0.24089(14) 0.0182(3) Uani 1 1 d . . .  
C35 C 0.52368(15) 0.35012(15) 0.12614(14) 0.0218(3) Uani 1 1 d . . .  
H35 H 0.4661 0.4003 0.0840 0.026 Uiso 1 1 calc R . . .  
C36 C 0.62811(16) 0.29419(15) 0.07406(15) 0.0251(4) Uani 1 1 d . . .  
H36 H 0.6418 0.3062 -0.0036 0.030 Uiso 1 1 calc R . . .  
C37 C 0.71231(17) 0.22086(16) 0.13533(16) 0.0293(4) Uani 1 1 d . . .  
H37 H 0.7834 0.1824 0.0997 0.035 Uiso 1 1 calc R . . .  
C38 C 0.69292(18) 0.20358(16) 0.24856(16) 0.0321(4) Uani 1 1 d . . .  
H38 H 0.7512 0.1537 0.2901 0.039 Uiso 1 1 calc R . . .  
C39 C 0.58810(16) 0.25918(15) 0.30190(15) 0.0255(4) Uani 1 1 d . . .  
H39 H 0.5749 0.2468 0.3796 0.031 Uiso 1 1 calc R . . .

loop\_  
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\_geom\_special\_details  
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All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.  
;

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_refine_special_details
;
    Refinement of F^2^ against ALL reflections. The weighted R-factor
wR and
    goodness of fit S are based on F^2^, conventional R-factors R are
based
    on F, with F set to zero for negative F^2^. The threshold expression
of
    F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc.
and is
    not relevant to the choice of reflections for refinement. R-factors
based
    on F^2^ are statistically about twice as large as those based on F,
and R-
    factors based on ALL data will be even larger.
;

_refine_ls_structure_factor_coef   Fsqd
_refine_ls_matrix_type              full
_refine_ls_weighting_scheme         calc
_refine_ls_weighting_details
    'calc w=1/[s^2^(Fo^2^)+(0.0440P)^2^+0.7175P] where
P=(Fo^2^+2Fc^2^)/3'
_atom_sites_solution_primary        direct
_atom_sites_solution_secondary      difmap
_atom_sites_solution_hydrogens      geom
_refine_ls_hydrogen_treatment       constr
_refine_ls_extinction_method       none
_refine_ls_extinction_coeff        ?
_refine_ls_number_reflns            8974
_refine_ls_number_parameters        467
```

_refine_ls_number_restraints	0
_refine_ls_R_factor_all	0.0482
_refine_ls_R_factor_gt	0.0372
_refine_ls_wR_factor_ref	0.0968
_refine_ls_wR_factor_gt	0.0905
_refine_ls_goodness_of_fit_ref	1.051
_refine_ls_restrained_S_all	1.051
_refine_ls_shift/su_max	0.001
_refine_ls_shift/su_mean	0.000
loop_	
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_atom_site_type_symbol	
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_atom_site_fract_y	
_atom_site_fract_z	
_atom_site_U_iso_or_equiv	
_atom_site_adp_type	
_atom_site_occupancy	
_atom_site_symmetry_multiplicity	
_atom_site_calc_flag	
_atom_site_refinement_flags	
_atom_site_disorder_assembly	
_atom_site_disorder_group	
Si1 Si 0.5000 1.0000 0.0000 0.02045(14) Uani 1 2 d S . . .	
P1 P 0.31840(4) 0.68070(3) 0.34130(3) 0.01538(9) Uani 1 1 d . . .	
P2 P 0.08796(4) 0.78241(3) 0.29552(3) 0.01475(9) Uani 1 1 d . . .	
S1 S 0.86494(4) 1.23891(3) 0.06448(3) 0.02817(11) Uani 1 1 d . . .	
S2 S 0.37092(7) 1.30078(5) 0.12107(4) 0.05078(17) Uani 1 1 d . . .	
S3 S 0.62359(5) 0.94311(4) 0.30871(3) 0.03425(12) Uani 1 1 d . . .	
N1 N 0.65492(13) 1.08199(11) 0.01900(10) 0.0245(3) Uani 1 1 d . . .	
N2 N 0.43045(13) 1.11319(11) 0.04125(10) 0.0248(3) Uani 1 1 d . . .	
N3 N 0.54388(14) 0.98198(11) 0.12461(10) 0.0256(3) Uani 1 1 d . . .	
N4 N 0.23412(12) 0.76877(10) 0.32877(9) 0.0184(3) Uani 1 1 d . . .	
N5 N 0.2588(2) 0.56236(19) 0.65821(13) 0.0547(5) Uani 1 1 d . . .	
C1 C 0.74334(15) 1.14802(13) 0.03735(11) 0.0204(3) Uani 1 1 d . . .	
C2 C 0.40439(17) 1.19218(15) 0.07365(12) 0.0262(4) Uani 1 1 d . . .	
C3 C 0.57710(15) 0.96524(13) 0.20195(12) 0.0233(4) Uani 1 1 d . . .	
C4 C 0.23049(14) 0.55580(12) 0.33733(11) 0.0173(3) Uani 1 1 d . . .	
C5 C 0.20931(15) 0.47607(13) 0.25557(11) 0.0198(3) Uani 1 1 d . . .	
H5 H 0.2446 0.4861 0.2023 0.024 Uiso 1 1 calc R . . .	
C6 C 0.13624(15) 0.38191(14) 0.25244(13) 0.0240(4) Uani 1 1 d . . .	
H6 H 0.1209 0.3280 0.1967 0.029 Uiso 1 1 calc R . . .	
C7 C 0.08589(16) 0.36676(14) 0.33052(13) 0.0268(4) Uani 1 1 d . . .	
H7 H 0.0360 0.3025 0.3280 0.032 Uiso 1 1 calc R . . .	
C8 C 0.10805(17) 0.44525(15) 0.41253(13) 0.0285(4) Uani 1 1 d . . .	
H8 H 0.0747 0.4341 0.4663 0.034 Uiso 1 1 calc R . . .	
C9 C 0.17876(16) 0.53985(14) 0.41584(12) 0.0227(3) Uani 1 1 d . . .	
H9 H 0.1921 0.5939 0.4713 0.027 Uiso 1 1 calc R . . .	
C10 C 0.43220(14) 0.71870(13) 0.45743(11) 0.0182(3) Uani 1 1 d . . .	
C11 C 0.48420(16) 0.64533(14) 0.50481(12) 0.0248(4) Uani 1 1 d . . .	
H11 H 0.4572 0.5742 0.4768 0.030 Uiso 1 1 calc R . . .	
C12 C 0.57520(16) 0.67641(15) 0.59272(12) 0.0288(4) Uani 1 1 d . . .	
H12 H 0.6102 0.6266 0.6247 0.035 Uiso 1 1 calc R . . .	
C13 C 0.61460(16) 0.77976(15) 0.63349(12) 0.0264(4) Uani 1 1 d . . .	
H13 H 0.6769 0.8007 0.6935 0.032 Uiso 1 1 calc R . . .	
C14 C 0.56391(16) 0.85354(15) 0.58746(12) 0.0257(4) Uani 1 1 d . . .	
H14 H 0.5915 0.9245 0.6161 0.031 Uiso 1 1 calc R . . .	
C15 C 0.47243(15) 0.82317(13) 0.49917(11) 0.0214(3) Uani 1 1 d . . .	
H15 H 0.4376 0.8734 0.4676 0.026 Uiso 1 1 calc R . . .	
C16 C 0.40912(14) 0.66232(12) 0.24983(11) 0.0174(3) Uani 1 1 d . . .	

C17 C 0.51261(16) 0.60794(14) 0.26247(13) 0.0266(4) Uani 1 1 d . . .  
H17 H 0.5367 0.5833 0.3202 0.032 Uiso 1 1 calc R . . .  
C18 C 0.57983(17) 0.59010(16) 0.19064(14) 0.0332(4) Uani 1 1 d . . .  
H18 H 0.6491 0.5521 0.1987 0.040 Uiso 1 1 calc R . . .  
C19 C 0.54620(17) 0.62765(15) 0.10671(13) 0.0309(4) Uani 1 1 d . . .  
H19 H 0.5931 0.6158 0.0580 0.037 Uiso 1 1 calc R . . .  
C20 C 0.44455(17) 0.68219(14) 0.09419(12) 0.0271(4) Uani 1 1 d . . .  
H20 H 0.4221 0.7080 0.0369 0.033 Uiso 1 1 calc R . . .  
C21 C 0.37500(15) 0.69941(13) 0.16523(11) 0.0210(3) Uani 1 1 d . . .  
H21 H 0.3047 0.7362 0.1562 0.025 Uiso 1 1 calc R . . .  
C22 C 0.08693(15) 0.89299(12) 0.24079(11) 0.0177(3) Uani 1 1 d . . .  
C23 C 0.19877(16) 0.93684(13) 0.21945(11) 0.0214(3) Uani 1 1 d . . .  
H23 H 0.2776 0.9125 0.2387 0.026 Uiso 1 1 calc R . . .  
C24 C 0.19447(18) 1.01672(13) 0.16960(12) 0.0265(4) Uani 1 1 d . . .  
H24 H 0.2704 1.0465 0.1545 0.032 Uiso 1 1 calc R . . .  
C25 C 0.07960(19) 1.05259(14) 0.14207(13) 0.0308(4) Uani 1 1 d . . .  
H25 H 0.0770 1.1067 0.1078 0.037 Uiso 1 1 calc R . . .  
C26 C -0.03166(18) 1.00994(15) 0.16429(13) 0.0311(4) Uani 1 1 d . . .  
H26 H -0.1098 1.0355 0.1458 0.037 Uiso 1 1 calc R . . .  
C27 C -0.02925(16) 0.92999(13) 0.21347(12) 0.0238(4) Uani 1 1 d . . .  
H27 H -0.1055 0.9006 0.2285 0.029 Uiso 1 1 calc R . . .  
C28 C -0.01296(14) 0.67712(12) 0.20528(11) 0.0173(3) Uani 1 1 d . . .  
C29 C -0.02909(16) 0.68146(14) 0.10721(12) 0.0246(4) Uani 1 1 d . . .  
H29 H 0.0058 0.7419 0.0898 0.030 Uiso 1 1 calc R . . .  
C30 C -0.09596(18) 0.59759(15) 0.03586(12) 0.0305(4) Uani 1 1 d . . .  
H30 H -0.1070 0.6007 -0.0305 0.037 Uiso 1 1 calc R . . .  
C31 C -0.14695(16) 0.50896(14) 0.06078(13) 0.0265(4) Uani 1 1 d . . .  
H31 H -0.1925 0.4517 0.0115 0.032 Uiso 1 1 calc R . . .  
C32 C -0.13158(15) 0.50388(13) 0.15761(12) 0.0222(3) Uani 1 1 d . . .  
H32 H -0.1663 0.4430 0.1744 0.027 Uiso 1 1 calc R . . .  
C33 C -0.06537(14) 0.58789(13) 0.23006(12) 0.0190(3) Uani 1 1 d . . .  
H33 H -0.0558 0.5847 0.2963 0.023 Uiso 1 1 calc R . . .  
C34 C 0.01524(15) 0.80780(12) 0.39729(11) 0.0180(3) Uani 1 1 d . . .  
C35 C -0.11777(15) 0.78501(13) 0.38744(12) 0.0216(3) Uani 1 1 d . . .  
H35 H -0.1726 0.7507 0.3266 0.026 Uiso 1 1 calc R . . .  
C36 C -0.16967(16) 0.81296(14) 0.46741(12) 0.0256(4) Uani 1 1 d . . .  
H36 H -0.2598 0.7965 0.4612 0.031 Uiso 1 1 calc R . . .  
C37 C -0.09035(17) 0.86452(14) 0.55565(12) 0.0253(4) Uani 1 1 d . . .  
H37 H -0.1264 0.8839 0.6096 0.030 Uiso 1 1 calc R . . .  
C38 C 0.04187(17) 0.88818(14) 0.56581(12) 0.0254(4) Uani 1 1 d . . .  
H38 H 0.0959 0.9239 0.6265 0.031 Uiso 1 1 calc R . . .  
C39 C 0.09495(16) 0.85937(13) 0.48694(11) 0.0222(3) Uani 1 1 d . . .  
H39 H 0.1854 0.8747 0.4940 0.027 Uiso 1 1 calc R . . .  
C40 C 0.3116(2) 0.76076(18) 0.71762(16) 0.0443(5) Uani 1 1 d . . .  
H40A H 0.3137 0.7927 0.6621 0.067 Uiso 1 1 calc R . . .  
H40B H 0.3957 0.7798 0.7640 0.067 Uiso 1 1 calc R . . .  
H40C H 0.2450 0.7846 0.7491 0.067 Uiso 1 1 calc R . . .  
C41 C 0.2826(2) 0.6499(2) 0.68462(14) 0.0384(5) Uani 1 1 d . . .

loop\_

\_atom\_site\_aniso\_label  
\_atom\_site\_aniso\_U\_11  
\_atom\_site\_aniso\_U\_22  
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Si1 0.0195(3) 0.0177(3) 0.0231(3) 0.0055(3) 0.0034(2) 0.0000(3)  
P1 0.01434(19) 0.0166(2) 0.01457(18) 0.00273(16) 0.00197(14)  
.00346(15)  
P2 0.01480(19) 0.0148(2) 0.01475(18) 0.00333(15) 0.00315(14)

0.00338(15)  
S1 0.0229(2) 0.0201(2) 0.0384(3) 0.00701(19) 0.00325(18) -0.00292(17)  
S2 0.0872(5) 0.0430(3) 0.0428(3) 0.0181(3) 0.0360(3) 0.0412(3)  
S3 0.0393(3) 0.0359(3) 0.0256(2) 0.0108(2) 0.00031(19) 0.0048(2)  
N1 0.0224(7) 0.0218(8) 0.0279(8) 0.0055(6) 0.0043(6) 0.0004(6)  
N2 0.0232(7) 0.0217(8) 0.0299(8) 0.0069(6) 0.0069(6) 0.0021(6)  
N3 0.0259(7) 0.0229(8) 0.0259(8) 0.0062(6) 0.0031(6) 0.0001(6)  
N4 0.0170(6) 0.0172(7) 0.0200(7) 0.0028(5) 0.0029(5) 0.0034(5)  
N5 0.0688(14) 0.0618(15) 0.0312(10) 0.0158(10) 0.0108(9) -0.0053(11)  
C1 0.0223(8) 0.0206(9) 0.0203(8) 0.0077(7) 0.0049(6) 0.0070(7)  
C2 0.0281(9) 0.0312(11) 0.0263(9) 0.0149(8) 0.0122(7) 0.0085(8)  
C3 0.0197(8) 0.0192(9) 0.0299(9) 0.0040(7) 0.0052(7) 0.0015(7)  
C4 0.0143(7) 0.0187(8) 0.0189(7) 0.0060(6) 0.0011(6) 0.0048(6)  
C5 0.0172(7) 0.0214(9) 0.0198(8) 0.0037(7) 0.0020(6) 0.0053(6)  
C6 0.0191(8) 0.0204(9) 0.0276(9) 0.0012(7) -0.0015(7) 0.0027(7)  
C7 0.0187(8) 0.0210(9) 0.0399(10) 0.0108(8) 0.0028(7) 0.0007(7)  
C8 0.0270(9) 0.0317(11) 0.0305(9) 0.0138(8) 0.0097(7) 0.0028(8)  
C9 0.0233(8) 0.0247(9) 0.0204(8) 0.0044(7) 0.0057(7) 0.0043(7)  
C10 0.0147(7) 0.0230(9) 0.0159(7) 0.0028(6) 0.0025(6) 0.0038(6)  
C11 0.0249(9) 0.0216(9) 0.0244(8) 0.0037(7) -0.0017(7) 0.0044(7)  
C12 0.0250(9) 0.0355(11) 0.0237(9) 0.0103(8) -0.0022(7) 0.0047(8)  
C13 0.0187(8) 0.0385(11) 0.0171(8) 0.0025(7) -0.0008(6) 0.0003(8)  
C14 0.0229(8) 0.0269(10) 0.0222(8) -0.0032(7) 0.0038(7) 0.0001(7)  
C15 0.0193(8) 0.0231(9) 0.0212(8) 0.0040(7) 0.0045(6) 0.0036(7)  
C16 0.0153(7) 0.0174(8) 0.0179(7) 0.0011(6) 0.0038(6) 0.0011(6)  
C17 0.0232(8) 0.0334(11) 0.0280(9) 0.0120(8) 0.0086(7) 0.0102(8)  
C18 0.0250(9) 0.0422(12) 0.0398(11) 0.0121(9) 0.0148(8) 0.0171(9)  
C19 0.0286(9) 0.0369(11) 0.0316(10) 0.0061(8) 0.0172(8) 0.0067(8)  
C20 0.0308(9) 0.0293(10) 0.0229(8) 0.0070(7) 0.0093(7) 0.0035(8)  
C21 0.0211(8) 0.0206(9) 0.0208(8) 0.0030(7) 0.0049(6) 0.0039(7)  
C22 0.0219(8) 0.0153(8) 0.0155(7) 0.0024(6) 0.0039(6) 0.0039(6)  
C23 0.0244(8) 0.0178(9) 0.0213(8) 0.0009(7) 0.0068(7) 0.0031(7)  
C24 0.0368(10) 0.0185(9) 0.0249(9) 0.0033(7) 0.0128(7) 0.0002(7)  
C25 0.0479(11) 0.0208(10) 0.0267(9) 0.0105(8) 0.0095(8) 0.0078(8)  
C26 0.0341(10) 0.0293(11) 0.0327(10) 0.0125(8) 0.0046(8) 0.0136(8)  
C27 0.0217(8) 0.0229(9) 0.0273(9) 0.0073(7) 0.0048(7) 0.0051(7)  
C28 0.0148(7) 0.0164(8) 0.0203(8) 0.0029(6) 0.0033(6) 0.0042(6)  
C29 0.0297(9) 0.0222(9) 0.0204(8) 0.0046(7) 0.0044(7) 0.0001(7)  
C30 0.0403(11) 0.0288(10) 0.0185(8) 0.0020(7) 0.0038(7) 0.0009(8)  
C31 0.0235(8) 0.0229(9) 0.0266(9) -0.0036(7) 0.0016(7) 0.0000(7)  
C32 0.0171(8) 0.0184(9) 0.0308(9) 0.0038(7) 0.0064(7) 0.0028(6)  
C33 0.0160(7) 0.0202(9) 0.0214(8) 0.0049(7) 0.0042(6) 0.0048(6)  
C34 0.0204(8) 0.0174(8) 0.0181(7) 0.0054(6) 0.0061(6) 0.0054(6)  
C35 0.0204(8) 0.0236(9) 0.0197(8) 0.0028(7) 0.0036(6) 0.0040(7)  
C36 0.0213(8) 0.0306(10) 0.0279(9) 0.0071(8) 0.0102(7) 0.0064(7)  
C37 0.0313(9) 0.0273(10) 0.0213(8) 0.0058(7) 0.0127(7) 0.0086(8)  
C38 0.0289(9) 0.0277(10) 0.0173(8) 0.0002(7) 0.0049(7) 0.0037(8)  
C39 0.0192(8) 0.0252(9) 0.0210(8) 0.0040(7) 0.0036(6) 0.0024(7)  
C40 0.0429(12) 0.0530(15) 0.0407(12) 0.0124(11) 0.0150(10) 0.0090(11)  
C41 0.0363(11) 0.0578(16) 0.0229(9) 0.0164(10) 0.0084(8) 0.0003(10)

\_geom\_special\_details

;

All esds (except the esd in the dihedral angle between two l.s. planes)

are estimated using the full covariance matrix. The cell esds are taken

into account individually in the estimation of esds in distances, angles

and torsion angles; correlations between esds in cell parameters are only

used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.  
;  
loop\_  
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Si1 N1 1.8232(14) . ?  
Si1 N1 1.8232(14) 2\_675 ?  
Si1 N2 1.8292(15) 2\_675 ?  
Si1 N2 1.8292(15) . ?  
Si1 N3 1.8311(14) . ?  
Si1 N3 1.8311(14) 2\_675 ?  
P1 N4 1.5854(13) . ?  
P1 C16 1.7996(15) . ?  
P1 C10 1.8029(16) . ?  
P1 C4 1.8086(16) . ?  
P2 N4 1.5868(13) . ?  
P2 C28 1.8006(16) . ?  
P2 C34 1.8036(15) . ?  
P2 C22 1.8044(16) . ?  
S1 C1 1.6070(17) . ?  
S2 C2 1.608(2) . ?  
S3 C3 1.6114(17) . ?  
N1 C1 1.170(2) . ?  
N2 C2 1.168(2) . ?  
N3 C3 1.170(2) . ?  
N5 C41 1.146(3) . ?  
C4 C5 1.400(2) . ?  
C4 C9 1.403(2) . ?  
C5 C6 1.394(2) . ?  
C5 H5 0.9500 . ?  
C6 C7 1.386(2) . ?  
C6 H6 0.9500 . ?  
C7 C8 1.391(3) . ?  
C7 H7 0.9500 . ?  
C8 C9 1.388(2) . ?  
C8 H8 0.9500 . ?  
C9 H9 0.9500 . ?  
C10 C15 1.397(2) . ?  
C10 C11 1.402(2) . ?  
C11 C12 1.390(2) . ?  
C11 H11 0.9500 . ?  
C12 C13 1.381(3) . ?  
C12 H12 0.9500 . ?  
C13 C14 1.391(3) . ?  
C13 H13 0.9500 . ?  
C14 C15 1.396(2) . ?  
C14 H14 0.9500 . ?  
C15 H15 0.9500 . ?  
C16 C21 1.398(2) . ?  
C16 C17 1.400(2) . ?  
C17 C18 1.386(2) . ?  
C17 H17 0.9500 . ?  
C18 C19 1.392(3) . ?  
C18 H18 0.9500 . ?

C19 C20 1.385(3) . ?  
C19 H19 0.9500 . ?  
C20 C21 1.394(2) . ?  
C20 H20 0.9500 . ?  
C21 H21 0.9500 . ?  
C22 C23 1.394(2) . ?  
C22 C27 1.405(2) . ?  
C23 C24 1.396(2) . ?  
C23 H23 0.9500 . ?  
C24 C25 1.386(3) . ?  
C24 H24 0.9500 . ?  
C25 C26 1.389(3) . ?  
C25 H25 0.9500 . ?  
C26 C27 1.391(2) . ?  
C26 H26 0.9500 . ?  
C27 H27 0.9500 . ?  
C28 C33 1.400(2) . ?  
C28 C29 1.404(2) . ?  
C29 C30 1.385(2) . ?  
C29 H29 0.9500 . ?  
C30 C31 1.389(3) . ?  
C30 H30 0.9500 . ?  
C31 C32 1.389(2) . ?  
C31 H31 0.9500 . ?  
C32 C33 1.392(2) . ?  
C32 H32 0.9500 . ?  
C33 H33 0.9500 . ?  
C34 C35 1.398(2) . ?  
C34 C39 1.399(2) . ?  
C35 C36 1.396(2) . ?  
C35 H35 0.9500 . ?  
C36 C37 1.383(2) . ?  
C36 H36 0.9500 . ?  
C37 C38 1.390(2) . ?  
C37 H37 0.9500 . ?  
C38 C39 1.392(2) . ?  
C38 H38 0.9500 . ?  
C39 H39 0.9500 . ?  
C40 C41 1.451(3) . ?  
C40 H40A 0.9800 . ?  
C40 H40B 0.9800 . ?  
C40 H40C 0.9800 . ?  
  
loop\_  
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N1 Si1 N1 180.00(9) . 2\_675 ?  
N1 Si1 N2 90.41(6) . 2\_675 ?  
N1 Si1 N2 89.59(6) 2\_675 2\_675 ?  
N1 Si1 N2 89.59(6) . . ?  
N1 Si1 N2 90.41(6) 2\_675 . ?  
N2 Si1 N2 180.00(9) 2\_675 . ?  
N1 Si1 N3 90.11(6) . . ?  
N1 Si1 N3 89.89(6) 2\_675 . ?  
N2 Si1 N3 89.94(7) 2\_675 . ?  
N2 Si1 N3 90.06(7) . . ?

N1 Si1 N3 89.89(6) . 2\_675 ?  
N1 Si1 N3 90.11(6) 2\_675 2\_675 ?  
N2 Si1 N3 90.06(7) 2\_675 2\_675 ?  
N2 Si1 N3 89.94(7) . 2\_675 ?  
N3 Si1 N3 180.00(9) . 2\_675 ?  
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C16 P1 C10 107.54(7) . . ?  
N4 P1 C4 115.85(7) . . ?  
C16 P1 C4 106.52(7) . . ?  
C10 P1 C4 106.75(7) . . ?  
N4 P2 C28 115.14(7) . . ?  
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C28 P2 C34 108.95(7) . . ?  
N4 P2 C22 107.06(7) . . ?  
C28 P2 C22 106.03(7) . . ?  
C34 P2 C22 107.66(7) . . ?  
C1 N1 Si1 168.20(14) . . ?  
C2 N2 Si1 170.13(15) . . ?  
C3 N3 Si1 175.12(15) . . ?  
P1 N4 P2 139.97(9) . . ?  
N1 C1 S1 179.03(15) . . ?  
N2 C2 S2 178.44(17) . . ?  
N3 C3 S3 179.53(18) . . ?  
C5 C4 C9 119.57(15) . . ?  
C5 C4 P1 120.90(12) . . ?  
C9 C4 P1 119.49(13) . . ?  
C6 C5 C4 119.84(15) . . ?  
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C4 C5 H5 120.1 . . ?  
C7 C6 C5 120.16(16) . . ?  
C7 C6 H6 119.9 . . ?  
C5 C6 H6 119.9 . . ?  
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C6 C7 H7 119.9 . . ?  
C8 C7 H7 119.9 . . ?  
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C7 C8 H8 120.0 . . ?  
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C8 C9 H9 120.0 . . ?  
C4 C9 H9 120.0 . . ?  
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C16 C17 H17 120.0 . . ?  
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C23 C22 C27 120.14(15) . . ?  
C23 C22 P2 120.08(12) . . ?  
C27 C22 P2 119.62(12) . . ?  
C22 C23 C24 119.70(16) . . ?  
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C25 C24 H24 120.0 . . ?  
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C24 C25 C26 120.36(16) . . ?  
C24 C25 H25 119.8 . . ?  
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C25 C26 C27 120.32(17) . . ?  
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C26 C27 C22 119.39(16) . . ?  
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C32 C33 H33 120.0 . . ?  
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C39 C34 P2 117.91(12) . . ?  
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H40A C40 H40C 109.5 . . ?  
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#### 4d) Crystallographic information file for compound 4 (data\_chppx70\_0m)

data\_ch1ppx70\_0n

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'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'  
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\_symmetry\_cell\_setting Orthorhombic  
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  '-x, -y, z+1/2'
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  'x+1/2, -y+1/2, z'

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  _cell_formula_units_Z      4
  _cell_measurement_temperature   120(2)
  _cell_measurement_reflns_used    2074
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  _exptl_special_details
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_computing_cell_refinement        'Bruker SMART'
_computing_data_reduction         'Bruker SAINT'
_computing_structure_solution     'Bruker SHELXTL'
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_refine_special_details
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    Refinement of F^2^ against ALL reflections. The weighted R-factor
wR and
    goodness of fit S are based on F^2^, conventional R-factors R are
based
    on F, with F set to zero for negative F^2^. The threshold expression
of
    F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc.
and is
    not relevant to the choice of reflections for refinement. R-factors
based
    on F^2^ are statistically about twice as large as those based on F,
and R-
    factors based on ALL data will be even larger.
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_refine_ls_matrix_type            full
_refine_ls_weighting_scheme       calc
_refine_ls_weighting_details
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P=(Fo^2^+2Fc^2^)/3'
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_refine_ls_solution_secondary      difmap
_refine_ls_solution_hydrogens      geom
_refine_ls_hydrogen_treatment      constr
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_refine_ls_wR_factor_ref            0.0862
_refine_ls_wR_factor_gt              0.0813
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_atom_site_type_symbol
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\_atom\_site\_occupancy  
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\_atom\_site\_refinement\_flags  
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O1 O 0.9965(4) 1.11185(12) -0.12840(17) 0.0396(6) Uani 1 1 d . . .  
O2 O 0.6724(7) 1.17369(15) 0.2283(2) 0.0936(15) Uani 1 1 d . . .  
O3 O 1.2098(4) 1.01558(15) 0.23571(17) 0.0377(6) Uani 1 1 d . . .  
O4 O 0.2697(4) 0.94817(13) -0.08445(17) 0.0368(6) Uani 1 1 d . . .  
N1 N 0.8277(4) 1.03364(13) -0.02370(18) 0.0261(6) Uani 1 1 d . . .  
N2 N 0.6936(4) 1.06010(13) 0.15093(18) 0.0265(6) Uani 1 1 d . . .  
N3 N 0.9848(4) 0.97105(12) 0.12362(17) 0.0231(6) Uani 1 1 d . . .  
N4 N 0.4999(4) 0.97998(12) 0.03167(18) 0.0248(6) Uani 1 1 d . . .  
N5 N 0.6607(4) 0.91359(13) 0.17638(15) 0.0193(5) Uani 1 1 d . . .  
N6 N 0.7874(4) 0.88855(13) 0.00752(17) 0.0201(5) Uani 1 1 d . . .  
N7 N 0.6940(5) 0.21322(17) -0.0030(2) 0.0422(8) Uani 1 1 d . . .  
C1 C 0.9120(5) 1.07256(15) -0.0746(2) 0.0249(7) Uani 1 1 d . . .  
C2 C 0.6841(6) 1.11665(17) 0.1879(2) 0.0383(9) Uani 1 1 d . . .  
C3 C 1.0938(5) 0.99370(15) 0.1811(2) 0.0233(6) Uani 1 1 d . . .  
C4 C 0.3881(5) 0.96417(15) -0.0272(2) 0.0233(7) Uani 1 1 d . . .  
C5 C 0.5927(5) 0.92958(16) 0.26193(19) 0.0239(7) Uani 1 1 d . . .  
H024 H 0.5770 0.9799 0.2794 0.029 Uiso 1 1 calc R . .  
C6 C 0.5437(5) 0.87430(16) 0.3271(2) 0.0240(7) Uani 1 1 d . . .  
H021 H 0.4952 0.8878 0.3875 0.029 Uiso 1 1 calc R . .  
C7 C 0.5649(5) 0.80120(15) 0.3047(2) 0.0223(6) Uani 1 1 d . . .  
H022 H 0.5340 0.7638 0.3494 0.027 Uiso 1 1 calc R . .  
C8 C 0.6338(5) 0.78258(15) 0.2136(2) 0.0206(6) Uani 1 1 d . . .  
C9 C 0.6535(4) 0.70856(15) 0.1787(2) 0.0217(6) Uani 1 1 d . . .  
H020 H 0.6242 0.6681 0.2191 0.026 Uiso 1 1 calc R . .  
C10 C 0.7139(4) 0.69512(16) 0.0880(2) 0.0225(7) Uani 1 1 d . . .  
H012 H 0.7239 0.6456 0.0662 0.027 Uiso 1 1 calc R . .  
C11 C 0.7622(4) 0.75486(16) 0.0258(2) 0.0205(6) Uani 1 1 d . . .  
C12 C 0.8234(5) 0.74652(17) -0.0691(2) 0.0239(7) Uani 1 1 d . . .  
H016 H 0.8341 0.6987 -0.0965 0.029 Uiso 1 1 calc R . .  
C13 C 0.8671(5) 0.80845(16) -0.1210(2) 0.0250(7) Uani 1 1 d . . .  
H017 H 0.9106 0.8035 -0.1844 0.030 Uiso 1 1 calc R . .  
C14 C 0.8479(5) 0.87919(16) -0.0808(2) 0.0241(7) Uani 1 1 d . . .  
H027 H 0.8789 0.9214 -0.1178 0.029 Uiso 1 1 calc R . .  
C15 C 0.7457(4) 0.82757(15) 0.05914(19) 0.0176(6) Uani 1 1 d . . .  
C16 C 0.6796(4) 0.84079(15) 0.15277(19) 0.0180(6) Uani 1 1 d . . .  
C17 C 0.7200(6) 0.3363(2) 0.0888(3) 0.0484(11) Uani 1 1 d . . .  
H11A H 0.8082 0.3696 0.0558 0.073 Uiso 1 1 calc R . .  
H11B H 0.7683 0.3264 0.1527 0.073 Uiso 1 1 calc R . .  
H11C H 0.5936 0.3594 0.0930 0.073 Uiso 1 1 calc R . .  
C18 C 0.7053(5) 0.26764(19) 0.0371(2) 0.0308(8) Uani 1 1 d . . .

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O1 0.0428(16) 0.0338(13) 0.0424(13) 0.0117(11) 0.0072(12) -0.0069(12)  
O2 0.219(5) 0.0284(15) 0.0331(14) -0.0080(12) -0.011(2) 0.024(2)  
O3 0.0391(17) 0.0482(15) 0.0258(12) -0.0096(10) -0.0056(12) -  
0.0045(12)  
O4 0.0418(16) 0.0389(14) 0.0298(12) 0.0051(11) -0.0080(12) -  
0.0087(11)  
N1 0.0370(17) 0.0186(12) 0.0228(12) 0.0015(10) -0.0004(13) -  
0.0025(12)  
N2 0.0348(17) 0.0175(13) 0.0271(13) -0.0004(10) 0.0015(13) 0.0012(11)  
N3 0.0295(16) 0.0192(13) 0.0205(12) -0.0009(10) 0.0003(12) -0.0021(11)  
N4 0.0290(16) 0.0216(13) 0.0237(12) 0.0014(10) -0.0032(13) 0.0035(11)  
N5 0.0231(14) 0.0172(11) 0.0175(11) -0.0004(9) 0.0002(11) 0.0008(10)  
N6 0.0217(15) 0.0192(12) 0.0194(11) -0.0002(9) -0.0005(10) -  
0.0015(10)  
N7 0.053(2) 0.0345(17) 0.0386(16) -0.0031(13) -0.0062(16) 0.0005(15)  
C1 0.0311(19) 0.0175(14) 0.0262(14) -0.0013(13) -0.0043(15)  
0.0016(13)  
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C6 0.0269(18) 0.0263(15) 0.0189(13) 0.0000(11) 0.0016(13) -0.0006(14)  
C7 0.0224(17) 0.0219(14) 0.0226(13) 0.0018(12) -0.0027(13) -  
0.0019(13)  
C8 0.0200(17) 0.0201(14) 0.0216(14) 0.0018(11) -0.0027(13) 0.0007(12)  
C9 0.0200(17) 0.0175(14) 0.0276(15) 0.0051(11) -0.0027(13) -  
0.0018(12)  
C10 0.0229(17) 0.0153(13) 0.0294(15) -0.0032(12) -0.0037(14)  
0.0018(11)  
C11 0.0209(16) 0.0180(14) 0.0226(14) -0.0044(11) -0.0021(14)  
0.0007(12)  
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0.0039(12)  
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0.0011(13)  
C14 0.0296(19) 0.0242(15) 0.0185(13) 0.0007(12) 0.0022(14) -  
0.0016(13)  
C15 0.0190(14) 0.0153(13) 0.0185(15) -0.0018(9) -0.0019(12) -  
0.0016(12)  
C16 0.0182(16) 0.0176(14) 0.0182(13) -0.0013(10) -0.0030(13)  
0.0004(11)  
C17 0.064(3) 0.0320(19) 0.049(2) -0.0076(17) -0.004(2) 0.0006(18)  
C18 0.033(2) 0.0299(19) 0.0295(16) 0.0063(14) -0.0005(15) 0.0042(14)

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All esds (except the esd in the dihedral angle between two l.s.  
planes)  
are estimated using the full covariance matrix. The cell esds are  
taken  
into account individually in the estimation of esds in distances,  
angles  
and torsion angles; correlations between esds in cell parameters are  
only  
used when they are defined by crystal symmetry. An approximate  
(isotropic)  
treatment of cell esds is used for estimating esds involving l.s.  
planes.  
;  
loop\_

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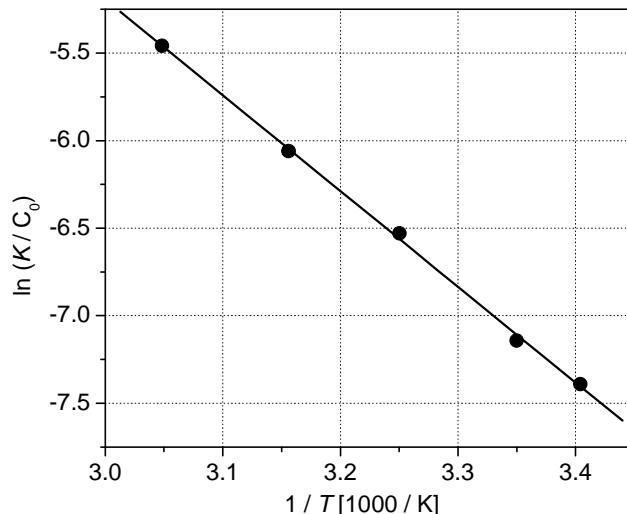
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## 6) Variable temperature NMR spectra of compound 3

$^1\text{H}$  NMR spectra were recorded of a solution of compound **3** in acetonitrile- $d^3$ ,  $c = 0.021(+/-0.006)$  mol dm $^{-3}$ . The thermodynamic equilibrium constants were calculated from the ratios of integrated peaks arising from **3** and bpy. The following data was obtained:

Exp. No	T [°C]	thermodynamic equilibrium constant
1	20.6	$6.17 \times 10^{-4}$
2	54.9	$4.26 \times 10^{-3}$
3	43.7	$2.33 \times 10^{-3}$
4	34.5	$1.46 \times 10^{-3}$
5	25.35	$7.90 \times 10^{-4}$



The standard enthalpy ( $\Delta H^\circ$ ), entropy ( $\Delta S^\circ$ ) and free energy ( $\Delta G^\circ$ ) of the equilibrium reaction,  $\text{Si}(\text{NCO})_4(\text{bpy}) \leftrightarrow \text{Si}(\text{NCO})_4 + \text{bpy}$ , were obtained from the slope and intercept of a van't Hoff plot,  $\Delta H^\circ = +45 \text{ kJ mol}^{-1}$ ,  $\Delta S^\circ = +93 \text{ J mol}^{-1} \text{ K}^{-1}$ ,  $\Delta G^\circ = +18 \text{ kJ mol}^{-1}$  ( $c_0 = 1 \text{ mol dm}^{-3}$ ). This results in  $K' = 7.9 \times 10^{-3} \text{ mol dm}^{-3}$  at 298 K and the complex stability constant  $\text{pK} = 3$  for  $\text{Si}(\text{NCO})_4(\text{bpy})$ , **3**.