# A Novel Heteroleptic Paddlewheel Diruthenium Bicyclic Guanidinate Complex: Synthesis, Structure, and Scope

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#### Procedure for the synthesis of TBO

## (1,1'-(benzylazanediyl)bis(3,3-dimethylbutan-2-one)), 1.<sup>1</sup>

Benzylamine (545 ml, 5 mmol, 1 equiv.) and 1-bromo-3,3-dimethylbutan-2-one (990mg, 5.5 mmol, 1.1 equiv.) were dissolved in 20 ml acetonitrile. Potassium carbonate (1.4 g, 10 mmol, 2.0 equiv.) was added and then brought to reflux for 6 hours. Then the mixture was filtered and the washed twice with MeCN. Removal of the solvent gave crude product that was further purified by column chromatography with gradient elution (hexane/EA; 50/1 to 20/1). After removing the solvent *in vacuo*, product (1.23 g) was obtained as colorless oil.

(1) colorless oil; 81% yield; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>, ppm):  $\delta$  = 7.35-7.19(m, 5H), 3.83 (s, 2H), 3.76 (s, 4H), 1.04 (s, 18H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>, ppm):  $\delta$  = 214.5, 138.8, 128.9, 128.2, 127.2, 77.4, 77.0, 76.6, 57.7, 56.1, 43.3, 26.3; IR (film): 3022, 1217, 760 cm<sup>-1</sup>; LRMS (ESI) m/z 326.2 (M+Na<sup>+</sup>); HRMS (ESI) m/z 326.2102 (M+Na<sup>+</sup>), calc. for C<sub>19</sub>H<sub>29</sub>O<sub>2</sub>N<sub>1</sub>Na<sub>1</sub> 326.2091.

## (1-(2-(hydroxyimino)-3,3-dimethylbutylamino)-3,3-dimethylbutan-2-one oxime), 2.<sup>2</sup>

tBu N H N

 $\ddot{N}$  H  $\ddot{N}$  A flame dried round bottom flask under nitrogen, was charged with OH OH OH Pd/C (424mg, 0.4mmol, 0.1equiv.), MeOH, and 1 (1.21g, 4mmol, 1equiv.). H<sub>2</sub> was bubbled into the reaction mixture for 10 minutes, sealed and stirred to react overnight. After the complete consumption of starting material, the suspension was filtered and the solvent removed to afford the oxime as colorless oil. The crude oil was used for the next step without further purification.

Hydroxylamine hydrochloride (1.4g, 20mmol, 5 equiv.), potassium hydroxide (1.12g, 20mmol, 5 equiv.) and debenzylated 2 (850mg, 4mmol, 1equiv.) were dissolved in EtOH. The solution was brought to reflux for 2 hours. Water was added after the subsequent removal of solvent and extracted with EA thrice. The combined organic layer was dried over sodium sulfate and solvent

<sup>&</sup>lt;sup>1</sup> H. Orsolya, C. Marco, Q. Silvio, S. Ian, P. Gianluca, Adv. Synth. Catal., 2005, 347, 677 –688.

<sup>&</sup>lt;sup>2</sup> Y. K. Joon, L. Tom, Org. Lett., 2005, 7, 4391-4393.

was removed. The crude was purified with column chromatography with gradient elution (hexane/EA; 8/1 to 2/1)to yield 855mg of product as white solid.

(2) white solid; 88% yield; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>, ppm):  $\delta = 3.65(s, 2H)$ , 3.51 (s, 1H), 1.17 (s, 9H), 1.15 (s, 9H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>, ppm):  $\delta = 162.9$ , 44.5, 37.1, 27.4; IR (film): 3022, 1217, 760 cm<sup>-1</sup>; LRMS (ESI) m/z 244.1 (M+H<sup>+</sup>); HRMS (ESI) m/z 244.2025 (M+H<sup>+</sup>), calc. for C<sub>12</sub>H<sub>26</sub>O<sub>2</sub>N<sub>3</sub> 244.2020.

#### (cis-3,7-di-tert-butyl)-1,4,6-triazabicyclo[3.3.0]oct-4-ene, 3.<sup>3</sup>

**2** (972mg, 4 mmol, 1 equiv.) was added slowly at  $-10^{\circ}$ C to a suspension of LiAlH<sub>4</sub> (605mg, 16 mmol, 4 equiv.) in 30 ml THF. The reaction mixture was heated to reflux overnight, then cooled to 0°C and quenched by sequential addition of water (3 ml) and 10% aqueous NaOH (3 ml). The mixture was then stirred at ambient temperature for 30 min. Filtration of the mixture and drying of the filtrate over over Na<sub>2</sub>SO<sub>4</sub> and removal of solvent to yield triamine as yellow oil. The triamine was used in the next step without further purification.

The crude free triamine (500 mg, 2.6 mmol) was dissolved in nitromethane (10 ml). Dimethyl trithiocarbonate (360  $\mu$ l, 3.3 mmol, 1.25 equiv.) in nitromethane (1 ml) was added to the mixture slowly, followed by refluxing at 110 °C for 2 h. The mixture was then cooled to room temperature. Acetic acid (610  $\mu$ l, 10.6 mmol, 4 equiv.) and MeI (490  $\mu$ l, 5.3 mmol, 2 equiv.) were added. The mixture was refluxed again at 110 °C for 3 h, then left stirring at room temperature overnight. CH<sub>2</sub>Cl<sub>2</sub> (10 ml) was added to dilute the reaction mixture and the solvent was removed under reduced pressure. CH<sub>2</sub>Cl<sub>2</sub> (10 ml) was added and the solution was loaded onto a plug of silica gel. It was eluted with copious amount of CH<sub>2</sub>Cl<sub>2</sub> to flush out a dark colored portion of impurities and then CH<sub>2</sub>Cl<sub>2</sub>/MeOH (19:1) was used to recover the product, which was assumed to be a HI salt. Acid-base workup gave the free guanidine as 1:1 *cis-trans* mixture (a pale yellow solid, 310 mg, 61 % yield calculated from guanidine). **3** (88mg) was obtained as pure *cis* isomer from recrystallization wth diethyl ether.

<sup>&</sup>lt;sup>3</sup> W.-P. Ye, D.-S. Leow, S.L.M. Goh, C.-T. Tan, C.-H. Chian, C.-H. Tan, *Tetrahedron Letters*, 2006, 47, 1007–1010.

(3) off-white solid; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>, ppm):  $\delta = 4.09(s, 2H)$ , 3. 94-3. 88 (t, 2H), 3.25-3.19 (s, 2H), 2.84-2.79 (t, 2H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>, ppm):  $\delta = 167.6, 73.5, 49.3, 33.4,$ 25.8; IR (film): 3022, 1217, 760 cm<sup>-1</sup>; LRMS (ESI) m/z 224.3 (M+H<sup>+</sup>); HRMS (ESI) m/z 224.2115 (M+H<sup>+</sup>), calc. for C<sub>13</sub>H<sub>26</sub>N<sub>3</sub> 224.2121.



110 1 (ppm)

100

170 160 150 140 130 120

190 180













Empirical formula	mpirical formula C36 H63 Cl N9 O4 Ru2		
ormula weight 923.54			
Temperature	223(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	P2(1)/n		
Unit cell dimensions	a = 11.3725(13) Å	α= 90°.	
	b = 34.635(4)  Å	β=117.046(2)°.	
	c = 12.3568(13) Å	$\gamma = 90^{\circ}$ .	
Volume	4334.9(8) Å <sup>3</sup>		
Ζ	4		
Density (calculated)	1.415 Mg/m <sup>3</sup>		
Absorption coefficient	0.804 mm <sup>-1</sup>		
F(000)	1916		
Crystal size	0.60 x 0.20 x 0.10 mm <sup>3</sup>		
Theta range for data collection	1.94 to 27.50°.		
Index ranges	-14<=h<=14, -45<=k<=40, -10<=l<=16		
Reflections collected	30059		
Independent reflections	9946 [R(int) = 0.0688]		
Completeness to theta = $27.50^{\circ}$	99.9 %		
Absorption correction	Semi-empirical from equivaler	nts	
Max. and min. transmission	0.9239 and 0.6439		
Refinement method	Full-matrix least-squares on F <sup>2</sup>	2	
Data / restraints / parameters	9946 / 0 / 472		
Goodness-of-fit on F <sup>2</sup>	1.236		
Final R indices [I>2sigma(I)]	R1 = 0.0861, $wR2 = 0.1767$		
R indices (all data)	) $R1 = 0.1056, wR2 = 0.1849$		
argest diff. peak and hole 2.182 and -1.668 e.Å <sup>-3</sup>			

Table 1. Crystal data and structure refinement for Ru<sub>2</sub>(TBO)<sub>2</sub>(OAc)<sub>2</sub>Cl•3MeCN (1•3MeCN).

	Х	у	Z	U(eq)
Ru(1)	1372(1)	1042(1)	4419(1)	18(1)
Ru(2)	3090(1)	1110(1)	3876(1)	18(1)
Cl(1)	-474(2)	895(1)	4938(2)	30(1)
O(1)	1901(4)	469(1)	4803(4)	26(1)
O(2)	3414(4)	520(1)	4128(4)	26(1)
O(3)	145(4)	891(1)	2644(4)	29(1)
O(4)	1732(5)	1005(1)	2098(4)	31(1)
N(1)	2697(5)	1153(1)	6198(4)	21(1)
N(2)	4456(5)	1166(1)	5647(4)	21(1)
N(3)	4770(5)	931(2)	7481(5)	28(1)
N(4)	871(5)	1602(1)	3980(5)	25(1)
N(5)	2742(5)	1684(1)	3634(5)	24(1)
N(6)	1664(7)	2200(2)	3925(6)	37(1)
N(7)	4719(7)	1006(2)	3159(6)	46(2)
C(1)	3945(6)	1103(2)	6411(6)	25(1)
C(2)	2653(7)	992(2)	7315(6)	27(1)
C(3)	3885(7)	735(2)	7891(7)	38(2)
C(4)	5779(7)	736(2)	7264(6)	35(2)
C(5)	5833(7)	991(2)	6255(5)	27(1)
C(6)	2588(8)	1305(2)	8166(6)	39(2)
C(7)	3872(9)	1543(3)	8759(8)	57(2)
C(8)	1468(9)	1583(2)	7438(8)	54(2)
C(9)	2340(10)	1111(3)	9125(7)	60(3)
C(10)	6921(7)	1300(2)	6749(6)	33(2)
C(11)	6831(10)	1553(3)	7715(9)	65(3)
C(12)	6842(9)	1547(3)	5701(8)	55(2)
C(13)	8255(8)	1097(3)	7293(8)	52(2)
C(14)	1755(7)	1807(2)	3812(6)	30(2)
C(15)	15(9)	1886(2)	4208(8)	48(2)
C(16)	912(9)	2248(2)	4609(9)	51(2)

Table 2. Atomic coordinates (  $x \ 10^4$ ) and equivalent isotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for 8610. U(eq) is defined as one third of the trace of the orthogonalized U<sup>ij</sup> tensor.

C(17)	3014(8)	2335(2)	4283(7)	42(2)
C(18)	3518(7)	2039(2)	3672(6)	30(2)
C(19)	-1320(8)	1953(3)	3119(9)	58(3)
C(20)	-2104(9)	1578(3)	2739(9)	63(3)
C(21)	-2131(11)	2233(3)	3454(13)	98(5)
C(22)	-1181(10)	2101(4)	2019(11)	99(5)
C(23)	3368(8)	2164(2)	2419(7)	35(2)
C(24)	3842(11)	1837(2)	1891(8)	58(3)
C(25)	4263(9)	2512(2)	2601(9)	52(2)
C(26)	1962(9)	2264(3)	1544(8)	59(2)
C(27)	2764(7)	327(2)	4543(6)	26(1)
C(28)	3052(8)	-95(2)	4750(7)	38(2)
C(29)	559(7)	918(2)	1858(5)	30(2)
C(30)	-409(9)	841(3)	564(7)	51(2)
C(31)	5366(7)	881(2)	2785(7)	36(2)
C(32)	6177(9)	712(3)	2268(9)	62(3)
N(1S)	3498(10)	-171(3)	-122(9)	96(3)
C(1S)	3119(10)	-3(3)	429(8)	53(2)
C(2S)	2598(11)	198(3)	1134(9)	64(3)
N(2S)	-208(10)	157(3)	8208(8)	76(3)
C(3S)	285(8)	89(2)	7616(8)	45(2)
C(4S)	912(8)	2(2)	6855(8)	44(2)

Ru(1)-N(4)	2.027(5)
Ru(1)-N(1)	2.059(5)
Ru(1)-O(3)	2.061(4)
Ru(1)-O(1)	2.066(4)
Ru(1)-Ru(2)	2.3493(7)
Ru(1)-Cl(1)	2.5097(17)
Ru(2)-N(5)	2.022(5)
Ru(2)-N(2)	2.037(5)
Ru(2)-O(4)	2.063(4)
Ru(2)-O(2)	2.074(4)
Ru(2)-N(7)	2.416(8)
O(1)-C(27)	1.262(7)
O(2)-C(27)	1.264(8)
O(3)-C(29)	1.260(8)
O(4)-C(29)	1.265(8)
N(1)-C(1)	1.332(8)
N(1)-C(2)	1.509(8)
N(2)-C(1)	1.331(8)
N(2)-C(5)	1.520(8)
N(3)-C(1)	1.362(8)
N(3)-C(4)	1.459(9)
N(3)-C(3)	1.483(8)
N(4)-C(14)	1.322(8)
N(4)-C(15)	1.496(8)
N(5)-C(14)	1.310(8)
N(5)-C(18)	1.503(8)
N(6)-C(14)	1.376(8)
N(6)-C(16)	1.459(9)
N(6)-C(17)	1.467(10)
N(7)-C(31)	1.118(10)
C(2)-C(3)	1.535(10)
C(2)-C(6)	1.536(9)
C(4)-C(5)	1.552(9)

Table 3. Bond lengths [Å] and angles  $[\circ]$  for 8610.

C(5)-C(10)	1.537(9)
C(6)-C(9)	1.497(10)
C(6)-C(8)	1.522(12)
C(6)-C(7)	1.542(11)
C(10)-C(12)	1.520(10)
C(10)-C(11)	1.521(11)
C(10)-C(13)	1.524(10)
C(15)-C(19)	1.520(12)
C(15)-C(16)	1.551(12)
C(17)-C(18)	1.531(10)
C(18)-C(23)	1.541(9)
C(19)-C(21)	1.518(13)
C(19)-C(20)	1.524(13)
C(19)-C(22)	1.526(13)
C(23)-C(26)	1.508(11)
C(23)-C(24)	1.523(10)
C(23)-C(25)	1.525(10)
C(27)-C(28)	1.494(9)
C(29)-C(30)	1.496(9)
C(31)-C(32)	1.462(11)
N(1S)-C(1S)	1.120(12)
C(1S)-C(2S)	1.438(13)
N(2S)-C(3S)	1.131(11)
C(3S)-C(4S)	1.447(11)
N(4)-Ru(1)-N(1)	95.0(2)
N(4)-Ru(1)-O(3)	89.2(2)
N(1)-Ru(1)-O(3)	175.07(19)
N(4)-Ru(1)-O(1)	177.46(19)
N(1)-Ru(1)-O(1)	86.50(18)
O(3)-Ru(1)-O(1)	89.20(18)
N(4)-Ru(1)-Ru(2)	89.81(15)
N(1)-Ru(1)-Ru(2)	89.23(14)
O(3)-Ru(1)-Ru(2)	88.23(13)
O(1)-Ru(1)-Ru(2)	88.18(12)

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N(4)-Ru(1)-Cl(1)	95.53(15)
N(1)-Ru(1)-Cl(1)	93.22(15)
O(3)-Ru(1)-Cl(1)	88.90(14)
O(1)-Ru(1)-Cl(1)	86.41(13)
Ru(2)-Ru(1)-Cl(1)	173.90(4)
N(5)-Ru(2)-N(2)	93.9(2)
N(5)-Ru(2)-O(4)	90.8(2)
N(2)-Ru(2)-O(4)	175.32(19)
N(5)-Ru(2)-O(2)	178.86(19)
N(2)-Ru(2)-O(2)	86.29(18)
O(4)-Ru(2)-O(2)	89.03(18)
N(5)-Ru(2)-Ru(1)	90.46(14)
N(2)-Ru(2)-Ru(1)	91.67(14)
O(4)-Ru(2)-Ru(1)	88.26(13)
O(2)-Ru(2)-Ru(1)	88.41(12)
N(5)-Ru(2)-N(7)	102.9(2)
N(2)-Ru(2)-N(7)	94.1(2)
O(4)-Ru(2)-N(7)	84.9(2)
O(2)-Ru(2)-N(7)	78.18(19)
Ru(1)-Ru(2)-N(7)	165.02(16)
C(27)-O(1)-Ru(1)	119.6(4)
C(27)-O(2)-Ru(2)	119.0(4)
C(29)-O(3)-Ru(1)	119.5(4)
C(29)-O(4)-Ru(2)	119.4(4)
C(1)-N(1)-C(2)	104.4(5)
C(1)-N(1)-Ru(1)	112.6(4)
C(2)-N(1)-Ru(1)	126.5(4)
C(1)-N(2)-C(5)	105.6(5)
C(1)-N(2)-Ru(2)	112.4(4)
C(5)-N(2)-Ru(2)	127.2(4)
C(1)-N(3)-C(4)	106.1(5)
C(1)-N(3)-C(3)	105.0(5)
C(4)-N(3)-C(3)	124.3(6)
C(14)-N(4)-C(15)	106.5(5)
C(14)-N(4)-Ru(1)	114.5(4)

C(15)-N(4)-Ru(1)	134.9(4)
C(14)-N(5)-C(18)	105.2(5)
C(14)-N(5)-Ru(2)	115.0(4)
C(18)-N(5)-Ru(2)	136.7(4)
C(14)-N(6)-C(16)	105.3(6)
C(14)-N(6)-C(17)	103.2(6)
C(16)-N(6)-C(17)	126.7(7)
C(31)-N(7)-Ru(2)	165.2(6)
N(2)-C(1)-N(1)	127.5(6)
N(2)-C(1)-N(3)	115.4(6)
N(1)-C(1)-N(3)	116.7(6)
N(1)-C(2)-C(3)	103.6(5)
N(1)-C(2)-C(6)	113.5(5)
C(3)-C(2)-C(6)	114.2(6)
N(3)-C(3)-C(2)	101.1(5)
N(3)-C(4)-C(5)	101.8(5)
N(2)-C(5)-C(10)	112.5(5)
N(2)-C(5)-C(4)	102.4(5)
C(10)-C(5)-C(4)	113.4(5)
C(9)-C(6)-C(8)	110.2(7)
C(9)-C(6)-C(2)	108.0(7)
C(8)-C(6)-C(2)	109.2(6)
C(9)-C(6)-C(7)	110.1(7)
C(8)-C(6)-C(7)	107.5(7)
C(2)-C(6)-C(7)	111.9(6)
C(12)-C(10)-C(11)	110.2(7)
C(12)-C(10)-C(13)	107.3(6)
C(11)-C(10)-C(13)	109.2(7)
C(12)-C(10)-C(5)	109.1(6)
C(11)-C(10)-C(5)	112.5(6)
C(13)-C(10)-C(5)	108.3(6)
N(5)-C(14)-N(4)	128.4(6)
N(5)-C(14)-N(6)	116.8(6)
N(4)-C(14)-N(6)	114.6(6)
N(4)-C(15)-C(19)	114.1(7)

N(4)-C(15)-C(16)	101.6(6)
C(19)-C(15)-C(16)	113.9(7)
N(6)-C(16)-C(15)	101.3(6)
N(6)-C(17)-C(18)	102.3(6)
N(5)-C(18)-C(17)	102.6(5)
N(5)-C(18)-C(23)	113.9(6)
C(17)-C(18)-C(23)	114.3(6)
C(21)-C(19)-C(15)	110.0(9)
C(21)-C(19)-C(20)	106.6(8)
C(15)-C(19)-C(20)	110.5(6)
C(21)-C(19)-C(22)	110.5(8)
C(15)-C(19)-C(22)	111.9(7)
C(20)-C(19)-C(22)	107.1(10)
C(26)-C(23)-C(24)	109.3(7)
C(26)-C(23)-C(25)	110.0(6)
C(24)-C(23)-C(25)	107.9(6)
C(26)-C(23)-C(18)	112.4(6)
C(24)-C(23)-C(18)	108.8(6)
C(25)-C(23)-C(18)	108.2(6)
O(1)-C(27)-O(2)	124.3(6)
O(1)-C(27)-C(28)	118.1(6)
O(2)-C(27)-C(28)	117.6(6)
O(3)-C(29)-O(4)	124.1(6)
O(3)-C(29)-C(30)	117.4(7)
O(4)-C(29)-C(30)	118.5(6)
N(7)-C(31)-C(32)	178.3(9)
N(1S)-C(1S)-C(2S)	177.4(12)
N(2S)-C(3S)-C(4S)	179.8(12)

Symmetry transformations used to generate equivalent atoms:

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Ru(1)	21(1)	17(1)	19(1)	1(1)	12(1)	2(1)
Ru(2)	24(1)	17(1)	19(1)	1(1)	15(1)	2(1)
Cl(1)	26(1)	35(1)	33(1)	3(1)	18(1)	0(1)
O(1)	32(3)	20(2)	31(2)	3(2)	19(2)	2(2)
O(2)	30(3)	18(2)	36(3)	1(2)	20(2)	3(2)
O(3)	30(3)	37(2)	19(2)	-2(2)	12(2)	-1(2)
O(4)	35(3)	42(3)	22(2)	-3(2)	18(2)	-2(2)
N(1)	22(3)	24(2)	20(2)	-1(2)	12(2)	-1(2)
N(2)	24(3)	23(2)	18(2)	-2(2)	13(2)	-2(2)
N(3)	12(3)	41(3)	26(3)	8(2)	4(2)	3(2)
N(4)	29(3)	22(2)	33(3)	8(2)	21(3)	8(2)
N(5)	23(3)	24(3)	32(3)	6(2)	19(2)	2(2)
N(6)	54(4)	24(3)	53(4)	5(3)	41(3)	6(3)
N(7)	49(4)	41(4)	47(4)	2(3)	21(4)	2(3)
C(1)	27(3)	23(3)	27(3)	-9(2)	13(3)	-6(2)
C(2)	32(4)	37(3)	19(3)	-3(3)	16(3)	-9(3)
C(3)	33(4)	50(4)	35(4)	12(3)	20(3)	-8(3)
C(4)	28(4)	43(4)	34(4)	14(3)	14(3)	6(3)
C(5)	34(4)	34(3)	13(3)	-1(2)	11(3)	-3(3)
C(6)	52(5)	53(4)	21(3)	-17(3)	26(4)	-13(4)
C(7)	68(6)	70(6)	48(5)	-29(4)	39(5)	-29(5)
C(8)	70(6)	54(5)	64(6)	-19(4)	53(5)	-6(4)
C(9)	75(7)	90(7)	38(5)	-6(4)	46(5)	-14(5)
C(10)	27(4)	46(4)	31(4)	1(3)	18(3)	-3(3)
C(11)	70(7)	63(6)	81(7)	-37(5)	50(6)	-31(5)
C(12)	43(5)	63(5)	56(5)	19(4)	20(4)	-9(4)
C(13)	30(4)	82(6)	45(5)	13(4)	18(4)	0(4)
C(14)	33(4)	28(3)	34(4)	10(3)	19(3)	6(3)
C(15)	64(6)	35(4)	74(6)	19(4)	56(5)	21(4)
C(16)	75(6)	31(4)	79(6)	1(4)	63(6)	7(4)

Table 4. Anisotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for 8610. The anisotropic displacement factor exponent takes the form:  $-2\pi^2$ [ h<sup>2</sup> a<sup>\*2</sup>U<sup>11</sup> + ... + 2 h k a<sup>\*</sup> b<sup>\*</sup> U<sup>12</sup> ]

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C(17)	60(5)	32(4)	49(5)	-5(3)	37(4)	-10(3)
C(18)	35(4)	25(3)	35(4)	7(3)	21(3)	1(3)
C(19)	41(5)	54(5)	95(7)	47(5)	44(5)	26(4)
C(20)	45(5)	86(7)	62(6)	41(5)	27(5)	29(5)
C(21)	71(8)	71(7)	194(14)	45(8)	97(9)	39(6)
C(22)	46(6)	147(11)	109(9)	94(9)	40(6)	32(7)
C(23)	45(5)	32(4)	43(4)	7(3)	33(4)	0(3)
C(24)	106(8)	50(5)	55(5)	6(4)	70(6)	-4(5)
C(25)	60(6)	38(4)	80(6)	18(4)	52(5)	6(4)
C(26)	63(6)	79(6)	40(5)	21(4)	28(5)	-1(5)
C(27)	28(4)	26(3)	27(3)	-1(3)	15(3)	3(3)
C(28)	44(5)	27(3)	52(5)	5(3)	30(4)	6(3)
C(29)	35(4)	36(4)	14(3)	2(3)	9(3)	3(3)
C(30)	48(5)	74(6)	25(4)	-4(4)	12(4)	-5(4)
C(31)	31(4)	45(4)	37(4)	-2(3)	20(3)	1(3)
C(32)	54(6)	86(7)	60(6)	-13(5)	37(5)	8(5)
N(1S)	87(7)	141(9)	67(6)	-18(6)	41(6)	26(7)
C(1S)	60(6)	63(6)	42(5)	3(4)	28(5)	4(5)
C(2S)	84(8)	57(6)	64(6)	-8(5)	46(6)	-2(5)
N(2S)	85(7)	92(7)	72(6)	0(5)	55(6)	4(5)
C(3S)	45(5)	46(5)	44(5)	2(4)	21(4)	-1(4)
C(4S)	42(5)	41(4)	51(5)	-2(4)	24(4)	-1(3)

Table 5. Hydrogen coordinates (  $x \ 10^4$ ) and isotropic displacement parameters (Å<sup>2</sup>x 10<sup>3</sup>) for 8610.

	х	у	Z	U(eq)
H(2A)	1862	826	7051	33
H(3A)	4262	732	8780	45
H(3B)	3690	469	7585	45
H(4A)	5519	470	6981	42
H(4B)	6631	732	7996	42
H(5A)	5964	825	5667	32
H(7A)	4607	1374	9235	85
H(7B)	3792	1739	9283	85
H(7C)	4027	1667	8131	85
H(8A)	643	1441	7053	81
H(8B)	1637	1707	6819	81
H(8C)	1408	1777	7976	81
H(9A)	1531	962	8743	90
H(9B)	2257	1304	9654	90
H(9C)	3072	940	9598	90
H(11A)	7532	1744	7993	98
H(11B)	6922	1394	8395	98
H(11C)	5982	1683	7373	98
H(12A)	7521	1744	6009	83
H(12B)	5980	1669	5300	83
H(12C)	6974	1386	5123	83
H(13A)	8956	1288	7613	78
H(13B)	8336	946	6668	78
H(13C)	8323	927	7943	78
H(15A)	-125	1794	4899	58
H(16A)	391	2487	4380	61
H(16B)	1489	2249	5487	61
H(17A)	3546	2332	5168	51
H(17B)	3016	2596	3979	51

4464	1989	4214	36
-2217	1476	3417	95
-1630	1392	2498	95
-2962	1627	2061	95
-2209	2135	4154	147
-3003	2259	2774	147
-1698	2482	3648	147
-671	1918	1811	149
-733	2349	2212	149
-2050	2131	1336	149
3750	1913	1101	86
3317	1608	1807	86
4762	1782	2430	86
4180	2595	1820	77
5172	2440	3125	77
4006	2720	2973	77
1917	2341	771	88
1663	2476	1873	88
1403	2041	1426	88
2490	-209	5064	57
3970	-131	5332	57
2883	-218	3988	57
-1263	779	516	77
-102	626	258	77
-488	1069	78	77
5885	451	2002	93
7094	709	2879	93
6092	866	1580	93
1766	82	998	96
2457	467	891	96
3220	181	1989	96
665	196	6221	66
629	-250	6488	66
1863	3	7344	66
	4464 -2217 -1630 -2962 -2209 -3003 -1698 -671 -733 -2050 3750 3317 4762 4180 5172 4006 1917 1663 1403 2490 3970 2883 -1263 -102 -488 5885 7094 6092 1766 2457 3220 665 629 1863	$\begin{array}{cccccc} 4464 & 1989 \\ -2217 & 1476 \\ -1630 & 1392 \\ -2962 & 1627 \\ -2209 & 2135 \\ -3003 & 2259 \\ -1698 & 2482 \\ -671 & 1918 \\ -733 & 2349 \\ -2050 & 2131 \\ 3750 & 1913 \\ 3317 & 1608 \\ 4762 & 1782 \\ 4180 & 2595 \\ 5172 & 2440 \\ 4006 & 2720 \\ 1917 & 2341 \\ 1663 & 2476 \\ 1403 & 2041 \\ 2490 & -209 \\ 3970 & -131 \\ 2883 & -218 \\ -1263 & 779 \\ -102 & 626 \\ -488 & 1069 \\ 5885 & 451 \\ 7094 & 709 \\ 6092 & 866 \\ 1766 & 82 \\ 2457 & 467 \\ 3220 & 181 \\ 665 & 196 \\ 629 & -250 \\ 1863 & 3 \end{array}$	4464 $1989$ $4214$ $-2217$ $1476$ $3417$ $-1630$ $1392$ $2498$ $-2962$ $1627$ $2061$ $-2209$ $2135$ $4154$ $-3003$ $2259$ $2774$ $-1698$ $2482$ $3648$ $-671$ $1918$ $1811$ $-733$ $2349$ $2212$ $-2050$ $2131$ $1336$ $3750$ $1913$ $1101$ $3317$ $1608$ $1807$ $4762$ $1782$ $2430$ $4180$ $2595$ $1820$ $5172$ $2440$ $3125$ $4006$ $2720$ $2973$ $1917$ $2341$ $771$ $1663$ $2476$ $1873$ $1403$ $2041$ $1426$ $2490$ $-209$ $5064$ $3970$ $-131$ $5332$ $2883$ $-218$ $3988$ $-1263$ $779$ $516$ $-102$ $626$ $258$ $-488$ $1069$ $78$ $5885$ $451$ $2002$ $7094$ $709$ $2879$ $6092$ $866$ $1580$ $1766$ $82$ $998$ $2457$ $467$ $891$ $3220$ $181$ $1989$ $665$ $196$ $6221$ $629$ $-250$ $6488$ $1863$ $3$ $7344$