

**Selective binding of benzenediol derivatives by simultaneous non-covalent interactions in bis-Pt(II)
aza-aromatic host-guest system**

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Supporting information

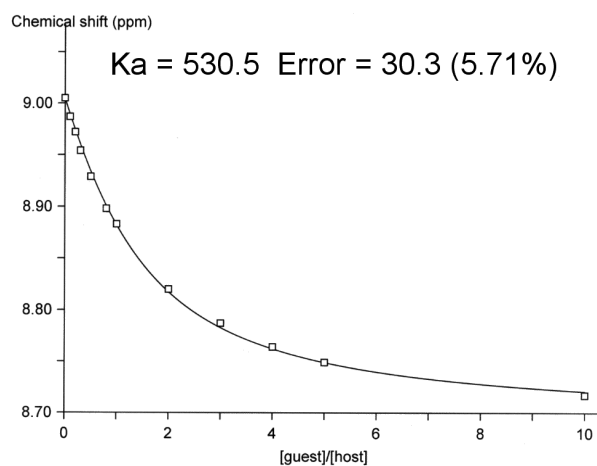


Fig. S1 Fit plot for the ^1H NMR titration of $1 \cdot \text{BF}_4$ with benzene-1,4-diol. $[1 \cdot \text{BF}_4] = 2 \text{ mM}$, $\text{DMSO-}d_6$, $25 \text{ }^\circ\text{C}$.

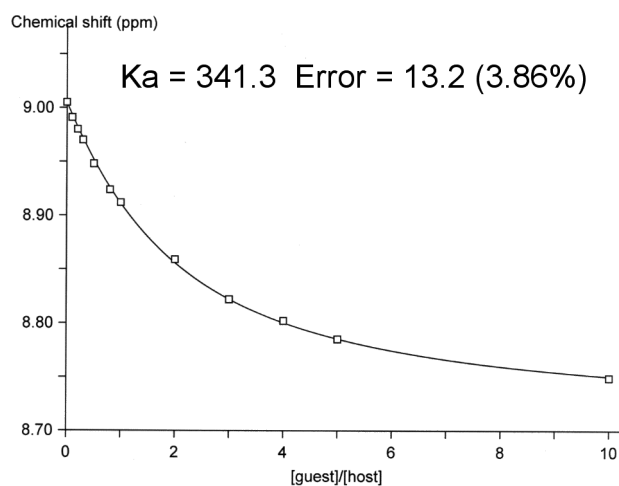


Fig. S2 Fit plot for the ^1H NMR titration of $1 \cdot \text{BF}_4$ with benzene-1,3-diol. $[1 \cdot \text{BF}_4] = 2 \text{ mM}$, $\text{DMSO-}d_6$, $25 \text{ }^\circ\text{C}$.

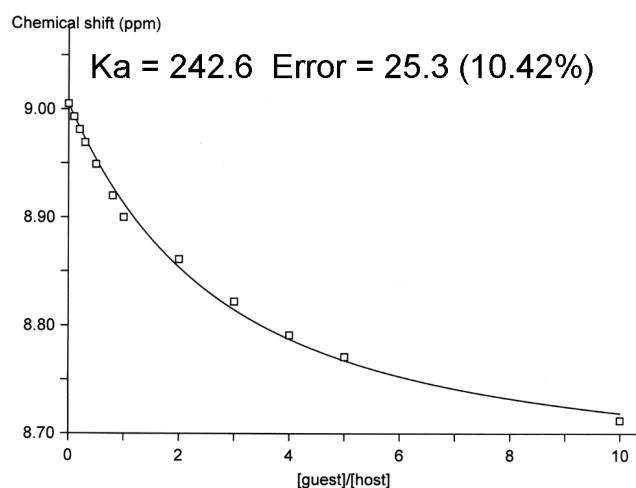


Fig. S3 Fit plot for the ^1H NMR titration of $1\cdot\text{BF}_4$ with benzene-1,2-diol. $[1\cdot\text{BF}_4] = 2$ mM, $\text{DMSO-}d_6$, 25°C .

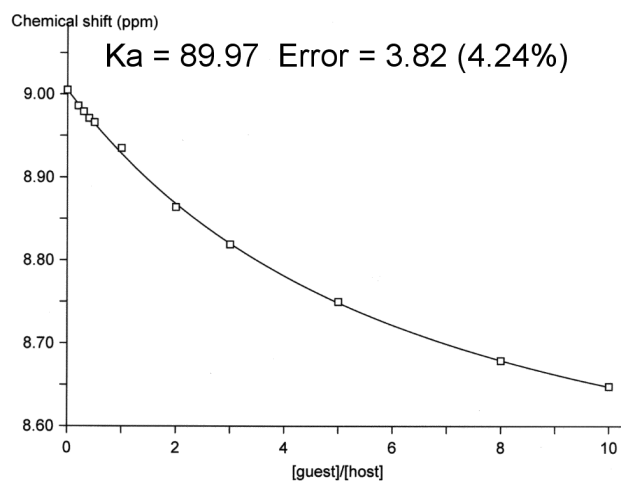


Fig. S4 Fit plot for the ^1H NMR titration of $1\cdot\text{BF}_4$ with 2-chloronaphthalene. $[1\cdot\text{BF}_4] = 2$ mM, $\text{DMSO-}d_6$, 25°C .

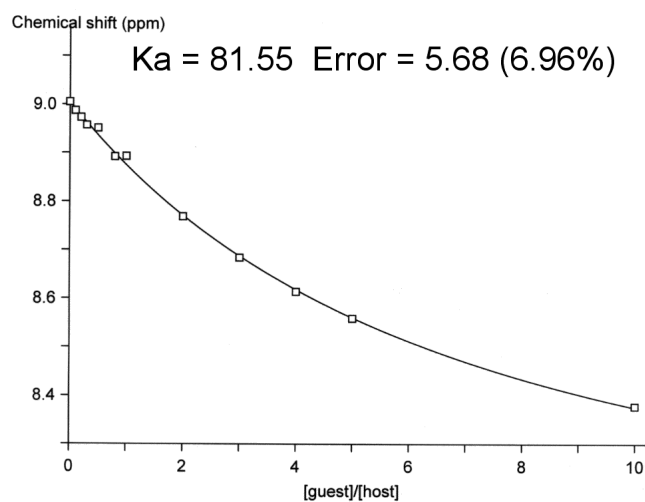


Fig. S5 Fit plot for the ^1H NMR titration of $1\cdot\text{BF}_4$ with pyrene. $[1\cdot\text{BF}_4] = 2$ mM, $\text{DMSO-}d_6$, 25°C .

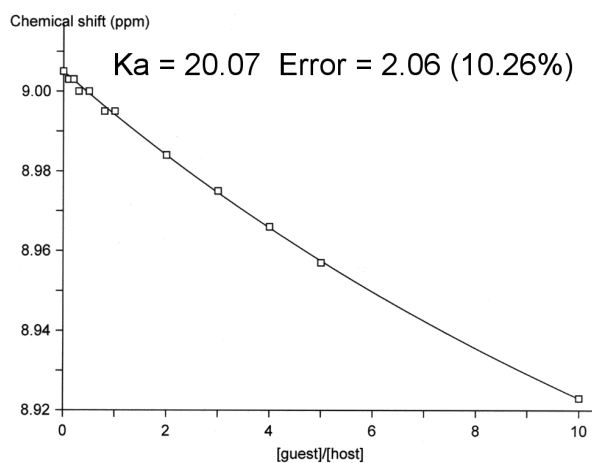


Fig. S6 Fit plot for the ^1H NMR titration of 1-BF_4 with 1,4-dichlorobenzene. $[1\text{-BF}_4] = 2$ mM, $\text{DMSO-}d_6$, 25°C .

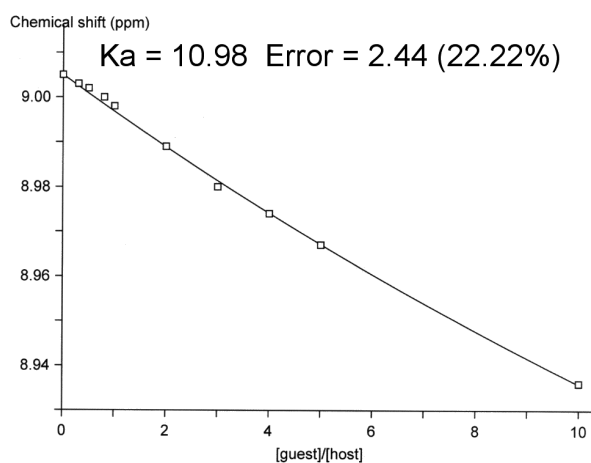


Fig. S7 Fit plot for the ^1H NMR titration of 1-BF_4 with 1,2-dichlorobenzene. $[1\text{-BF}_4] = 2$ mM, $\text{DMSO-}d_6$, 25°C .

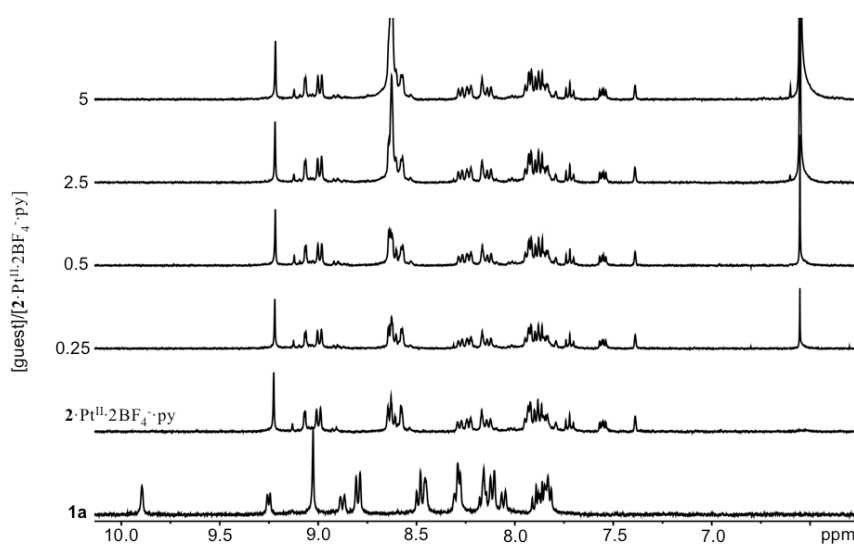


Fig. S8 Titration of $2\text{-Pt}^{\text{II}}\cdot 2\text{BF}_4\cdot\text{py}$ with benzene-1,4-diol ($[2\text{-Pt}^{\text{II}}\cdot 2\text{BF}_4\cdot\text{py}] = 2$ mM, $\text{DMSO-}d_6/(0.75\%$ pyridine- d_5), 25°C).

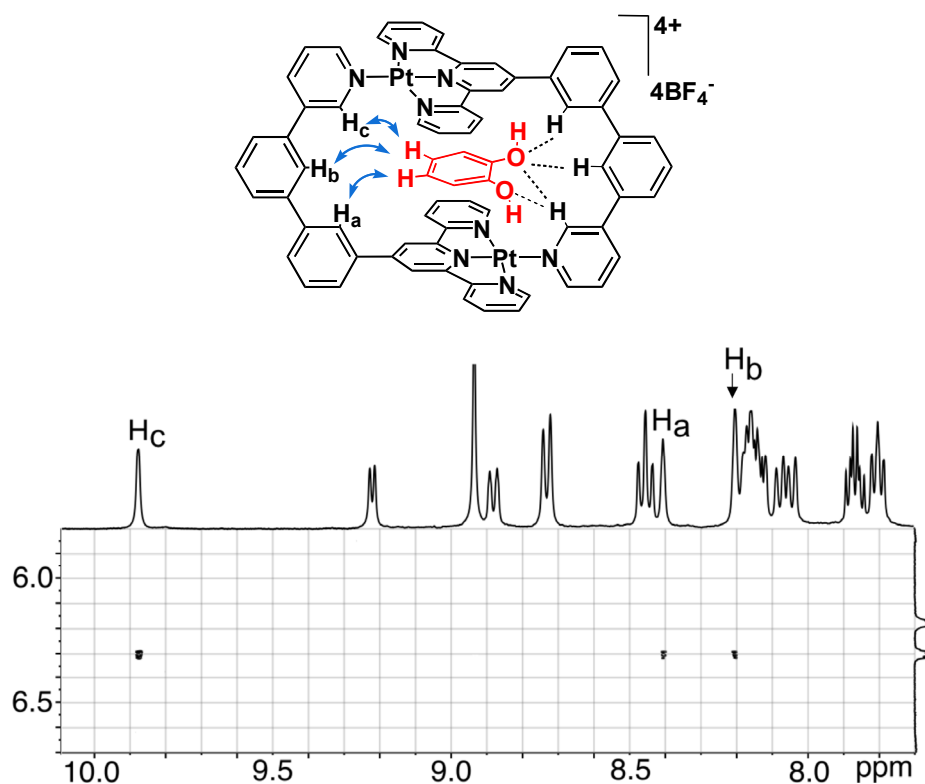


Fig. S9 Portion of the ¹H NOESY spectrum of a mixture of **1**·BF₄ with benzene-1,2-diol ([**1**·BF₄] = 9 mM, [benzene-1,2-diol] = 3 mM in DMSO-*d*₆, at 25 °C with a mixing time of 2.0 sec. The assignment and illustration of observed intermolecular NOE cross peaks are also reported.

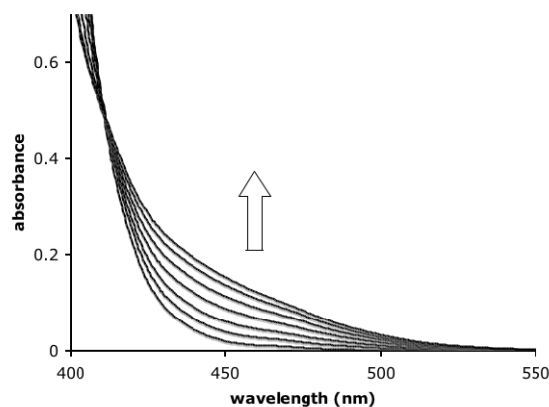


Fig. S10 Portion of the UV-vis titration of **1**·BF₄ ([**1**·BF₄] = 2 mM) by addition of various concentrations of benzene-1,3-diol (0-100 mM) in DMSO at 25 °C.

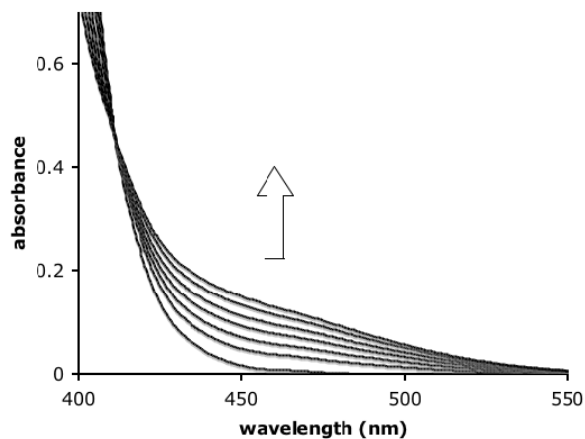


Fig. S11 Portion of the UV-vis titration of 1·BF₄ ([1·BF₄] = 2 mM) by addition of various concentrations of benzene-1,2-diol (0-100 mM) in DMSO at 25 °C.

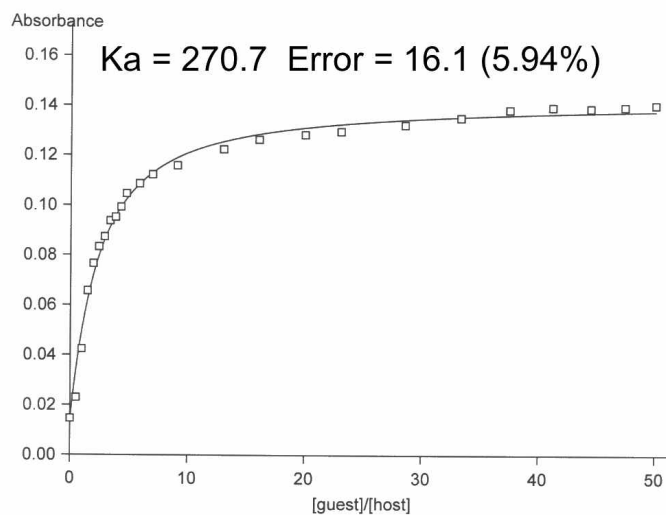


Fig. S12 Fit plot for the UV-vis titration of 1·BF₄ with benzene-1,4-diol. [1·BF₄] = 2 mM, DMSO, 25 °C.

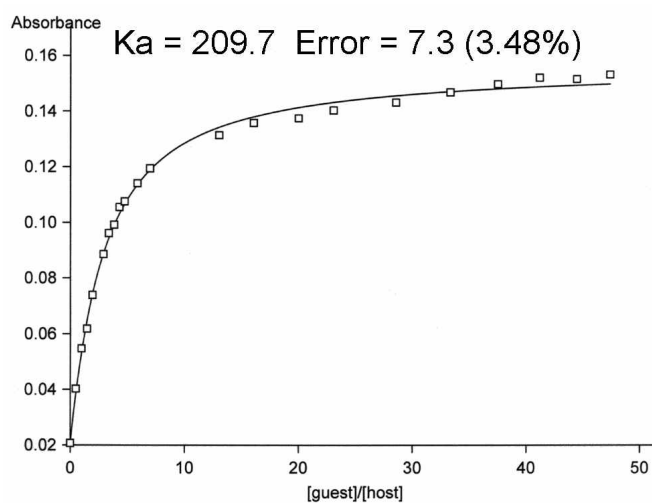


Fig. S13 Fit plot for the UV-vis titration of 1·BF₄ with benzene-1,3-diol. [1·BF₄] = 2 mM, DMSO, 25 °C.

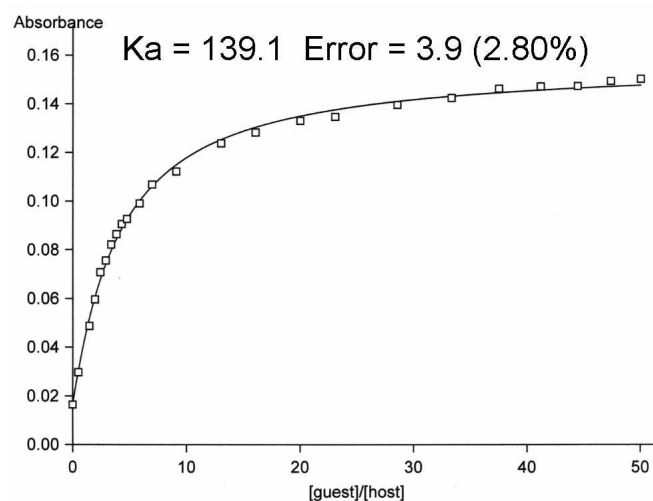


Fig. S14 Fit plot for the UV-vis titration of $1 \cdot \text{BF}_4$ with benzene-1,2-diol. $[1 \cdot \text{BF}_4] = 2 \text{ mM}$, DMSO, $25 \text{ }^\circ\text{C}$.

Calculation of Electrostatic Potential Surfaces

The structure of $1 \cdot \text{BF}_4$ was entered into *Spartan '04* by using the crystal structure data of $1 \cdot \text{BF}_4 \cdot \text{benzene-1,3-diol}$ ($1 \cdot \text{meta} \cdot \text{BF}_4$) saved in pdb format.

Electrostatic Potential Surfaces were calculated using the following settings:

Properties: total charge – 4+; multiplicity – singlet;

Calculations: calculate – single point energy; at – ground state; with – Density Functional – B3LYP – 6-31G* – pseudopotential; start from – initial geometry.