

B918336A (Revised Version)

Supporting Information

for

1,4-Alkyl migration associated with simultaneous S-C bond cleavage and N-C bond formation in platinum complexes of 2-aminothioethers. Characterization of intramolecular interligand charge transfer phenomenon

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Table S1. The bond distances (in Å) obtained at the B3LYP/sdd level optimized geometry [1][OTf]

Bond	X-ray	Calculated
Pt(1)-S(1)	2.2783(7)	2.323
Pt(1)-N(1)	2.020(2)	2.056
Pt(1)-N(3)	1.9645(19)	2.028
N(2)-N(3)	1.292(3)	1.289
Pt(1)-N(4)	1.941(2)	1.981
S(1)-C(17)	1.771(2)	1.777
N(4)-C(12)	1.376(3)	1.375
S(1)-C(18)	1.817(3)	1.831
S(1)-Pt(1)-N(1)	175.47(6)	178.5
S(1)-Pt(1)-N(3)	99.56(6)	101.8
S(1)-Pt(1)-N(4)	84.34(8)	82.38
N(1)-Pt(1)-N(3)	77.70(8)	77.04
N(1)-Pt(1)-N(4)	98.68(9)	98.69
N(3)-Pt(1)-N(4)	174.15(10)	174.8
Pt(1)-S(1)-C(17)	99.49(9)	99.25
Pt(1)-N(1)-C(1)	129.29(16)	129.1
Pt(1)-N(1)-C(5)	112.33(17)	111.7
Pt(1)-N(3)-N(2)	120.52(16)	118.4
Pt(1)-N(3)-C(6)	125.37(15)	127.1
Pt(1)-N(4)-C(12)	122.24(18)	123.6

Table S2. The bond distances (in Å) obtained at the B3LYP/sdd level optimized geometry **2**

	X-ray	Calculated
Pt(1)-S(1)	2.2707(16)	2.329
Pt(1)-N(1)	2.045(5)	2.061
Pt(1)-N(3)	1.968(5)	2.004
Pt(1)-N(4)	1.988(5)	1.980
N(2)-N(3)	1.315(7)	1.313
S(1)-C(17)	1.738(7)	1.744
N(4)-C(12)	1.364(8)	1.366
S(1)-Pt(1)-N(1)	176.27(14)	175.7
S(1)-Pt(1)-N(3)	99.79(15)	100.9
S(1)-Pt(1)-N(4)	83.54(15)	82.63
N(1)-Pt(1)-N(3)	77.61(19)	77.07
N(1)-Pt(1)-N(4)	99.03(19)	99.17
N(3)-Pt(1)-N(4)	176.5(2)	176.3
Pt(1)-S(1)-C(17)	99.9(2)	99.47
Pt(1)-N(1)-C(1)	129.6(4)	129.5
Pt(1)-N(1)-C(5)	111.6(4)	111.6
Pt(1)-N(3)-N(2)	119.9(4)	119.0
Pt(1)-N(3)-C(6)	127.7(4)	127.5
Pt(1)-N(4)-C(12)	121.5(4)	123.6

Table S3. The bond distances (in Å) obtained at the B3LYP/sdd level optimized geometry **3**

	X-ray	Calculated
Pt(1)-S(1)	2.242(3)	2.313
Pt(1)-N(1)	2.055(8)	2.097
Pt(1)-N(2)	1.969(8)	2.009
Pt(1)-N(4)	1.981(8)	2.023
N(2)-N(3)	1.295(11)	1.308
S(1)-C(17)	1.721(11)	1.738
N(4)-C(12)	1.366(13)	1.383
S(1)-Pt(1)-N(1)	172.6(2)	173.3
S(1)-Pt(1)-N(3)	96.3(2)	97.1
S(1)-Pt(1)-N(4)	84.0(2)	83.4
N(1)-Pt(1)-N(3)	77.0(3)	76.2
N(1)-Pt(1)-N(4)	102.4(3)	103.1
N(3)-Pt(1)-N(4)	175.9(3)	174.5
Pt(1)-S(1)-C(17)	101.0(4)	99.7
Pt(1)-N(4)-C(12)	120.8(7)	120.6
Pt(1)-N(1)-C(5)	111.9(6)	110.6
Pt(1)-N(3)-N(2)	120.2(6)	119.4
Pt(1)-N(3)-C(6)	127.4(7)	127.0
Pt(1)-N(1)-C(1)	131.6(7)	131.4

Table S4. The bond distances (in Å) obtained at the B3LYP/sdd level optimized geometry [5][OTf]

	X-ray	Calculated
Pt(1)-S(1)	2.259(9)	2.316
Pt(1)-N(1)	2.036(3)	2.066
Pt(1)-N(2)	2.025(3)	2.050
Pt(1)-N(3)	1.981(3)	2.001
S(3)-C(17)	1.814(4)	1.832
S(3)-C(16)	1.774(4)	1.770
N(3)-C(11)	1.355(5)	1.372
S(1)-Pt(1)-N(1)	177.4(9)	177.7
S(1)-Pt(1)-N(2)	97.38(8)	99.54
S(1)-Pt(1)-N(3)	84.43(10)	82.18
N(1)-Pt(1)-N(2)	80.08(12)	78.99
N(1)-Pt(1)-N(3)	98.12(13)	99.32
N(2)-Pt(1)-N(3)	178.0(13)	177.7
Pt(1)-S(1)-C(16)	99.57(12)	99.46
Pt(1)-N(3)-C(11)	121.0(3)	121.8

Table S5. Some Selected Molecular Orbitals along with their energies and compositions of compound [1][OTf] in gas phase b3lyp/sdd level calculation

MO	Energy, eV	Composition		
		Pt	pap	(H ₂ N^SMe)
LUMO+5	-3.28	06	86	08
LUMO+4	-3.46	02	98	0
LUMO+3	-3.69	13	11	76
LUMO+2	-4.61	17	71	12
LUMO+1	-4.70	30	49	21
LUMO	-6.48	10	79(N=N, 38)	11
HOMO	-8.23	08	14	78
HOMO-1	-9.73	02	04	94
HOMO-2	-9.87	05	96	01
HOMO-3	-10.06	0	95	05
HOMO-4	-10.45	61	13	26
HOMO-5	-10.49	64	11	25
HOMO-6	-10.78	58	23	19
HOMO-7	-11.24	06	74(N=N, 36)	20
HOMO-8	-11.38	04	80(N=N, 39)	16
HOMO-9	-11.51	24	24	52
HOMO-10	-11.99	55	21	24
HOMO - LUMO = 1.75 eV				

Table S6. Some Selected Molecular Orbitals along with their energies and compositions of compound **2** in gas phase b3lyp/sdd level calculation

MO	Energy, eV	Composition		
		Pt	pap	(HN^S)
LUMO+5	-0.05	82	12	06
LUMO+4	-0.13	06	89	05
LUMO+3	-0.22	05	94	01
LUMO+2	-0.79	41	28	31(S,22)
LUMO+1	-1.35	02	98	0
LUMO	-3.02	13	62(N=N, 29)	25
HOMO	-4.72	06	30	64(S,13)
HOMO-1	-5.44	14	02	84(S,30)
HOMO-2	-6.60	08	85	07
HOMO-3	-6.82	73	12	15
HOMO-4	-6.92	09	88	03
HOMO-5	-7.04	31	17	52
HOMO-6	-7.10	41	23	36
HOMO-7	-7.48	30	16	54(S,42)
HOMO-8	-7.85	06	74(N=N, 56)	20
HOMO-9	-8.02	13	57	30(S,13)
HOMO-10	-8.15	14	52	34(S,13)

Table S7. Some Selected Molecular Orbitals along with their energies and compositions of compound **3** in gas phase b3lyp/sdd level calculation

MO	Energy, eV	Composition		
		Pt	pap	(BzN^S)
LUMO+5	-0.10	06	68	26
LUMO+4	-0.18	04	73	23
LUMO+3	-0.26	07	23	70
LUMO+2	-0.87	39	28	33(S,22)
LUMO+1	-1.28	04	94	02
LUMO	-2.98	12	65(N=N, 31)	23
HOMO	-4.65	06	28	66(S,14)
HOMO-1	-5.36	13	02	85(S,30)
HOMO-2	-6.53	10	45	45
HOMO-3	-6.65	01	43	56
HOMO-4	-6.74	74	08	18
HOMO-5	-6.85	03	10	87
HOMO-6	-6.91	11	81	08
HOMO-7	-6.99	30	13	57
HOMO-8	-7.05	30	21	49
HOMO-9	-7.54	19	36	45(S,34)
HOMO-10	-7.63	22	38	40(S,13)

Table S8. Some Selected Molecular Orbitals along with their energies and compositions of compound [5][OTf] in gas phase b3lyp/sdd level calculation.

MO	Energy, eV	Composition		
		Pt	bpy	(H ₂ N^SMe)
LUMO+10	-1.75	64	06	30
LUMO+9	-1.82	26	01	73
LUMO+8	-2.08	82	08	10
LUMO+7	-2.53	39	01	60
LUMO+6	-2.75	87	03	10
LUMO+5	-3.21	02	93	05
LUMO+4	-3.38	21	07	72(S,23)
LUMO+3	-4.39	46	20	34(S,22)
LUMO+2	-4.58	02	97	01
LUMO+1	-4.73	02	98	0
LUMO	-5.77	05	92	03
HOMO	-7.69	09	05	86
HOMO-1	-9.38	01	0	99
HOMO-2	-9.92	84	02	14
HOMO-3	-9.96	51	09	40
HOMO-4	-10.36	54	32	04
HOMO-5	-10.57	15	80	05
HOMO-6	-11.08	44	09	47(S,24)
HOMO-7	-11.39	20	21	59
HOMO-8	-11.61	38	20	42(S,22)
HOMO-9	-11.83	19	14	67
HOMO-10	-12.02	01	99	0

Table S9. Selected list of vertical excitations computed at the TD-DFT/B3LYP/6-31+G**//B3LYP/6-31G* level of theory for [1][OTf]

Excited State	Wavelength (nm)	f	Energy (eV)	Transition	Character
1	790.4	0.216	1.5685	(67%)HOMO → LUMO	H₂N^SMe(π)→pap(π[*]_{N=N}), ILCT
5	425.2	0.174	2.9161	(63%)HOMO-2 → LUMO	pap(π) → pap(π [*] _{N=N}), ILCT
6	419.5	0.037	2.9555	(53%)HOMO → LUMO+1 (31%)HOMO → LUMO+2	H ₂ N^SMe(π)→pap(π [*]), ILCT
8	382.9	0.063	3.2374	(40%)HOMO-6 → LUMO (23%)HOMO-5 → LUMO	Pt(dπ) → pap(π [*] _{N=N}), MLCT
10	351.1	0.038	3.5304	(37%)HOMO-6 → LUMO (33%)HOMO-4 → LUMO	Pt(dπ) → pap(π [*] _{N=N}), MLCT
14	281.1	0.061	4.4101	(53%)HOMO-8 → LUMO	pap(n, _{N=N}) → pap(π [*] _{N=N}), ILCT
18	268.5	0.068	4.6165	(53%)HOMO-1 → LUMO+1	H ₂ N^SMe(π)→pap(π [*]), ILCT

Table S10. Selected list of vertical excitations computed at the TD-DFT/B3LYP/6-31+G**//B3LYP/6-31G* level of theory for **2**

Excited State	Wavelength (nm)	f	Energy (eV)	Transition	Character
1	833.2	0.230	1.4880	(50%)HOMO → LUMO (25%)HOMO-1 → LUMO	HN^S(π)→pap(π[*]_{N=N}), ILCT
3	722.3	0.103	1.7165	(34%)HOMO-1 → LUMO (33%)HOMO → LUMO	HN^S(π)→pap(π [*] _{N=N}), ILCT
5	413.2	0.111	3.0002	(59%)HOMO-2 → LUMO	pap(π) → pap(π [*] _{N=N}), ILCT
7	383.6	0.068	3.2322	(52%)HOMO-5 → LUMO	Pt(dπ)/HN^S(π) → pap(π [*] _{N=N}), MLCT
10	362.6	0.076	3.4186	(66%)HOMO-1 → LUMO+1	HN^S(π) → pap(π [*]), ILCT
14	311.2	0.113	3.9834	(87%)HOMO → LUMO+3	HN^S(π) → pap(π [*]), ILCT
19	283.7	0.063	4.3701	(79%)HOMO → LUMO+4	HN^S(π) → pap(π [*]), ILCT

Table S11. Selected list of vertical excitations computed at the TD-DFT/B3LYP/6-31+G**//B3LYP/6-31G* level of theory for **3**

Excited State	Wavelength (nm)	f	Energy (eV)	Transition	Character
1	839.4	0.192	1.4770	(36%)HOMO → LUMO (36%)HOMO-1 → LUMO	BzN [^] S(π)→pap(π [*] _{N=N}), ILCT
2	717.3	0.134	1.7284	(48%)HOMO-1 → LUMO (25%)HOMO → LUMO	BzN [^] S(π)→pap(π [*] _{N=N}), ILCT
6	390.3	0.105	3.1767	(65%)HOMO-3 → LUMO	pap(π)/BzN [^] S(π)→pap(π [*] _{N=N}), ILCT
8	367.3	0.046	3.3757	(69%)HOMO-6 → LUMO	pap(π) → pap(π [*] _{N=N}), ILCT
10	355.3	0.102	3.4894	(38%)HOMO-1 → LUMO+1 (27%)HOMO → LUMO+3	BzN [^] S(π) → pap(π [*]), ILCT BzN [^] S(π) → L(π [*]), ILCT
17	301.9	0.065	4.1060	(91%)HOMO → LUMO+4	BzN [^] S(π) → pap(π [*]), ILCT
19	294.0	0.048	4.2162	(84%)HOMO → LUMO+5	BzN [^] S(π) → pap(π [*]), ILCT
23	279.1	0.054	4.4422	(53%)HOMO-11 → LUMO	pap(π)/BzN [^] S(π)→pap(π [*] _{N=N}), ILCT

Table S12. Selected list of vertical excitations computed at the TD-DFT/B3LYP/6-31+G**//B3LYP/6-31G* level of theory for [5][OTf]

Excited State	Wavelength (nm)	f	Energy (eV)	Transition	Character
1	580.5	0.097	3.1356	(94%)HOMO → LUMO	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{bpy}(\pi^*)$, ILCT
4	391.5	0.043	3.1667	(88%)HOMO → LUMO+3	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{Pt(d),LMCT}$ $\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{H}_2\text{N}^{\wedge}\text{SMe}(\pi^*),\text{ILCT}$
6	349.7	0.021	3.5446	(65%)HOMO → LUMO+4	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{H}_2\text{N}^{\wedge}\text{SMe}(\pi^*),\text{ILCT}$
10	302.7	0.125	4.0961	(54%)HOMO-4 → LUMO (32%)HOMO-5 → LUMO	$\text{Pt(d}\pi\text{)} \rightarrow \text{bpy}(\pi^*)$, MLCT $\text{bpy}(\pi) \rightarrow \text{bpy}(\pi^*)$, ILCT
11	293.0	0.298	4.2462	(49%)HOMO-5 → LUMO (29%)HOMO-4 → LUMO	$\text{bpy}(\pi) \rightarrow \text{bpy}(\pi^*)$, ILCT $\text{Pt(d}\pi\text{)} \rightarrow \text{bpy}(\pi^*)$, MLCT
14	282.9	0.084	4.3824	(79%)HOMO → LUMO+7	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{H}_2\text{N}^{\wedge}\text{SMe}(\pi^*),\text{ILCT}$
21	252.6	0.068	4.9077	(54%)HOMO-1 → LUMO+1	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{bpy}(\pi^*)$, ILCT
22	247.3	0.056	5.0127	(42%)HOMO-1 → LUMO+3	$\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{Pt(d),LMCT}$ $\text{H}_2\text{N}^{\wedge}\text{SMe}(\pi) \rightarrow \text{H}_2\text{N}^{\wedge}\text{SMe}(\pi^*),\text{ILCT}$

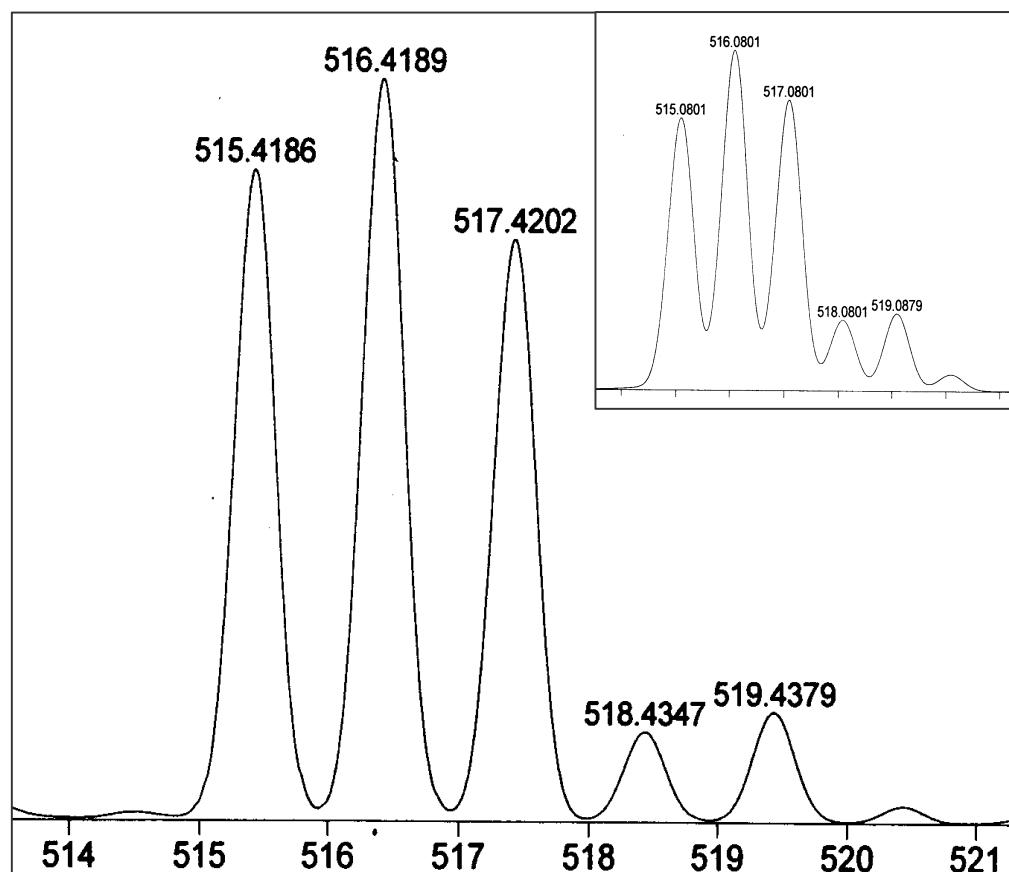


Figure S1. ESI-MS spectrum of the compound **[1]Cl**. Inset: simulated isotopic pattern.

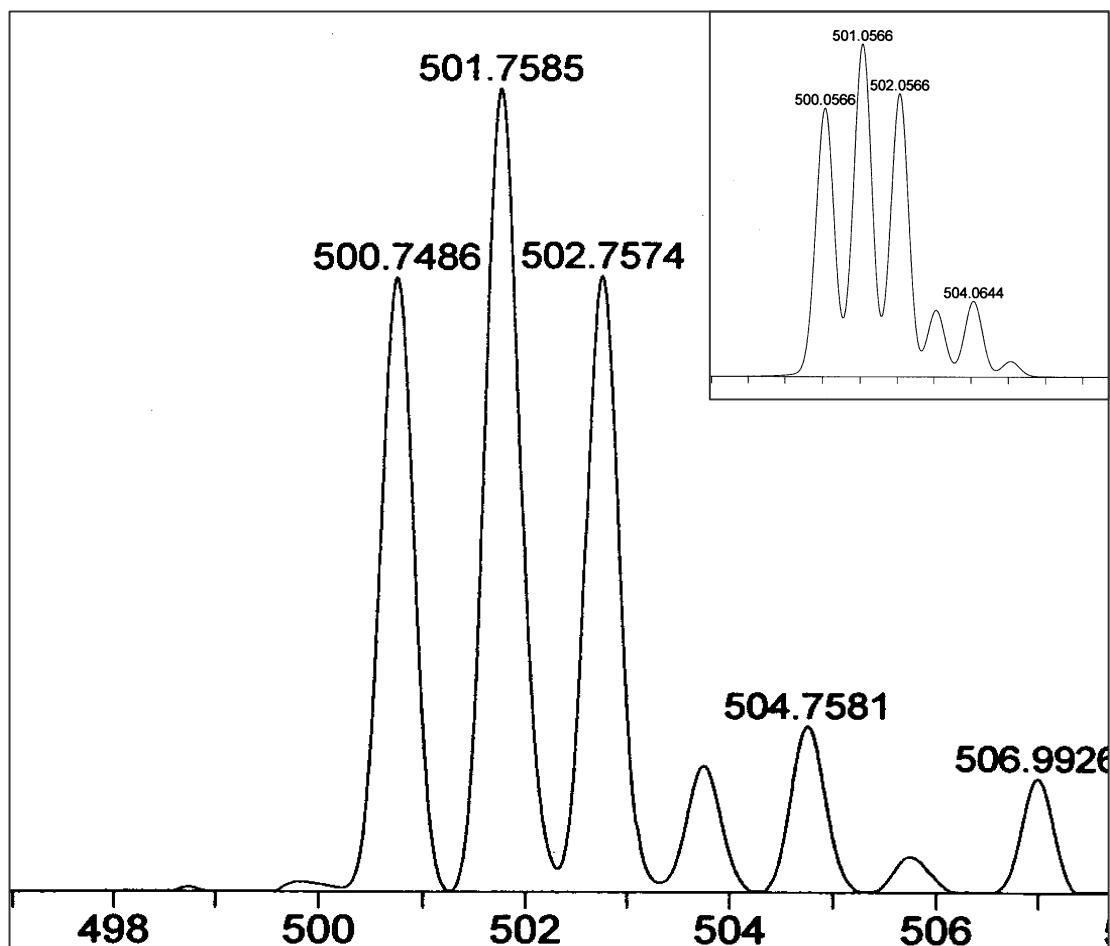


Figure S2. ESI-MS spectrum of the compound **2**. Inset: simulated isotopic pattern.

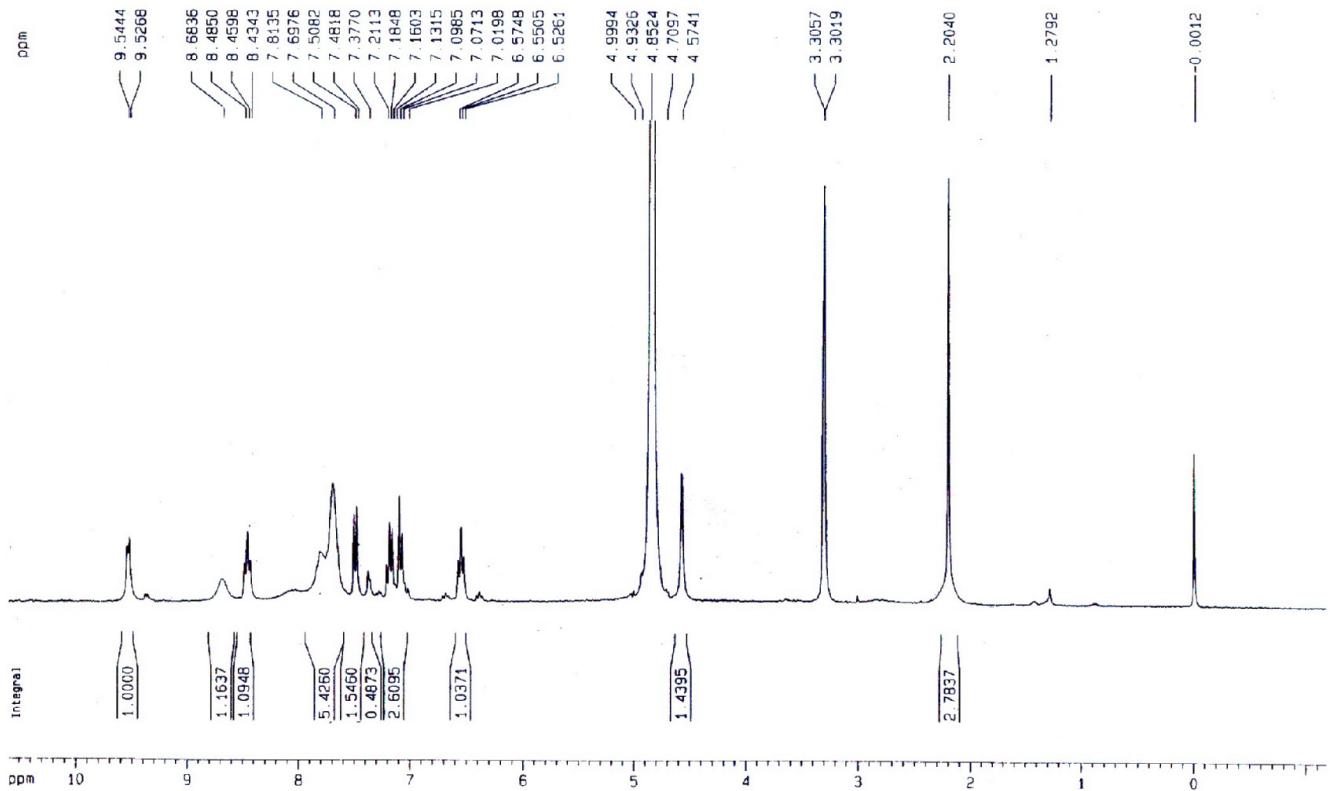


Figure S3. ^1H NMR spectrum of the compound **[1]Cl**.

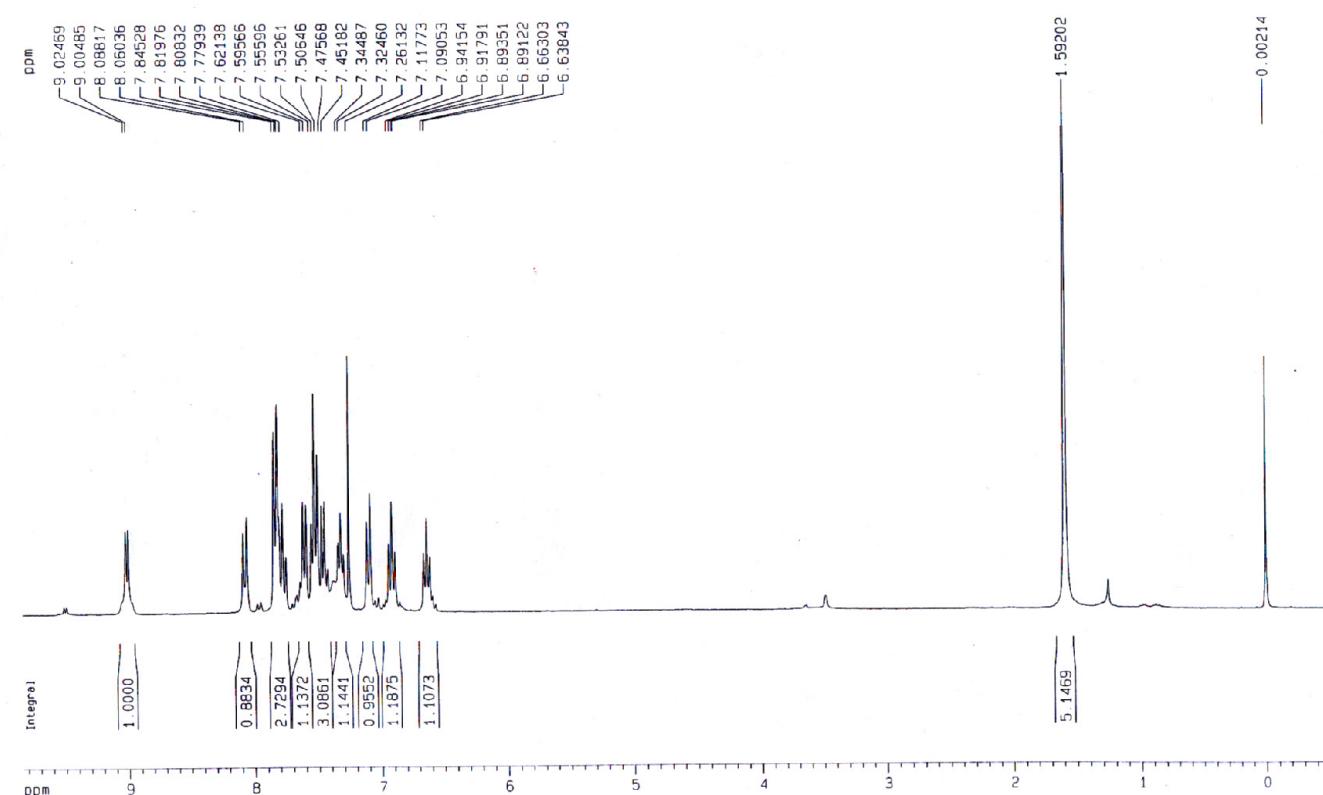


Figure S4. ¹H NMR spectrum of the compound 2

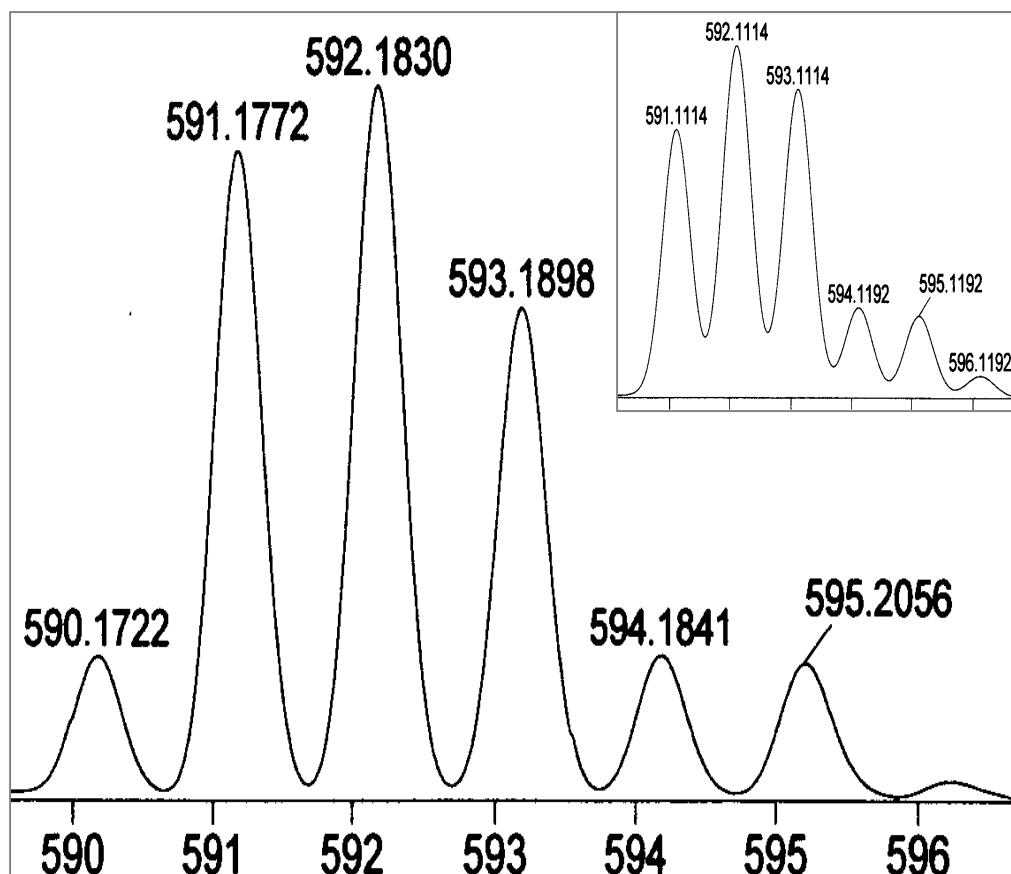


Figure S5. ESI-MS spectrum of the compound 3. Inset: simulated isotopic pattern.

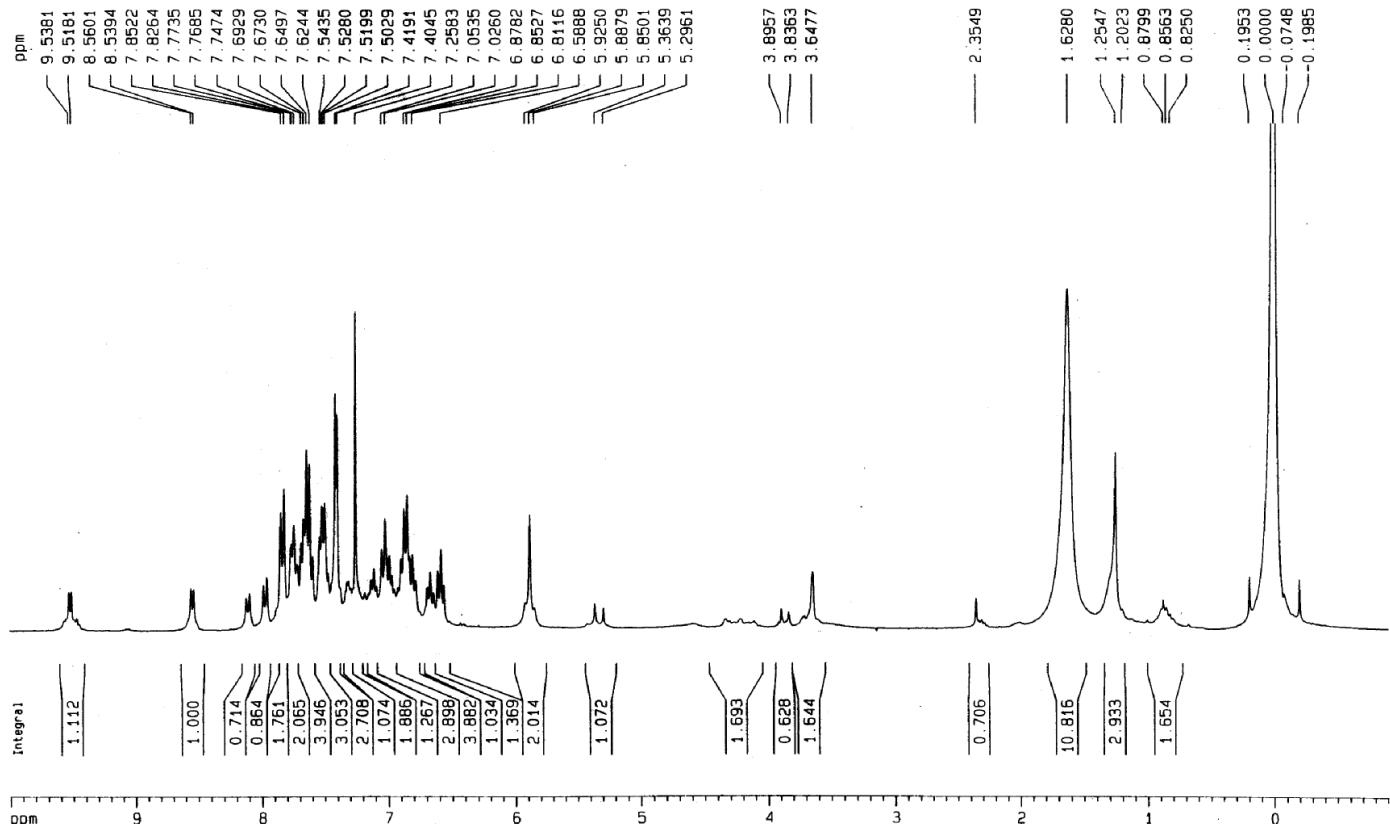


Figure S6. ¹H NMR spectrum of the compound 3

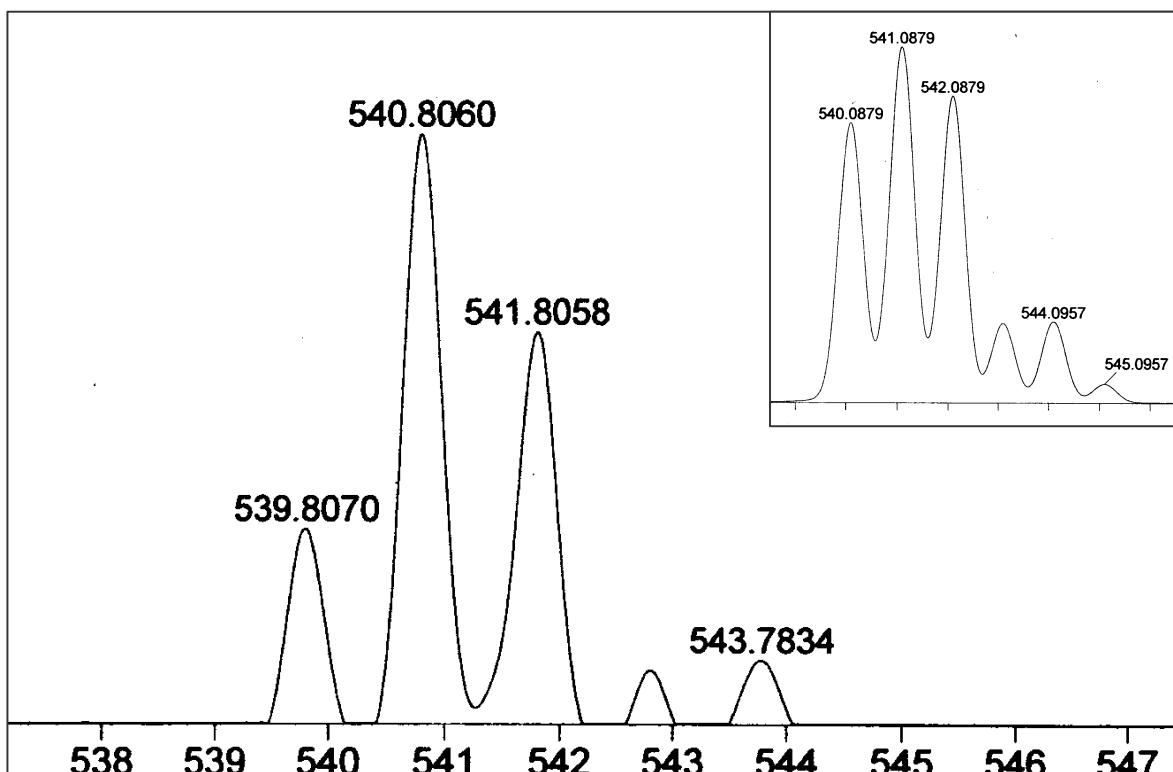


Figure S7. ESI-MS spectrum of the compound 4. Inset: simulated isotopic pattern

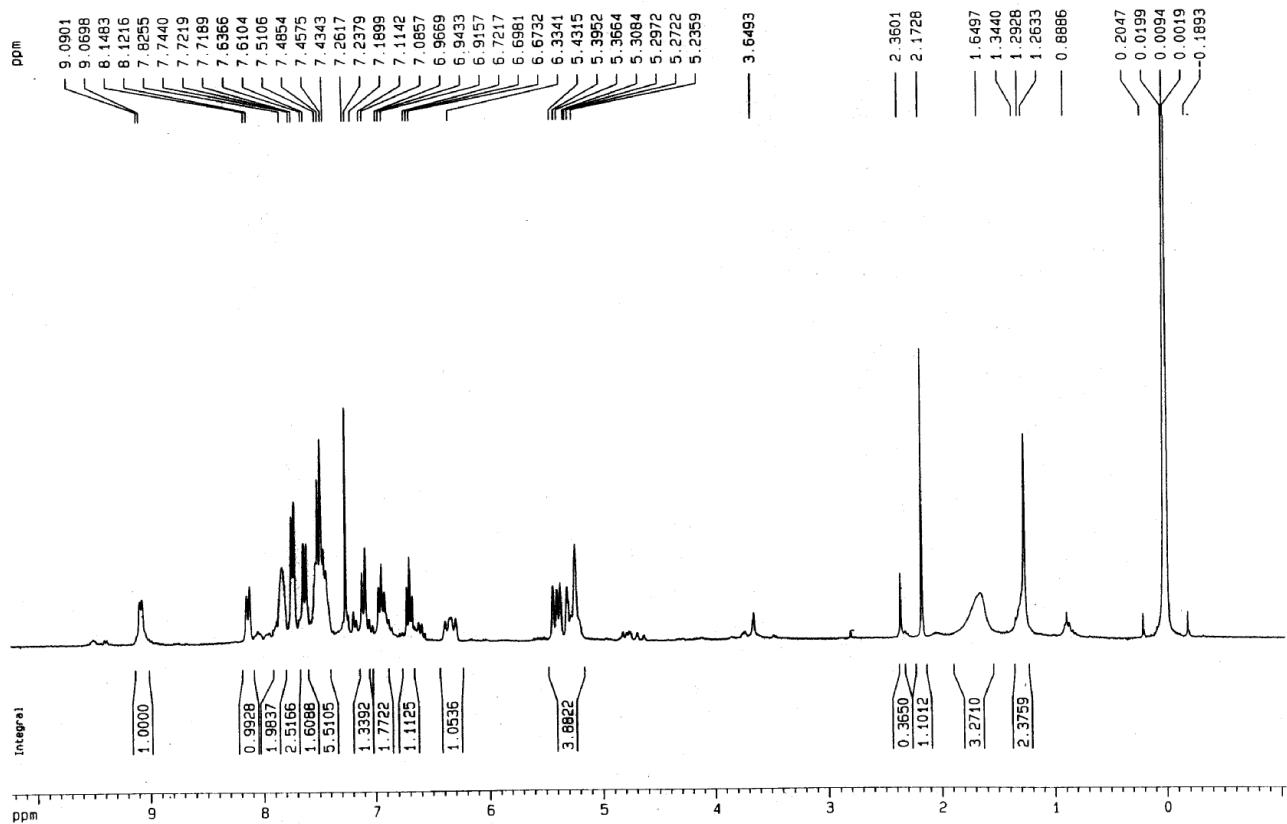


Figure S8. ¹H NMR spectrum of the compound 4

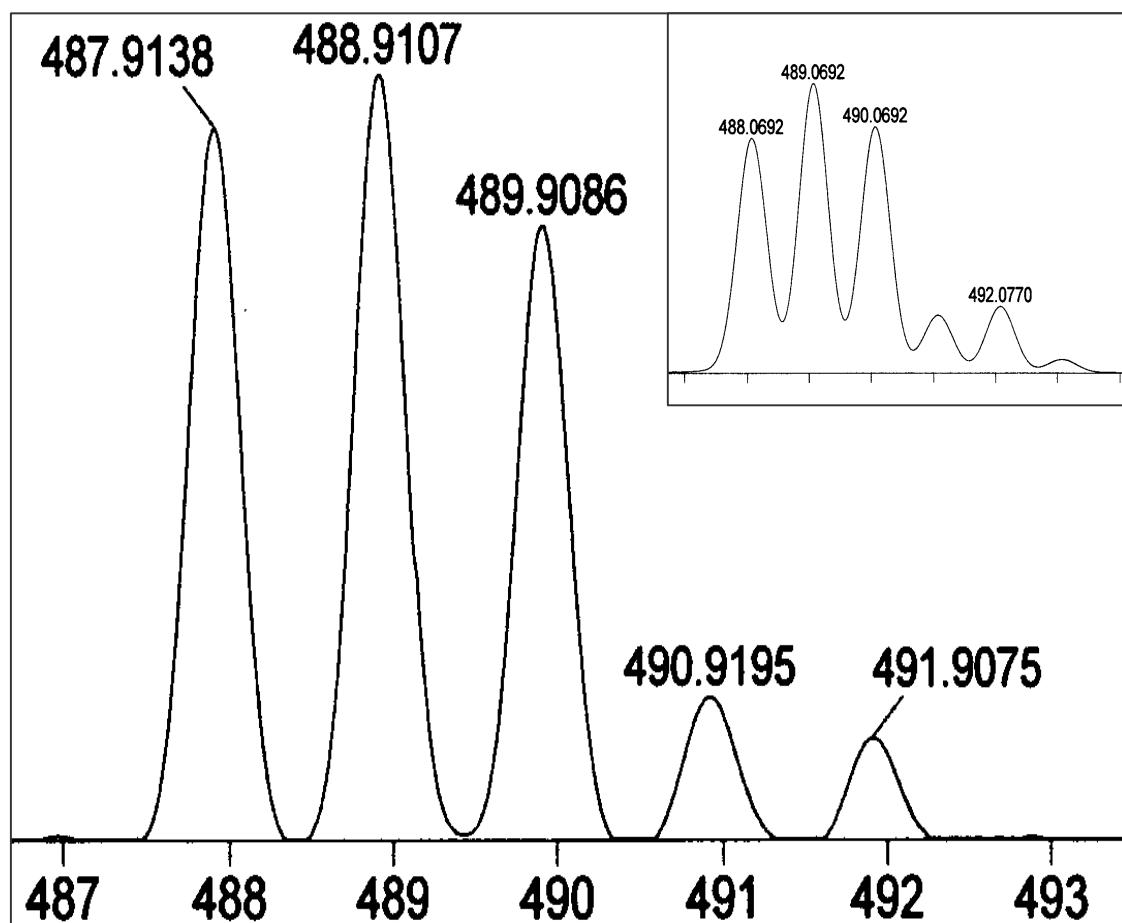


Figure S9. ESI-MS spectrum of the compound [5]Cl. Inset: simulated isotopic pattern.

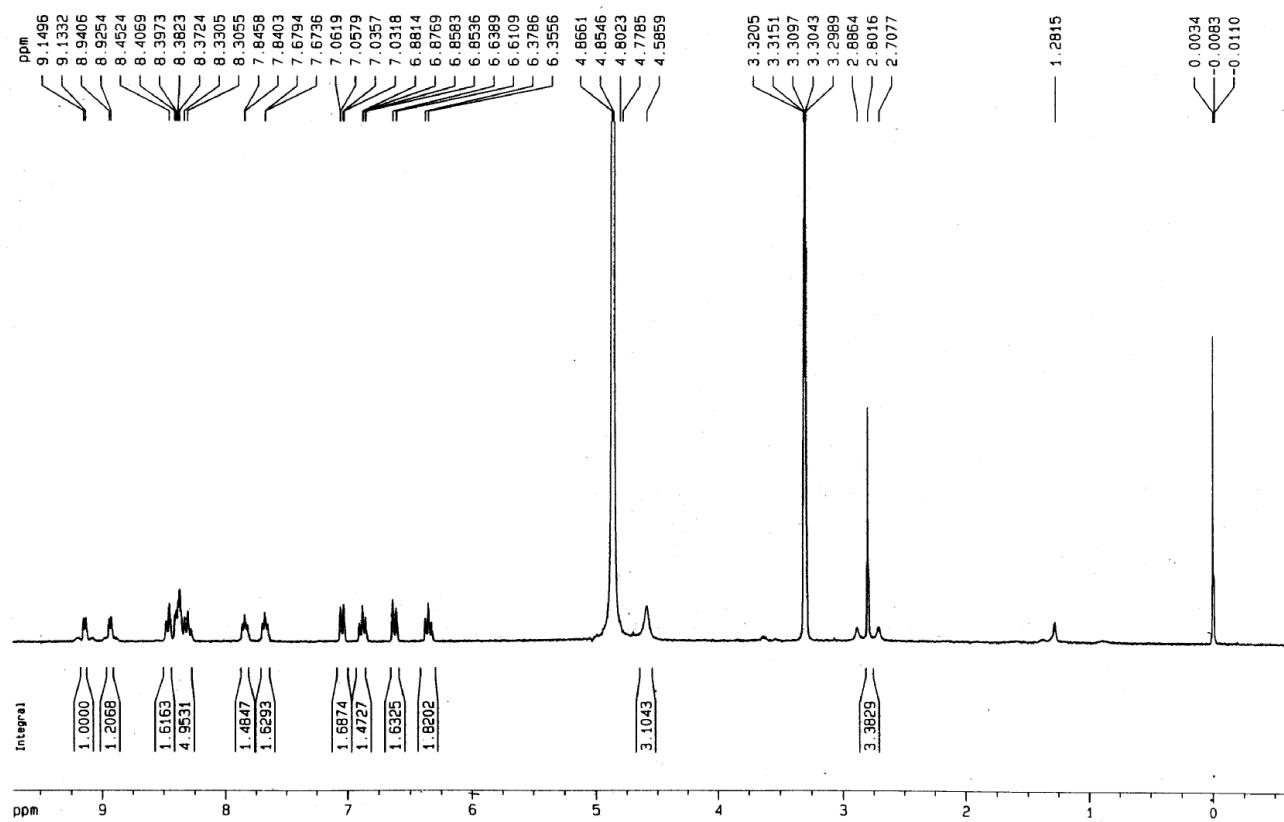


Figure S10. ¹H NMR spectrum of the compound [5]Cl

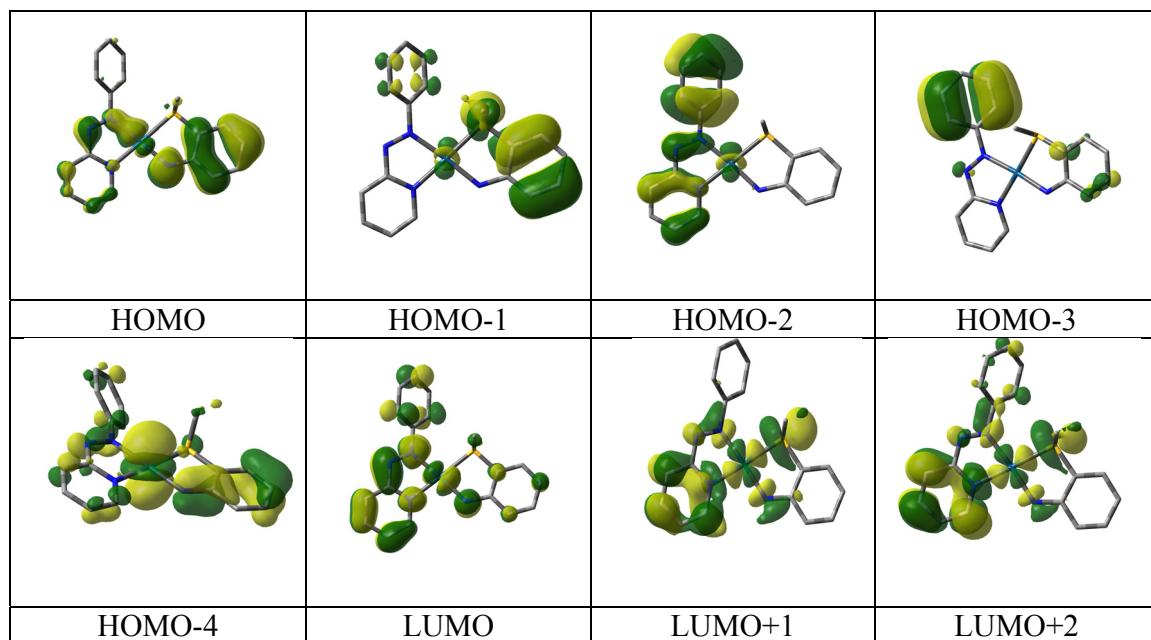


Figure S11. DFT calculated frontier orbitals for the compound [1][OTf]

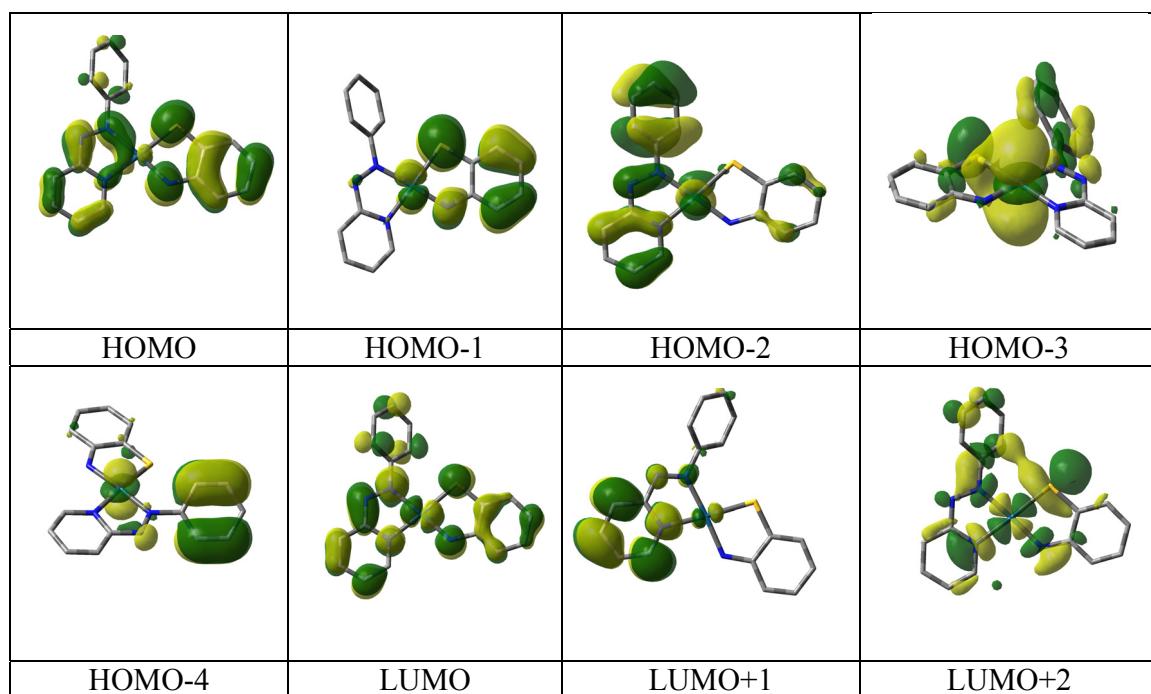


Figure S12. DFT calculated frontier orbitals for the compound **2**

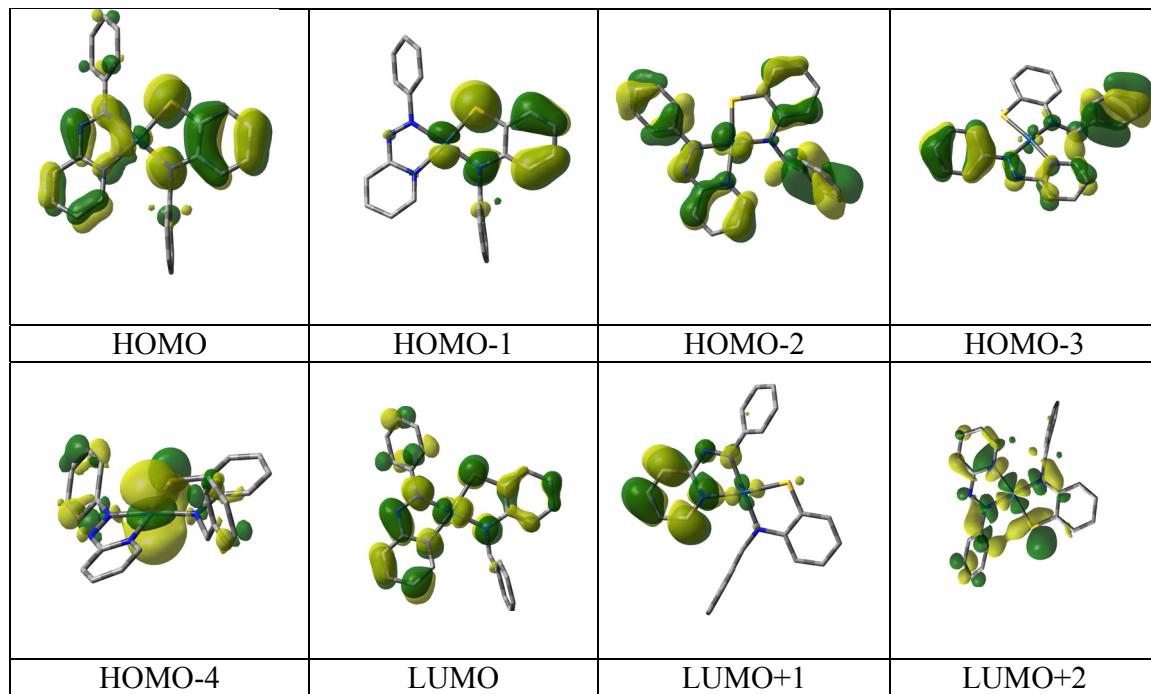


Figure S13. DFT calculated frontier orbitals for the compound **3**

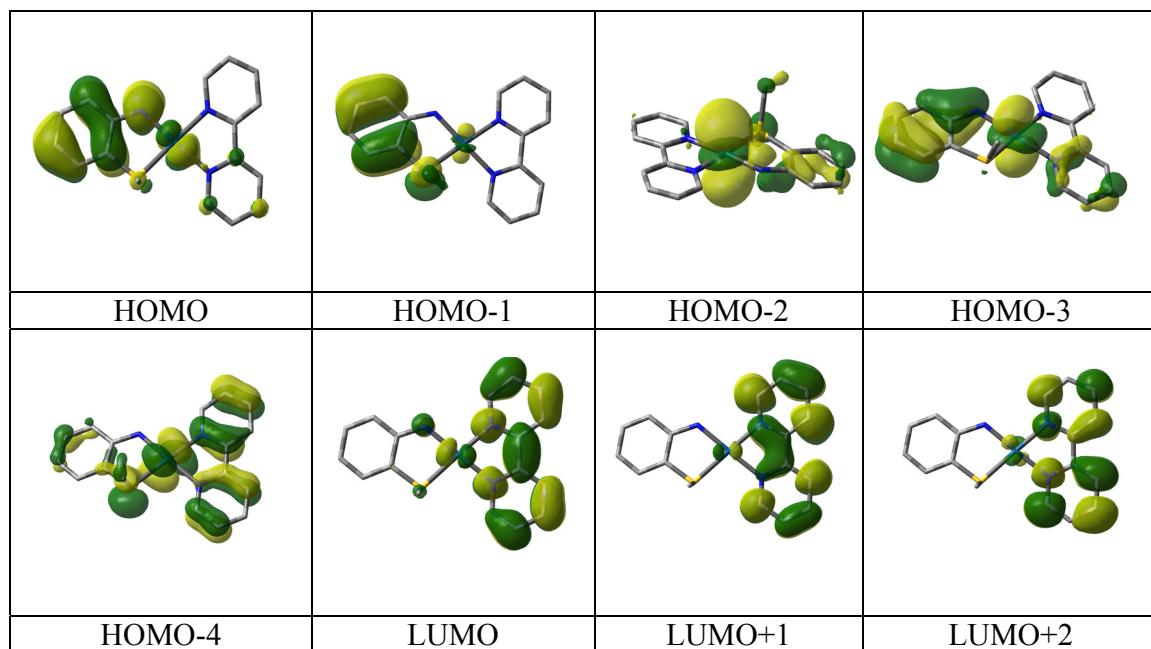


Figure S14. DFT calculated frontier orbitals for the compound [5][OTf]

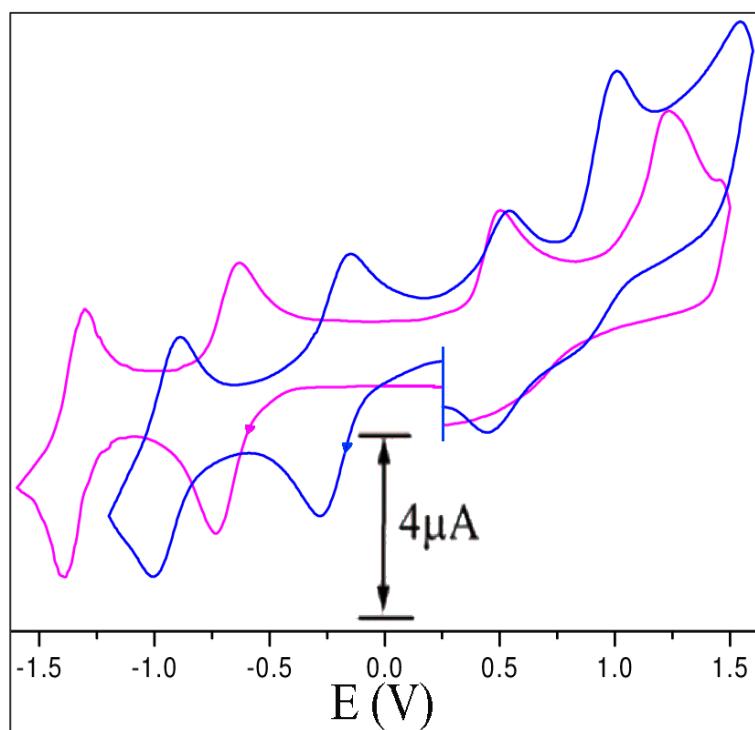


Figure S15. Cyclic voltammograms of the compound **[1][OTf]** (blue) and **2** (pink) in $\text{CH}_3\text{CN} / 0.1 \text{ M Et}_4\text{NClO}_4$.

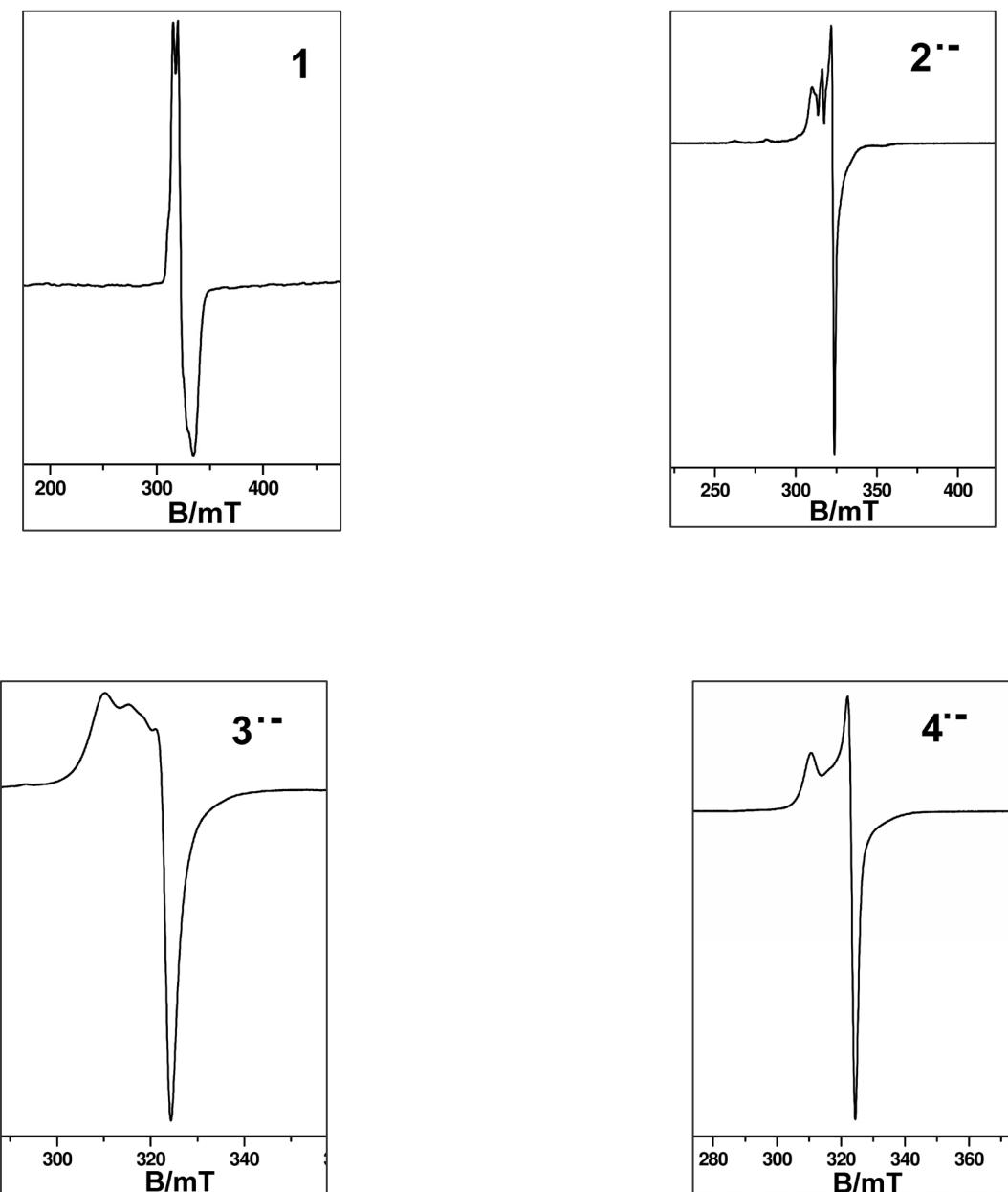


Figure S16. X-band EPR spectrum of compound **1**, **2^{·-}**, **3^{·-}**, **4^{·-}** in CH₃CN/0.1 M Et₄NClO₄ at 120K.

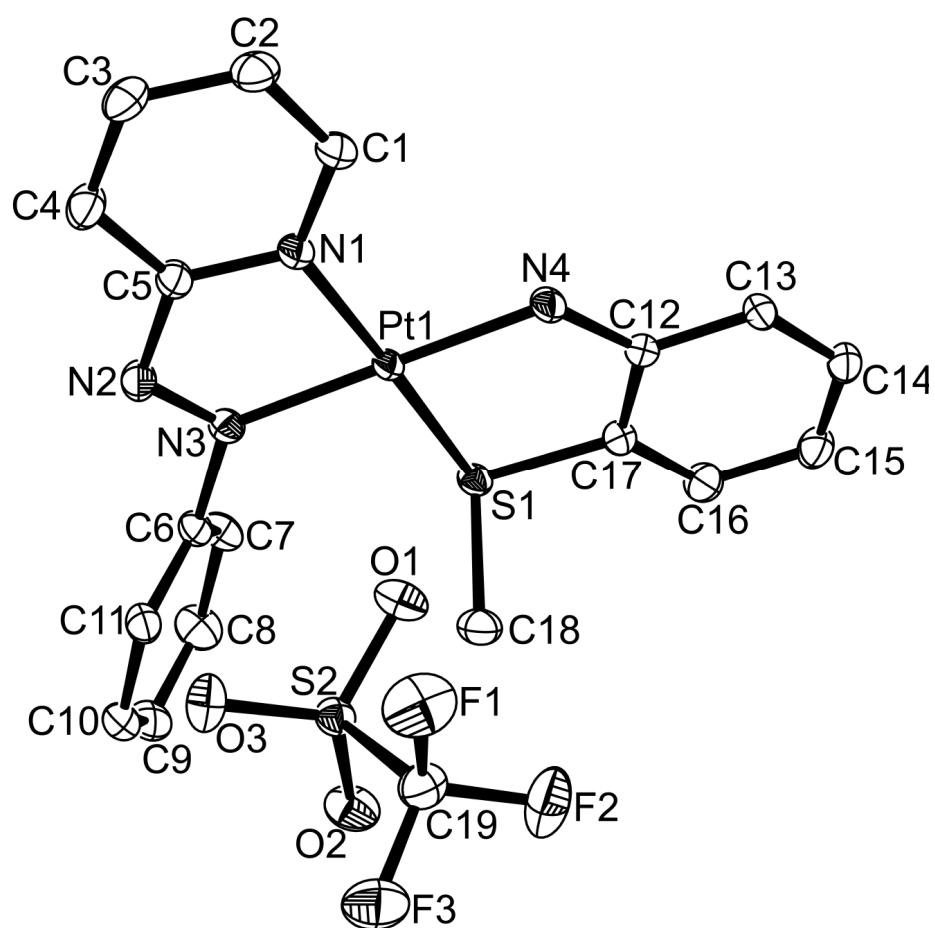


Figure S17. ORTEP and atom numbering scheme of the compound **[1][OTf]**

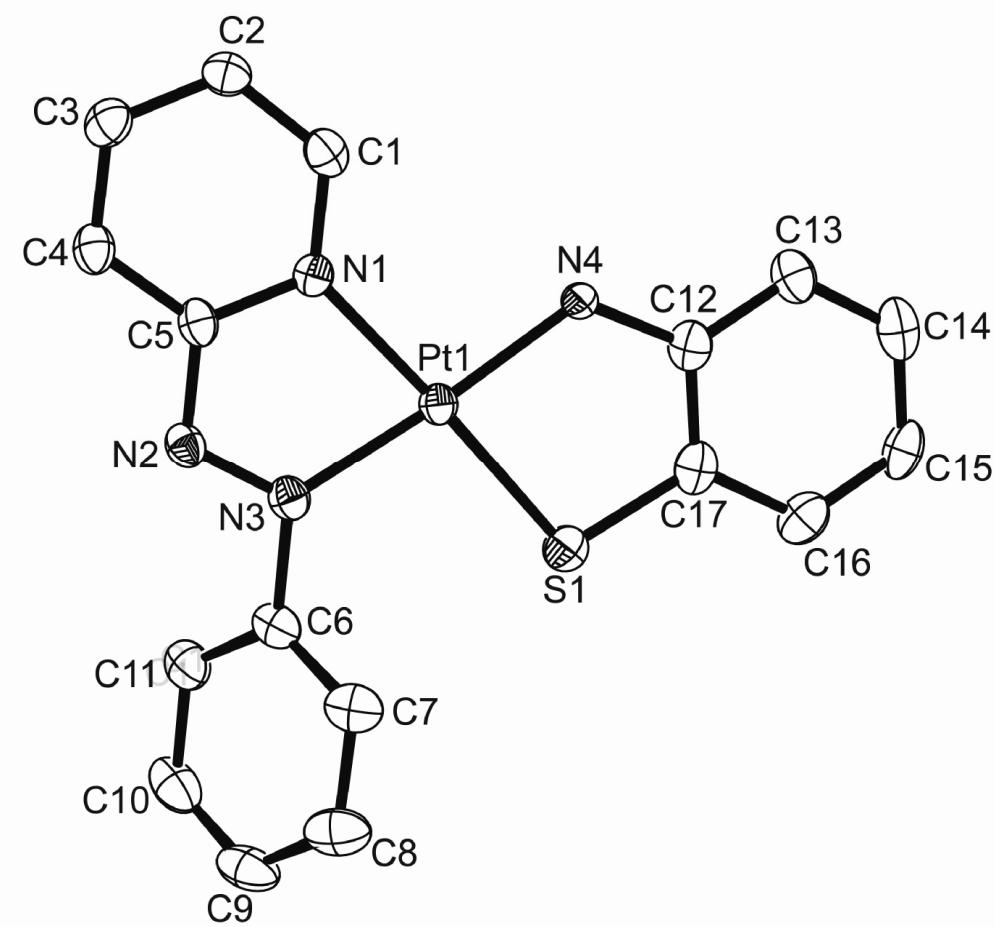


Figure S18. ORTEP atom numbering scheme of the compound 2

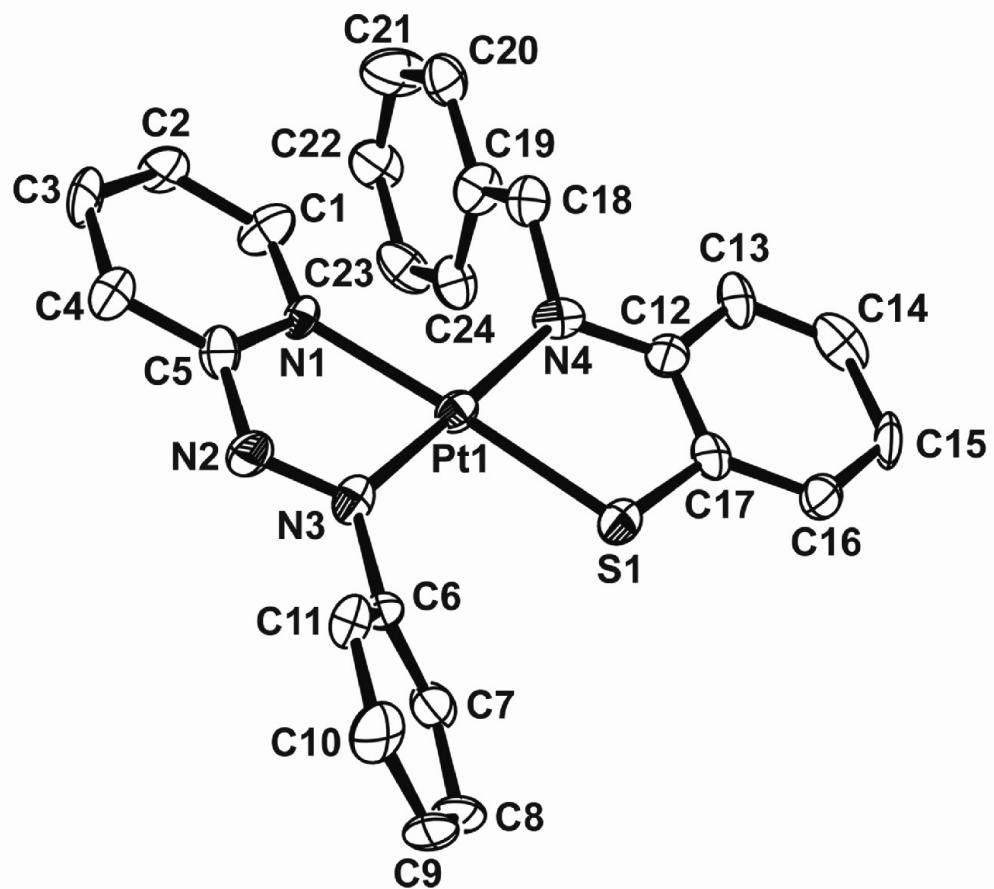


Figure S19. ORTEP atom numbering scheme of the compound 3

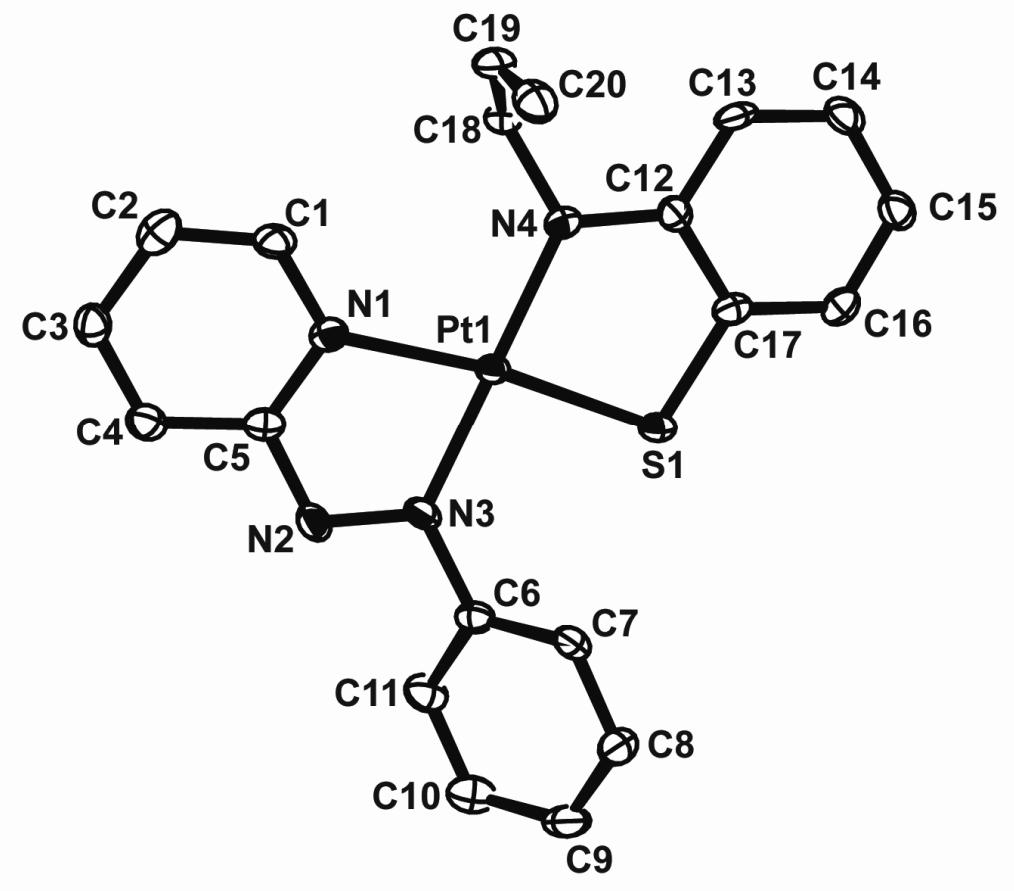


Figure S20. ORTEP atom numbering scheme of the compound 4

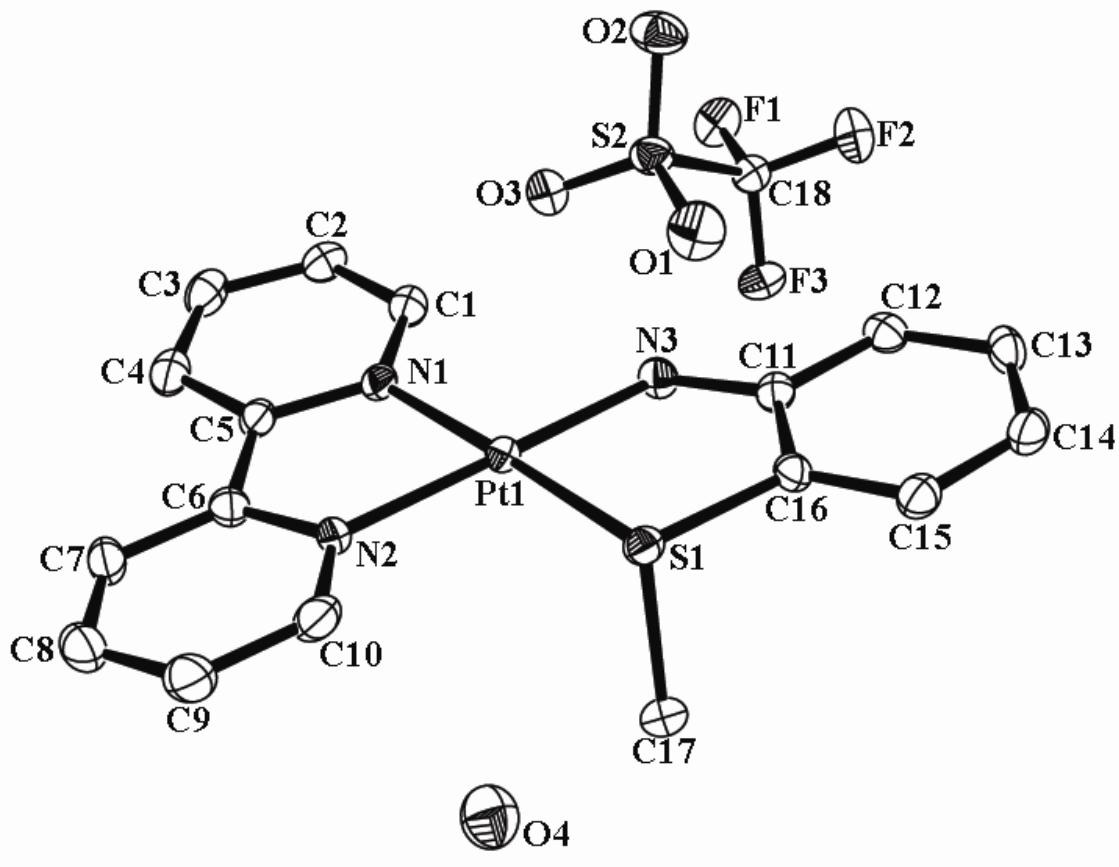


Figure S21. ORTEP atom numbering scheme of the compound **[5][OTf]**