Comparative investigation of the solution species $[U(CO_3)_5]^{6-}$ and the crystal structure of Na₆[U(CO₃)₅]·12H₂O

C. Hennig^{1,*}, A. Ikeda-Ohno^{1,2}, F. Emmerling³, W. Kraus³, G. Bernhard¹

¹Forschungszentrum Dresden-Rossendorf, Institute of Radiochemistry, P.O. Box 510119, 01314 Dresden, Germany

² Japan Atomic Energy Agency, Synchrotron Radiation Research Center (Spring-8), Kouto 1-1-1, Sayo-cho, Sayo-gun, 679-5148 Hyogo-ken, Japan.

> ³BAM Federal Institute for Materials Research and Testing, Richard-Willstätter-Str. 11, 12489 Berlin, Germany





Figure S1. Photo and electrode arrangement of electrochemical cell for electroreduction. Working electrode (WE) = Hg (Inner diameter = 3.0 mm), Counter electrode (CE) = Pt wire, Reference electrode (RE) = Ag/AgCl (with Vycor glass liguid junction). Sample volume > 2.0 ml



<Before electrolysis (U(VI))> <After electrolysis (U(IV))>
Figure S2. Color variation of uranium in 1.0 M NaHCO3 before and after the electrolysis.

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Empirical formula	$Na_{6}[U(CO_{3})_{5}] \cdot 12 H_{2}O$
Formula weight	892.19
Temperature [K]	293(2)
Wavelength [Å]	0.71073
Crystal system, space group	triclinic, $P\overline{1}$ (No. 2)
Unit cell dimensions [Å,°]	a = 9.563(2)
	b = 9.901(2)
	c = 13.658(3)
	$\alpha = 90.452(3)$
	$\beta = 104.692(3)$
	$\gamma = 95.427(3)$
Volume [Å ³]	1244.6(5)
Z / Calculated density [g/cm ³]	2 / 2.658
Absorption coefficient [mm ⁻¹]	6.740
F(000)	856
Crystal size	0.2 x 0.14 x 0.3 mm
θ range for data collection [°]	1.54 to 30.83
Limiting indices	$-13 \le h \le 13$,
-	$-13 \le k \le 14$,
	$-19 \le 1 \le 17$
Reflections collected / unique	9812 / 6887
R(int)	0.1258
Absorption correction	Empirical, ψ-scan
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	6887/0/353
Goodness-of-fit on F^2	1.019
Final R indices [I>2sigma(I)]	R1 = 0.0690, wR2 = 0.1775
R indices (all data)	R1 = 0.0718, $wR2 = 0.1807$
Largest diff. peak and hole $[e / Å^3]$	5.89 and -6.68

Table S3. Crystal data and structure refinement.

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Figure S4: Crystal structure drawn along the b-axis. The coordination polyhedron of *U(IV)* is blue, yellow - Na, grey - C, red - O, white - H.



Figure S5: The counter ion arrangement in the structure. The coordination polyhedron of Na⁺ is yellow, blue - U, grey - C, red - O, white - H.