

Supporting Informations

Effects of Axial Pyridine Coordination on a Saddle-Distorted Porphyrin Macrocycle: Stabilization of Hexa-coordinated High-Spin Fe(III) and Air-stable Low-Spin Iron(II) Porphyrinates

Ranjan Patra, Susovan Bhowmik, Sudip Kumar Ghosh and Sankar Prasad Rath*

Department of Chemistry, Indian Institute of Technology Kanpur, Kanpur-208016.

Email: sprath@iitk.ac.in

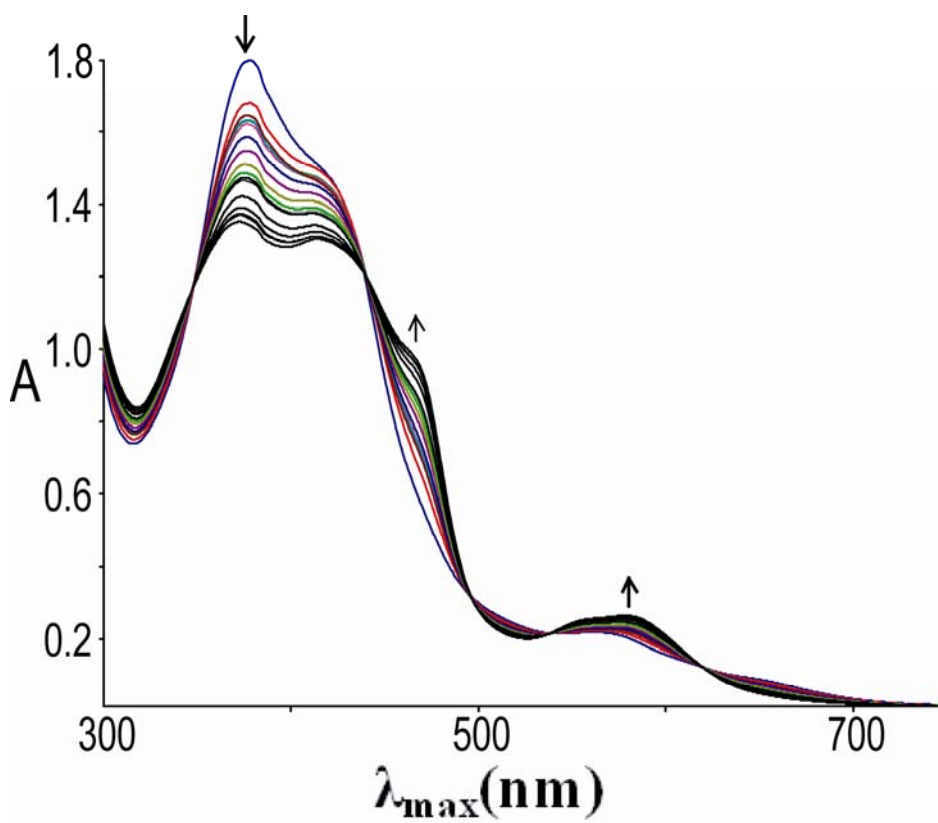


Figure S1. Time-evolution spectral changes (at 298K) of $\text{Fe}^{\text{III}}(\text{tn-OEP})\text{ClO}_4$ in chloroform in presence of excess pyridine in air showing first the appearance of six-coordinated $\text{Fe}^{\text{III}}(\text{tn-OEP})(\text{py})_2\text{ClO}_4$ which, on longer exposure, again converts to $\text{Fe}^{\text{II}}(\text{tn-OEP})(\text{py})_2$. Arrows indicate increase or decrease of band intensity.

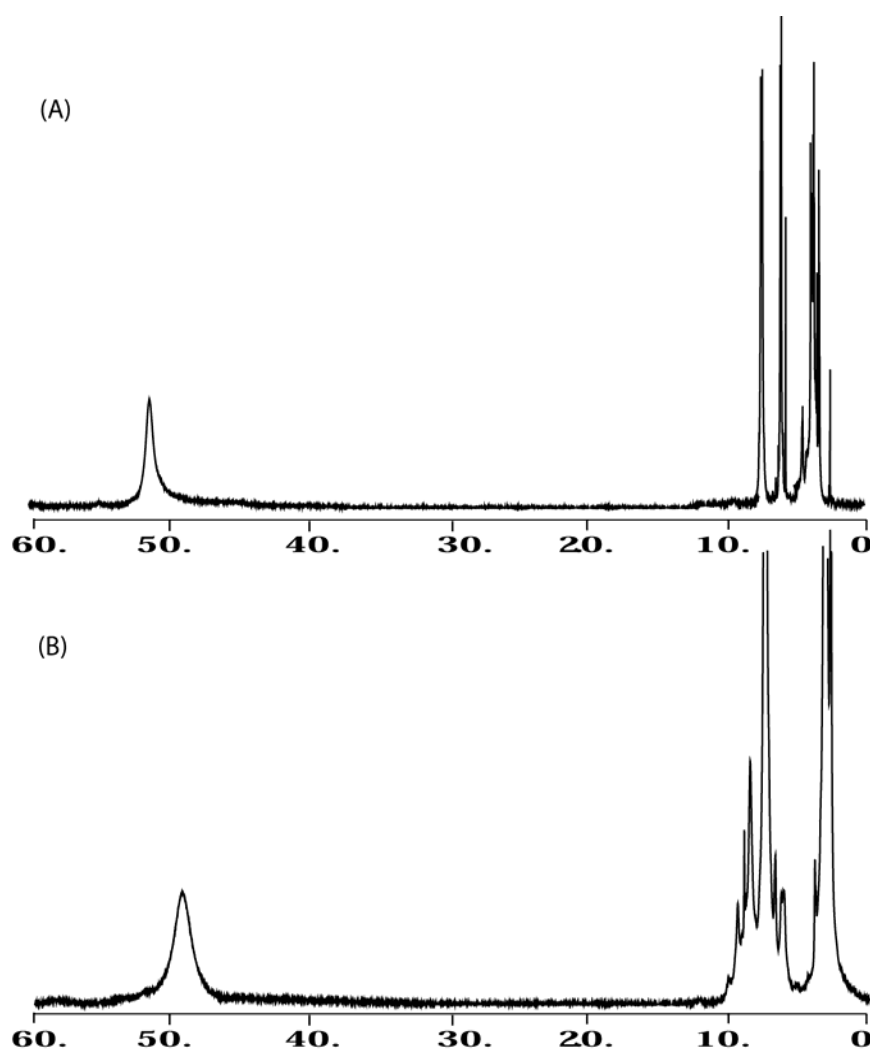


Figure S2: ^1H NMR spectra in CDCl_3 at 295K for (A) for $\text{Fe}^{\text{III}}(\text{tn-OEP})(4\text{-CNpy})_2\cdot\text{ClO}_4$ (B) for $\text{Fe}^{\text{III}}(\text{tn-OEP})(3\text{-Clpy})_2\cdot\text{ClO}_4$.

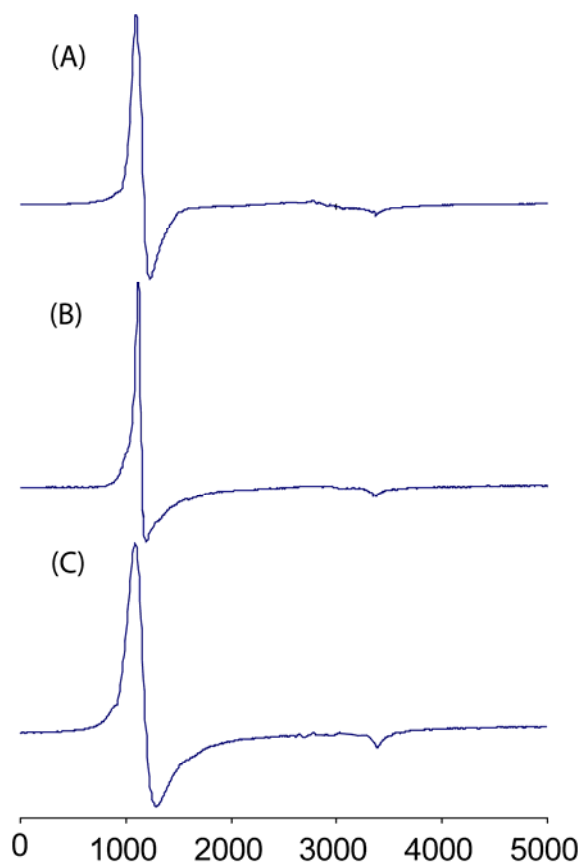


Figure S3. X-band EPR spectra of the isolated solid (at 120K) of (A) $\text{Fe}^{\text{III}}(tn\text{-OEP})(4\text{-CNpy})_2\text{ClO}_4$; (B) $\text{Fe}^{\text{III}}(tn\text{-OEP})(3\text{-Clpy})_2\text{ClO}_4$ and (C) $\text{Fe}^{\text{III}}(tn\text{-OEP})(\text{py})_2\text{ClO}_4$.

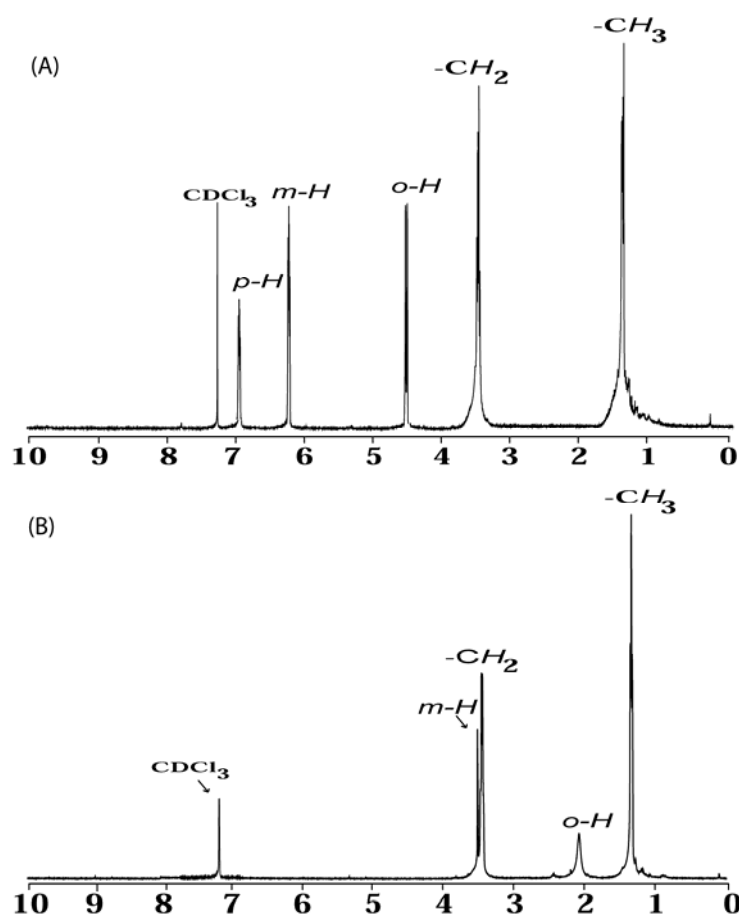


Figure S4: ^1H NMR spectra in CDCl_3 at 295K (A) for $\text{Fe}^{\text{II}}(\text{tn-OEP})(\text{py})_2$ and (B) for $\text{Fe}^{\text{II}}(\text{tn-OEP})(4\text{-CNpy})_2$

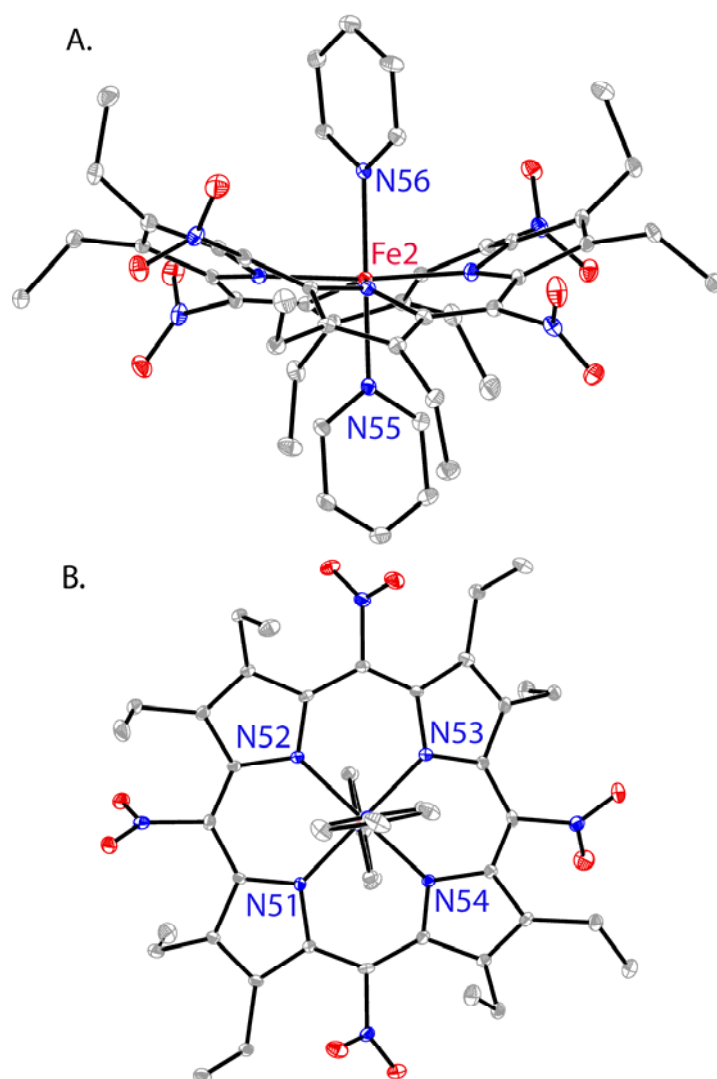


Figure S5. Two perspective views [A, side view; B, top view] for molecule 2 of Fe^{II}(*m*-OEP)(py)₂ showing 50% thermal contours for all non-hydrogen atoms at 100K (H-atoms have been omitted for clarity).

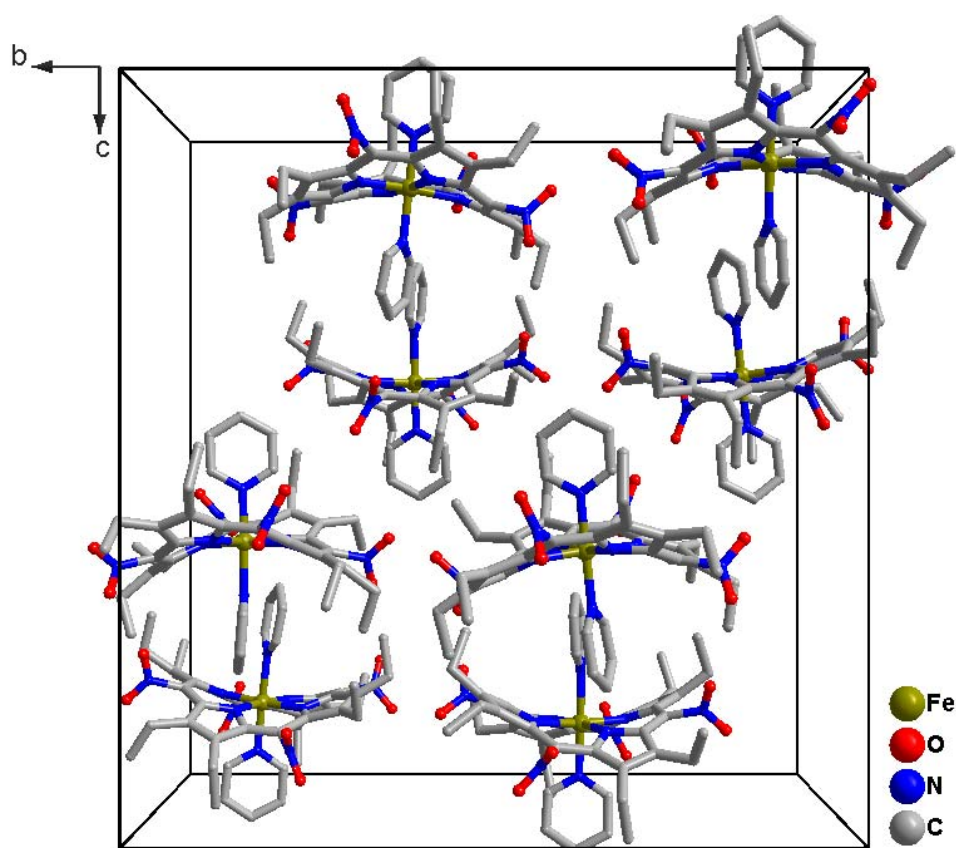


Figure S6. Diagram illustrating the packing of the $\text{Fe}^{\text{II}}(\text{tn-OEP})(\text{py})_2$ molecules in the unit cell.

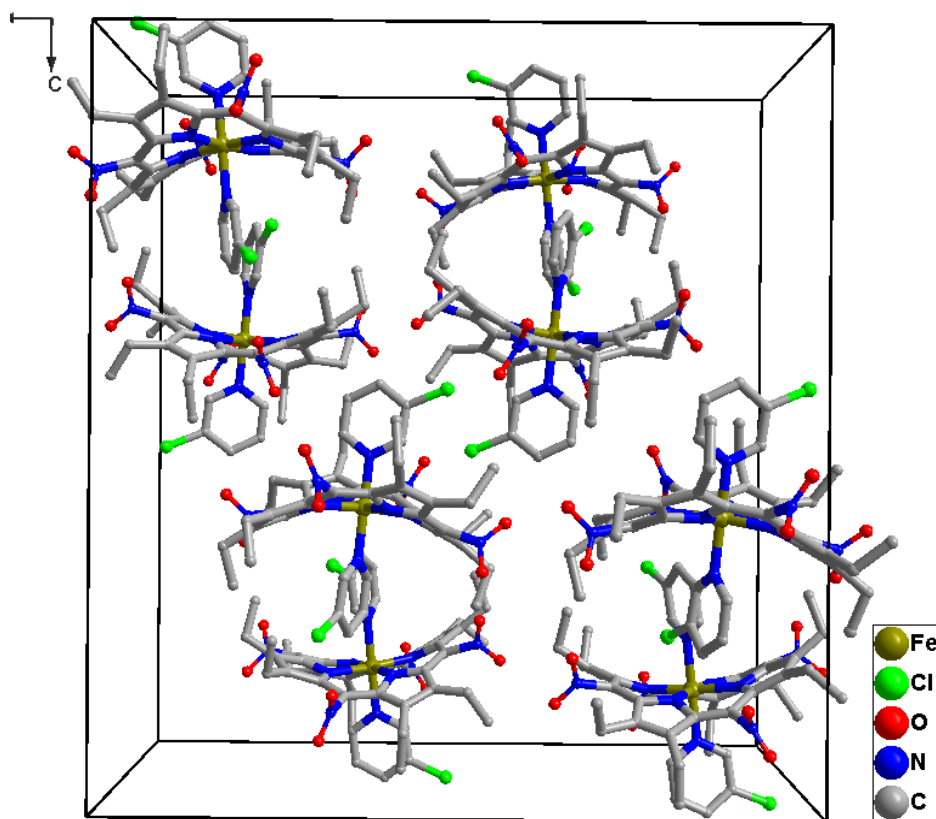


Figure S7. Diagram illustrating the packing of the $\text{Fe}^{\text{II}}(\text{mn-OEP})(3\text{-Clpy})_2$ molecules in the unit cell.

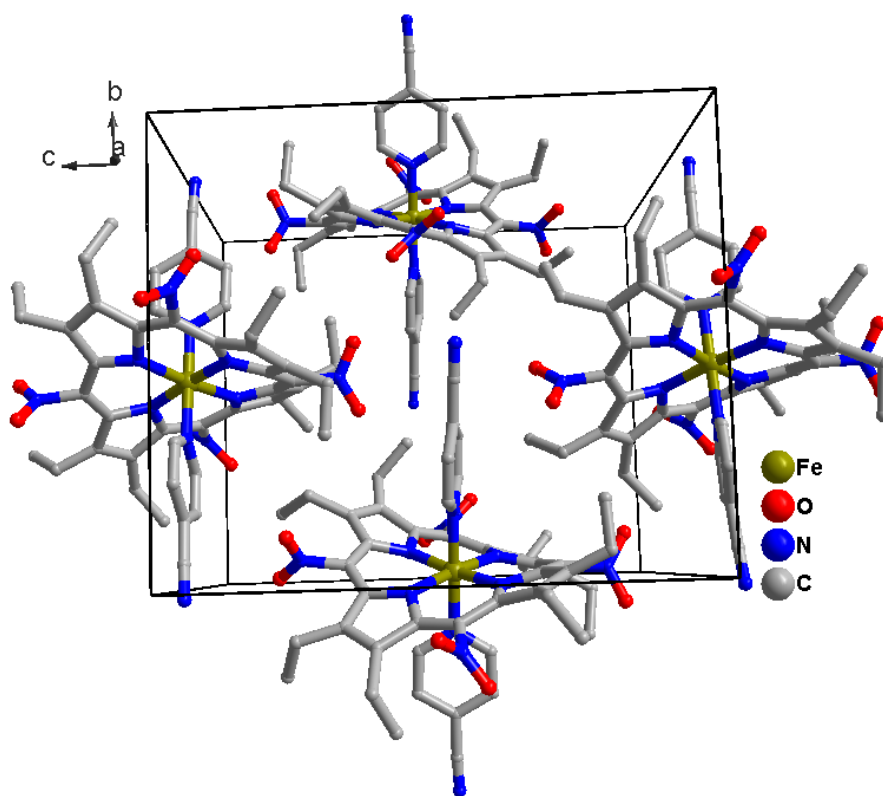


Figure S8. Diagram illustrating the packing of the $\text{Fe}^{\text{II}}(\text{tn-OEP})(4\text{-CNpy})_2$ molecules in the unit cell.

Table S1. Crystal Data and Data Collection Parameters

	Fe ^{II} (<i>tn</i> -OEP)(Py) ₂	Fe ^{II} (<i>tn</i> -OEP)(4-CNpy) ₂	Fe ^{II} (<i>tn</i> -OEP)(3-Clpy) ₂
T, K	100(2)	100(2)	100(2)
Formula	C ₄₆ H ₅₀ Fe N ₁₀ O ₈	C ₄₈ H ₄₈ Fe N ₁₂ O ₈	C ₄₆ H ₄₈ Cl ₂ Fe N ₁₀ O ₈
Formula weight	926.81	976.83	995.69
Color and Habit	Dark green	Dark green	Dark green
Crystal system	Orthorhombic	Monoclinic	Orthorhombic
Space group	P2(1)2(1)2(1)	P2(1)/c	Pbca
a, Å	13.4183(11)	19.4563(17)	13.3561(10)
b, Å	25.469(2)	14.1225(12)	25.439(2)
c, Å	26.501(2)	16.7267(14)	27.699(2)
α, deg	90	90	90
β, deg	90	93.694(2)	90
γ, deg	90	90	90
V, Å ³	9056.9(13)	4586.5(7)	9411.1(13)
Radiation (λ, Å)	Mo Kα	Mo Kα	Mo Kα
	(0.71073)	(0.71073)	(0.71073)
Z	8	4	8
d _{calcd} , g•cm ⁻³	1.359	1.415	1.405
μ, mm ⁻¹	0.398	0.398	0.498
F(000)	3888	2040	4144
No. of unique data	15904	8767	9247
No. of restraints	0	0	0
No. of params. Refined	1188	630	622
GOF on F ²	1.045	1.021	1.032
R1 ^a [<i>I</i> > 2σ(<i>I</i>)]	0.0544	0.0468	0.0558
R1 ^a (all data)	0.0726	0.0730	0.0905
wR2 ^b (all data)	0.1288	0.1219	0.1381

$$^a R1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}; \quad ^b wR2 = \sqrt{\frac{\sum [w(F_o^2 - F_c^2)^2]}{\sum [w(F_o^2)^2]}}$$

