# Supporting Information

## Coordination between $[H_2P_2W_{12}O_{48}]^{12-}$ and Antimony (III): Synthesis and Characterization of Sandwich Complex Derived from the New $\{P_2W_{13}O_{51}\}$ Fragment

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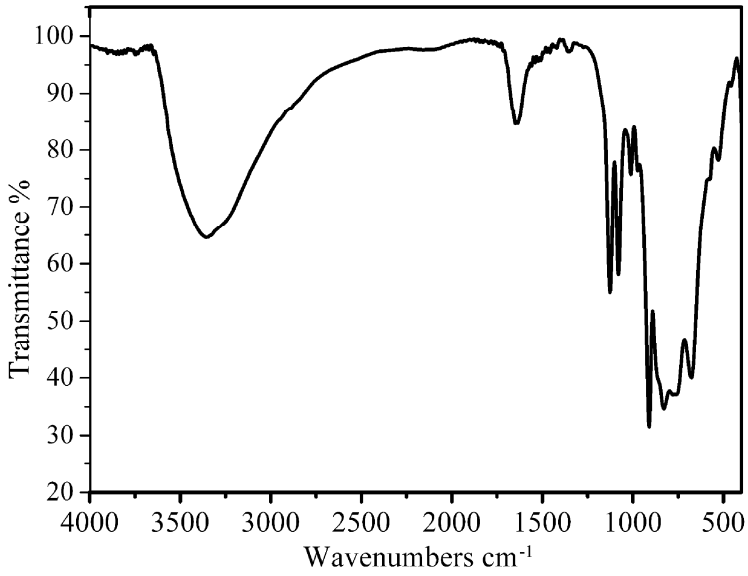


**Figure S1.** IR spectra of compound $K_{12}[H_2P_2W_{12}O_{48}]\cdot 24H_2O$. The P−O bands are at 1126, 1080 and 1011 $cm^{-1}$.

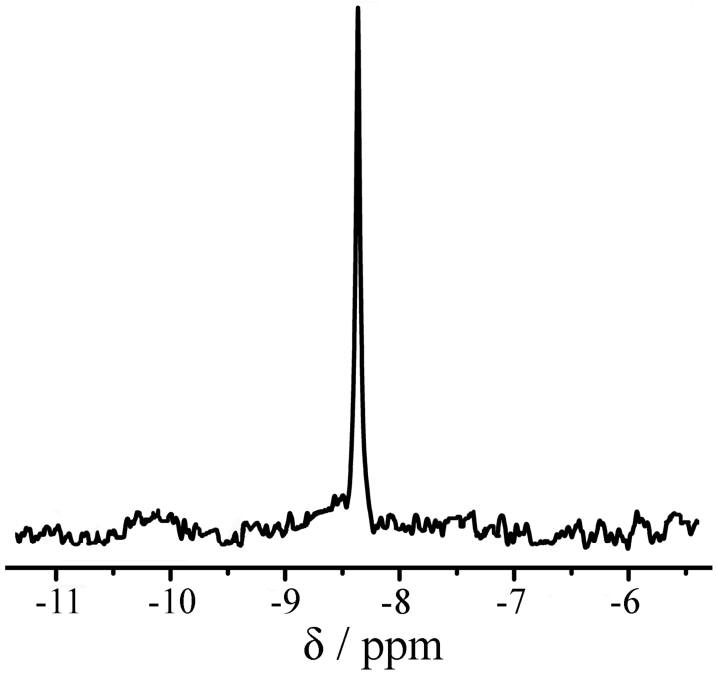


**Figure S2.** $^{31}P$ NMR spectrum of a freshly prepared solution ($K_{12}[H_2P_2W_{12}O_{48}]\cdot 24H_2O$) in lithium chloride exhibits a single resonance at −8.4 ppm at room temperature.

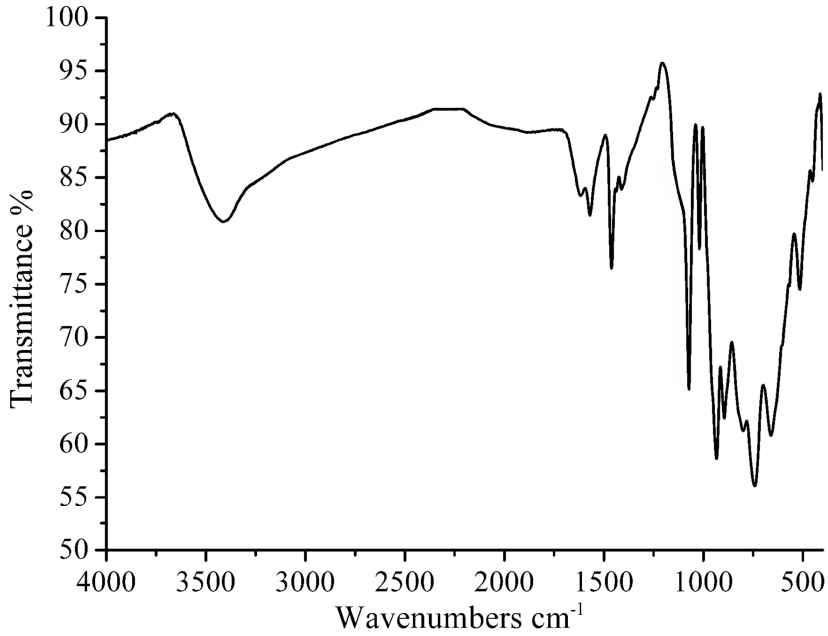


**Figure S3.** IR spectra of compound $K_{10}$−**1**$\cdot 21H_2O$**.**

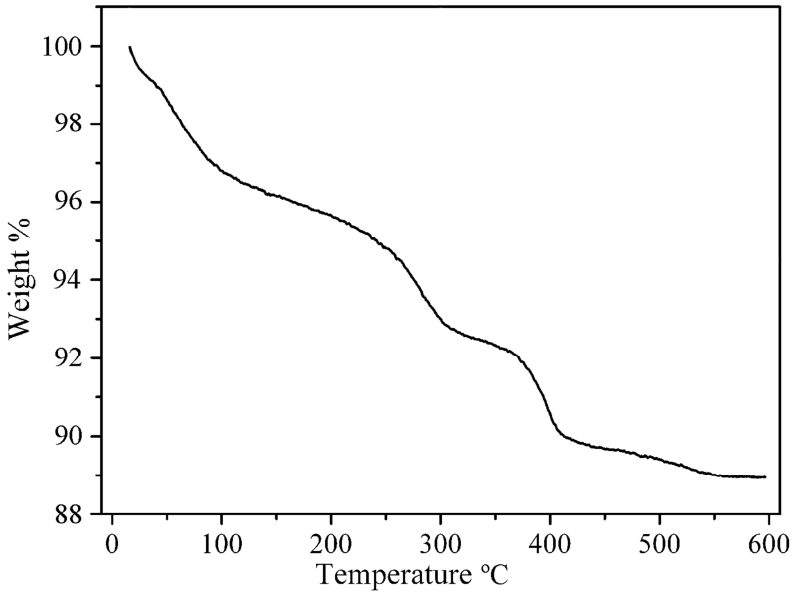


**Figure S4.** TG curve of compound $K_{10}$–**1**·$21H_2O$. Heating rate: 10 deg $min^{-1}$, $N_2$ atmosphere.

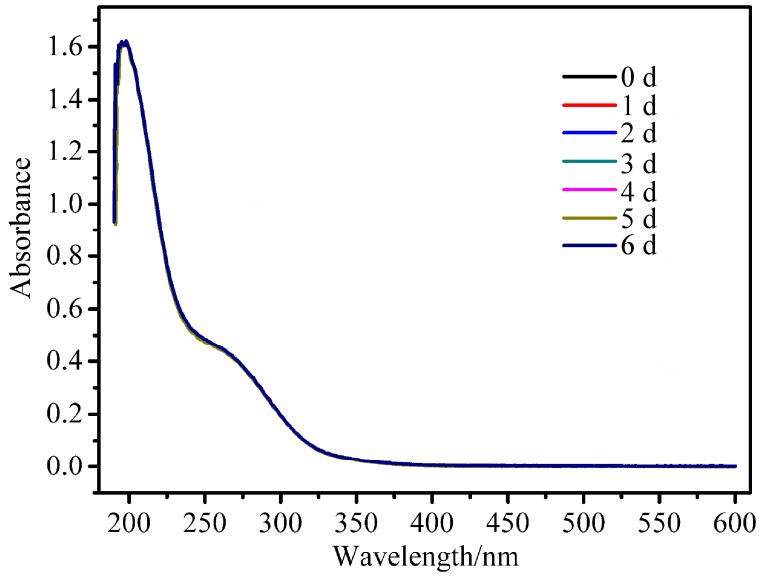


**Figure S5.** The influence of time on the stability of **1** in the aqueous solution. Conditions: the pH value that **1** was dissolved in water ($4 \times 10^{-4}$ mol $L^{-1}$) is 6.0.

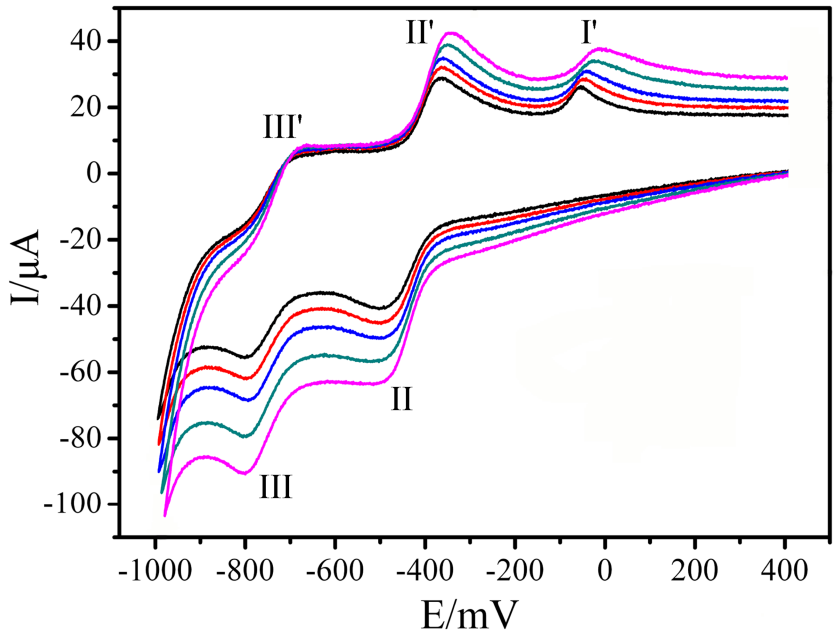


**Fig. S6.** Cyclic voltammograms of **1** (1.0 mM) in the pH 4.7 (0.5 M $CH_3COOK$ + $CH_3COOH$) buffer solution at different scan rates (from inner to outer: 40, 50, 60, 80, 100 $mV \cdot s^{-1}$), the working electrode was glassy carbon, and the reference electrode was Ag/AgCl.