

Tuning the Electronic Properties of dppz-Ligands and their Palladium(II) Complexes

Katharina Butsch,^a Ronald Gust,^b Axel Klein^{a*}, Ingo Ott,^c and Marco Romanski^a

Supplementary Information

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Figures S2-S7. Selected optical spectra (absorption, emission and excitation).

Supplementary Tables

Table S1. Complete data of absorption and luminescence measurements.

Table S2. Complete electrochemical data of free ligands and Pd^{II} complexes. Potentials in V vs. ferrocene/ferrocinium from cyclic voltammetry or square wave voltammetry in 0.1 M DMF/*n*Bu₄NPF₆ solutions at 298 K; Half wave potentials $E_{1/2}$ given for reversible processes, cathodic peak potentials E_{pc} for irreversible processes.

Table S3. Short H...F contacts in the crystal structure of [(dppz)PdMe(Py)][SbF₆]

In Situ measurement of [(tfmdppz)Pd(Me)(Acetone)]⁺: [(tfmdppz)Pd(Me)Cl] was suspended in 1 mL of (CD₃)₂CO and an excess of AgSbF₆ was added as a solid. After 5 min reaction time at room temperature the mixture was filtered and the solution was analysed with NMR spectroscopy. δ_{H} (300 MHz; (CD₃)₂CO): 9.79 (1 H, m, H_{dppz8}), 9.69 (1 H, m, H_{dppz1}), 9.04 (1 H, m, H_{dppz6}), 8.74 (1 H, s, H_{dppz10}), 8.69 (1 H, m, H_{dppz3}), 8.61 (1 H, m, H_{dppz13}), 8.34 (1 H, m, H_{dppz12}), 8.24 (2 H, m, H_{dppz2,7}), 1.04 (3 H, s, H_{Pd-CH3}). δ_{C} (300 MHz; (CD₃)₂CO): 153.4 (1 C, d, C_{dppz6}), 153.2 (1 C, d, C_{dppz6'}), 150.9 (1 C, d, C_{dppz3}), 150.6 (1 C, d, C_{dppz3'}), 137.3 (1 C, d, C_{dppz8}), 137.2 (1 C, d, C_{dppz8'}), 136.5 (1 C, d, C_{dppz1}), 136.3 (1 C, d, C_{dppz1'}), 132.2 (1 C, d, C_{dppz2}), 132.0 (1 C, d, C_{dppz2}), 132.0 (1 C, d, C_{dppz13}), 128.4 (1 C, d, C_{dppz7}), 128.2 (1 C, d, C_{dppz7'}), 128.0 (1 C, d, C_{dppz10}), 127.8 (1 C, d, C_{dppz12}), 3.3 (1 C, q, C_{Pd-CH3}).

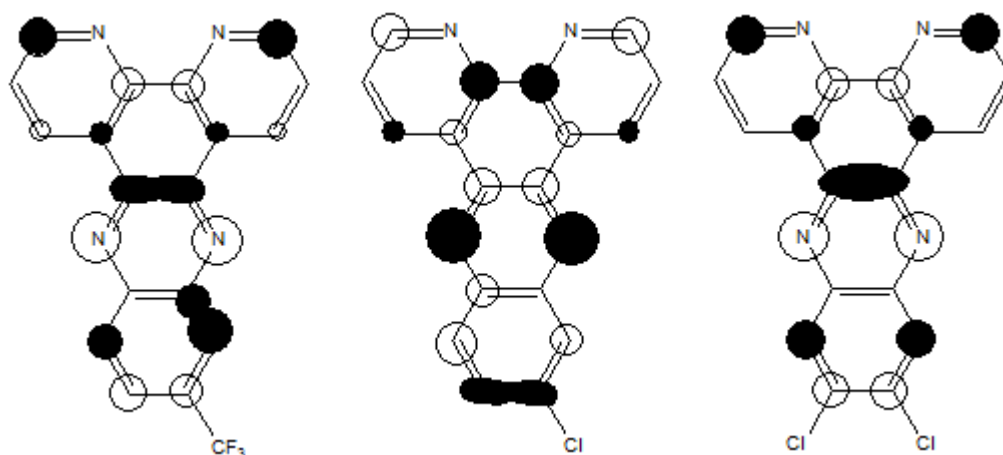
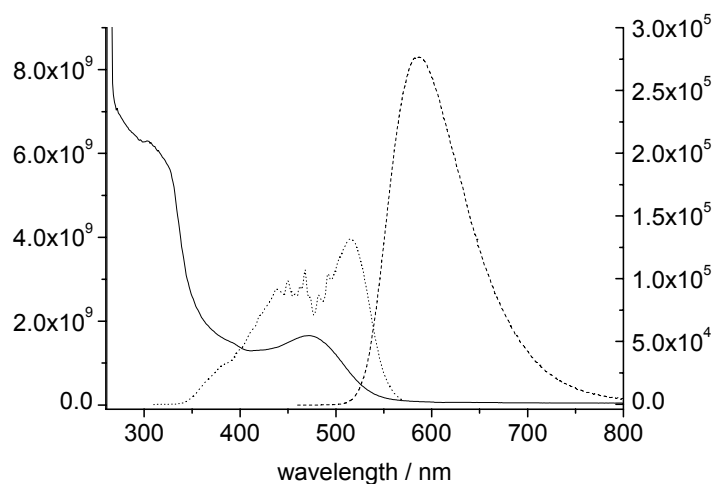


Figure S1. Calculated (B3-LYP, TZVP) electron density for the LUMOs of tfmdppz (left), cdppz (middle) and dcdppz (right).



Figures S2. Absorption (solid), emission (dashed) and excitation (dotted) spectra of $[(adppz)Pd(Me)(Caf)][SbF_6]$ measured in DMF solution at room temperature.

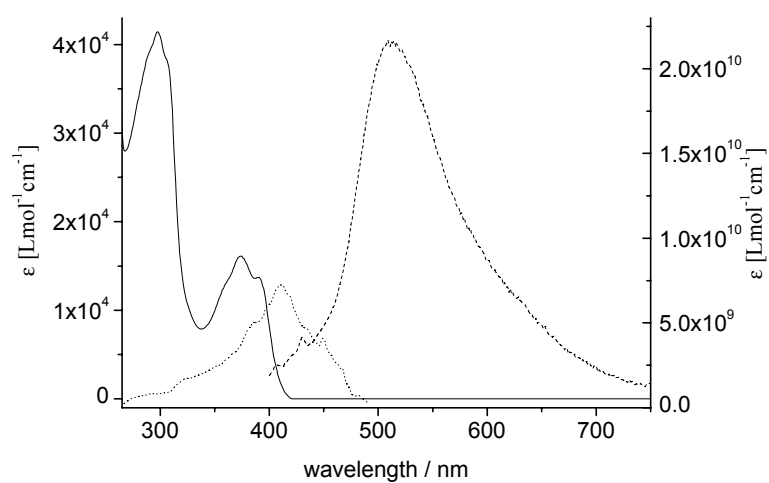


Figure S3. Absorption (solid), emission (dashed) and excitation (dotted) spectra of ndppz measured in DMF solution at room temperature

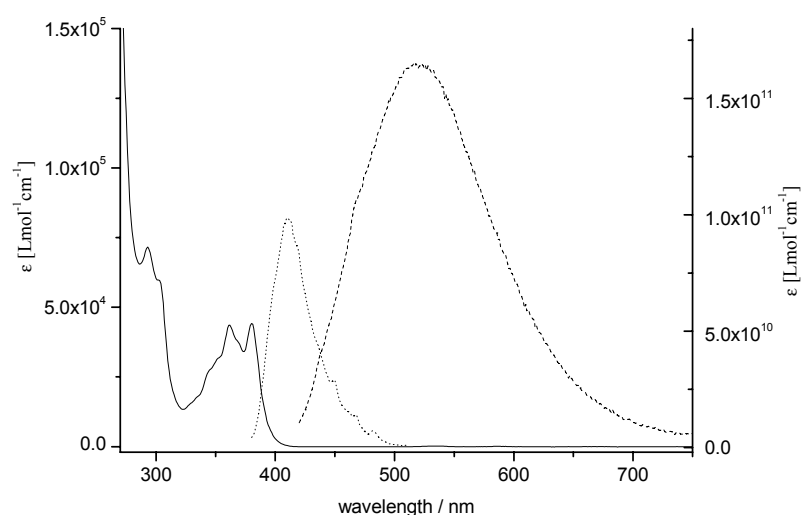


Figure S4. Absorption (solid), emission (dashed) and excitation (dotted) spectra of tfmdppz measured in DMF solution at room temperature

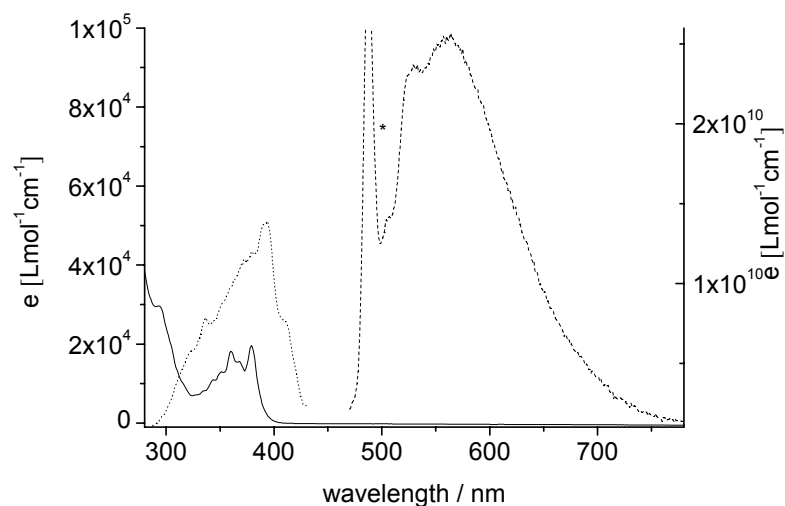


Figure S5. Absorption (solid), emission (dashed) and excitation (dotted) spectra of $[(\text{dppz})\text{Pd}(\text{Me})(\text{Caf})][\text{SbF}_6]$ measured in DMF solution at room temperature.

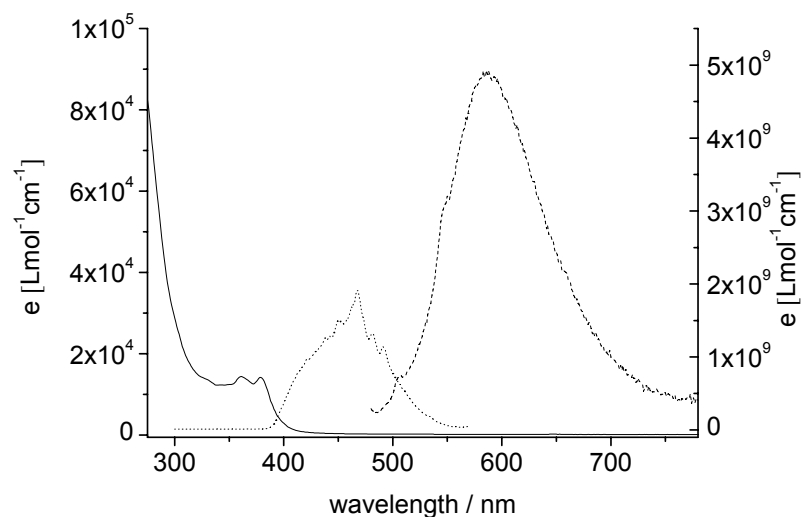


Figure S6. Absorption (solid), emission (dashed) and excitation (dotted) spectra of $[(\text{tfmdppz})\text{Pd}(\text{Me})(\text{Caf})][\text{SbF}_6]$ measured in DMF solution at room temperature.

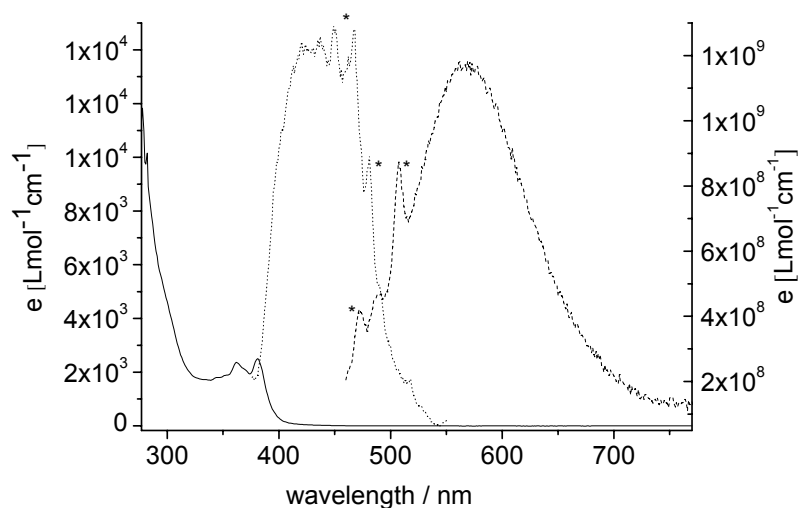


Figure S7. Absorption (solid), emission (dashed) and excitation (dotted) spectra of [(dppz)Pd(Me)(1MeUra)] measured in DMF solution at room temperature.

Table S1. Complete data of absorption and luminescence measurements

compound	$\tilde{\nu}_{\text{Abs.max}}^a$	$\tilde{\nu}_{\text{Exc.max}}^a$	$\tilde{\nu}_{\text{Em.max}}^a$	Stokes-shift (1) ^b	Stokes-shift (2) ^b
adppz	34130 / 32468 / 22371	21277	17900	4471	3377
[(adppz)Pd(Me)(Caf)] ⁺	32895 / 21186	22222	17065	4121	5157
dppz	34130 / 27855 / 26385	25000	22222	4163	2778
[(dppz)Pd(Me)(Caf)] ⁺	34247 / 27778 / 26385	22222	17700	8685	4522
tfindppz	34130 / 27778 / 26385	24390	19200	7185	5190
[(tfindppz)Pd(Me)(Caf)] ⁺	36765 / 27778 / 26455	21277	17100	9355	4177
cdppz	34247 / 27322 / 25907	23810	19400	6507	4410
[(cdppz)Pd(Me)(Caf)] ⁺	36363 / 27100 / 25707	22222	16000	9707	6222
dcdppz	33900 / 26954 / 25510	20000	18400	7110	1600
[(dcdppz)Pd(Me)(Caf)] ⁺	35971 / 26882 / 25445	21739	17300	8145	4439
ndppz	33557 / 26810 / 25575	24390	19500	6075	4890
[(ndppz)Pd(Me)(Caf)] ⁺	34965 / 26954 / 28169 / 25575	22222	17400	8175	4822

^a Absorption (Abs.max), Excitation (Exc.max) and Emission maxima (Em.max) / cm^{-1} .

^b Stokes-shift (1) = $\tilde{\nu}_{(\text{Abs.max})} - \tilde{\nu}_{(\text{Em.max})} / \text{cm}^{-1}$; Stokes-shift (2) = $\tilde{\nu}_{(\text{Exc.max})} - \tilde{\nu}_{(\text{Em.max})} / \text{cm}^{-1}$.

Table S2. Complete electrochemical data of free ligands and Pd^{II} complexes Potentials in V vs. ferrocene/ferrocinium from cyclic voltammetry or square wave voltammetry in 0.1 M DMF/*n*Bu₄NPF₆ solutions at 298 K; Half wave potentials E_{1/2} given for reversible processes, cathodic peak potentials E_{pc} given for irreversible processes.

Compound	E _{1/2}	E _{1/2} or E _{pc2}	ΔE _{Red1} – E _{Red2}	E _{pc3}	E _{pc4}	solvent
ndppz	-1.08	-1.53	0.44	-2.73	-2.80	CH ₃ CN
tfindppz	-1.45	-2.29	0.78	-2.55	-2.94	CH ₃ CN
dcdppz	-1.45	-2.20	0.73	-2.34	-2.87	CH ₃ CN
cdppz	-1.61	-2.42	0.79	-2.48	-2.93	CH ₃ CN
dppz	-1.64	-2.46	0.78	-2.94	/	CH ₃ CN
ndppz	-1.18	-1.73	0.53	-2.91	/	DMF
tfindppz	-1.47	-2.31	0.78	-2.87	/	DMF
dcdppz	-1.35	-2.16	0.76	-2.77	/	DMF
cdppz	-1.53	-2.33	0.76	-2.33	-2.85	DMF
dppz	-1.66	-2.46	0.73	-2.89	/	DMF
adppz	-1.84	-1.88	0.51	-2.84	/	DMF
[(ndppz)Pd(Me)(Caf)] ⁺	-1.03	-1.56	0.57	-2.39	/	DMF
[(tfindppz)Pd(Me)(Caf)] ⁺	-1.20	-2.02	0.72	-2.38	/	DMF
[(dcdppz)Pd(Me)(Caf)] ⁺	-1.23	-2.06	0.79	-2.33	/	DMF
[(cdppz)Pd(Me)(Caf)] ⁺	-1.27	-2.01	0.73	-2.30	/	DMF
[(dppz)Pd(Me)(Caf)] ⁺	-1.44	-2.21	0.86	-2.89	/	DMF
[(adppz)Pd(Me)(Caf)] ⁺	-1.38	-2.14	0.78	-3.18	/	DMF
[(ndppz)Pd(Me)Cl]	-1.05	-1.54 (irr)	0.49	-2.38	-2.91	DMF
[(tfindppz)Pd(Me)Cl]	-1.25	-2.08 (irr)	0.73	-2.33	-2.92	DMF
[(dcdppz)Pd(Me)Cl]	-1.33	-2.12 (irr)	0.79	-2.33	-2.51	DMF
[(cdppz)Pd(Me)Cl]	-1.33	-2.07 (irr)	0.64	-2.30	-2.81	DMF
[(dppz)Pd(Me)Cl]	-1.35	-2.06 (irr)	0.71	/	/	DMF

Table S3. Short H...F contacts in the crystal structure of [(dppz)PdMe(Py)][SbF₆]

F1...H8	2.586(7)	F4...H4	2.591(4)
F1...H9	2.610(8)	F4...H15	2.621(6)
F2...H5	2.695(5)	F5...H2	2.420(5)
F2...H6	2.629(5)	F5...H6	2.845(5)
F3...H3	2.665(5)	F5...H9	2.691(6)
F3...H14	2.538(6)	F6...H5	2.573(5)
F3...H23	2.439(5)	F6...H22	2.662(6)
F4...H3	2.594(4)		