

**Supporting Information
to the paper entitled**

**Efficient, stable, tunable, and easy to synthesize, handle and recycle luminescent materials:
[H₂NMe₂]₃[Ln(III)(2,6-dipicolinolate)₃] (Ln = Eu, Tb, or its solid solutions)**

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Table S1 Crystal data and structure refinement data for [H₂N(CH₃)₂]₃[Eu(2,6-dpa)₃] (**1^{Eu}**) and [H₂N(CH₃)₂]₃[Tb(2,6-dpa)₃]•2H₂O (**2^{Tb}**).

Compound	1^{Eu}	2^{Tb}
Empirical formulae	C ₂₁ H ₉ EuN ₃ O ₁₂ , 3(C ₂ H ₈ N)	C ₂₁ H ₉ TbN ₃ O ₁₂ , 3(C ₂ H ₈ N), 2H ₂ O
Formula weight	785.55	828.54
Temperature (K)	293(2)	293(2)
Wavelength (Å)	0.71073	0.71073
Crystal system	triclinic	orthorhombic
Space group	P-1	Pc2 ₁ n
Unit cell dimensions		
<i>a</i> (Å)	12.0026(2)	10.6792(10)
<i>b</i> (Å)	12.0692(2)	17.0739(3)
<i>c</i> (Å)	12.4073(2)	18.7096(3)
α (°)	78.6380(7)	90
β (°)	69.8886(7)	90
γ (°)	72.6756(8)	90
Volume (Å ³)	1602.19(5)	3411.43
Z	2	4
Calculated density (Mg/m ³)	1.628	1.613
Absorption coefficient (mm ⁻¹)	2.027	2.146
F(000)	792	1672
Crystal size (mm)	0.06 x 0.08 x 0.15	0.3 x 0.1 x 0.1
Theta range for data collection (°)	3.4 – 27.0	2.18 – 27.48
Index ranges	-13<=h<=15, -15<=k<=15, -15<=l<=15	-13<=h<=13, -22<=k<=22, -24<=l<=24
Reflections collected	11707	63289
Independent reflections	6922	4033
Refinement method	Full-matrix least-squares on F^2	Full-matrix least-squares on F
Data / restraints / parameters	3899 / 0 / 433	3899 / 7 / 433
<i>wR</i> (F^2)	0.0286	0.024
Goodness-of-fit on F^2	1.090	1.114
Largest diff. peak and hole (e.Å ⁻³)	0.96 and -0.67	1.044 and -2.211

Table S2 Selected bond distances (in Å) for $[\text{H}_2\text{N}(\text{CH}_3)_2]_3[\text{Eu}(2,6\text{-dpa})_3]$ ($\mathbf{1}^{\text{Eu}}$) and $[\text{H}_2\text{N}(\text{CH}_3)_2]_3[\text{Tb}(2,6\text{-dpa})_3]\bullet 2\text{H}_2\text{O}$ ($\mathbf{2}^{\text{Tb}}$).
 $[\text{Eu}(2,6\text{-dpa})_3]$ ($\mathbf{1}^{\text{Eu}}$) | $[\text{Tb}(2,6\text{-dpa})_3]$ ($\mathbf{2}^{\text{Tb}}$)

Coordination sphere			
Eu(1) – O(1)	2.409(2)	Tb(1) – O(2)	2.440(5)
Eu(1) – O(2)	2.477(2)	Tb(1) – O(4)	2.419(10)
Eu(1) – O(5)	2.446(2)	Tb(1) – O(8)	2.450(4)
Eu(1) – O(6)	2.427(2)	Tb(1) – O(11)	2.440(5)
Eu(1) – O(9)	2.455(2)	Tb(1) – O(16)	2.440(5)
Eu(1) – O(10)	2.466(2)	Tb(1) – O(24)	2.441(10)
Eu(1) – N(1)	2.553(2)	Tb(1) – N(3)	2.480(6)
Eu(1) – N(2)	2.521(3)	Tb(1) – N(5)	2.540(6)
Eu(1) – N(3)	2.560(2)	Tb(1) – N(28)	2.513(6)
<i>C–O(Eu/Tb)</i>			
C(1) – O(1)	1.260(4)	C(23) – O(2)	1.297(13)
C(7) – O(2)	1.262(4)	C(10) – O(4)	1.250(6)
C(8) – O(5)	1.250(4)	C(18) – O(8)	1.232(12)
C(14) – O(6)	1.259(5)	C(12) – O(11)	1.280(6)
C(15) – O(9)	1.260(4)	C(21) – O(16)	1.250(6)
C(21) – O(10)	1.259(4)	C(32) – O(24)	1.310(6)
<i>C–O</i>			
C(1) – O(3)	1.237(4)	C(23) – O(15)	1.230(4)
C(7) – O(4)	1.245(4)	C(10) – O(26)	1.226(16)
C(8) – O(7)	1.257(4)	C(18) – O(17)	1.240(3)
C(14) – O(8)	1.245(5)	C(12) – O(36)	1.220(5)
C(15) – O(11)	1.244(5)	C(21) – O(34)	1.250(5)
C(21) – O(12)	1.247(4)	C(32) – O(29)	1.228(14)

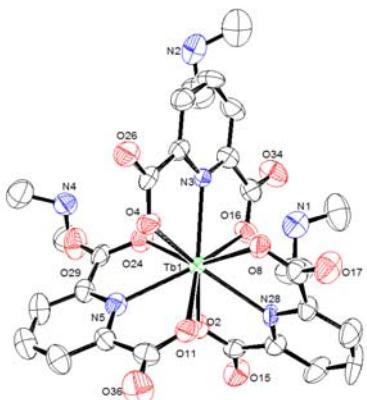


Fig. S1 Perspective view of $[\text{H}_2\text{N}(\text{CH}_3)_2]_3[\text{Tb}(2,6\text{-dpa})_3]\bullet 2\text{H}_2\text{O}$ ($\mathbf{2}^{\text{Tb}}$). Carbon = white, Oxygen = red, Nitrogen = blue, Europium = green. Hydrogen atoms are omitted for clarity, and ellipsoids are represented at the 50% probability level. **Note:** One water molecule (O2w) resides unbound within the crystal lattice. The other water molecule (O3w) is bridging two complexes via hydrogen bonding with two COO moieties ($\text{O3w} – \text{O}17 = 2.917 \text{ \AA}$; $\text{O3w} – \text{O}24 = 2.984 \text{ \AA}$). O3w also hydrogen bonded to a $\text{H}_2\text{N}(\text{CH}_3)_2$ cation ($\text{O3w} – \text{H}4\text{a} = 2.373 \text{ \AA}$; $\text{N}4 – \text{H}4\text{a} = 0.969 \text{ \AA}$).

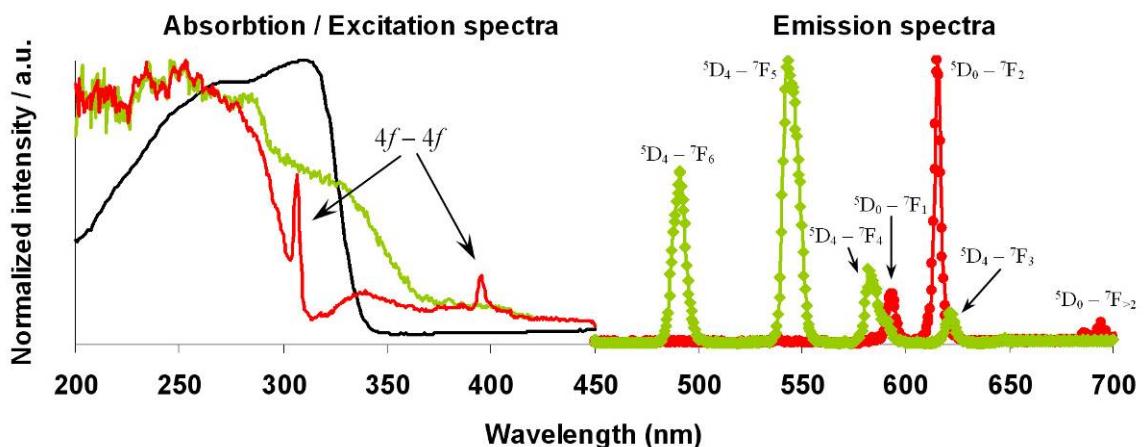


Fig. S2 Left: Absorption spectrum of the ligand **2,6-H₂dpa** (black line) and excitation spectra of **1^{Eu}** ($[H_2N(CH_3)_2]_3[Eu(2,6\text{-dpa})_3]$, red line, recorded at 615 nm), and **2^{Tb}** ($[H_2N(CH_3)_2]_3[Tb(2,6\text{-dpa})_3]$, green line, recorded at 545 nm). Right: emission spectra of **1^{Eu}** (red dots, excited at 254 nm), and **2^{Tb}** (green rhomboids, excited at 254 nm). The transitions between two specific energy levels are specified in the figure.



Fig. S3 Artistic impression of complex **1^{Eu}** (left), accompanied by early Christmas greetings painted with the compounds **1^{Eu} - 3^{Eu/Tb}** and visualized by excitation at 254 nm (right).