

Supplemental Material

Synthesis, Characterization and Structures of Cyclic Organorhodium Complexes of the Type $[\text{Rh}\{\text{CH}(\text{SO}_2\text{Ph})\text{CH}_2\text{CH}_2\text{YR}_2-\kappa\text{C},\kappa\text{Y}\}\text{L}_2]$ ($\text{YR}_2 = \text{PPh}_2, \text{NMe}_2$; $\text{L}_2 = \text{Diphosphine, Cyclooctadiene}$)

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1. Preparation of Starting Materials

$\text{PhSO}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$ (**1**) was synthesized by conversion of NaPPh_2 with $\text{PhSO}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}^1$ which was prepared according to a conventional method from KSPH and $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{Br}^2$ followed by oxidation with acetic acid/ H_2O_2 .³ The amino compound $\text{PhSO}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{NMe}_2$ (**2**) was obtained by oxidation of $\text{PhSCH}_2\text{CH}_2\text{CH}_2\text{NMe}_2^4$ with acetic acid/ H_2O_2 . $[\{\text{Rh}(\text{cod})\}_2(\mu\text{-Cl})_2]$ (**5**) and $[\{\text{Rh}(\text{C}_2\text{H}_4)_2\}_2(\mu\text{-Cl})_2]$ (**6**) were obtained according to established procedures.⁵ The dimeric Rh(I) complexes of the type of $[\{\text{Rh}(\text{P}^{\wedge}\text{P})\}_2(\mu\text{-Cl})_2]$ ($\text{P}^{\wedge}\text{P} = \text{dmpe}$, **7a**; dppm , **7b**; dppe , **7c**; dppp , **7d**) were received by reaction of either $[\{\text{Rh}(\text{cod})\}_2(\mu\text{-Cl})_2]$ (**5**) or $[\{\text{Rh}(\text{C}_2\text{H}_4)_2\}_2(\mu\text{-Cl})_2]$ (**6**) with the respective diphosphine in toluene.^{6,7} To avoid a predominant formation of the cationic bis(diphosphine)rhodium(I) complexes $[\text{Rh}(\text{P}^{\wedge}\text{P})_2]^+\text{Cl}^-$ the addition of the diphosphine has to be performed slowly. Best results were obtained using complex **5** for synthesis of the dppe complex **7c** and using complex **6** for complexes **7a**, **7b** and **7d**. Whereas the dinuclear olefin complexes **5** and **6** are relatively air stable, the respective diphosphine complexes $[\{\text{Rh}(\text{P}^{\wedge}\text{P})\}_2(\mu\text{-Cl})_2]$ (**7a–d**) are highly air and moisture sensitive and have to be stored under argon.

2. NMR Spectroscopic Data of Starting Materials

2.1. $\text{PhSO}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$ (**1**)

^1H NMR (400 MHz, CDCl_3): δ 1.83 (m, 2H, $\text{CH}_2\text{CH}_2\text{CH}_2$), 2.11 (m, 2H, CH_2PPh_2), 3.20 (m, 2H, $\text{CH}_2\text{SO}_2\text{Ph}$), 7.28–7.35 (m, 10H, PPh_2), 7.48–7.52 (m, 2H, *m-H*, SO_2Ph), 7.56–7.62 (m, 1H, *p-H*, SO_2Ph), 7.81–7.82 (m, 2H, *o-H*, SO_2Ph). ^{13}C NMR (100 MHz, CDCl_3): δ 19.5 (d, $^2J(^{13}\text{C}, ^{31}\text{P}) = 18.8$ Hz, $\text{CH}_2\text{CH}_2\text{CH}_2$), 26.7 (d, $^1J(^{13}\text{C}, ^{31}\text{P}) = 12.5$ Hz, CH_2PPh_2), 56.7 (d, $^3J(^{13}\text{C}, ^{31}\text{P}) = 13.6$ Hz, $\text{CH}_2\text{SO}_2\text{Ph}$), 127.9–138.9 (m, C_{Ar}). ^{31}P NMR (81 MHz, CDCl_3): δ –16.4 (s, PPh_2).

2.2. PhSO₂CHCH₂CH₂NMe₂ (**3**)

¹H-NMR (400 MHz, CDCl₃): δ 1.77–1.85 (m, 2H, CH₂CH₂CH₂), 2.07 (s, 6H, N(CH₃)₂), 2.24 (m, 2H, CH₂NMe₂), 3.09–3.13 (m, 2H, CH₂SO₂Ph), 7.48–7.52 (m, 1H, *m*-H, SO₂Ph), 7.56–7.61 (m, 1H, *p*-H, SO₂Ph), 7.84–7.86 (m, 2H, *o*-H, SO₂Ph). ¹³C-NMR (125 MHz, CDCl₃): δ 20.7 (s, CH₂CH₂CH₂), 45.0 (s, N(CH₃)₂), 54.0 (s, CH₂SO₂Ph), 57.2 (s, CH₂NMe₂), 127.9 (s, *o*-C, SO₂Ph), 129.2 (s, *m*-C, SO₂Ph), 133.6 (s, *p*-C, SO₂Ph), 139.2 (s, *i*-C, SO₂Ph).

2.3. [{Rh(cod)}₂(μ-Cl)₂] (**5**)

¹H NMR (400 MHz, CDCl₃): δ 1.71/2.46 (m/m, 8H/8H, CH₂), 4.18 (s, br, 8H, CH). ¹³C NMR (100 MHz, CDCl₃): δ 31.0 (s, CH₂), 78.7 (d, ¹J(¹³C, ¹⁰³Rh) = 13.8 Hz, CH).

2.4. [{Rh(C₂H₄)₂}₂(μ-Cl)₂] (**6**).

¹H NMR (200 MHz, THF-*d*₈): δ 3.10 (s, br, 16H, CH₂). ¹³C NMR (50 MHz, THF-*d*₈): δ 60.7 (s, br, CH₂).

2.5. [{Rh(P[∧]P)}₂(μ-Cl)₂] (**7a–d**)

P[∧]P = dmpe (**7a**): ¹H NMR (400 MHz, CD₃NO₂): δ 1.41 (m, 24H, PCH₃), 1.51 ('d', *N* = 17.0 Hz, 8H, CH₂P). ¹³C NMR (50 MHz, CD₃NO₂): δ 16.2 ('t', *N* = 26.6 Hz, P(CH₃)₂), 29.2 ('t'd', *N* = 51.2/6.1 Hz, CH₂PMe₂). ³¹P NMR (81 MHz, CD₃NO₂): δ 55.8 (d, ¹J(³¹P, ¹⁰³Rh) = 196.6 Hz, PMe₂).

P[∧]P = dppm (**7b**): ¹H NMR (400 MHz, THF-*d*₈): δ 3.92 (td, ²J(¹H, ³¹P) = 10.6 Hz, ³J(¹H, ¹⁰³Rh) = 2.5 Hz, 4H, CH₂P), 7.15–7.30 (m, 24H, *m*-H/*p*-H PPh₂), 7.91–8.02 (m, 16H, *o*-H PPh₂). ¹³C NMR (100 MHz, THF-*d*₈): 51.2 (t, ¹J(¹³C, ³¹P) = 23.7 Hz, CH₂P), 131.3 ('t', *N* = 10.0 Hz, *o*-C, PPh₂), 132.4 (s, *p*-C, PPh₂), 135.9 ('t', *N* = 11.8 Hz, *m*-C, PPh₂), 139.6 ('t', *N* = 38.4 Hz, *i*-C, PPh₂). ³¹P NMR (81 MHz, THF-*d*₈): δ -27.4 (d, ¹J(³¹P, ¹⁰³Rh) = 169.0 Hz, PPh₂).

P[∧]P = dppe (**7c**): ¹H NMR (400 MHz, THF-*d*₈): δ 2.00 ('d', *N* = 19.2 Hz, 8H, CH₂P), 7.19–7.43 (m, 24H, *m*-H/*p*-H PPh₂), 7.80–7.89 (m, 16H, *o*-H PPh₂). ¹³C NMR (100 MHz, THF-*d*₈): δ 28.9 ('t'd', *N* = 57.2/3.8 Hz, CH₂PPh₂), 128.3 (m, *m*-C, PPh₂), 129.6 (s, *p*-C, PPh₂), 134.1 (m, *o*-C, PPh₂), 137.6 (m, *i*-C, PPh₂). ³¹P NMR (81 MHz, THF-*d*₈): δ 73.6 (d, ¹J(³¹P, ¹⁰³Rh) = 197.8 Hz, PPh₂).

P[∧]P = dppp (**7d**): ¹H NMR (400 MHz, THF-*d*₈): δ 1.73 (s, br, 4H, CH₂CH₂CH₂), 2.14 (s, br, 8H, CH₂PPh₂), 7.00–7.10 (m, 16H, *m*-H, PPh₂), 7.13–7.18 (m, 8H, *p*-H PPh₂), 7.65–7.69 (m, 16H, *o*-H PPh₂). ¹³C NMR (100 MHz, THF-*d*₈): 20.7 (s, br, CH₂CH₂CH₂), 29.3 (t, ²J(¹³C, ³¹P) = 19.6 Hz, CH₂PPh), 127.8 (s, br, *m*-C, PPh₂), 129.0 (s, *p*-C, PPh₂), 134.4 (s, br, *m*-C, PPh₂),

138.3 (t', $N = 42.0$ Hz, *i*-C, PPh₂). ³¹P NMR (81 MHz, THF-*d*₈): δ 32.5 (d, $^1J(^{31}\text{P}, ^{103}\text{Rh}) = 212.8$ Hz, PPh₂).

References

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