# **Supporting Information**

# Highly Efficient Palladium Precatalysts of Homoscorpionate Bispyrazolyl Ligands for the More Challenging Suzuki–Miyaura Cross–

# **Coupling of Aryl Chlorides**

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**Figure S1**. ORTEP of **2** shown with 50 % probability ellipsoids. Selected bond lengths (Å) and angles (°): Pd1-N1 2.0448(18), Pd1-N5 2.0459(18), Pd1-Cl1 2.2810(7), Pd1-Cl2 2.2824(8), Pd1-H6A 2.8436, Pd1-H7B 2.8463, N1-N2 1.366(2), N1-Pd1-N5 92.23(7), N1-Pd1-Cl1 88.42(5), N5-Pd1-Cl1 178.84(6), N1-Pd1-Cl2 178.65(5), N5-Pd1-Cl2 88.01(5).



Figure S2. ORTEP of 3 shown with 50 % probability ellipsoids. Selected bond lengths (Å) and angles (°): Pd1-N1 2.0217(15), Pd1-Cl1 2.2866(5), Pd1-H6A 2.8458, N1-N2 1.362(2), N1-Pd1-N1 90.76(9), N1-Pd1-Cl1 177.99(4), Cl1-Pd1-Cl1 92.76(2).



Figure S3. Detailed orbital interaction diagram showing major contributions to the pyrazole-palladium bond in 1.



Figure S4. Detailed orbital interaction diagram showing major contributions to the pyrazole-palladium bond in 2.



**Figure S5**. Simplified orbital interaction diagram showing major contributions to the pyrazole—palladium bonding orbital HOMO-27 in **2**.



Figure S6. Detailed orbital interaction diagram showing major contributions to the pyrazole-palladium bond in 3.



**Figure S7**. Simplified orbital interaction diagram showing major contributions to the pyrazole—palladium bonding orbital HOMO-26 in **3**.



Figure S8. Detailed orbital interaction diagram showing major contributions to the pyrazole-palladium bond in 4.



**Figure S9**. Simplified orbital interaction diagram showing major contributions to the pyrazole—palladium bonding orbital HOMO-71 in **4**.



**Figure S10**. Temperature dependence of the coupling of p-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>Cl with PhB(OH)<sub>2</sub> at 0.5 mol % loading of **3**.

compound	1	2	3	4
lattice	Orthorhombic	Triclinic	Orthorhombic	Triclinic
formula	$C_{22}H_{30}Cl_2N_6Pd$	$C_{25}H_{35}Cl_2N_7Pd$	$C_{19}H_{24}Cl_2N_5Pd$	$C_{44}H_{61}Cl_4N_{11}Pd_2$
formula weight	555.82	610.90	499.73	1098.64
Space group	P 21 21 21	P-1	Pcmn <sup>[a]</sup>	P-1
a/Å	11.6343(2)	10.688(4)	8.1966(2)	12.8252(1)
b/Å	12.5462(3)	12.313(2)	13.6939(3)	12.8252(1)
c/Å	16.8066(4)	12.6717(12)	18.0046(4)	17.6659(2)
$lpha/^{\circ}$	90	110.583(14)	90	98.513(9)
β/°	90	105.572(19)	90	111.107(8)
$\gamma/^{\circ}$	90	101.64(2)	90	107.462(8)
$V/Å^3$	2453.20(9)	1420.4(6)	2020.90(8)	2476.5(3)
Ζ	4	2	4	2
temperature (K)	123(2)	123(2)	123(2)	123(2)
radiation ( $\lambda$ ,Å)	0.71073	0.71073	0.71073	0.71073
$\rho$ (calcd.), g cm <sup>-3</sup>	1.505	1.428	1.642	1.473
$\mu$ (Mo K $\alpha$ ), mm <sup>-1</sup>	0.995	0.868	1.197	0.985
$\theta$ max, deg.	32.0906	32.4264	32.4887	32.2149
No. of data	8066	9378	3588	13369
No. of parameters	259	325	138	559
Flack Parameter	-0.02(3)			
<b>R</b> <sub>1</sub>	0.0426	0.0350	0.0275	0.0549
$wR_2$	0.0871	0.0562	0.0652	0.1277
GOF	1.077	0.823	1.032	0.956

 Table S1. X-ray crystallographic data for palladium 1–4 complexes.

<sup>[a]</sup> The reported space group Pcmn is a non-standard setting of Pmna.

The density functional theory calculations were performed for the palladium 1-4 complexes by using the GAUSSIAN  $03^{[1]}$  suite of quantum chemical programs.

 Table S2.
 B3LYP/SDD, 6-31G(d) optimized coordinates of 1.

Energy = -2100.5472514 hartree/particle.

Ν	-0.97522	-1.50413	0.487914
Ν	0.225668	-1.92987	0.006342
Ν	1.499391	0.000214	-0.86528
Ν	0.217392	1.930913	-0.00508
Ν	-0.9811	1.501609	0.478973
С	-2.60241	-2.17716	2.230229
Н	-2.66508	-1.21702	2.754865
Н	-2.70832	-2.97702	2.968535
Н	-3.4398	-2.22794	1.526607
С	-1.30576	-2.31691	1.497234
С	-0.28685	-3.2739	1.672803
Н	-0.26468	-4.07066	2.40251
С	0.671941	-3.00953	0.709007
С	1.959187	-3.7153	0.413553
Н	2.108452	-4.51366	1.144916
Н	2.821657	-3.04162	0.466995
Н	1.953342	-4.17662	-0.5818
С	0.848054	-1.25182	-1.14255
Н	1.570438	-1.96659	-1.54377
Н	0.051429	-1.10248	-1.87633
С	2.869049	0.002955	-0.42403
С	3.195502	0.01871	0.948358
С	2.15395	0.040772	2.045313
Н	2.220117	0.969347	2.625635
Н	2.314155	-0.78607	2.747304
Н	1.138629	-0.037	1.657325
С	4.552914	0.018355	1.310701
Η	4.814285	0.028853	2.366455
С	5.557025	0.004323	0.349602
Η	6.600872	0.004076	0.651888
С	5.222013	-0.00999	-1.00503
Η	6.006053	-0.02056	-1.75787
С	3.88522	-0.01107	-1.41081
С	3.545178	-0.0269	-2.88455
Η	4.456863	-0.01504	-3.48909
Н	2.941055	0.838956	-3.17984

Н	2.973477	-0.91855	-3.16848
С	0.844069	1.24861	-1.14903
Н	0.049166	1.092618	-1.88335
Н	1.564821	1.963533	-1.55307
С	1.937579	3.732521	0.386549
Н	2.804904	3.065841	0.449309
Н	2.079425	4.540203	1.109107
Н	1.93078	4.182487	-0.61396
С	0.654743	3.02081	0.687097
С	-0.30707	3.287717	1.647349
Н	-0.29158	4.091696	2.369267
С	-1.31898	2.321834	1.479581
С	-2.61853	2.183422	2.207463
Н	-3.45409	2.289293	1.507456
Н	-2.70322	2.950403	2.982503
Н	-2.7083	1.201035	2.683965
Pd	-2.09719	-0.00616	-0.43854
Cl	-3.30146	-1.68749	-1.48678
Cl	-3.30666	1.665138	-1.49717

 Table S3.
 B3LYP/SDD, 6-31G(d) optimized coordinates of 2.

Energy = -2139.8649961 hartree/particle.

Ν	-1.1728	1.500606	0.485306
N	0.006876	1.932972	-0.04049
N	1.257412	0.001974	-0.94241
Ν	0.009754	-1.93179	-0.04266
Ν	-1.17311	-1.50654	0.482047
С	-2.74687	2.175065	2.274622
Н	-3.60688	2.306057	1.609224
Н	-2.79367	2.924805	3.06953
Н	-2.8304	1.182631	2.730769
С	-1.47407	2.316265	1.50141
С	-0.45663	3.281154	1.637097
Н	-0.41463	4.081314	2.362183
С	0.469757	3.018713	0.641415
С	1.741706	3.731073	0.298709
Η	2.608439	3.061023	0.31776
Η	1.914997	4.528608	1.025686
Н	1.69572	4.194333	-0.69469
С	0.596457	1.251541	-1.20462
Η	-0.22033	1.098841	-1.91509
Η	1.305511	1.966508	-1.62877
С	0.596835	-1.24752	-1.20631

Н	1.305762	-1.96107	-1.63309
Η	-0.22116	-1.0939	-1.91519
С	1.752394	-3.72295	0.294372
Н	1.707016	-4.18739	-0.69852
Н	1.930272	-4.51899	1.021881
Н	2.616099	-3.049	0.311529
С	0.477728	-3.01602	0.638253
С	-0.44798	-3.28435	1.632803
Η	-0.40271	-4.0855	2.356607
С	-1.47024	-2.32446	1.497743
С	-2.73825	-2.18486	2.279506
Η	-2.75785	-1.24812	2.84849
Н	-2.83996	-3.01355	2.985977
Н	-3.59969	-2.18495	1.603904
С	2.633978	0.001814	-0.5237
С	3.638114	0.001645	-1.52391
С	3.276872	0.001915	-2.99268
Н	4.179303	0.001982	-3.61108
Н	2.684188	0.880827	-3.27286
Η	2.684143	-0.87685	-3.27323
С	4.978242	0.001098	-1.13922
Н	5.746964	0.000785	-1.90954
С	5.35972	0.000951	0.210227
С	6.82054	0.001044	0.596685
Н	6.947294	-0.00598	1.683598
Н	7.336532	0.88632	0.20474
Н	7.339919	-0.87676	0.192726
С	4.354047	0.00122	1.174526
Η	4.629172	0.001027	2.227699
С	2.989509	0.001527	0.83893
С	1.973979	0.000624	1.960263
Η	0.947768	0.007448	1.593877
Η	2.107808	0.877694	2.60485
Η	2.098929	-0.88461	2.595499
Pd	-2.32588	-0.00144	-0.39485
Cl	-3.57375	1.677318	-1.39603
Cl	-3.57686	-1.67501	-1.4005

 Table S4.
 B3LYP/SDD, 6-31G(d) optimized coordinates of 3.

Energy = -2061.2262513 hartree/particle.

Ν	-0.73694	-1.49678	0.356871
Ν	0.249352	-1.90526	-0.49061
Ν	1.136738	-1.3E-05	-1.80625

Ν	0.249432	1.905201	-0.49053
Ν	-0.73661	1.496478	0.357133
С	-1.69004	-2.18164	2.539121
Η	-1.55301	-1.24641	3.094163
Η	-1.57294	-3.0127	3.240444
Η	-2.71088	-2.19013	2.143785
С	-0.70436	-2.30686	1.420387
С	0.33248	-3.24583	1.254435
Η	0.608301	-4.03502	1.939116
С	0.918926	-2.97329	0.03033
С	2.061967	-3.65261	-0.65655
Η	2.389738	-4.50254	-0.05274
Η	2.920124	-2.98194	-0.78138
Η	1.781674	-4.0383	-1.64423
С	0.422993	-1.24547	-1.79661
Н	0.946438	-1.97142	-2.42517
Н	-0.58507	-1.09986	-2.19363
С	2.584765	-1.6E-05	-1.93756
С	0.423007	1.245455	-1.79655
Н	-0.58506	1.099864	-2.19355
Н	0.946439	1.971426	-2.4251
С	2.061789	3.652808	-0.65664
Н	2.920082	2.982293	-0.78141
Н	2.389389	4.502882	-0.05294
Н	1.781401	4.038318	-1.64437
С	0.918912	2.973316	0.030343
С	0.332641	3.245698	1.254572
Н	0.608452	4.034908	1.939238
С	-0.70398	2.306526	1.420675
С	-1.68944	2.181084	2.539579
Н	-2.71037	2.189953	2.144474
Н	-1.57201	3.011851	3.241201
Н	-1.5525	1.245611	3.094216
С	3.431077	-3.1E-05	-0.66325
С	4.828623	-4.5E-05	-0.79635
С	2.878039	-3.3E-05	0.620957
С	5.653887	-6.2E-05	0.326838
С	3.705405	-5.2E-05	1.748794
С	5.092289	-6.7E-05	1.607206
Н	5.273774	-4.2E-05	-1.79019
Н	1.800256	-1.4E-05	0.742871
Н	6.733731	-7.6E-05	0.204045
Н	3.258288	-5.5E-05	2.7393
Н	5.733064	-8.3E-05	2.484674
Н	2.866567	-0.87394	-2.54275
Н	2.866581	0.873902	-2.54275

Pd	-2.10586	0.000035	-0.13297
Cl	-3.60023	-1.67545	-0.71327
Cl	-3.59982	1.67599	-0.71295

Table S5. B3LYP/SDD, 6-31G(d) optimized coordinates of 4.

Energy = -4279.7339465 hartree/particle.

Ν	-3.53407	1.399762	1.410156
Ν	-2.29914	1.898826	1.69477
Ν	-0.44771	2.980823	-0.55666
Ν	3.3453	2.875467	-0.90876
Ν	3.999458	2.103403	0.002489
Ν	3.219354	-1.70766	1.340263
Ν	2.030958	-2.35485	1.505082
Ν	-0.03864	-3.47182	-0.76455
Ν	-3.75984	-2.86546	-1.3639
Ν	-4.49599	-2.10624	-0.50626
С	-5.9054	1.95913	1.889528
Η	-6.22043	1.879565	0.843926
Η	-6.45713	2.776804	2.362246
Н	-6.17512	1.025521	2.396557
С	-4.43383	2.213286	1.980627
С	-3.75659	3.253549	2.63813
Η	-4.20105	4.061721	3.200803
С	-2.40114	3.029387	2.440012
С	-1.21542	3.805627	2.919646
Η	-0.5246	3.180838	3.497771
Н	-1.55314	4.620346	3.565523
Н	-0.65549	4.237493	2.083811
С	-1.1043	1.268018	1.15
Н	-1.23709	0.191801	1.277151
Η	-0.25544	1.565772	1.76927
С	-0.87269	1.600456	-0.33309
Η	-0.14768	0.866636	-0.72523
Η	-1.80882	1.430938	-0.87257
С	-0.9642	3.573386	-1.79851
Η	-2.01768	3.27915	-1.87206
Η	-0.47208	3.174541	-2.70488
С	-0.8702	5.088248	-1.81273
С	-1.44092	5.843302	-0.77745
Η	-1.93388	5.326469	0.041303
С	-1.38282	7.23608	-0.79815
Н	-1.83289	7.807468	0.00964
С	-0.75462	7.899272	-1.85696

Η	-0.71147	8.985027	-1.87295
С	-0.18863	7.159035	-2.8948
Η	0.296685	7.66496	-3.72548
С	-0.24637	5.761988	-2.86889
Η	0.188842	5.189063	-3.68548
С	0.967993	3.222894	-0.29264
Н	1.216687	2.845053	0.704378
Η	1.134949	4.305499	-0.27377
С	1.969462	2.588414	-1.29565
Η	1.873455	1.501991	-1.34477
Η	1.828091	2.980797	-2.30569
С	3.625982	5.02239	-2.18401
Η	3.54253	4.608921	-3.19708
Н	4.359321	5.83221	-2.2157
Н	2.654897	5.457676	-1.9228
С	4.076444	3.984465	-1.20482
С	5.244902	3.902048	-0.46387
Н	6.061614	4.609359	-0.46598
С	5.161793	2.711189	0.277188
С	6.157668	2.122981	1.226308
Н	5.68448	1.883285	2.184193
Н	6.971305	2.832232	1.403972
Н	6.587974	1.199906	0.821633
С	5.607965	-2.06949	1.9132
Н	5.950058	-1.79913	0.909003
Н	6.224917	-2.89269	2.285193
Н	5.761382	-1.20411	2.568445
С	4.168748	-2.47822	1.889454
С	3.573547	-3.63936	2.409393
Н	4.072914	-4.45256	2.915769
С	2.215244	-3.53607	2.149434
Č	1.091373	-4.46761	2.473496
Н	0.37175	-4.00816	3.162571
Н	1.4891	-5.36597	2.952461
Н	0.55284	-4.76036	1.566988
С	0.810376	-1.80731	0.928466
H	0.891078	-0.72518	1.037984
Н	-0.03024	-2.1347	1.544879
C	0.61267	-2.17546	-0.55367
H	0.05299	-1 3551	-1 03576
Н	1 59717	-2 19183	-1 02886
C	0 381374	-4 1345	-2.0126
Ĥ	0.364083	-3.44576	-2.87577
Н	-0.35762	-4.91784	-2.22163
C	1 75451	-4 77095	-1 90687
č	2 864825	-4 21774	-2 55226
$\sim$	2.001023	1. <u>~</u> 1//T	2.55220

Н	2.752011	-3.30867	-3.13754
С	4.125509	-4.81131	-2.43892
Η	4.978429	-4.35895	-2.9367
С	4.285169	-5.97293	-1.68455
Η	5.263633	-6.43812	-1.59834
С	3.180351	-6.53815	-1.0393
Η	3.296955	-7.44701	-0.45433
С	1.927268	-5.93908	-1.14959
Н	1.069555	-6.37961	-0.64587
С	-1.4837	-3.46782	-0.5454
Η	-1.70388	-3.09313	0.459337
Η	-1.83145	-4.5078	-0.57163
С	-2.33107	-2.64572	-1.55678
Η	-2.15993	-1.57181	-1.45798
Η	-2.09583	-2.9259	-2.58758
С	-3.99339	-4.87952	-2.85021
Η	-3.66977	-4.40259	-3.78367
Η	-4.78398	-5.59224	-3.09769
Η	-3.14328	-5.44789	-2.45501
С	-4.51652	-3.87957	-1.86764
С	-5.77986	-3.74972	-1.31402
Η	-6.6364	-4.37994	-1.50478
С	-5.72904	-2.62787	-0.46935
С	-6.81906	-2.02149	0.356764
Н	-6.51782	-1.94731	1.407116
Η	-7.72257	-2.63501	0.295793
Η	-7.05967	-1.01327	0.001893
Pd	-3.91618	-0.36417	0.422129
Pd	3.486213	0.206444	0.621441
Cl	-3.18813	-1.59671	2.29814
Cl	-4.54475	0.838868	-1.50261
Cl	2.584018	1.095949	2.615575
Cl	4.253979	-0.6393	-1.43335

compound/specie	N <sub>pyrazole</sub>	Pd	compound/specie	N <sub>pyrazole</sub>	Pd
CI Pd CI		0.654	CI Pd CI		0.654
	-0.296 -0.296			-0.296 -0.296	
	-0.319 -0.319	0.621		-0.319 -0.319	0.622

 Table S6.
 Natural charge analyses of the palladium 1 and 2 complexes.

compound/specie	N <sub>pyrazole</sub>	Pd	compound/specie	N <sub>pyrazole</sub>	Pd
CI Pd CI		0.654	CI-Pa <sup>.CI</sup> CI-Pa <sup>.CI</sup>		0.975 0.975
	-0.297 -0.297		$Ph$ $N^{-2}N^{-2}N$ $N^{-2}N$	-0.319 -0.313 -0.320 -0.309	
Ph N N N N N N N N N N N N N N N N N N N	-0.319 -0.319	0.621	$\begin{array}{c} Ph \\ \hline N \\ Ph \\ \hline N \\ Ph \\ \hline N \\ Ph \\ \hline Cl \\ Pd \\ Pd \\ Cl \\ Pd \\ Cl \\ Pd \\ Pd \\ Cl \\ Pd \\ Pd \\ Pd \\ Cl \\ Pd \\ P$	-0.305 -0.300 -0.306 -0.296	0.651 0.651

# Table S7. Natural charge analyses of the palladium 3 and 4 complexes.

compound/specie	N <sub>pyrazole</sub>	Pd	compound/specie	N <sub>pyrazole</sub>	Pd
CI Pd CI		0.269	CI Pd CI		0.269
	-0.285 -0.286			-0.286 -0.285	
	-0.319 -0.319	0.022		-0.319 -0.319	0.023

# Table S8. Mulliken charge analyses of the palladium 1 and 2 complexes.

compound/specie	N <sub>pyrazole</sub>	Pd	compound/specie	N <sub>pyrazole</sub>	Pd
CI Pd CI		0.269	CI-Pd <sup>-CI</sup> CI-Pd <sup>-CI</sup>		0.448 0.448
	-0.286 -0.286		Ph $Ph$ $Ph$ $Ph$ $Ph$ $Ph$ $N$	-0.315 -0.309 -0.315 -0.307	
$\mathbf{B}_{\mathbf{C}_{1}}^{\mathbf{Ph}} \mathbf{N}_{\mathbf{N}_{1}}^{\mathbf{N}} \mathbf{N}_{\mathbf{N}_{2}}^{\mathbf{N}}$	-0.320 -0.320	0.027	$\begin{array}{c} Ph \\ Ph $	-0.306 -0.298 -0.306 -0.295	-0.006 -0.002

# Table S9. Mulliken charge analyses of the palladium 3 and 4 complexes.

compound/specie	5 <i>s</i>	4d	5p	6р	6d	7p	compound/specie	5 <i>s</i>	4d	5p	6р	6d	7p
Pd <sup>2+</sup>		8.00					Pd <sup>2+</sup>		8.00				
CI Pd CI	0.28	9.03	0.03				CI Pd CI	0.28	9.03	0.03			
	0.42	8.92		0.01	0.01	0.01		0.42	8.92		0.01	0.01	0.01

**Table S10.** Electronic configuration of Pd in the palladium 1 and 2 complexes.

compound/specie	5s	4d	5p	6р	6d	7p	compound/specie	5 <i>s</i>	4d	5p	5d	6р	6d
Pd <sup>2+</sup>		8.00					$Pd^{2+}$		8.00				
CI Pd CI	0.28	9.03	0.03				CI-Pd <sup>-CI</sup> CI-Pd <sup>-CI</sup>	0.06 0.06	8.91 8.91	0.04 0.04			
Ph N N CI Pd CI	0.42	8.92		0.01	0.01	0.01	$\begin{array}{c} Ph \\ & & \\ & & \\ Ph \\ & & \\ Ph \\ & & \\ Ph \\ & & \\ Cl - Pd - Cl \\ & & \\ Cl - Pd - Cl \\ & & \\ Pd - Cl \\ & & \\ Ph \\ & & \\ Ph \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ $	0.41 0.41	8.91 8.90		0.01	0.01 0.01	0.01 0.01

**Table S11.** Electronic configuration of Pd in the palladium 3 and 4 complexes.

compound	d(Pd-N <sub>pyrazole</sub> ) (Å)	D <sub>e</sub> (Pd-N <sub>pyrazole</sub> ) (kcal/mol)
	2.09	41.98
	2.09	42.05
Ph N N CI Pd CI 3	2.09	41.87
$\begin{array}{c} Ph \\ N^{-2}N^{-2}N \\ N^{-2}N^{-2}N \\ CI^{-Pd} \\ $	2.06	50.20

# Table S12. Bond distance & bond energy of Pd-N<sub>pyrazole</sub> in 1-4.

					yield <sup>[b]</sup>	
entry	Reagent <sup>[a]</sup>	reagent <sup>[a]</sup>	cross-coupled product	PdCl <sub>2</sub>	(COD)PdCl <sub>2</sub>	Hg(0)
						drop
1	O <sub>2</sub> N-CI	(HO) <sub>2</sub> B	0 <sub>2</sub> N-	>99	91	>99
2	NC-CI	(HO) <sub>2</sub> B		77	51	>99
3	PhOC-CI	(HO) <sub>2</sub> B	PhOC-	78	86	>99
4	ОНССІ	(HO) <sub>2</sub> B	ОНС	26	39	>99
	СНО		СНО		49	
5	СІ	(HO) <sub>2</sub> B		67		>99
6	F <sub>3</sub> C-CI	(HO) <sub>2</sub> B	F <sub>3</sub> C	62	24	>99
7		(HO) <sub>2</sub> B	H <sub>3</sub> COC	15	15	69
8	CI CI	(HO) <sub>2</sub> B		41	12	70
9	CI	(HO) <sub>2</sub> B		11	10	29
10	— Сі	(HO) <sub>2</sub> B		4	2	10
11	CI	(HO) <sub>2</sub> B		14	13	15
12	CI	(HO) <sub>2</sub> B		32	49	>99
13	CI	(HO) <sub>2</sub> B		0	0	15

Table S13. Selected results for Suzuki–Miyaura cross-coupling reaction of chlorides catalyzed by PdCl<sub>2</sub> and by 3 under Hg(0) conditions.

<sup>[a]</sup> Reaction conditions: 1.00 mmol of aryl chloride, 1.20 mmol of boronic acid, 1.50 mmol of Cs<sub>2</sub>CO<sub>3</sub>, 1.50 mmol of TBAB, 2 mol % of PdCl<sub>2</sub> or (COD)PdCl<sub>2</sub> or **3** in 8 mL of DMF:H<sub>2</sub>O (9:1), at 120 °C for 5 hours. <sup>[b]</sup> The yields (%) were determined by GC using diethylene glycol di-n-butyl ether as an internal standard.

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