

Supplementary Information

Synthesis and Reactivity of Molybdenum Imido Alkylidene bis-Pyrazolide Complexes.

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Anne Lesage,^b Lyndon Emsley,^b Richard R. Schrock^{c,*}**

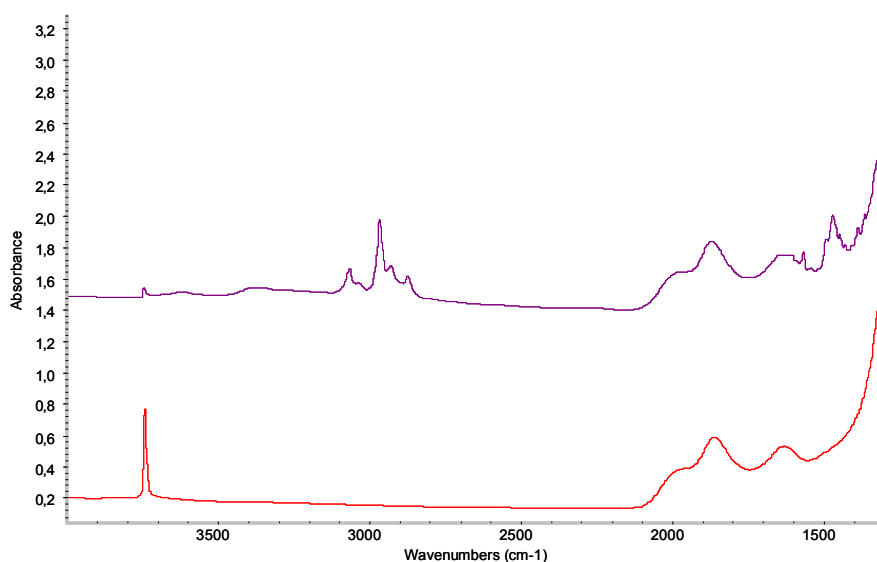
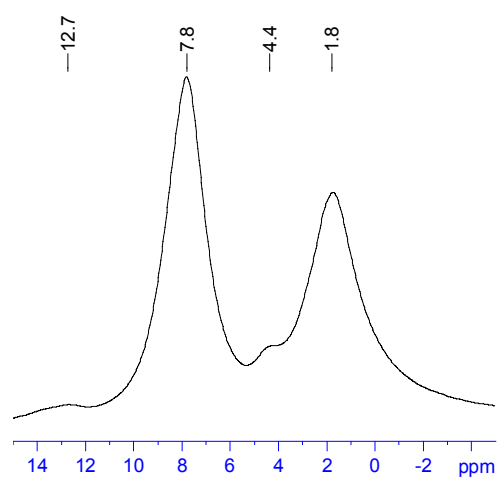


Figure S1. Grafting of $[\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dppz})_2]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) **1-dppz-Ph** on a silica partially dehydroxylated at 700 °C, $\text{SiO}_{2-(700)}$, monitored by IR spectroscopy. $\text{SiO}_{2-(700)}$ pellet (30 mg) (bottom spectrum) After impregnation of $[\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dppz})_2]$ (7 mg) in benzene (10 mL) (3 h, 25 °C), followed by three washings (30 min, 25 °C) in benzene and a drying step under vacuum (1.34 Pa, 1 h, 25 °C) (top spectrum).

a)



b)

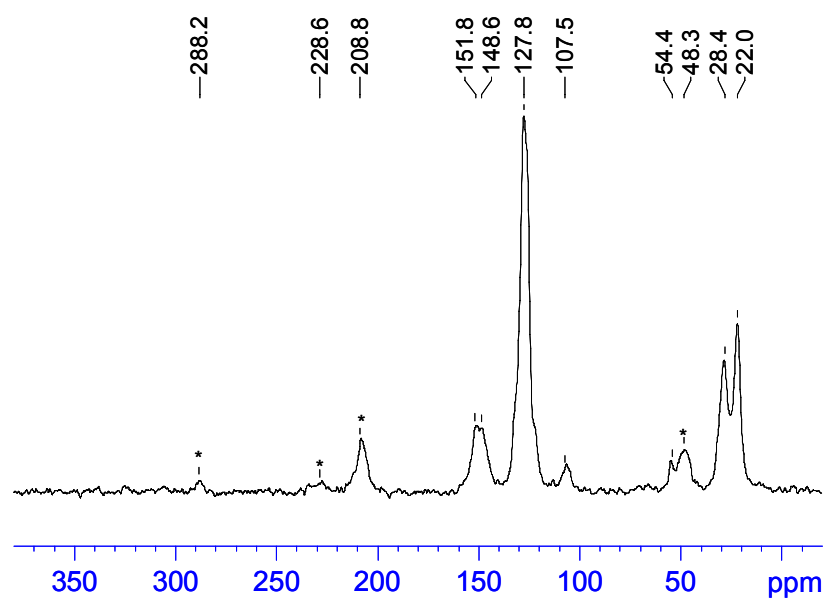
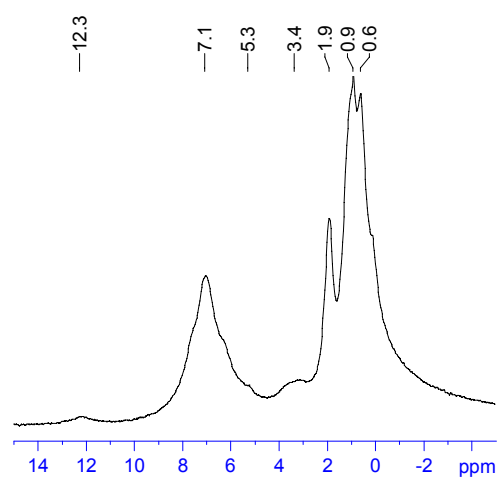


Figure S2. a) One-dimensional solid state proton single pulse spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dppz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1-dppz-Ph/SiO₂**) recorded under MAS frequency of 10 kHz. 8 scans were coadded at a recycle rate of 64 s. b) ¹³C CP solid state NMR spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dppz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1-dppz-Ph/SiO₂**) recorded under MAS frequency of 10 kHz. 38193 scans were coadded at a recycle rate of 2 s.

a)



b)

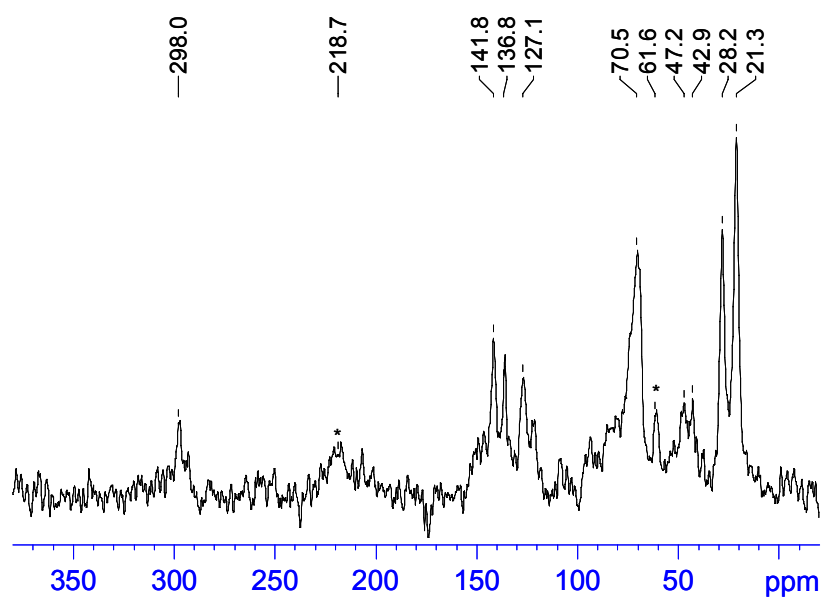


Figure S3. a) One-dimensional solid state proton single pulse spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_3)(\text{dtpz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1^{*}-dppz-Me/SiO₂**) recorded under MAS frequency of 10 kHz. 8 scans were coadded at a recycle rate of 4 s. b) ^{13}C CP solid state NMR spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_3)(\text{dtpz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1^{*}-dppz-Me/SiO₂**) recorded under MAS frequency of 10 kHz. 30720 scans were coadded at a recycle rate of 2 s.

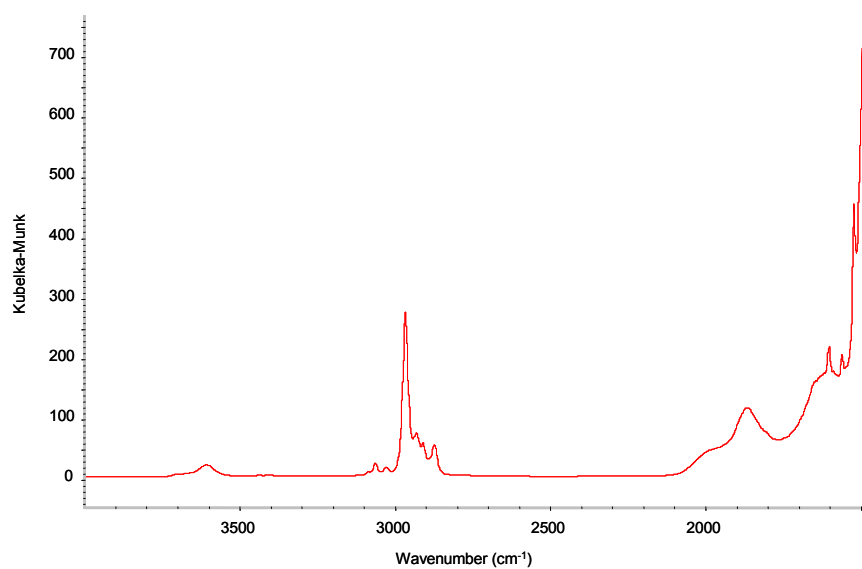
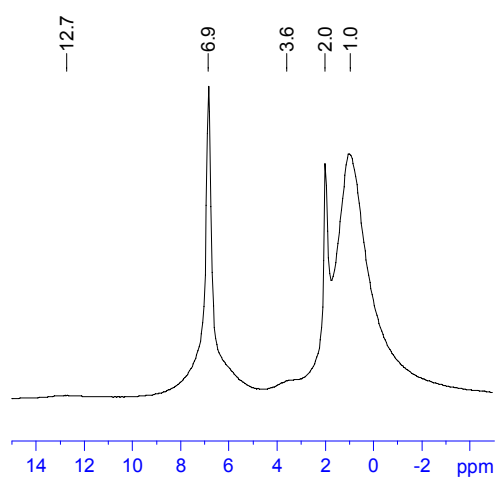


Figure S4. DRIFT spectroscopy of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dtpz})_2]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1-dtpz-Ph/SiO₂**).

a)



b)

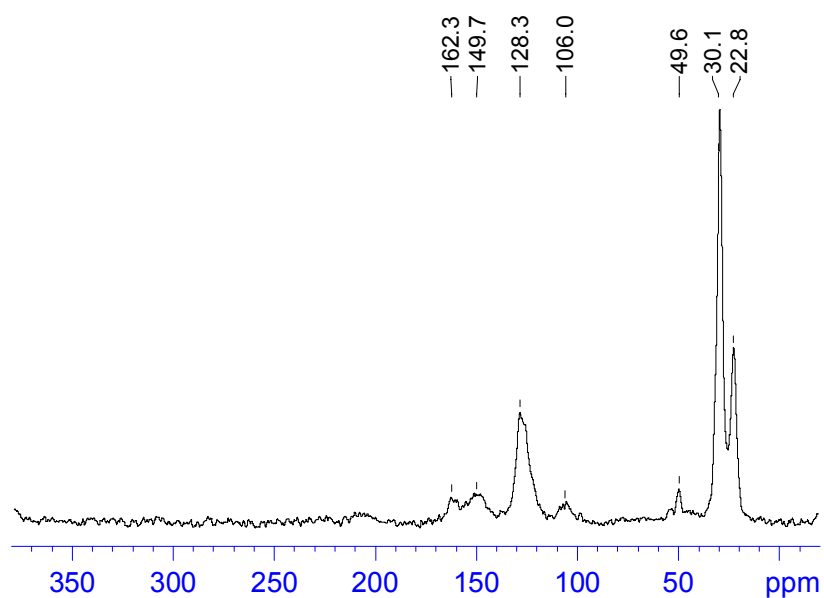


Figure S5. a) One-dimensional solid state proton single pulse spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dtpz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1-dtpz-Ph/SiO₂**) recorded under MAS frequency of 10 kHz. 8 scans were coadded at a recycle rate of 4 s. b) ¹³C CP solid state NMR spectrum of $[(\equiv\text{SiO})\text{Mo}(\equiv\text{NAr})(=\text{CHCMe}_2\text{Ph})(\text{dtpz})]$ ($\text{Ar} = 2,6\text{-}i\text{Pr}_2\text{C}_6\text{H}_3$) (**1-dtpz-Ph/SiO₂**) recorded under MAS frequency of 10 kHz. 30720 scans were coadded at a recycle rate of 2 s.

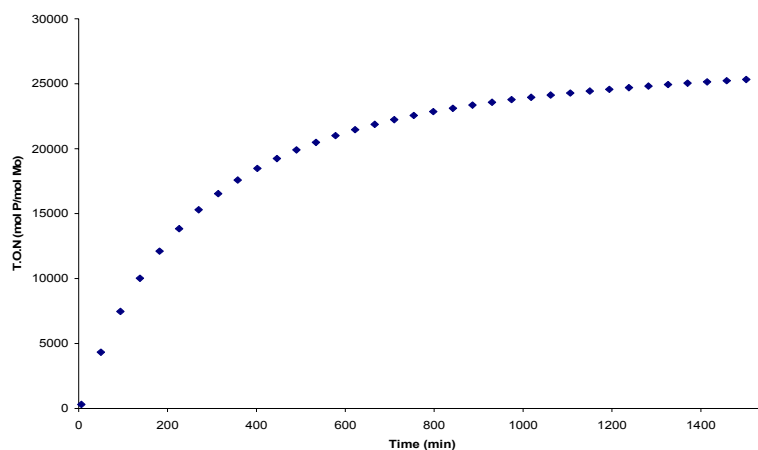


Figure S6. Activity in propene metathesis catalysed by **1-dppz-Ph/SiO₂** cumulative TON as a function of time. 17.4 μmol Mo was used, propene was flowed at $16 \text{ mol} \times \text{mol}(\text{Mo})^{-1} \times \text{s}^{-1}$ at a temperature of 30 °C. A initial $\text{TOF}_{(6\text{min})}$ 1.73 s^{-1} and a cumulative $\text{TON}_{(1550\text{min})}$ of 25400 were obtained. $\text{TOF}_{(1550 \text{ min})}$ 0.03 s^{-1} .

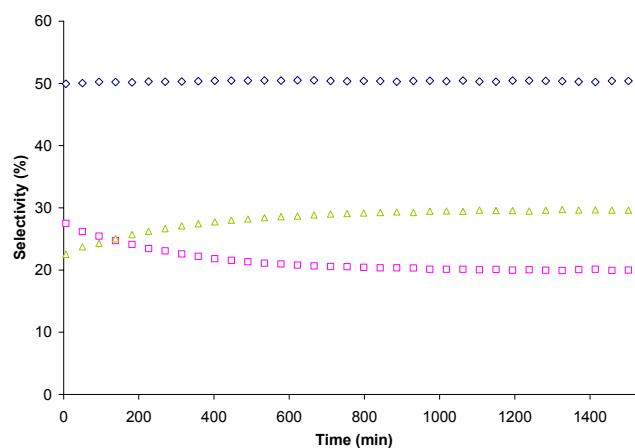


Figure S7. Selectivity in propene metathesis catalysed by **1-dppz-h/SiO₂**. Product selectivity as a function of time. Ethylene ($\diamond$), E-2-butene ($\square$), Z-2-butene (Δ). Selectivity in ethene and 2-butenes > 99.9%.

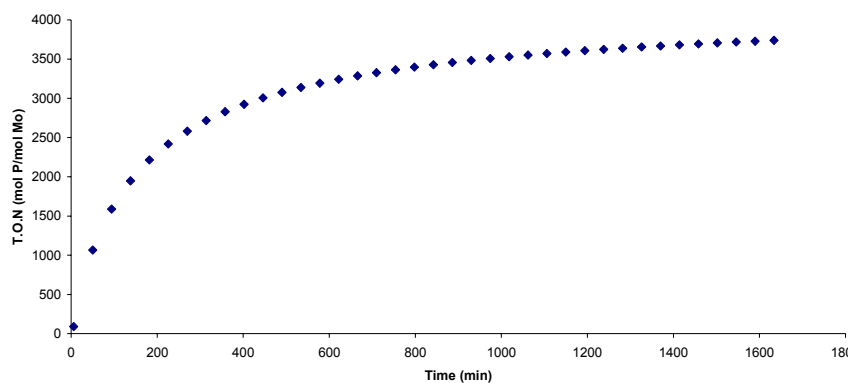


Figure S8. Activity in propene metathesis catalysed by **1-dtpz-Ph/SiO₂** cumulative TON as a function of time. 17.5 μmol Mo was used, propene was flowed at $16 \text{ mol} \times \text{mol}(\text{Mo})^{-1} \times \text{s}^{-1}$ at a temperature of 30 °C. A initial $\text{TOF}_{(6\text{min})}$ 1.72 s^{-1} and a cumulative $\text{TON}_{(1550\text{min})}$ of 25300 were obtained. $\text{TOF}_{(1550 \text{ min})}$ 0.03 s^{-1} .

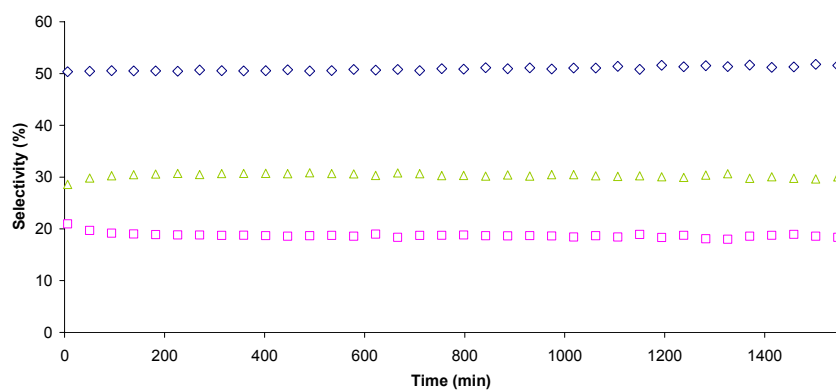


Figure S9. Selectivity in propene metathesis catalysed by **1-dtpz-Ph/SiO₂**. Product selectivity as a function of time. Ethylene ($\diamond$), E-2-butene ($\square$), Z-2-butene (Δ). Selectivity in ethene and 2-butenes > 99.9%.