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Gas–Phase Synthesis and Reactivity of Dimethylaurate

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Supporting Information

Complete citation for reference 19: Gaussian 03, Revision B.04, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, Gaussian, Inc., Wallingford CT, 2004.

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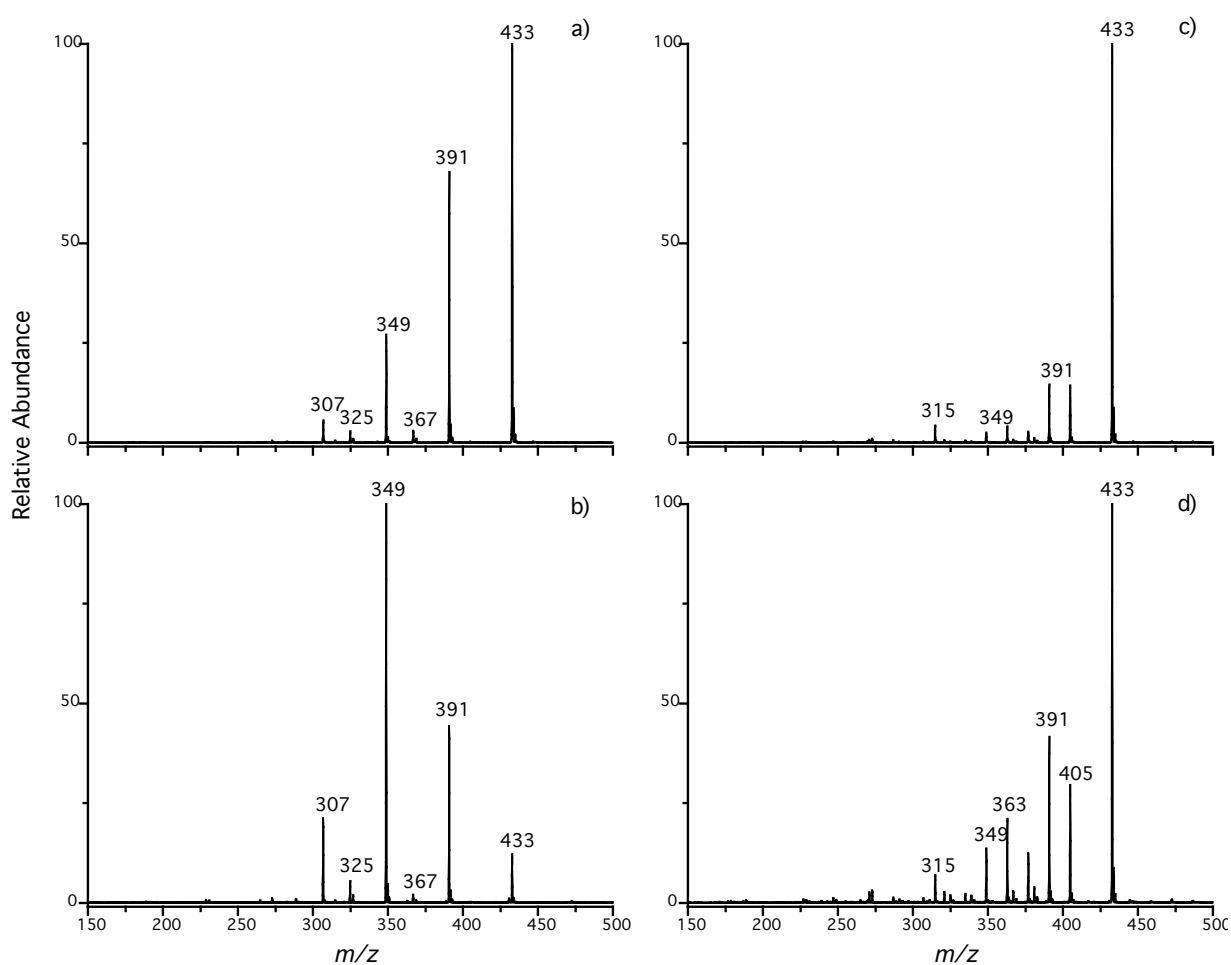


Figure S1: ESI/MS of gold(III) acetate (1mM) dissolved in water a) fresh sample; b) sample after 2 hours; and methanol:water (50:50 v/v) c) fresh sample; d) sample after 2 hours.

<i>m/z</i>	Assignment
307	$[\text{CH}_3\text{CO}_2\text{Au}(\text{OH})_3]^-$
349	$[(\text{CH}_3\text{CO}_2)_2\text{Au}(\text{OH})_2]^-$ or $[\text{CH}_3\text{CO}_2\text{Au}(\text{OCH}_3)_3]^-$
377	$[(\text{CH}_3\text{CO}_2)_2\text{Au}(\text{OCH}_3)_2]^-$
391	$[(\text{CH}_3\text{CO}_2)_3\text{AuOH}]^-$
405	$[(\text{CH}_3\text{CO}_2)_3\text{AuOCH}_3]^-$
433	$[(\text{CH}_3\text{CO}_2)_4\text{Au}]^-$

Table S1: Ions formed via ESI.

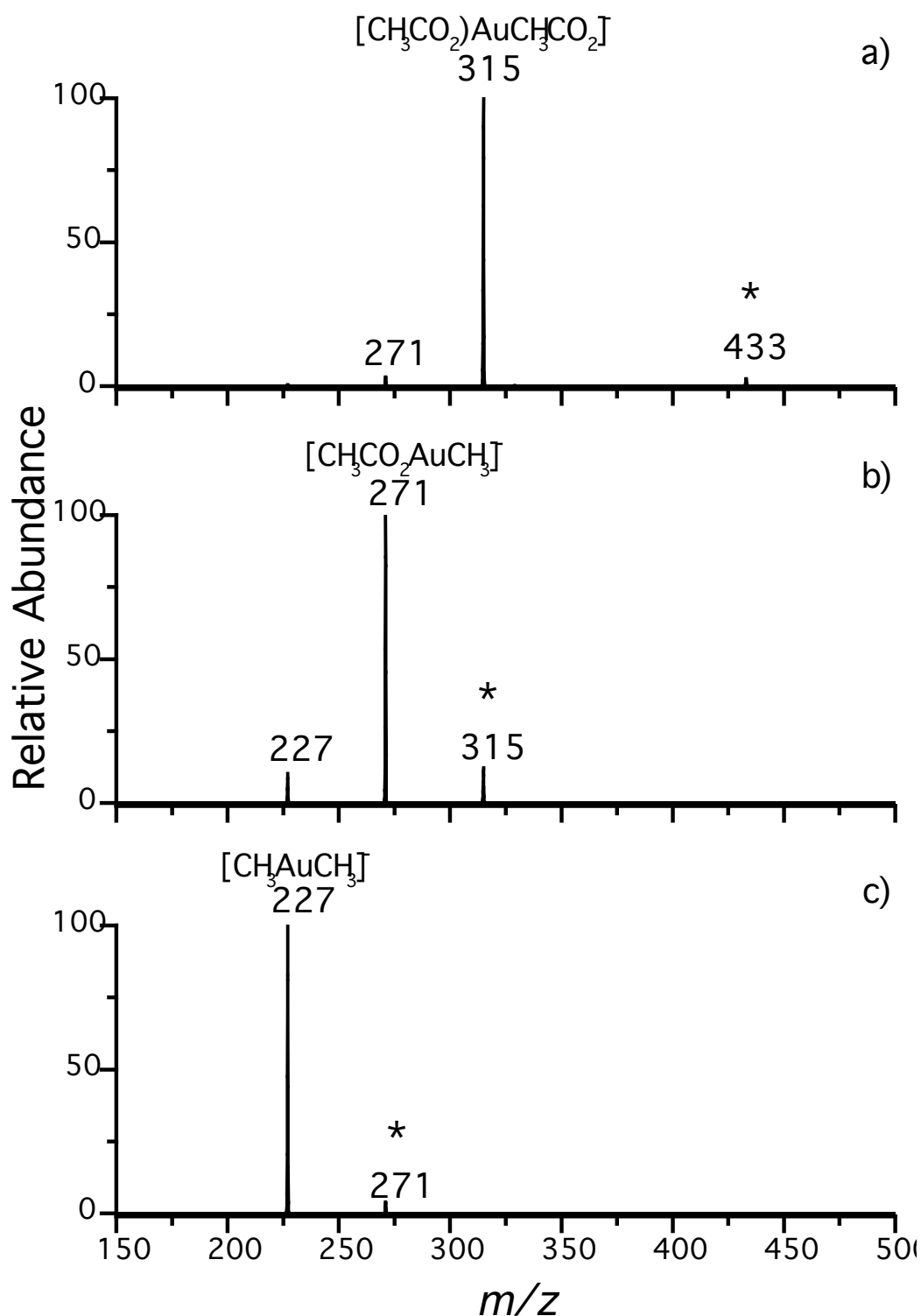
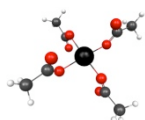


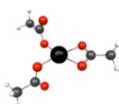
Figure S2: Mass spectra showing the formation of $[\text{CH}_3\text{AuCH}_3]^-$ via a series of collision induced dissociation experiments (CID): **(a)** MS² CID spectrum of $[(\text{CH}_3\text{CO}_2)_4\text{Au}]^-$, m/z 433; **(b)** MS³ CID spectrum of $[\text{CH}_3\text{CO}_2\text{AuO}_2\text{CCH}_3]^-$, m/z 315; **(c)** MS⁴ CID spectrum of $[\text{CH}_3\text{CO}_2\text{AuCH}_3]^-$, m/z 271. The mass selected precursor ion is marked with an * in each case. **4**



$[(\text{CH}_3\text{CO}_2)_4\text{Au}]^- \text{C}_{2v}$

Au	79.0	0.00027000	-0.00032900	0.00000000
C	6.0	-1.94446301	1.93390298	0.96314001
C	6.0	-1.93495703	-1.94420099	-0.96274000
C	6.0	1.93475997	1.94440103	-0.96262801
O	8.0	-1.64174104	1.62454295	2.10869908
O	8.0	-1.43754101	1.43592894	-0.14566900
O	8.0	1.43671000	1.43724799	0.14605001
O	8.0	1.62586200	1.64153802	-2.10825896
O	8.0	1.64193499	-1.62640095	2.10789108
O	8.0	1.43844700	-1.43613505	-0.14639100
O	8.0	-1.43570101	-1.43838203	0.14600800
O	8.0	-1.62656796	-1.64063895	-2.10832405
C	6.0	-2.99412990	-3.00165296	-0.65042102
H	1.0	-2.55955410	-3.80020809	-0.03879200
H	1.0	-3.81059790	-2.55445504	-0.07224800
H	1.0	-3.38387609	-3.41700792	-1.58300698
C	6.0	-3.00486898	2.99018407	0.65099603
H	1.0	-3.81621599	2.54434299	0.06460500
H	1.0	-3.40199900	3.39799404	1.58380198
H	1.0	-2.56886196	3.79415607	0.04756700
C	6.0	2.98966908	3.00605202	-0.65009302
H	1.0	2.53803205	3.82424688	-0.07770000
H	1.0	3.78617311	2.57594991	-0.03278900
H	1.0	3.40798092	3.39305997	-1.58249497
C	6.0	1.94546700	-1.93448901	0.96221799
C	6.0	3.00617504	-2.99028897	0.64952803
H	1.0	2.56886697	-3.79631090	0.04972600
H	1.0	3.81511211	-2.54538703	0.05915500
H	1.0	3.40649700	-3.39543104	1.58212197

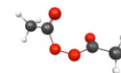
Sum of electronic and zero-point Energies = -1049.623871



$[(\text{CH}_3\text{CO}_2)_3\text{Au}]$

Au	79.0	0.21650700	-0.00405900	-0.00324000
C	6.0	-2.65466309	3.06007910	-0.44248101
H	1.0	-2.54190207	3.20748711	-1.51987398
H	1.0	-3.61198807	2.55490804	-0.26978299
H	1.0	-2.65777397	4.02008104	0.07791300
C	6.0	-2.71042705	-3.01139498	0.45929900
H	1.0	-2.59007907	-3.16372609	1.53518403
H	1.0	-3.65875912	-2.48612189	0.29724699
H	1.0	-2.73857999	-3.96985292	-0.06318600
C	6.0	4.20589304	-0.01785000	0.01827900
H	1.0	4.59609985	0.55126202	-0.82916403
H	1.0	4.53894520	0.46753499	0.94398302
H	1.0	4.58712006	-1.04186904	0.00693900
C	6.0	-1.53644204	2.19886994	0.11700100
C	6.0	2.71536708	-0.02336100	-0.00964400
C	6.0	-1.58033097	-2.17200398	-0.10951200
O	8.0	2.04231095	0.94924700	-0.50869000
O	8.0	2.03005695	-0.98944497	0.48531899
O	8.0	-0.98917103	2.38872695	1.19084299
O	8.0	-1.23894501	1.19416296	-0.71043903
O	8.0	-1.25395799	-1.17575002	0.71728301
*8.0	-1.04823101	-2.37007594	-1.18943202	

*Sum of electronic and zero-point Energies = -821.053037

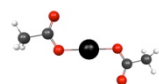


$[\text{CH}_3\text{CO}_2\text{MeO}_2\text{CH}_3]^-$

O	8.0	1.12844300	-1.17209101	-0.80468601
O	8.0	0.65623200	0.29733101	0.88501799
O	8.0	-0.65623099	-0.29735401	0.88501102
O	8.0	-1.12844205	1.17211199	-0.80466002
C	6.0	-2.81507611	-0.40178901	-0.02177500
H	1.0	-2.81130290	-1.21404600	-0.75765800
H	1.0	-3.02454710	-0.82642299	0.96284199
H	1.0	-3.58723402	0.31888899	-0.30008000
C	6.0	1.47317803	-0.28677699	-0.06953700
C	6.0	2.81507492	0.40178701	-0.02176600
H	1.0	2.81135702	1.21394897	-0.75775498
H	1.0	3.02448606	0.82655102	0.96280801
H	1.0	3.58724499	-0.31893101	-0.29993099

Sum of electronic and zero-point Energies = -456.779641

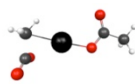
Figure S3: Calculated Cartesian coordinates and energies (Hartrees) for species relevant to the fragmentation of $[(\text{CH}_3\text{CO}_2)_4\text{Au}]^-$



[CH₃CO₂AuO₂CCH₃]⁻

C	6.0	-2.96340895	-0.16013899	-0.00089400
C	6.0	-4.30895805	0.57811701	0.00116900
O	8.0	-1.95624304	0.66709101	-0.00067400
O	8.0	-2.92190099	-1.39311898	-0.00109700
H	1.0	-4.38517189	1.21433794	0.89126199
H	1.0	-5.13524580	-0.13804300	-0.01012900
H	1.0	-4.37785387	1.23563004	-0.87378103
C	6.0	2.96340704	0.16014700	0.00001500
C	6.0	4.30895901	-0.57810700	0.00045700
O	8.0	2.92189002	1.39312696	-0.00093600
O	8.0	1.95624804	-0.66709000	0.00088200
H	1.0	5.13526392	0.13811900	0.00272900
H	1.0	4.38257313	-1.22322595	-0.88338202
H	1.0	4.38042784	-1.22681701	0.88179499
Au	79.0	0.00000100	-0.00000200	0.00002100

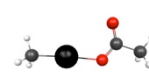
Sum of electronic and zero-point Energies = -592.784884



TS [CH₃CO₂AuO₂CCH₃]⁻ → [CH₃CO₂AuCH₃]⁻ + CO₂ (-338 cm⁻¹)

Au	79.0	0.11505100	0.09118800	-0.01150100
O	8.0	-1.76429403	-0.78206003	-0.09157700
C	6.0	-2.81187201	-0.00865000	0.00709100
O	8.0	-2.81764197	1.21767700	0.13290900
C	6.0	-4.11904907	-0.80582303	-0.04093400
H	1.0	-4.17137003	-1.48087704	0.82168800
H	1.0	-4.97597980	-0.12741300	-0.02617000
H	1.0	-4.15061188	-1.42817199	-0.94245398
O	8.0	3.10008597	-0.85848701	-1.01708198
C	6.0	2.70456910	-0.39400399	0.05620200
O	8.0	2.74215102	-0.75137401	1.24211299
C	6.0	2.15248609	1.24751997	-0.13708200
H	1.0	1.39704704	1.77924705	0.49133399
H	1.0	3.07279611	1.64844799	0.30190399
H	1.0	2.09985304	1.56463599	-1.18026805

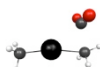
Sum of electronic and zero-point Energies = -592.718659



[CH₃CO₂AuCH₃]⁻

Au	79.0	-0.67999297	-0.07546600	0.00002700
O	8.0	1.40479100	-0.62586600	0.00076700
C	6.0	2.38917994	0.21059801	0.00038200
O	8.0	2.33777690	1.44981205	0.00013900
C	6.0	3.76233912	-0.49057701	-0.00057800
H	1.0	3.85434508	-1.13050401	-0.88739902
H	1.0	4.57253504	0.24510300	0.00573000
H	1.0	3.85043502	-1.14235103	0.87789899
C	6.0	-2.69337392	0.31615800	-0.00051400
H	1.0	-2.88921189	1.35727000	-0.29861900
H	1.0	-3.22941399	-0.34149900	-0.70194799
H	1.0	-3.12864304	0.16514800	0.99926001

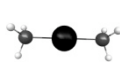
Sum of electronic and zero-point Energies = -404.197803



TS [CH₃CO₂AuCH₃]⁻ → [CH₃AuCH₃]⁻ + CO₂ (-341 cm⁻¹)

Au	79.0	0.50680703	0.09676400	-0.00000300
C	6.0	2.40405989	-0.72521198	0.00006000
O	8.0	-2.31946802	-0.76111501	1.15023899
C	6.0	-2.08299994	-0.43602800	0.00001300
O	8.0	-2.31934190	-0.76138502	-1.15016401
C	6.0	-1.42029297	1.44550800	-0.00010200
H	1.0	-1.05092800	1.93434095	-0.91175097
H	1.0	-2.50490189	1.60551906	-0.00034300
H	1.0	-1.05128896	1.93432105	0.91169697
H	1.0	2.55204201	-1.34820104	0.89324403
H	1.0	2.54895711	-1.35523498	-0.88868302
H	1.0	3.17428207	0.05925800	-0.00437000

Sum of electronic and zero-point Energies = -404.137166



[CH₃AuCH₃]⁻

Au	79.0	0.00000000	0.00010900	-0.00013700
C	6.0	-2.12720299	-0.00040400	0.00052500
H	1.0	-2.53713489	-0.36932200	0.95716602
H	1.0	-2.53817892	1.01211405	-0.15818900
H	1.0	-2.53784704	-0.64466500	-0.79670697
C	6.0	2.12720299	-0.00040500	0.00052500
H	1.0	2.53817797	1.01211500	-0.15818700
H	1.0	2.53713489	-0.36932299	0.95716500
H	1.0	2.53784704	-0.64466399	-0.79670900

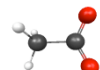
Sum of electronic and zero-point Energies = -215.585267



CO₂

C	6.0	0.00000000	0.00000000	0.00000000
O	8.0	0.00000000	0.00000000	1.16979396
O	8.0	0.00000000	0.00000000	-1.16979396

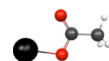
Sum of electronic and zero-point Energies = -188.575850



CH₃CO₂⁻

O	8.0	0.70457202	1.16242397	0.00002000
O	8.0	0.80539101	-1.11075401	0.00002000
C	6.0	0.20919301	0.00116600	-0.00009800
C	6.0	-1.35404205	-0.04939700	-0.00004100
H	1.0	-1.72823596	-1.08034205	-0.00285600
H	1.0	-1.74091601	0.47551799	0.88480002
H	1.0	-1.74146199	0.48085099	-0.88142800

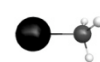
Sum of electronic and zero-point Energies = -228.487580



[CH₃CO₂Au]

C	6.0	1.95726800	0.13221800	-0.00034900
C	6.0	3.39722300	-0.36717600	-0.00000600
O	8.0	1.08345997	-0.85304701	-0.00025200
O	8.0	1.66684496	1.32613504	-0.00006700
H	1.0	3.57797289	-0.97818601	0.89024198
H	1.0	4.07762289	0.48575699	-0.01111200
H	1.0	3.57312489	-0.99909002	-0.87629801
Au	79.0	-0.82731801	-0.01118300	0.00002300

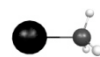
Sum of electronic and zero-point Energies = -364.190197



[CH₃Au]

Au	79.0	0.22006799	0.00000000	0.00000000
C	6.0	-1.82346404	-0.00000100	-0.00001000
H	1.0	-2.14818501	0.99234599	-0.32082400
H	1.0	-2.14819193	-0.77404600	-0.69889098
H	1.0	-2.14820290	-0.21829100	1.01976299

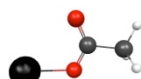
Sum of electronic and zero-point Energies = -175.647426



[CH₃Au]⁻

Au	79.0	-0.23313800	0.00000100	0.00000000
C	6.0	1.92640996	0.00000400	-0.00000100
H	1.0	2.28639007	-0.41409799	0.95212299
H	1.0	2.28643298	-0.61756700	-0.83462900
H	1.0	2.28664899	1.03154802	-0.11746100

Sum of electronic and zero-point Energies = -175.672986



[CH₃CO₂Au]⁻

C	6.0	-2.20323205	0.14356600	-0.00029600
C	6.0	-3.59156895	-0.53700000	0.00022500
O	8.0	-1.23603296	-0.70928401	-0.00048500
O	8.0	-2.13788295	1.38260198	-0.00006700
H	1.0	-3.69090295	-1.18486297	-0.87968600
H	1.0	-4.38802624	0.21389900	-0.00469800
H	1.0	-3.69391394	-1.17566299	0.88656700
Au	79.0	0.93079698	-0.01113100	0.00003400

Sum of electronic and zero-point Energies = -364.277241



Me.

C	6.0	0.00000400	-0.00000200	-0.00001800
H	1.0	1.07251894	-0.15388601	0.00003600
H	1.0	-0.40299401	1.00575805	0.00003600
H	1.0	-0.66954798	-0.85185802	0.00003600

Sum of electronic and zero-point Energies = -39.811971

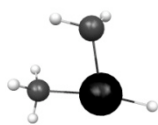


Me-

C	6.0	-0.00008700	-0.00000200	0.12094500
H	1.0	0.51700997	0.90736097	-0.24182799
H	1.0	0.52777100	-0.90114200	-0.24183100
H	1.0	-1.04426003	-0.00620700	-0.24201100

Sum of electronic and zero-point Energies = -39.811839

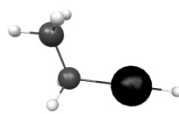
Figure S4: Calculated Cartesian coordinates and energies (Hartrees) for species relevant to the fragmentation of [CH₃CO₂AuO₂CCH₃]⁻ and [CH₃CO₂AuCH₃]⁻.



TS $[\text{CH}_3\text{AuCH}_3]^- \rightarrow [\text{CH}_3\text{CH}_2\text{AuH}]^-$ (-117 cm^{-1})

Au	79.0	-0.26557499	-0.09935900	-0.00294700
C	6.0	0.64091903	1.83990502	-0.13521200
H	1.0	1.75050604	1.76833797	-0.17594500
H	1.0	0.44400400	2.19124603	0.90722299
H	1.0	-1.94596601	0.07437000	0.05653200
C	6.0	1.76045406	-0.75286299	0.02710400
H	1.0	2.30145502	-0.43307701	0.92622900
H	1.0	1.70560300	-1.85382497	0.02617300
H	1.0	2.31660104	-0.41992000	-0.85871202

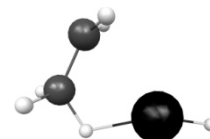
Sum of electronic and zero-point Energies = -215.493737



$[\text{CH}_3\text{CH}_2\text{AuH}]^-$

Au	79.0	0.45343000	-0.01650400	-0.00000800
C	6.0	-1.58419096	0.66687697	-0.00001000
H	1.0	-1.74387097	1.31839204	0.87817103
H	1.0	-1.74375403	1.31800699	-0.87849700
C	6.0	-2.66783190	-0.43148601	0.00016200
H	1.0	-2.57749391	-1.08487701	-0.88002503
H	1.0	-2.57767105	-1.08442104	0.88070601
H	1.0	-3.70459890	-0.03526400	-0.00004500
H	1.0	2.03857589	-0.54039800	-0.00059900

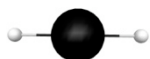
Sum of electronic and zero-point Energies = -215.578504



TS $[\text{CH}_3\text{CH}_2\text{AuH}]^- \rightarrow [\text{HAuH}]^- + \text{C}_2\text{H}_4$ (-577 cm^{-1})

Au	79.0	-0.43505099	-0.00698700	-0.00000200
H	1.0	-1.90572298	0.57692999	-0.00004900
C	6.0	2.25933003	0.80757898	-0.00005200
C	6.0	2.10297608	-0.66780502	0.00006200
H	1.0	2.01094294	1.34616804	-0.91671300
H	1.0	2.01093197	1.34631205	0.91652101
H	1.0	1.01488495	-1.27063501	0.00010000
H	1.0	2.53205204	-1.14263701	0.89122099
H	1.0	2.53206491	-1.14277601	-0.89101601

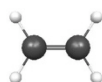
Sum of electronic and zero-point Energies = -215.493737



$[\text{HAuH}]^-$

Au	79.0	0.00000000	0.00000000	0.00000000
H	1.0	0.00000000	0.00000000	1.68150496
H	1.0	0.00000000	0.00000000	-1.68151498

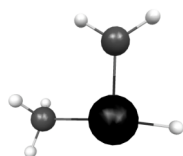
Sum of electronic and zero-point Energies = -137.012661



C_2H_4

C	6.0	-0.66759598	0.00000000	-0.00016000
C	6.0	0.66759300	0.00000000	-0.00019000
H	1.0	-1.24046302	0.92477602	0.00050000
H	1.0	-1.24046397	-0.92477602	0.00050000
H	1.0	1.24047303	-0.92476797	0.00054900
H	1.0	1.24047303	0.92476797	0.00054900

Sum of electronic and zero-point Energies = -78.540582



Triplet $\text{CH}_3\text{Au}(\text{CH}_2)\text{H}^-$

Au	79.0	-0.07623100	-0.24994500	-0.00044700
C	6.0	-1.19329202	1.59079695	0.00007500
H	1.0	-0.67845601	2.55840898	-0.03212900
H	1.0	-2.28687310	1.61546004	0.03325300
H	1.0	-1.45645905	-1.18979394	0.00140700
C	6.0	1.83981800	0.73816401	0.00210600
H	1.0	2.03737998	1.18722904	0.98865700
H	1.0	2.67039204	0.05147200	-0.22608399
H	1.0	1.85711503	1.54911900	-0.74289900

Sum of electronic and zero-point Energies = -215.457801

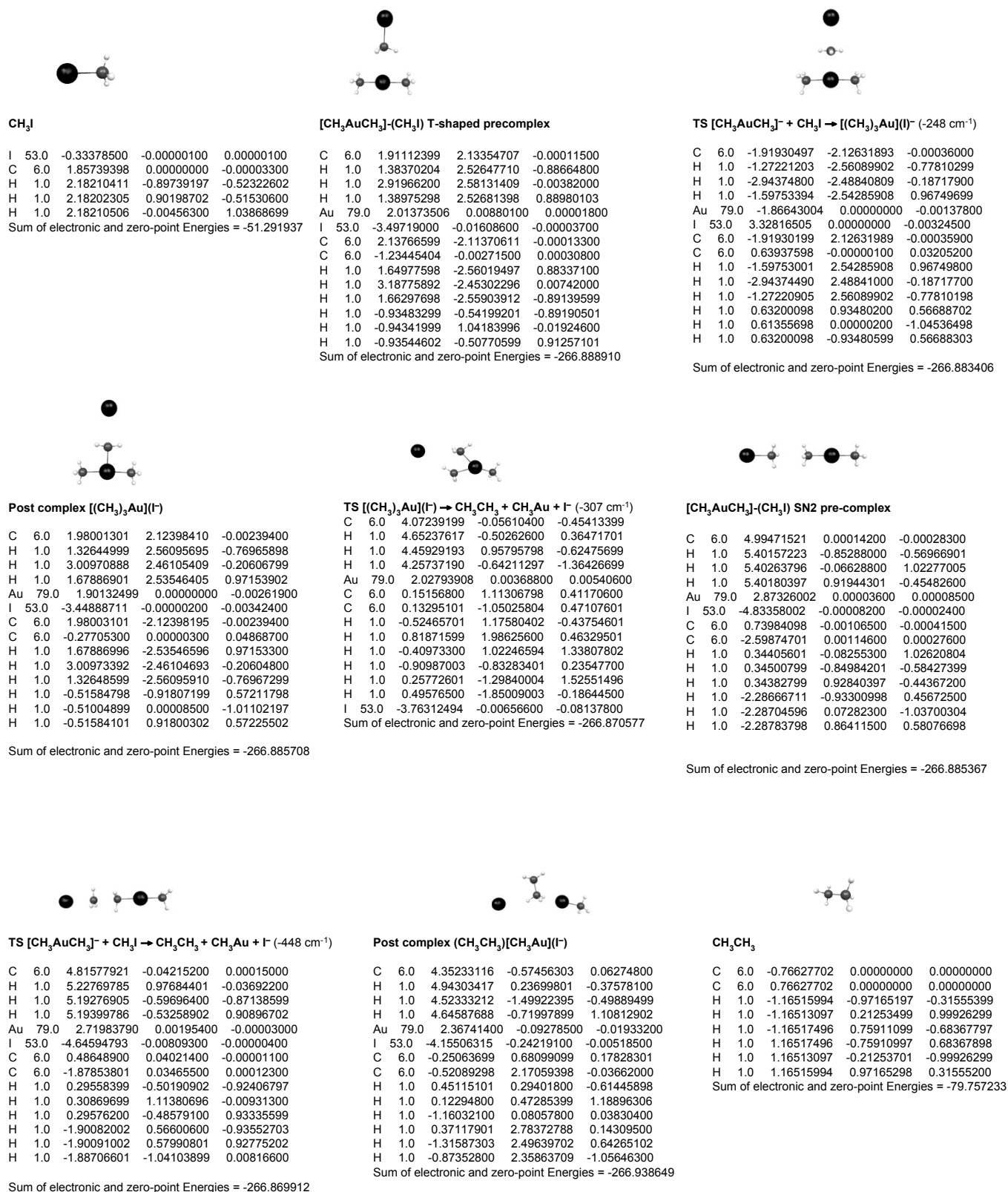


Figure S6: Calculated Cartesian coordinates and energies (Hartrees) for [CH₃AuCH₃]⁺ reaction with CH₃I.

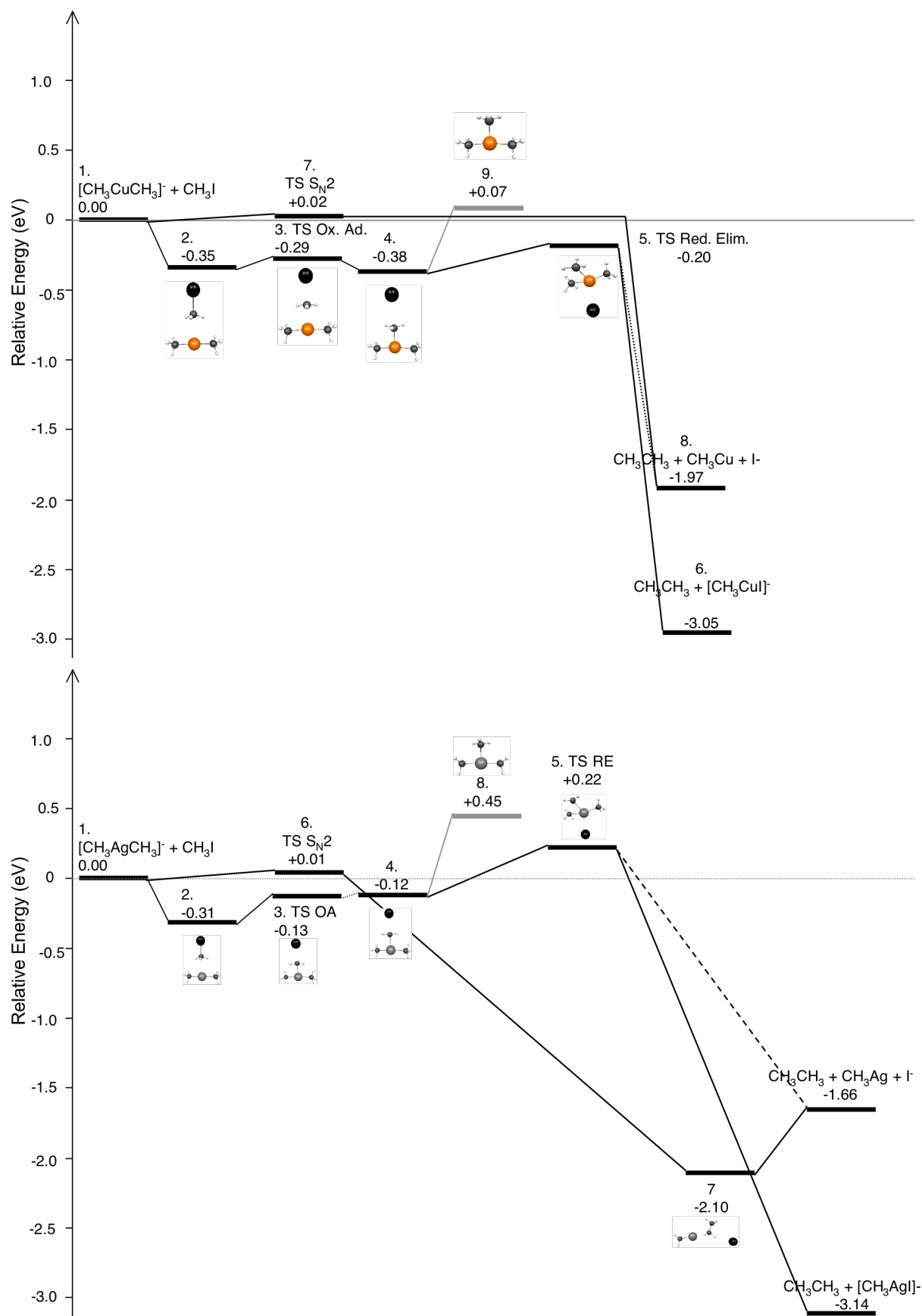
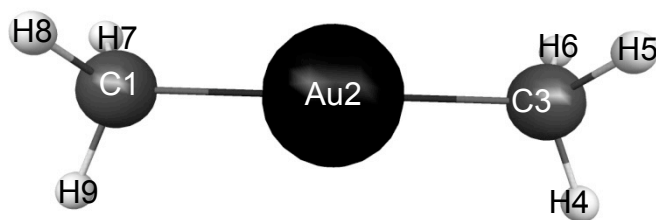


Figure S8: Comparison of $\text{CH}_3\text{MCH}_3^-$ ($\text{M} = \text{Cu}$ and Ag) reaction with CH_3I at B3LYP.

Crystal Structure Comparison:

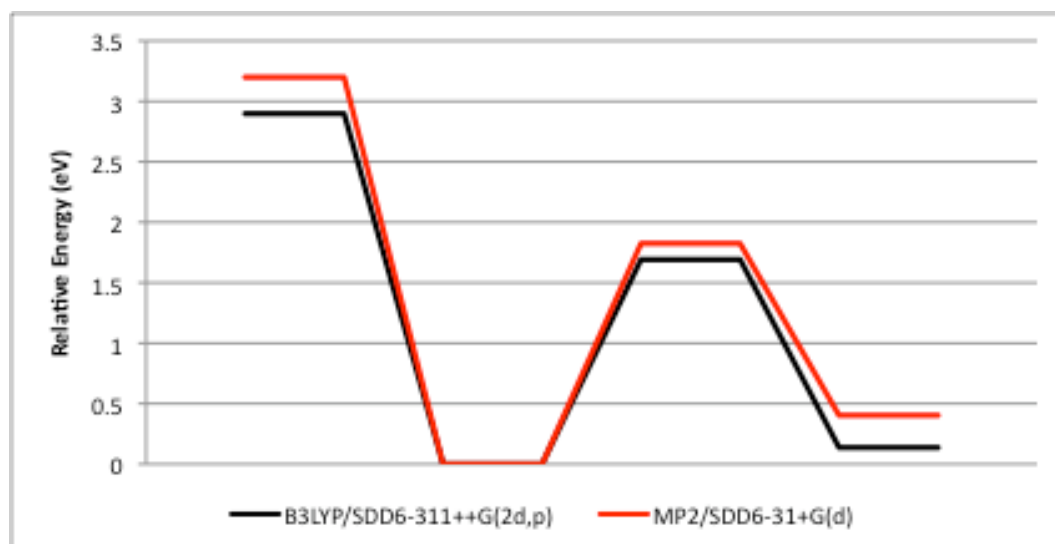


Bond length (Å)/angle (deg)	DFT value: B3LYP/SDD6-31+G(d)	Experimental value ¹	Previous theory ^{2,3}
C1- Au2, C3-Au2	2.13, 2.13	2.087(5), 2.076(6), 2.084(5), 2.053(5)	2.150 ² , 2.109 ² , 2.13 ³
C2-Au2-C3	179.5	178.2(2), 178.2(2)	180.0 ³

1. Zhu, D; Lindeman, S. V.; Kochi, J. K., *Organometallics*, 1999, 18, 2241-2248
2. Tossell, J. A.,. *Chem. Phys. Lett.* **1998**, 286, (1-2), 73-78.
3. Nakanishi, W.; Yamanaka, M.; Nakamura, E.,. *J. Am. Chem. Soc.* **2005**, 127, (5), 1446-1453.

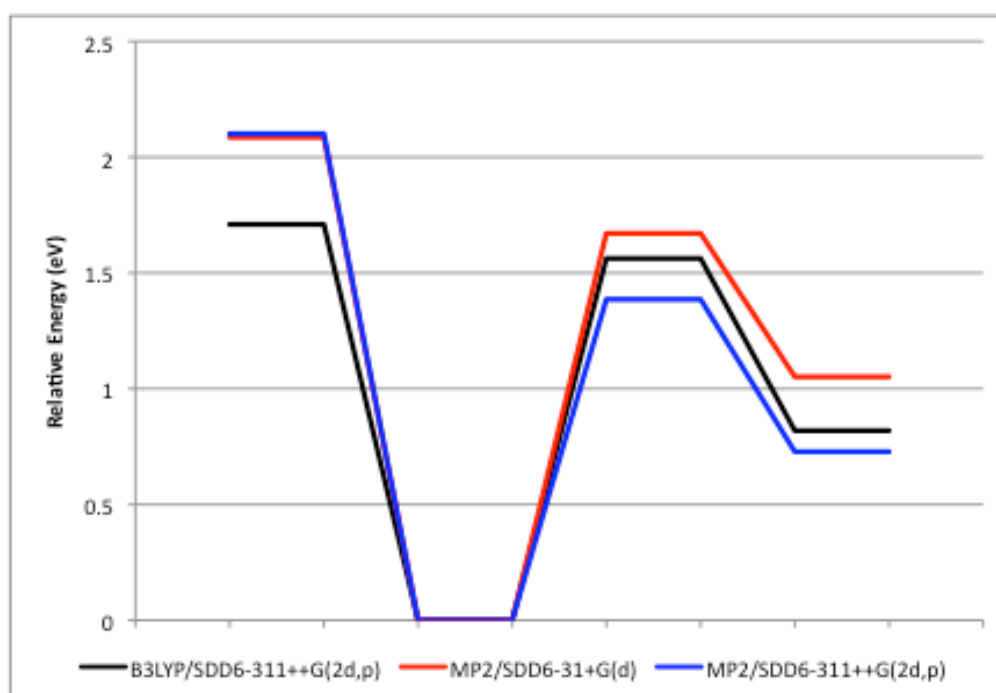
Figure S9: Comparison of experimental and computed bond lengths and geometries.

(a)



$[\text{CH}_3\text{CO}_2\text{AuO}_2\text{CCH}_3]^-$	B3LYP/SDD6-31+G(d)	B3LYP/SDD6-311+G(2d,p)	MP2/SDD6-31+G(d)
Acetate loss	2.91	2.90	3.20
Decarboxylation	1.80	1.69	1.82

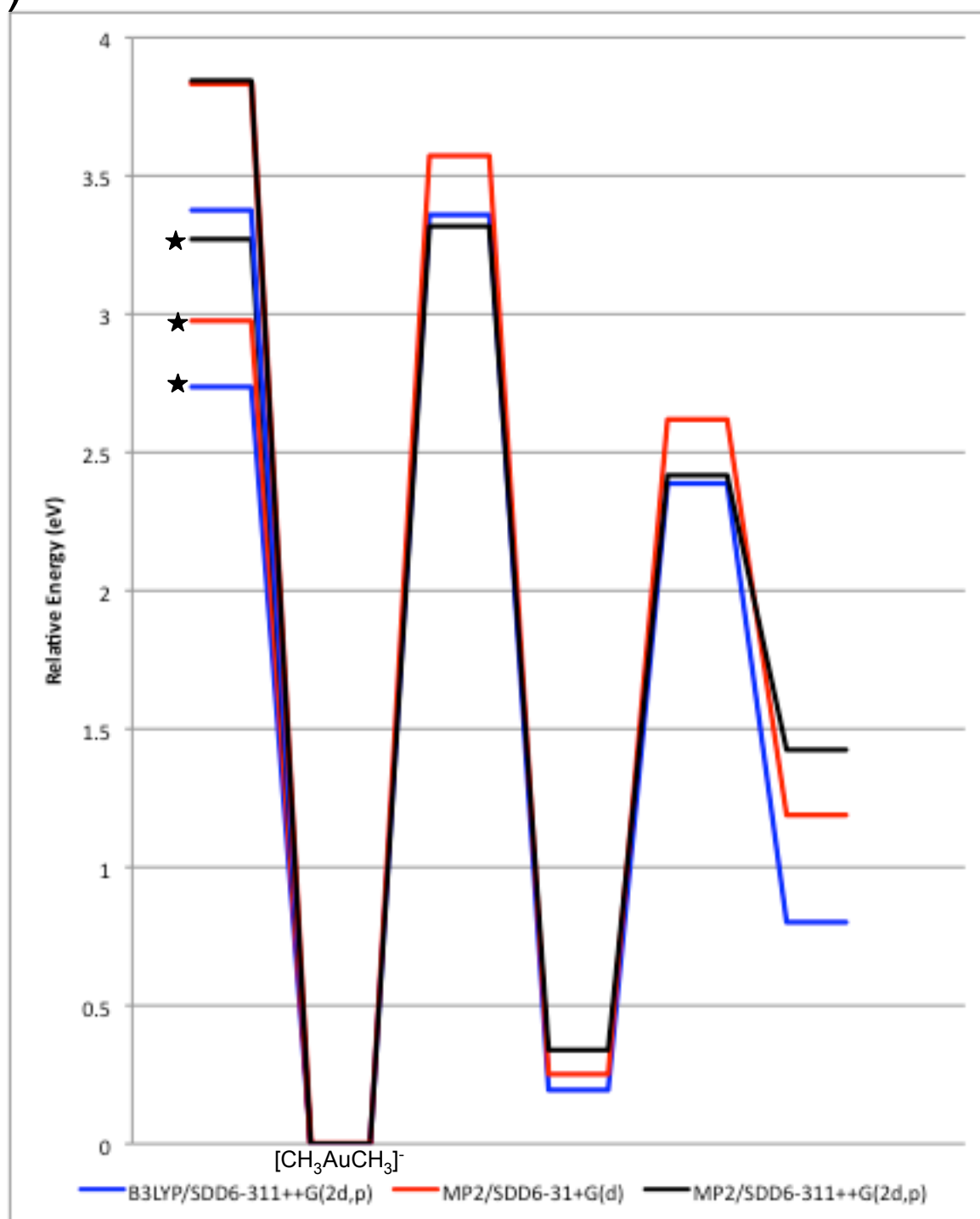
(b)



$[\text{CH}_3\text{AuO}_2\text{CCH}_3]^-$	B3LYP/SDD6-31+G(d)	B3LYP/SDD6-311+G(2d,p)	MP2/SDD6-31+G(d)	MP2/SDD6-311+G(2d,p)
Acetate loss	1.72	1.71	2.09	2.10
Decarboxylation	1.65	1.56	1.67	1.39
CH_3 Heterolysis	5.33	5.25	5.60	5.56
CH_3 Homolysis	2.95	2.74	2.98	3.27

Figure S10: Comparison of computational methods and basis sets for fragmentation of, a) $[\text{CH}_3\text{CO}_2\text{AuO}_2\text{CCH}_3]^-$; b) $[\text{CH}_3\text{AuO}_2\text{CCH}_3]^-$ and c) $[\text{CH}_3\text{AuCH}_3]^-$. Table a), b) and c) contains key energies (eV).

(c)



$[\text{CH}_3\text{AuCH}_3]^-$	B3LYP/SDD6-31+G(d)	B3LYP/SDD6-311+G(2d,p)	MP2/SDD6-31+G(d)	MP2/SDD6-311+G(2d,p)
CH_3 Heterolysis	3.42	3.38	3.83	3.84
★ CH_3 Homolysis	2.73	2.74	2.98	3.27
Rearrangement	3.43	3.36	3.57	3.32
beta hydride	2.49	2.39	2.62	2.42

Figure S10: Comparison of computational methods and basis sets for fragmentation^{13f}; a) $[\text{CH}_3\text{CO}_2\text{AuO}_2\text{CCH}_3]^-$; b) $[\text{CH}_3\text{AuO}_2\text{CCH}_3]^-$ and c) $[\text{CH}_3\text{AuCH}_3]^-$. Table a), b) and c) contains key energies (eV).

Table S11: Intensity comparison at different q values and reaction times: trapped $[\text{CH}_3\text{CO}_2\text{AuCH}_3]^-$ versus its CID fragments

MS parameter	Reaction time	q = 0.25						q = 0.35					
		30 ms			100 ms			30 ms			100 ms		
		NCE_a	Sum Intensity^b	R^c	NCE^a	Sum Intensity^b	R^c	NCE^a	Sum Intensity^b	R^c	NCE^a	Sum Intensity^b	R^c
$\text{CH}_3\text{Au}(\text{CH}_3\text{CO}_2)^-$		0	3102716										
CID of $\text{CH}_3\text{Au}(\text{CH}_3\text{CO}_2)^-$													
		18	3194757	1.03	15	3241014	1.04	18	2911279	0.94	15	3242356	1.04

- a: normalised collision energy, which is the relative CID energy used (the ion-trap software).
- b: The sum of the integration of the intensities for all peaks present in the mass spectrum
- c: ratio between the Sum of the intensities upon CID and the intensity upon trapping.