Electronic Supplementary Information for Dalton Transactions

Electronic Supplementary Information

Novel Platinum(II) Complexes of 3-(Aminomethyl)naphthoquinone Mannich Bases: Synthesis, Crystal Structure and Cytotoxic activities

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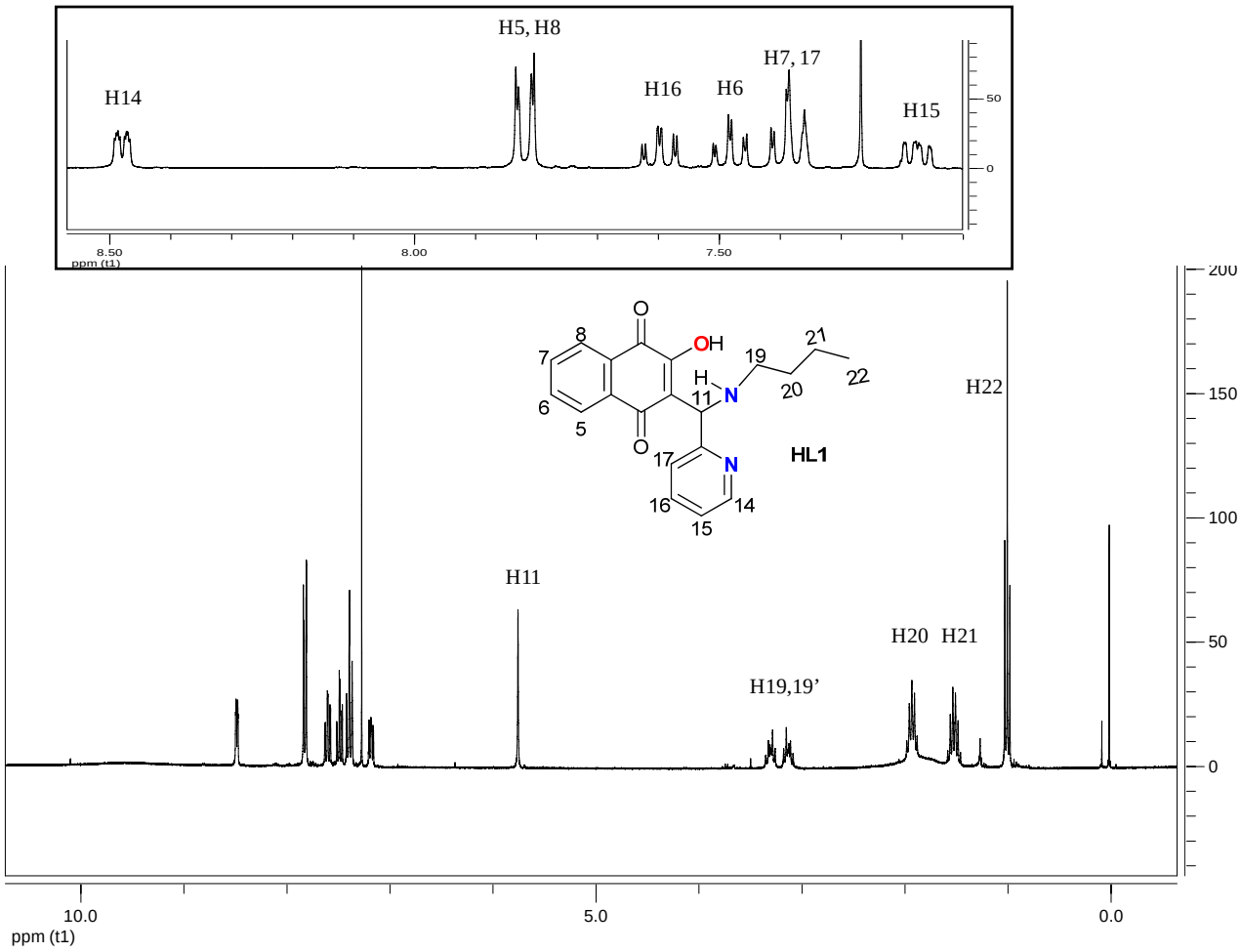

Figure S1. ^{1}H NMR spectrum of **HL1** in $CDCl_3$.

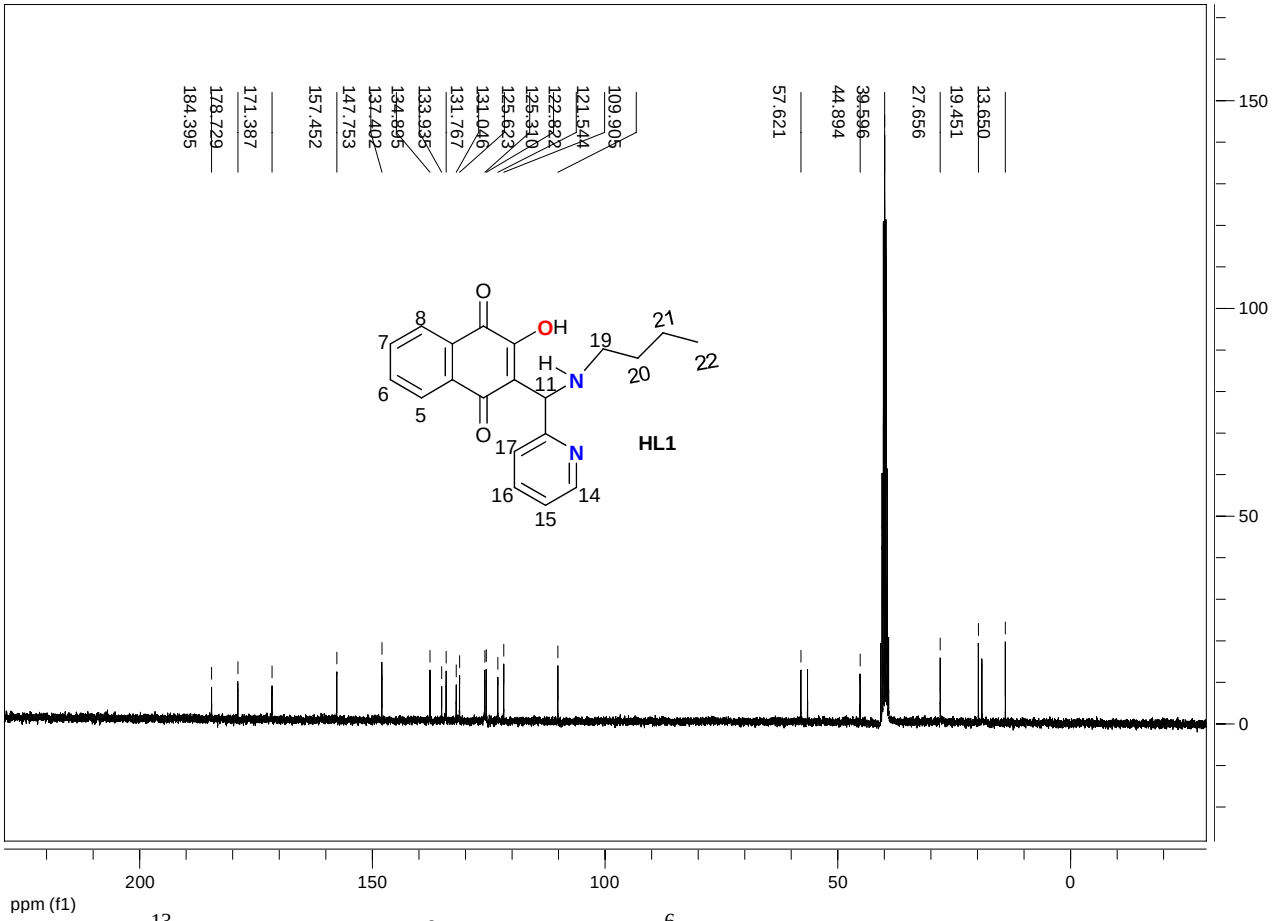

Figure S2. ^{13}C NMR spectrum of **HL1** in DMSO-d^6.

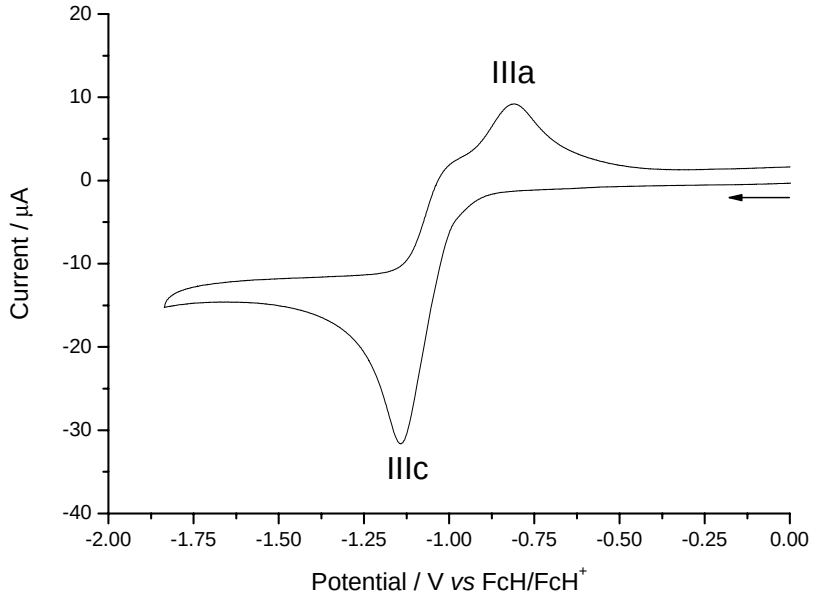

Figure S3. Cyclic voltammogram of **HL1** in 0.1 $molL^{-1}$ Bu_4ClO_4/CH_3OH obtained at $0.1Vs^{-1}$ with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

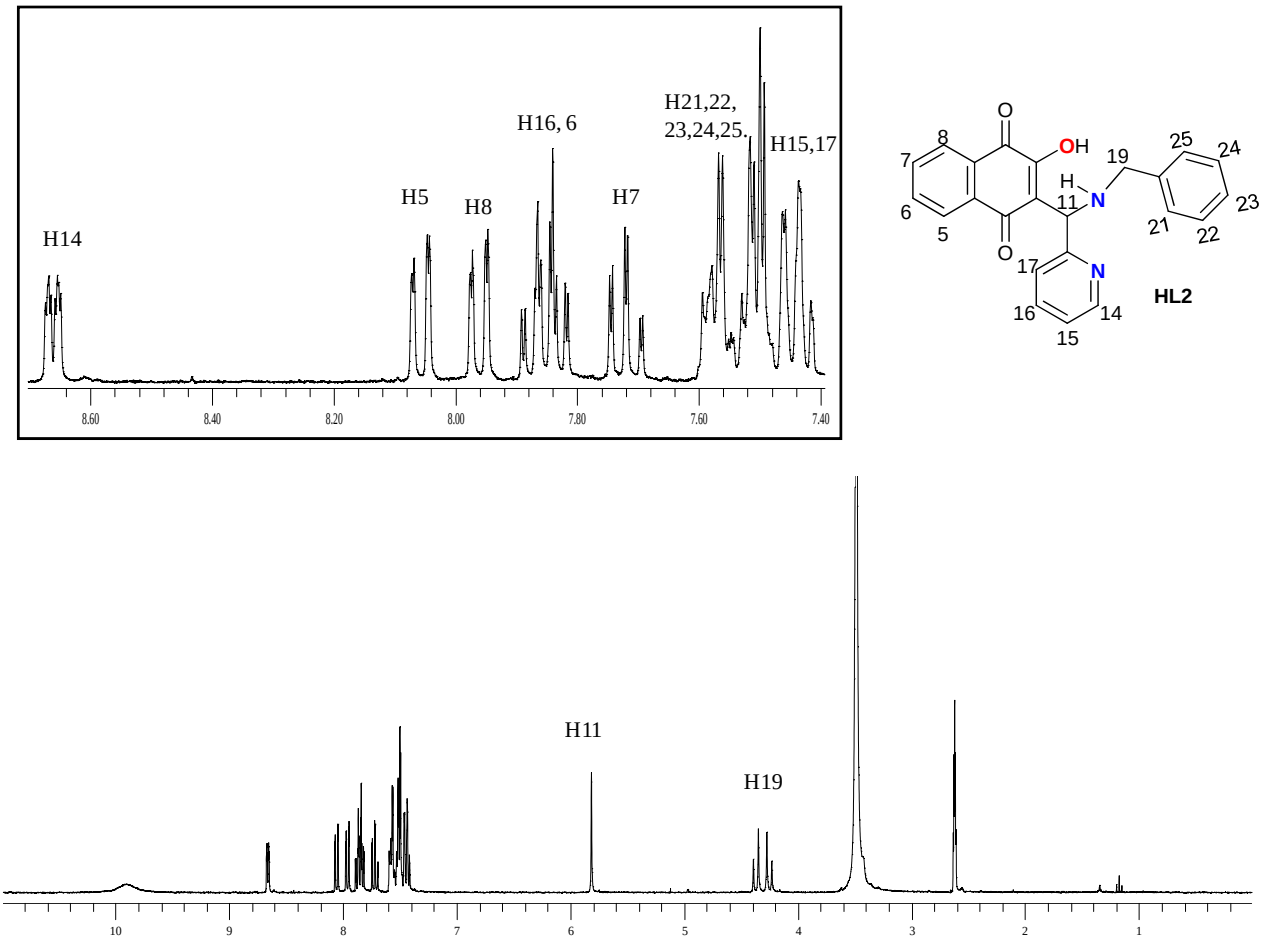

Figure S4. ^{1}H NMR spectrum of **HL2** in DMSO-d^{6}.

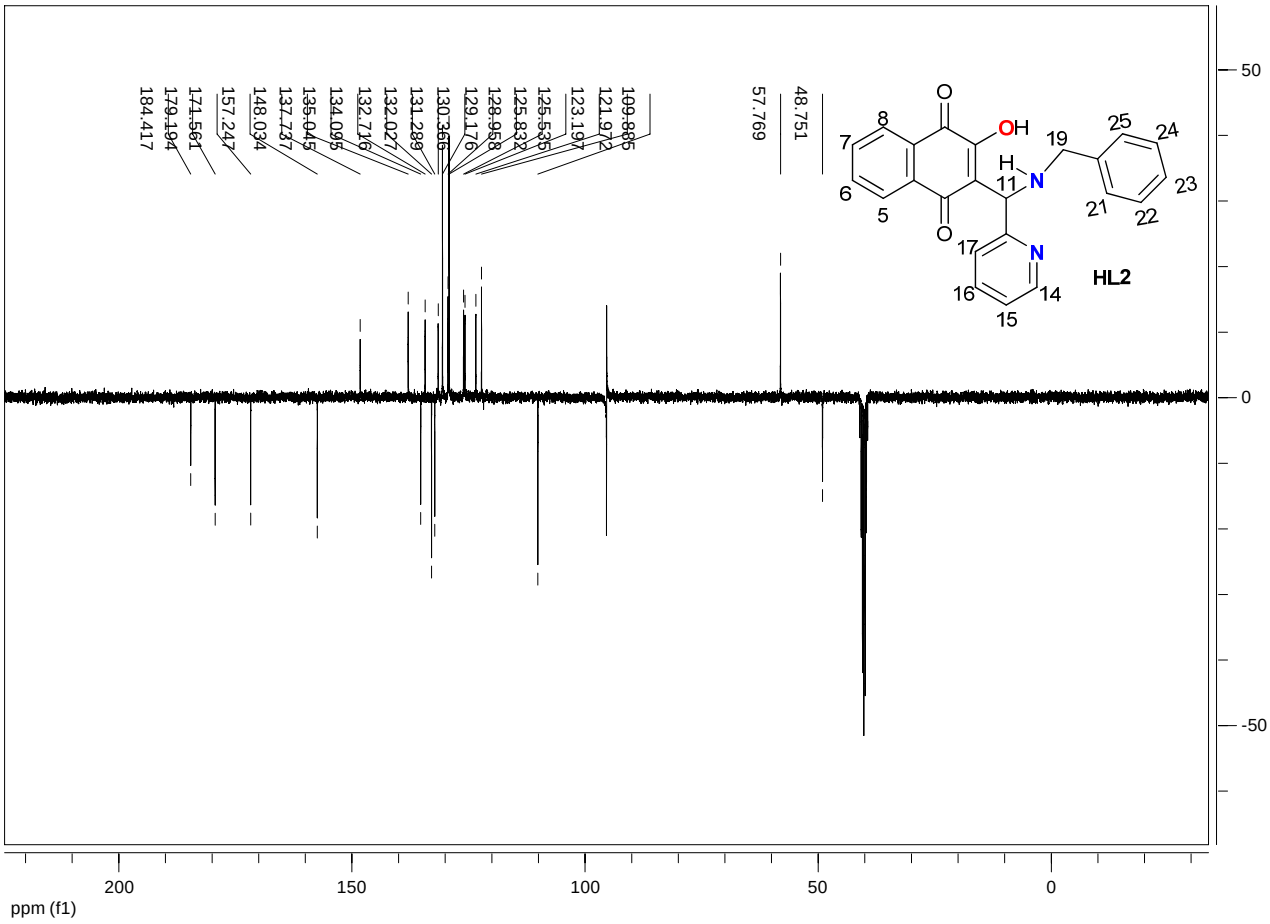

Figure S5. ^{13}C NMR spectrum (APT) of **HL2** in DMSO-d^{6}.

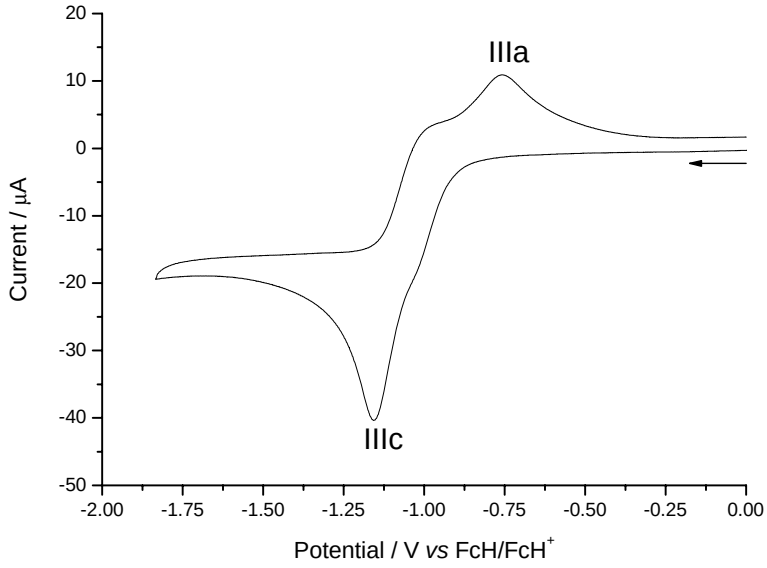

Figure S6. Cyclic voltammogram of **HL2** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

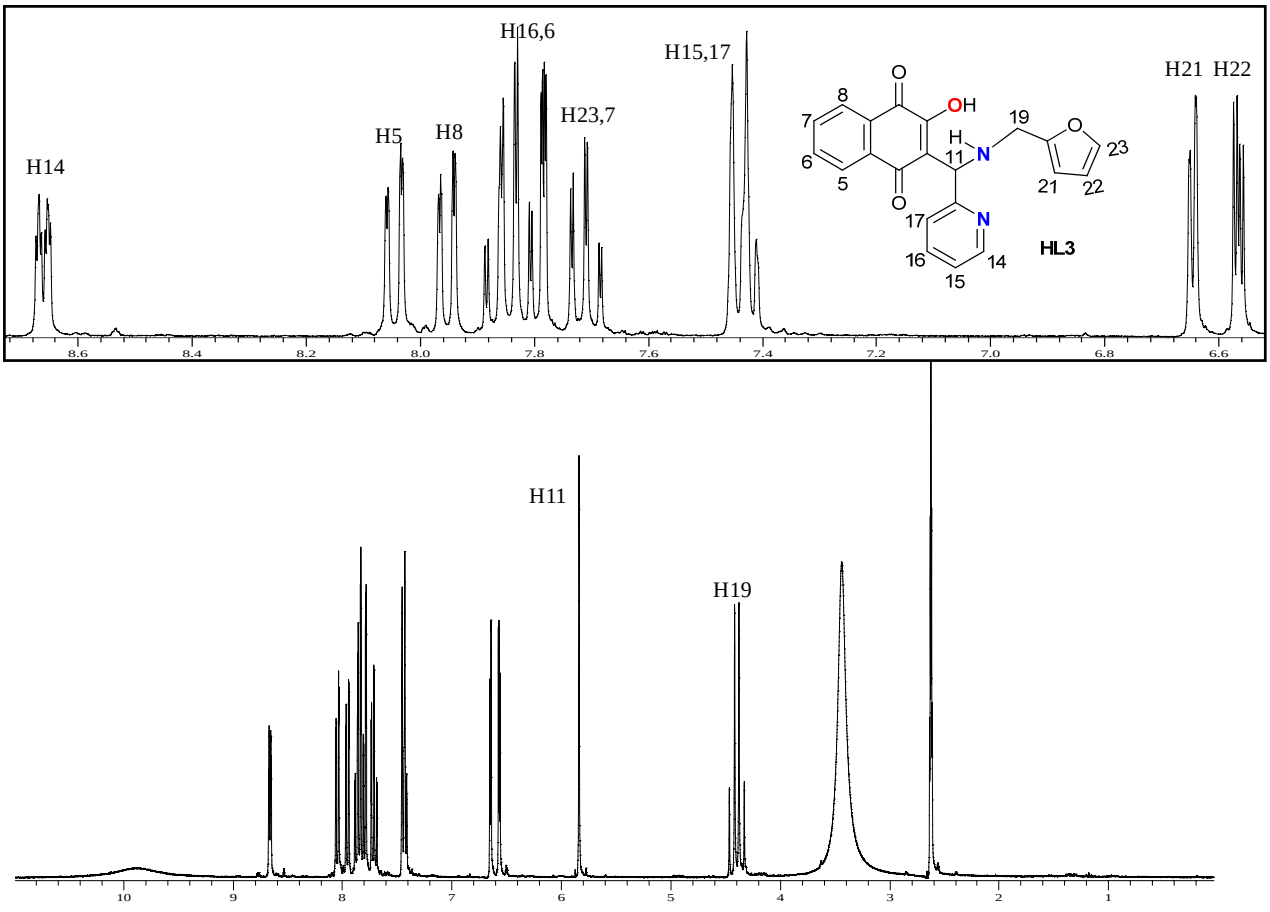

Figure S7. ^{1}H NMR spectrum of **HL3** in DMSO-d^{6}.

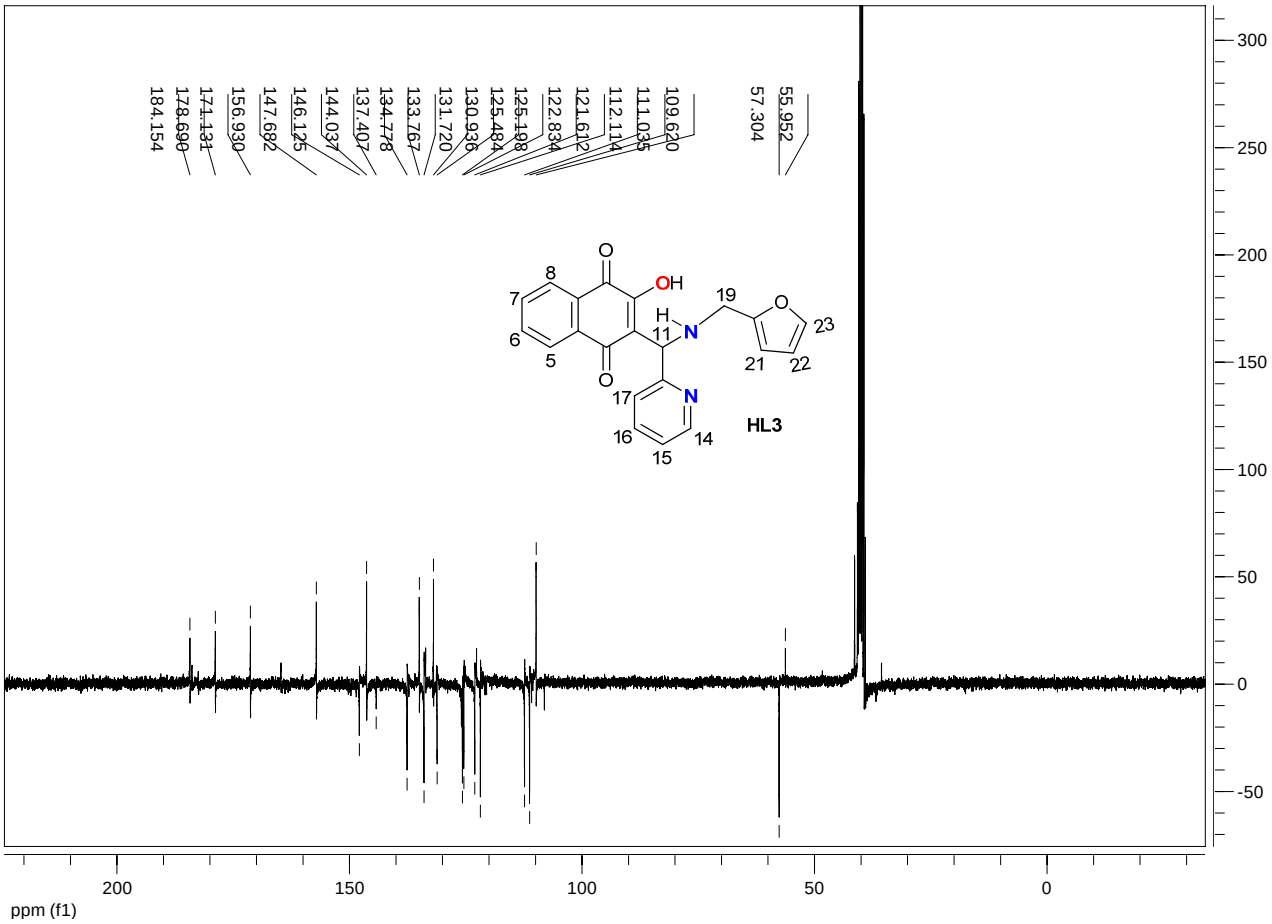

Figure S8. ^{13}C NMR spectrum (APT) of **HL3** in DMSO-d^6.

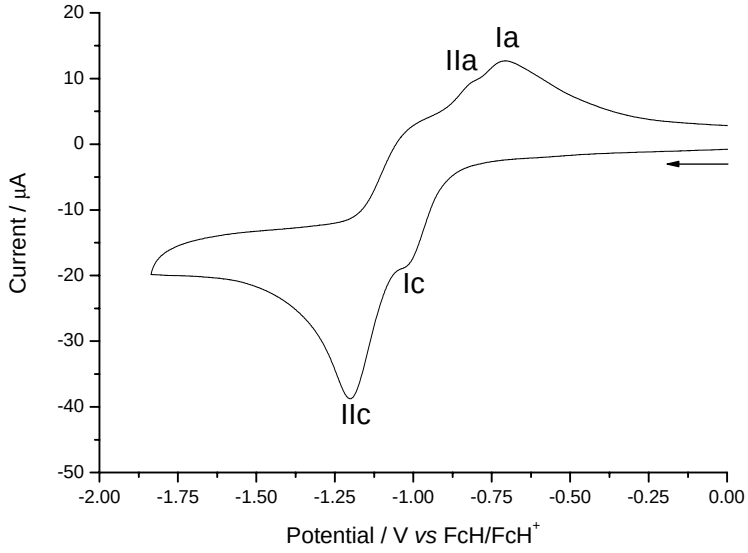

Figure S9. Cyclic voltammogram of **HL3** in 0.1 $molL^{-1}$ Bu_4ClO_4/CH_3OH obtained at $0.1Vs^{-1}$ with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

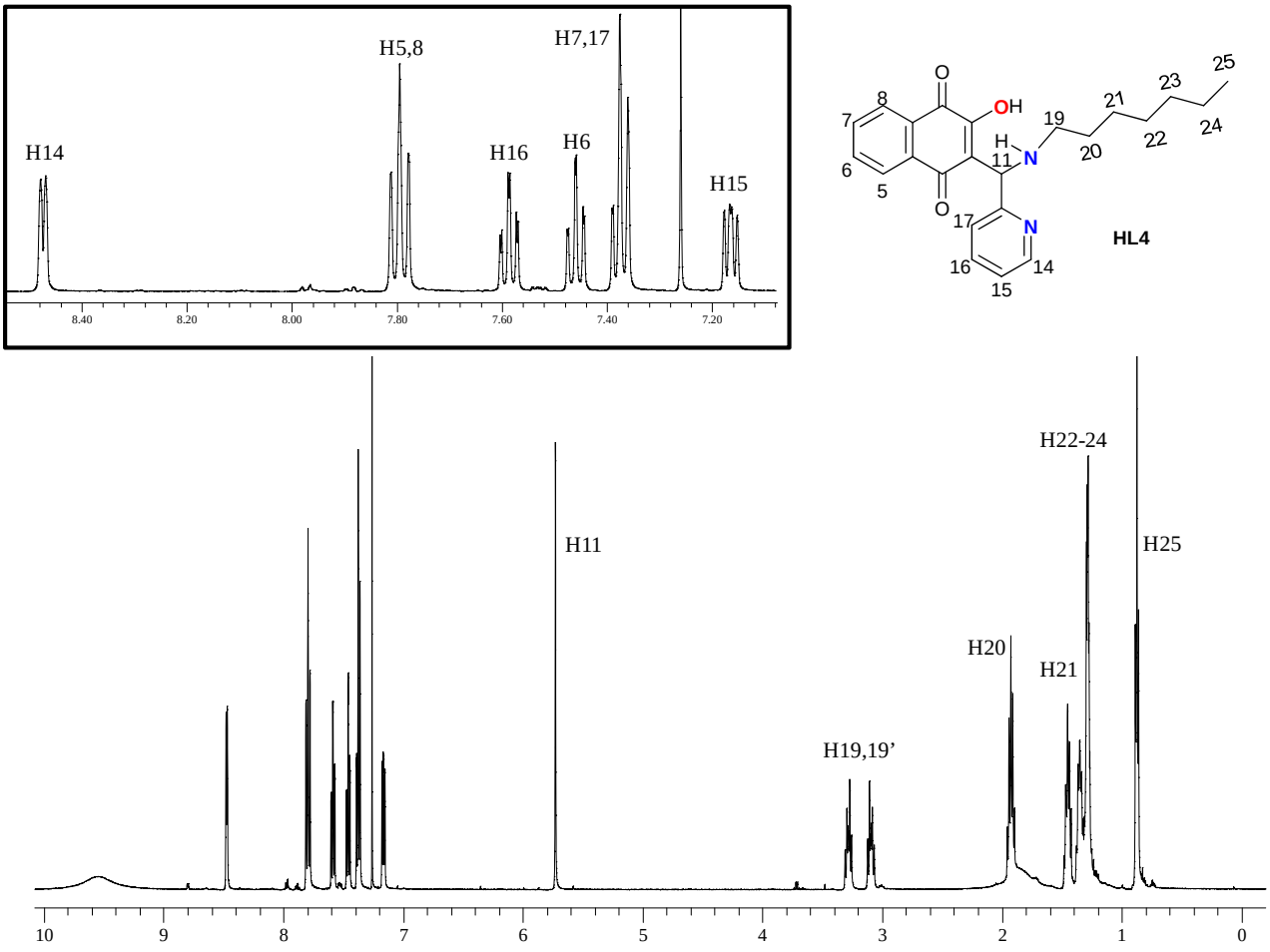

Figure S10. ^{1}H NMR spectrum of **HL4** in $CDCl_3$.

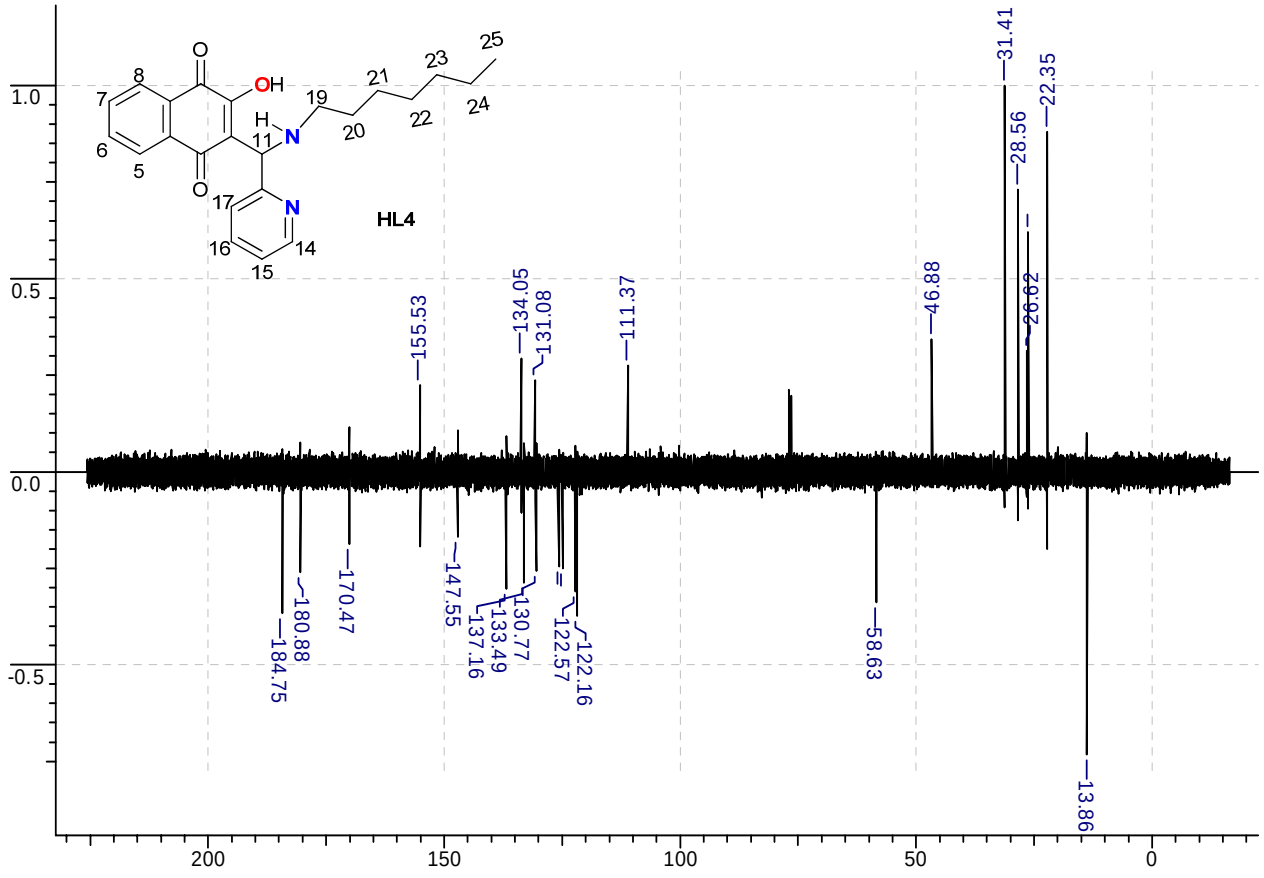

Figure S11. ^{13}C NMR spectrum (APT) of **HL4** in $CDCl_3$.

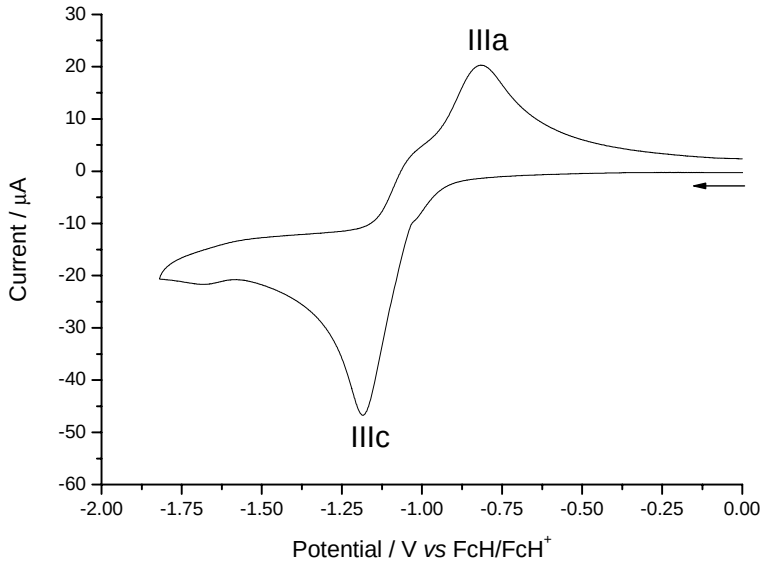

Figure S12. Cyclic voltammogram of **HL4** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

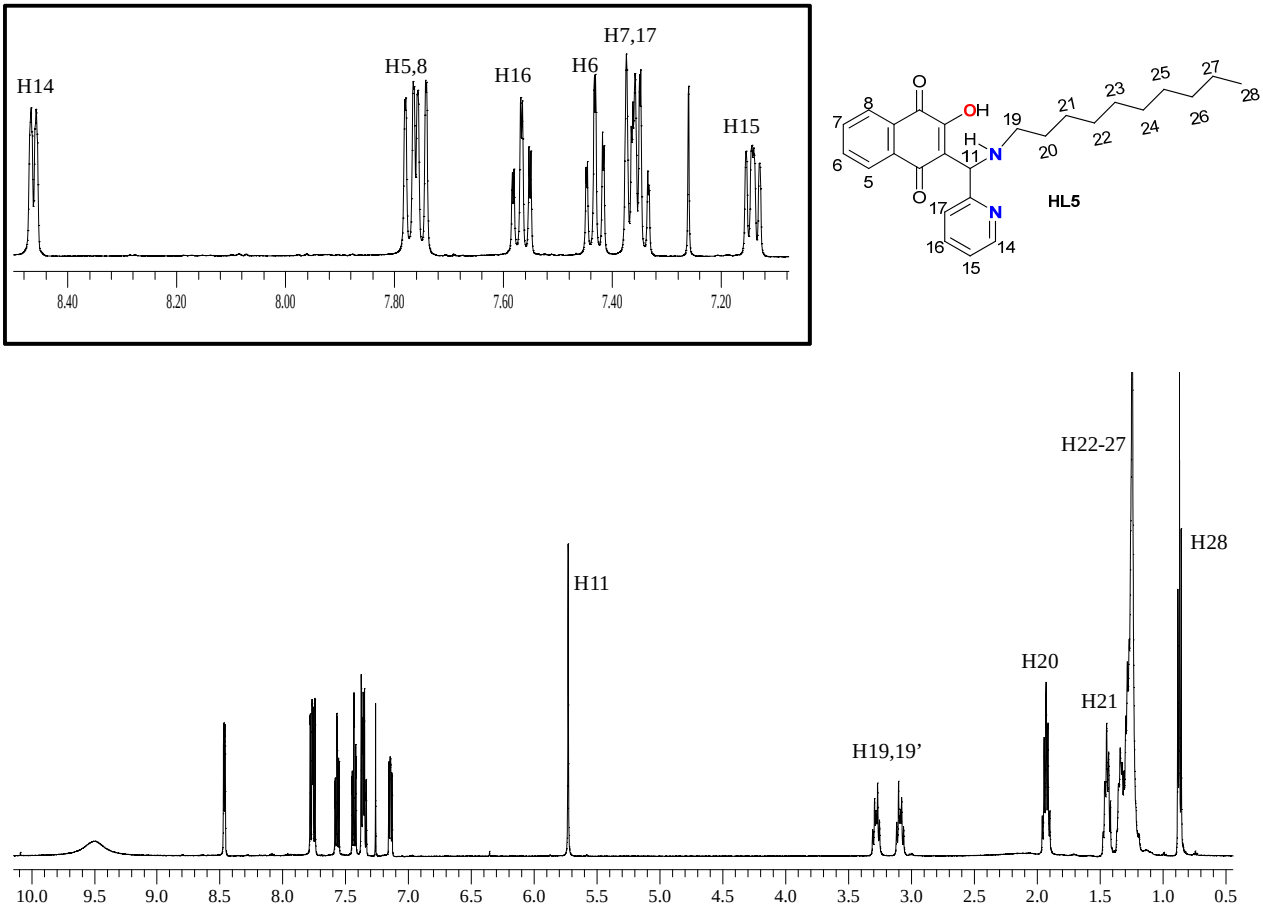

Figure S13. ^{1}H NMR spectrum of **HL5** in $CDCl_3$.

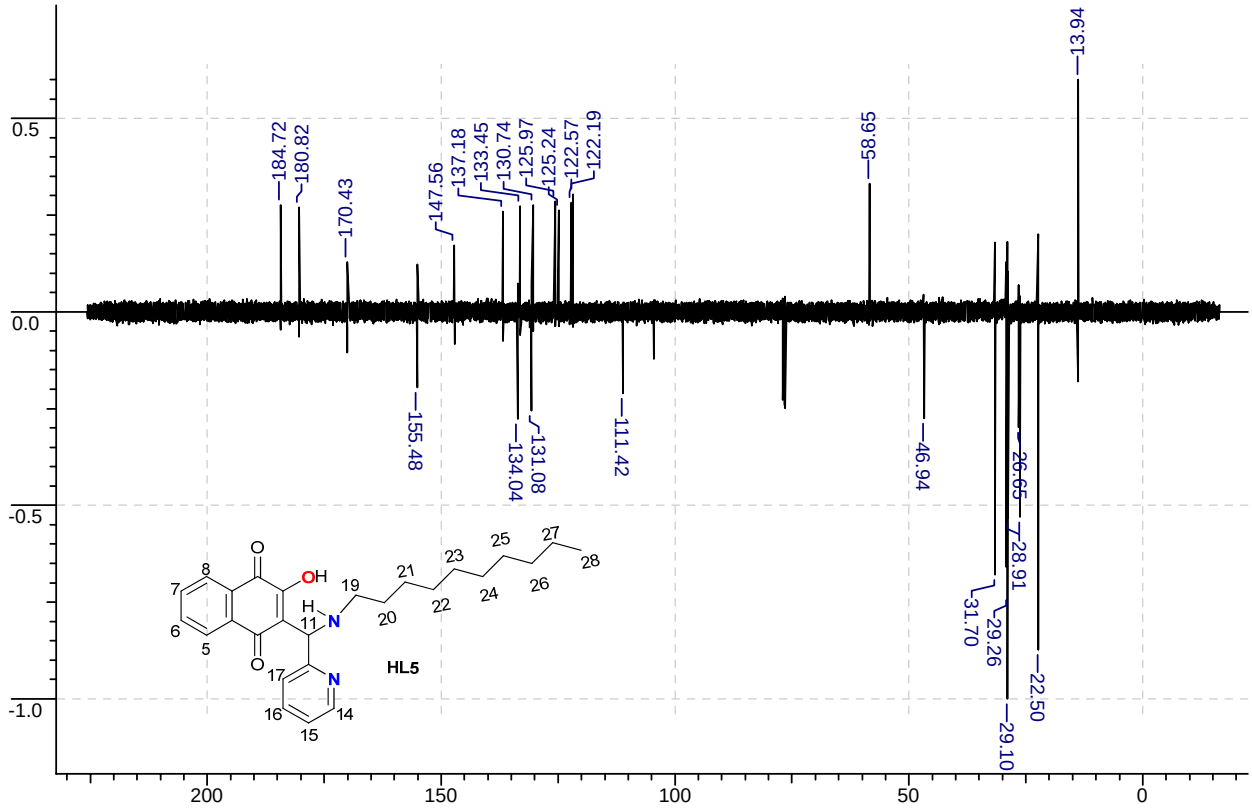

Figure S14. ^{13}C NMR spectrum (APT) of **HL5** in $CDCl_3$.

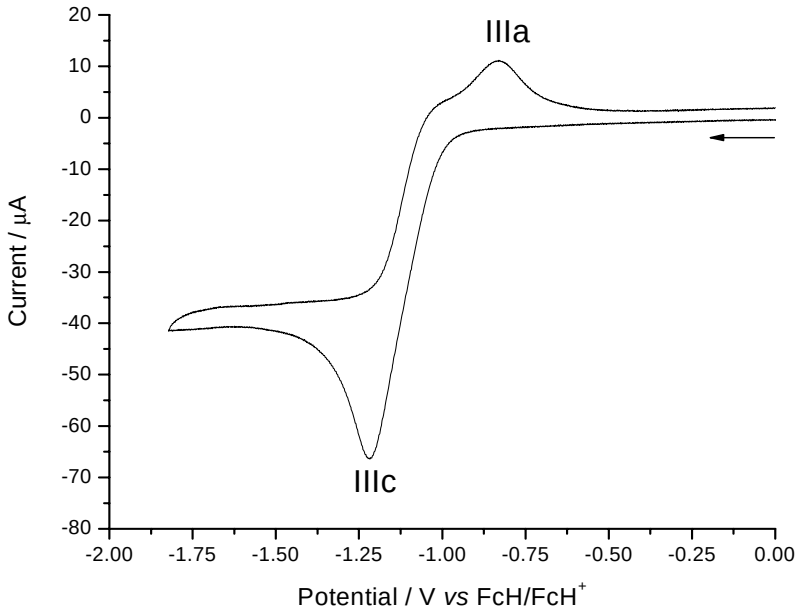

Figure S15. Cyclic voltammogram of **HL5** in 0.1 $molL^{-1}$ Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

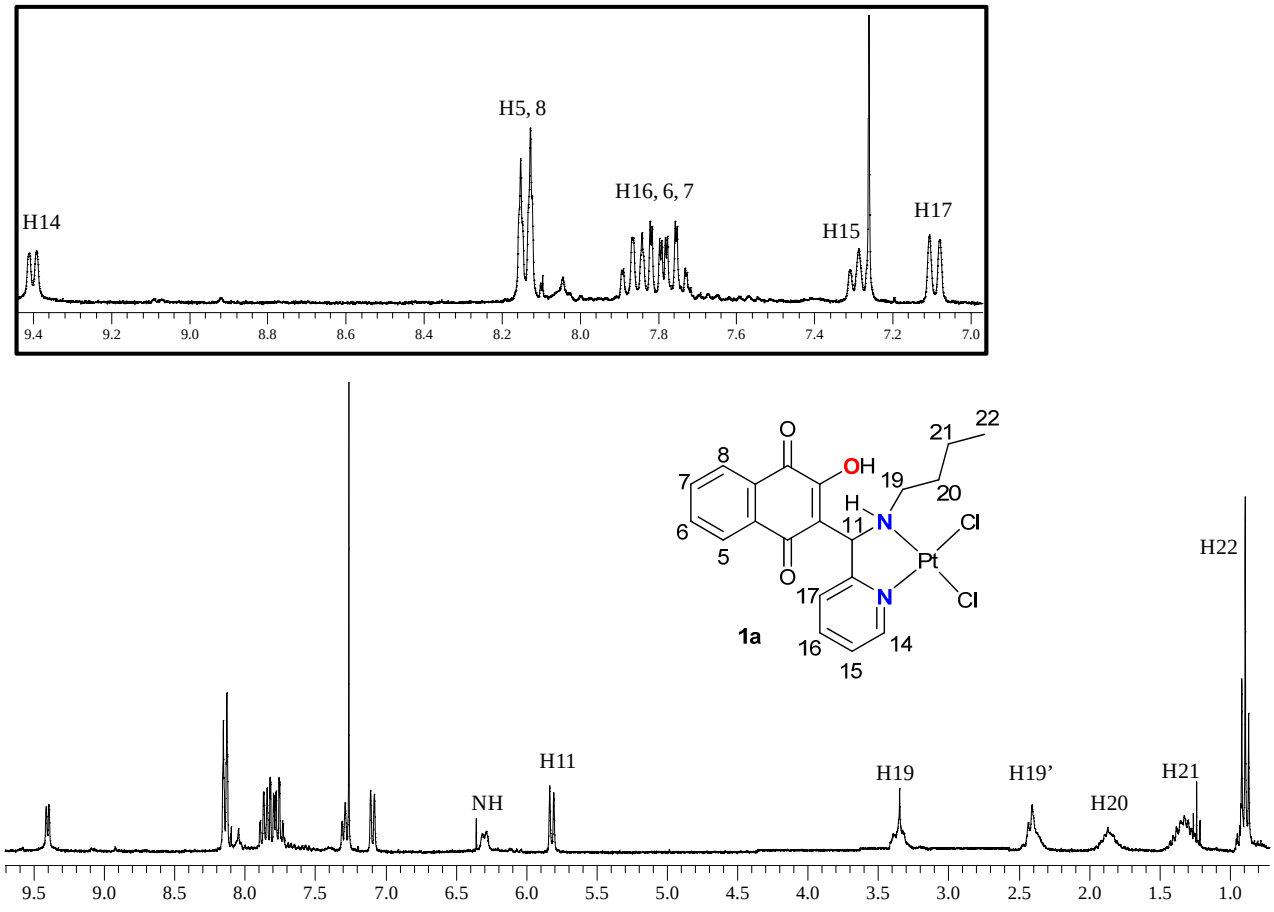

Figure S16. ^{1}H NMR spectrum of **1a** in $CDCl_3$.

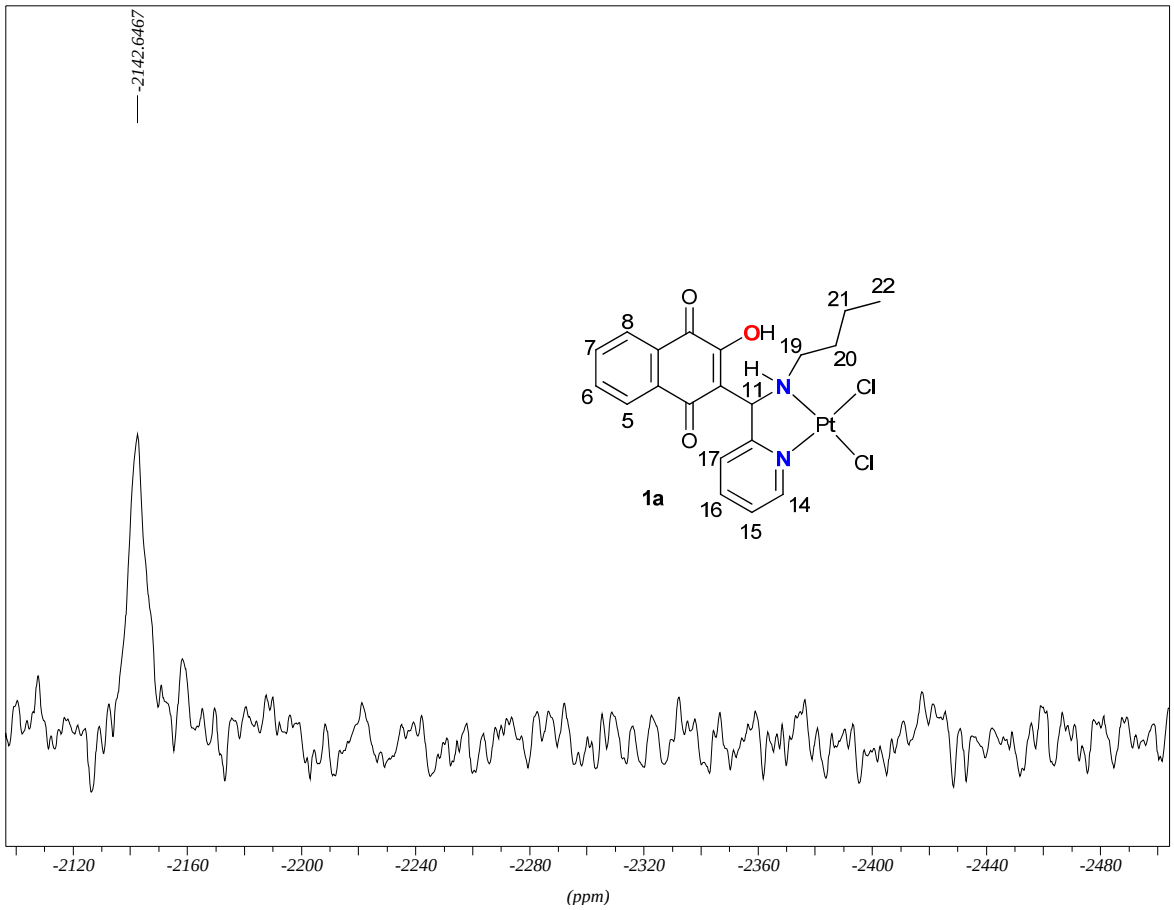

Figure S17. ^{195}Pt NMR spectrum of **1a** in $CDCl_3$.

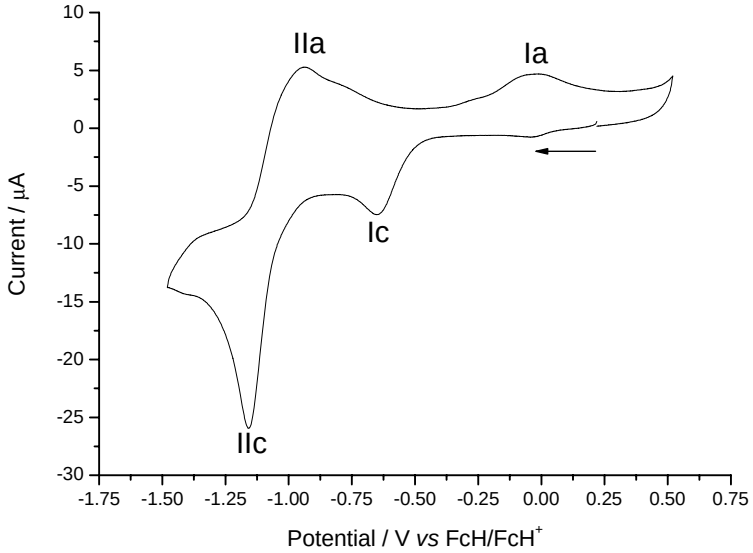

Figure S18. Cyclic voltammogram of **1a** in 0.1 $molL^{-1}$ Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

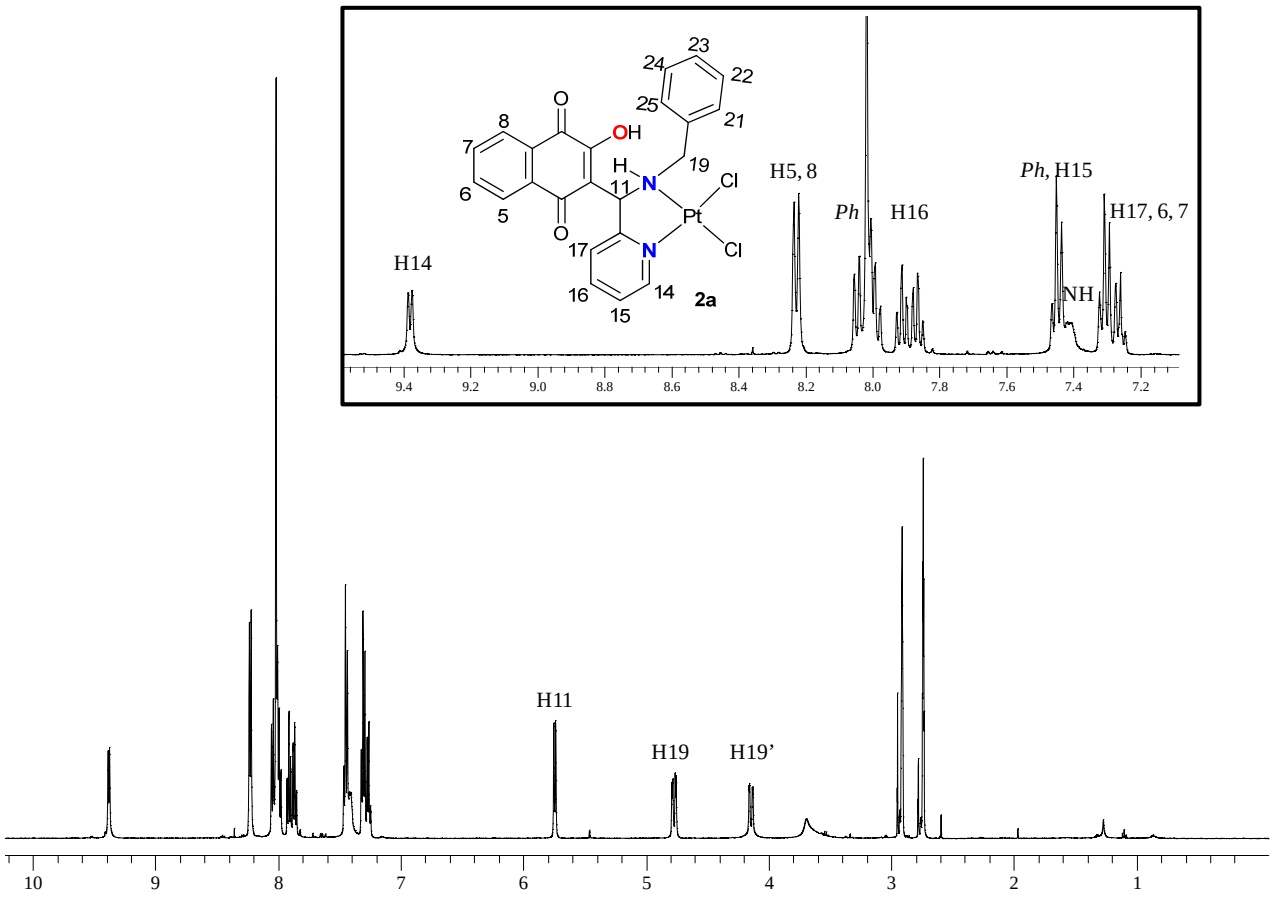

Figure S19. ^{1}H NMR spectrum of **2a** in DMF-d^{7}.

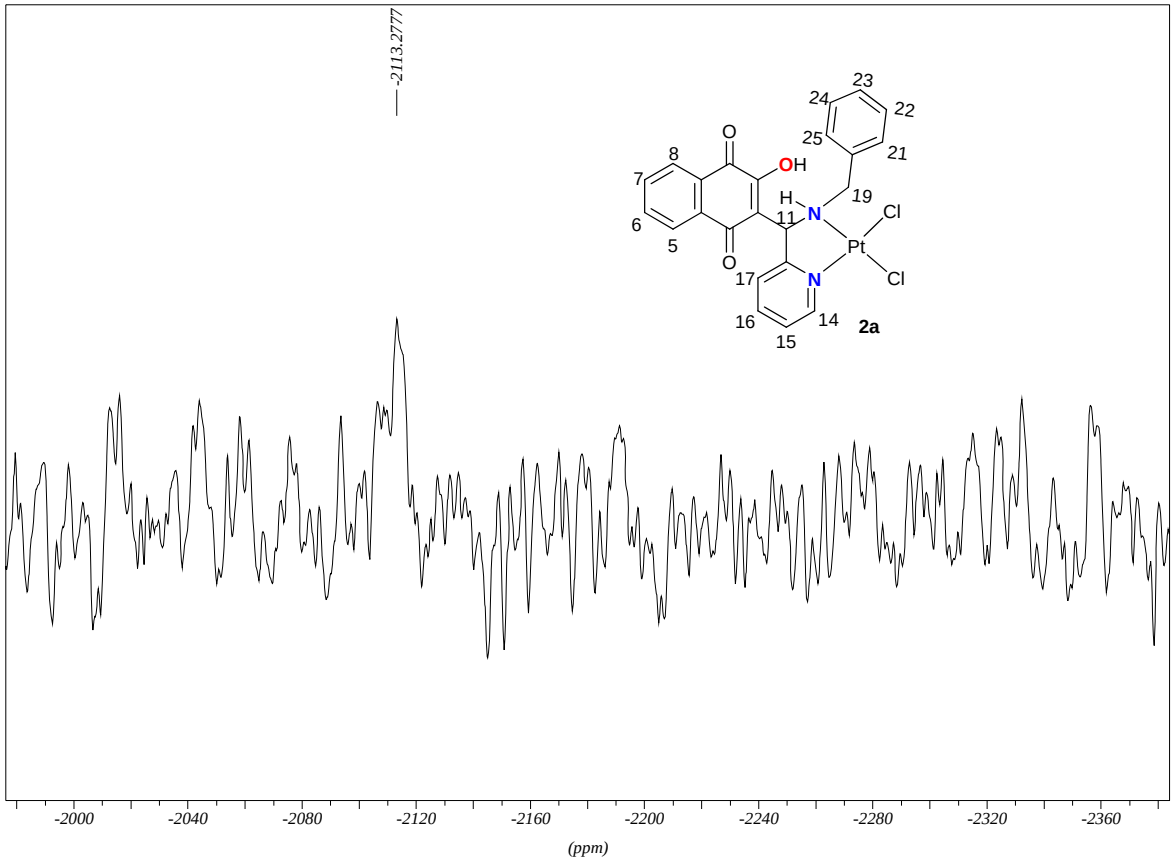

Figure S20. ^{195}Pt NMR spectrum of **2a** in DMF-d^{7}.

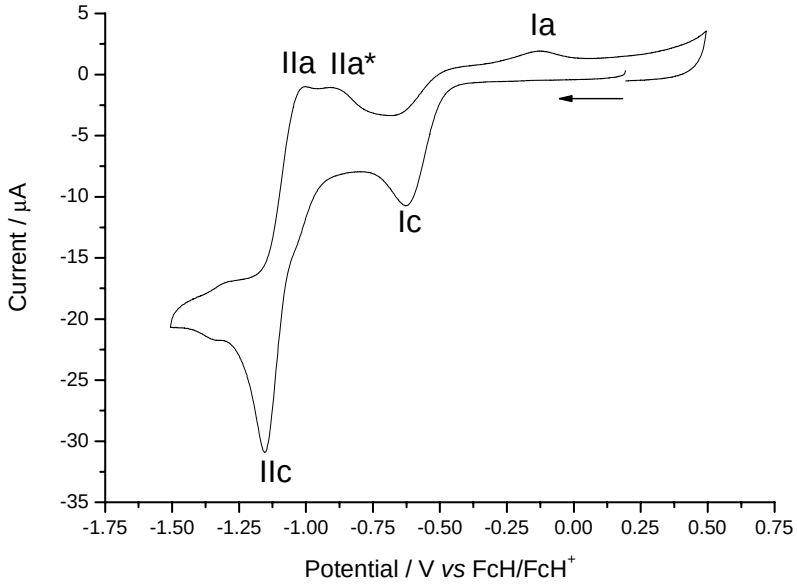

Figure S21. Cyclic voltammogram of **2a** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

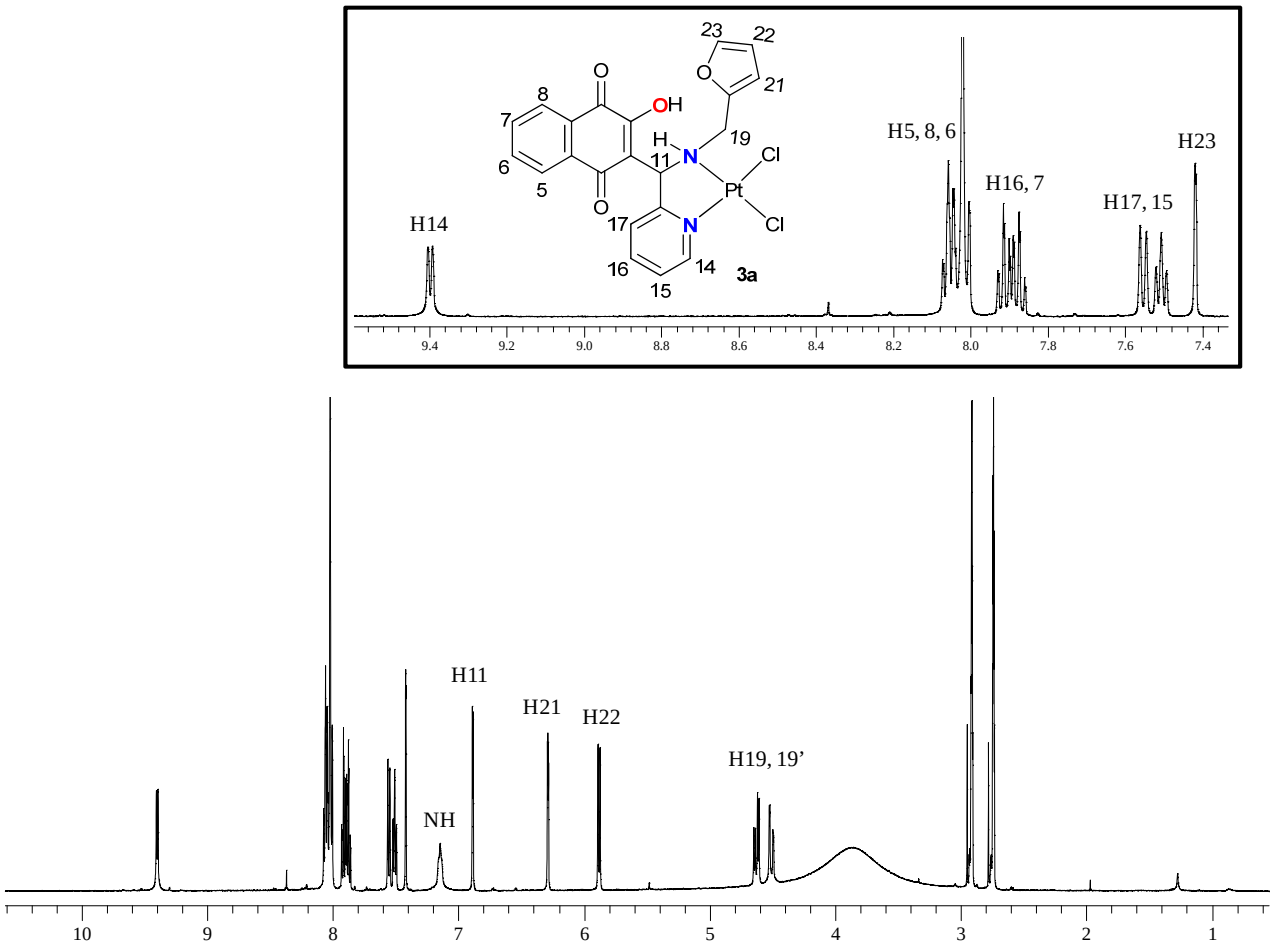

Figure S22. ^{1}H NMR spectrum of **3a** in DMF-d^{7}.

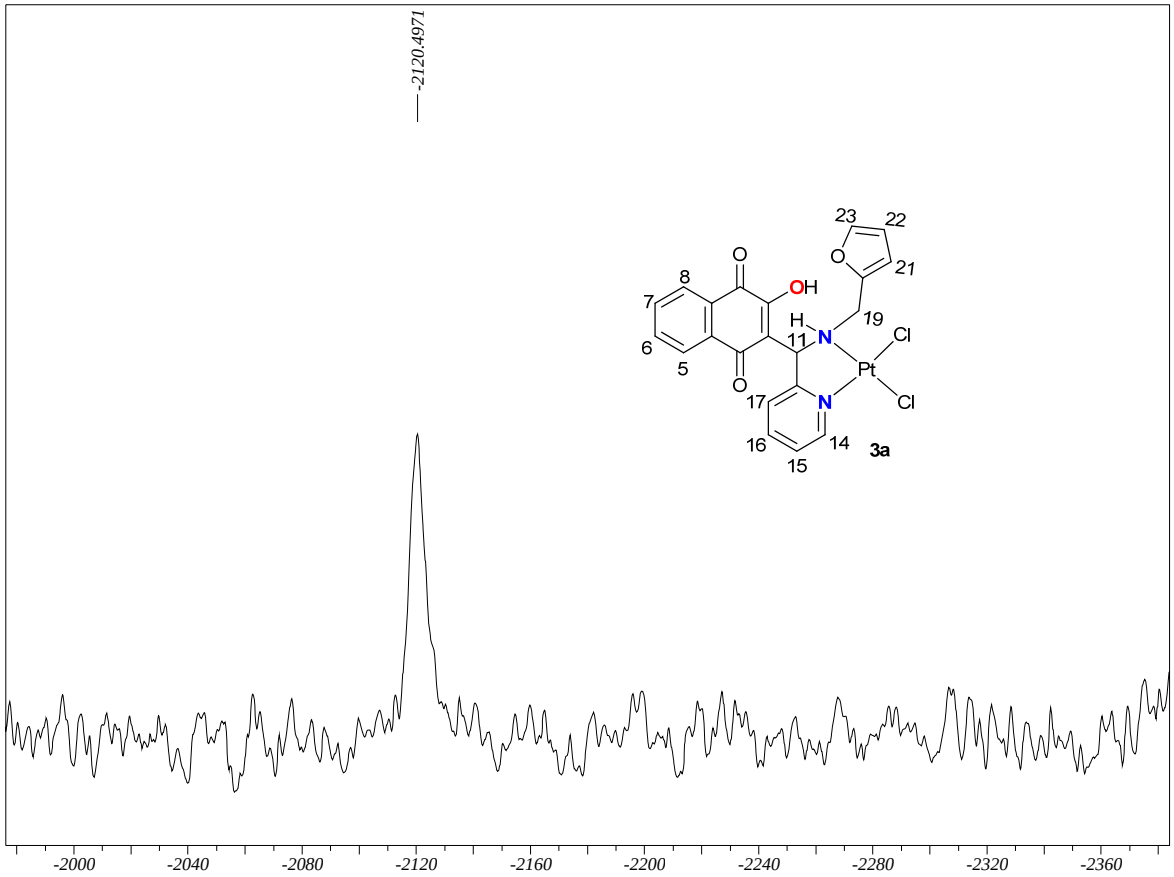

Figure S23. ^{195}Pt NMR spectrum of **3a** in DMF-d^{7}.

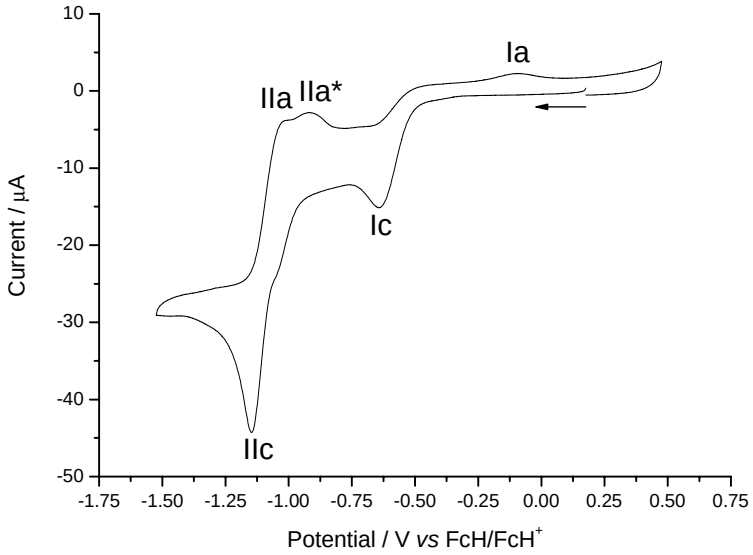

Figure S24. Cyclic voltammogram of **3a** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

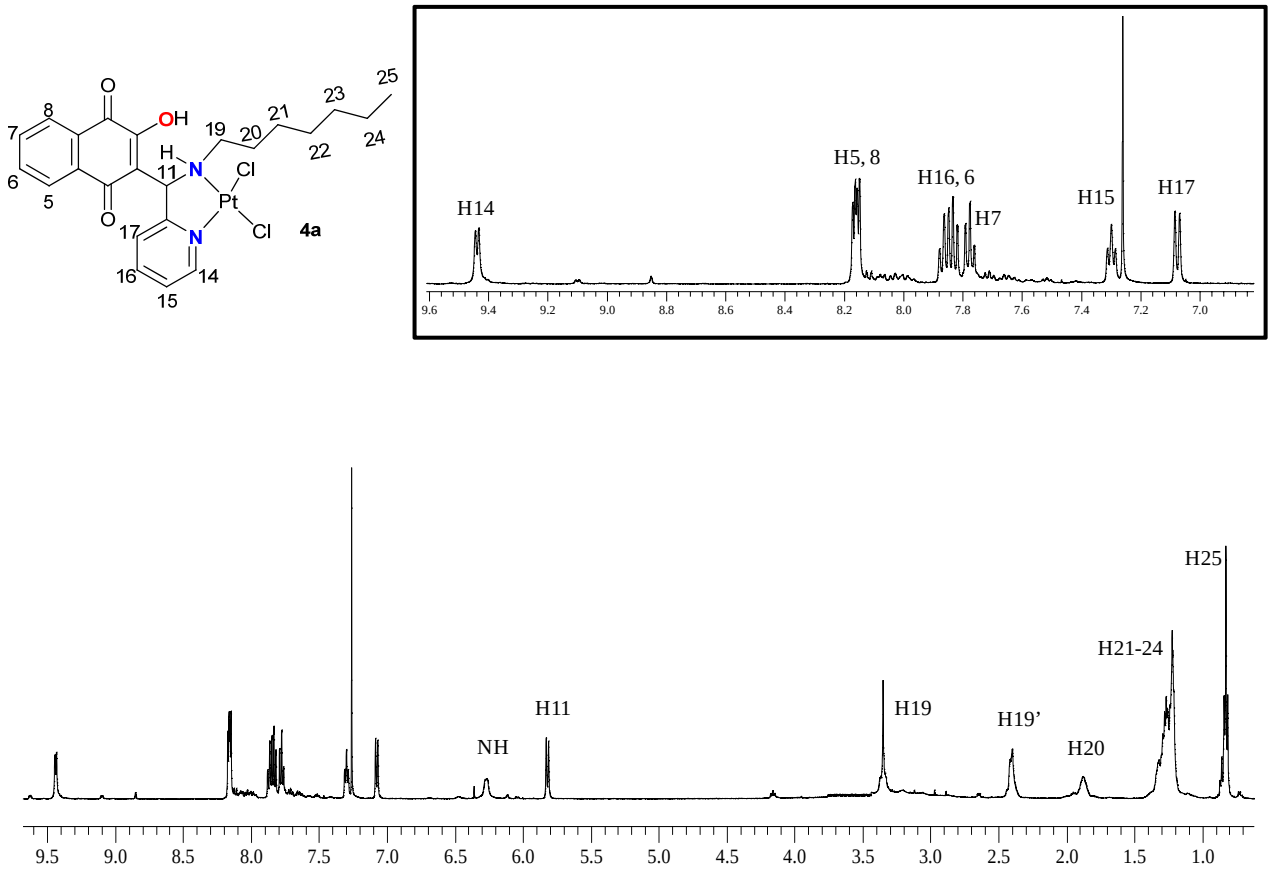

Figure S25. ^{1}H NMR spectrum of **4a** in $CDCl_3$.

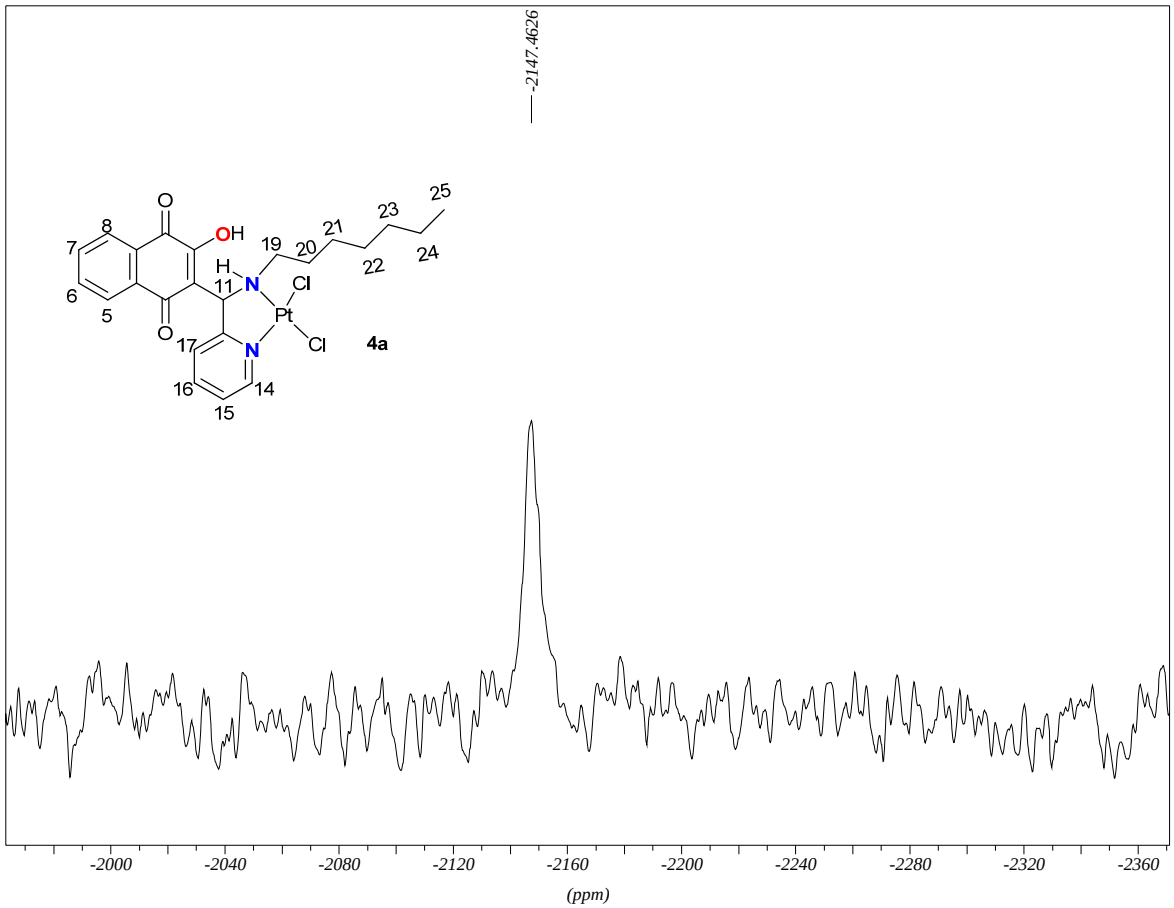

Figure S26. ^{195}Pt NMR spectrum of **4a** in $CDCl_3$.

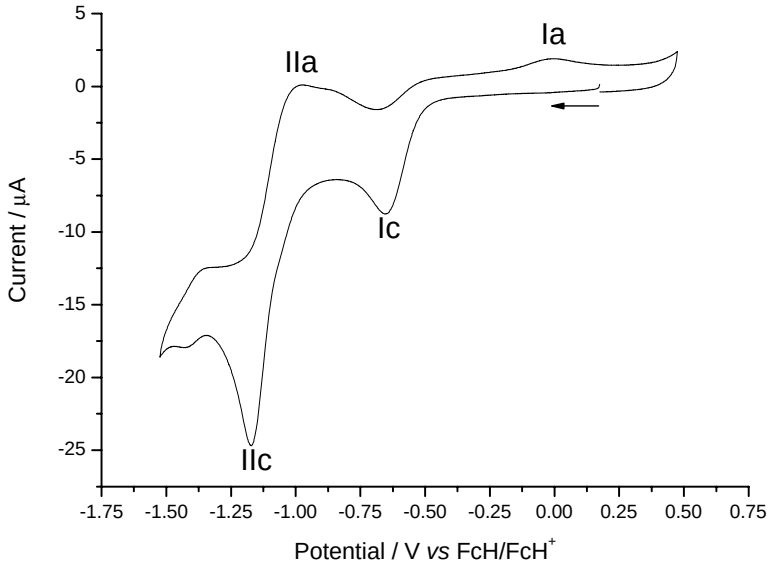

Figure S27. Cyclic voltammogram of **4a** in 0.1 $molL^{-1}$ Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

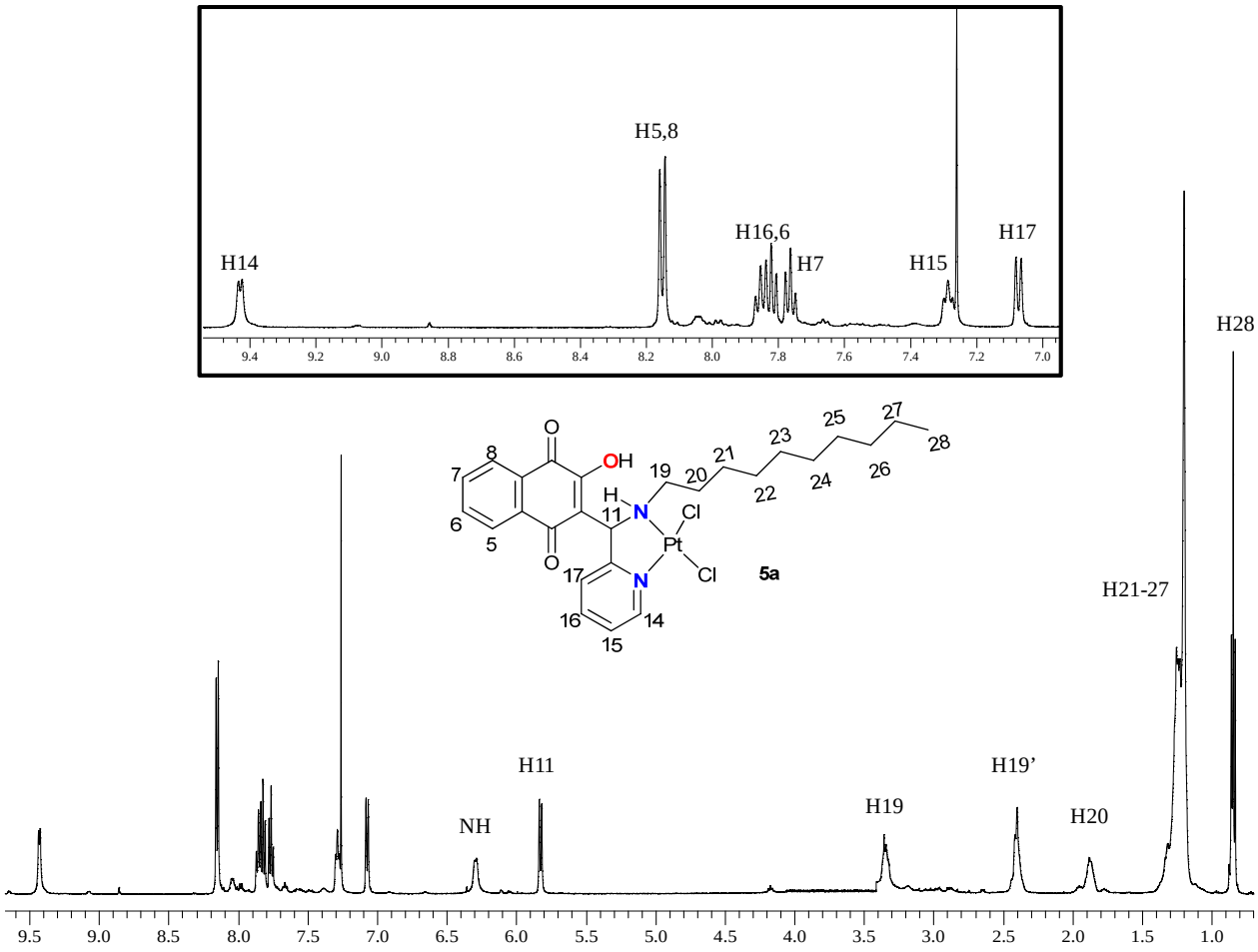

Figure S28. ^{1}H NMR spectrum of **5a** in $CDCl_3$.

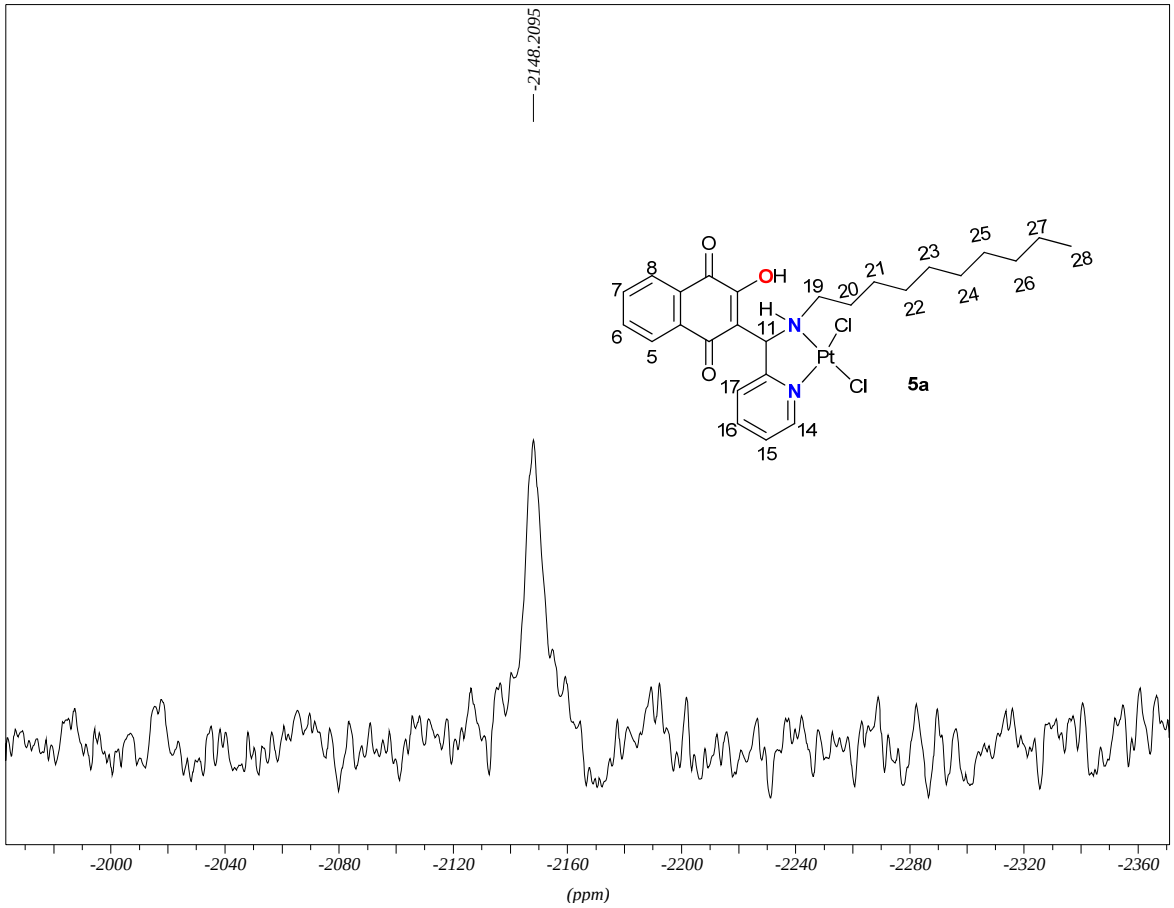

Figure S29. ^{195}Pt NMR spectrum of **5a** in $CDCl_3$.

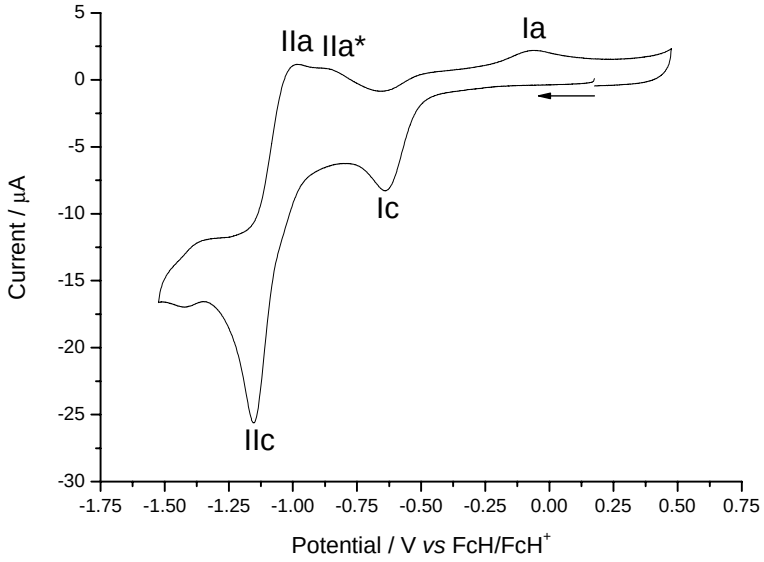

Figure S30. Cyclic voltammogram of **5a** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

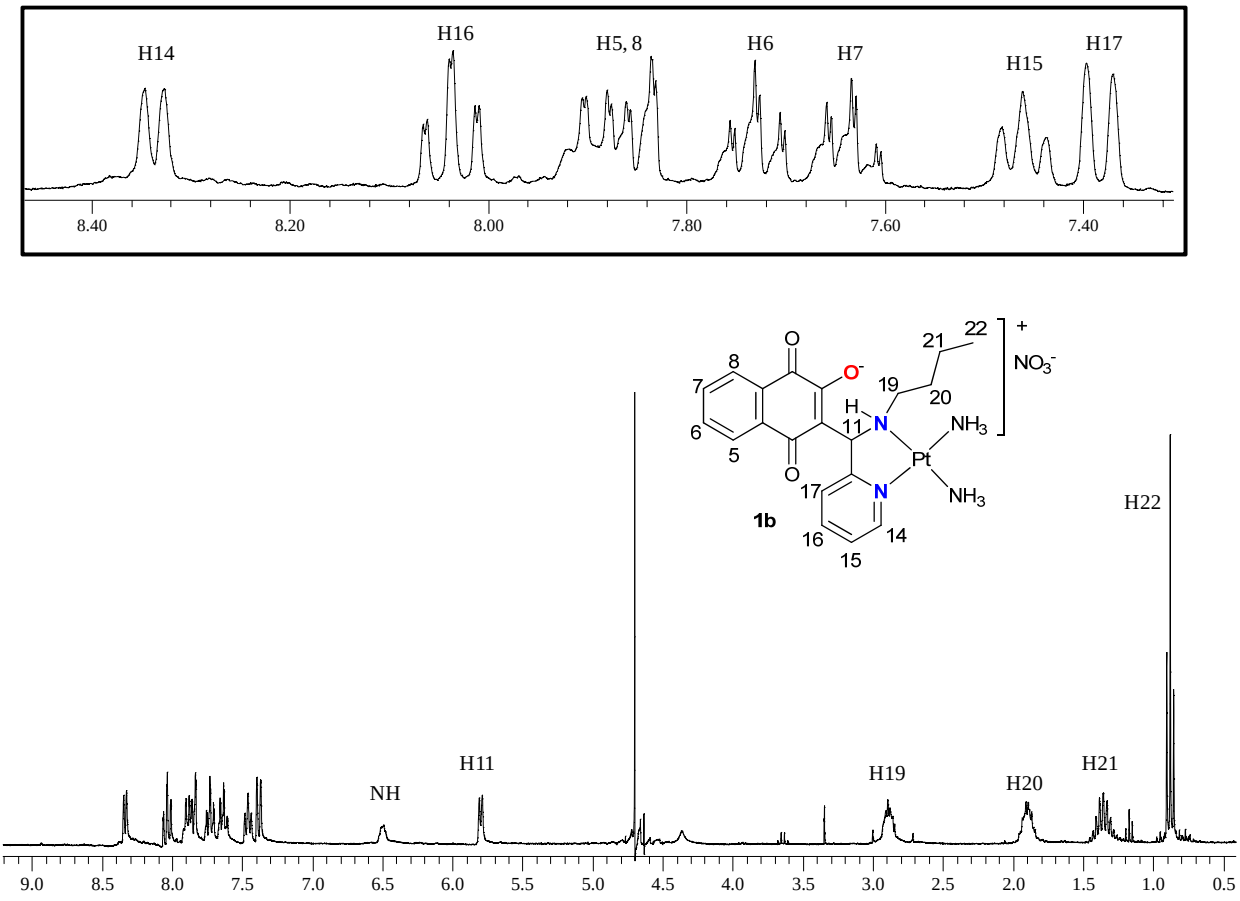

Figure S31. ^{1}H NMR spectrum of **1b** in D_2O.

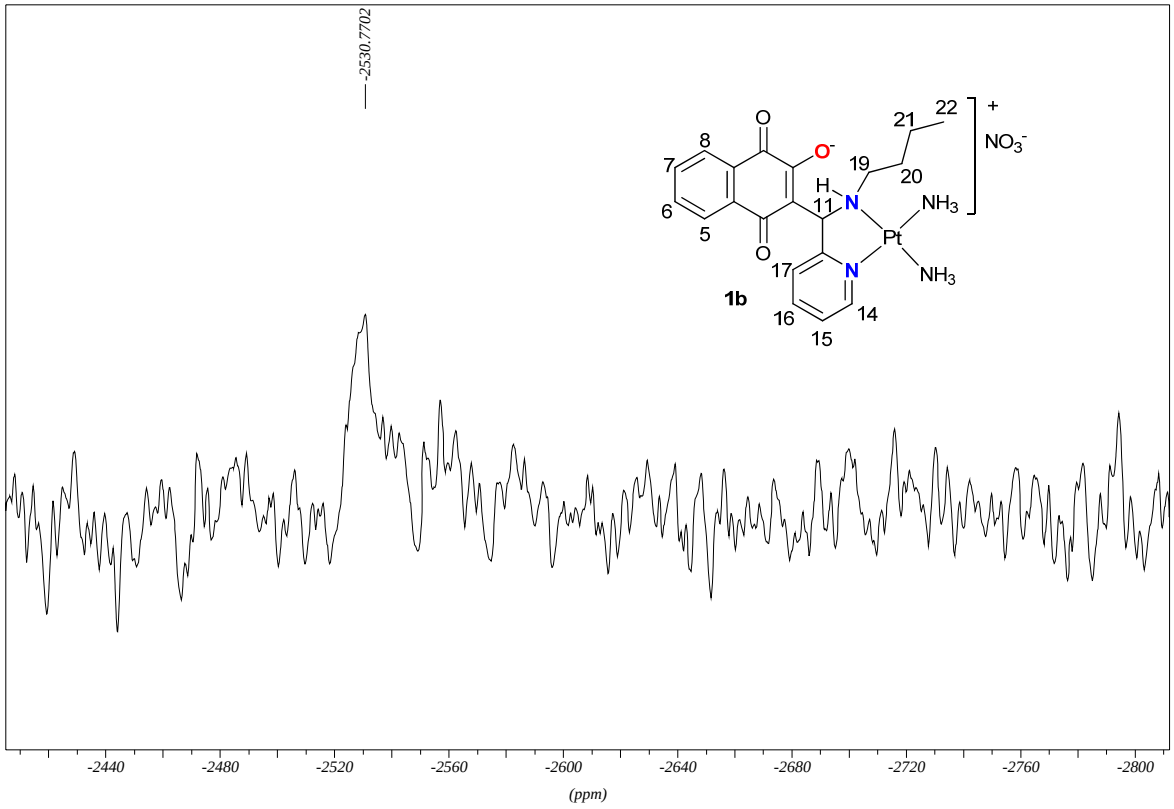

Figure S32. ^{195}Pt NMR spectrum of **1b** in D_2O.

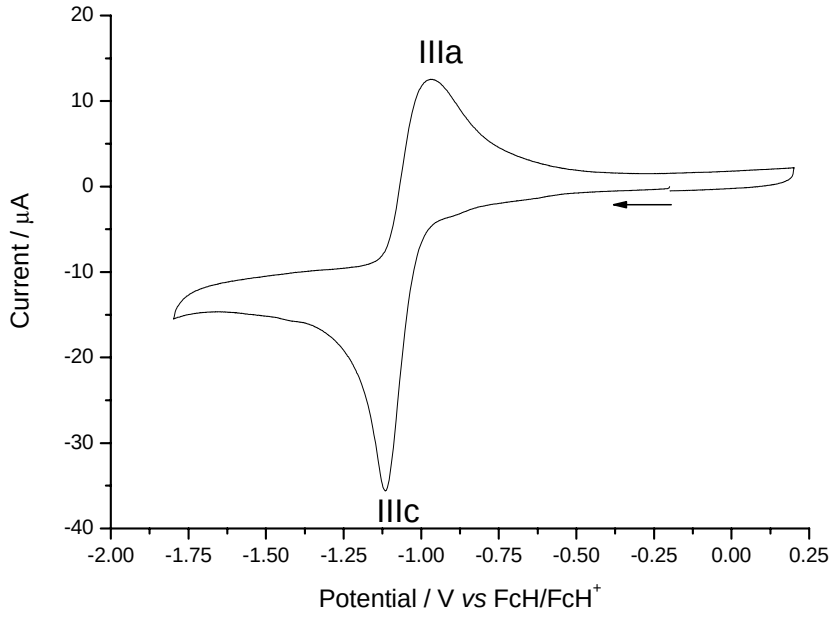

Figure S33. Cyclic voltammogram of **1b** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

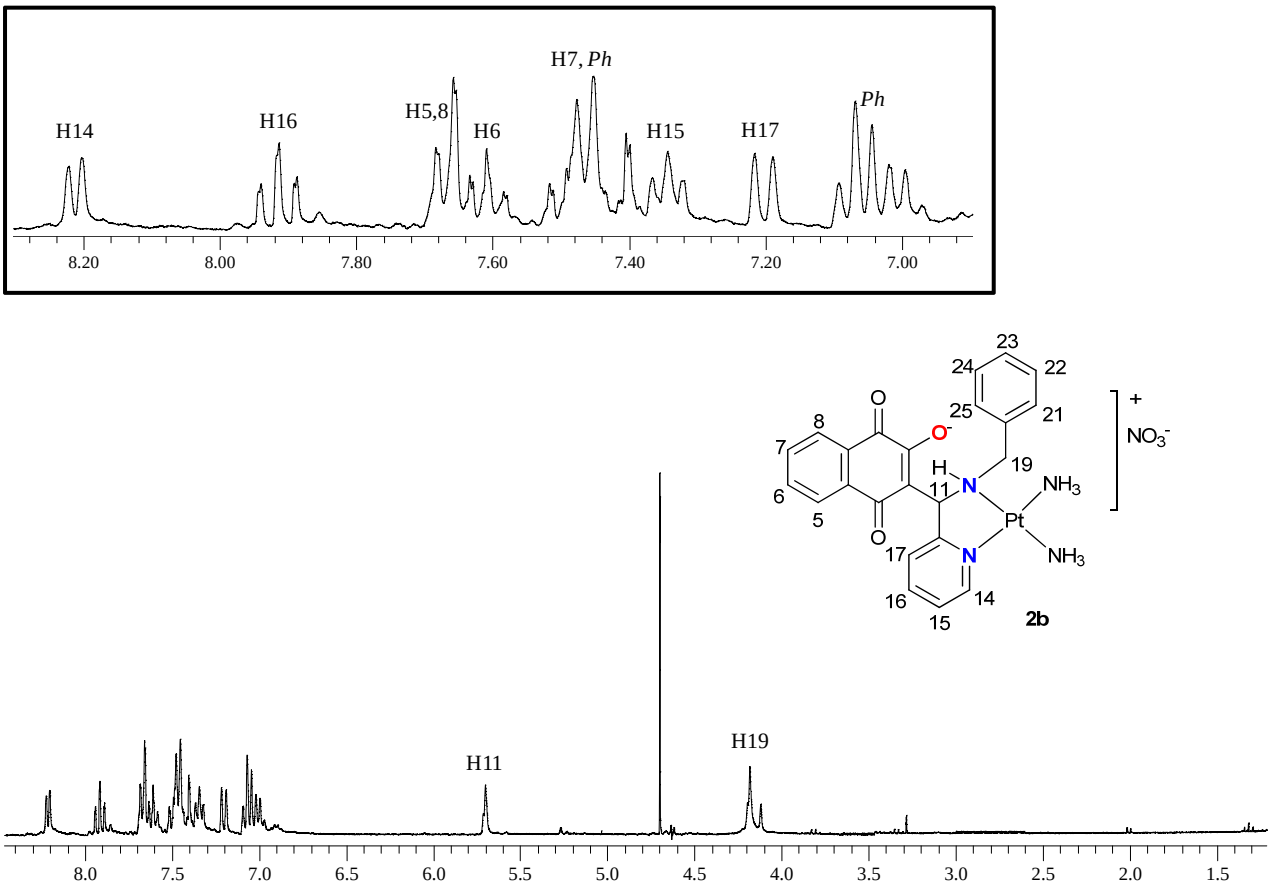

Figure S34. ^{1}H NMR spectrum of **2b** in D_2O.

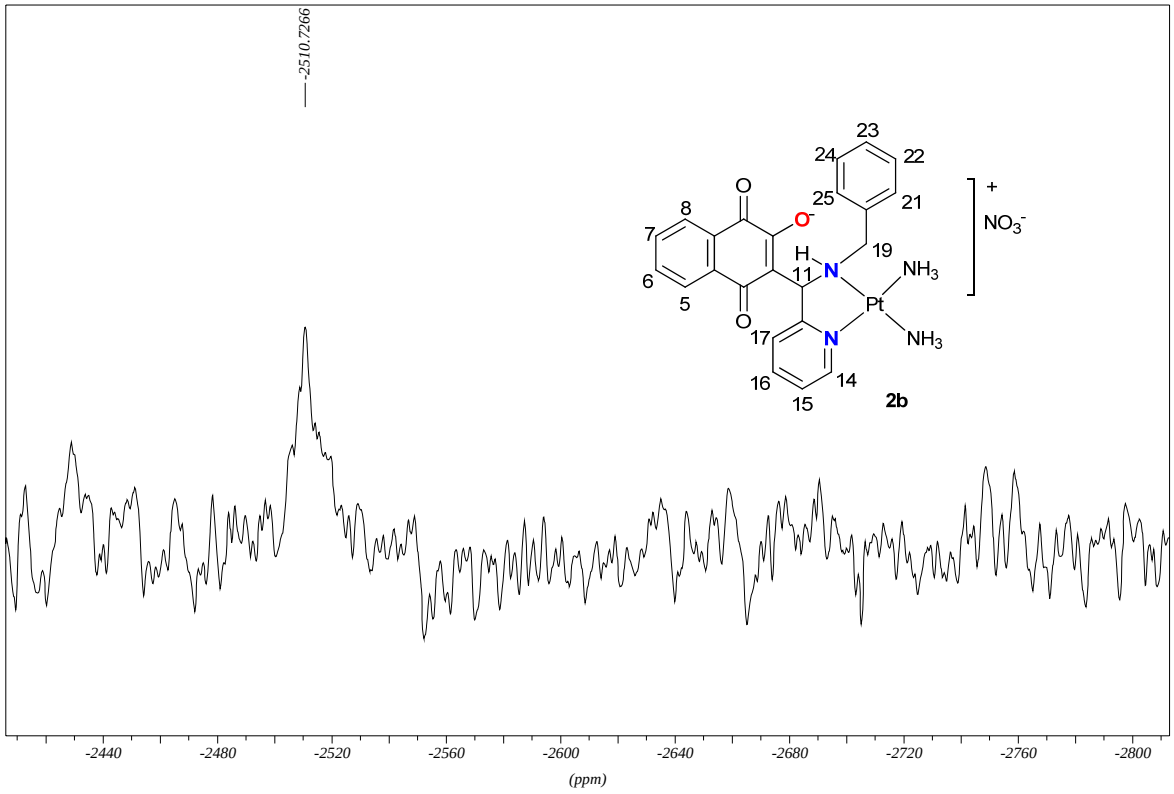

Figure S35. ^{195}Pt NMR spectrum of **2b** in D_2O.

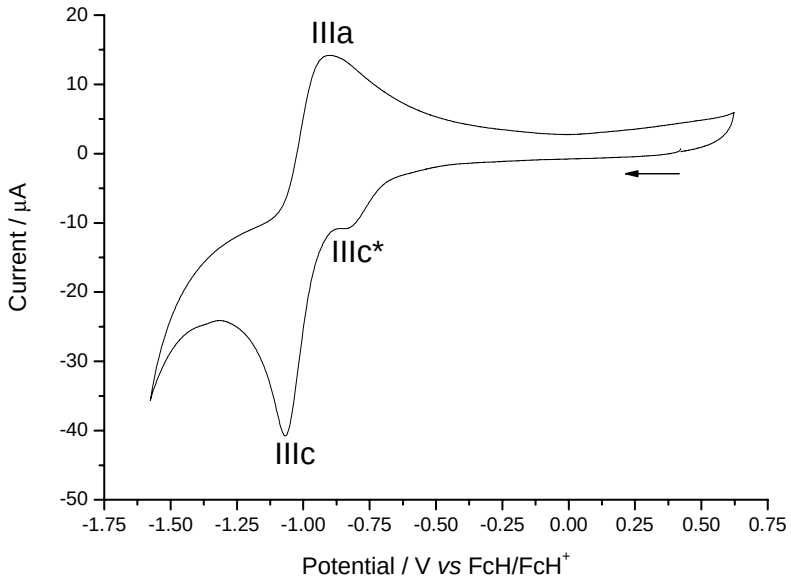

Figure S36. Cyclic voltammogram of **2b** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

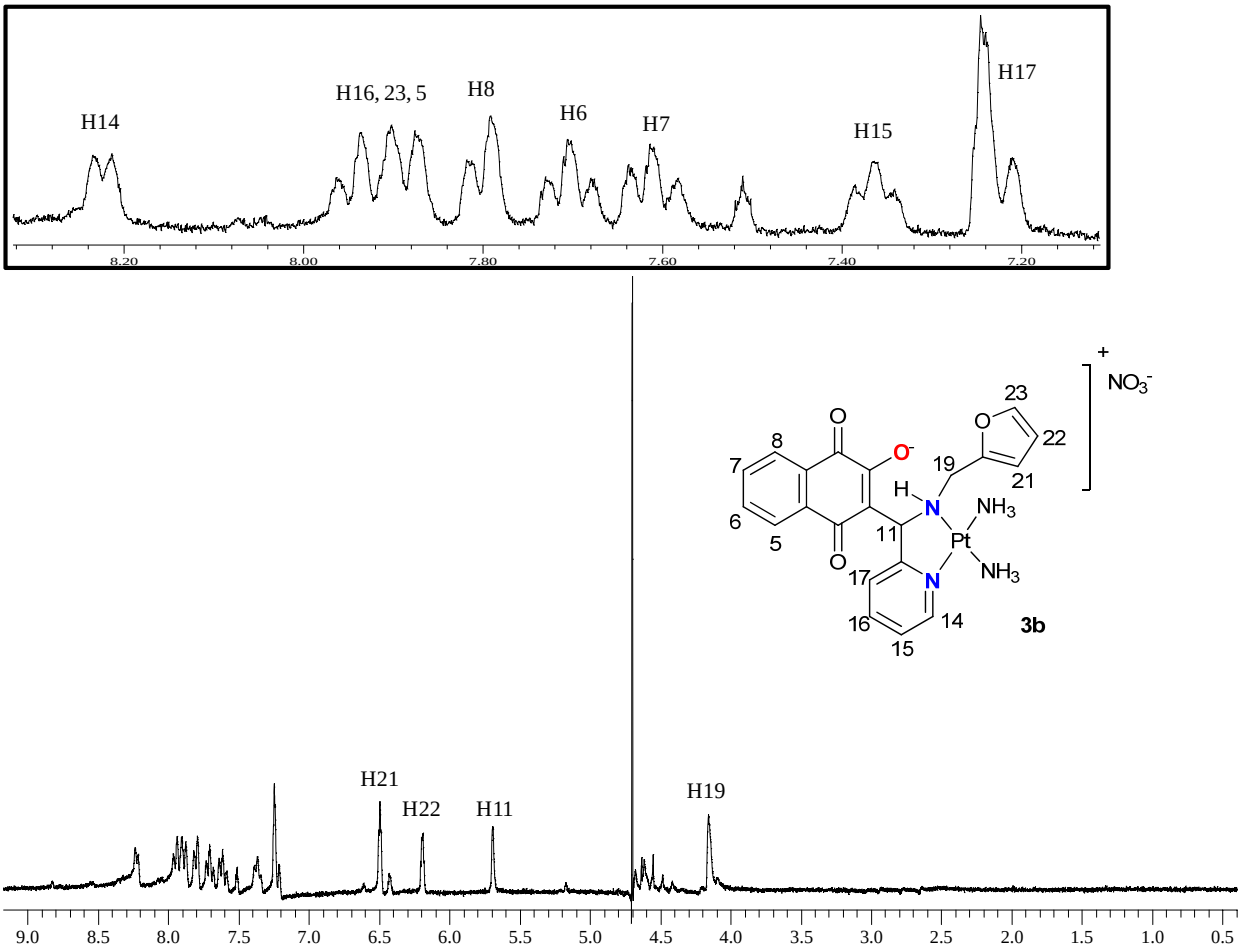

Figure S37. ^{1}H NMR spectrum of **3b** in D_2O.

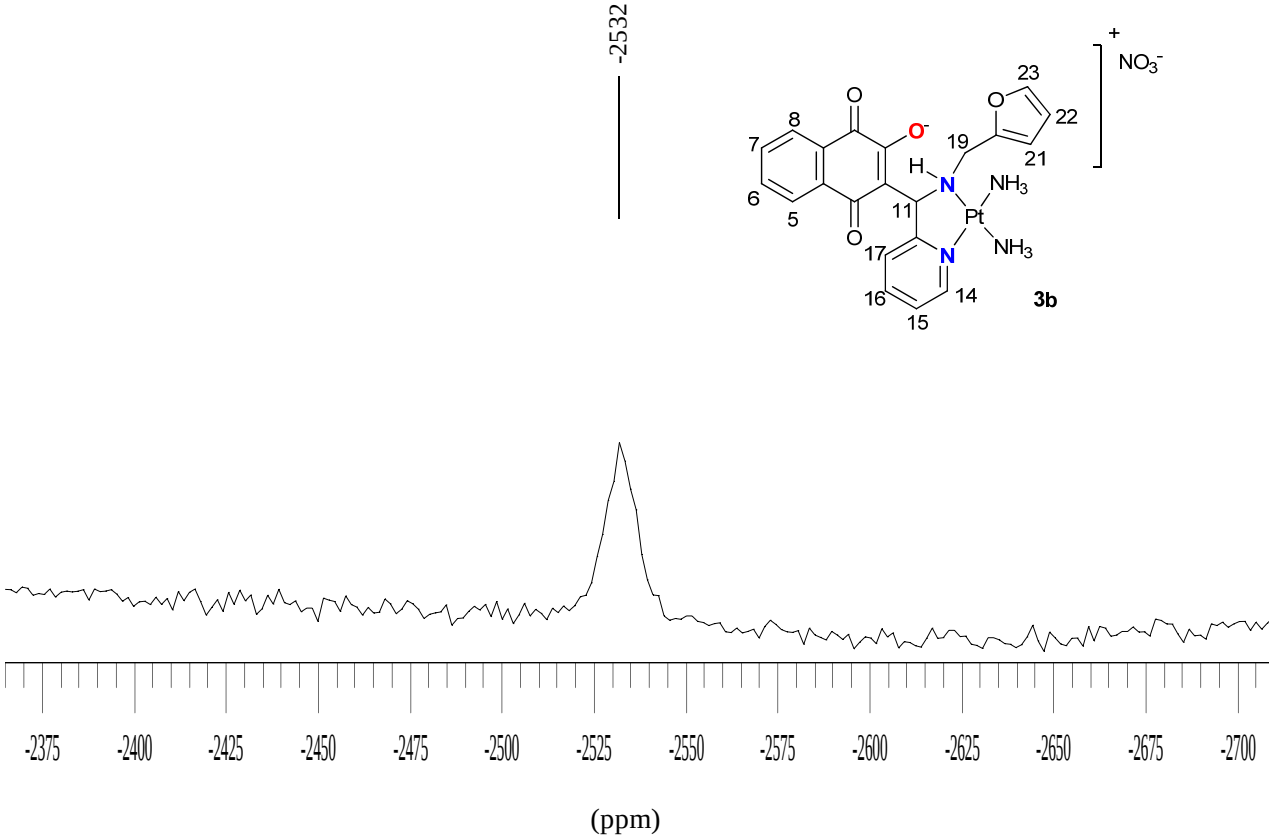

Figure S38. ^{195}Pt NMR spectrum of **3b** in DMF-d^7.

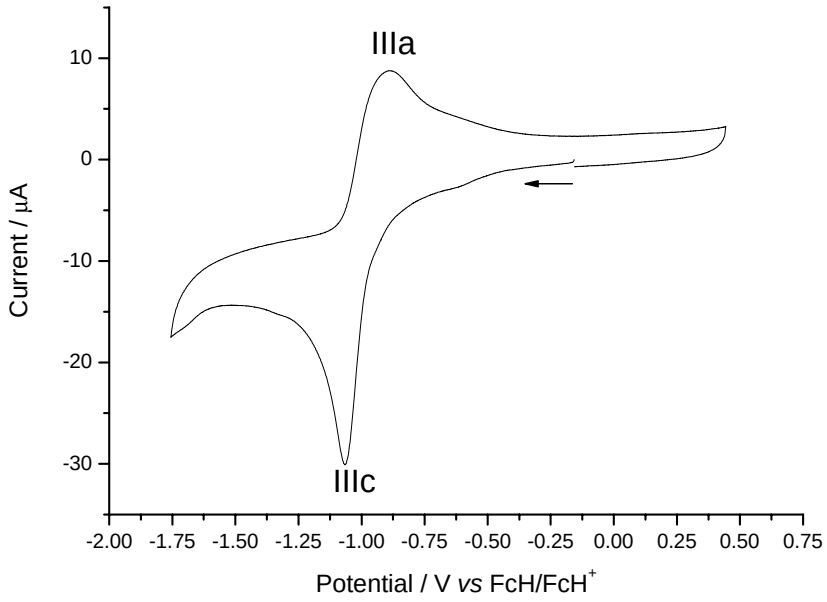

Figure S39. Cyclic voltammogram of **3b** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

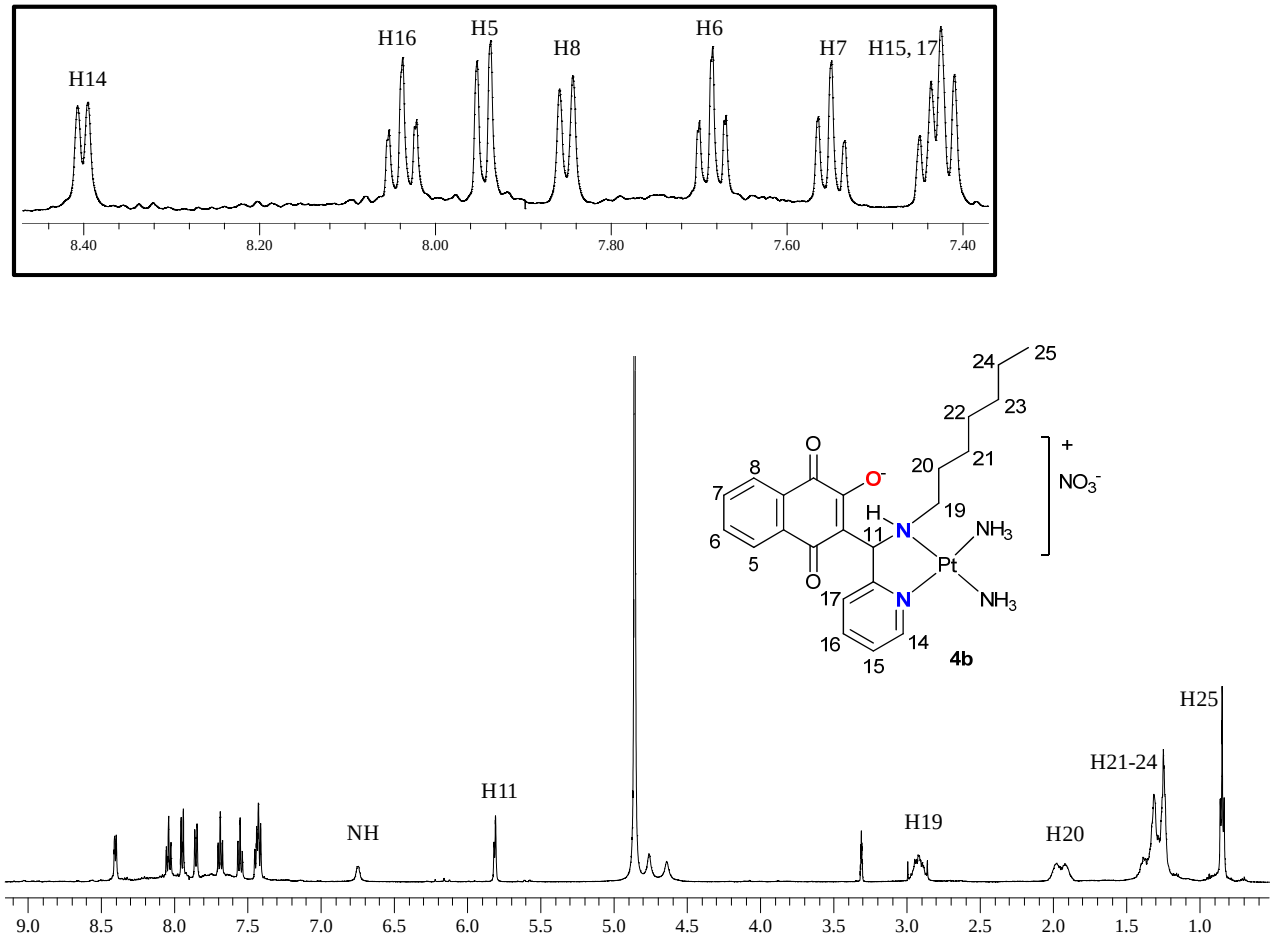

Figure S40. ^{1}H NMR spectrum of **4b** in CD_3OD.

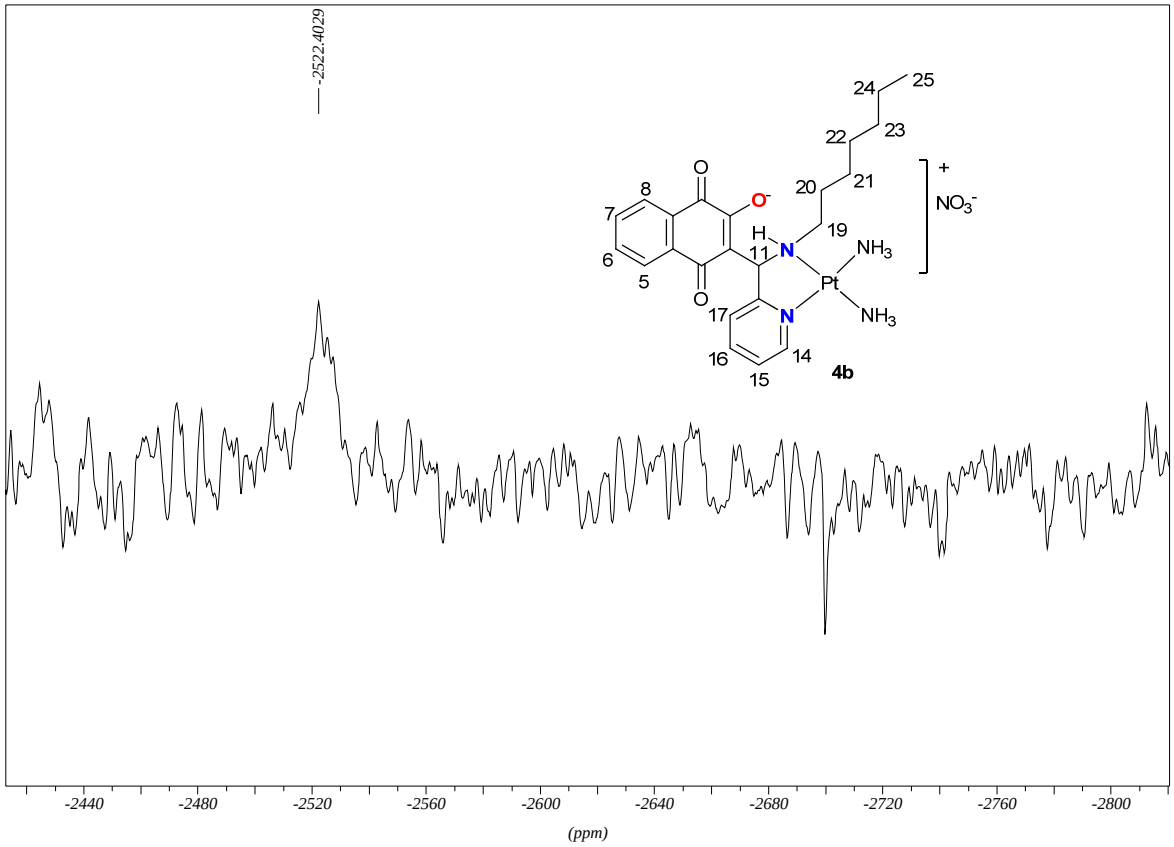

Figure S41. ^{195}Pt NMR spectrum of **4b** in DMF-d^{7}.

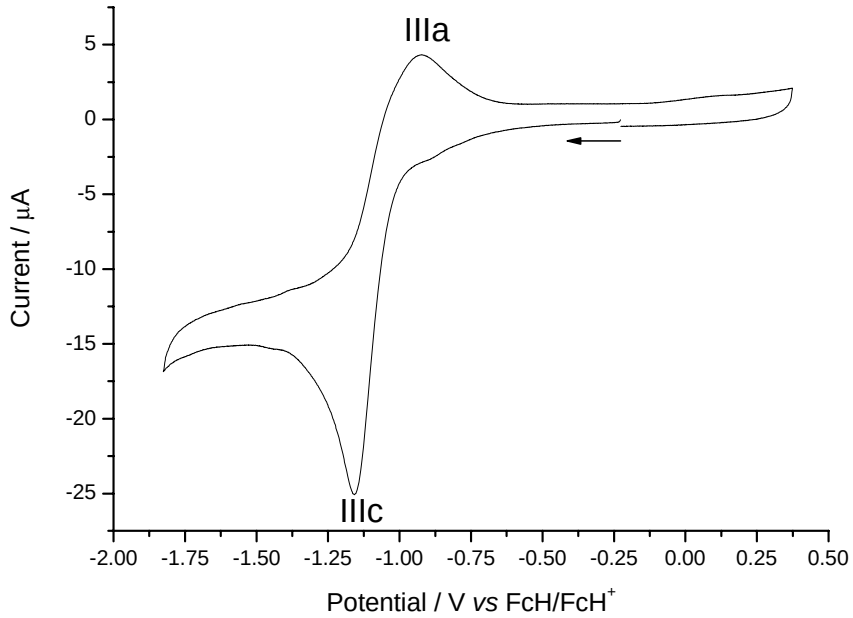

Figure S42. Cyclic voltammogram of **4b** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.

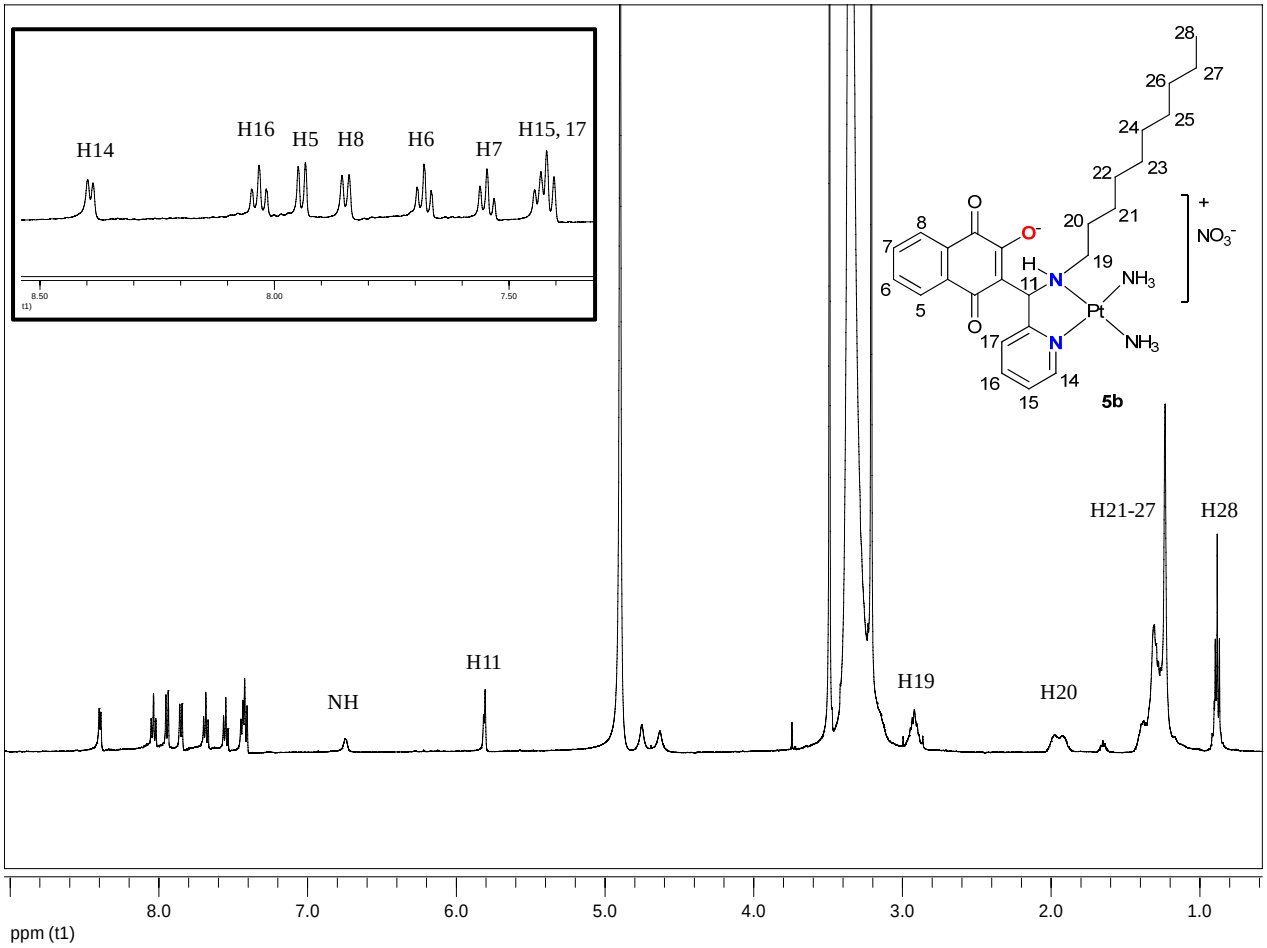

Figure S43. ^{1}H NMR spectrum of **5b** in CD_3OD.

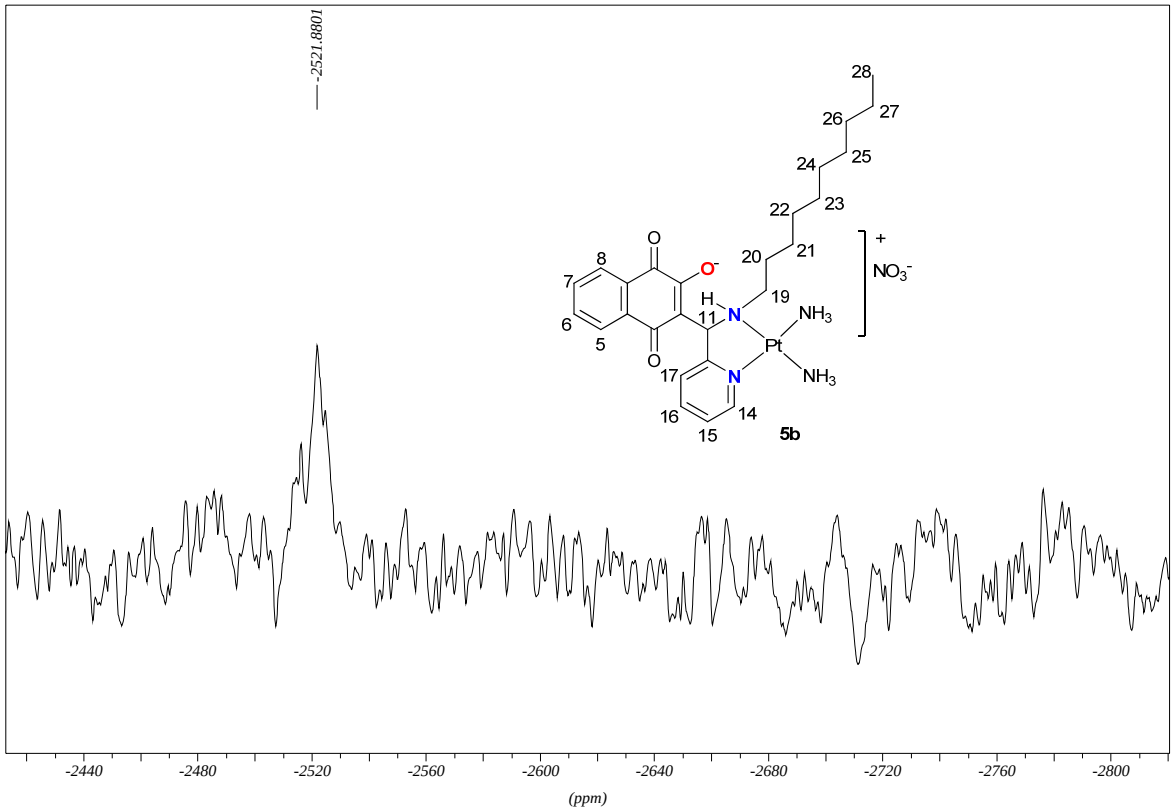

Figure S44. ^{195}Pt NMR spectrum of **5b** in DMF-d^7.

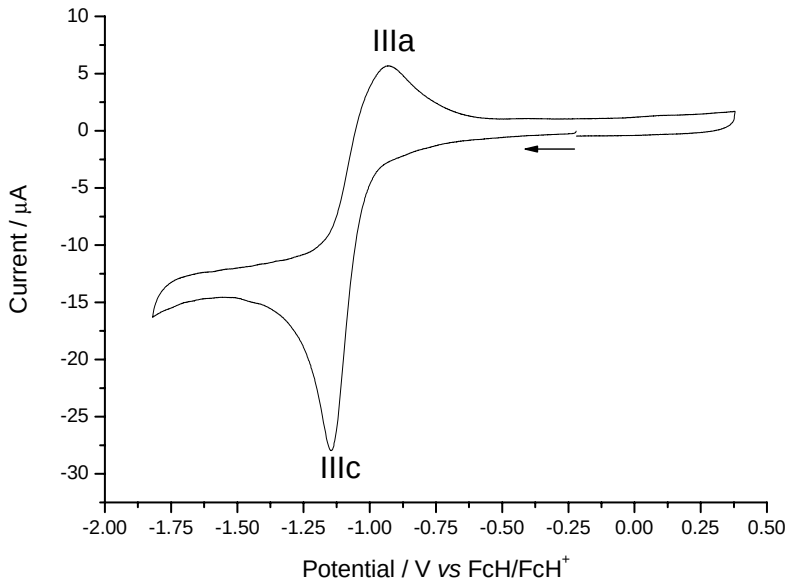

Figure S45. Cyclic voltammogram of **5b** in 0.1 molL^{-1} Bu_4ClO_4/CH_3OH obtained at 0.1Vs^{-1} with a glassy carbon electrode, the potentials being referred to the ferrocene/ferrocenium (FcH/FcH^+) pair internal standard.