

Supporting Information for

Selective heterolytic P-P bond cleavage of white phosphorus by frustrated carbene-borane Lewis pairs

**Dirk Holschumacher, Thomas Bannenberg, Kerstin Ibrom, Constantin G.
Daniliuc, Peter G. Jones and Matthias Tamm***

* *Institut für Anorganische und Analytische Chemie, Technische Universität
Carolo-Wilhelmina, Hagenring 30, D-38106 Braunschweig, Germany.*

Fax: +49 (251) 391-5309; Tel: +49 (251) 391-5387; E-mail: m.tamm@tu-bs.de

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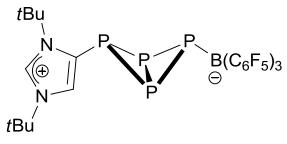
Energies of all optimized structures

Structures and XYZ coordinates

1. Experimental Details

Materials and Methods: All operations with air and moisture-sensitive compounds were performed in a glove box under a dry argon atmosphere (MBraun 200B) or on a high vacuum line using Schlenk techniques. All solvents were purified by a solvent purification system from MBraun and stored over molecular sieve (4 Å) prior to use. Elemental analysis (C, H, N) succeeded by combustion MS on a Finnigan MAT 95 (EI), Finnigan MAT 95 XL (ESI), respectively and HR-MS from Bruker-Demo QTOF micro. Unless otherwise indicated, all starting materials were obtained from Aldrich and were used without further purification. 1,3-Di-*tert*-butylimidazolin-2-ylidene (**1**)^[1] and B(C₆F₅)₃^[2] were prepared according to literature procedures.

- ^[1] A. J. Arduengo III., *US Patent 5 077 414*, **1991**, b) A. J. Arduengo, III; H. Bock, H. Chen, M. Denk, D. A. Dixon, J. C. Green, W. A. Herrmann, N. L. Jones, M. Wagner, R. West *J. Am. Chem. Soc.* **1994**, *116*, 6641-6649
- ^[2] C. Wang, G. Erker, G. Kehr, K. Wedeking, R. Fröhlich, *Organometallics* **2005**, *24*, 4760-4773



(3): To 20 ml of a solution of P_4 (0.98 mmol, 121 mg) in toluene/benzene (5:1) were added $B(C_6F_5)_3$ (0.98 mmol, 0.500 g) and 1,3-Di-*tert*-butylimidazolin-2-ylidene **1** (0.98 mmol 0.177 g). The solution was mixed at ca. 5 °C and stirred at rt for 4 hours. During this time the colour of the solution changed from yellow to purple and a smooth pale yellow precipitate was formed. The solvent was decanted carefully and the solid was washed 4 times with 3 ml toluene and following 3 times with 5 ml pentane. The crude product was dried in vacuo and recrystallized from THF/pentane at -35 °C. The product was collected as a white solid after filtration and drying in vacuo. Yield 523 mg (67 %, based on elemental analysis). Crystals for **3** were obtained from a mixture of THF/pentane at -35 °C.

elemental analysis: theoretical for $C_{29}H_{20}BF_{15}N_2P_4 \cdot 2$ THF: N 2.92, C 46.28, H 3.78, found: N 2.99, C 45.76, H 3.65.

2. NMR Characterisation

The NMR spectra were obtained on Bruker Avance II 600, DRX 400 and Avance II 300 instruments. Frequencies and referencing methods are listed in tables 1-5. For referencing the ^{19}F , ^{31}P and ^{11}B spectra the observation frequencies of a dilute solution of the reference compound in THF-d₈ (^{19}F , virtual internal referencing) or the neat external standard (^{31}P , ^{11}B , virtual external referencing) had been determined earlier. The β -deuterium atoms of THF-d₈ were always used as the lock signal. Standard two-dimensional techniques (HSQC, HMBC) as well as ^{13}C triple resonances spectra (selective decoupling for all three ^{31}P signals under ^1H broadband decoupling conditions) were used to assign the spectra.

Table 1 ^1H -NMR spectrum of **3**, 600.1 MHz, THF-d₈, $\delta(\text{TMS}) = 0$

δ [ppm]	multiplicity	no. of H	J [Hz]	assignment
8.45	d	1	$J_{1\text{-H},3\text{-H}} = 1.9$	1-H
7.47	d	1		3-H
1.84	s	9		$\text{C}(\text{CH}_3)_3$ at N-1
1.54	s	9		$\text{C}(\text{CH}_3)_3$ at N-2

Table 2 ^{13}C -NMR spectrum of **3**, 150.9 MHz, THF-d₈, $\delta(\text{THF-d}_8) = 25.3$

δ [ppm]	Type of C	multiplicity ¹	J [Hz] ¹	assignment
148.80	C_q	“dm”	Splitting: 240.5	$\text{B}(\text{C}_6\text{F}_5)_3$: C_{ortho}
≈ 139.2	C_q	“dt”	Splittings: 246.4, ≥ 13.7	$\text{B}(\text{C}_6\text{F}_5)_3$: C_{para}
≈ 137.6	C_q	“ddd”	Splittings: 246.7, $\geq 20.9, \geq 11.3$	$\text{B}(\text{C}_6\text{F}_5)_3$: C_{meta}
≈ 136.3	C_q	dd	$J_{\text{P-1,C}} \approx 106, J_{\text{P-4,C}} \approx 44^*$	C-2
133.98	CH	s		C-1
127.72	CH	br. s	$J_{\text{P-2,C}} = 8.9^{**}$	C-3
≈ 123.4	C_q	v br. m	($\nu_{1/2} \approx 155$ Hz)	$\text{B}(\text{C}_6\text{F}_5)_3$: C_{ipso}
63.30	C_q	s		$\text{C}(\text{CH}_3)_3$ at N-1
61.25	C_q	s		$\text{C}(\text{CH}_3)_3$ at N-2
30.33	q	d	$J_{\text{P-2,C}} = 9.3^2$	$\text{C}(\text{CH}_3)_3$ at N-1
29.21	q	s		$\text{C}(\text{CH}_3)_3$ at N-2

*derived from HMBC; ** derived from $^{13}\text{C}\{\text{P}_{\text{sel}}, ^1\text{H}\}$, irradiation at $\delta(^{31}\text{P}) = -200.2$ ppm

¹ Where spectra of higher order are observed merely a description of the multiplicity and splittings instead of couplings constants are given.

Table 3 ^{19}F -NMR spectrum of **3**, 376.4 MHz, THF-d₈, $\delta[(\text{CFCl}_3)_{\text{virt.,int.}}] = 0$

δ [ppm]	multiplicity	Splitting [Hz]	assignment
-126.0	“br. d”	≈ 68 ($\nu_{1/2} \approx 50$ Hz)	$\text{B}(\text{C}_6\text{F}_5)_3$: F _{ortho}
-162.6	“t”	20.4	$\text{B}(\text{C}_6\text{F}_5)_3$: F _{meta}
-166.1	“≈ t”	19.0	$\text{B}(\text{C}_6\text{F}_5)_3$: F _{para}

Table 4 ^{31}P -NMR spectrum of **3**, 121.5 MHz, THF-d₈, $\delta[(85\% \text{ H}_3\text{PO}_4)_{\text{virt., ext.}}] = 0$

δ [ppm]	multiplicity	Splitting [Hz]	assignment
-159.5	dtsept	$J_{\text{P-1,P-4}} = 285.3, J_{\text{P-2,P-4}} = 172.7, J_{\text{P-4,Fortho}} = 63.3$	P-4
-200.2	br. m	$J_{\text{P-1,P-2}} = 173.2$	P-1
-327.4	tsept	$J_{\text{P-2,Fortho}} = 12.3$	P-2

Table 5 ^{11}B -NMR spectrum of **3**, 96.3 MHz, THF-d₈, $\delta[(\text{BF}_3 \cdot \text{O}(\text{CH}_2\text{CH}_3)_2)_{\text{virt., ext.}}] = 0$

δ [ppm]	multiplicity	Splitting [Hz]
-13.9	br. d	36 ($\nu_{1/2} \approx 79$ Hz)

Please note:

The following spectra, which were recorded on a 600 MHz-NMR spectrometer show impurities due to decomposition in solution during prolonged measurements within two days.

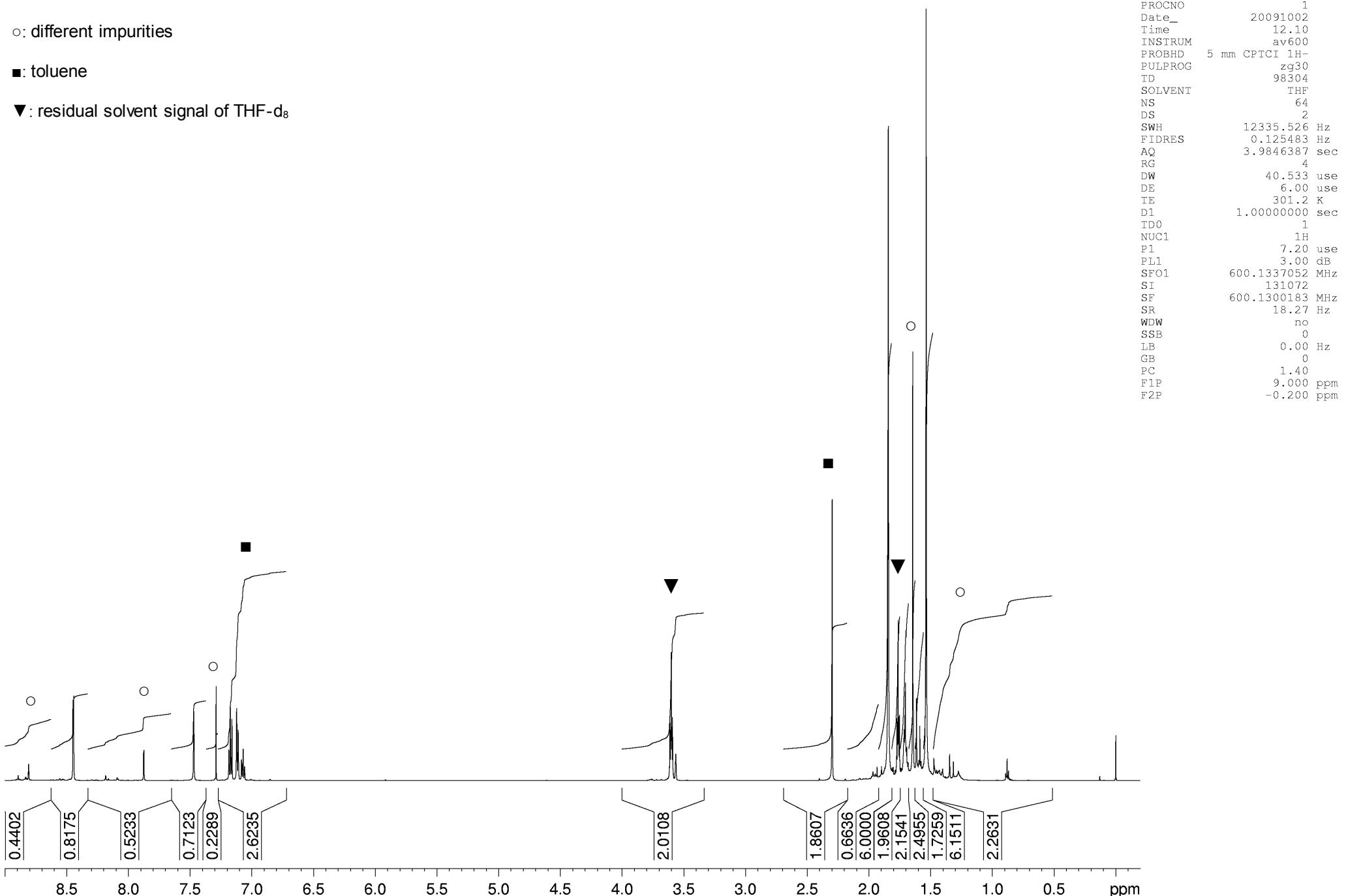
NMR spectra of freshly prepared samples measured on a 400 MHz NMR-spectrometer are also shown; these show no impurities.

¹H-NMR spectrum of **3**

○: different impurities

■: toluene

▼: residual solvent signal of THF-d₈

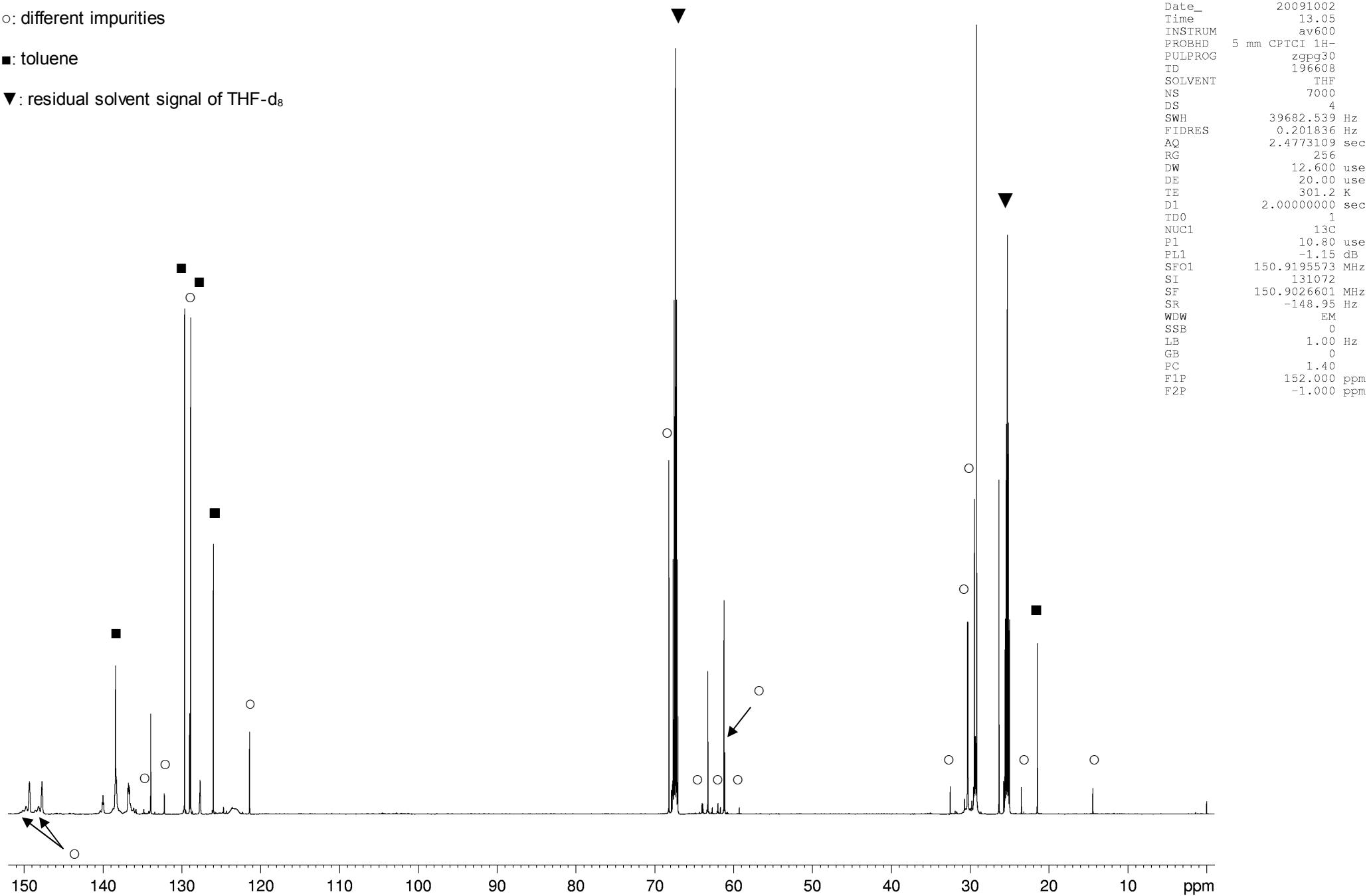


¹³C-NMR spectrum of 3

○: different impurities

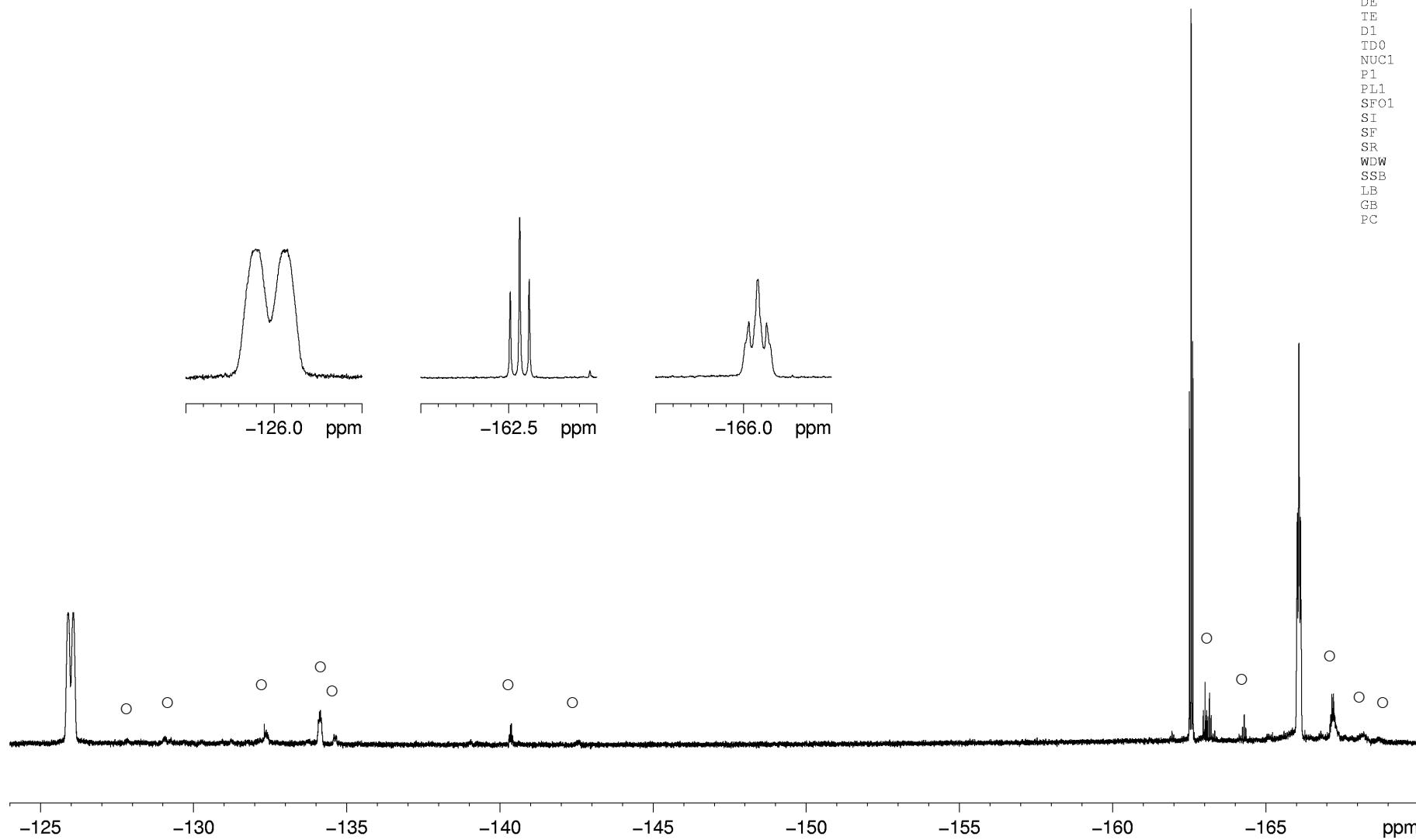
■: toluene

▼: residual solvent signal of THF-d₈



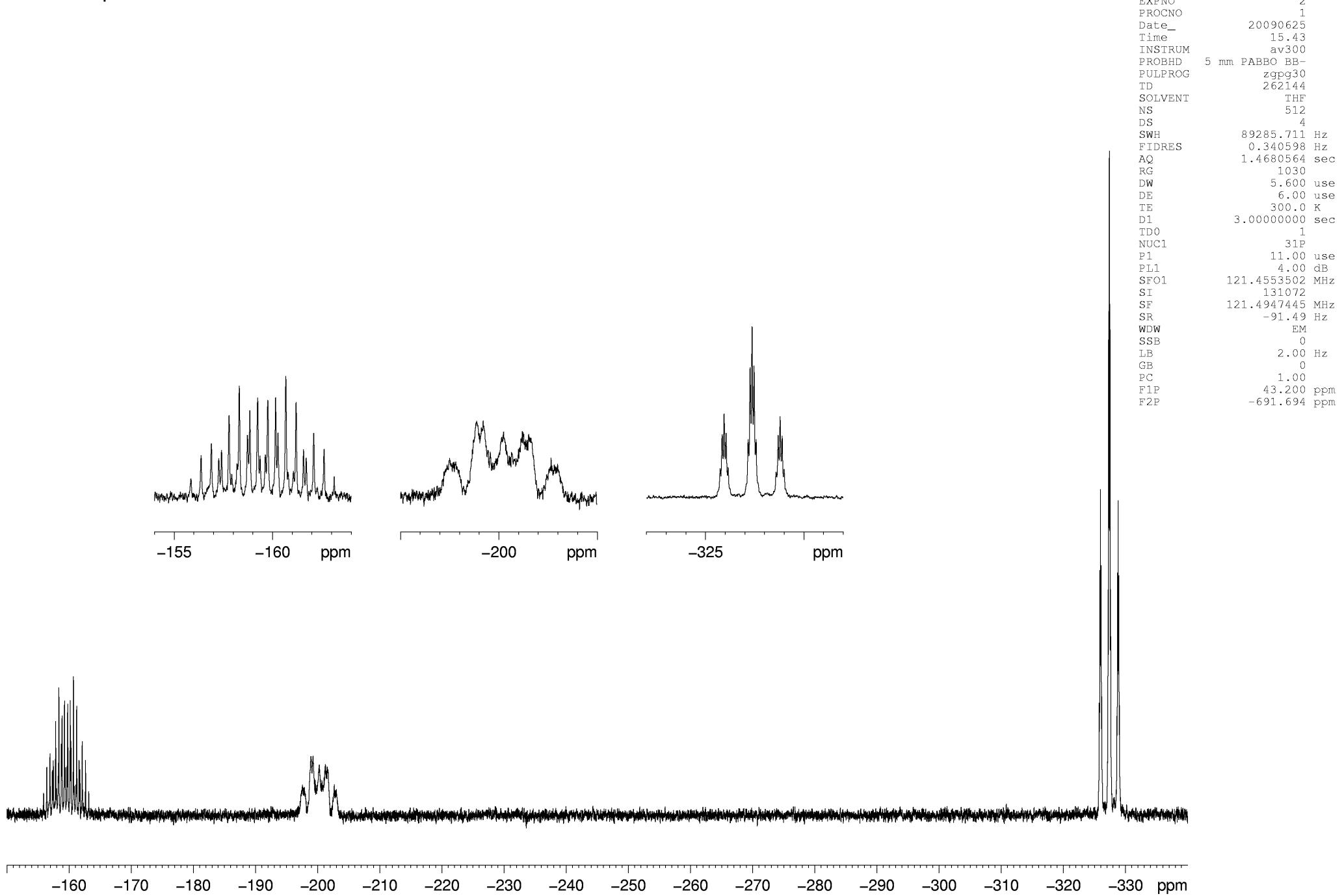
¹⁹F-NMR spectrum of 3

○: different impurities



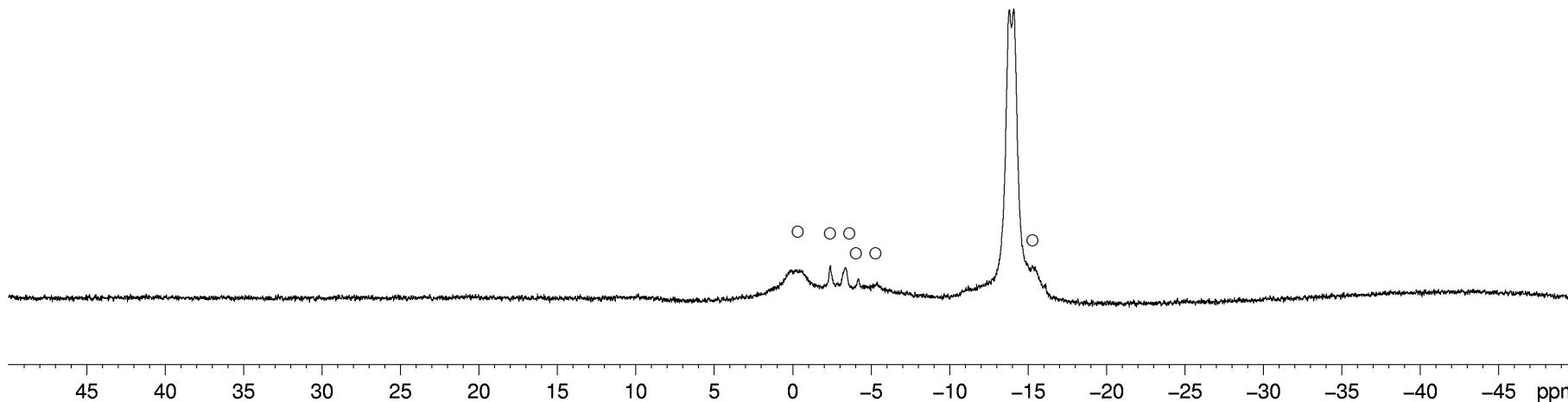
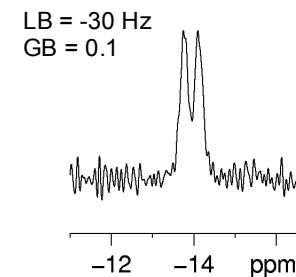
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TD 262144
SOLVENT THF
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DS 4
SWH 75187.969 Hz
FIDRES 0.286819 Hz
AQ 1.7433076 sec
RG 3251
DW 6.650 use
DE 6.00 use
TE 299.2 K
D1 2.0000000 sec
TDO 1
NUC1 ¹⁹F
P1 9.40 use
PL1 -4.00 dB
SFO1 376.4043252 MHz
SI 131072
SF 376.4415429 MHz
SR -356.11 Hz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

³¹P-NMR spectrum of **3**



¹¹B-NMR spectrum of **3**

○: different impurities and probehead background signal resp.



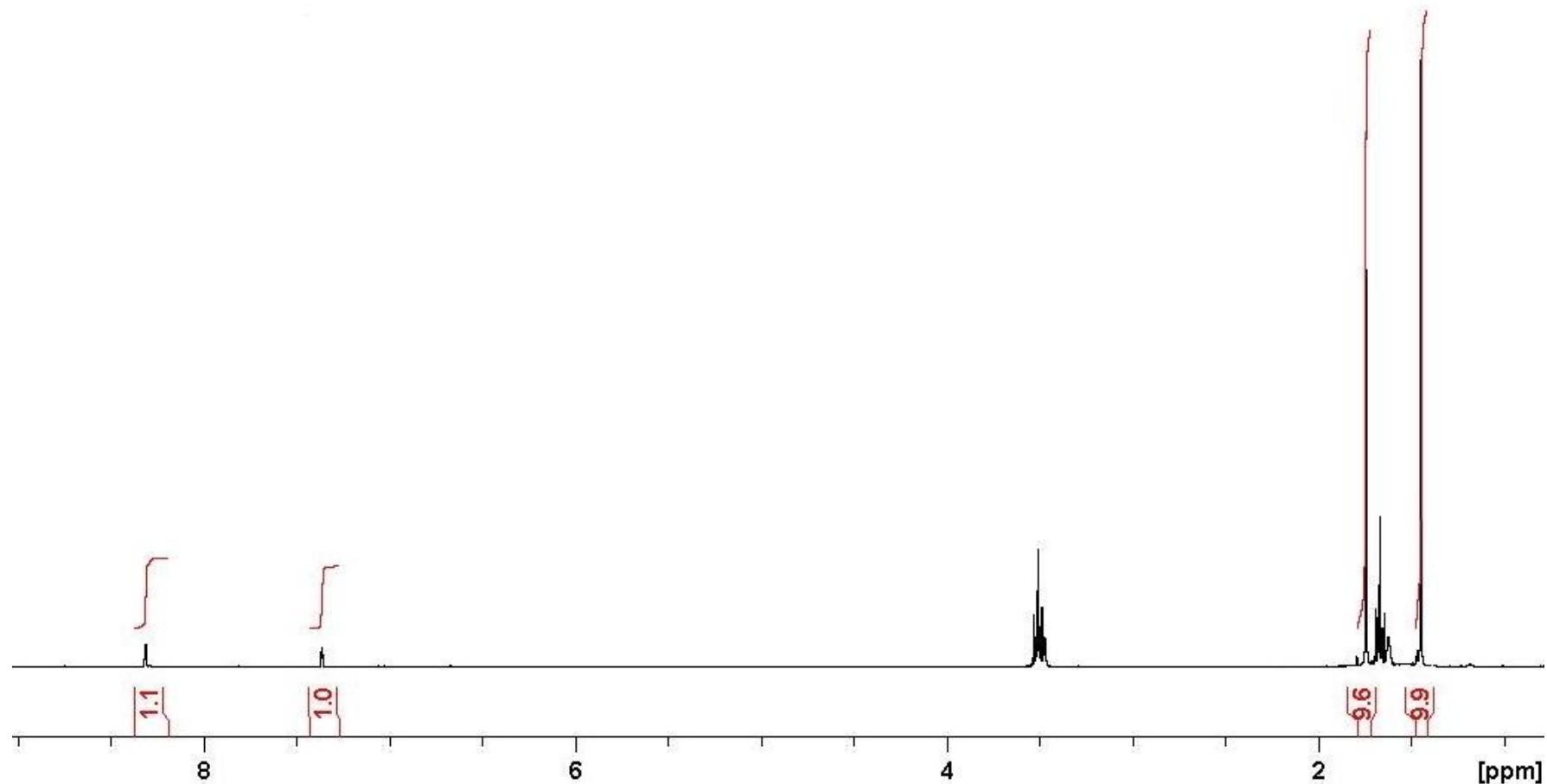
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PULPROG  zgpg
TD        32768
SOLVENT   THF
NS        512
DS        4
SWH      21008.404 Hz
FIDRES   0.641126 Hz
AQ        0.7799284 sec
RG        287
DW        23.800 use
DE        6.00 use
TE        300.0 K
D1        1.0000000 sec
TD0       1
NUC1     11B
P1        9.70 use
PL1       4.00 dB
SFO1     96.2936310 MHz
SI        16384
SF        96.2935819 MHz
SR        -49.11 Hz
WDW      EM
SSB      0
LB        1.00 Hz
GB        0
PC        1.40
F1P      50.865 ppm
F2P      -49.119 ppm

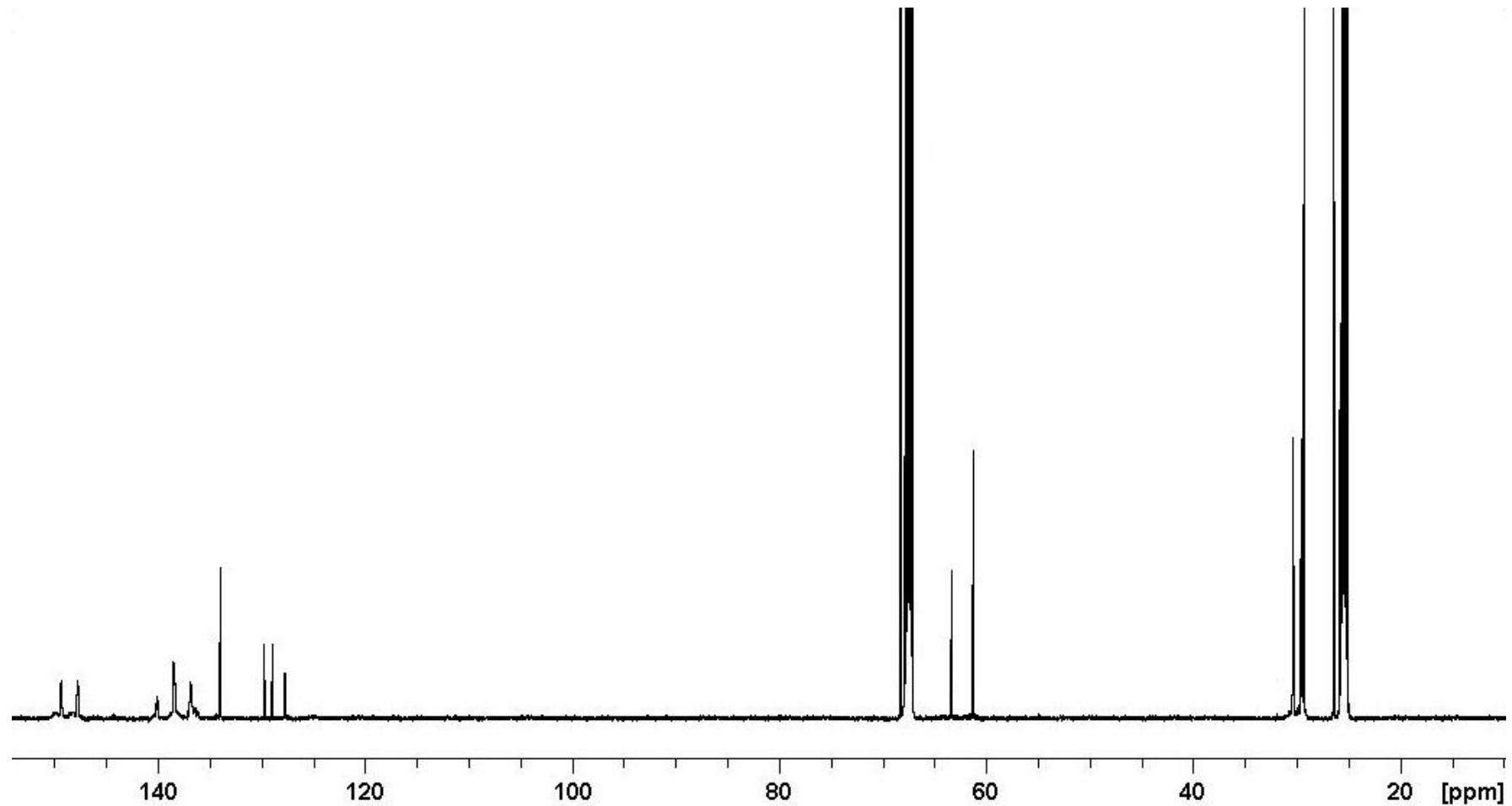
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NMR spectroscopic illustration of the compound **3**, which were directly measured after synthesis on a DRX 400 spectrometer. In solution slow decomposition takes place after some hours.

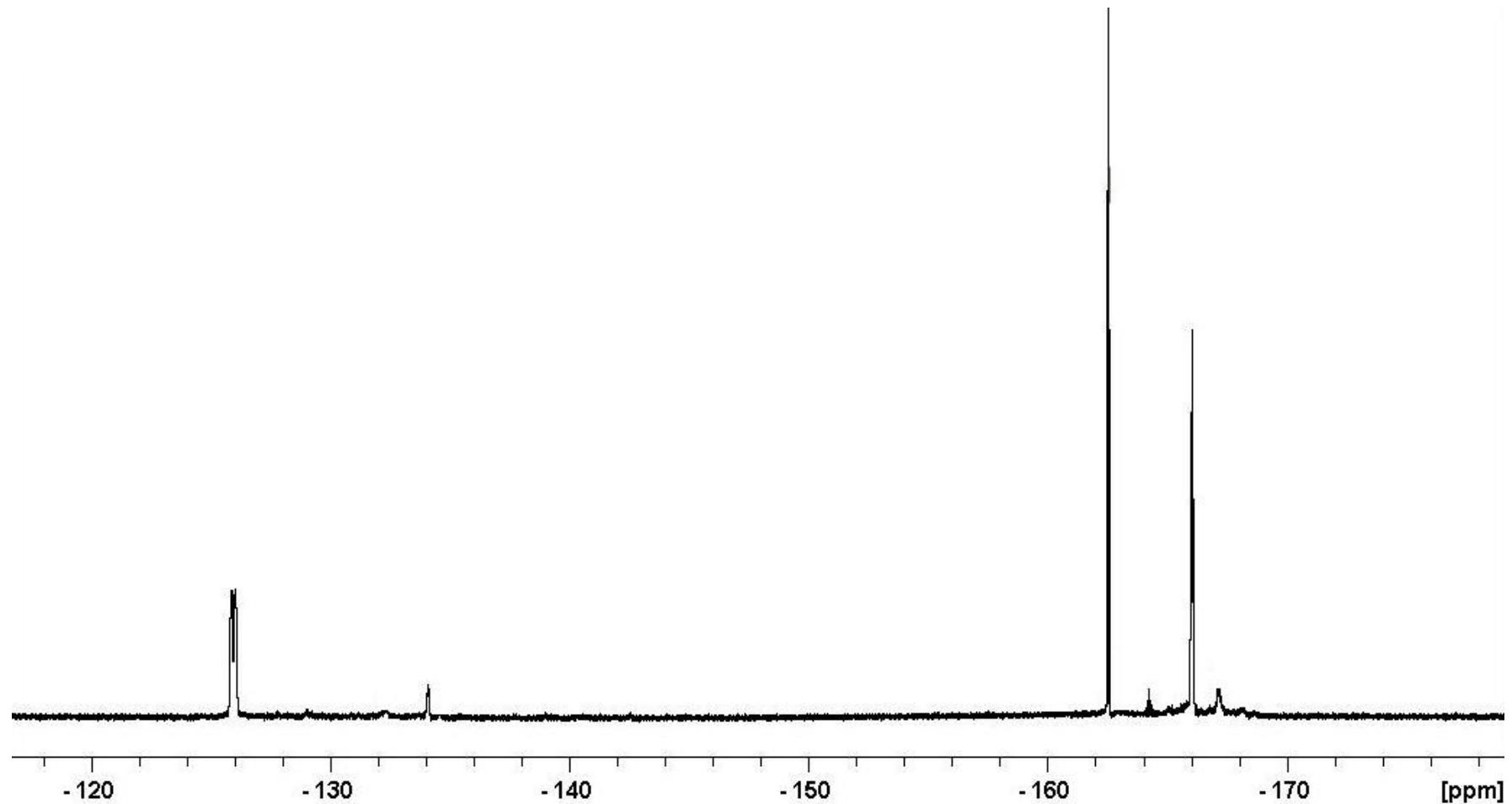
¹H NMR



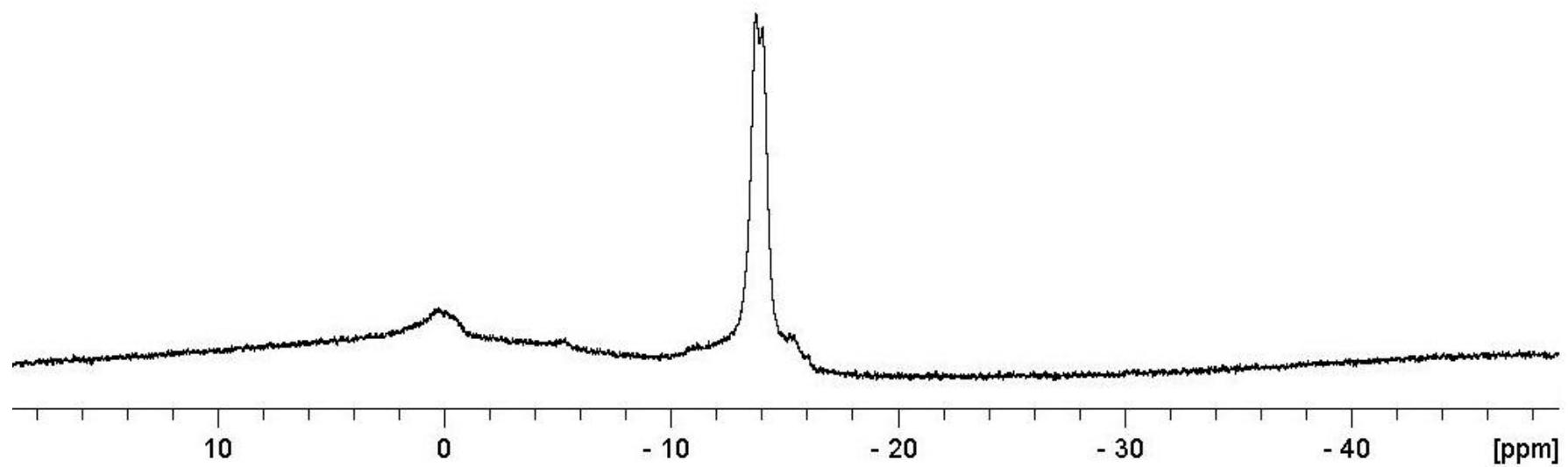
¹³C NMR



¹⁹F NMR



^{11}B NMR



3. X-Ray structure determination of 3·2THF

Crystal data: $C_{37}H_{36}BF_{15}N_2O_2P_4$, $M = 960.37$, monoclinic, $P2_1/n$, $a = 15.2628(2)$, $b = 12.8453(2)$, $c = 21.2228(4)$ Å, $\beta = 94.111(2)^\circ$, $Z = 4$, $V = 4150.14(2)$ Å³, $D_x = 1.537$ Mg m⁻³, $\mu = 0.29$ mm⁻¹, $T = 100$ K. Data collection: A crystal $0.26 \times 0.16 \times 0.09$ mm was used to record 160687 intensities to $2\theta = 56.6^\circ$ on an Oxford Diffraction Xcalibur E diffractometer using monochromated Mo $K\alpha$ radiation. An absorption correction was performed on the basis of multi-scans. Structure refinement: The structure was refined anisotropically on F^2 using the program SHELXL-97 (G.M. Sheldrick, University of Göttingen, Germany). Hydrogen atoms were included using rigid methyl groups or a riding model. The final $wR2$ was 0.0631 for 556 parameters and all 9886 unique reflections, with conventional $R1$ ($F > 4\sigma(F)$) 0.0298; max. $\Delta\rho$ 0.43 e Å⁻³; S 0.87.

4. Computational Details

All computations were performed using the hybrid density functional method M05-2X implemented in the Gaussian09 program.^[1] For all main-group elements (C, H, N, B, F) the all-electron triple- ζ basis set (6-311G**) was used.^[2]

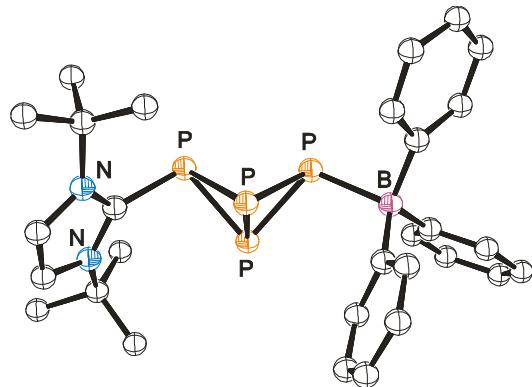
- [1] Gaussian 09, Revision A.1, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. RagHAVACHARI, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.
- [2] X. Cao and M. Dolg, *J. Chem. Phys.* 2001, **115**, 7348.

Energies for all optimized structure:

Compound	E(0 K) ^a [Ha]	H(298 K) ^b [Ha]	G(298 K) ^b [Ha]
<i>t</i> Bu ₂ ImC-P ₄ -BR ₃ (4) (<i>trans-trans</i>)	-4114.484986	-4114.434616	-4114.570734
<i>t</i> Bu ₂ ImCH-P ₄ -BR ₃ (3) (<i>trans-trans</i>)	-4114.508198	-4114.457935	-4114.593018
<i>t</i> Bu ₂ ImCH-P ₄ -BR ₃ (5) (<i>trans-cis</i>)	-4114.504674	-4114.454632	-4114.587323
<i>t</i> Bu ₂ ImCH-P ₄ -BR ₃ (6) (<i>cis-trans</i>)	-4114.511446	-4114.460983	-4114.597554

^aDFT energy incl. ZPE.

^bstandard conditions T = 298.15 K and p = 1 atm.



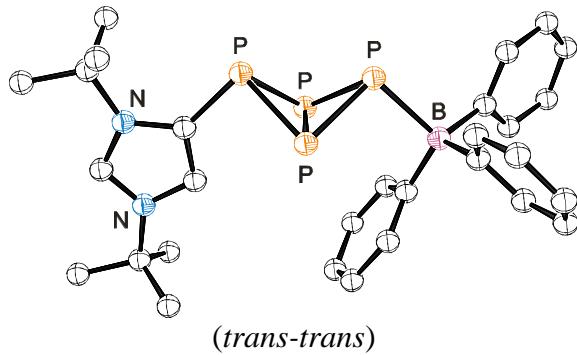
(*trans-trans*)

Structure of the carben-P₄-boran adduct (**4**) (atom, x-, y-, z-positions in Å):

(Hydrogen and fluorine atoms are omitted for clarity)

B	1.638200	0.064600	-0.138200
N	-5.102400	1.019100	-0.047300
N	-5.124500	-1.144100	-0.341100
P	-2.876800	0.285400	-1.655900
P	-1.603300	0.530900	0.163600
P	-1.512500	-1.340200	-0.926300
P	-0.017700	0.257100	-1.356500
F	4.001100	0.968200	-1.792600
F	4.738600	3.453400	-2.213600
F	3.363500	5.528000	-1.122000
F	1.190000	5.018400	0.436800
F	0.419600	2.535400	0.900100
F	2.230200	1.575000	2.474800
F	1.466900	0.797300	4.874200
F	0.188300	-1.568600	5.235100
F	-0.295900	-3.173200	3.083500
F	0.465100	-2.419300	0.654700
F	4.155700	-0.340200	1.021000
F	5.964700	-2.228600	0.441300
F	5.496800	-3.989800	-1.576700
F	3.182400	-3.832600	-2.991100
F	1.371000	-1.982800	-2.425400
C	-6.078800	0.497700	0.769400
C	-6.087300	-0.838100	0.594500
C	-4.473700	-0.007500	-0.684600
C	-4.994400	2.508000	-0.333600
C	-3.663300	3.103000	0.126100
C	-5.224100	2.719900	-1.830000
C	-6.113800	3.220100	0.428200
C	2.188200	1.592700	-0.376200
C	3.272900	1.915400	-1.184200
C	3.681800	3.216100	-1.435000
C	2.987100	4.272800	-0.885700
C	1.886400	4.009400	-0.093700
C	1.514000	2.700100	0.130900
C	1.264100	-0.319000	1.404500
C	1.547800	0.426900	2.543600
C	1.185200	0.029500	3.820200
C	0.545900	-1.179600	4.012000
C	0.298400	-1.987900	2.920500
C	0.681900	-1.554100	1.666300
C	2.674600	-1.079600	-0.656900
C	3.873200	-1.205300	0.033300
C	4.826800	-2.161000	-0.252000
C	4.593600	-3.055900	-1.280900
C	3.418000	-2.971400	-1.997800
C	2.489800	-1.994400	-1.676500
C	-5.045600	-2.542000	-0.917200
C	-4.232300	-3.426800	0.022400

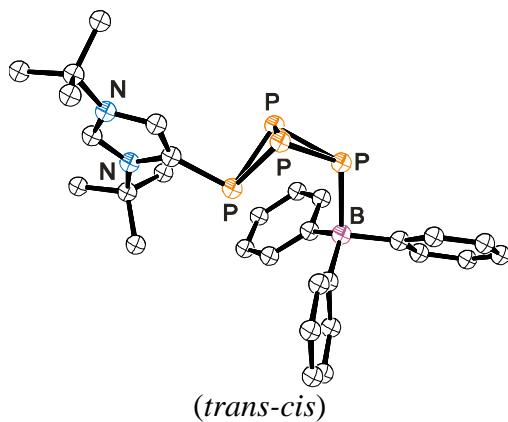
C	-4.492400	-2.513200	-2.340000
C	-6.480700	-3.080600	-1.008600
H	-6.914300	-3.294500	-0.034000
H	-7.127400	-2.390800	-1.551100
H	-6.446900	-4.021800	-1.554000
H	-6.709300	-1.578800	1.054100
H	-3.411300	2.770700	1.132200
H	-3.770300	4.187500	0.140300
H	-2.843800	2.863900	-0.543200
H	-6.696800	1.097700	1.404700
H	-3.207400	-3.079700	0.116900
H	-4.689400	-3.450300	1.012500
H	-4.218800	-4.442600	-0.373100
H	-3.434400	-2.291500	-2.386100
H	-4.637400	-3.502700	-2.771100
H	-5.038600	-1.793000	-2.950900
H	-7.100600	2.853300	0.147000
H	-6.064300	4.273800	0.162000
H	-5.981900	3.148100	1.508000
H	-4.436700	2.272100	-2.431700
H	-5.229000	3.791100	-2.030900
H	-6.187100	2.305100	-2.129900



Structure of the abnormal carben-P₄-boran adduct (3) (atom, x-, y-, z-positions in Å):
(Hydrogen and fluorine atoms are omitted for clarity)

B	-1.806600	-0.102400	-0.138100
N	5.320800	-1.405600	-0.643300
N	5.258100	0.564800	0.276100
P	2.482700	-2.034100	-1.118100
P	1.164500	-1.213000	0.470800
P	1.248600	-0.214700	-1.476600
P	-0.490900	-1.511900	-0.964900
F	-0.690700	0.502200	2.572800
F	1.137700	2.326400	3.114600
F	2.038900	4.015300	1.192600
F	1.045500	3.821000	-1.329900
F	-0.764600	1.978700	-1.929600
F	-3.075300	2.558800	-0.946000
F	-5.045300	2.840400	-2.677500
F	-6.141500	0.682700	-3.906400
F	-5.208000	-1.809200	-3.331100
F	-3.227000	-2.129000	-1.582100
F	-3.607700	1.335600	1.471900
F	-5.113500	0.448900	3.492100
F	-4.960400	-2.152600	4.272100
F	-3.269400	-3.843700	2.980000
F	-1.770900	-2.994500	0.977500
C	-0.892800	1.175500	0.296600
C	-0.354700	1.328000	1.570500
C	0.608700	2.272200	1.887500
C	1.067700	3.134700	0.913700
C	0.555600	3.039400	-0.365200
C	-0.398500	2.077500	-0.642600

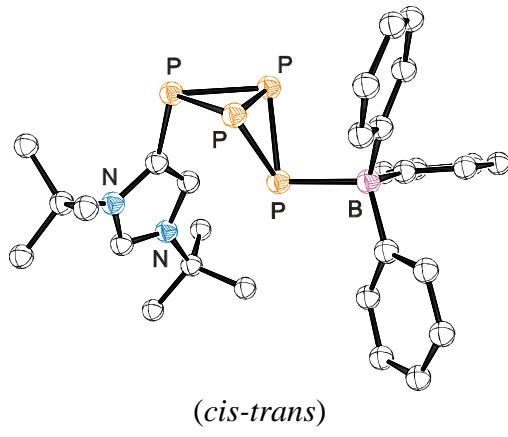
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C	-3.521100	1.441700	-1.536300
C	-4.568900	1.619100	-2.425000
C	-5.135200	0.524600	-3.047500
C	-4.654300	-0.737700	-2.759200
C	-3.609500	-0.870200	-1.864100
C	-2.609400	-0.773600	1.124200
C	-3.499600	0.041900	1.813900
C	-4.287300	-0.388800	2.862400
C	-4.211400	-1.710800	3.262500
C	-3.351900	-2.565000	2.604300
C	-2.578500	-2.087600	1.557400
C	6.058500	-0.414800	-0.134800
C	3.975900	-1.049500	-0.560000
C	3.967600	0.189000	0.021800
C	5.878800	-2.681700	-1.197000
C	5.329700	-3.846500	-0.375100
C	5.487100	-2.782600	-2.670000
C	7.400200	-2.656800	-1.080200
C	5.632900	1.878800	0.882100
C	4.928300	1.989900	2.233200
C	5.168800	2.978100	-0.072300
C	7.143600	1.929300	1.063800
H	7.493900	1.142000	1.732600
H	7.399700	2.886400	1.514100
H	7.666500	1.862800	0.108800
H	7.127200	-0.406800	-0.065200
H	3.130200	0.815100	0.265400
H	5.588600	-3.725900	0.676900
H	5.773400	-4.772700	-0.739600
H	4.249200	-3.932200	-0.466300
H	4.408900	-2.848400	-2.797700
H	5.932600	-3.683800	-3.090400
H	5.858200	-1.919300	-3.222600
H	7.840200	-1.844600	-1.660100
H	7.778700	-3.592700	-1.486700
H	7.727600	-2.593000	-0.041800
H	3.844500	1.990100	2.127200
H	5.212600	2.931300	2.701600
H	5.218500	1.169300	2.889000
H	5.649700	2.872300	-1.044800
H	5.433800	3.947400	0.348300
H	4.088300	2.960600	-0.208300



Structure of the abnormal carben-P₄-boran adduct (5) (atom, x-, y-, z-positions in Å):
(Hydrogen and fluorine atoms are omitted for clarity)

B	-1.630400	-0.060400	-0.322200
N	4.432500	0.446900	0.421200
N	5.665900	-0.934100	-0.721400
P	1.692000	-0.339600	-0.228600
P	1.273600	0.639300	-2.189100

P	1.259900	-1.549300	-2.018000
P	-0.654000	-0.455600	-2.131100
F	-0.181500	0.755600	2.263300
F	0.802700	3.147700	2.751000
F	0.556300	5.158100	0.960600
F	-0.776500	4.715100	-1.385700
F	-1.783200	2.309800	-1.914400
F	-3.787400	2.121000	-0.045200
F	-6.279700	1.894400	-0.896800
F	-7.134700	-0.368900	-2.121000
F	-5.407400	-2.442100	-2.479100
F	-2.882600	-2.244700	-1.631000
F	-2.897700	0.336900	2.139800
F	-3.186700	-1.367100	4.170300
F	-2.129000	-3.870100	4.017200
F	-0.792500	-4.624400	1.774400
F	-0.506500	-2.973200	-0.253000
C	-1.073700	1.402100	0.144100
C	-0.410300	1.694800	1.328100
C	0.114400	2.943400	1.616800
C	-0.001200	3.972200	0.706300
C	-0.660100	3.738500	-0.485700
C	-1.172100	2.479000	-0.731600
C	-3.184900	-0.036200	-0.849900
C	-4.112800	0.979800	-0.668700
C	-5.428600	0.884900	-1.092100
C	-5.870500	-0.266600	-1.712800
C	-4.990700	-1.316300	-1.896300
C	-3.688300	-1.181000	-1.457600
C	-1.638100	-1.194400	0.857700
C	-2.346100	-0.880700	2.015000
C	-2.516600	-1.742600	3.078900
C	-1.982700	-3.016200	3.004400
C	-1.302000	-3.392000	1.866400
C	-1.155700	-2.491900	0.821300
C	5.681600	0.075900	0.144600
C	3.556900	-0.357800	-0.308500
C	4.352300	-1.215800	-1.012100
C	4.064200	1.567400	1.355000
C	3.279100	2.609900	0.562500
C	3.256700	0.989000	2.514900
C	5.339700	2.205300	1.896700
C	6.838100	-1.648300	-1.305400
C	6.803300	-1.442100	-2.818600
C	6.714600	-3.127100	-0.944600
C	8.122100	-1.071300	-0.723000
H	3.873100	2.985900	-0.271000
H	3.029300	3.440400	1.222600
H	2.348700	2.204400	0.173100
H	2.319400	0.547300	2.185500
H	3.012100	1.795500	3.204600
H	3.839900	0.234300	3.043500
H	5.931500	1.501800	2.484000
H	5.042100	3.016000	2.558800
H	5.950400	2.636500	1.102200
H	5.895900	-1.856600	-3.255800
H	7.657800	-1.946800	-3.267900
H	6.854200	-0.381300	-3.063100
H	6.699600	-3.259700	0.136900
H	7.570400	-3.664400	-1.351500
H	5.810300	-3.566100	-1.363300
H	8.241100	-0.016900	-0.974300
H	8.962800	-1.610600	-1.155800
H	8.165900	-1.200200	0.359100
H	6.565700	0.521900	0.551800
H	4.064300	-1.995900	-1.692100



Structure of the abnormal carben-P₄-boran adduct (6) (atom, x-, y-, z-positions in Å):
 (Hydrogen and fluorine atoms are omitted for clarity)

B	1.250400	0.021100	0.067500
N	-4.665900	0.955600	0.138400
N	-4.387200	-1.1130500	-0.420200
P	-2.729300	2.331400	-1.586600
P	-0.792800	1.350600	-2.078300
P	-0.905700	2.625600	-0.307200
P	-0.758800	0.421600	-0.058500
F	2.673900	0.920900	-2.437400
F	3.768600	3.298900	-2.804000
F	3.704900	5.181000	-0.846600
F	2.475600	4.615500	1.516800
F	1.359700	2.233900	1.915800
F	3.435900	0.611200	2.134800
F	3.709000	-0.610500	4.454600
F	2.020400	-2.604700	5.191900
F	0.024700	-3.368100	3.495400
F	-0.266800	-2.172600	1.145200
F	3.890800	-0.820700	-0.365300
F	4.786300	-2.828000	-1.877700
F	3.043300	-4.347300	-3.309800
F	0.376700	-3.801300	-3.186600
F	-0.538000	-1.825700	-1.724700
C	2.020000	1.423600	-0.206600
C	2.626600	1.778200	-1.403500
C	3.196200	3.020700	-1.631600
C	3.166100	3.980300	-0.640500
C	2.551500	3.686300	0.562600
C	1.994500	2.437200	0.746000
C	1.504000	-0.646200	1.545400
C	2.532400	-0.331000	2.426100
C	2.712200	-0.968300	3.643900
C	1.861200	-1.988200	4.021600
C	0.848900	-2.368500	3.162400
C	0.709700	-1.709400	1.956800
C	1.644700	-1.182800	-0.981600
C	2.990900	-1.526200	-1.064100
C	3.481200	-2.562100	-1.833600
C	2.598100	-3.338500	-2.563300
C	1.250700	-3.055600	-2.501600
C	0.808600	-1.999700	-1.721300
C	-5.047800	-0.299100	0.381000
C	-3.688700	0.932800	-0.855500
C	-3.544200	-0.381900	-1.196400
C	-5.224500	2.155400	0.842700
C	-4.110200	2.814500	1.651600
C	-5.806800	3.100100	-0.206100
C	-6.335600	1.709300	1.789700
C	-4.413800	-2.625200	-0.390000
C	-3.095100	-3.095400	0.218600
C	-4.567900	-3.122100	-1.825300

C	-5.594900	-3.084300	0.455500
H	-3.661000	2.098000	2.339700
H	-4.533100	3.638700	2.225600
H	-3.330500	3.211900	1.009200
H	-5.042300	3.471000	-0.885500
H	-6.252000	3.955300	0.301500
H	-6.580300	2.596100	-0.786000
H	-7.151400	1.218100	1.257800
H	-6.740900	2.600100	2.265400
H	-5.959800	1.058300	2.579600
H	-2.245000	-2.757300	-0.370300
H	-3.085600	-4.185000	0.249100
H	-2.973900	-2.714600	1.232200
H	-5.465700	-2.710200	-2.286000
H	-4.647400	-4.208300	-1.812300
H	-3.701600	-2.864000	-2.432800
H	-5.482200	-2.788300	1.498900
H	-5.626800	-4.171900	0.429200
H	-6.541300	-2.708600	0.063700
H	-5.787700	-0.596200	1.095200
H	-2.864400	-0.827300	-1.897500