## Supporting information

## **Crystal Structure and Carrier Transport Properties of a New 3D Mixed-Valence Cu(I)-Cu(II) Coordination Polymer Including Pyrrolidine Dithiocarbamate Ligand**

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Figure S1. Comparison of XRPD patterns of product 1 (a) with the respective calculated ones (b) form the single-crystal data.

Table S1. Estimated parameters ( $R_1$ ,  $C_1$ ,  $R_2$ ,  $C_2$ ,  $R_3$ , CPE<sub>3</sub>-A and CPE<sub>3</sub>- $\alpha$ ) by the fittings to the impedance data of **1** at different temperatures with the equivalent circuit in the inset of Figure 6 using ZView software, where  $R_1$ ,  $R_2$  and  $R_3$  are the resistances, CPE<sub>3</sub>-A is the parameter concerned with the capacitance, and CPE<sub>3</sub>- $\alpha$  is corresponding to the dispersion of the relaxation time of the electrode interface, respectively; the CPE impedance is given by the relationship:  $Z_{CPE} = 1/[A(j\omega)^{\alpha}]$ , and for  $\alpha = 1$  the CPE describes an ideal capacitor (A = C).

$T(\mathbf{K})$	$R_1\left(\Omega\right)$	$C_1(\mathbf{F})$	$R_2\left(\Omega\right)$	$C_2(\mathbf{F})$	$R_3(\Omega)$	CPE <sub>3</sub> -A	CPE <sub>3</sub> - <i>a</i>	$\chi^2$
250	3.59E+05	5.24E-11	1.82E+06	7.18E-11	1.80E+08	3.88E-11	0.97654	0.002277
255	3.33E+05	4.73E-11	1.65E+06	7.84E-11	1.93E+08	4.16E-11	0.9738	0.0019584
260	2.35E+05	4.80E-11	1.16E+06	8.05E-11	1.56E+08	4.48E-11	0.96724	0.001954

265	2.05E+05	4.69E-11	9.49E+05	9.05E-11	1.64E+08	4.86E-11	0.96142	0.0016443
270	1.59E+05	4.94E-11	6.54E+05	1.04E-10	1.34E+08	5.41E-11	0.95008	0.0011515
275	1.25E+05	4.91E-11	5.14E+05	1.09E-10	1.04E+08	5.73E-11	0.94643	0.0011083
280	1.04E+05	4.87E-11	4.32E+05	1.19E-10	9.28E+07	6.20E-11	0.94117	0.0017478
285	78234	5.11E-11	3.03E+05	1.33E-10	8.41E+07	6.87E-11	0.93136	0.0013741
290	60215	5.44E-11	2.11E+05	1.53E-10	7.37E+07	7.56E-11	0.92229	0.00096505
295	47979	5.72E-11	1.55E+05	1.83E-10	6.33E+07	8.48E-11	0.91238	0.0012443
300	40506	5.79E-11	1.39E+05	2.07E-10	5.61E+07	9.17E-11	0.90762	0.0011505

Table S2. Calculated electric conductivity  $\sigma$  and dielectric constant  $\mathcal{E}$  from the parameters in Table S1 by using the following relationships:  $\sigma = 1/R \times d/S$  and  $\mathcal{E} \approx A/\mathcal{E}_0 \times d/S$ , where dis the thickness of the pellet sample, S is the electrode area and  $\mathcal{E}_0$  is the permittivity of vacuum,  $8.855 \times 10^{-12}$  F/m. The  $\sigma_1$  and  $\sigma_2$  values are corresponding to the electric conductivity of the bulk sample and grain boundary of **1**, respectively.

	$\sigma_1$ (S cm <sup>-1</sup> )	$\mathcal{E}_1$	$\sigma_2$ (S cm <sup>-1</sup> )	$\mathcal{E}_2$
250	5.90E-08	12.541	1.16E-08	17.165
255	6.36E-08	11.314	1.29E-08	18.753
260	9.03E-08	11.489	1.82E-08	19.26
265	1.04E-07	11.221	2.23E-08	21.65
270	1.34E-07	11.809	3.24E-08	24.972
275	1.69E-07	11.742	4.12E-08	26.145
280	2.04E-07	11.644	4.90E-08	28.537
285	2.71E-07	12.221	6.98E-08	31.718
290	3.52E-07	13.01	1.00E-07	36.598
295	4.41E-07	13.692	1.37E-07	43.869
300	5.23E-07	13.85	1.53E-07	49.538

Table S3. Crystal Data and Structure Refinement Parameters for 1.

**Empirical Formula** 

C21H33N4S8Cu6Br4Cl3

Formula Weight	1405.25
Crystal System	tetragonal
Lattice Parameters	a = 15.6452(5) Å
	c = 15.9953(6) Å
	$V = 3915.2(2) \text{ Å}^3$
Space Group	I4 <sub>1</sub> /a (#88)
Z value	4
$D_{\text{calc}}$	2.384 g/cm <sup>3</sup>
$F_{000}$	2720.00
$\mu$ (MoK $\alpha$ )	$79.515 \text{ cm}^{-1}$
No. Observations (All reflections)	2236
No. Variables	109
Reflection/Parameter Ratio	20.51
Residuals: $R_1 (I > 2.00 \sigma(I))^a$	0.0365
Residuals: R (All reflections)	0.0385
Residuals: $wR_2$ (All reflections) <sup>b</sup>	0.0872
Goodness of Fit Indicator	1.045
Max Shift/Error in Final Cycle	0.002
Maximum peak in Final Diff. Map	2.78 $e/Å^3$
Minimum peak in Final Diff. Map	-2.04 $e/Å^3$

<sup>*a*</sup>  $R_1 = \Sigma ||F_0| - |F_c|| / \Sigma |F_0|$ <sup>*b*</sup>  $wR_2 = [\Sigma w (F_0^2 - F_c^2)^2 / \Sigma w (F_0^2)^2]^{1/2}$ 

Table S4. Atomic coordinates and  $B_{iso}/B_{eq}$  and occupancy for 1.

atom	Х	у	Z	B <sub>eq</sub>	occ
Br1	0.22059(2)	0.52393(3)	0.00067(3)	1.445(9)	
Cu1	0.5000	0.5000	0.0000	1.181(12)	1/2
Cu2	0.30162(3)	0.53620(3)	0.13061(3)	1.364(11)	
Cl1	0.9235(2)	0.32420(19)	0.08138(19)	7.54(9)	3/4
S1	0.57147(6)	0.58857(6)	0.09172(6)	1.333(16)	
S2	0.43655(6)	0.46530(6)	0.12531(6)	1.230(15)	
N1	0.5064(2)	0.5628(2)	0.2450(2)	1.30(5)	

C1	0.5059(2)	0.5416(2)	0.1667(2)	1.17(5)	
C2	0.5603(2)	0.6318(2)	0.2811(2)	1.59(6)	
C3	0.5296(3)	0.6373(3)	0.3714(3)	2.25(7)	
C4	0.4982(3)	0.5471(3)	0.3906(2)	2.25(8)	
C5	0.4536(2)	0.5210(3)	0.3101(2)	1.73(6)	
C6	1.0000	0.2500	0.1250	10.5(7)	1/4

 $B_{eq} = 8/3 \pi^2 (U_{11}(aa^*)^2 + U_{22}(bb^*)^2 + U_{33}(cc^*)^2 + 2U_{12}(aa^*bb^*)\cos\gamma + 2U_{13}(aa^*cc^*)\cos\beta + 2U_{23}(bb^*cc^*)\cos\alpha)$ 

Table S5. Anisotropic displacement parameters for 1.

atom	U11	U22	U33	U12	U13	U23
Br1	0.0151(2)	0.0205(2)	0.0193(2)	0.00190(14)	-0.00176(14)	-0.00510(15)
Cu1	0.0124(3)	0.0158(3)	0.0167(3)	0.0007(2)	0.0008(2)	0.0009(2)
Cu2	0.0173(2)	0.0166(2)	0.0179(2)	-0.00086(18)	-0.00034(19)	-0.00068(19)
Cl1	0.174(3)	0.0541(15)	0.0581(16)	-0.0517(19)	-0.0107(19)	0.0012(12)
S1	0.0130(4)	0.0191(4)	0.0186(4)	-0.0019(3)	-0.0003(3)	0.0031(3)
S2	0.0141(4)	0.0144(4)	0.0182(4)	-0.0010(3)	0.0017(3)	-0.0012(3)
N1	0.0131(15)	0.0160(16)	0.0202(17)	0.0009(12)	-0.0001(12)	0.0006(13)
C1	0.0107(16)	0.0134(17)	0.0204(19)	0.0012(13)	-0.0009(14)	0.0010(14)
C2	0.0171(19)	0.0165(18)	0.027(2)	-0.0016(14)	-0.0011(16)	-0.0034(16)
C3	0.024(2)	0.034(2)	0.028(2)	-0.0063(18)	0.0026(18)	-0.011(2)
C4	0.027(2)	0.040(2)	0.019(2)	-0.009(2)	0.0014(17)	-0.0039(19)
C5	0.0186(19)	0.028(2)	0.019(2)	-0.0061(16)	0.0020(16)	-0.0001(17)
C6	0.083(9)	0.083(9)	0.23(3)	0.0000	0.0000	0.0000
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The general temperature factor expression:  $exp(-2\pi^2(a^{*2}U_{11}h^2 + b^{*2}U_{22}k^2 + c^{*2}U_{33}l^2 + 2a^{*}b^{*}U_{12}hk + 2a^{*}c^{*}U_{13}hl + 2$ 2b\*c\*U23kl))

Table S6.	Bond lengths (Å) of 1.
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	1. /

atom	atom	distance	atom	atom	distance
Br1	Cu2	2.4421(6)	Br1	$Cu2^{1)}$	2.4776(6)
Cu1	S1	2.3071(9)	Cu1	S1 <sup>2)</sup>	2.3071(9)
Cu1	S2	2.3017(9)	Cu1	$S2^{2}$	2.3017(9)
Cu2	S1 <sup>3)</sup>	2.3428(10)	Cu2	S2	2.3862(10)
Cl1	C6	1.807(3)	S1	C1	1.741(4)
S2	C1	1.744(3)	N1	C1	1.295(5)
N1	C2	1.486(5)	N1	C5	1.482(5)
C2	C3	1.525(6)	C3	C4	1.526(7)
C4	C5	1.520(6)			

Symmetry Operators:

(1) Y+3/4-1, -X+3/4, Z+3/4-1(3) Y+3/4-1, -X+1/4+1, -Z+1/4

(2) -X+1,-Y+1,-Z

Table S7. Bond angles (<sup>0</sup>) of **1**.

atom	atom	atom	angle	atom	atom	atom	angle
Cu2	Br1	$Cu2^{1)}$	118.82(2)	S1	Cu1	S1 <sup>2)</sup>	180.00(4)
S1	Cu1	S2	78.28(3)	S1	Cu1	$S2^{2)}$	101.72(3)
S1 <sup>2)</sup>	Cu1	S2	101.72(3)	S1 <sup>2)</sup>	Cu1	$S2^{2}$	78.28(3)
S2	Cu1	$S2^{2)}$	180.00(4)	Br1	Cu2	$Br1^{3)}$	112.30(2)
Br1	Cu2	S1 <sup>4)</sup>	111.32(3)	Br1	Cu2	S2	113.11(3)
$Br1^{3)}$	Cu2	S1 <sup>4)</sup>	111.41(3)	$Br1^{3)}$	Cu2	S2	104.89(3)
S1 <sup>4)</sup>	Cu2	S2	103.31(3)	Cu1	S1	$Cu2^{5)}$	115.63(4)
Cu1	S1	C1	84.20(13)	$Cu2^{5)}$	S1	C1	109.03(13)
Cu1	S2	Cu2	107.63(3)	Cu1	S2	C1	84.30(13)
Cu2	S2	C1	102.62(13)	C1	N1	C2	124.5(3)
C1	N1	C5	124.2(3)	C2	N1	C5	111.3(3)
S1	C1	S2	113.2(2)	S1	C1	N1	123.6(2)
S2	C1	N1	123.1(3)	N1	C2	C3	103.4(3)
C2	C3	C4	103.8(3)	C3	C4	C5	103.1(3)
N1	C5	C4	102.7(3)	Cl1	C6	Cl1 <sup>6)</sup>	98.57(15)
Cl1	C6	Cl1 <sup>7)</sup>	134.58(14)	Cl1	C6	Cl1 <sup>8)</sup>	98.57(17)
Cl1 <sup>6)</sup>	C6	Cl1 <sup>7)</sup>	98.57(17)	Cl1 <sup>6)</sup>	C6	Cl1 <sup>8)</sup>	134.58(14)
Cl1 <sup>7)</sup>	C6	Cl1 <sup>8)</sup>	98.57(15)				

Symmetry Operators:

- (1) Y+3/4-1,-X+3/4,Z+3/4-1(3) -Y+3/4,X+1/4,Z+1/4(5) -Y+1/4+1,X+1/4,-Z+1/4(7) -X+2,-Y+1/2,Z

- (2) -X+1,-Y+1,-Z