

## Kinetico-mechanistic studies of the acidolysis of Rh-C bonds in monocyclometallated dirhodium(II) acetato complexes; influence of electronic and steric effects.

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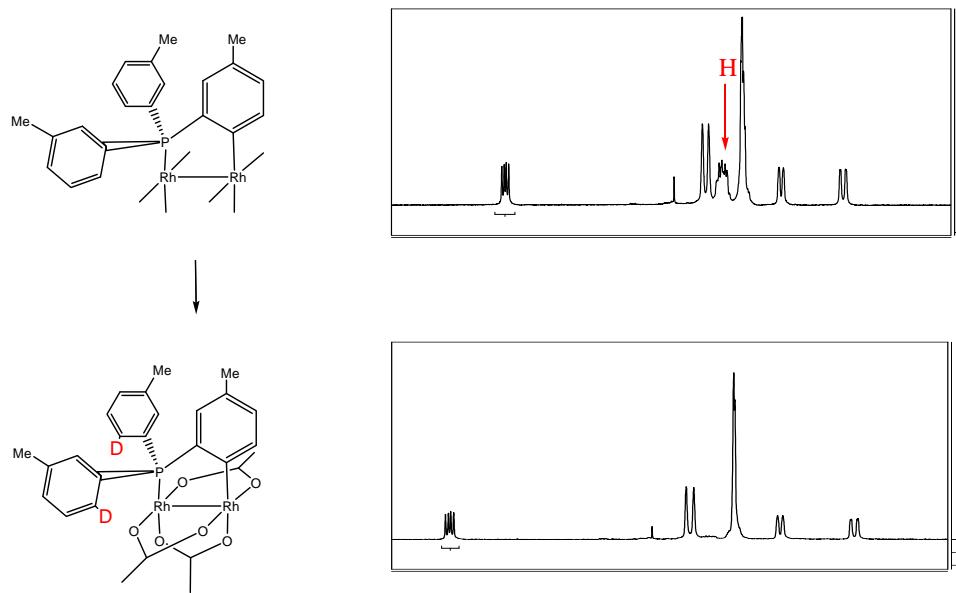
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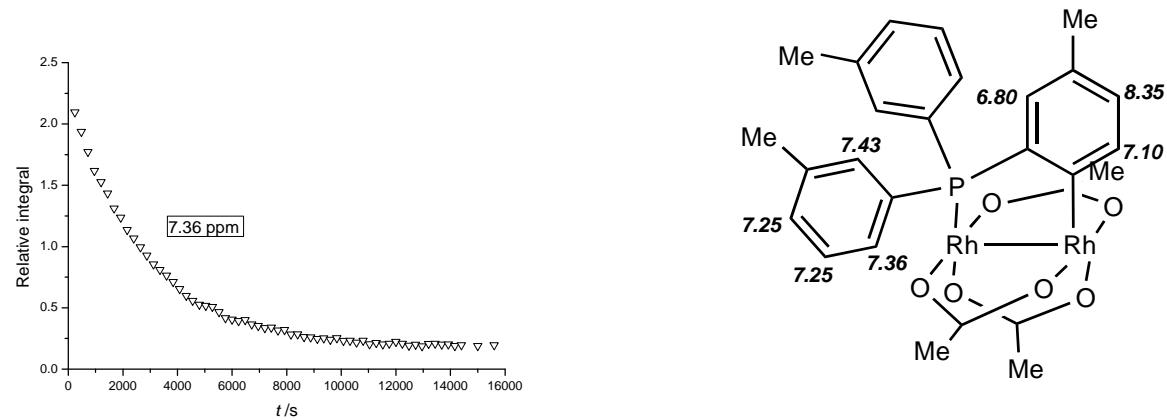
Supplementary material

**Figure S1.-** Observed  $^1\text{H}$  NMR spectral changes (aromatic zone,  $\text{CD}_3\text{COOD}$ , 65 °C) with time on dissolution of compound  $\text{Rh}_2\text{P}(4\text{-MeC}_6\text{H}_4)_3$ . Top initial spectrum, bottom after *ca.* 7 h.

FIGURE S1



**Figure S2.-** Changes in the relative integral of selected proton signals for the relevant *ortho* hydrogens on complex Rh<sub>2</sub>P(4-MeC<sub>6</sub>H<sub>4</sub>)<sub>3</sub> with time (CD<sub>3</sub>COOD, 400 MHz, 65 °C).



**Table S1.-** Observed rate constants for the acid cleavage of the Rh-C bonds of the different compounds studied as a function of the temperature.

Compound	T /K	$k_{\text{obs}} / \text{s}^{-1}$
$\text{Rh}_2\text{PPh}_3$	335	$4.6 \times 10^{-6}$
	344	$1.1 \times 10^{-5}$
	353	$2.7 \times 10^{-5}$
	362	$8.2 \times 10^{-5}$
$\text{Rh}_2\text{P}(5\text{-MeC}_6\text{H}_4)_3$	343	$7.7 \times 10^{-6}$
	353	$1.7 \times 10^{-5}$
	363	$3.2 \times 10^{-5}$
	373	$8.8 \times 10^{-5}$
$\text{Rh}_2\text{P}(4\text{-MeC}_6\text{H}_4)_3$	335	$4.1 \times 10^{-5}$
	344	$1.2 \times 10^{-4}$
	353	$2.9 \times 10^{-4}$
	362	$4.4 \times 10^{-4}$
$\text{Rh}_2\text{P}(5\text{-CF}_3\text{C}_6\text{H}_4)_3$	353	$1.2 \times 10^{-6}$
	363	$2.6 \times 10^{-6}$
	373	$4.2 \times 10^{-6}$
	383	$6.2 \times 10^{-6}$
$\text{Rh}_2\text{P}(4\text{-CF}_3\text{C}_6\text{H}_4)_3$	363	$2.4 \times 10^{-6}$
	373	$3.2 \times 10^{-6}$
	378	$3.7 \times 10^{-6}$
	383	$4.5 \times 10^{-6}$

**Table S2.-** Characterization data before H/D scrambling exchange for all the complexes studied in this work and indicated in Chart 1.

Compound	<sup>1</sup> H (300 MHz) in CD <sub>3</sub> COOD	<sup>31</sup> P (121 MHz) in CDCl <sub>3</sub>	<sup>19</sup> F (282 MHz) in CDCl <sub>3</sub>
Rh <sub>2</sub> PPh <sub>3</sub>	2.26 (s, 3), 2.32 (s, 6), 6.80 (m, 1), 7.07 (m, 1), 7.36 (m, 4), 7.35 (m, 2), 7.42 (m, 2), 8.35 (m, 1)	18.2 ( $J_{\text{Rh-P}} = 150$ , $^2J_{\text{Rh-P}} = 6$ )	-
Rh <sub>2</sub> P(5-MeC <sub>6</sub> H <sub>4</sub> ) <sub>3</sub>	2.04 (s, 3), 2.06 (s, 6), 2.39 (s, 6), 2.42 (s, 3), 6.79 (m, 1), 7.16 (m, 1), 7.39 (m, 4), 7.42(m, 4), 7.46 (m, 4), 8.29 (m, 1)	16.8 ( $J_{\text{Rh-P}} = 158$ , $^2J_{\text{Rh-P}} = 6$ )	-
Rh <sub>2</sub> P(4-MeC <sub>6</sub> H <sub>4</sub> ) <sub>3</sub>	2.06 (s, 3), 2.16 (s, 6), 2.26 (s, 3), 2.32 (s, 6), 6.80 (m, 1), 7.10 (m, 1), 7.25 (m, 4), 7.36 (m, 2), 7.43 (m, 2), 8.35 (m, 1)	17.7 ( $J_{\text{Rh-P}} = 150$ , $^2J_{\text{Rh-P}} = 6$ )	-
Rh <sub>2</sub> P(5-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> ) <sub>3</sub>	0.94 (broad, 3), 1.35 (broad, 6), 2.15 (s, 6), 2.05 (s, 3), 6.83 (m, 1), 7.10 (m, 1), 7.68 (m, 4), 7.79 (m, 4), 8.86 (broad signal, 1)	19.8 ( $J_{\text{Rh-P}} = 154$ )	-66.89 ppm (s, 1), -66.39 ppm (s, 2)
Rh <sub>2</sub> P(4-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub> ) <sub>3</sub>	2.03 (s, 3), 2.16 (s, 6), 7.27 (m, 1), 7.55 (m, 1), 7.72 (m, 4), 7.88 (d, 2), 8.06 (d, 2), 8.89 (m, 1)	20.7 ( $J_{\text{Rh-P}} = 156$ )	-66.82 ppm (s, 1), -66.39 ppm (s, 2)