

Diboron-Containing Fluorophores with Extended Ladder-Type π -Conjugated Skeleton

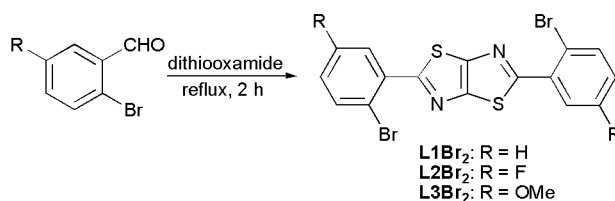
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Synthesis of the starting materials



Scheme S1: Synthetic procedure of the starting materials

2,5-bis(2-bromophenyl)thiazolo[5,4-d]thiazole (L1Br₂). A mixture of dithiooxamide (300 mg, 2.5 mmol) and 2-bromobenzaldehyde (4.63 g, 25 mmol) was heated to reflux temperature, and then it was cooled and diluted with ethanol after heating for 2 hours. The crude product was washed with ether and recrystallized from cyclohexanone to give yellow crystals (400 mg, 35% yield). ¹H NMR (300 MHz, CDCl₃, ppm): δ 8.06 (d, J = 8.1 Hz, 2 H), 7.76 (d, J = 7.8 Hz, 2 H), 7.46 (t, J = 7.5 Hz, 2 H), 7.33 (t, J = 7.2 Hz, 2 H). Ms m/z: 451.8 [M]⁺ (calcd: 451.9). Anal. Calcd (%) for C₁₆H₈Br₂N₂S₂: C, 42.50; H, 1.78; N, 6.20; S, 14.18. Found: C, 42.37; H, 1.93; N, 6.21; S, 14.41.

2,5-bis(2-bromo-5-fluorophenyl)thiazolo[5,4-d]thiazole (L2Br₂). In the same manner described for L1Br₂, compound L2Br₂ was provided as green crystals (20% yield) using 2-bromo-5-fluorobenzaldehyde as the initial material. Ms m/z: 487.8 [M]⁺ (calcd: 487.8). Anal. Calcd (%) for C₁₆H₆Br₂F₂N₂S₂: C, 39.37; H, 1.24; N, 5.74; S, 13.14. Found: C, 39.67; H, 1.43; N, 5.92; S, 13.35. ¹H NMR spectrum of L2Br₂ was not recorded due to its poor solubility in organic solvents.

2,5-bis(2-bromo-5-methoxyphenyl)thiazolo[5,4-d]thiazole (L3Br₂). In the same manner described for L1Br₂, compound L3Br₂ was provided using 2-bromo-5-methoxybenzaldehyde as the initial material and purified by vacuum sublimation to give yellow green solids (10% yield). Ms m/z: 512.8 [M]⁺ (calcd: 511.9). Anal. Calcd (%) for C₁₈H₁₂Br₂N₂O₂S₂: C, 42.21; H, 2.36; N, 5.47; S, 12.52. Found: C, 42.47; H, 2.34; N, 5.54; S, 12.56. ¹H NMR spectrum of L3Br₂ was not recorded due to its poor solubility in organic solvents.

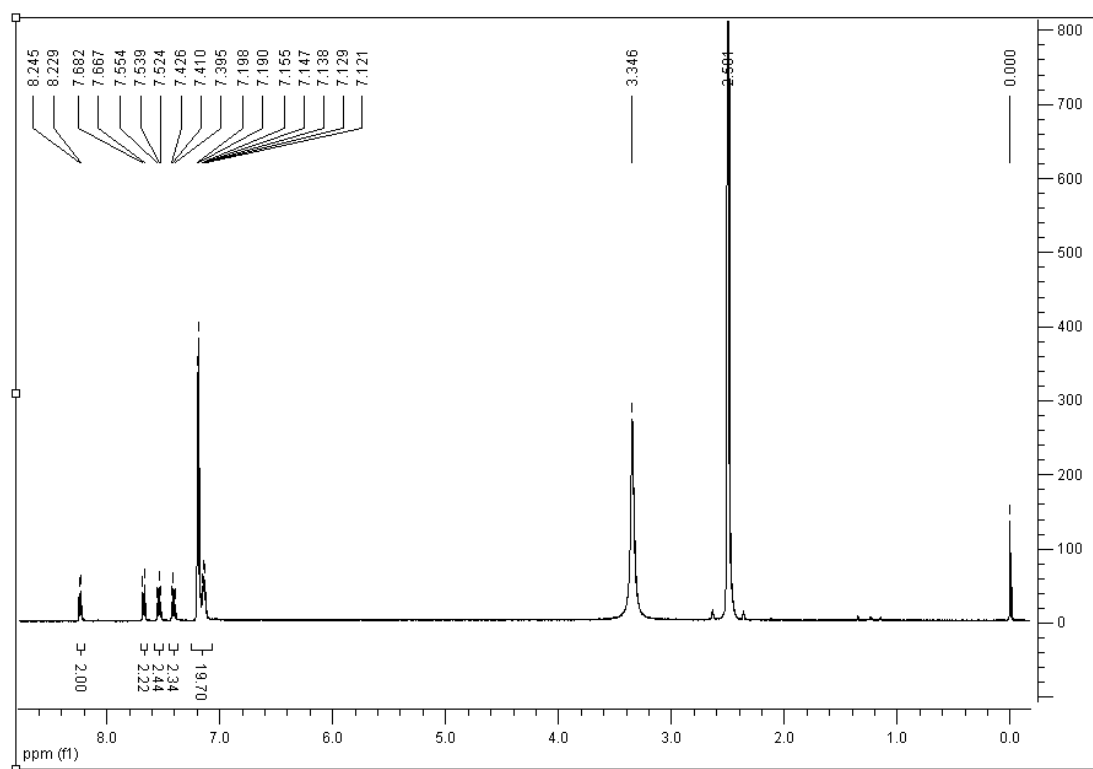


Fig. S1 ^1H NMR spectrum of compound **1**.

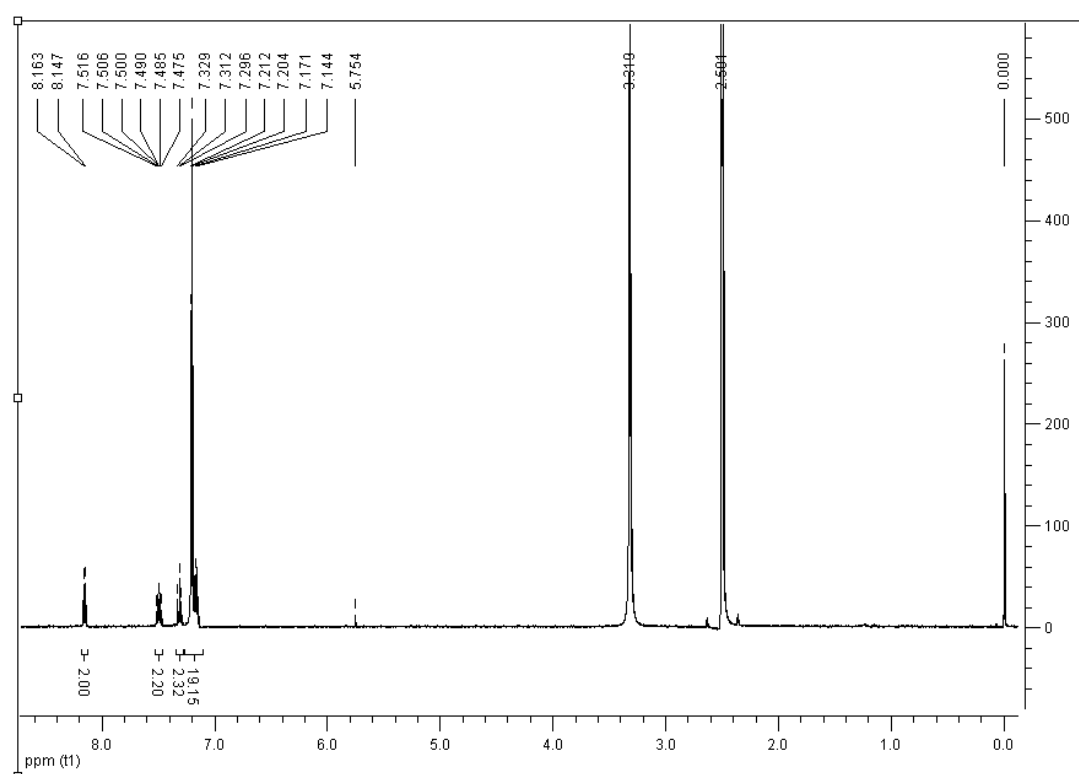


Fig. S2 ^1H NMR spectrum of compound **2**.

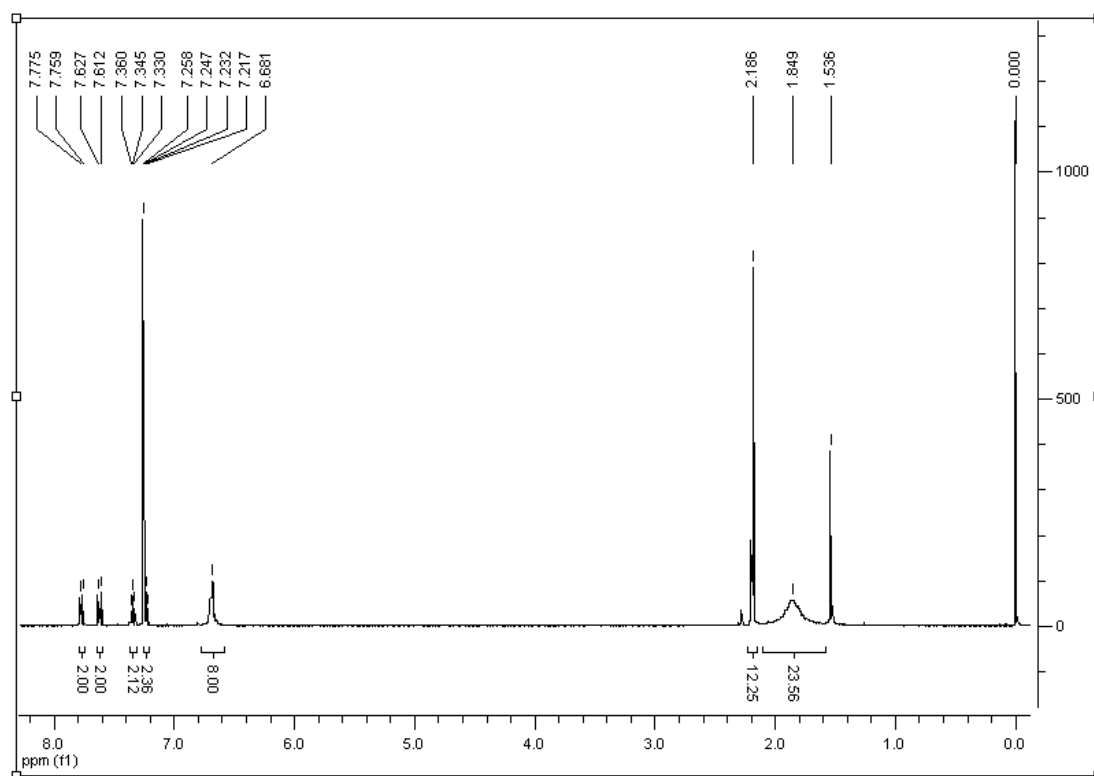


Fig. S3 ^1H NMR spectrum of compound **3**.

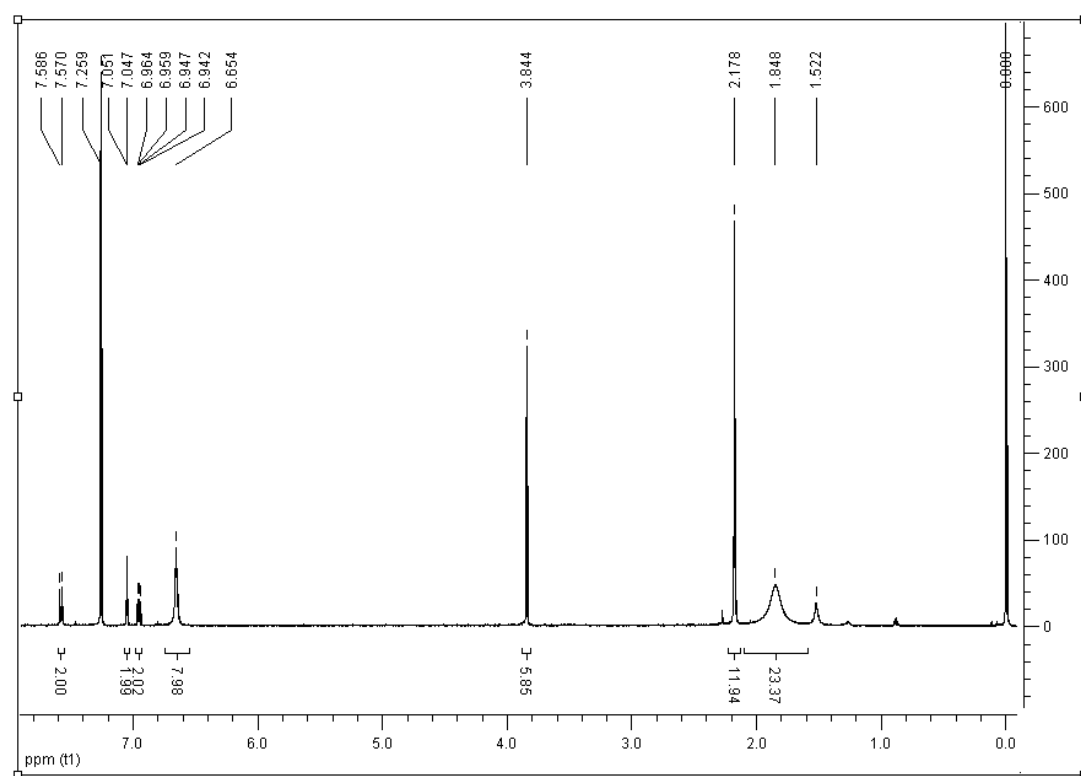


Fig. S4 ^1H NMR spectrum of compound **4**.

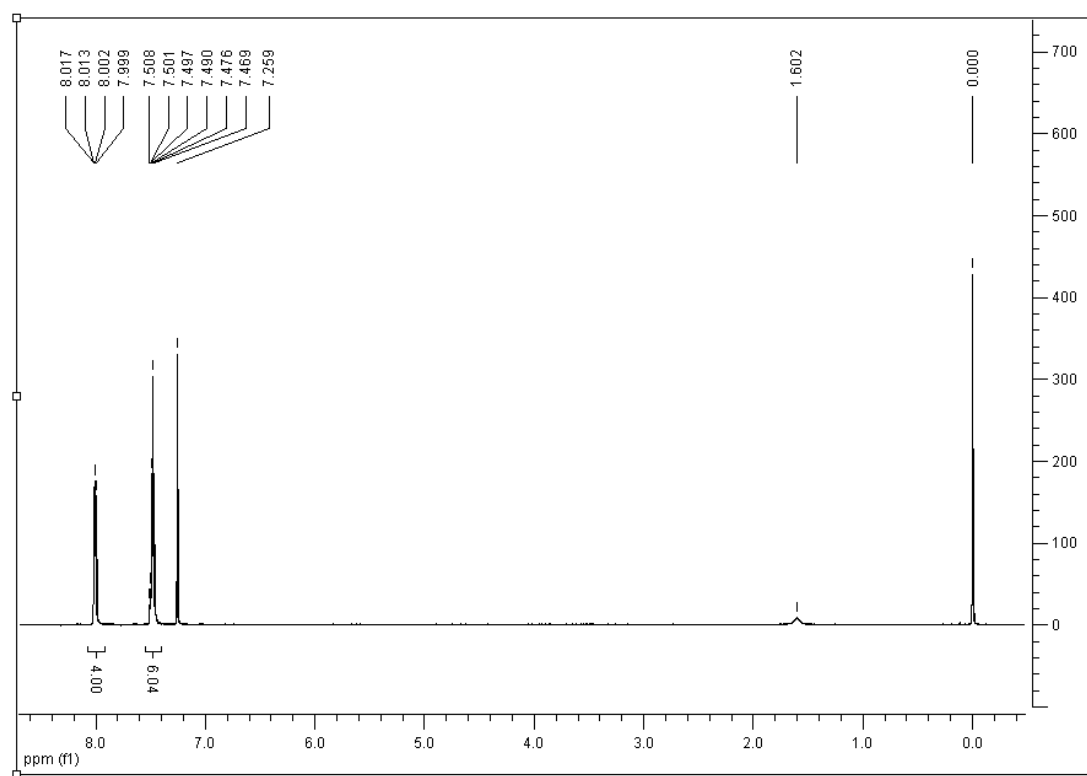


Fig. S5 ^1H NMR spectrum of compound **5**.

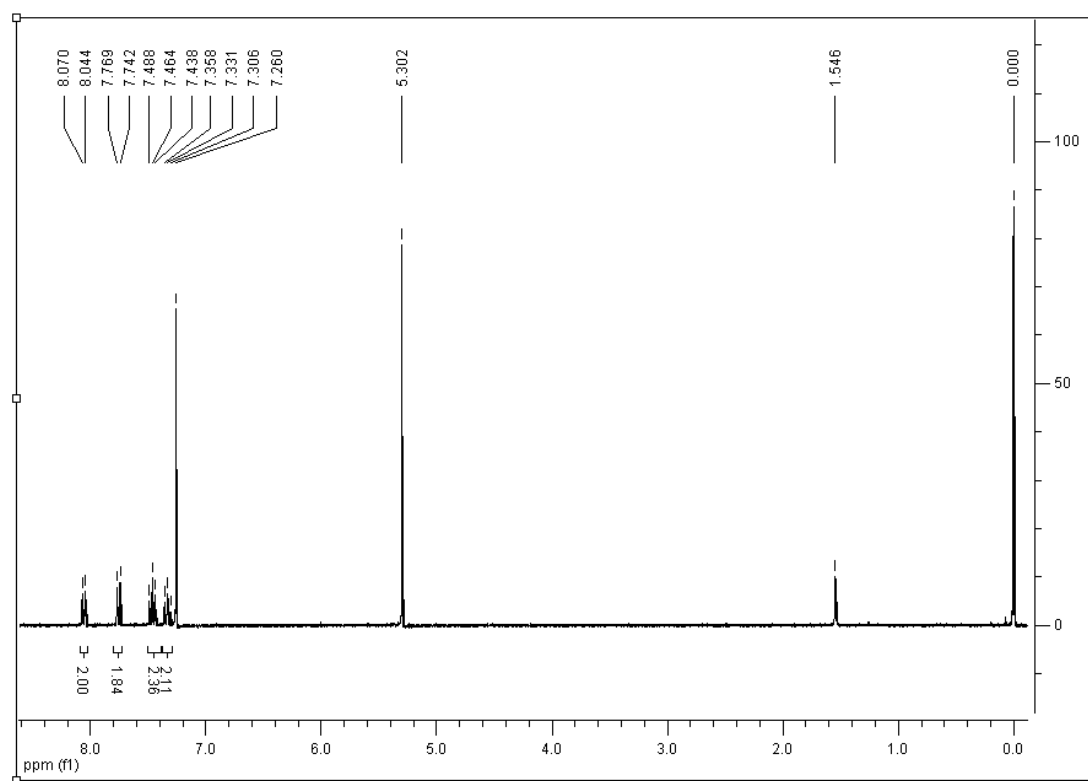


Fig. S6 ^1H NMR spectrum of compound **L1Br₂**.

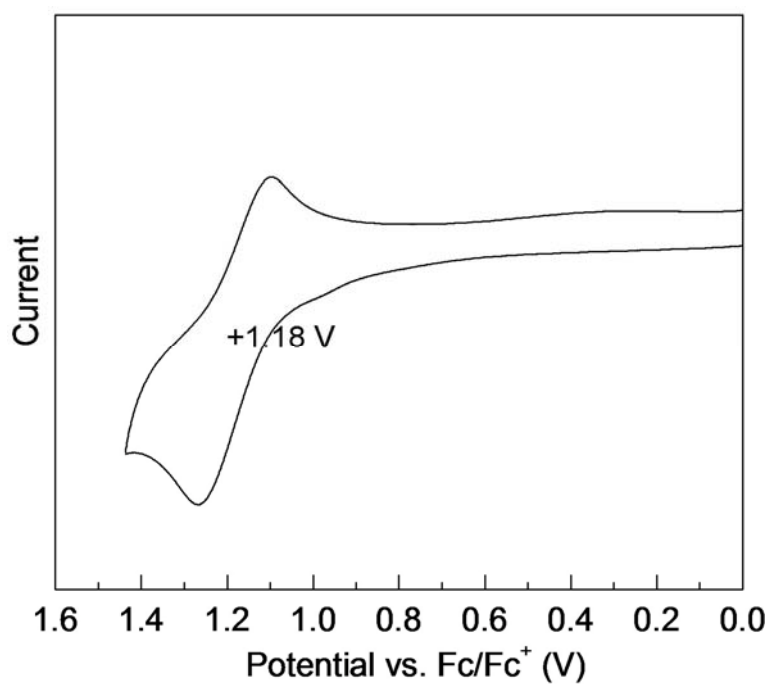


Fig. S7 Cyclic voltammogram for oxidation measurement of **5** in CH₂Cl₂ (1 mM), measured with TBAP (0.1 M) as a supporting electrolyte at a scan rate of 100 mV s⁻¹.

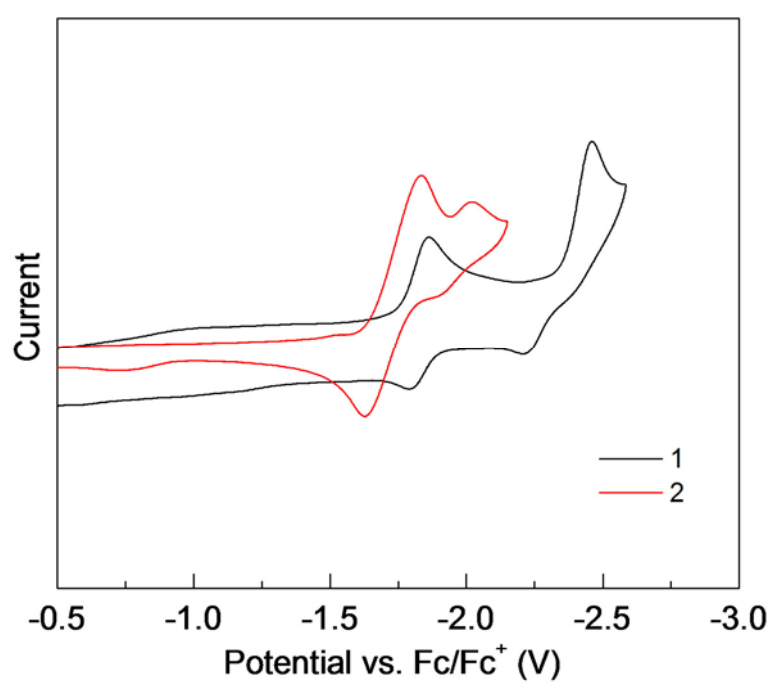


Fig. S8 Cyclic voltammograms of **3** and **5** with two sequential reduction peaks in THF (1 mM), measured with TBAP (0.1 M) as a supporting electrolyte at a scan rate of 100 mV s⁻¹.

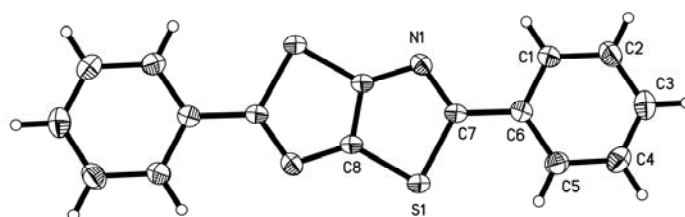


Fig. S9 ORTEP drawing of **5** with 30% thermal ellipsoids.

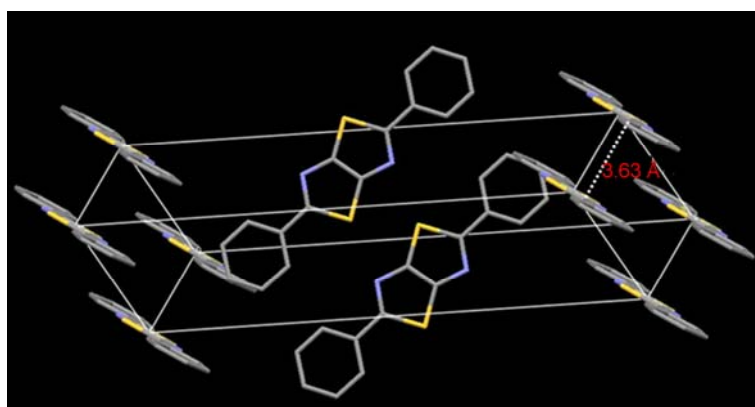


Fig. S10 Crystal packing structure of **5**.

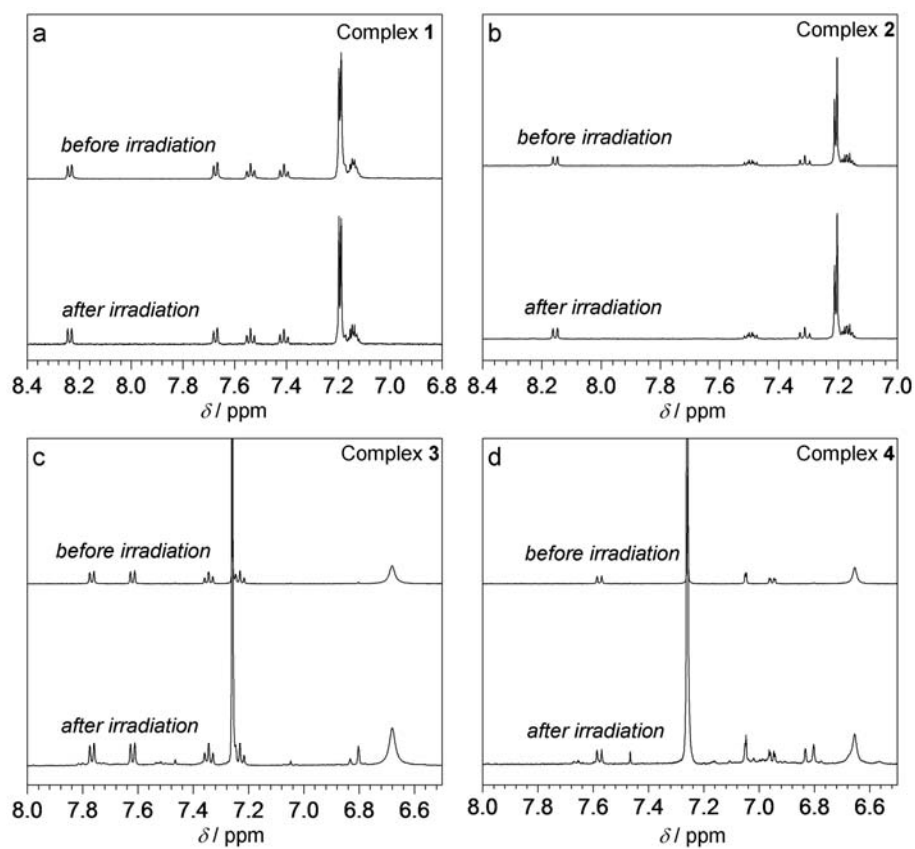


Fig. S11 Proton NMR spectra of boron complexes (**1** and **2** in d_6 -DMSO; **3** and **4** in $CDCl_3$) before and after UV irradiation (365 nm) for 2 hours.

Table S1. Crystal Data for Compounds **3** and **5**

	3	5
formula	C ₅₂ H ₅₂ B ₂ N ₂ S ₂	C ₁₆ H ₁₀ N ₂ S ₂
fw	790.72	294.40
crystal system	monoclinic	monoclinic
space group	P2 ₁ /c	P2 ₁ /c
<i>a</i> (Å)	9.0889(18)	5.735(4)
<i>b</i> (Å)	14.676(3) Å	5.124(3)
<i>c</i> (Å)	16.295(3)	23.1380(16)
α (deg)	90	90
β (deg)	94.18(3)	101.01(3)
γ (deg)	90	90
<i>V</i> (Å ³)	2167.8(7)	667.5(7)
<i>Z</i>	2	2
<i>D_c</i> (g cm ⁻³)	1.211	1.465
θ_{max} (deg)	27.48	27.45
no. of reflns meads	20384	6138
no. of reflns used	4954	1529
no. of parameters	268	91
<i>R</i> _{int}	0.1747	0.0174
final <i>R</i> [<i>I</i> > 2σ(<i>I</i>)]		
<i>R</i> 1	0.1086	0.0395
w <i>R</i>	0.2000	0.1135
<i>R</i> (all data)		
<i>R</i> 1	0.2464	0.0439
w <i>R</i> 2	0.2605	0.1176
GOF on <i>F</i> ²	1.064	1.069

Table S2. Selected Bond Lengths (Å) and Angles (deg) for **3^a** and **5^b**

Complex 3			
B(1)-C(18)	1.628(8)	C(6)#1-B(1)-N(3)	93.8(3)
B(1)-C(9)	1.630(8)	C(1)-C(6)-B(1)#1	131.8(4)
B(1)-C(6)#1	1.630(7)	C(5)-C(6)-B(1)#1	112.8(4)
B(1)-N(3)	1.704(6)	N(3)#1-C(7)-C(5)	113.2(4)
C(5)-C(7)	1.450(6)	N(3)#1-C(7)-S(1)	115.8(3)
C(6)-B(1)#1	1.630(7)	C(5)-C(7)-S(1)	131.0(3)
C(7)-N(3)#1	1.335(5)	C(8)#1-C(8)-N(3)	114.8(5)
C(7)-S(1)	1.712(4)	C(8)#1-C(8)-S(1)	110.8(4)
C(8)-C(8)#1	1.371(8)	N(3)-C(8)-S(1)	134.3(3)
C(8)-N(3)	1.378(5)	C(10)-C(9)-C(14)	116.1(5)
C(8)-S(1)	1.725(4)	C(10)-C(9)-B(1)	120.6(5)
C(18)-B(1)-C(9)	116.5(4)	C(14)-C(9)-B(1)	123.2(5)
C(18)-B(1)-C(6)#1	102.9(4)	C(7)#1-N(3)-C(8)	109.7(4)
C(9)-B(1)-C(6)#1	124.7(5)	C(7)#1-N(3)-B(1)	110.9(3)
C(18)-B(1)-N(3)	117.6(4)	C(8)-N(3)-B(1)	138.6(3)
C(9)-B(1)-N(3)	100.0(4)	C(7)-S(1)-C(8)	88.8(2)
Complex 5			
S(1)-C(8)	1.7105(19)	C(8)-S(1)-C(7)	88.46(8)
S(1)-C(7)	1.7593(17)	C(7)-N(1)-C(8)#1	107.10(14)
N(1)-C(7)	1.330(2)	N(1)-C(7)-C(6)	123.44(14)
N(1)-C(8)#1	1.376(2)	N(1)-C(7)-S(1)	116.30(13)
C(8)-C(8)#1	1.372(3)	C(6)-C(7)-S(1)	120.26(12)
C(6)-C(7)	1.471(2)	C(8)#1-C(8)-S(1)	109.71(16)

Symmetry codes: ^a #1, -x+1, -y, -z ; ^b #1, -x + 2, -y + 2, -z + 2.