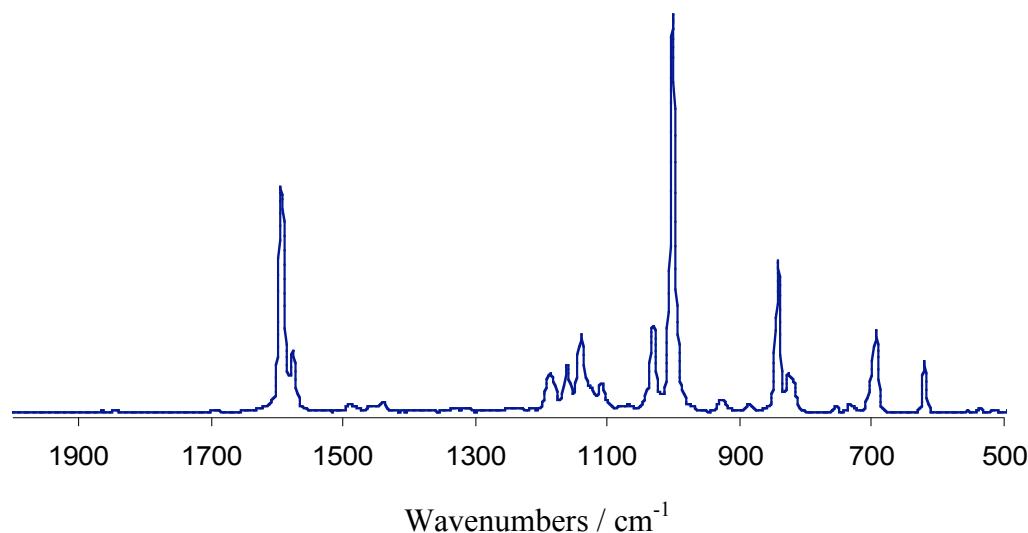


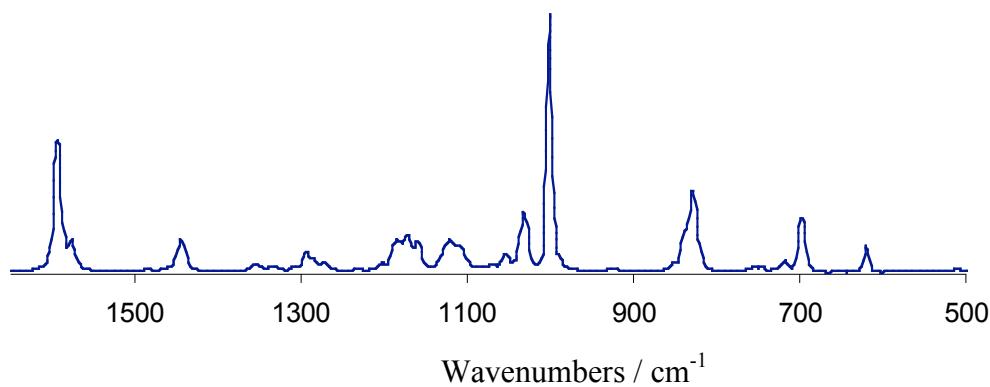
## Probing the local coordination environment and nuclearity of uranyl(VI) complexes in non-aqueous media by emission spectroscopy

Michael P. Redmond,<sup>a</sup> Stephanie M. Cornet,<sup>\*a</sup> Sean D. Woodall,<sup>a</sup> Daniel Whittaker,<sup>a</sup> David Collison,<sup>a</sup> Madeleine Helliwell,<sup>a</sup> and Louise S. Natrajan,<sup>\*a</sup>

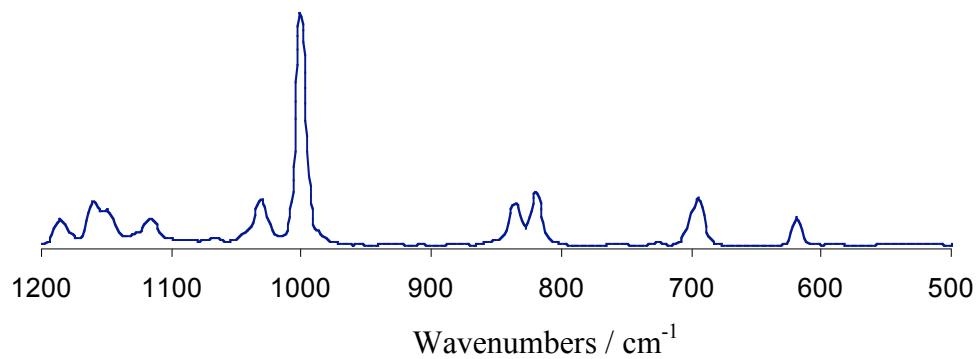
### SUPPLEMENTARY INFORMATION



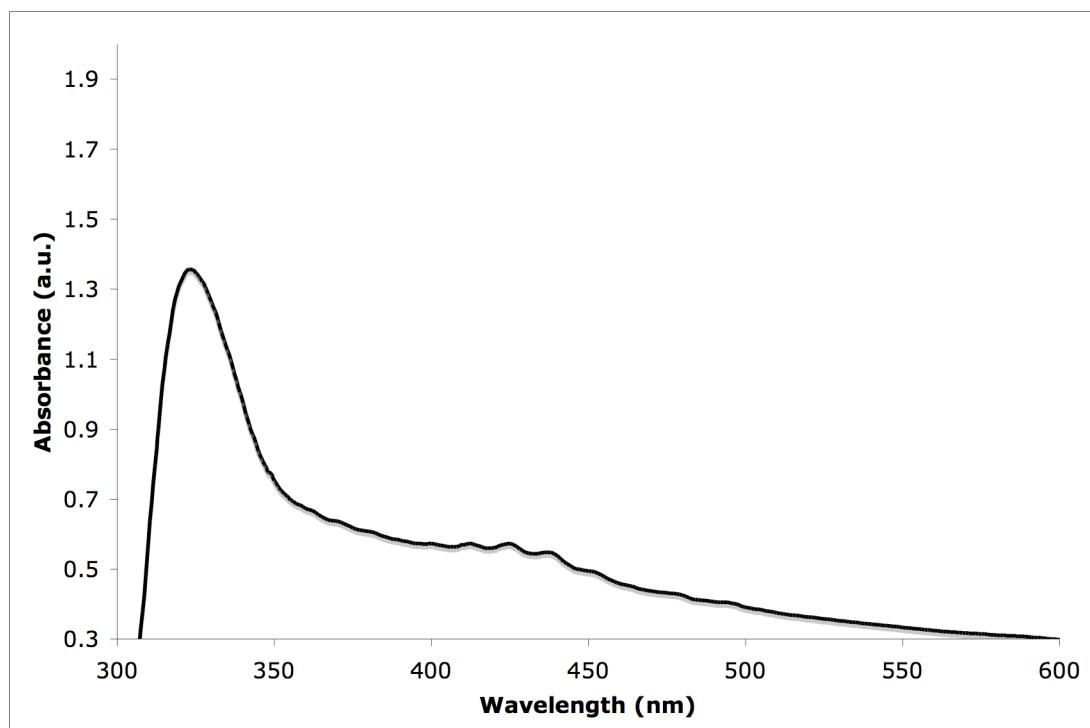
**Figure S1.** Raman spectrum (in absorbance) between 2000–500 cm<sup>-1</sup> of crystalline [UO<sub>2</sub>(TPIP)<sub>2</sub>(THF)] (**4**).



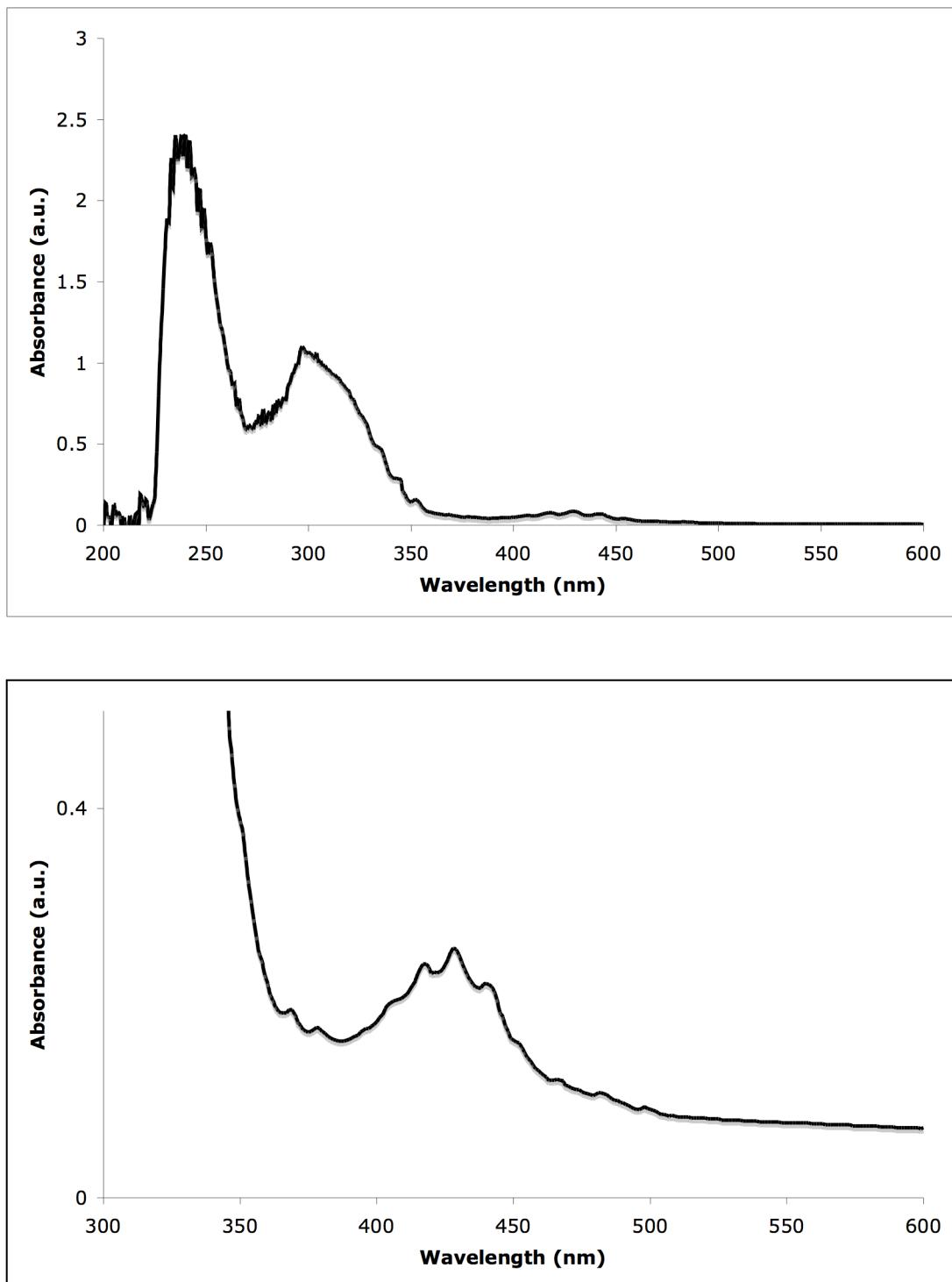
**Figure S2.** Raman spectrum (in absorbance) between 1650–500 cm<sup>-1</sup> of crystalline [UO<sub>2</sub>(TPIP)<sub>2</sub>(OPCy<sub>3</sub>)] (**5**).



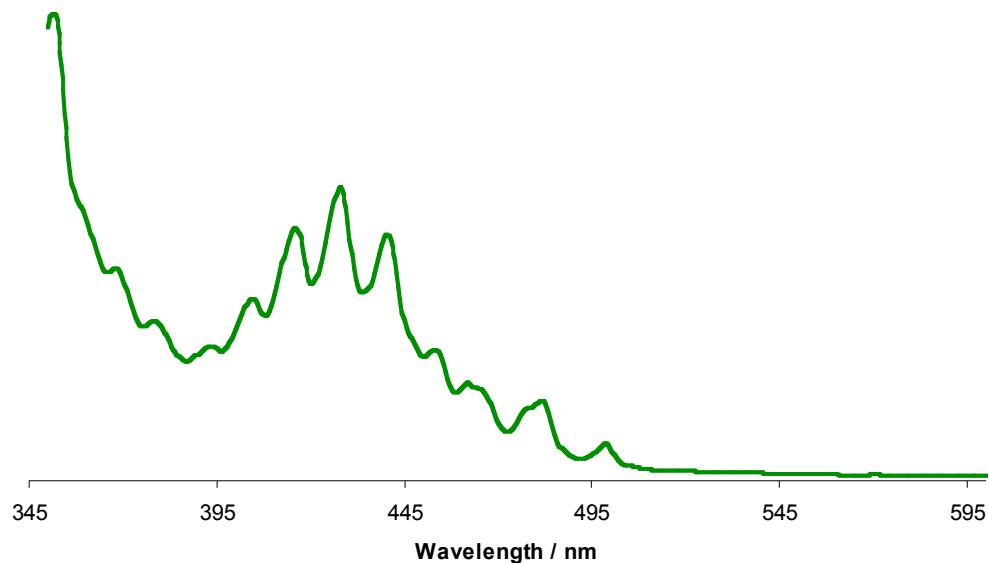
**Figure S3.** Raman spectrum (in absorbance) between 1200–500  $\text{cm}^{-1}$  of powdered  $[\text{UO}_2(\text{TPIP})_2]_3$  (**7**)



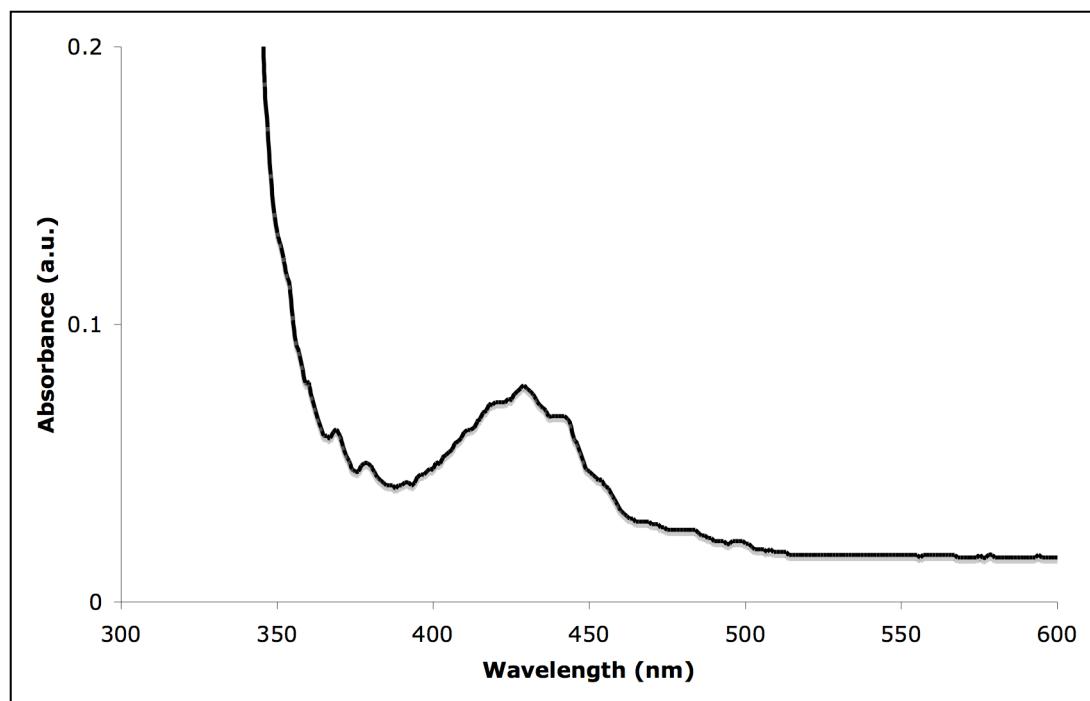
**Figure S4.** UV-vis spectrum of  $[\text{UO}_2\text{Cl}_2(\text{Ph}_3\text{AsO})]$  (**2**) in  $\text{CH}_2\text{Cl}_2$  (295 K)



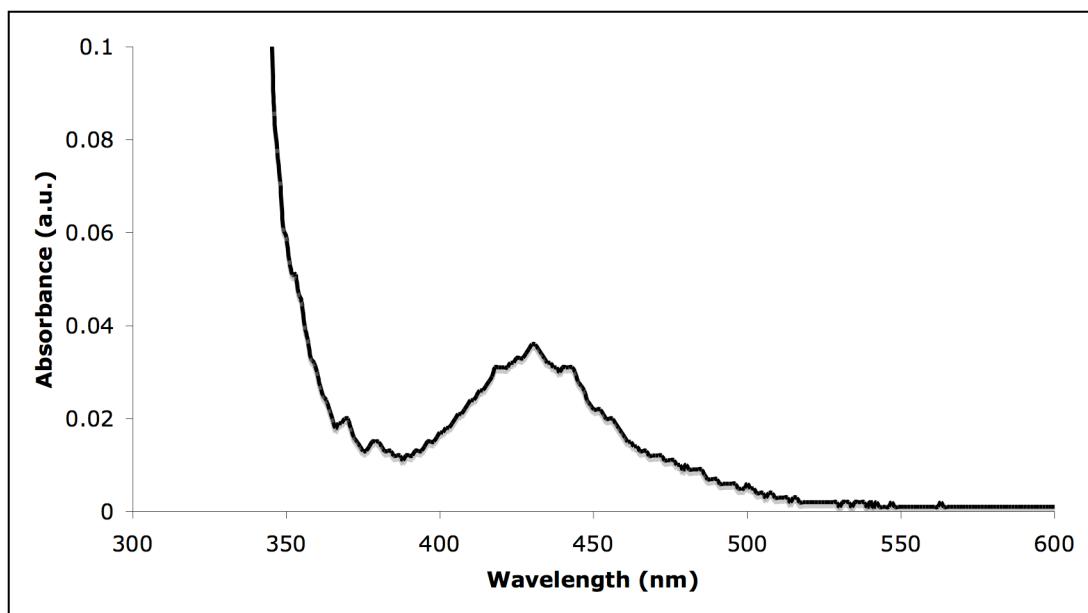
**Figure S5.** UV-vis spectrum of  $[UO_2(\text{TPIP})_2(\text{thf})]$  (**4**) in  $\text{CH}_2\text{Cl}_2$  (295 K)



**Figure S6.** UV-vis spectrum of  $[\text{UO}_2(\text{TPIP})_2(\text{Cy}_3\text{PO})]$  (**5**) in  $\text{CH}_2\text{Cl}_2$  (295 K)

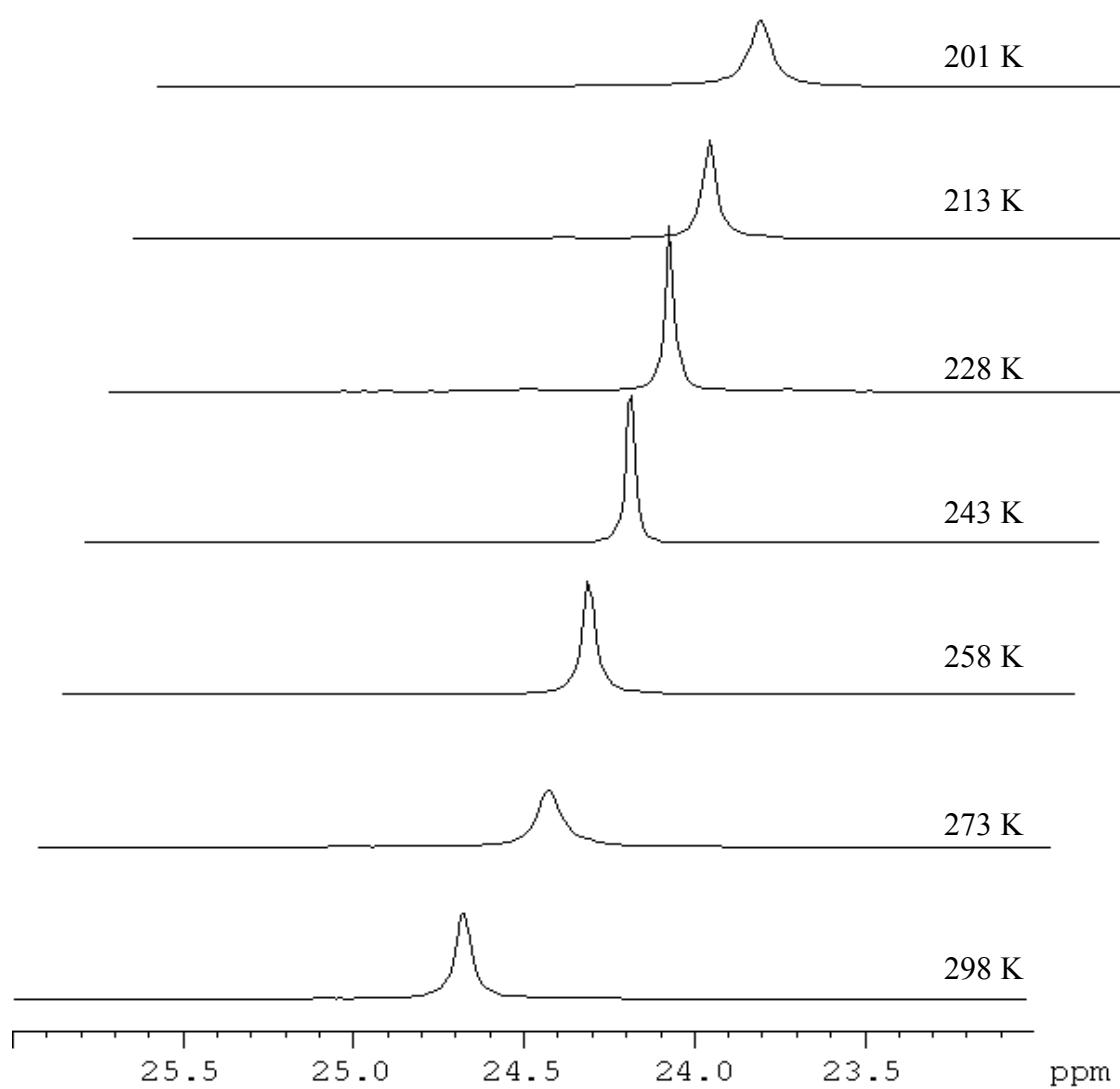


**Figure S7.** UV-vis spectrum of  $[\text{UO}_2(\text{TPIP})_2]_2$  (**6**) in  $\text{CH}_2\text{Cl}_2$  (295 K)

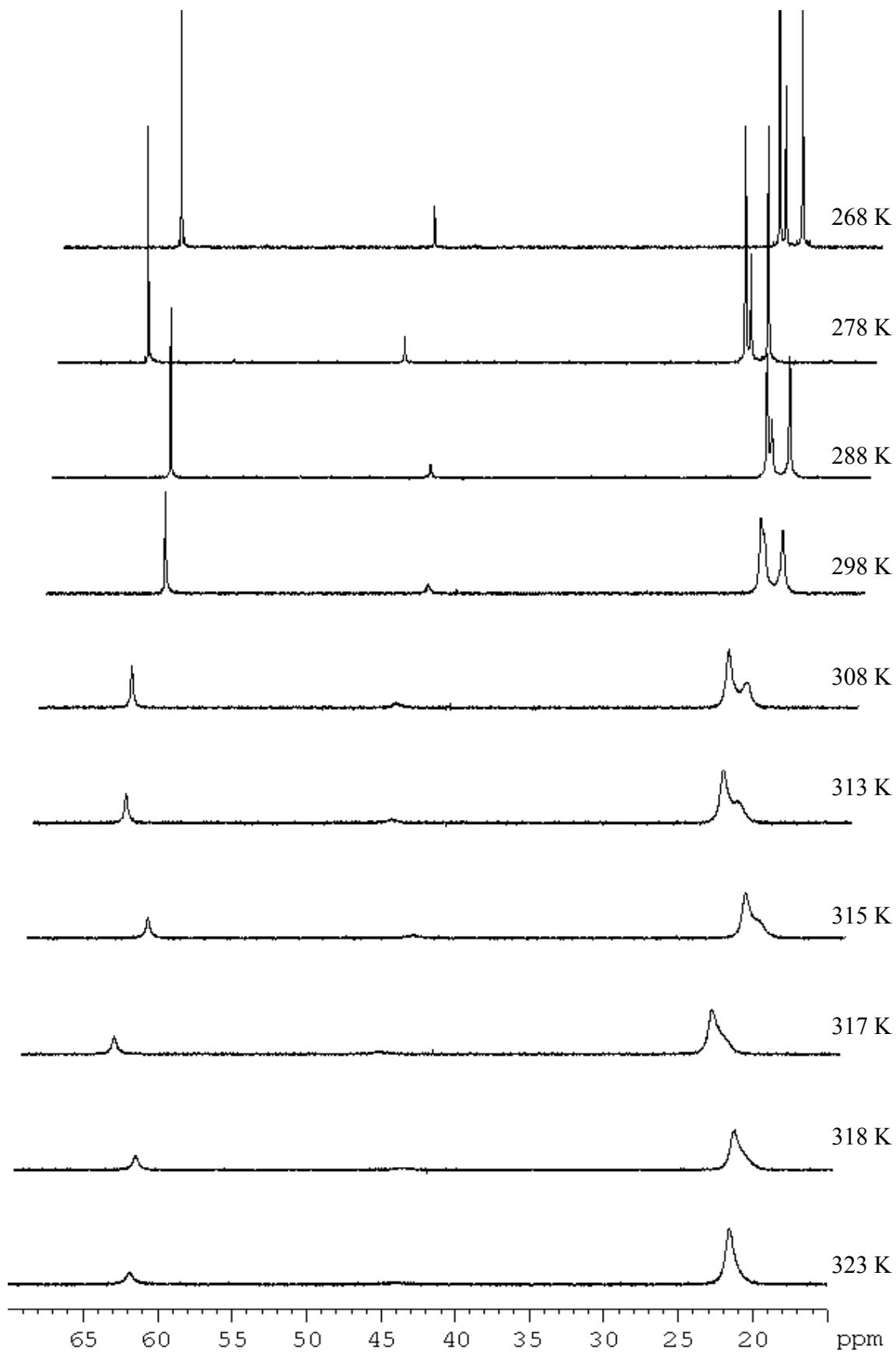


**Figure S8.** UV-vis spectrum of  $[\text{UO}_2(\text{TPIP})_2]_3$  (**7**) in  $\text{CH}_2\text{Cl}_2$  (295 K)

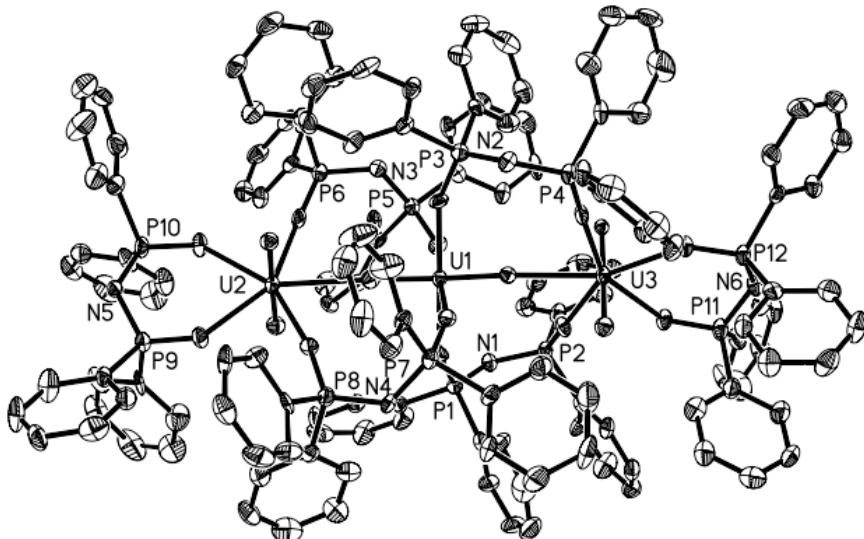
**Figure S9.** VT (298–201 K)  $^{31}\text{P}\{\text{H}\}$  NMR spectra of  $[\text{UO}_2(\text{TPIP})_2(\text{THF})]$  (**4**) in  $\text{CD}_2\text{Cl}_2$  (spectra offset).



**Figure S10.** VT (298–213 K)  $^1\text{H}$  NMR spectra of  $[\text{UO}_2(\text{TPIP})_2(\text{OPCy}_3)]$  (**5**) in  $\text{CD}_2\text{Cl}_2$ .



**Figure S11.** Thermal ellipsoid drawing of  $[\text{UO}_2(\text{TPIP})_2]_3$  (**7**) at the 50 % probability level; H atoms and lattice solvent molecules not shown.



**Table S1.** Data collection and structural refinement for **7**

| Complex  | 7.0.5C <sub>6</sub> H <sub>14</sub>  |
|--|--|
| Empirical formula  | C <sub>144</sub> H <sub>120</sub> N <sub>6</sub> O <sub>18</sub> P <sub>12</sub> U <sub>3</sub> , 0.5(C <sub>6</sub> H <sub>14</sub> ) |
| Formula weight   | 3351.28  |
| Temperature (K)  | 100(2)   |
| Wavelength (Å)   | 0.71073  |
| Crystal system , space group                                 | Monoclinic, P2 <sub>1</sub> /c   |
| a, b, c (Å)  | 20.5618(15),<br>22.3000(16), 29.998(2)   |
| α, β, γ (°).   | 90, 92.282(1), 90  |
| Volume (Å <sup>3</sup> )                                     | 13744.1(17)  |
| Z  | 4  |
| Density (calculated) (Mg/m <sup>3</sup> )                    | 1.620  |
| Absorption coefficient (mm <sup>-1</sup> )                   | 3.734  |
| Crystal form, colour   | Plate, yellow  |
| Crystal size (mm <sup>3</sup> )                              | 0.20 x 0.20 x 0.05   |
| Theta range for data collection (°)                          | 2.08 to 25.03  |
| Index ranges   | -24<=h<=24, -26<=k<=26, -35<=l<=35   |
| No. of measured, independent, observed reflections (I>2σ(I)) | 97209, 24254, 15597  |
| R(int)   | 0.1161   |
| Absorption correction  | Semi-empirical from equivalents  |

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|   |   |
|---|---|
| T <sub>max</sub> , T <sub>min</sub>                             | 1.000, 0.668                                |
| Refinement method   | Full-matrix least-squares on F <sup>2</sup> |
| Data / restraints / parameters                                  | 24254 / 96 / 1676                           |
| R[F <sup>2</sup> > 2σ(F <sup>2</sup> )], wR(F <sup>2</sup> ), S | 0.0506, 0.0805, 1.109                       |
| Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )      | 1.316, -1.359                               |

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