Electronic Supplementary Information

New metal-anion radical framework materials: Co^{II} compounds showing ferromagnetic to antiferromagnetic phase transition at about 344 K, and Zn^{II} compounds exhibiting terminal anion ligand induced direct white-light-emission

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Table S1 Selected bond distances (Å) and angles (°) for compounds 1, 2 and 4^a

1^{b}		2 ^c		4 ^{<i>c</i>}	
Zn(1)-O(1)	1.944(3)	Zn(1)-O(1)	1.946(2)	Co(1)-O(1)	1.9271(15)
Zn(1)-O(2)	1.966(4)	Zn(1)-O(2)	1.962(2)	Co(1)-O(2)	1.9561(14)
Zn(1)-N(1)	2.008(4)	Zn(1)-N(1)	2.015(2)	Co(1)-N(1)	2.0197(18)
Zn(1)-N(3)#2	1.981(4)	Zn(1)-N(3)#2	1.979(2)	Co(1)-N(3)#2	2.0044(18)
C(14)-O(1)	1.284(6)	C(14)-O(1)	1.270(3)	C(14)-O(1)	1.288(2)
O(1)-Zn(1)-O(2)	99.42(19)	O(1)-Zn(1)-O(2)	102.86(10)	O(1)-Co(1)-O(2)	104.04(7)
O(1)-Zn(1)-N(1)	96.24(16)	O(1)-Zn(1)-N(1)	96.45(9)	O(1)-Co(1)-N(1)	96.45(7)
O(1)-Zn(1)-N(3)#2	108.39(16)	O(1)-Zn(1)-N(3)#2	107.53(9)	O(1)-Co(1)-N(3)#2	108.85(7)
O(2)-Zn(1)-N(1)	113.96(19)	O(2)-Zn(1)-N(1)	116.21(10)	O(2)-Co(1)-N(1)	119.28(7)
O(2)-Zn(1)-N(3)#2	115.06(19)	O(2)-Zn(1)-N(3)#2	110.61(10)	O(2)-Co(1)-N(3)#2	110.62(7)
N(1)-Zn(1)-N(3)#2	119.45(17)	N(1)-Zn(1)-N(3)#2	120.11(10)	N(1)-Co(1)-N(3)#2	115.39(8)

^{*a*} According that data/crystal quality of 3 is not good, it is only given as a partial determination, the bond distances and angles of compound 3 are not provided.

^{*b*} Symmetry code for compound 1: #2 = -x + 1/2, y + 1/2, -z + 1/2.

^{*c*} Symmetry code for compounds **2** and **4**: #2 = -x, y+1/2, -z+1/2.



Scheme S1 Synthesis of isomorphous zinc-anion radical frameworks from zwitterionic radical.



Fig. S1 PXRD patterns of simulated 1 from the X-ray single-crystal structure, and synthesized 1, 1a and 1b.



Fig. S2 IR spectra of 1, 1a and 1b, showing the characteristic band of the SCN⁻ ligand at 2069 cm⁻¹ and N_3^- ligand at 2059 cm⁻¹, respectively, attributed to the stretching vibration of the SCN⁻ and N_3^- group, respectively.



Fig. S3 PXRD patterns of the simulated and synthesized 3.



Fig. S4 PXRD patterns of the simulated and synthesized 4.



Fig. S5 X-band EPR spectra of 1 (a) and 2 (b) at room temperature.



Fig. S6 X-band EPR spectra of 3 (a) and 4 (b) at room temperature.

Noteworthy, except for one broad high field line (g = 2.240), compound **3** also shows another broad lower field satellite line (Fig. S6a), which probably arises from the magnetic field lying on different orientation of crystals of 1D chain structure.



Fig. S7 2D network of 3 showing the π - π stacking interactions between adjacent chains.



Fig. S8 A topological illustration of the 2D (6, 3) network of compound 2.



Fig. S9 3D framework of 2 as viewed down the *b*-axis, showing the offset π - π stacking interactions between 2D layers.



Fig. S10 3D framework of **4** as viewed down the *b*-axis, showing the offset π - π stacking interactions between 2D layers.



Fig. S11 Solid photoluminescence spectrum of Hbipo⁻⁻ radical.



Fig. S12 Solid state PL spectra of 1 by variation of excitation light under same metrical condition.



Fig. S13 Solid state PL spectra of 1b by variation of excitation light under same metrical condition.



Fig. S14 Solid excitation spectra of 1, 1a and 1b at room temperature.



Fig. S15 Solid PL spectra of compounds 1 (a), 2(b), 3(c) and 4(d).