

2,2'-biquinolines as test pilots for tuning the colour emission of luminescent mesomorphic silver(I) complexes

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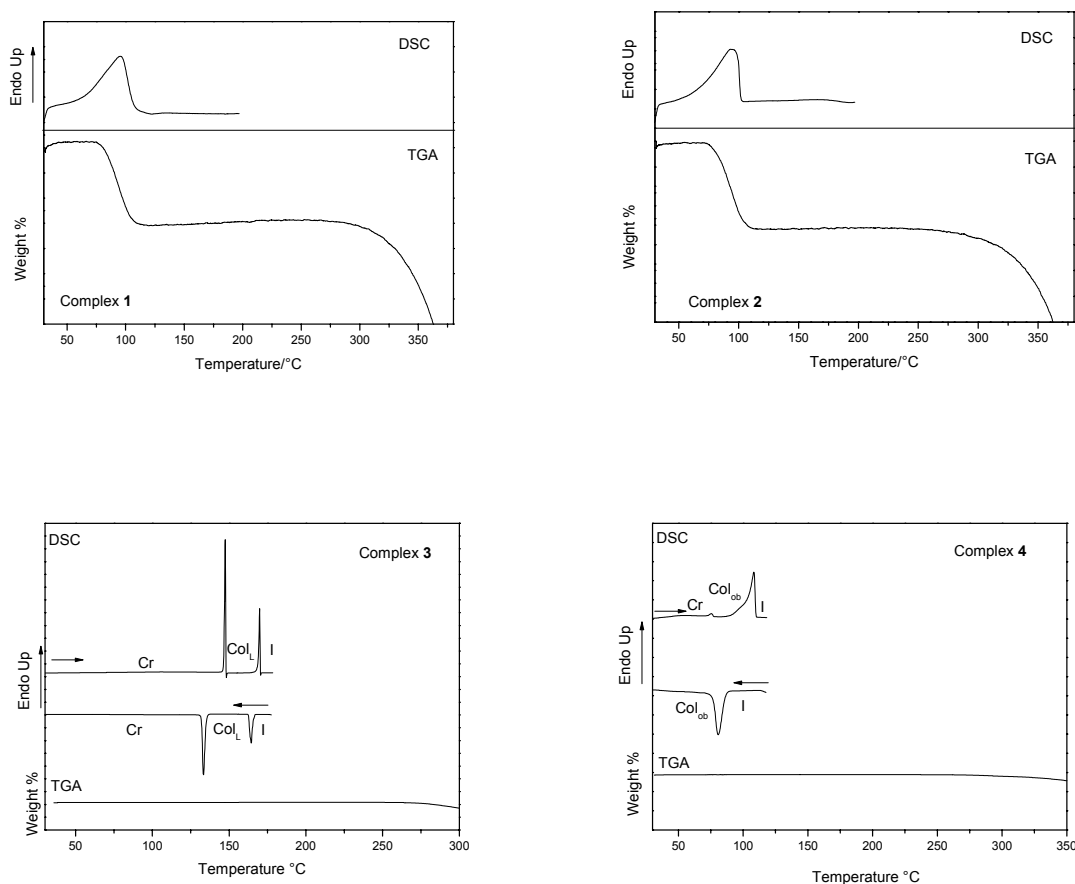
1. DSC and TGA thermograms of Complexes 1 and 2

The thermal stability of complexes **1** - **4** was tested both by thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC).

The TGA scans of both complexes **1** and **2** reveal a desolvation process of one water molecule in a single narrow step (90-100°C), in correspondence with an endotherm peak ($\Delta H = 57.8$ KJ/mol for complex **1** and $\Delta H = 59.4$ KJ/mol for complex **2**) observed by DSC (Figure S1). Thermogravimetric data for the complex **1**: experimental mass loss (3.3%) corresponds to one water molecule (calcd. 3.4%). Thermogravimetric data for the complex **2**: experimental mass loss (2.7%) corresponds to one water molecule (calcd. 2.8%).

For the liquid crystalline complexes **3** and **4**, no thermal processes were detected on TGA scans until 250°C, where decomposition occurs.

Figure S1.



2. PXRD data for complexes 3 and 4.

Table S1

Complex	Mesophase lattice constants/Å	$d_{\text{meas}}/\text{Å}$ ($d_{\text{calcd}}/\text{Å}$)	Miller indice ¹ hk
3	Col _L at 155°C (on heating) $a = 9.79 \text{ Å}$ $b = 8.96 \text{ Å}$ $c = 31.9 \text{ Å}$ $\gamma = 85.6^\circ$	31.5 (31.9)	001
		15.8 (15.9)	002
		10.6(10.6)	003
		9.7 (9.8)	100
		8.9 (8.9)	010
		8.4 (8.3)	102
		7.8 (7.8)	012
		7.2 (7.2)	103
		6.7 (6.7)	111
		6.2 (6.2)	104
		4.9 (4.9)	200
		4.3	h_{CH}
		3.4	h
		4	Col _{obp} at 30°C (on cooling) $a = 42.01 \text{ Å}$ $b = 35.78 \text{ Å}$ $\gamma = 112.3^\circ$
33.3 (33.3)	010		
21.7 (21.7)	110		
13.9 (14.0)	$3\bar{1}0$		
13.0 (13.0)	300		
11.9 (11.9)	$1\bar{3}0$		
10.5 (10.5)	$4\bar{1}0$		
8.8 (8.8)	$1\bar{4}0$		
6.6 (6.6)	050		
6.0 (6.0)	340		
4.4	001, h_{ch}		
3.4	h_0		

3. Absorption data and spectra of L¹, L³, 1 and 3

Table S2. Absorption maxima recorded in dichloromethane solution at room temperature

compound	Abs, $\lambda/nm(\epsilon/M^1cm^{-1})$
L ¹	365(5000), 338 (21200), 326(25000), 315 (21200), 300(sh), 290(sh), 260(69200)
L ²	360(sh), 350 (sh) ,340 (24900), 325(sh), 270(43800)
L ³	380(1270), 350(sh), 340 (25160), 320(sh), 270(46500)
L ⁴	366(1180), 340(sh), 330(25800), 310(sh), 300(sh), 280(sh), 262(83300)
1	380(sh), 356(26200), 340(21950), 330(sh), 295(10260), 284(13080), 265(73400)
2	390(sh), 370 (24200), 360(sh), 312 (10600), 300(sh), 274(60200)
3	400(sh), 367(24150), 360(sh), 312(10920), 305(sh), 274(61300)
4	380(sh), 358 (16400), 330(sh), 310(sh), 300(31200), 280(sh), 265(92600)

Figure S2. Absorption spectra of L¹ (a), L³ (b), 1 (c) and 3 (d) in dichloromethane solution

