SUPPORTING INFORMATION

Bridging, Terminal, Free, Anti, and Gauche BPA Ligands in a 1D Hybrid Organic–Inorganic Mn^{II}-Cyanate System

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Figure S1. Asymmetric unit of compound **1**.

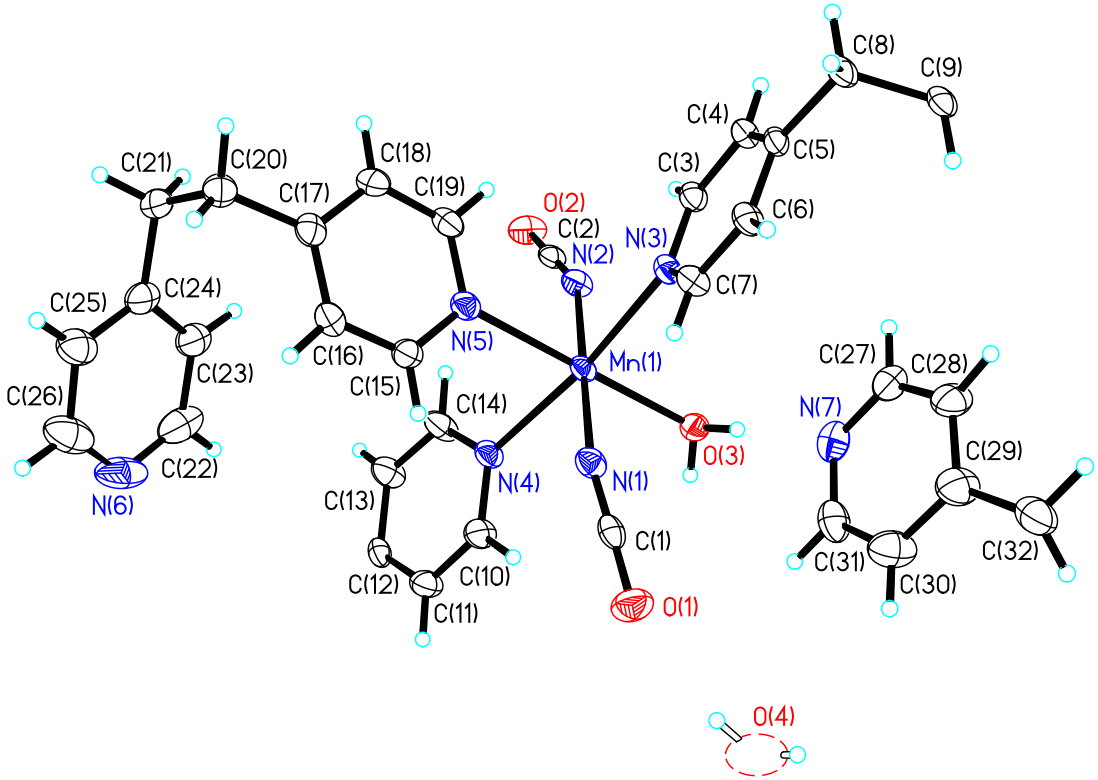

Figure S2. IR Spectrum for compound **1**.

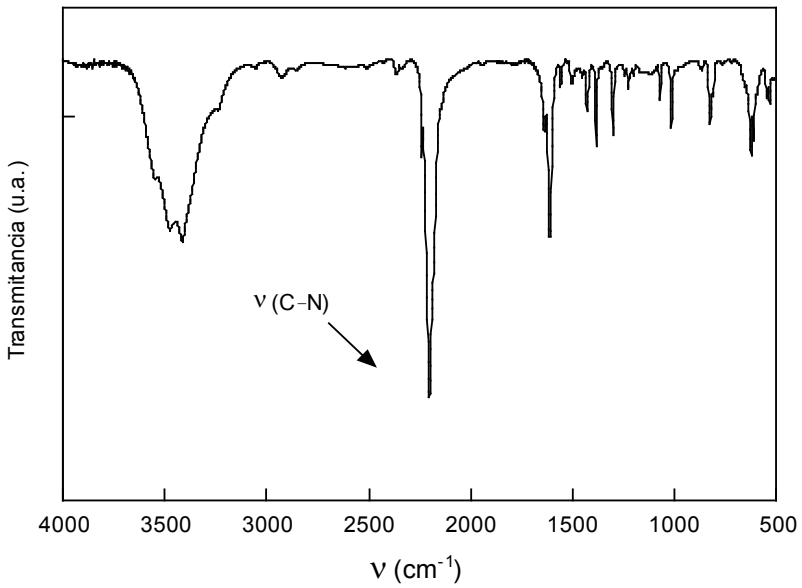

Figure S3. ESR spectra at 200 K and 100 K for compound **1**.

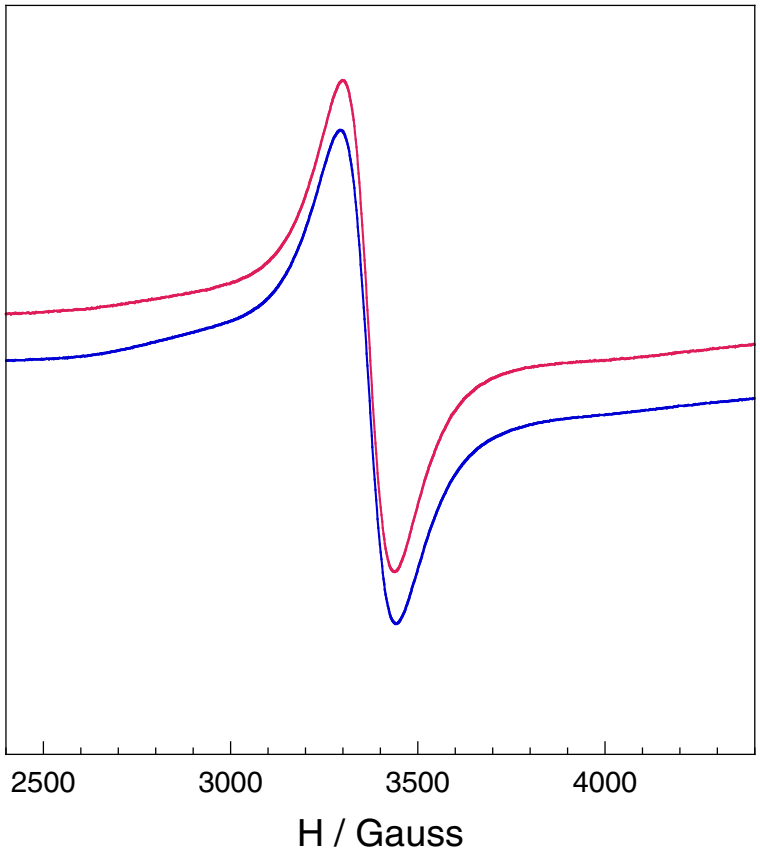

Figure S4. Thermal evolution of $1/\chi_m$ and Curie-Weiss law for compound **1**.

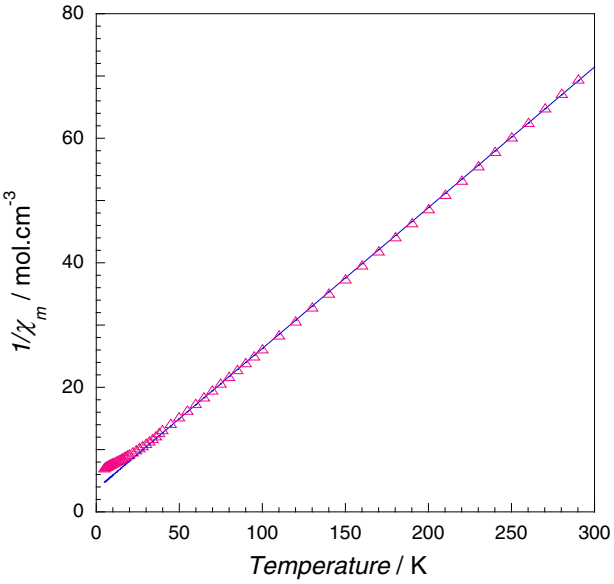

Table S1. Thermogravimetric data for compound **1**.

COMPOUND	STEP	T_i [°C]	T_f [°C]	ΔT	$\%\Delta m$ experiment	$\%\Delta m$ theoretic	ASIGNATION (per unit formula)
(1)	1	54	115	61	4.06	3.63	Loss of water molecules
	2	115	235	120	42.62	44.43	Loss of 1.5 bpa molecules
	3	235	335	100	38.20	38.0	Pyrolysis of the other bpa molecule and the two cyanate groups
	TOTAL	54	335	281	84.88	86.06	

T_i = Initial temperature, T_f = Final temperature, $\Delta T = T_f - T_i$, $\%\Delta m_{experiment}$ = Experimental mass loss percentage, $\%\Delta m_{theortetic}$ = Theoretical mass loss percentage.