

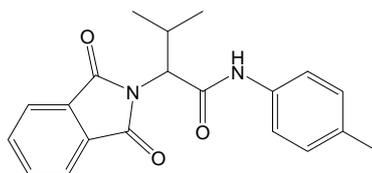
Experimental

General methods and instrumentation:

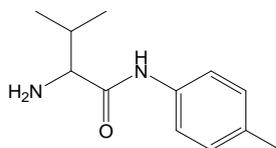
All manipulations of air and moisture sensitive species were performed under an atmosphere of argon or dinitrogen using standard Schlenk and glove box techniques. Solvents were dried by passing through an alumina drying column incorporated into a MBraun SPS800 solvent purification system, except in the case of tetrahydrofuran (THF), which was dried over potassium and distilled under argon. All solvents were degassed and stored under argon in Teflon valve ampoules. Deuterated chloroform was passed through a column of basic alumina before being stored over 4Å molecular sieves prior to use. Benzene-d₆ and THF-d₈ were dried over potassium under an argon atmosphere before being vacuum transferred, freeze pump thaw degassed and stored in a glove box. All other reagents were purchased from commercial suppliers and used as received unless otherwise stated.

Air sensitive samples for NMR spectroscopy were prepared in a glovebox under a dinitrogen atmosphere using 5 mm Nolan NMR tubes equipped with J. Young Teflon valves. All other samples were prepared in Wilmad 5 mm NMR tubes. NMR spectra were recorded on Bruker Avance DPX 250, 400, 500 or Jeol Eclipse 300 spectrometers. NMR spectra are quoted in ppm and were referenced internally relative to the residual protio-solvent (¹H) or solvent (¹³C) resonances, or externally to CFCl₃ (¹⁹F); all coupling constants are quoted in Hertz. In all cases, NMR assignments were confirmed by the use of two-dimensional ¹H-¹H or ¹H-¹³C correlation experiments (HSQC and HMBC). Mass spectra were recorded on a Waters LCT Premier XE or Waters GCT Premier mass spectrometer by the mass spectrometry service at the School of Chemistry, Cardiff University. Infrared spectra were prepared as liquid films on NaCl plates, or KBr pellets and were recorded on a Jasco 660-Plus FT/IR spectrometer. Infrared data are quoted in wavenumbers (cm⁻¹).

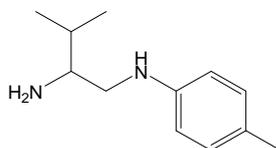
Preparation of Diamines (1a – c): Diamines **1b** and **1c** were prepared according to literature procedures.¹ Herein one new diamine analogue has been prepared from modified literature methods and in a similar fashion, data are reported for each synthetic step for H₂NCH(ⁱPr)CH₂NHC₆H₄CH₃ (HNN^{ArMe}, **1a**).



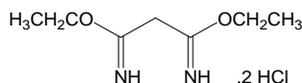
Protected HNN^{ArMe}: ¹H NMR data (CDCl₃, 400.1 MHz, 293 K): δ_H = 7.89 (2H, dd, ³J_{HH} = 5.4 Hz, ³J_{HH} = 5.4 Hz, PhH), 7.77 (2H, dd, ³J_{HH} = 5.4 Hz, ³J_{HH} = 5.4 Hz, PhH), 7.44 (2 H, d, ³J_{HH} = 8.3 Hz, 4-C₆H₄CH₃), 7.10 (2 H, d, ³J_{HH} = 8.3 Hz, 4-C₆H₄CH₃), 4.51 (1 H, d, ³J_{HH} = 11.6 Hz, PhNCH), 2.91 (1 H, m, CH(CH₃)₂), 2.29 (3 H, s, C₆H₄CH₃), 1.16 (3 H, d, ³J_{HH} = 6.5 Hz, CH(CH₃)₂), 0.89 (3 H, d, ³J_{HH} = 6.5 Hz, CH(CH₃)₂) ppm.



Deprotected HNN^{ArMe}: ¹H NMR data (D₂O, 400.1 MHz, 293 K): δ_H = 7.37 (2 H, d, ³J_{HH} = 8.0 Hz, 4-C₆H₄CH₃), 7.29 (2 H, d, ³J_{HH} = 8.0 Hz, 4-C₆H₄CH₃), 3.97 (1 H, d, ³J_{HH} = 6.2 Hz, H₂NCH), 2.35 (1 H, m, CH(CH₃)₂), 2.34 (3 H, s, C₆H₄CH₃), 1.12 (6 H, d, ³J_{HH} = 6.9 Hz, CH(CH₃)₂) ppm.

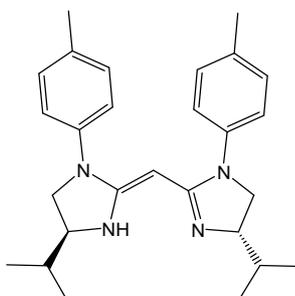


Diamine [HNN^{ArMe}] 1a; ¹H NMR data (CDCl₃, 500.1 MHz, 293 K): δ_H = 7.00 (2 H, d, ³J_{HH} = 8.2 Hz, *o*-C₆H₄CH₃), 6.58 (2 H, d, ³J_{HH} = 8.2 Hz, *m*-C₆H₄CH₃), 3.23 (1 H, dd, ³J_{HH} = 3.4 Hz, ²J_{HH} = 12.0 Hz, NCH₂), 2.84 (1 H, dd, ³J_{HH} = 3.4 Hz, ²J_{HH} = 12.0 Hz, NCH₂), 2.75 (1 H, m, H₂NCH), 2.25 (3H, s, C₆H₄CH₃), 1.69 (1 H, m, (CH₃)₂CH), 0.97 (6 H, dd, ³J_{HH} = 9.4 Hz, ³J_{HH} = 6.8 Hz, CH(CH₃)₂), ppm. ¹³C{¹H} NMR data (CDCl₃, 100.6 MHz, 293 K): δ_C = 146.3 (*ipso*-C₆H₄CH₃), 129.6 (*o*-C₆H₄CH₃), 126.4 (*p*-C₆H₄CH₃), 113.1 (*m*-C₆H₄CH₃), 56.1 (H₂NCH), 48.2 (CH₂), 32.3 ((CH₃)₂CH), 20.3 (C₆H₄CH₃), 19.2 (CH₃ isopropyl), 17.8 (CH₃ isopropyl) ppm. IR data: (NaCl disc, cm⁻¹) 3365 (m), 3029 (w), 2956 (s), 2905 (s), 2871 (s), 2831 (s), 2356 (w), 2065 (w), 1844 (w), 1733 (w), 1618 (m), 1588 (m), 1511 (s), 1464 (s), 1408 (m), 1386 (m), 1376 (m), 1926 (m), 1234 (s), 1179 (m), 1109 (w), 1076 (w), 1038 (s), 911 (w), 820 (s), 731 (w). HR-MS (ES) data: [M]⁺, found (calc. for C₁₂H₂₁N₂) 193.1697 (193.1705).



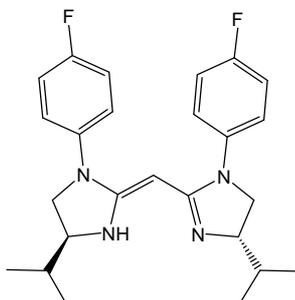
Diethyl malonimidate dihydrochloride; Prepared from a modified literature procedure.² To a solution of malononitrile (10 g, 0.15 mol), in dry 1,4-dioxane (200 ml), ethanol (19.70 ml, 0.33 mol) was added. Dry HCl_(g) was then passed through the solution for 6 hours where upon the system was sealed and allowed to stir under an HCl atmosphere for a further 18 hours. The solution was filtered away from the resulting diethyl malonimidate dihydrochloride precipitate. The precipitate was washed with dry 1,4 dioxane (2 × 20 mls) and dry diethyl ether (20 ml). The desired product was then dried *in vacuo* and stored in a glove box, giving near quantitative yields. Data were consistent with that previously reported.²

General procedure for bisimidazolines (2a–c); Prepared according to modified literature procedures.³ Under an inert atmosphere, diethyl malonimidate dihydrochloride (26 mmol *ca.* 0.6 g) and HNN^R (1 g, 52 mmol) were dissolved in dry dichloromethane and heated at vigorous reflux for 4 days. Proceeding such time the solution was allowed to cool and washed with NaHCO₃ (10%, 2 × 25 ml). The organic solution was subsequently reduced in volume under reduced pressure. The relevant R-BIM was realized by column chromatography over silica gel (typical solvent systems were composed of ethyl acetate/methanol/Et₃N (9:1:1 v/v)) to yield the desired R-BIM as a white solid in approximately 35% yield.

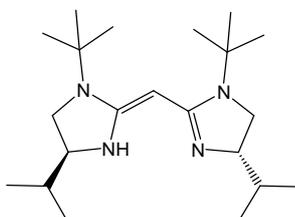


[Ar^{Me}-BIM] 2a; ¹H NMR data (CDCl₃, 500.1 MHz, 293 K): δ_H = 7.04 (8 H, app d, ³J_{HH} = 1.5 Hz, *m*-C₆H₄CH₃ and *o*-C₆H₄CH₃), 3.79 (2 H, m, NCH₂), 3.63 (2 H, m, NCH₂CH), 3.44 (2 H, m, NCH₂), 2.28 (6 H, s, C₆H₄CH₃), 1.72 (2 H, m, (CH₃)₂CH), 1.09 (6 H, br d, ³J_{HH} = 6.5Hz, CH(CH₃)₂), 0.93 (6 H, dd, ³J_{HH} = 6.5Hz, CH(CH₃)₂) ppm. ¹³C{¹H} NMR data (CDCl₃, 100.6 MHz, 293 K): δ_C = 159.9 (C=N), 139.9 (*ipso*-C₆H₄CH₃), 132.5 (*p*-C₆H₄CH₃), 129.3 (*o*-C₆H₄CH₃), 122.2 (*m*-C₆H₄CH₃), 65.0 (C=NCH imidazoline), 55.5 (CH₂ imidazoline), (CH₂ bridge not observed), 33.8 ((CH₃)₂CH), 20.7 (C₆H₄CH₃) 19.6 (CH₃ isopropyl), 19.1 (CH₃ isopropyl) ppm. IR data: (KBr disc, cm⁻¹) 3031 (w), 2963 (s), 2917 (w), 2867 (w), 2823 (w), 1895 (w), 1623 (s), 1601 (s), 1571 (m), 1533 (s), 1509 (s), 1471 (m), 1385 (m), 1324 (m), 1260 (s), 1194 (m), 1171

(m), 1105 (s), 1023 (s), 862 (m), 801 (s), 722 (m), 711 (m), 631 (w), 550 (m), 502 (w).
Accurate mass ES-MS for $[\text{Ar}^{\text{Me}}\text{-BIM} + \text{H}]^+$: $m/z = 417.3018$ (calcd for $\text{C}_{27}\text{H}_{37}\text{N}_4$: 417.3006).

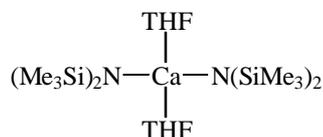


[Ar^F-BIM] 2b; ¹H NMR data (CDCl₃, 500.1 MHz, 293 K): $\delta_{\text{H}} = 7.06$ (4 H, dd, ³J_{HH} = 8.5 Hz, *o*-C₆H₄F), 6.93 (4 H, app t, ³J_{HH} = 8.5 Hz, *m*-C₆H₄F), 3.75 (2 H, m, NCH₂), 3.64 (2 H, m, NCH₂CH), 3.39 (2 H, m, NCH₂), 1.72 (2 H, m, (CH₃)₂CH), 1.09 (6 H, br d, ³J_{HH} = 6.3 Hz, CH(CH₃)₂), 0.93 (6 H, dd, ³J_{HH} = 6.3 Hz, CH(CH₃)₂) ppm. ¹⁹F NMR data (CDCl₃, 282.8 MHz, 293 K): $\delta_{\text{F}} = -118.45$ (br s, *p*-C₆H₄F) ppm. ¹³C{¹H} NMR data (CDCl₃, 100.6 MHz, 293 K): $\delta_{\text{C}} = 160.3$ (C=N), 159.1 (d, ¹J_{CF} = 243.1 Hz, *p*-C₆H₄F), 138.4 (*ipso*-C₆H₄F), 124.3 (d, ³J_{CF} = 7.5 Hz, *o*-C₆H₄F), 115.5 (d, ²J_{CF} = 22.4 Hz, *m*-C₆H₄F), 65.2 (C=NCH imidazoline), 55.8 (CH₂ imidazoline), (CH₂ bridge not observed), 33.8 ((CH₃)₂CH), 19.6 (CH₃ isopropyl), 19.1 (CH₃ isopropyl) ppm. IR data: (KBr disc, cm⁻¹) 3111 (w), 3078 (w), 3051 (w), 2959 (s), 2831 (s), 2608 (w), 1876 (m), 1616 (s), 1593 (s), 1536 (s), 1504 (s), 1469 (m), 1425 (m), 1390 (m), 1363 (m), 1306 (s), 1256 (s), 1232 (s), 1194 (s), 1168 (m), 1119 (s), 1090 (m), 1046 (m), 1010 (m), 904 (w), 832 (s), 799 (s), 723 (s), 638 (m), 593 (m), 564 (m), 551 (s), 503 (m), 435 (m). Accurate mass ES-MS for $[\text{Ar}^{\text{F}}\text{-BIM} + \text{H}]^+$: $m/z = 425.2509$ (calcd. for $\text{C}_{25}\text{H}_{31}\text{N}_2\text{F}_2$: 425.2517).



[^tBu-BIM] 2c; ¹H NMR data (CDCl₃, 500.1 MHz, 293 K): $\delta_{\text{H}} = 3.42$ (2 H, app t, ³J_{HH} = 7.9 Hz, NCH₂), 3.25 (2 H, m, NCH₂CH) 2.85 (2 H, app t, ³J_{HH} = 8.8 Hz, NCH₂), 1.54 (2 H, m, (CH₃)₂CH), 1.33 (9 H, s, C(CH₃)₃), 1.00 (6 H, d, ³J_{HH} = 6.5 Hz, CH(CH₃)₂), 0.83 (6 H, d, ³J_{HH} = 6.5 Hz, CH(CH₃)₂) ppm. ¹³C{¹H} NMR data (CDCl₃, 100.6 MHz, 293 K): $\delta_{\text{C}} = 162.2$ (C=N), 63.8 (C=NCH imidazoline), 52.4 (CH₂ imidazoline), 52.0 (C(CH₃)₃), (CH₂ bridge not observed), 33.7 ((CH₃)₂CH), 28.7 (C(CH₃)₃) 19.9 (CH₃ isopropyl), 19.2 (CH₃ isopropyl) ppm. IR data: (KBr disc, cm⁻¹) 3465 (m), 2964 (s), 2867 (m), 2810 (m), 1604 (s), 1529 (s), 1469 (m), 1391 (m), 1362

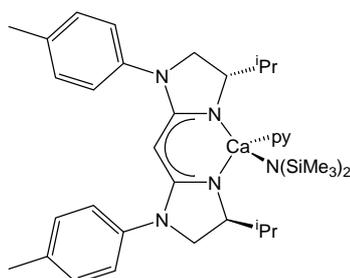
(m), 1294 (m), 1264 (s), 1227 (s), 1166 (w), 1122 (m), 1095 (m), 1065 (m), 1017 (m), 973 (w), 921 (w), 802 (s), 732 (s), 698 (w), 646 (w), 547 (w), 468 (w). Accurate mass ES-MS for $[\text{tBu-BIM} + \text{H}]^+$: $m/z = 349.3344$ (calcd. for $\text{C}_{21}\text{H}_{41}\text{N}_4$: 349.3331).



[Ca{N(SiMe₃)₂}₂(THF)₂]: CaI₂ (1.47g, 5 mmol) was added to K(N(SiMe₃)) (1.99g, 10 mmol) and dissolved in dry THF and allowed to stir for 18 hours under an inert atmosphere. The solution was filtered away from the resulting salt precipitate and the solvent removed under reduced pressure to yield the white solid [Ca{N(SiMe₃)₂}₂(THF)₂]. This was further dried *in vacuo* over night to approximately 2×10^{-2} mbar.^{4, 5} Data were consistent with that previously reported.⁶

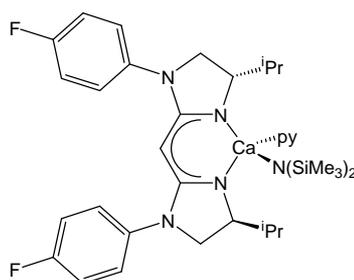
General procedure for the preparation of [Ca(R-BIM){N(SiMe₃)₂}(py)] (3a–c):

[Ca{N(SiMe₃)₂}₂(py)₂] was prepared from the corresponding analogue [Ca{N(SiMe₃)₂}₂(THF)₂] (*vide supra*) via literature procedures.⁶ R-BIM (30mg, 0.07 mmol) in dry toluene (0.5ml) was added to 1 equivalent of [Ca{N(SiMe₃)₂}₂(py)₂] (0.07 mmol *ca.* 0.037 g). The solution was agitated briefly and allowed to react for 18 hours at ambient temperature under an inert atmosphere. The resulting complex was typically realised as an orange powder when dried *in vacuo* to approximately 4×10^{-2} mbar.

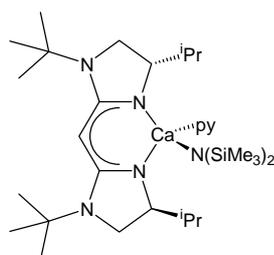


[Ca(Ar^{Me}-BIM){N(SiMe₃)₂}(py)] 3a: ¹H NMR data (*d*₈-THF, 500.1 MHz, 293 K): $\delta_{\text{H}} = 8.56$ (4 H, m, 2,6-NC₅H₅), 7.64 (2 H, m, 4-NC₅H₅), 7.24 (4 H, m, 3,5-NC₅H₅), 6.97-6.89 (8 H, m, *o*-C₆H₄CH₃ and *m*-C₆H₄CH₃), 3.89 (2 H, m, NCH₂CH), 3.69 (1 H, NCH₂), 3.67 (1 H, s, CH bridge), 3.51 (1 H, NCH₂), 3.34 (1 H, NCH₂), 3.24 (1 H, NCH₂), 2.22 (3 H, br. s, C₆H₅CH₃), 2.20 (3 H, br. s, C₆H₅CH₃), 1.86 (2 H, m, (CH₃)₂CH), 0.87 (6 H, m, CH(CH₃)₂), 0.77 (6 H, app t, ³J_{HH} = 6.7 Hz, CH(CH₃)₂), 0.11 (18 H, s, N(Si(CH₃)₃) ppm. ¹³C{¹H} NMR data (*d*₈-THF, 75.5 MHz, 293 K): $\delta_{\text{C}} = 164.8$ (C=N), 164.1 (C=N), 150.8 (2,6-NC₅H₅), 142.7 (*ipso*-C₆H₄CH₃),}

142.1 (*ipso*-C₆H₄CH₃), 136.3 (4-NC₅H₅), 132.0 (*p*-C₆H₄CH₃), 131.3 (*p*-C₆H₄CH₃), 129.5 (*o*-C₆H₄CH₃), 129.4 (*o*-C₆H₄CH₃), 124.3 (3,5-NC₅H₅), 123.9 (*m*-C₆H₄CH₃), 123.2 (*m*-C₆H₄CH₃), 66.5 (C=NCH_{imidazoline}), 66.1 (C=NCH_{imidazoline}), 58.7 (CH_{bridge}), 51.8 (CH_{2imidazoline}), 51.2 (CH_{2imidazoline}), 34.9 ((CH₃)₂CH), 34.3 ((CH₃)₂CH), 20.8 (C₆H₄CH₃), 20.2 (CH_{3isopropyl}), 15.4 (CH_{3isopropyl}), 2.7 (N(Si(CH₃)₃)₂) ppm. IR data: (KBr disc, cm⁻¹) 3027 (w), 2936 (s), 2867 (w), 2241 (w), 2179 (w), 2129 (w), 1652 (m), 1515 (s), 1455 (w), 1417 (w), 1315 (m), 1261 (s), 1100 (s), 1020 (s), 934 (w), 800 (s), 703 (w), 667 (w), 618 (w), 552 (w), 479 (m).



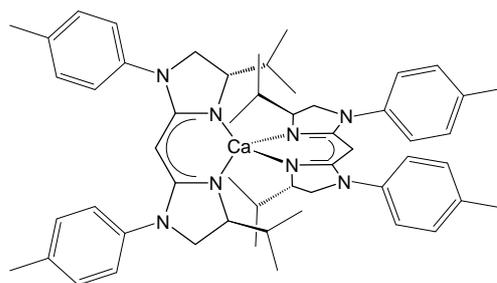
[Ca(Ar^F-BIM){N(SiMe₃)₂}(py)] **3b**: ¹H NMR data (d₈-THF, 500.1 MHz, 293 K): δ_H = 8.54 (2 H, m, 2,6-NC₅H₅), 7.64 (1 H, m, 4-NC₅H₅), 7.24 (2 H, m, 3,5-NC₅H₅), 7.10-7.02 (4 H, m, *o*-C₆H₄F), 6.87 (4 H, app t, ³J = 8.7 Hz, *m*-C₆H₄F), 3.90 (2 H, m, NCH₂CH), 3.52 (2 H, NCH₂), 3.48 (1 H, s, CH_{bridge}), 3.40 (2 H, NCH₂), 2.04 (2 H, m, (CH₃)₂CH), 0.89 (6 H, d, ³J_{HH} = 6.8 Hz, CH(CH₃)₂), 0.78 (6 H, d, ³J_{HH} = 6.8 Hz, CH(CH₃)₂), 0.10 (18 H, s, N(Si(CH₃)₃)₂) ppm. ¹³C{¹H} NMR data (d₈-THF, 75.5 MHz, 293 K): δ_C = 164.4 (C=N), 159.2 (d, ¹J_{CF} = 239.9 Hz, *p*-C₆H₄F), 150.8 (2,6-NC₅H₅), 141.9 (d, ⁴J_{CF} = 2.3 Hz, *ipso*-C₆H₄F), 136.3 (4-NC₅H₅), 125.8 (d, ³J_{CF} = 7.5 Hz, *o*-C₆H₄F), 124.3 (3,5-NC₅H₅), 115.4 (d, ²J_{CF} = 22.5 Hz, *m*-C₆H₄F), 66.5 (C=NCH_{imidazoline}), 57.8 (CH_{bridge}), 52.1 (CH_{2imidazoline}), 34.2 ((CH₃)₂CH), 20.7 (CH_{3isopropyl}), 15.7 (CH_{3isopropyl}), 2.7 (N(Si(CH₃)₃)₂) ppm. ¹⁹F NMR data (d₈-THF, 282.8 MHz, 293 K): δ_F = -121.80 (s, *p*-C₆H₄F) ppm. IR data: (KBr disc, cm⁻¹) 3419 (br. w) 2960 (m), 2873 (w), 2240 (w), 1621 (m) 1593 (w), 1544 (s), 1479 (w), 1428 (w), 1387 (w), 1316 (w), 1260 (m), 1222 (m), 1097 (m), 1018 (m), 837 (m), 798 (m), 720 (w), 668 (w), 594 (w), 551 (m), 422 (w), 417 (w).



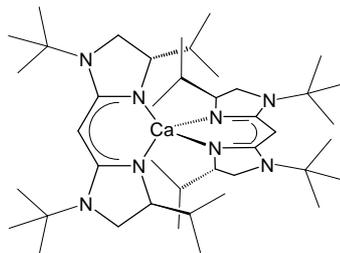
[Ca(^tBu-BIM){N(SiMe₃)₂}(py)] **2c**: ¹H NMR data (d₈-THF, 500.1 MHz, 293 K): δ_H = 8.55 (2 H, m, 2,6-NC₅H₅), 7.64 (1 H, m, 4-NC₅H₅), 7.23 (2 H, m, 3,5-NC₅H₅), 3.83 (1 H, s, CH_{bridge}),

3.63-3.56 (2 H, m, partially obscured by solvent peak, NCH_2CH), 3.20-3.12 (2 H, m, NCH_2), 3.10-3.02 (2 H, m, NCH_2), 1.84-1.70 (2 H, m, partially obscured by solvent peak $(\text{CH}_3)_2\text{CH}$), 1.37 (18 H, s, $\text{C}(\text{CH}_3)_3$), 0.85 (3 H, d, $^3J_{\text{HH}} = 6.9$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.81 (3 H, app t, $^3J_{\text{HH}} = 6.9$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.78 (3 H, d, $^3J_{\text{HH}} = 6.9$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.74 (3 H, app t, $^3J_{\text{HH}} = 6.9$ Hz, $\text{CH}(\text{CH}_3)_2$), 0.10 (18 H, $\text{N}(\text{Si}(\text{CH}_3)_2)$ ppm. $^{13}\text{C}\{^1\text{H}\}$ NMR data (d_8 -THF, 75.5 MHz, 293 K): $\delta_{\text{C}} = 167.0$ (C=N), 150.8 (2,6- NC_5H_5), 136.3 (4- NC_5H_5), 124.3 (3,5- NC_5H_5), 64.5 (C=NCH imidazoline), 62.5 (CH bridge), 53.0 ($\text{C}(\text{CH}_3)_3$), 47.7 (CH_2 imidazoline), 35.9 ($(\text{CH}_3)_2\text{CH}$), 29.87 ($\text{C}(\text{CH}_3)_2$), 20.2 (CH_3 isopropyl), 15.8 (CH_3 isopropyl), 2.7($\text{N}(\text{Si}(\text{CH}_3)_3)_2$) ppm. IR data: (KBr disc, cm^{-1}) 2960 (s), 2871 (m), 1597 (s), 1528 (m), 1503 (w), 1464 (m), 1444 (w), 1396 (m), 1361 (m), 1312 (w), 1258 (s), 1229 (s), 1175 (w), 1101 (s), 1047 (s), 1018 (s), 929 (w), 861 (w), 802 (s), 748 (w), 699 (w), 670 (m), 640 (w), 615 (w), 527 (w), 473 (w).

General procedure for the preparation of the $[\text{Ca}(\text{R-BIM})_2]$ (4a-c): R-BIM (50mg, 0.31 mmol) in dry toluene (0.5ml) was added to $[\text{Ca}\{\text{N}(\text{SiMe}_3)_2\}_2(\text{py})_2]$ (82.0mg, 0.16 mmol). The solution was agitated briefly and allowed to react for 18 hours at ambient temperature under an inert atmosphere. The resulting complex was typically realised as a dark orange/magenta powder when dried *in vacuo* to approximately 4×10^{-2} mbar. Despite obtaining spectroscopically pure complexes, and repeated recrystallization attempts, satisfactory elemental analyses could not be obtained. Given the highly air and moisture sensitive nature of these compounds, and literature precedence, this is unsurprising.^{7,12}



$[\text{Ar}^{\text{Me}}\text{Ca}(\text{BIM})_2]$ 4a: ^1H NMR data (d_8 -THF, 500.1 MHz, 293 K): $\delta_{\text{H}} = 7.04$ -6.97 (8 H, m, *o*- $\text{C}_6\text{H}_4\text{CH}_3$), 6.96-6.89 (8 H, m, *m*- $\text{C}_6\text{H}_4\text{CH}_3$), 3.96-3.82 (4 H, m, NCH_2CH), 3.72 (1 H, s, CH bridge), 3.70 (1 H, s, CH bridge), 3.68 (2 H, app t, $^3J = 4.6$ Hz, NCH_2), 3.52 (2 H, app t, $^3J = 8.4$ Hz, NCH_2), 3.35 (2 H, m, NCH_2), 3.26-3.91 (2 H, m, NCH_2), 2.21 (6 H, s, $\text{C}_4\text{H}_5\text{CH}_3$), 2.20 (6 H, s, $\text{C}_4\text{H}_5\text{CH}_3$), 2.10-1.96 (2 H, m, $(\text{CH}_3)_2\text{CH}$), 1.94-1.79 (2 H, m, $(\text{CH}_3)_2\text{CH}$), 0.89-0.82 (12 H, m, $\text{CH}(\text{CH}_3)_2$), 0.81-0.72 (12 H, m, $\text{CH}(\text{CH}_3)_2$) ppm. $^{13}\text{C}\{^1\text{H}\}$ NMR data (d_8 -THF, 75.5 MHz, 293 K): $\delta_{\text{C}} = 164.4$ (C=N), 164.2 (C=N), 163.5 (C=N), 163.4 (C=N), 143.25 (*ipso*- $\text{C}_6\text{H}_4\text{CH}_3$), 143.18 (*ipso*- $\text{C}_6\text{H}_4\text{CH}_3$), 142.5 (*ipso*- $\text{C}_6\text{H}_4\text{CH}_3$), 142.3 (*ipso*- $\text{C}_6\text{H}_4\text{CH}_3$), 132.7 (p-



[Ca(^tBu-BIM)₂] **4c**: ¹H NMR data (d₈-THF, 500.1 MHz, 293 K): δ_H = 3.70 (2 H, s, CH_{bridge}), 3.50 (4 H, d t, ³J_{HH} = 9.1 Hz, ⁴J_{HH} = 2.7 Hz, NCH₂CH), 3.14 (4 H, dd, ³J_{HH} = 8.0 Hz, ⁴J_{HH} = 2.7 Hz, NCH₂), 3.05 (4 H, app t, ³J_{HH} = 9.1 Hz, NCH₂), 1.83-1.75 (4 H, m, (CH₃)₂CH), 1.37 (36 H, br s, C(CH₃)₃), 0.82 (12 H, d, ³J_{HH} = 6.9 Hz, CH(CH₃)₂), 0.73 (12 H, d, ³J_{HH} = 6.9 Hz, CH(CH₃)₂), ppm. ¹³C{¹H} NMR data (d₈-THF, 75.5 MHz, 293 K): δ_C = 166.6 (C=N), 166.4 (C=N), 65.3 (C=NCH_{imidazoline}), 64.3 (C=NCH_{imidazoline}), 62.0 (CH_{bridge}), 61.8 (CH_{bridge}), 52.9 (C(CH₃)₃), 52.7 (C(CH₃)₃), 47.9 (CH₂ imidazoline), 47.8 (CH₂ imidazoline), 35.5 ((CH₃)₂CH), 35.2 ((CH₃)₂CH), 30.0 (C(CH₃)₂), 29.82 (C(CH₃)₂), 20.9 (CH₃ isopropyl), 20.2 (CH₃ isopropyl), 16.2 (CH₃ isopropyl), 15.7 (CH₃ isopropyl) ppm. IR data: (KBr disc, cm⁻¹) 2955 (s), 2866 (s), 1601 (s), 1526 (s), 1467 (m), 1442 (w), 1393 (m), 1363 (m), 1313 (w), 1259 (s), 1229 (s), 1174 (w), 1100 (s), 1016 (s), 927 (w), 863 (m), 793 (s), 734 (w), 699 (w), 659 (w), 545 (w), 520 (w), 466 (w).

Equilibrium coefficients were determined by integration of the ¹H NMR spectra; ΔG values were obtained using the equation ΔG = -RT ln K.

General procedure for hydroamination catalysis: Both the 1-amino 2,2-dimethylpent-4-ene and 1-amino 2,2-diphenylpent-4-ene were prepared according to literature methods.^{8,9} In a dinitrogen filled glovebox R-BIM (0.04 mmol) was predissolved in C₆D₆ (0.7 ml) and successively added to [Ca{N(SiMe₃)₂}₂(THF)₂] (0.04 mmol). The resulting mixture was agitated by hand and left for two minutes. To this solution was added the corresponding amino olefin (0.4 mmol). The solution was transferred to a J. Young Teflon valve equipped NMR tube and sealed. All catalyst reactions were monitored *via* ¹H NMR periodically to monitor conversion (conversion was determined from notable resonances in the spectra corresponding to the pyrrolidine product).¹⁰ Upon conversion ceasing, a solution of (R)-(-)-*O*-acetylmandelic acid (0.41 mmol) predissolved in a minimal amount of CDCl₃ was added to the reaction mixture producing the diastereomeric salts. The resulting enantioexcess (e.e) was then determined from ¹H NMR spectroscopy.¹¹

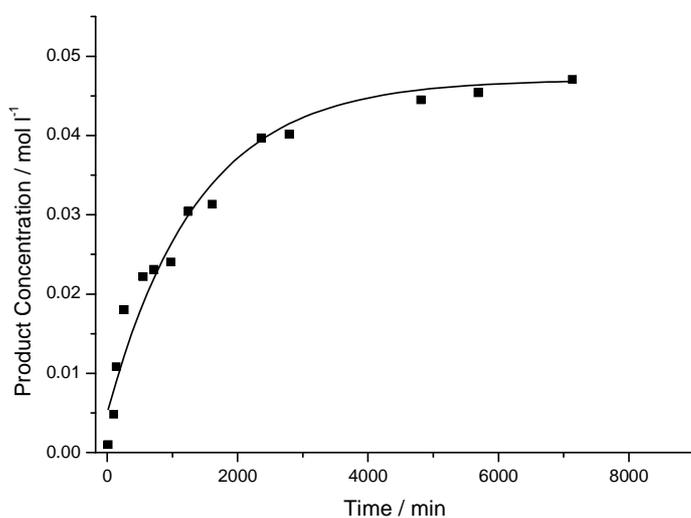
Determination of Rate Constants from Conversion Curves:

The conversion curves were curve fitted to an exponential decay function $y=Ae^{-bx}$. The initial rate, given by the slope at the origin ($x=0$), were determined by differentiation of the equation: $dy/dx = -A/b$.

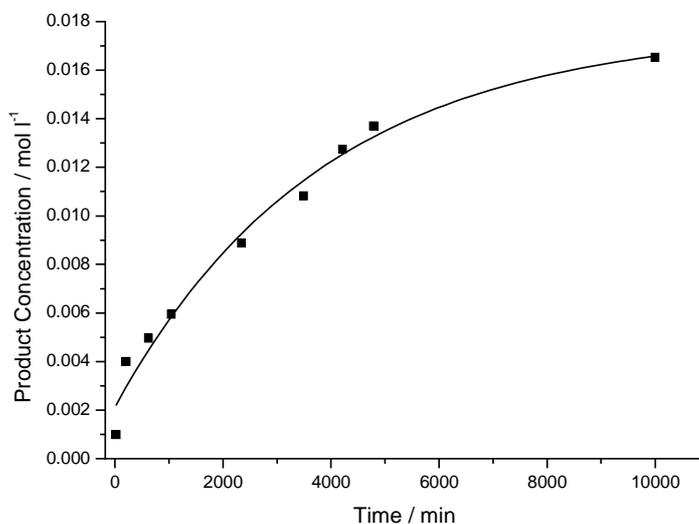
Product Conversion Curves



10 mol%, catalytic substrate A, Initial Rate = $5.1(5) \times 10^{-7} \text{ mol L}^{-1}\text{s}^{-1}$.

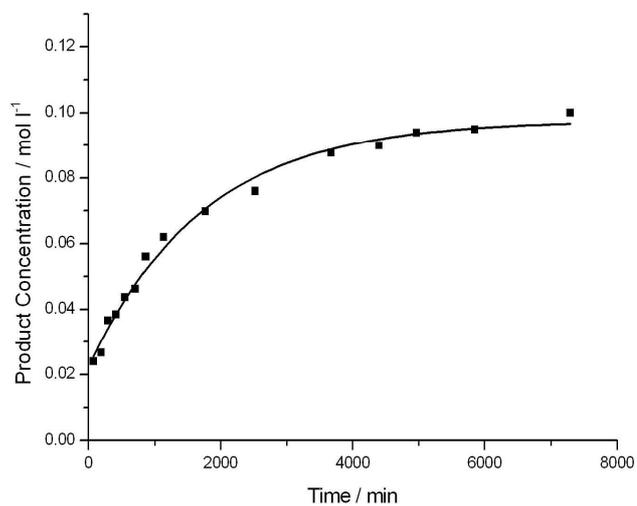


10 mol%, catalytic substrate A, Initial Rate = $6.8(3) \times 10^{-8} \text{ mol L}^{-1}\text{s}^{-1}$.



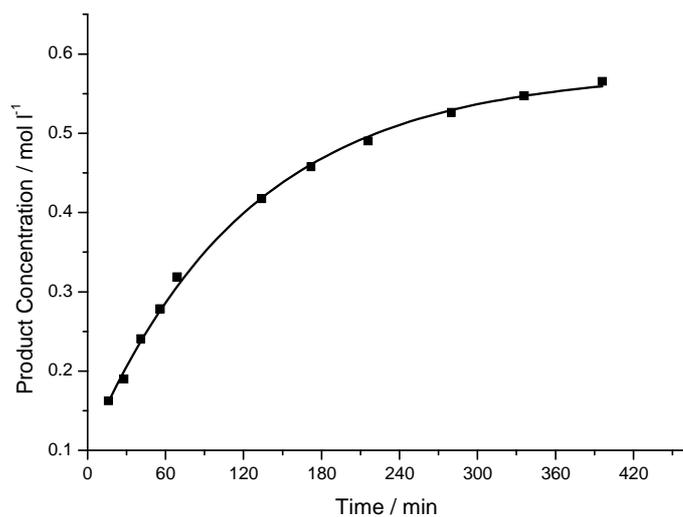
[Ca(^tBu-BIM){N(SiMe₃)₂}(THF)]

10 mol%, catalytic substrate A, Initial Rate = $7.3(4) \times 10^{-7} \text{ mol L}^{-1} \text{ s}^{-1}$.



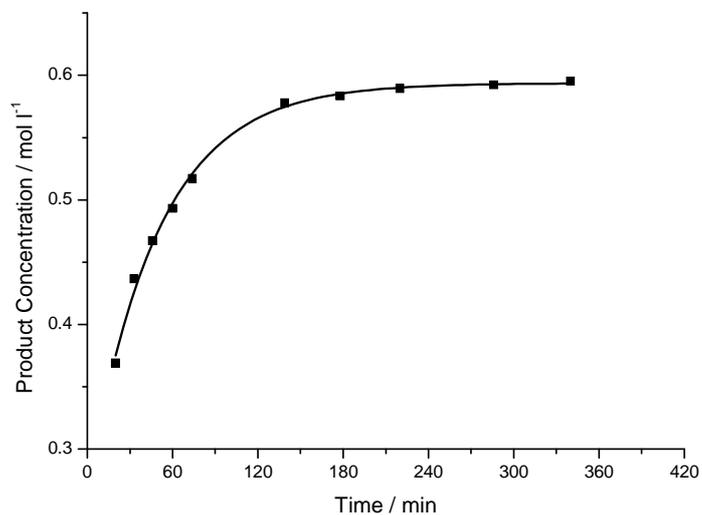
[Ca(^tBu-BIM){N(SiMe₃)₂}(THF)]

10 mol%, catalytic substrate B, Initial Rate = $6.4(3) \times 10^{-5} \text{ mol L}^{-1} \text{ s}^{-1}$.



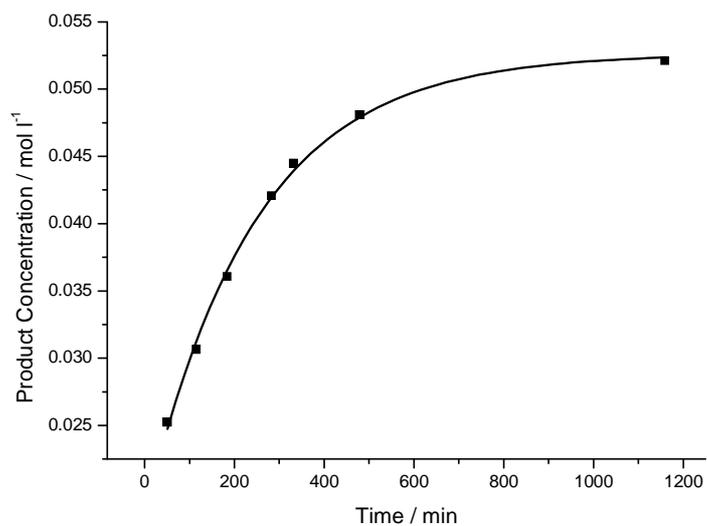
[Ca(^tBu-BIM){N(SiMe₃)₂}(THF)]

20 mol%, catalytic substrate B, Initial Rate = $1.1(1) \times 10^{-4} \text{ mol L}^{-1}\text{s}^{-1}$



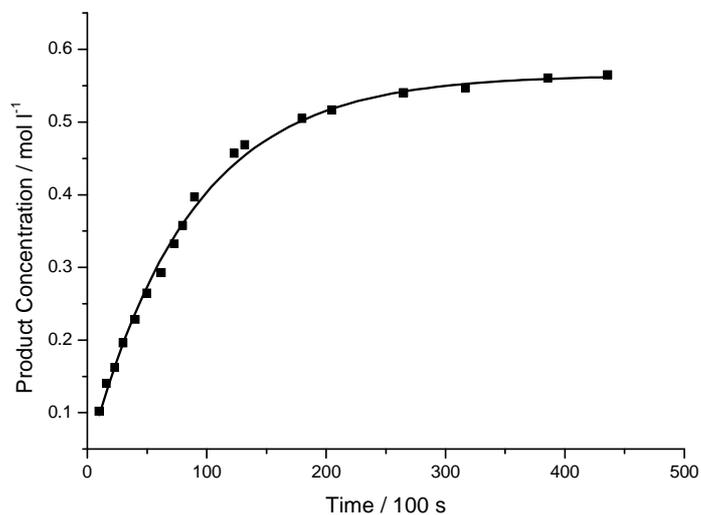
[Ca(Ar^{Me}-BIM){N(SiMe₃)₂}(THF)]

5 mol%, catalytic substrate B, Initial Rate = $2.5(1) \times 10^{-6} \text{ mol L}^{-1}\text{s}^{-1}$

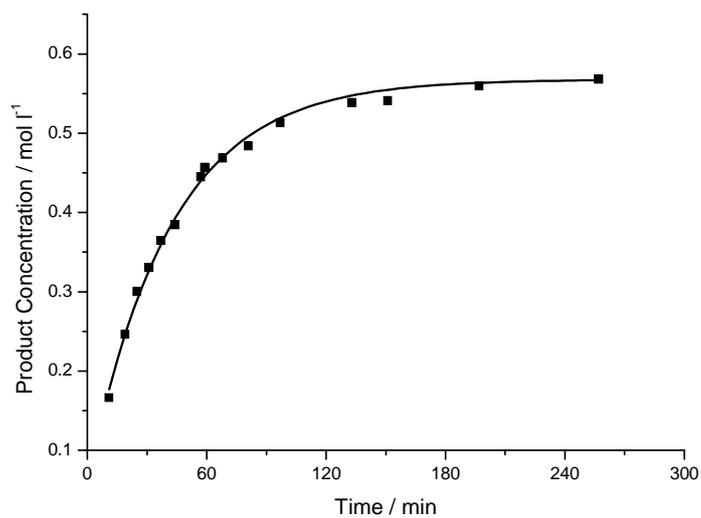




10 mol%, catalytic substrate B, Initial Rate = $1.0(1) \times 10^{-4} \text{ mol L}^{-1} \text{ s}^{-1}$.

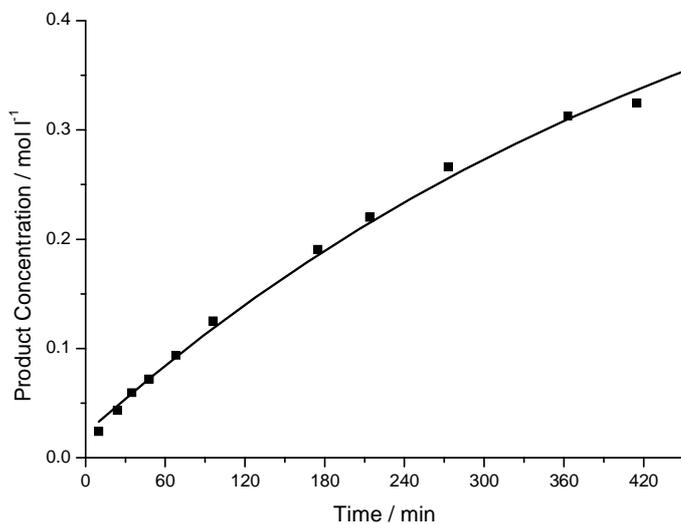


20 mol%, catalytic substrate B, Initial Rate = $2.1(2) \times 10^{-4} \text{ mol L}^{-1} \text{ s}^{-1}$.



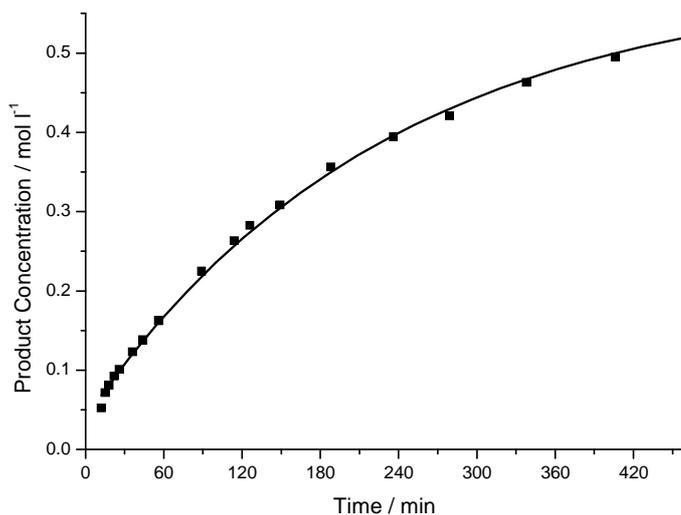
[Ca(Ar^F-BIM){N(SiMe₃)₂}(THF)]

10 mol%, catalytic substrate B, Initial Rate = $1.8(1) \times 10^{-5} \text{ mol L}^{-1}\text{s}^{-1}$



[Ca(Ar^F-BIM){N(SiMe₃)₂}(THF)]

20 mol%, catalytic substrate B, Initial Rate = $4.0(3) \times 10^{-5} \text{ mol L}^{-1}\text{s}^{-1}$



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