# Electronic Supplementary Information 

for

# Spin crossover in polymeric and heterometallic $\mathrm{Fe}^{\mathrm{II}}$ species containing polytopic dipyridylamino-substituted-triazine ligands 

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ppm (f1)
Figure S1 ${ }^{1} \mathrm{H}$ NMR of ddta (DMSO = dimethylsulfoxide).

ppm (f1)
Figure S2 ${ }^{13} \mathrm{C}$ NMR of ddta (DCM $=$ dichloromethane $)$.


Figure S3 ${ }^{1} \mathrm{H}$ NMR of tptd (THF = tetrahydrofuran).


1pm (f1)
Figure $\mathbf{S 4}{ }^{13} \mathrm{C}$ NMR of tptd.


Figure S5 Labelled asymmetric unit of 1a, with ellipsoids shown at 50\% probability. Hydrogen atoms have been omitted for clarity.


Figure S6 Labelled asymmetric unit of 1b, with ellipsoids shown at $50 \%$ probability. Hydrogen atoms have been omitted for clarity.


Figure S7 Labelled asymmetric unit of 2, with ellipsoids shown at 50\% probability. Hydrogen atoms have been omitted for clarity.


Figure S8 Labelled asymmetric unit of 3, with ellipsoids shown at 50\% probability. Hydrogen atoms have been omitted for clarity.

## Special Refinement Details for Single Crystal Data

1a: The hydroxyl group of one of the 1-propanol molecules within the structure of $\mathbf{1 a}$ is disordered over two positions, with $65 \%$ occupancy for O14A and $35 \%$ occupancy for O14B. O14A and O14B have been restrained to have the same thermal displacement parameters. The distances between C40 and O14A and C40 and O14B have been restrained to 1.430(1) A. Hydrogen atoms have not been assigned to O14A or O14B.
$\mathbf{1 b}$ : Within the structure of $\mathbf{1 b}$ the disorder in the 1-propanol molecule that has been described for 1a has increased. The hydroxyl group of the previously described 1-propanol molecule is disordered over two positions at $50 \%$ occupancy each. C38 and C39 of the same 1-propanol molecule are also found to be disordered over two positions, with $50 \%$ occupancy at each position for both disordered atoms. The distances between C40 and O14A, and C40 and O14B, have been restrained to be $1.430(1) \AA$. The distances between C40 and C39A, and C40 and C39B, have been restrained to be 1.50 (1) $\AA$. The distances between C39A and C38A, and C39B and C38B, have been restrained to be 1.50 (1) $\AA$. The distances between O14A and C39A, and O14B and C39B, have been restrained to be 2.50 (1) $\AA$. The distances between C40 and C38A, and C40 and C38B, have been restrained to be $2.50(1) \AA$. The atoms O14A, O14B, C40, C39A, C39B, C38A and C38B have been restrained to have the same thermal displacement parameters. The distance between C36 and C37 within the second unique 1-propanol molecule has been restrained to be $1.50(1) \AA$. Hydrogen atoms have not been assigned to O14A, O14B, C40, C39A, C39B, C38A or C38B. The atoms O13, C35, C36, C37, O14A, O14B, C40, C39A, C39B, C38A and C38B have been refined isotropically. ISOR restraints have been applied to C23, C24 and O4.

2: Within one of the two unique $\mathrm{NCS}^{-}$ligands in the structure of 2 the sulfur atom is disordered over two positions. Here, S2A is found to be at $70 \%$ occupancy, while S2B is found to be at $30 \%$ occupancy. The distances between C35 and S2A, and C35 and S2B, have been restrained to be $1.640(5) \AA$. The distance between O5 and C36, within the lattice methanol molecule, has been restrained to be $1.40(1) \AA$. ISOR restraints have also been applied to C36 and O5. Within the monoaza-15-crown-5 moiety of the tptd ligand disorder has been found in O 2 and C 31 , both of which are disordered over two positions. Here, O2A and O2B are each at $50 \%$ occupancy, while C31A is at $40 \%$ occupancy and C31B is at $60 \%$ occupancy. Within the monoaza-15-crown-5 ring the distances between C25 and C26, C27
and C28, C29 and C30, C31A and C32, C31B and C32, and C33 and C34 have been restrained to be the same. Further to this, the distances between C26 and O1, C27 and O1, C 28 and $\mathrm{O} 2 \mathrm{~A}, \mathrm{C} 28$ and $\mathrm{O} 2 \mathrm{~B}, \mathrm{C} 29$ and $\mathrm{O} 2 \mathrm{~A}, \mathrm{C} 29$ and $\mathrm{O} 2 \mathrm{~B}, \mathrm{C} 30$ and $\mathrm{O} 3, \mathrm{C} 31 \mathrm{~A}$ and O 3 , C 31 B and O 3 , and C 32 and O 4 have been restrained to be the same. ISOR restraints have been applied to C31A, C31B, C30, C29, O2A, O2B, C36, O5 and C28. Hydrogen atoms have not been assigned to C28, C29, C31A, C31B or C32.

3: Within the structure of $\mathbf{3}$ disorder is found in the mono-aza-15-crown-5 moiety of the tptd ligand. Here, C32, O4 and C33 are each disordered over two positions (labelled C32A and C32B, O4A and O4B, C33A and C33B, respectively), where each position was found to be at $50 \%$ occupancy for each atom. Atoms C31, C32A, C32B, O4A, O4B, C33A and C33B have been refined isotropically. Within this disordered part of the monoaza-15-crown-5 moiety of the tptd ligand the distances between O3 and C32A, O3 and C32B, C31 and O4A, C31 and O4B, C32A and C33A, C32B and C33B, O4A and C34, and O4B and C34 have been restrained to be 2.40 (1) $\AA$. Within the monoaza-15-crown- 5 moiety of the tptd ligand the distances between C25 and C26, C27 and C28, C29 and C30, C31 and C32A, C31 and C32B, C33A and C34, and C33B and C34 have been restrained to be 1.50 (1) A. The distances between C 26 and $\mathrm{O} 1, \mathrm{C} 27$ and $\mathrm{O} 1, \mathrm{C} 28$ and $\mathrm{O} 2, \mathrm{C} 29$ and $\mathrm{O} 2, \mathrm{C} 30$ and $\mathrm{O} 3, \mathrm{C} 31$ and O 3 , C 32 A and $\mathrm{O} 4 \mathrm{~A}, \mathrm{C} 32 \mathrm{~B}$ and $\mathrm{O} 4 \mathrm{~B}, \mathrm{C} 33 \mathrm{~A}$ and O 4 A , and C 33 B and O 4 B have been restrained to be the same. Hydrogen atoms have not been assigned to C31, C32A, C32B, C33A, C33B or C34. The atoms C32A and C32B have been restrained to have the same thermal displacement parameters.

Table S1 Unit cell and refinement details for variable temperature powder X-Ray diffraction data collected on $\mathbf{1}$
(trans-[ $\left.\left.\mathrm{Fe}^{\mathrm{II}}(\mathrm{NCS})_{2}(\mathrm{ddta})_{2} \mathrm{Na}_{2}\right]\left(\mathrm{ClO}_{4}\right)_{2} \cdot 4 \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$

| $\mathbf{T} / \mathbf{K}$ | $\boldsymbol{a}(\mathbf{\AA})$ | $\boldsymbol{b}(\mathbf{\AA})$ | $\boldsymbol{c}(\AA \mathbf{\AA})$ | $\boldsymbol{a}\left({ }^{\boldsymbol{}}\right)$ | $\boldsymbol{\beta}\left({ }^{\boldsymbol{}}\right)$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | $\boldsymbol{\gamma}\left({ }^{\boldsymbol{}}\right)$ | Volume <br> $\left(\AA^{3}\right)$ | $\boldsymbol{\chi}^{\mathbf{2}}$ |  |  |
| 150 | $12.2216(1)$ | $13.2399(2)$ | $16.5525(2)$ | $78.3172(7)$ | $89.1164(7)$ | $64.8491(6)$ | $2366.7(8)$ | 4.179 |
| 160 | $12.2257(1)$ | $13.2468(1)$ | $16.5601(2)$ | $78.3528(7)$ | $89.1267(6)$ | $64.8342(6)$ | $2369.8(8)$ | 3.918 |
| 170 | $12.2306(1)$ | $13.2542(1)$ | $16.5689(2)$ | $78.3872(7)$ | $89.1374(6)$ | $64.8189(6)$ | $2373.4(8)$ | 3.958 |
| 180 | $12.2353(1)$ | $13.2636(1)$ | $16.5770(2)$ | $78.4312(6)$ | $89.1388(6)$ | $64.8035(6)$ | $2377.2(8)$ | 3.917 |
| 190 | $12.2418(1)$ | $13.2718(1)$ | $16.5876(1)$ | $78.4830(6)$ | $89.1482(6)$ | $64.7915(6)$ | $2381.8(8)$ | 3.288 |
| 200 | $12.2481(1)$ | $13.2802(1)$ | $16.5986(1)$ | $78.5408(6)$ | $89.1534(5)$ | $64.7789(6)$ | $2386.3(7)$ | 3.220 |
| 205 | $12.2534(1)$ | $13.2857(1)$ | $16.6059(1)$ | $78.5741(6)$ | $89.1532(6)$ | $64.7657(6)$ | $2389.5(7)$ | 3.063 |
| 210 | $12.2570(1)$ | $13.2893(1)$ | $16.6130(1)$ | $78.6097(6)$ | $89.1509(5)$ | $64.7665(5)$ | $2392.3(7)$ | 2.950 |
| 215 | $12.2604(1)$ | $13.2935(1)$ | $16.6208(1)$ | $78.6507(6)$ | $89.1490(5)$ | $64.7699(5)$ | $2395.3(7)$ | 2.920 |
| 220 | $12.2651(1)$ | $13.2976(1)$ | $16.6291(1)$ | $78.6939(6)$ | $89.1442(5)$ | $64.7673(5)$ | $2398.5(7)$ | 2.819 |
| 225 | $12.2687(1)$ | $13.3038(1)$ | $16.6407(1)$ | $78.7410(6)$ | $89.1568(5)$ | $64.7688(5)$ | $2402.5(7)$ | 2.648 |
| 230 | $12.2725(1)$ | $13.3078(2)$ | $16.6500(1)$ | $78.7963(6)$ | $89.1529(6)$ | $64.7741(6)$ | $2406.0(8)$ | 3.582 |
| 235 | $12.2787(1)$ | $13.3125(1)$ | $16.6639(1)$ | $78.8510(6)$ | $89.1470(5)$ | $64.7777(5)$ | $2410.7(7)$ | 2.637 |
| 240 | $12.2825(1)$ | $13.3166(1)$ | $16.6775(2)$ | $78.9112(6)$ | $89.1454(6)$ | $64.7808(6)$ | $2414.8(7)$ | 2.608 |
| 245 | $12.2885(1)$ | $13.3229(1)$ | $16.6922(1)$ | $78.9761(6)$ | $89.1409(6)$ | $64.7847(6)$ | $2420.0(7)$ | 2.588 |
| 250 | $12.2930(1)$ | $13.3278(2)$ | $16.7058(2)$ | $79.0506(7)$ | $89.1339(6)$ | $64.7849(6)$ | $2424.5(8)$ | 3.291 |
| 255 | $12.2986(1)$ | $13.3332(2)$ | $16.7226(2)$ | $79.1289(7)$ | $89.1377(7)$ | $64.7947(7)$ | $2430.0(9)$ | 3.489 |
| 260 | $12.3033(1)$ | $13.3390(2)$ | $16.7382(2)$ | $79.2002(7)$ | $89.1353(7)$ | $64.7976(7)$ | $2435.0(9)$ | 3.409 |
| 265 | $12.3086(1)$ | $13.3453(2)$ | $16.7556(2)$ | $79.2755(7)$ | $89.1368(6)$ | $64.7951(6)$ | $2440.4(8)$ | 3.161 |
| 270 | $12.3126(1)$ | $13.3500(2)$ | $16.7711(2)$ | $79.3510(7)$ | $89.1435(6)$ | $64.8017(6)$ | $2445.1(8)$ | 3.102 |
| 275 | $12.3177(1)$ | $13.3559(2)$ | $16.7862(2)$ | $79.4318(7)$ | $89.1488(7)$ | $64.7907(7)$ | $2450(1)$ | 3.617 |
| 280 | $12.3236(1)$ | $13.3604(2)$ | $16.8012(2)$ | $79.5118(6)$ | $89.1617(7)$ | $64.7896(7)$ | $2454.8(9)$ | 2.812 |
| 290 | $12.3354(2)$ | $13.3740(2)$ | $16.8312(2)$ | $79.6567(7)$ | $89.1746(7)$ | $64.7813(7)$ | $2465.1(9)$ | 3.055 |
| 300 | $12.3412(2)$ | $13.3851(2)$ | $16.8530(2)$ | $79.8110(8)$ | $89.2216(9)$ | $64.7643(8)$ | $2473(1)$ | 3.369 |



Figure S9 Le Bail fit to synchrotron powder X-Ray data on $\mathbf{1}$ (trans-[ $\left.\left[\mathrm{Fe}^{\mathrm{II}}(\mathrm{NCS})_{2}(\mathrm{ddta})_{2} \mathrm{Na}_{2}\right]\left(\mathrm{ClO}_{4}\right)_{2} \bullet 4 \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$ at 150 K . Observed data are shown as black crosses, the calculated data are shown as a red line, the difference between observed and calculated data is shown as a green line, and the predicted, possible location of peaks based on the symmetry of the lattice are shown as vertical blue lines.


Figure S10 Le Bail fit to synchrotron powder X-Ray data on $\mathbf{1}\left(\right.$ trans- $\left.\left[\mathrm{Fe}^{\mathrm{II}}(\mathrm{NCS})_{2}(\mathrm{ddta})_{2} \mathrm{Na}_{2}\right]\left(\mathrm{ClO}_{4}\right)_{2} \bullet 4 \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OH}\right)$ at 300 K . Observed data are shown as black crosses, the calculated data are shown as a red line, the difference between observed and calculated data is shown as a green line, and the predicted, possible location of peaks based on the symmetry of the lattice are shown as vertical blue lines.








Figure S11 (a) Temperature dependent behaviour of the unit cell volume for $\mathbf{1}$; (b) temperature dependent behaviour of the unit cell volume for $\mathbf{1}$ given as a proportion of its highest magnitude; (c) temperature dependent behaviour of $a$ for $\mathbf{1}$; (d) temperature dependent behaviour of $b$ for $\mathbf{1}$; (e) temperature dependent behaviour of $c$ for $\mathbf{1 ; ~ ( f ) ~ t e m p e r a t u r e ~ d e p e n d e n t ~ b e h a v i o u r ~ o f ~ t h e ~ a n g l e ~} \alpha$ for $\mathbf{1 ; ( g )}$ temperature dependent behaviour of the angle $\beta$ for $\mathbf{1 ;}$ (h) temperature dependent behaviour of the angle $\gamma$ for $\mathbf{1}$. In (a) the vertical axis has units of $\AA^{3}$, in (b) the vertical axis is a proportion of the highest value of the unit cell volume, in (c) - (d) the vertical axis has units of $\AA$, and in (f) - (h) the vertical axis is given in ${ }^{\circ}$. In all Figures the horizontal axis is temperature (K). For Figures (a) and (c) - (h) error bars to three times the calculated error are given, but in most cases are eclipsed by the position markers and cannot be seen here; no error bars are given for (b).

