

Electronic Supplementary Information

Novel heterometal-organic complexes as first single source precursors for up- converting NaY(Ln)F₄ (Ln = Yb, Er, Tm) nanomaterials

Shashank Mishra,^{*a,b} Gilles Ledoux,^b Erwann Jeanneau,^c Stéphane Daniele^{*a}

and Marie-France Joubert^{*b}

^aUniversity of Lyon 1, IRCELYON, 2 Avenue A. Einstein, 69626 Villeurbanne, France.

Fax: 33 472445399; Tel: 33 472445329; E-mail: mishrashashank74@rediffmail.com

^bUniversity of Lyon 1, Laboratoire de Physico Chimie des Matériaux Luminescent, 10 rue A.

M. Ampère, 69622 Villeurbanne, France.

^cUniversity of Lyon 1, Centre de Diffractométrie, 69622 Villeurbanne, France.

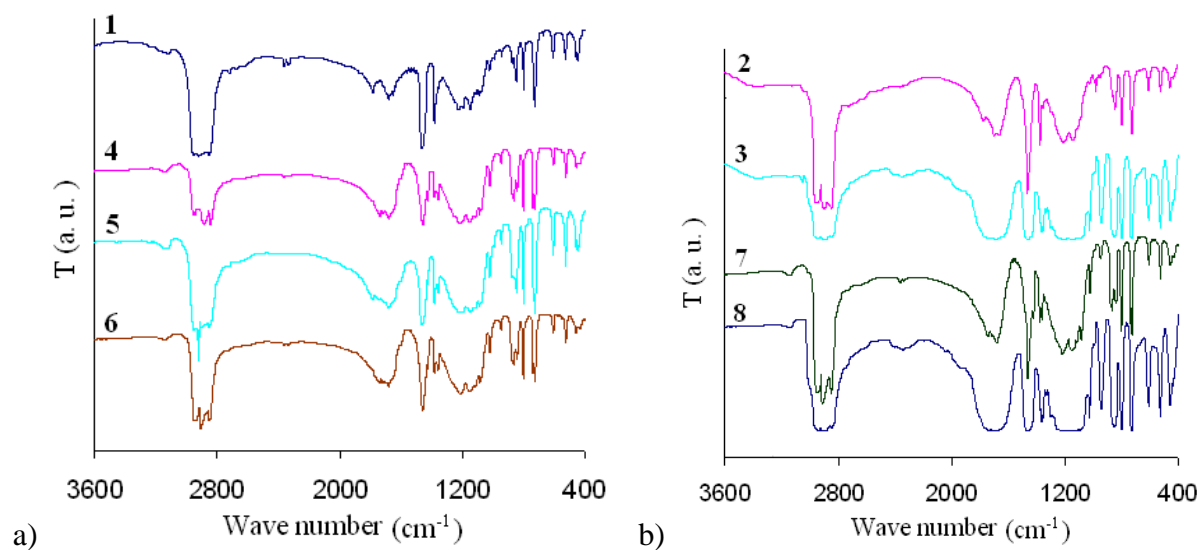


Fig. S1 FT-IR spectra of (a) diglyme complexes **1** & **4-6**, and (b) triglyme (**2**) and tetraglyme complexes (**3, 7, 8**).

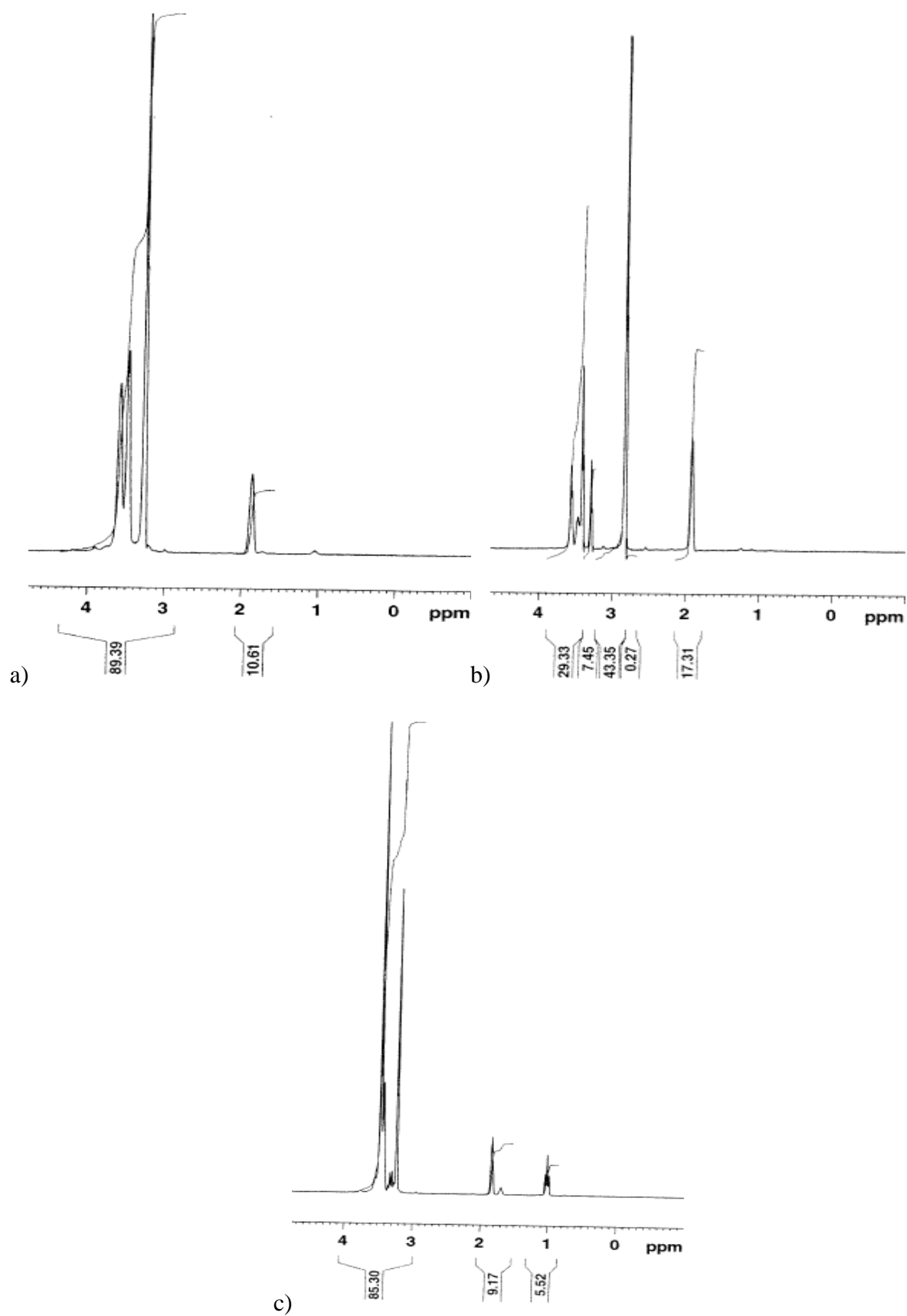


Fig. S2 ¹H NMR spectra of **1** (a), **2** (b) and **3** (c).

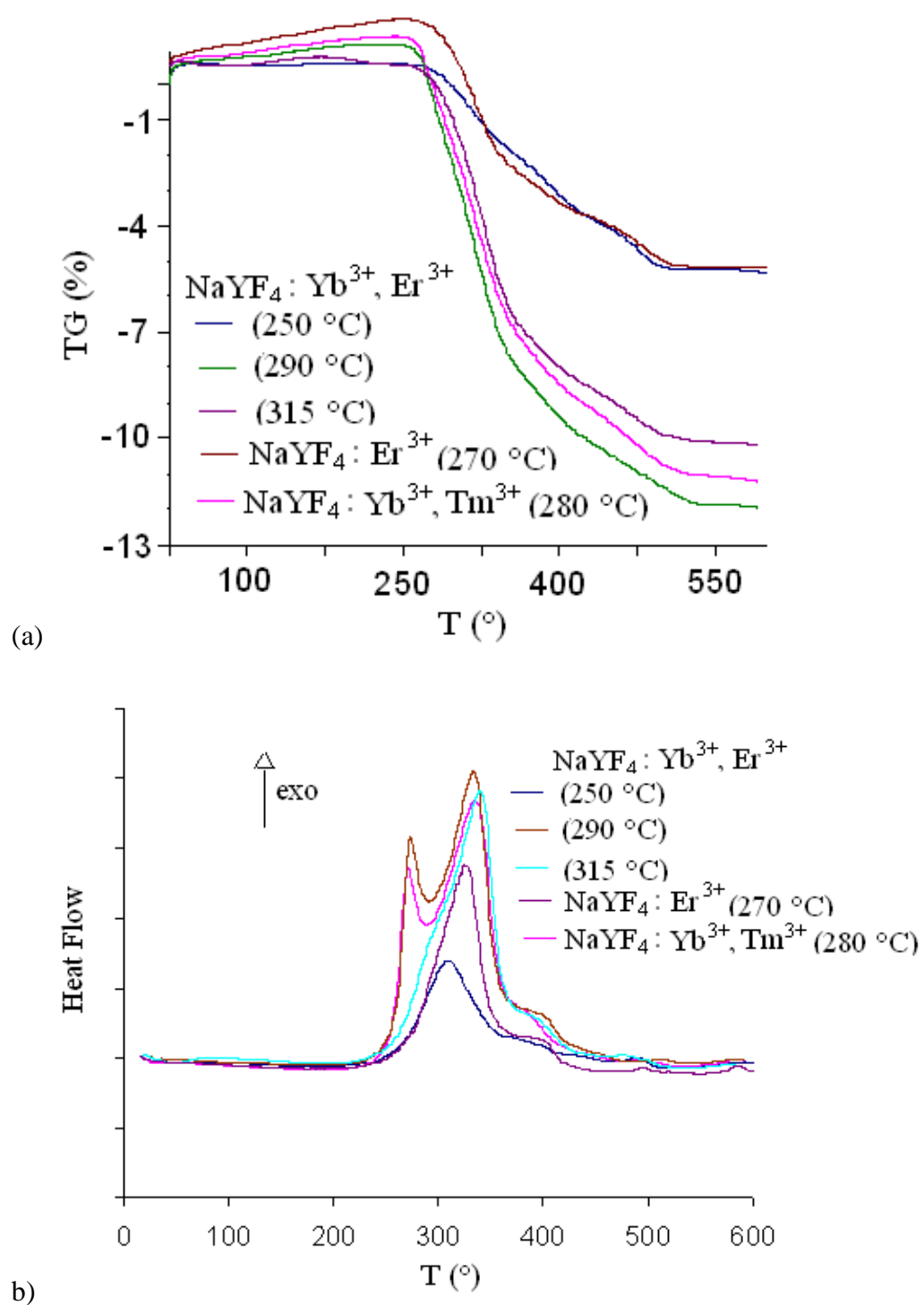


Fig. S3 TGA (a) and DTA (b) of NaYF₄: Yb³⁺, Er³⁺/Tm³⁺ NCs obtained at different temperature.

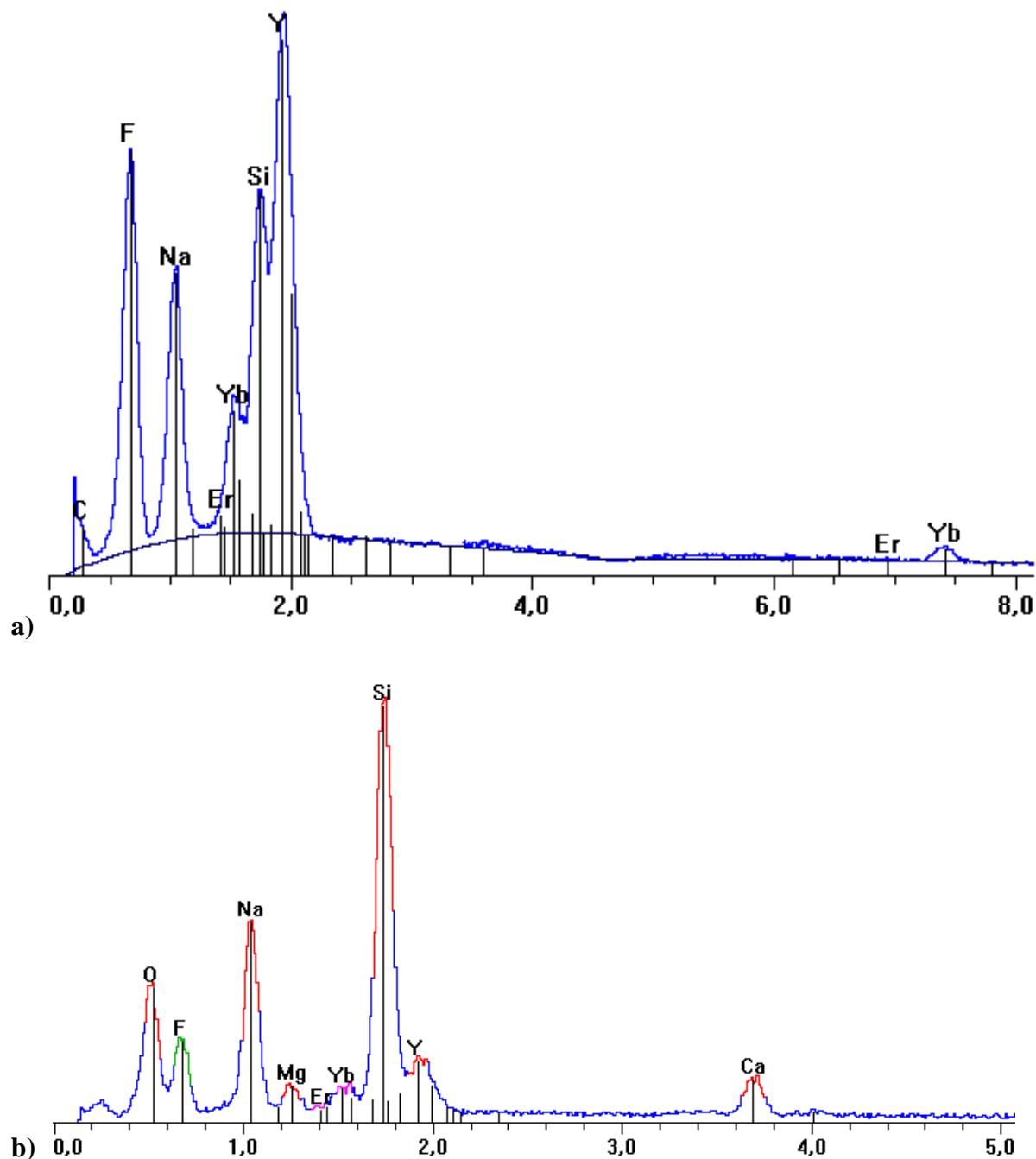


Fig. S4 EDX analysis of the thin films obtained by spin-coating of the diglyme precursors **1**, **4** and **6**. Above: NaYF₄: Yb³⁺, Er³⁺ films on Si wafer & calcined at 400 °C under nitrogen. Below: thin films consisting of the NaF and Y(Ln)OF phases on glass substrate (calcined at 400 °C in air).

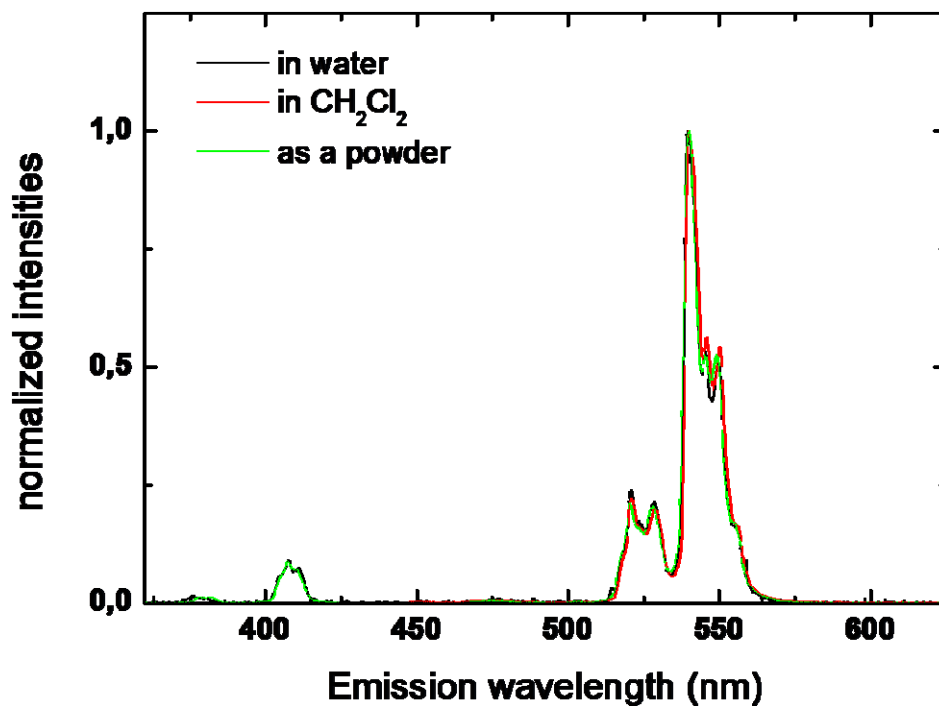


Fig. S5 Emission spectra of NaYF₄: Yb³⁺, Er³⁺ NCs when taken in solid or solution states.

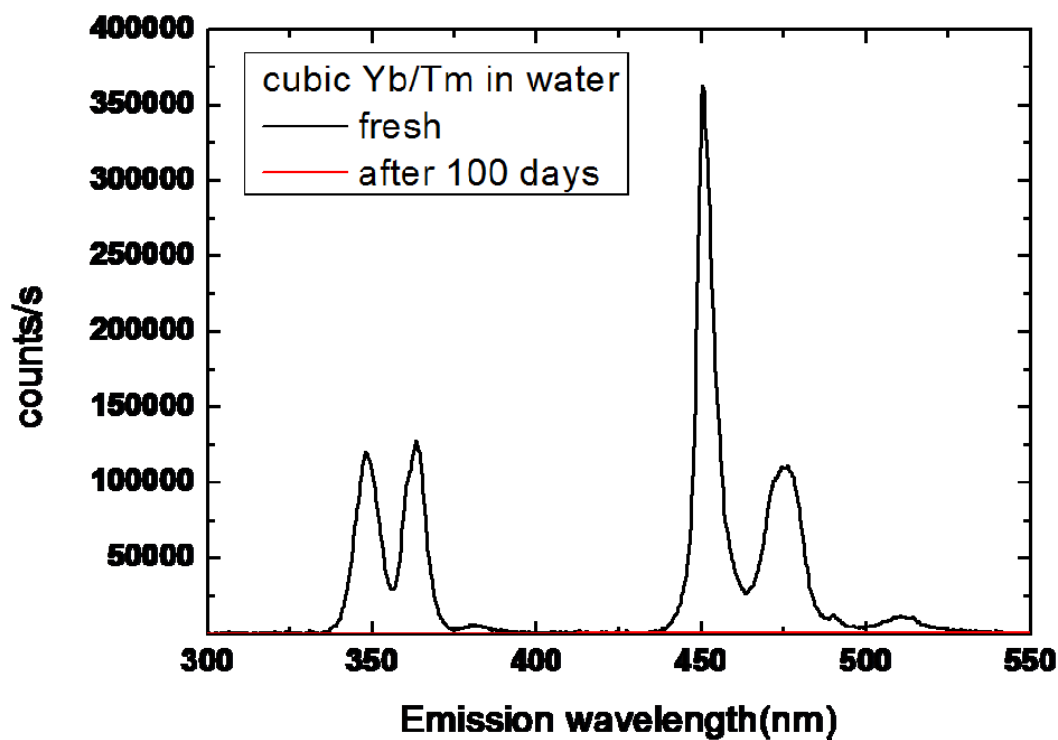


Fig. S6 Effect of aging of NaYF₄: 20% Yb³⁺, 2% Tm³⁺ NCs.

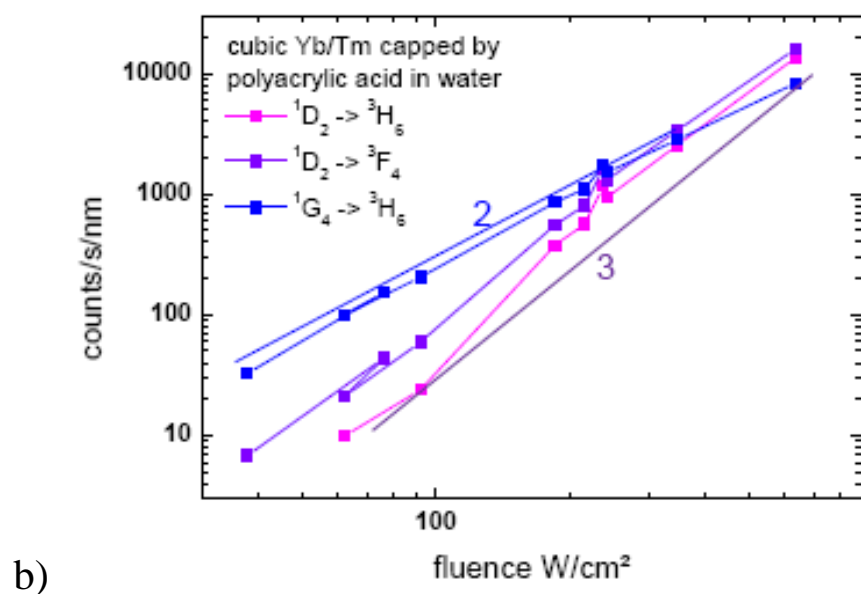
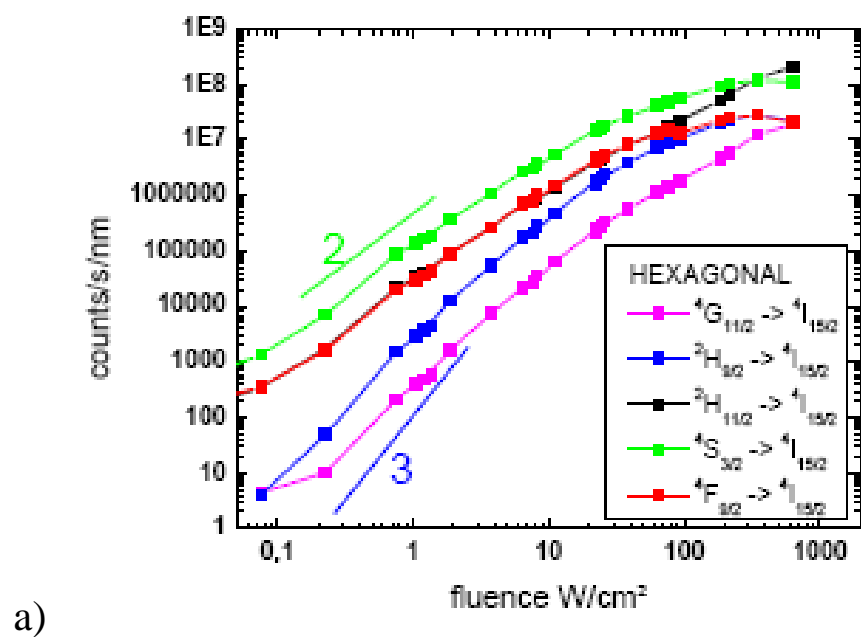


Fig. S7: Evolution of the intensity of the difference bands with the excitation power: (a) hexagonal phase of NaYF₄: Yb³⁺, Ln³⁺ NCs taken as a solid powder, and b) the cubic phase of NaYF₄: Yb³⁺, Tm³⁺ NCs taken in water.