Supporting Information

New Terphenyl based Ensemble for the Detection of Acetate Ions in Plasma like System

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General Experimental Procedures

UV-Vis and fluorescence Titrations

UV-Vis and fluorescence titrations were performed with $1x10^{-5}$ M solution of receptor **3** in THF. Typically, aliquots of freshly prepared standard solutions $(10^{-1} - 10^{-3} \text{ M in THF})$ of metal perchlorates $(Pb^{2+}, Hg^{2+}, Ba^{2+}, Cd^{2+}, Ag^+, Zn^{2+}, Cu^{2+}, Ni^{2+}, Co^{2+}, Fe^{3+}, Fe^{2+}, K^+, Mg^{2+}, Na^+ and Li^+)$ were added and the UV-Vis spectra of the samples were recorded. Similarly, aliquots of freshly prepared solutions of anions $(F^{-}, CI^{-}, Br^{-}, \Gamma, HSO_{4}^{-}, H_{2}PO_{4}^{-}, CH_{3}COO^{-}, NO_{3}^{-} N_{3}^{-}, SO_{4}^{2-}, SO_{3}^{2-}, Cr_{2}O_{7}^{2-})$ were added to record the absorption behaviour of **3**.

¹H NMR Experiments

The stock solution of compound **3** (10^{-2} M) and perchlorate salts of Hg²⁺ (2 x 10^{-2} M) were prepared in CDCl₃:CD₃CN (8:2) for ¹H NMR titration experiments. Tetrabutylammonium salts of fluoride and acetate were prepared in CDCl₃.

Experimental Procedure for synthesis of 4¹

A solution of dansyl chloride (0.27 g, 1.02 mmol) in dichloromethane (5 ml) was added to the solution of diamine **1** (0.05 g, 0.22 mmol) in minimum amount of dichloromethane. To the above solution triethylamine (90 μ L) was added. The resultant reaction mixture was stirred for overnight at room temperature. The mixture so obtained was treated with dichloromethane and dried over anhydrous Na₂SO₄. The organic layer was evaporated under reduced pressure and the crude product was purified by column chromatography (CH₂Cl₂) to give 70 mg of **4** (yield 70%).



¹ Kavallieratos, K.; Rosenberg, J. M., Chen, W.-Z., Ren, T. J.Am. Chem. Soc. 2005, 127, 6514.

¹H NMR of Compound 3



¹³C NMR of Compound 3



S4

Mass spectrum of 3









¹H NMR of Compound 3 in CDCl₃





¹H NMR of Compound 3 in the presence of CH₃COO⁻ ions in CDCl₃



¹H NMR of ensemble 3 + Hg in CDCl₃:CD₃CN (8:2)





¹H NMR of Ensemble 3-Hg in the presence of CH₃COO⁻ ions in CDCl₃

¹H NMR of Ensemble 3-Hg in the presence of F⁻ ions in CDCl₃







Fig. 1. UV-Vis spectra of receptor 3 (5 μ M) upon addition of CH₃COO⁻ions (0.1 – 10.0 equiv) in THF



Figure 2. UV-Vis spectra of receptor 3 (5 μ M) upon addition of F⁻ ions (0.1 – 10.0 equiv) in THF





Figure 2. UV-Vis spectra of receptor 3 (5 μ M) upon addition of OH⁻ ions (0.1 – 200.0 equiv) in THF

Compiled UV Titration data of receptor 3 with Hg²⁺



Figure 3. UV-Vis spectra of receptor 3 (5 μ M) upon addition of Hg²⁺ ions (0.1 – 20.0 equiv) in THF





Figure 4. UV-Vis spectra of receptor 1 (5 μ M) upon addition of of OAc⁻ ions (0.1 – 30.0 equiv) in the presence of PLAS- BSA (10 μ l)



Figure 5. UV-Vis spectra of receptor **1** (5 μ M) upon addition of of OAc⁻ ions (0.1 – 30.0 equiv) in the absence of PLAS- BSA (10 μ l)



Figure 6. UV-Vis spectra of receptor 1 (5 μ M) upon addition of F⁻ions (0.1 – 30.0 equiv) in THF

Compiled fluorescence behavior of receptor 3 in the presence of various metal ions





Compiled fluorescence behavior of ensemble 3-Hg in the presence of various anions

Fluorescence behavior of ensemble 3-Hg in the presence of CH₃COOK



Figure 7. Fluorescence spectra of 3+Hg (10 μ M) upon addition of aqueous solution of CH₃COOK (5 equiv)

Fluorescence behavior of reference compound 4 in the presence of Hg



Figure 8. Fluorescence spectra of 4 (2 μ M) upon addition of Hg²⁺ ions in THF (50 equiv)