### **Supporting Information**

### Onset of Three-Centre, Four-Electron Bonding in peri-Substituted

### Acenaphthenes:

### A Structural and Computational Investigation

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#### **1. Experimental Section**

All experiments were carried out under an oxygen- and moisture-free nitrogen atmosphere using standard Schlenk <sup>10</sup> techniques and glassware. Reagents were obtained from commercial sources and used as received. Dry solvents were collected from a MBraun solvent system. Elemental analyses were performed by the University of St. Andrews School of Chemistry Microanalysis Service. Infra-red spectra were recorded as KBr discs in the range 4000-300 cm<sup>-1</sup> on a Perkin-Elmer System 2000 Fourier transform spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Jeol GSX 270 MHz spectrometer with δ(H) and δ(C) referenced to external tetramethylsilane. <sup>77</sup>Se and <sup>125</sup>Te NMR spectra were recorded on a <sup>15</sup> Jeol GSX 270 MHz spectrometer with δ(Se) and δ(Te) referenced to external dimethylselenide and diphenyl ditelluride respectively. Assignments of <sup>13</sup>C and <sup>1</sup>H NMR spectra were made with the help of H-H COSY and HSQC experiments. All measurements were performed at 25 °C. All values reported for NMR spectroscopy are in parts per million (ppm). Coupling constants (*J*) are given in Hertz (Hz). Mass spectrometry was performed by the University of St. Andrews Mass Spectrometry Service. Electron impact mass spectrometry (EIMS) and Chemical Ionisation Mass Spectrometry (CIMS) <sup>20</sup> was carried out on a Micromass GCT orthogonal acceleration time of flight mass spectrometer. Electrospray Mass Spectrometry (ESMS) was carried out on a Micromass LCT orthogonal accelerator time of flight mass spectrometer. 5,6dibromoacenaphthene **A13**<sup>1,2,3</sup> and 5,6-diiodoacenaphthene **A14** were prepared following standard literature procedures

- starting from acenaphthene and 5,6-dibromoacenaphthene respectively.
- <sup>25</sup> 5-Bromo-6-(phenylsulfanyl)acenaphthene [Acenap(Br)(SPh)] (A1): *Synthesis of* A1 *from* A13: A solution of *n*-butyllithium (2.5 M) in hexane (3.85 mL, 9.62 mmol) was added dropwise to a solution of 5,6-dibromoacenaphthene (3.002 g, 9.62 mmol) in diethyl ether (40 mL) at -78°C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (2.100 g, 9.62 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78°C for a further 1 h, then washed with 0.1 N sodium hydroxide (3 x 60 mL). The organic layer was dried <sup>30</sup> with magnesium sulfate and concentrated under reduced pressure affording a yellow solid. Addition of hexane afforded the product as a white precipitate which was collected by filtration and recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (2.8 g, 87 %); mp 108-110 °C; elemental analysis (Found C, 63.3, H 3.7. Calc. for C<sub>18</sub>H<sub>13</sub>BrS: C 63.35; H, 3.8 %); IR (KBr disk): *v*<sub>max</sub> cm<sup>-1</sup> 2918s, 2825w, 1597vs, 1579vs, 1562s, 1486s, 1475vs, 1436vs, 1417s, 1406s, 1321vs, 1256s, 1229s, 1204w, 1197w, 1107s, 1081vs, 1069s, 1024vs, 996s, <sup>35</sup> 944w, 864s, 839vs, 813vs, 733vs, 687vs, 625s, 603s, 507s, 492s, 466s, 459s, 407w; δ<sub>H</sub>(270 MHz; CDCl<sub>3</sub>, 25 °C; Me<sub>4</sub>Si)
- 7.68 (1 H, d,  ${}^{3}J_{HH}$  7.4 Hz, Acenap 4-H), 7.52 (1 H, d,  ${}^{3}J_{HH}$  7.3 Hz, Acenap 7-H), 7.20-7.11 (3 H, m, Acenap 8-H, SPh 12,16-H), 7.11-7.02 (4 H, m, Acenap 3-H, SPh 13,15-H), 3.33-3.22 (4 H, m, 2 x CH<sub>2</sub>);  $\delta_{C}$ (67.9 MHz; CDCl<sub>3</sub>; 25 °C;

Me<sub>4</sub>Si) 148.2(q), 147.0(q), 141.9(q), 139.2(q), 137.8(s), 135.5(s), 130.1(q), 129.1(s), 126.7(q), 125.9(s), 120.90s), 120.6(s), 114.5(q), 30.2(s, CH<sub>2</sub>), 30.0(s, CH<sub>2</sub>); MS (EI<sup>+</sup>): *m/z* 341.95 (M<sup>+</sup>, 25 %), 260.15 (100, M - Br).

- 5-Bromo-6-(phenylselanyl)acenaphthene [Acenap(Br)(SePh) (A2): Synthesis of A2 from A13: A solution of ns butyllithium (2.5 M) in hexane (2.7 mL, 6.76 mmol) was added dropwise to a solution of 5,6-dibromoacenaphthene (1.054 g, 3.378 mmol) in diethyl ether (40 mL) at -78°C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (2.109 g, 6.76 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78°C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure affording an orange oil. Addition of hexane afforded the <sup>10</sup> product as a white precipitate which was collected by filtration and recrystallised from dichloromethane to give colourless crystals (0.9 g, 66 %); mp 77-78 °C; elemental analysis (Found C, 56.05; H, 3.2. Calc. for C<sub>18</sub>H<sub>13</sub>BrSe: C, 55.7; H, 3.4 %); IR (KBr disk): v<sub>max</sub> cm<sup>-1</sup> 3039w, 2915s, 2822w, 2341w, 1953w, 1886w, 1829w, 1759w, 1651w, 1597s, 1556s, 1470w, 1434s, 1408s, 1344w, 1328s, 1300s, 1248s, 1230s, 1196w, 1173w, 1150w, 1106s, 1060w, 1044w, 1015s, 995s, 969w, 912w, 832vs, 809vs, 740vs, 698vs, 688vs, 600s, 538w, 504w, 473s, 461s, 396w; δ<sub>H</sub>(270 MHz, CDCl<sub>3</sub>, 25 °C, Me<sub>4</sub>Si) 7.59 <sup>15</sup> (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 4-H), 7.54-7.45 (2 H, m, SePh 12-16-H), 7.30-7.19 (3H, m, SePh 13-15-H), 7.10 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 7-H), 6.95 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 3-H), 6.87 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 8-H), 3.15 (4 H, s, 2 x CH<sub>2</sub>);  $\delta_{C}(67.9 \text{ MHz}; \text{CDCl}_{3}; 25 \,^{\circ}\text{C}; \text{Me}_{4}\text{Si})$  146.8(q), 145.7(q), 142.0(q), 135.7(s), 134.2(s), 133.0(s), 132.0(q), 130.1(q), 1 129.7(s), 128.3(s), 126.5(q), 120.8(s), 120.6(s), 115.4(q), 30.1(s,  $CH_2$ ), 29.9(s,  $CH_2$ );  $\delta_{se}(51.5 \text{ MHz}; CDCl_3; 25 °C;$ PhSeSePh) 423.7(s); MS (EI<sup>+</sup>): *m/z* 387.94 (M<sup>+</sup>, 100 %).
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5-Bromo-6-(phenyltelluro)acenaphthene [Acenap(Br)(TePh)] (A3): Synthesis of A3 from A13: A solution of nbutyllithium (2.5 M) in hexane (2.9 mL, 9.7 mmol) was added dropwise to a solution of 5,6-dibromoacenaphthene (2.026 g, 9.7 mmol) in diethyl ether (40 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (3.971 g, 9.7 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred 25 at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (3 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure affording a vellow oil. Addition of hexane afforded the product as a yellow precipitate which was collected by filtration and recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give yellow crystals (0.7 g, 53 %); mp 135-137 °C; elemental analysis (Found: C, 49.6; H, 2.8. Calc. for C<sub>18</sub>H<sub>13</sub>BrTe: C, 49.5; H, 3.0 %); IR (KBr disk): v<sub>max</sub> cm<sup>-1</sup> 2923s, 2839w, 1825w, 1602s, 30 1555s, 1472w, 1434s, 1408s, 1347w, 1328s, 1250w, 1229s, 1112w, 1097s, 1062w, 1039s, 1014s, 996s, 982w, 913w, 832vs, 803vs, 733vs, 693vs, 598s, 534w, 489w, 456s;  $\delta_{\rm H}$ (270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 7.91-7.83 (2 H, m, TePh 12,16-H), 7.58 (1 H, d, <sup>4</sup>J<sub>HH</sub> 7.4 Hz, Acenap 4-H), 7.41-7.33 (1 H, m, TePh 14-H), 7.31-7.22 (2 H, m, TePh 13,15-H), 7.17 (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 7-H), 6.99 (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 3-H), 6.81 (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 8-H), 3.18 (4 H, s, 2 x CH<sub>2</sub>);  $\delta_{C}(67.9 \text{ MHz}; \text{CDCl}_{3}; 25 \text{ °C}; \text{Me}_{4}\text{Si})$  147.6(q), 146.1(q), 142.5(q), 141.8(s), 136.0(s), 133.6(s), 132.9(q),  $_{35}$  130.4(s), 129.3(s), 121.9(s), 120.9(s), 119.4(q), 117.3(q), 30.4(s, CH<sub>2</sub>), 30.2(s, CH<sub>2</sub>);  $\delta_{Te}(81.2 \text{ MHz}; \text{CDCl}_3; 25 \text{ °C};$ PhTeTePh) 696.0(s); MS (ES<sup>+</sup>): m/z 468.89 (100 %, M + OMe,).

**5-Iodo-6-(phenylsulfanyl)acenaphthene [Acenap(I)(SPh)] (A4):** Synthesis of A4 from A1: A solution of *n*-butyllithium (2.5 M) in hexane (1.07 mL, 2.66 mmol) was added dropwise to a solution of 5-bromo-6-(phenylsulfanyl)acenaphthene

(0.909 g, 2.66 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of iodine (0.676 g, 2.66 mmol) in diethyl ether (30 mL) was added dropwise whilst the temperature of the mixture was maintained between -10-0 °C. The resulting mixture was stirred at -10 °C for 1/2 h and for a further 1 h without cooling. Aqueous 5% sodium thiosulfate (60 mL) was added to the mixture with vigorous stirring. The organic layer was swahed with aqueous sodium thiosulfate and then water, dried with magnesium sulfate and concentrated under reduced pressure affording a cream solid. The crude solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.8 g, 76 %); mp 92-94 °C; elemental analysis (Found: C, 55.7; H, 3.0. Calc. for C<sub>18</sub>H<sub>13</sub>IS: C, 55.7, H, 3.4 %); IR (KBr disk): v<sub>max</sub> cm<sup>-1</sup> 2919s, 2819w, 1936w, 1872w, 1789w, 1722w, 1654w, 1595s, 1578vs, 1474vs, 1434vs, 1415s, 1403s, 1360w, 1341w, 1318vs, 1268w, 1254w, 1228s, 10 1202s, 1178w, 1150w, 1105s, 1078s, 1022s, 998w, 941w, 894w, 858s, 837vs, 811s, 733vs, 685vs, 624s, 598s, 478w, 463s, 401w; δ<sub>H</sub>(270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 8.12 (1 H, d, <sup>3</sup>*J*<sub>HH</sub> 7.4 Hz, Acenap 4-H), 7.71 (1 H, d, <sup>3</sup>*J*<sub>HH</sub> 7.2 Hz, Acenap 7-H), 7.19 (1 H, d, <sup>3</sup>*J*<sub>HH</sub> 7.2 Hz, Acenap 8-H), 7.17-7.09 (2 H, m, SPh 13,15-H), 7.07-7.00 (1 H, m, SPh 14-H), 6.98-6.93 (2 H, m, SPh 12,16-H), 6.91 (1 H, d, <sup>3</sup>*J*<sub>HH</sub> 7.4 Hz, Acenap 3-H), 3.39-3.20 (4 H, m, 2 x CH<sub>2</sub>); δ<sub>C</sub>(67.9 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 150.2(q), 148.5(q), 144.8(s), 141.5(q), 140.8(q), 140.4(s), 132.5(q), 129.4(s), 128.1(s), 126.2(q), 125.7(s), 122.2(s), 15 120.9(s), 85.4(q), 30.6(s, CH<sub>2</sub>), 30.5(s, CH<sub>2</sub>); MS (ES<sup>+</sup>): m/z 410.98 (40 %, M + Na).

*Synthesis of* **A4** *from* **A14**: A solution of *n*-butyllithium (2.5 M) in hexane (0.52 mL, 1.29 mmol) was added dropwise to a solution of 5,6-diiodoacenaphthene (0.525 g, 1.29 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (0.282 g, 1.29 mmol) in diethyl ether (20 mL) was added <sup>20</sup> dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (3 x 40 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure affording a yellow oil. The crude compound was purified by column chromatography on silica gel (hexane) and the resulting yellow solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.3 g, 55 %).

**5-Iodo-6-(phenylselanyl)acenaphthene [Acenap(I)(SePh)] (A5):** *Synthesis of* **A5** *from* **A2**: A solution of *n*-butyllithium (2.5 M) in hexane (1.10 mL, 2.74 mmol) was added dropwise to a solution of 5-bromo-6-(phenylselenyl)acenaphthene (1.065 g, 2.74 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of iodine (0.695 g, 2.74 mmol) in diethyl ether (30 mL) was added dropwise whilst the temperature of the mixture <sup>30</sup> was maintained between -10-0 °C. The resulting mixture was stirred at -10 °C for 1/2 h and for a further 1 h without cooling. Aqueous 5% sodium thiosulfate (60 mL) was added to the mixture with vigorous stirring. The organic layer was washed with aqueous sodium thiosulfate and then water, dried with magnesium sulfate and concentrated under reduced pressure affording a cream solid. The crude solid was purified by column chromatography on silica gel (hexane) and the resulting yellow crystalline solid was recrystallised by diffusion of hexane into a saturated solution of the compound in <sup>35</sup> dichloromethane to give the product as colourless crystals (0.7 g, 60 %); mp 82-84 °C; elemental analysis (Found: C, 49.7; H, 2.7. Calc. for C<sub>18</sub>H<sub>13</sub>ISe: C, 49.7; H, 3.0 %); IR (KBr disk):  $v_{max}$  cm<sup>-1</sup> 2914s, 2867w, 2824w, 1971w, 1950w, 1898w, 1874w, 1860w, 1803w, 1651w, 1597w, 1552w, 1472s, 1434s, 1403s, 1339w, 1320s, 1299w, 1270w, 1249s, 1230s, 1157w, 1105s, 1000s, 967w, 953w, 936w, 915w, 835vs, 809vs, 738vs, 690s, 598s, 536w, 498w, 463s, 385w,  $\delta_{\rm H}(270 \text{ MHz}; \text{ CDCl}_3; 25 °C; Me_4Si) 8.00 (1 H, d.<sup>3</sup>J_{\rm HH} 7.4 Hz, Acenap 4-H), 7.42-7.33 (2 H, m, Se$ *Ph*12,16-H), 7.31 (1 H,

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d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 7-H), 7.22-7.14 (3 H, m, Se*Ph* 13-15-H), 6.90 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 8-H), 6.81 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.4 Hz, Acenap 3-H), 3.21-3.08 (4 H, m, 2 x *CH*<sub>2</sub>);  $\delta_{\text{C}}(67.9 \text{ MHz}; \text{CDCl}_3; 25 ^{\circ}\text{C}; \text{Me}_4\text{Si})$  148.3(q), 147.5(q), 143.5(s), 141.8(q), 135.9(s), 134.7(s), 134.2(q), 133.2(q), 130.0(s), 128.1(s), 126.6(q), 122.0(s), 121.2(s), 86.9(q), 30.5(s, *C*H<sub>2</sub>), 30.3(s, *C*H<sub>2</sub>);  $\delta_{\text{Se}}(51.5 \text{ MHz}; \text{CDCl}_3; 25 ^{\circ}\text{C}; \text{PhSeSePh})$  400.9(s); MS (ES<sup>+</sup>): *m/z* 435.91 (3 %, M + H), 308.01 (100, M - I).

*Synthesis of* **A5** *from* **A14**: A solution of *n*-butyllithium (2.5 M) in hexane (0.58 mL, 1.45 mmol) was added dropwise to a solution of 5,6-diiodoacenaphthene (0.589 g, 1.45 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (0.453 g, 1.45 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (3 x 40 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure affording a yellow oil. The crude compound was purified by column chromatography on silica gel (hexane) and the resulting yellow solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.2 g, 30 %).

<sup>15</sup> **5-Iodo-6-(phenyltelluro)acenaphthene [Acenap(I)(TePh)] (A6):** *Synthesis of* **A6** *from* **A14**: A solution of *n*-butyllithium (2.5 M) in hexane (0.50 mL, 1.26 mmol) was added dropwise to a solution of 5,6-diiodoacenaphthene (0.510 g, 1.26 mmol) in diethyl ether (40 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (0.514 g, 1.26 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (3 x 60 mL). The organic layer was dried with magnesium <sup>20</sup> sulfate and concentrated under reduced pressure affording a yellow oil. The crude compound was purified by column chromatography on silica gel (hexane/dichloromethane 9:1) and the resulting yellow solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.1 g, 22 %); mp 77-79 °C; IR (KBr disk):  $v_{max}$  cm<sup>-1</sup> 2922w, 2834w, 1599w, 1547s, 1469w, 1431s, 1405s, 1341w, 1316s, 1263w, 1248s, 1228s, 1176w, 1154w, 1112w, 1095s, 1062w, 1036w, 1009s, 996s, 984w, 951w, 932w, 913w, 832vs, 801vs, 732vs, <sup>25</sup> 691vs, 594s, 532w, 484w, 454s, 378w, 359w;  $\delta_{\rm H}(270 \text{ MHz; CDCl}_3; 25 °C; Me_4Si) 8.08 (1 H, d, <sup>3</sup><sub>JHH</sub> 7.3 Hz, Acenap 4-H), 7.96-7.89 (2 H, m, Te$ *Ph*12,16-H), 7.49-7.31 (4 H, m, Acenap 7-H, Te*Ph* $13-15-H), 7.00 (1 H, d, <sup>3</sup><sub>JHH</sub> 7.3 Hz, Acenap 3-H), 6.92 (1 H, d, <sup>3</sup><sub>JHH</sub> 7.3 Hz, Acenap 8-H), 3.36-3.26 (4 H, m, 2 x CH<sub>2</sub>); <math>\delta_{\rm C}(67.9 \text{ MHz; CDCl}_3; 25 °C; Me_4Si) 147.0(q), 141.9(s), 141.2(s), 138.4(q), 137.3(s), 130.4(s), 129.2(s), 128.0(q), 121.9(s), 121.8(s), 121.7(q), 113.9(q), 30.4(s, CH<sub>2</sub>), 30.0(s, CH<sub>2</sub>); <math>\delta_{\rm Te}(81.2 \text{ MHz; CDCl}_3; 25 °C; PhTeTePh) 662.0(s); MS (ES<sup>+</sup>):$ *m/z*516.75 (100 %, M + OMe).

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5-Iodo-6-bromoacenaphthene [Acenap(I)(Br)] (A15): *Synthesis of* A15 *from* A3: A solution of *n*-butyllithium (2.5 M) in hexane (0.33 mL, 0.829 mmol) was added dropwise to a solution of 5-bromo-6-(phenyltelluro)acenaphthene (0.362 g, 0.829 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of iodine (0.210 g, 0.829 mmol) in diethyl ether (30 mL) was added dropwise whilst the temperature of the mixture was maintained between -10-0°C. The resulting mixture was stirred at -10 °C for 1/2 h and for a further 1 h without cooling. Aqueous 5% sodium thiosulfate (60 mL) was added to the mixture with vigorous stirring. The organic layer was washed with aqueous sodium thiosulfate and then water, dried with magnesium sulfate and concentrated under reduced pressure affording a cream solid. The crude solid was washed with hexane and the white precipitate was collected by filtration. Recrystallisation by diffusion of hexane into a saturated solution of the compound in dichloromethane gave the product as

colourless crystals (0.05 g, 17 %); mp 144-146 °C; elemental analysis (Found: C, 40.7; H, 2.0. Calc. for C<sub>12</sub>H<sub>8</sub>IBr: C, 40.15; H, 2.25 %); IR (KBr disk):  $v_{max}$  cm<sup>-1</sup> 2938w, 2924w, 2819w, 1855w, 1595s, 1554w, 1434w, 1403s, 1344w, 1316s, 1249w, 1225s, 1142w, 1100s, 1067w, 1041w, 1014s, 939w, 887w, 834vs, 809s, 768w, 733w, 695w, 681w, 595s, 539w, 489w, 456w, 361w, 328w;  $\delta_{H}$ (270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 8.16 (1 H, d,  ${}^{3}J_{HH}$  7.4 Hz, Acenap 7-H), 7.74 (1 H, d,  ${}^{3}J_{HH}$  5 7.4 Hz, Acenap 4-H), 7.03 (1 H, d,  ${}^{3}J_{HH}$  7.4 Hz, Acenap 3-H), 6.87 (1 H, d,  ${}^{3}J_{HH}$  7.4 Hz, Acenap 8-H), 3.30-3.16 (4 H, m, 2 x CH<sub>2</sub>);  $\delta_{C}$ (67.9 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 148.6(q), 148.1(q), 145.0(s), 141.6(q), 136.1(s), 129.7(q), 122.3(s), 121.3(s), 116.6(q), 85.6(q), 30.5(s, CH<sub>2</sub>), 30.3(s, CH<sub>2</sub>); MS (ES<sup>-</sup>): *m/z* 380.68 (100 %, M + Na – H).

**5,6-Bis(phenylsulfanyl)acenaphthene [Acenap(SPh)<sub>2</sub>] (A7):** *Synthesis of* **A7** *from* **A1**: A solution of *n*-butyllithium (2.5 M) in hexane (0.66 mL, 1.65 mmol) was added dropwise to a solution of 5-bromo-6-(phenylsulfanyl)acenaphthene (0.562 g, 1.65 mmol) in diethyl ether (20 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (0.360 g, 1.65 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 20 mL). The organic layer was dried with magnesium sulphate and concentrated under reduced pressure. The residual crude oil was purified by column <sup>15</sup> chromatography on silica gel (hexane/dichloromethane 9:1) and the resulting white solid was recrystallised by diffusion of pentane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.4 g, 69%); mp 108-110 °C; elemental analysis (Found: C, 77.9; H, 4.9. Calc. for C<sub>24</sub>H<sub>18</sub>S<sub>2</sub>: C, 77.8; H, 4.9 %); IR (KBr disk):  $v_{max}$  cm<sup>-1</sup> 2908w, 1655w, 1637w, 1597w, 1580s, 1475s, 1436s, 1415s, 1341w, 1322s, 1296w, 1256w, 1230w, 1199w, 1084s, 1067w, 1022s, 955w, 882w, 839s, 733vs, 686s, 622w, 569w, 505w, 463w;  $\delta_{H}$ (270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) <sup>20</sup> 7.40 (2 H, d, <sup>3</sup>J<sub>HH</sub> 7.3 Hz, Acenap 4,7-H), 7.16-7.10 (6 H, m, Acenap 3,8-H, SPh 12,16,18,22-H), 7.10-7.02 (6 H, m, SPh 13-15,19-21-H), 3.31 (4 H, s, 2 x CH<sub>2</sub>);  $\delta_{C}$ (67.9 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 147.5(q), 141.4(q), 139.2(q), 136.4(s), 129.9(s), 129.1(s), 129.0(s), 127.8(q), 126.1(s), 120.4(s), 30.2(s, 2 x CH<sub>2</sub>); MS (ES<sup>+</sup>): *m/z* 393.02 (100, M + Na).

*Synthesis of* **A7** *from* **A4**: A solution of *n*-butyllithium (2.5 M) in hexane (0.2 mL, 0.51 mmol) was added dropwise to a <sup>25</sup> solution of 5-iodo-6-(phenylsulfanyl)acenaphthene (0.199 g, 0.51 mmol) in diethyl ether (20 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (0.112 g, 0.51 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 20 mL). The organic layer was dried with magnesium sulphate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane/dichloromethane 9:1) and the resulting <sup>30</sup> white solid was recrystallised by diffusion of pentane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.05 g, 28%).

Synthesis of A7 from A13: A solution of 5,6-dibromoacenaphthene (0.592 g, 1.897 mmol) in diethyl ether (40 mL) was cooled to -10 - 0 °C on an ice-ethanol bath and to this was added a solution of TMEDA (0.8 mL, 5.04 mmol). The mixture <sup>35</sup> was allowed to stir for 15 min before a solution of *n*-butyllithium (2.5 M) in hexane (1.8 mL, 4.56 mmol) was added dropwise over a period of 15 min. During these operations, the temperature of the mixture was maintained at -10 - 0 °C. The mixture was stirred at this temperature for a further 1 h, before being cooled to -78 °C. A solution of diphenyl disulfide (0.829 g, 3.79 mmol) in diethyl ether (30 mL) was then added dropwise and the resulting solution was stirred at -78 °C for a further 1 h. The mixture was allowed to warm to room temperature and then washed with 0.1 N sodium hydroxide (3 x 60

mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure to afford a yellow solid. The crude product was purified by column chromatography on silica gel (hexane/dichloromethane) and the resulting cream powder was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.4 g, 54 %).

5,6-Bis(phenylselanyl)acenaphthene [Acenap(SePh)<sub>2</sub>] (A8): Synthesis of A8 from A2: A solution of n-butyllithium (2.5 M) in hexane (0.94 mL, 2.35 mmol) was added dropwise to a solution of 5-bromo-6-(phenylselenyl)acenaphthene (0.912 g, 2.35 mmol) in diethyl ether (20 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (0.734 g, 2.35 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at 10-78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 20 mL). The organic layer was dried with magnesium sulphate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting white solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.7 g, 60%); mp 104-106 °C; elemental analysis (Found: C, 62.12; H, 3.68. Calc. for  $C_{24}H_{18}Se_2$ : C, 62.08; H, 3.91 %); IR (KBr disk):  $v_{max}$  cm<sup>-1</sup> 15 2924w, 2872w, 2829w, 1957w, 1881w, 1857w, 1651w, 1637w, 1597s, 1573s, 1545w, 1474s, 1433s, 1410s, 1326s, 1301w, 1254w, 1232w, 1171w, 1157w, 1103s, 1069w, 1019s, 998w, 913w, 901w, 856w, 836vs, 811vs, 745vs, 732vs, 691s, 666w, 605s, 574w, 541w, 503w, 469s, 337w;  $\delta_{\rm H}$ (270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 7.45 (2 H, d,  ${}^{3}J_{\rm HH}$  7.3 Hz, Acenap 4,7-H), 7.40-7.34 (4 H, m, SePh 12,16,18,22-H), 7.34-7.20 (6 H, m, SePh 13-15,19-21-H), 7.04 (2 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 3,8-H), 3.27 (4 H, s, 2 x CH<sub>2</sub>);  $\delta_{\rm C}(67.9 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{Me}_4\text{Si})$  147.0(q), 141.3(q), 136.4(s), 134.6(q), 133.5(q), 133.3(s),  $_{20}$  129.3(s), 127.2(s), 125.6(q), 120.6(s), 30.1(s, CH<sub>2</sub>);  $\delta_{se}$ (51.5 MHz; CDCl<sub>3</sub>; 25 °C; PhSeSePh) 408.3(s); MS (ES<sup>+</sup>): *m/z* 465.96 (100 %, M + H).

*Synthesis of* **A8** *from* **A5**: A solution of *n*-butyllithium (2.5 M) in hexane (0.17 mL, 0.427 mmol) was added dropwise to a solution of 5-iodo-6-(phenylselanyl)acenaphthene (0.186 g, 0.427 mmol) in diethyl ether (30 mL) at -78 °C. The mixture <sup>25</sup> was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (0.133 g, 0.427 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 20 mL). The organic layer was dried with magnesium sulphate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane/dichloromethane 9:1) and the resulting white solid was recrystallised by diffusion of pentane into a saturated solution of the compound in dichloromethane to give <sup>30</sup> the product as colourless crystals (0.06 g, 28%).

Synthesis of A8 from A13: A solution of 5,6-dibromoacenaphthene (0.676 g, 2.167 mmol) in diethyl ether (40 mL) was cooled to -10 - 0 °C on an ice-ethanol bath and to this was added a solution of TMEDA (0.9 mL, 5.76 mmol). The mixture was allowed to stir for 15 min before a solution of *n*-butyllithium (2.5 M) in hexane (2.1 mL, 5.21 mmol) was added <sup>35</sup> dropwise over a period of 15 min. During these operations, the temperature of the mixture was maintained at -10 - 0 °C. The mixture was stirred at this temperature for a further 1 h, before being cooled to -78 °C. A solution of diphenyl diselenide (1.353 g, 4.33 mmol) in diethyl ether (30 mL) was then added dropwise and the resulting solution was stirred at -78 °C for a further 1 h. The mixture was allowed to warm to room temperature and then washed with 0.1 N sodium hydroxide (3 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure to

afford a red solid. The crude product was purified by column chromatography on silica gel (hexane/dichloromethane) and the resulting cream powder was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.5 g, 52 %).

- <sup>5</sup> **5,6-Bis(phenyltelluro)acenaphthene** [Acenap(TePh)<sub>2</sub>] (A9): *Synthesis of* A9 *from* A13: A solution of 5,6dibromoacenaphthene (0.548 g, 1.756 mmol) in diethyl ether (40 mL) was cooled to -10 - 0 °C on an ice-ethanol bath and to this was added a solution of TMEDA (0.7 mL, 4.67 mmol). The mixture was allowed to stir for 15 min before a solution of *n*-butyllithium (2.5 M) in hexane (1.7 mL, 4.22 mmol) was added dropwise over a period of 15 min. During these operations, the temperature of the mixture was maintained at -10 - 0 °C. The mixture was stirred at this temperature for a <sup>10</sup> further 1 h, before being cooled to -78 °C. A solution of diphenyl ditelluride (1.438 g, 3.51 mmol) in diethyl ether (30 mL) was then added dropwise and the resulting solution was stirred at -78 °C for a further 1 h. The mixture was allowed to warm to room temperature and then washed with 0.1 N sodium hydroxide (3 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure to afford a red solid. The crude product was purified by column chromatography on silica gel (hexane/dichloromethane) and the resulting cream powder was recrystallised by <sup>15</sup> diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.5 g, 55 %); mp 146-148 °C; elemental analysis (Found: C, 51.5; H, 2.95. Calc. for C<sub>24</sub>H<sub>18</sub>Te<sub>2</sub>: C, 51.3; H, 3.2 %); IR (KBr disk):  $v_{max}$ cm<sup>-1</sup> 2914w, 2867w, 2824w, 1599w, 1588w, 1567s, 1483w, 1469s, 1429s, 1403s, 1320s, 1261w, 1247w, 1230s, 1176w, 1150w, 1112w, 1097s, 1057s, 1013s, 996s, 896w, 834s, 806s, 729vs, 690vs, 650w, 598s, 534w, 484w, 449s, 311w;  $\delta_{\rm H}$  (270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 7.81 (2 H, d, <sup>3</sup>J<sub>HH</sub> 7.3 Hz, Acenap 4,7-H), 7.64-7.54 (4 H, m, Te*Ph* 12,16,18,22-H), 7.24-7.07 (6 H m; T, Ph 12, 15, 16, 21 H) 607 (0 H H  $^{3}$  Hz, 72 H have a result H) 2, 27 (4 H, m, Te*Ph* 12,16,18,22-H), 7.24-7.07
- <sup>20</sup> (6 H, m, Te*Ph* 13-15,19-21-H), 6.97 (2 H, d,  ${}^{3}J_{HH}$  7.3 Hz, Acenap 3,8-H), 3.27 (4 H, s, 2 x C*H*<sub>2</sub>);  $\delta_{C}$ (67.9 MHz; CDCl<sub>3</sub>; 25°C; Me<sub>4</sub>Si) 148.5(q), 142.5(s), 141.2(q), 139.0(q), 138.1(s), 129.9(s), 128.2(s), 122.9(q), 121.5(s), 113.5(q), 30.3(s, 2 x CH<sub>2</sub>);  $\delta_{Te}$ (81.2 MHz; CDCl<sub>3</sub>; 25 °C; PhTeTePh) 585.9(s); MS (ES<sup>+</sup>): *m/z* 592.94 (100, M + OMe).
- 5-(Phenylsulfanyl)-6-(phenylselanyl)acenaphthene [Acenap(SePh)(SPh)] (A10): Synthesis of A10 from A1: A solution 25 of *n*-butyllithium (2.5 M) in hexane (0.54 mL, 1.35 mmol) was added dropwise to a solution of 5-bromo-6-(phenylsulfanyl)acenaphthene (0.460 g, 1.35 mmol) in diethyl ether (20 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (0.421 g, 1.35 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 20 mL). The organic layer was dried with magnesium sulphate and concentrated under reduced pressure. The residual crude 30 oil was purified by column chromatography on silica gel (hexane) and the resulting white solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give the product as colourless crystals (0.2 g, 37%); mp 106-108 °C; elemental analysis (Found: C, 69.15; H, 4.6. Calc. for C<sub>24</sub>H<sub>18</sub>SSe: C, 69.1; H, 4.35 %); IR (KBr disk): v<sub>max</sub> cm<sup>-1</sup> 2914w, 2829w, 1959w, 1936w, 1860w, 1654w, 1595s, 1578s, 1475s, 1436s, 1408s, 1331vs, 1301w, 1254w, 1230w, 1216w, 1176w, 1152w, 1107w, 1082s, 1062w, 1021s, 998w, 936w, 915w, 894w, 865w, 836vs, 816s, <sup>35</sup> 778w, 747vs, 732vs, 690vs, 624w, 608w, 579w, 555w, 510w, 466s, 404w, 311w; δ<sub>H</sub>(270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 7.79  $(1 \text{ H}, d, {}^{3}J_{\text{HH}} 7.2 \text{ Hz}, \text{Acenap 4-H}), 7.64-7.59 (2 \text{ H}, \text{m}, \text{SePh 12,16-H}), 7.41-7.33 (3 \text{ H}, \text{m}, \text{SePh 13-15-H}), 7.31 (1 \text{ H}, d, \text{H})$ <sup>3</sup>J<sub>HH</sub> 7.2 Hz, Acenap 3-H), 7.26-7.19 (2 H, m, SPh 18,22-H), 7.16-7.09 (4 H, m, Acenap 7-H, SPh 19-21-H), 7.06 (1 H, d,  ${}^{3}J_{\text{HH}}$  7.0 Hz, Acenap 8-H), 3.46-3.40 (2 H, m, CH<sub>2</sub>), 3.40-3.33 (2 H, m, CH<sub>2</sub>);  $\delta_{\text{C}}$ (67.9 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 149.3(q), 145.0(q), 141.4(q), 140.6(q), 139.7(s), 136.3(s), 132.9(q), 132.4(q), 131.7(s), 129.5(s), 128.9(s), 128.2(s),

128.0(q), 127.4(s), 125.3(s), 125.1(q), 120.7(s), 120.3(s), 30.4(s, CH<sub>2</sub>), 29.8(s, CH<sub>2</sub>);  $\delta_{Se}(51.5 \text{ MHz}; \text{ CDCl}_3; 25 \text{ °C};$  PhSeSePh) 433.7(s); MS (ES<sup>+</sup>): *m/z* 441.05 (100 %, M + Na), 418.05 (25, M + H).

*Synthesis of* **A10** *from* **A2**: A solution of *n*-butyllithium (2.5 M) in hexane (0.30 mL, 0.783 mmol) was added dropwise to a solution of 5-bromo-6-(phenylselanyl)acenaphthene (0.304 g, 0.783 mmol) in diethyl ether (40 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (0.171 g, 0.783 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting white solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.03 g, 36 %).

*Synthesis of* **A10** *from* **A4**: A solution of *n*-butyllithium (2.5 M) in hexane (0.23 mL, 0.583 mmol) was added dropwise to a solution of 5-iodo-6-(phenylsulfanyl)acenaphthene (0.226 g, 0.583 mmol) in diethyl ether (20 mL) at -78 °C. The mixture <sup>15</sup> was stirred at this temperature for 1 h, after which a solution of diphenyl diselenide (0.182 g, 0.583 mmol) in diethyl ether (20 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 40 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting white solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless <sup>20</sup> crystals (0.05 g, 21 %).

*Synthesis of* **A10** *from* **A5**: A solution of *n*-butyllithium (2.5 M) in hexane (0.37 mL, 0.783 mmol) was added dropwise to a solution of 5-iodo-6-(phenylselanyl)acenaphthene (0.397 g, 0.912 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl disulfide (0.199 g, 0.912 mmol) in diethyl ether <sup>25</sup> (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 40 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting white solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.2 g, 52 %).

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**5-(Phenyltelluro)-6-(phenylsulfanyl)acenaphthene [Acenap(TePh)(SPh)] (A11):** *Synthesis of* **A11** *from* **A1**: A solution of *n*-butyllithium (2.5 M) in hexane (0.59 mL, 1.47 mmol) was added dropwise to a solution of 5-bromo-6-(phenylsulfanyl)acenaphthene (0.501 g, 1.47 mmol) in diethyl ether (40 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (0.602 g, 1.47 mmol) in diethyl ether (30 mL) was added <sup>35</sup> dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting cream powder was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.2 g, 30 %); mp 110-112 °C; elemental analysis (Found: C, 62.12; H, 3.68. Calc. for C<sub>24</sub>H<sub>18</sub>STe: C, 61.85; H, 3.89 %); IR (KBr disk):

 $v_{\text{max}}$  cm<sup>-1</sup> 2923w, 2824w, 1957w, 1931w, 1855w, 1637w, 1595s, 1578s, 1475s, 1434s, 1427s, 1407s, 1331s, 1299w, 1254w, 1232w, 1202w, 1178w, 1154w, 1105w, 1078w, 1058w, 1020s, 996w, 913w, 863w, 836s, 809s, 736vs, 693s, 624w, 602w, 487w, 455w, 397w;  $\delta_{\text{H}}(270 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{Me}_4\text{Si})$  7.88-7.81 (2 H, m, Te*Ph* 12,16-H), 7.72 (1 H, d,  $^3J_{\text{HH}}$  7.2 Hz, Acenap 4-H), 7.37-7.30 (1 H, m, Te*Ph* 14-H), 7.28-7.19 (3 H, m, Acenap 3-H, Te*Ph* 13,15-H), 7.18-7.07 (3 H, m, Acenap 5 7-H, S*Ph* 18,22-H), 7.05-6.95 (3 H, m, S*Ph* 19-21-H), 6.90 (1 H, d,  $^3J_{\text{HH}}$  7.4 Hz, Acenap 8-H), 3.39-3.22 (4 H, m, 2 x C*H*<sub>2</sub>);  $\delta_{\text{C}}(67.9 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{Me}_4\text{Si})$  150.2(q), 145.4(q), 141.4(q), 141.3(s), 140.4(q), 139.9(s), 135.6(q), 134.9(s), 129.7(s), 129.1(q), 129.0(s), 128.6(s), 126.9(s), 125.7(s), 124.8(q), 121.5(s), 120.2(s), 111.7(q), 30.5(s, CH<sub>2</sub>), 29.8(s, CH<sub>2</sub>);  $\delta_{\text{Te}}(81.2 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{PhTeTePh})$  689.4(s); MS (ES<sup>+</sup>): *m/z* 499.03 (100 %, M + OMe).

- <sup>10</sup> *Synthesis of* **A11** *from* **A4**: A solution of *n*-butyllithium (2.5 M) in hexane (0.29 mL, 0.737 mmol) was added dropwise to a solution of 5-iodo-6-(phenylsulfanyl)acenaphthene (0.286 g, 0.737 mmol) in diethyl ether (30 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (0.302 g, 0.737 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 40 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting yellow solid was recreated by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless.
- recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.1 g, 35 %).

5-(Phenvltelluro)-6-(phenvlselanvl)acenaphthene [Acenap(TePh)(SePh)] (A12): Synthesis of A12 from A2: A solution 20 of *n*-butyllithium (2.5 M) in hexane (0.75 mL, 1.88 mmol) was added dropwise to a solution of 5-bromo-6-(phenylselanyl)acenaphthene (0.731 g, 1.88 mmol) in diethyl ether (40 mL) at -78 °C. The mixture was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (0.771 g, 1.88 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 60 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil 25 was purified by column chromatography on silica gel (hexane) and the resulting yellow solid was recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.6 g, 62%); mp 138-140 °C; elemental analysis (Found: C, 56.4; H, 3.5. Calc. for C<sub>24</sub>H<sub>18</sub>SeTe: C, 56.2; H, 3.5 %); IR (KBr disk): v<sub>max</sub> cm<sup>-1</sup> 2947w. 2909w. 2834w. 1950w. 1876w. 1855w. 1595s. 1573s. 1474s. 1432s. 1405s. 1324s. 1299w. 1249w. 1230s. 1154w, 1112w, 1097s, 1065s, 1018s, 998w, 896w, 837vs, 807w, 764w, 729vs, 687vs, 664s, 614w, 600s, 539w, 489w, <sup>30</sup> 451s, 399w, 333w; δ<sub>H</sub>(270 MHz; CDCl<sub>3</sub>; 25 °C; Me<sub>4</sub>Si) 7.98 (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.2 Hz, Acenap 4-H), 7.96-7.91 (2 H, m, TePh 12,16-H), 7.46-7.41 (1 H, m, TePh 14-H), 7.38-7.31 (3 H, m, Acenap 7-H, TePh 13,15-H), 7.31-7.24 (3 H, m, Acenap 3-H, SePh 18,22-H), 7.24-7.14 (3 H, m, SePh 19-21-H), 7.00 (1 H, d, <sup>3</sup>J<sub>HH</sub> 7.4 Hz, Acenap 8-H), 3.44-3.34 (4 H, m, 2 x CH<sub>2</sub>);  $\delta_{\rm C}(67.9 \text{ MHz}; \text{ CDCl}_3; 25 \text{ °C}; \text{ Me}_4\text{Si}) 149.8(q), 145.8(q), 141.3(q), 141.1(s), 140.8(s), 136.0(q), 135.8(s), 135.7(q), 145.8(q), 141.3(q), 141.1(q), 140.8(q), 140.8(q), 145.8(q), 145.8(q), 141.8(q), 141.8(q), 140.8(q), 140.8(q), 145.8(q), 145.$ 129.8(s), 129.7(s), 129.2(s), 128.4(s), 126.4(s), 123.2(q), 122.3(q), 121.3(s), 120.4(s), 113.0(q), 30.4(s, CH<sub>2</sub>), 29.7(s, CH<sub>2</sub>);  $_{35} \delta_{Se}(51.5 \text{ MHz}; \text{CDCl}_3; 25 \text{ }^{\circ}\text{C}; \text{PhSeSePh}) 340.7 (s, {}^{4}J_{SeTe} \pm 715.6 \text{ Hz}); \delta_{Te}(81.2 \text{ MHz}; \text{CDCl}_3; 25 \text{ }^{\circ}\text{C}; \text{PhTeTePh}) 663.4 (s, s)$  ${}^{4}J_{\text{TeSe}} \pm 715.6 \text{ Hz}$ ; MS (ES<sup>+</sup>): *m/z* 544.87 (100 %, M + OMe).

Synthesis of A12 from A5: A solution of *n*-butyllithium (2.5 M) in hexane (0.26 mL, 0.657 mmol) was added dropwise to a solution of 5-iodo-6-(phenylselanyl)acenaphthene (0.286 g, 0.657 mmol) in diethyl ether (40 mL) at -78 °C. The mixture

was stirred at this temperature for 1 h, after which a solution of diphenyl ditelluride (0.269 g, 0.657 mmol) in diethyl ether (30 mL) was added dropwise. The resulting mixture was stirred at -78 °C for a further 1 h, then washed with 0.1 N sodium hydroxide (2 x 30 mL). The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The residual crude oil was purified by column chromatography on silica gel (hexane) and the resulting yellow solid was <sup>5</sup> recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (0.1 g, 32%).

- **5,6-diiodoacenaphthene**(Acenap(I)<sub>2</sub>] (A14): *Synthesis of* A14 *from* A13: A solution of TMEDA (6.42 mL, 42.7 mmol) was added to a solution of 5,6-dibromoacenaphthene (5.005 g, 16.04mmol) in diethyl ether (300 mL). To this *n*-<sup>10</sup> butyllithium (2.5 M) in hexane (15.4 mL, 38.50 mmol) was added dropwise at -78 °C and stirred at this temperature for a further 1 h. The temperature of the reaction mixture was allowed to warm to -10 °C after which a solution of iodine (9.038 g, 35.61 mmol) in diethyl ether (80 mL) was added dropwise. The resulting mixture was stirred at -10 °C for a further 1/2 h and for a further 1 h without cooling. Aqueous 5% sodium thiosulfate (300 mL) was added to the mixture with vigorous stirring. The organic layer was washed with aqueous sodium thiosulfate and then water, dried with magnesium sulfate and <sup>15</sup> concentrated under reduced pressure affording a pale brown solid. The title compound was purified by column chromatography on silica gel (hexane) and the resulting brown crystalline solid recrystallised by diffusion of hexane into a saturated solution of the compound in dichloromethane to give colourless crystals (1.9 g, 28 %); mp 158-159 °C (lit, <sup>ref</sup>, 159-160 °C); IR (KBr disk): *v*<sub>max</sub> cm<sup>-1</sup> 2938s, 2924s, 2862s, 2815s, 2625w, 2563w, 2436w, 2308w, 2232w, 2127w, 2023w, 1933w, 1857s, 1808w, 1770w, 1739w, 1649w, 1595s, 1550s, 1486w, 1431s, 1405s, 1355w, 1339w, 1311s, 1261s, 1249s, <sup>20</sup> 1223s, 1142w, 1097s, 1007s, 955w, 941w, 882w, 832vs, 806vs, 738s, 678s, 648w, 593s, 534w, 482s, 458w, 394w, 380w;
- $δ_{\rm H}(270 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{Me}_4\text{Si}) 8.23 (2 \text{ H}, \text{d}, {}^{3}J_{\rm HH} 7.5 \text{ Hz}, \text{Acenap 4,7-H}), 6.91 (2 \text{ H}, \text{d}, {}^{3}J_{\rm HH} 7.5 \text{ Hz}, \text{Acenap 3,8-H}),$ 3.26 (4 H, s, 2 x *CH*<sub>2</sub>);  $δ_{\rm C}(67.9 \text{ MHz}; \text{CDCl}_3; 25 \,^{\circ}\text{C}; \text{Me}_4\text{Si}) 149.0(\text{q}), 144.7(\text{s}), 131.3(\text{q}), 121.9(\text{s}), 121.0(\text{q}), 89.5(\text{q}),$ 30.0(s, 2 x *C*H<sub>3</sub>).

#### 2. Computational Details

- <sup>25</sup> Geometries were fully optimised in the gas phase at the B3LYP level<sup>[4]</sup> using Curtis and Binning's 962(d) basis<sup>[5]</sup> on Se and Br (augmented with a set of diffuse s and p functions on Br), the Stuttgart-Dresden effective core potentials along with their double zeta valence basis sets for Te and I<sup>[6]</sup> (augmented with d-polarisation functions with exponents of 0.237 and 0.241, respectively),<sup>[7]</sup> and 6-31+G(d) basis elsewhere. Wiberg bond indices<sup>8</sup> were obtained in a natural bond orbital analysis<sup>9</sup> at the same level. The optimisations were started from a number of different conformers for those compounds containing a
- <sup>30</sup> phenylchalcogen substituent. For the mono-phenyl chalcogens these were A and B. For the di-phenylchalcogens these were AAc, AAt, AB, and BB. Mixed chalcogen structures also included the BA conformer. For the dichalcogen compounds, experimental structures from X-ray crystallography were used as one of the starting conformers. The computations were performed using the Gaussian 03 suite of programs.<sup>[10]</sup>
- <sup>35</sup> The *J*(<sup>77</sup>Se, <sup>125</sup>Te) spin-spin coupling constants of **A12** and **N12** were computed<sup>[11]</sup> at the BP86 level<sup>[12]</sup> using the relativistic zeroth-order regular approximation both in its scalar formulation (ZORA)<sup>[13]</sup> and with spin-orbit coupling (ZORA-SO),<sup>[14]</sup> together with a TZ2P basis of Slater-type orbitals. A fine integration grid was used (Integration 6), as well as a finite nuclear size. These calculations were performed with the ADF program.<sup>[15,16]</sup> Employing the B3LYP-optimised geometries

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(BA conformations), *J* values of -490 Hz and -658 Hz were obtained for A12 and N12, respectively, at the ZORA level (-480 Hz and -647 Hz, respectively, at ZORA-SO). While the observed values of |J| = 715.6 Hz (A12) and 834.0 Hz (N12) are underestimated in absolute terms, the signs (which stem from the negative sign of the <sup>125</sup>Te gyromagnetic ratio) should be reliable and were used to assign a negative sign to the observed  $J(^{77}Se,^{125}Te)$  values.

Table S1: Selected optimised parameters WBIs, and relative energies (kJ/mol) of mono-phenylchalcogen acenaphthene derivatives A1-A6 (B3LYP level),
including structural data from X-ray crystallography, where available)

Compound	A1 XRD-A	A1-A	A1-B	A2 XRD-B	A2-A	A2-B
X, EPh	Br, S	Br, S	Br, S	Br, Se	Br, Se	Br, Se
Energy(rel)* / kJ/mol	-	0.00	0.32	-	6.05	0.00
X(1)…E(1) / Å	3.297(3)	3.3538	3.1888	3.1588(16)	3.4301	3.2231
WBI X(1)⋯E(1)	-	0.0060	0.0254	-	0.0079	0.0413
X(1)-C(1) / Å	1.926(7)	1.9255	1.9293	1.905(5)	1.9250	1.9295
E(1)-C(9) / Å	1.777(7)	1.7963	1.8093	1.943(5)	1.9456	1.9631
C(10)-C(9)-E(1)-C(13); θ1 / °	-89.44(1)	-81.399	170. 🗆 85	159.92(1)	-80.291	179.726
C(9)-E(1)-C(13)-C(14); γ1 / °	170.06(1)	168.816	102.258	95.94(1)	-2□.195	92.910
Compound	A3 XRD-B	A3-A	АЗ-В	A4 XRD-A	A4-A	A4-C
X, EPh	Br, Te	Br, Te	Br, Te	I, S	I, S	I, S
En [] rgy(rel)* / kJ/mol	-	15.11	0.00	-	0.00	1. 🗆 4
X(1)…E(1) / Å	3.2503(14)	3.5625	3.3189	3.428(4)	3.4799	3.3620
WBI X(1)…E(1)	-	0.0111	0.0620	-	0.0115	0.0290
X(1)-C(1) / Å	1.919(10)	1.9255	1.9309	2.129(13)	2.1514	2.1512
E(1)-C(9) / Å	2.148(10)	2.1499	2.1661	1.773(14)	1.7947	1.8074
C(10)-C(9)-E(1)-C(13); θ1 / °	162.39(1)	-77.267	175.394	90.15(1)	-85.000	150.547
C(9)-E(1)-C(13)-C(14); y1 / °	100.03(1)	153.061	96.172	151.86(1)	174.347	116.508
Compound	A5 XRD-B	A5-A	А5-В	A6 XRD-B	A6-A	A6-B
X, EPh	I, Se	I, Se	I, Se	I, Te	I, Te	I, Te
Ener [] y(rel)* / kJ/mol	-	4.82	0.00	-	15.13	0.00
X(1)…E(1) / Å	3.3291(11)	3.5653	3.3842	3.3721(17)	3.7054	3.4553
WBI X(1)…E(1)	-	0.0141	0.0504	-	0.0181	0.0841
X(1)-C(1) / Å	2.110(7)	2.1517	2.1497	2.058(16)	2.1517	2.1497
E(1)-C(9) / Å	1.930(7)	1.9435	1.9622	2.176(17)	2.1474	2.1693
C(10)-C(9)-E(1)-C(13); θ1 / °	-165.87(1)	-80.612	161.047	-161.42(1)	-84.399	179.978
$C(9)-E(1)-C(13)-C(14): \gamma 1 / °$	-92.60(1)	163 542	-80.713	-101.61(1)	160,161	92.576

 Table S2 Selected optimised parameters WBIs, and relative energies (kJ/mol) of di-phenylchalcogen acenaphthene derivatives A7-A12 (B3LYP level), including structural data from X-ray crystallography, where available)

Compound	A7 XRD-AAt	A7-AAt	A7-AB	A7-CCt
Eph, E'Ph	S, S	S, S	S, S	S, S
Energy(rel)* / kJ/mol	-	3.21	0.00	3.21
E(1)…E(2) / Å	3.274(4)	3.3227	3.1351	3.0862
WBI $E(1) \cdots E(2)$	-	0.0065	0.0314	0.0364
E(1)-C(1) / Å	1.812(12)	1.7912	1.8068	1.8060
E(2)-C(9) / Å	1.782(11)	1.7912	1.7952	1.8070
C(10)-C(1)-E(1)-C(13)□ θ1 / °	75.64(1)	82.320	164.523	139.201
C(10)-C(9)-E(2)-C(19); θ2 / °	92.07(1)	82.337	-86.587	144.087
C(1)-E(1)-C(13)-C(14); γ1 / °	-4.51(1)	8.688	-79.244	-57.640
C(9)-E(2)-C(19)-C(20); y2 / °	30.56(1)	8.635	-9.636	-62.374
Compound	A8 XRD-AB	A8-AAt	A8-AB	
Eph, E'Ph	Se, Se	Se, Se	Se, Se	
Energy(rel)* / kJ/mol	-	11.00	0.00	
E(1)…E(2) / Å	3.1834(10)	3.4720	3.2170	
WBI E(1)…E(2)	-	0.0123	0.0660	
E(1)-C(1) / Å	1.928(9)	1.9358	1.9382	
E(2)-C(9) / Å	1.932(8)	1.9358	1.9614	
C(10)-C(1)-E(1)-C(13); θ1 / °	- 7.70(1)	81.607	92.012	
C(10)-C(9)-E(2)-C(19); θ2 / °	163.05(1)	81.610	-174.155	
C(1)-E(1)-C(13)-C(14); γ1 / °	-148.68(1)	-168.996	-177.769	
C(9)-E(2)-C(19)-C(20); y2 / °	-84.31(1)	-168.992	-97.683	
Compound	A9 XRD-AB	A9-AAt	A9-AB	
Eph. E'Ph	Te. Te	Te□ Te	Te. Te	
Energy(rel)* / kJ/mol	-	22.77	0.00	
E(1)…E(2) / Å	3.3674(19)	3.7237	3.4228	
WBI $E(1) \cdots E(2)$	-	0.0234	0.1404	
E(1)-C(1) / Å	2.180(17)	2.1320	2.1674	
E(2)-C(9) / Å	2.155(17)	2.1346	2.1426	
C(10)-C(1)-E(1)-C(13); θ1 / °	166.60(1)	74.708	161.011	
C(10)-C(9)-E(2)-C(19); θ2 / °	79.56(1)	75.924	95.269	
C(1)-E(1)-C(13)-C(14); γ1 / °	86.37(1)	22.353	103.489	
C(9)-E(2)-C(19)-C(20); y2 / °	-167.34(1)	-149.017	113.643	
Compound	A10 XRD-BA	A10-AAt	A10-AB	A10-BA
Eph, E'Ph	Se, S	Se, S	Se, S	Se, S
Energy( el)* / kJ/mol	-	11.53	8.74	0.00
E(1)…E(2) / Å	3.113(4)	3.3946	3.2127	3.1313
WBI $E(1) \cdots E(2)$	-	0.0107	0.0386	0.0565
E(1)-C(1) / Å	1.953(10)	1.9452	1.9418	1.9600
E(2)-C(9) / Å	1.798(11)	1.7962	1.8076	1.7932
С(10)-С(1)-Е(1)-С(13); ө1 / °	156.26(1)	79.□07	85.260	-176.246
C(10)-C(9)-E(2)-C(19); θ2 / °	-100.46(1)	84.009	-168.158	93.960
C(1)-E(1)-C(13)-C(14); y1 / °	-82.12(1)	-167.224	-166.930	-96.424
C(9)-E(2)-C(19)-C(20); y2 / °	5.36(1)	5.482	83.269	-1.233

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Compound	A11 XRD-BA	A11-AAt	A11-BA	A11-AC
Eph, E'Ph	Te, S	Te, S	Te, S	Te, S
Energy(rel)* / kJ/mol	-	22.70	0.00	22.2534978
E(1)…E(2) / Å	3.1576(15)	3.5077	3.1908	3.3768
WBI $E(1) \cdots E(2)$	-	0.0149	0.0905	0.0387
E(1)-C(1) / Å	2.138(6)	2.1477	2.1608	2.1438
E(2)-C(9) / Å	1.781(7)	1.7960	1.7945	1.8071
C(10)-C(1)-E(1)-C(13); θ1 / °	166.03(1)	73.351	-179.857	83.607
C(10)-C(9)-E(2)-C(19); θ2 / °	-95.67(1)	88.092	100.995	-154.366
C(1)-E(1)-C(13)-C(14); γ1 / °	-83.57(1)	30.354	91.648	30.520
C(9)-E(2)-C(19)-C(20); γ2 / °	-150.20(1)	176.838	159.248	-114.342
Compound	A12 XRD-BA	A12-AAt	A12AB	A12-BA
Eph, E'Ph	Te, Se	Te, Se	Te, Se	Te, Se
Energy(rel)* / kJ/mol	-	23.88	16.22	0.00
E(1)…E(2) / Å	3.2479(19)	3.5863	3.3733	3.2668
WBI $E(1) \cdots E(2)$	-	0.0176	0.0704	0.1156
E(1)-C(1) / Å	2.152(10)	2.1484	2.1386	2.1638
E(2)-C(9) / Å	1.943(11)	1.9439	1.9638	1.9405
C(10)-C(1)-E(1)-C(13); θ1 / °	-168.14(1)	75.092	81.843	-173.618
C(10)-C(9)-E(2)-C(19); θ2 / °	101.19(1)	85.768	169.922	-98.243
C(1)-E(1)-C(13)-C(14); γ1 / °	83.63(1)	-151.654	-152.943	88.938
C(9)-E(2)-C(19)-C(20); γ2 / °	160.85(1)	176.127	99.097	-149.412
<sup><i>a</i></sup> Distances in Å, angles in degrees, relative of	energies in kJ/mol.			



Figure S1: Plot of optimised (B3LYP) vs. observed (X-ray diffraction) peri-distances in acenaphthene derivatives.

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Figure S2: Plot of the increase in the Wiberg bond index on going from AAt to AB conformers vs the concomitant contraction of the peri-distance (in Å, acenapthene series). The more pronounced this bond contraction, the higher the covalent contribution.



Figure S3: Plot of the electrostatic potential of A9-AAt (left) and A9-AB (right), mapped onto isodensity surfaces ( $\rho = 4 \cdot 10^{-3}$  au); color code: red 50 kJ/mol; green: zero; blue -50 kJ/mol.

 Table S3: Selected optimised parameters, WBIs, and relative energies of naphthalene derivatives N1-N6 (B3LYP level), including structural data from X-ray crystallography, where available).

Compound	N1 (Oil)	N1-A	N1-B		
X, EPh	Br, SPh	Br, SPh	Br, SPh		
Energy(rel)* / kJ/mol	-	3.04	0.00		
$X(1)\cdots E(1)$	-	3.2879	3.1314		
WBI X(1)…E(1)	-	0.0066	0.0290		
X(1)-C(1)	-	1.9280	1.9303		
E(1)-C(9)	-	1.8001	1.8109		
C(10)-C(9)-E(1)-C(11); 01 / °	-	76.897	158.707		
C(9)-E(1)-C(11)-C(12); γ1 / °	-	18.536	109.227		
Compound	N2 XRD-B	N2-A	N2-B	N3 XRD-B	N3-B
X, EPh	Br, SePh	Br, SePh	Br, SePh	Br, TePh	Br, TePh
Energy(rel)* / kJ/mol	-	9.85	0.00	-	0.00
$X(1)\cdots E(1)$	3.1136(6)	3.3689	3.1731	3.1909(10)	3.2664
WBI X(1)…E(1)	-	0.0086	0.0454	-	0.0675
X(1)-C(1)	1.919(3)	1.9268	1.9297	1.917(6)	1.9316
E(1)-C(9)	1.948(3)	1.9505	1.9655	2.153(6)	2.1704
C(10)-C(9)-E(1)-C(11); 01 / °	-159.2(3)	-71.119	159.200	-160.0(6)	162.873
C(9)-E(1)-C(11)-C(12); γ1 / °	-96.0(3)	-25.459	105.765	-99.4(5)	100.731
Compound	N4 XRD-A	N4-C	N5 XRD-B	N5-C	
X, EPh	I, Sph	I, SPh	I, SePh	I, SePh	
Energy(rel)* / kJ/mol	-	0.00	-	0.00	
$X(1)\cdots E(1)$	3.338(11)	3.2973	3.2524(8)	3.3355	
WBI X(1)…E(1)	-	0.0354	-	0.0566	
X(1)-C(1)	2.118(11)	2.1538	2.122(4)	1.9641	
E(1)-C(9)	1.765(11)	1.8096	1.958(4)	2.1519	
C(10)-C(9)-E(1)-C(11); θ1 / °	83.6(10)	148.508	-157.0(4)	151.829	
C(9)-E(1)-C(11)-C(12); γ1 / °	1.4(10)	116.268	-89.4(4)	-76.273	
Compound	N6 XRD-B	N6-B			
X EPh	I TePh	I TePh			
Energy(rel)* / kJ/mol	-,	0.00			
$X(1)\cdots F(1)$	3 3146(6)	3 4253			
WBLX(1)E(1)	-	0.0867			
X(1)-C(1)	2,108(6)	2.1518			
E(1)-C(9)	2.151(6)	2.1687			
C(10)-C(9)-E(1)-C(11): 01 / °	-158.1(6)	152.606			
C(9)-E(1)-C(11)-C(12); γ1 / °	-86.5(5)	-78.210			
<sup><i>a</i></sup> Distances in Å, angles in degrees, relative e	nergies in kJ/mol.				

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Table S4: Selected optimised parameters, WBIs, and relative energies of naphthalene derivatives N7-N12 (B3LYP level), including structural data from X-ray crystallography, where available).

Compound	N7 XRD-AB	N7-AAt	N7-AB	N7-CCt
Eph, E'Ph	SPh, SPh	SPh, SPh	SPh, SPh	SPh, SPh
Energy(rel)* / kJ/mol	-	5.93	0.00	0.69
E(1)…E(2)	3.0036(13)	3.2533	3.0498	3.0071
WBI $E(1) \cdots E(2)$	-	0.0099	0.0396	0.0446
E(1)-C(1)	1.794(3)	1.8000	1.8099	1.8092
E(2)-C(9)	1.783(4)	1.8000	1.7977	1.8095
C(10)-C(1)-E(1)-C(11); θ1 / °	159.8(2)	76.041	-179.798	138.628
C(10)-C(9)-E(2)-C(17); θ2 / °	95.0(3)	76.040	81.680	138.761
C(1)-E(1)-C(11)-C(12); γ1 / °	-73.62(1)	13.345	92.056	-60.353
C(9)-E(2)-C(17)-C(18); γ2 / °	-24.10(1)	13.351	12.741	124.984
Compound	N8 XRD-AB	N8-AAt	N8-AB	N8-CCt
Eph, E'Ph	SePh, SePh	SePh, SePh	SePh, SePh	n SePh, SePh
Energy(rel)* / kJ/mol	-	15.18	1.52	0.00
E(1)…E(2)	3.1332(9)	3.4140	3.1522	3.1171
WBI E(1)…E(2)	-	0.0140	0.0□81	0.0821
E(1)-C(1)	1.922(7)	1.9494	1.9660	1.9594
E(2)-C(9)	1.930(7)	1.9494	1.9449	1.9576
C(10)-C(1)-E(1)-C(11); θ1 / °	155.35(1)	65.404	159.999	136.310
C(10)-C(9)-E(2)-C(17); 02 / °	-109.990	65.402	-107.051	131.088
C(1)-E(1)-C(11)-C(12); γ1 / °	-83.57(1)	35.143	-80.441	-63.715
C(9)-E(2)-C(17)-C(18); γ2 / °	10.88(1)	35.068	-155.136	-59.079
Compound	N9 XRD-CCt	N9-CCc	N9-CCt	
Eph. E'Ph	TePh. TePh	TePh. TePh	TePh. TePl	1
Energy(rel)* / kJ/mol	-	3.25	0.00	
E(1)…E(2)	3.287(1)	3.3907	3.3408	
WBI E(1)…□(2)	-	0.1407	0.1497	
E(1)-C(1)	2.14(1)	2.1720	2.1574	
E(2)-C(9)	2.14(2)	2.1449	2.1575	
C(10)-C(1)-E(1)-C(11); θ1 / °	-124.79(1)	-118.985	-129.438	
C(10)-C(9)-E(2)-C(17); 02 / °	-132.46(1)	152.358	-129.469	
C(1)-E(1)-C(11)-C(12); γ1 / °	100.88(1)	-127.785	□117.102	
C(9)-E(2)-C(17)-C(18); y2 / °	57.21(1)	-77.057	65.924	
Compound	N10 YRD_RA	N10-AAt	N10_RA	
Enh F'Ph	SePh SPh	SePh SPh	SePh SPh	
Energy(rel)* / k I/mol	-	15 11	0.00	
$F(1)\cdots F(2)$	3 ()63(2)	3 3326	3 0567	
$WBIF(1)\cdots F(2)$	5.005(2)	0.0118	0.0676	
$\mathbf{F}(1) - \mathbf{C}(1 \square$	1 907(9)	1 9517	1 9642	
E(2)-C(9)	1.813(8)	1,7985	1 7959	
С(10)-С(1)-Е(1)-С(11): Ө1 / °	-156 4(7)	66 787	168 716	
С(10)-С(9)-Е(2)-С(17): θ2 / °	111 3(7)	69 547	91 079	
$C(1)-E(1)-C(11)-C(12): \gamma 1 / °$	-98.0(8)	-145.834	100.267	
C(9)-E(2)-C(17)-C(18); $\gamma 2 / ^{\circ}$	-9.3(9)	-157.852	174.732	

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Compound	N11 XRD-BA	N11-BA	N12 XRD-BA	N12-BA
Eph, E'Ph	TePh, SPh	TePh, SPh	TePh, SePh	TePh, SePh
Energy(rel)* / kJ/mol	-	0.00	-	0.00
E(1)…E(2)	3.0684(13)	3.1202	3.1919(11)	3.2035
WBI E(1)…E(2)	-	0.1028	-	0.1295
E(1)-C(1)	2.141(5)	2.1649	2.137(11)	2.1681
E(2)-C(9)	1.770(5)	1.7954	1.920(12)	1.9444
C(10)-C(1)-E(1)-C(11); θ1 / °	-160.5(11)	169.653	-165.3(5)	164.033
C(10)-C(9)-E(2)-C(17); θ2 / °	-83.1(11)	94.898	-84.0(5)	95.389
C(1)-E(1)-C(11)-C(12); γ1 / °	108.0(11)	97.895	-89.5(4)	-83.532
C(9)-E(2)-C(17)-C(18); γ2 / °	-12.3(9)	-10.877	-12.5(□)	-32.494
<sup>a</sup> Distances in Å, angles in degrees, relative	energies in kJ/mol.			



Figure S4: Plot of optimised (B3LYP) vs. observed (X-ray diffraction) peri-distances in naphthalene derivatives.



Figure S5: Plot of B3LYP optimised peri-distances vs. sum of van der Waals radii in the naphthalene series. Note that no A or AA minima could be located for any of the I and Te compounds - the optimisations went directly to B, C, AB or CC conformers.

Peri-distance: DFT conformers vs. sum of van der Waals radii

#### 3. Crystal structure analyses

X-ray crystal structures for A1-A7, A9, A10, A12, A14 and A15 were determined at -148(1) °C on the St Andrews Robotic Diffractometer<sup>17</sup> a Rigaku ACTOR-SM, Saturn 724 CCD area detector with graphite monochromated Mo-K $\alpha$  radiation ( $\lambda = 50.71073$  Å). Data for compound A8 were collected at -180(1) °C by using a Rigaku MM007 High brilliance RA generator (Mo K $\alpha$  radiation, confocal optic) and Mercury CCD system. At least a full hemisphere of data was collected using  $\omega$  scans. Data for compound A11 were collected at -180(1) °C by using a Rigaku MM007 High brilliance RA generator (Mo K $\alpha$  radiation, confocal optic) and Saturn CCD system. At least a full hemisphere of data was collected using  $\omega$  scans. Data for compound A11 were collected at -180(1) °C by using a Rigaku MM007 High brilliance RA generator (Mo K $\alpha$  radiation, confocal optic) and Saturn CCD system. At least a full hemisphere of data was collected using  $\omega$  scans. Data for compound A16 were collected at -148(1) °C on a Rigaku SCXmini CCD area detector with graphite monochromated <sup>10</sup> Mo K $\alpha$  radiation ( $\lambda = 0.71073$  Å). All data had intensities corrected for Lorentz, polarisation and absorption. The data for the complexes was collected and processed using CrystalClear (Rigaku). <sup>18</sup> The structures were solved by Patterson or direct methods and expanded using Fourier techniques. Non-hydrogen atoms were refined anisotropically, and hydrogen atoms were refined using a riding model. All calculations were performed using the CrystalStructure<sup>19</sup> and SHELXL-97.<sup>20</sup> These X-ray data can be obtained free of charge via www.ccdc.cam.ac.uk/conts/retrieving.html or from the Cambridge

<sup>15</sup> Crystallographic Data centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax (+44) 1223-336-033; e-mail: deposit@ccdc.cam.ac.uk CCDC Nos: A1-A12 816080-81691, A14-A16 816091-816094.



Fig. S6 Molecular structures of A3-A6 (H atoms omitted for clarity).



Fig. S7 Molecular structures of A10-A12 (H atoms omitted for clarity).

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Fig. S8 Molecular structures of A14 and A16 (H atoms omitted for clarity).

Compound	A1	A2	A3	A4	A5	A6
X, EPh	Br,SPh	Br,SePh	Br,TePh	I,SPh	I,SePh	I,TePh
Peri-region-distances						
X(1)…E(1)	3.297(3)	3.1588(16)	3.2503(14)	3.428(4)	3.3291(11)	3.3721(17) [3.3774(17)]
$\Sigma r_{vdW}$ - X···E <sup><i>a</i></sup>	0.353	0.5912	0.6597	0.352	0.5509	0.6679 [0.6626]
$2 n \Sigma r_{vdW}^{a}$	90	84	83	91	86	83 [84]
X(1)-C(1)	1.926(7)	1.905(5)	1.919(10)	2.129(13)	2.110(7)	2.058(16) [2.067(16)]
E(1)-C(9)	1.777(7)	1.943(5)	2.148(10)	1.773(14)	1.930(7)	2.176(17) [2.151(18)]
Acenaphthene bond len	gths					
C(1)-C(2)	1.383(10)	1.361(8)	1.369(14)	1.388(17)	1.360(10)	1.37(2) [1.40(2)]
C(2)-C(3)	1.387(10)	1.429(8)	1.389(16)	1.42(2)	1.407(11)	1.37(3) [1.37(3)]
C(3)-C(4)	1.355(11)	1.362(8)	1.376(16)	1.399(18)	1.368(11)	1.39(3) [1.45(2)]
C(4)-C(5)	1.411(9)	1.416(8)	1.415(14)	1.423(17)	1.401(9)	1.41(2) [1.39(2)]
C(5)-C(10)	1.421(9)	1.423(7)	1.434(14)	1.438(17)	1.417(9)	1.43(2) [1.42(2)]
C(5)-C(6)	1.434(10)	1.405(8)	1.420(15)	1.433(17)	1.428(9)	1.42(2) [1.40(2)]
C(6)-C(7)	1.358(10)	1.367(8)	1.364(15)	1.344(19)	1.361(10)	1.40(2) [1.37(2)]
C(7)-C(8)	1.385(10)	1.405(8)	1.409(14	1.422(20)	1.390(10)	1.44(2) [1.39(2)]
C(8)-C(9)	1.396(11)	1.377(8)	1.396(14)	1.378(18)	1.391(10)	1.34(2) [1.38(2)]
C(9)-C(10)	1.437(9)	1.438(8)	1.434(13)	1.444(17)	1.444(9)	1.44(2) [1.47(2)]
C(10)-C(1)	1.442(10)	1.431(8)	1.411(13)	1.444(17)	1.433(10)	1.44(2) [1.43(2)]
C(4)-C(11)	1.536(10)	1.514(8)	1.505(15)	1.53(2)	1.513(10)	1.50(3) [1.50(2)]
C(11)-C(12)	1.536(11)	1.562(9)	1.538(17)	1.563(19)	1.545(11)	1.48(3) [1.53(3)]
C(12)-C(6)	1.535(9)	1.510(8)	1.488(17)	1.529(17)	1.510(10)	1.45(2) [1.52(2)]
Peri-region bond angles	5					
X(1)-C(1)-C(10)	122.3(5)	121.8(4)	121.6(7)	125.3(9)	123.8(5)	124.6(12) [123.7(11)]
C(1)-C(10)-C(9)	132.5(6)	131.4(5)	131.7(9)	132.6(12)	131.4(6)	130.5(15) [130.3(14)]
E(1)-C(9)-C(10)	125.5(6)	121.4(4)	123.5(7)	124.4(10)	122.3(5)	123.2(12) [124.4(10)]
$\Sigma$ of bay angles	380.3(11)	74.6(9)	376.8(15)	382.3(20)	377.5(11)	378.3(32) [378.4(28)]
Splay angle <sup>b</sup>	20.3	14.6	16.8	22.3	17.5	18.3 [18.4]
C(4)-C(5)-C(6)	112.1(6)	111.4(5)	110.8(9)	111.9(10)	111.1(6)	110.4(15) [109.8(14)]
X(1)-E(1)-C(13)	87.4(1)	172.83(1)	169.0(1)	86.46(1)	177.69(1)	171.31(1) [171.48(1)]
Out-of-plane displacem	ent					
X(1)	0.149(1)	0.287(1)	0.307(1)	-0.201(1)	-0.439(1)	-0.425(1) [0.333(1)]
E(1)	-0.121(1)	-0.359(1)	-0.298(1)	0.092(1)	0.364(1)	0.334(1) [-0.395(1)]
Central naphthalene rir	ng torsion angles					
C:(6)-(5)-(10)-(1)	178.76(1)	175.27(1)	178.17(1)	179.40(1)	174.34(1)	175.29(1) [175.81(1)]
C:(4)-(5)-(10)-(9)	176.84(1)	177.24(1)	176.39(1)	177.51(1)	177.91(1)	-177.33(1) [178.08(1)]

а

<sup>*a*</sup> van der Waals radii used for calculations:  $r_{vdW}(S)$  1.80Å,  $r_{vdW}(Se)$  1.90Å,  $r_{vdW}(Te)$  2.06Å,  $r_{vdW}(Br)$  1.85 Å,  $r_{vdW}(I)$  1.98 Å, <sup>21</sup> b Splay angle:  $\Sigma$  of the three bay region angles – 360.

**Table S6** Selected interatomic distances [Å] and angles [°] for Acenap[EPh][E`Ph] compounds **A7-A12** [values in parentheses are for independent s molecules]

Compound	A7	A8	A9	A10	A11	A12
EPh, E`Ph	SPh,SPh	SePh,SePh	TePh,TePh	SePh,SPh	TePh,SPh	TePh,SePh
Peri-region-distances						
E(1)…E(2)	3.274(4) [3.289(4)]	3.1834(10)	3.3674(19)	3.113(4)	3.1576(15)	3.2479(19)
$\Sigma \mathbf{r}_{vdW}$ - $\mathbf{E} \cdots \mathbf{E}^{a}$	0.326 [0.311]	0.6166	0.7526	0.5870	0.7024	0.7121
$\% \Sigma r_{\rm vdW}^{a}$	91 [91]	84	82	84	82	82
E(1)-C(1)	1.812(12) [1.776(11)]	1.928(9)	2.180(17)	1.953(10)	2.138(6)	2.152(10)
E(2)-C(9)	1.782(11) [1.804(12)]	1.932(8)	2.155(17)	1.798(11)	1.781(7)	1.943(11)
Acenaphthene bond lengths	,					
C(1)-C(2)	1.393(16) [1.402(15)]	1.372(12)	1.367(19)	1.393(15)	1.363(8)	1.397(14)
C(2)-C(3)	1.411(17) [1.405(17)]	1.411(12)	1.43(3)	1.424(14)	1.417(9)	1.404(15)
C(3)-C(4)	1.313(18) [1.380(17)]	1.371(12)	1.36(2)	1.350(16)	1.371(9)	1.351(16)
C(4)-C(5)	1.432(16) [1.406(15)]	1.420(12)	1.410(17)	1.404(15)	1.420(8)	1.422(16)
C(5)-C(10)	1.437(17) [1.431(15)]	1.423(12)	1.43(2)	1.450(14)	1.427(8)	1.428(14)
C(5)-C(6)	1.431(16) [1.403(16)]	1.425(12)	1.443(20)	1.423(16)	1.415(8)	1.409(16)
C(6)-C(7)	1.372(18) [1.383(17)]	1.364(12)	1.318(19)	1.367(16)	1.354(9)	1.371(17)
C(7)-C(8)	1.415(18) [1.458(18)]	1.412(12)	1.42(3)	1.406(15)	1.416(9)	1.405(17)
C(8)-C(9)	1.411(17) [1.365(17)]	1.372(12)	1.38(2)	1.361(17)	1.371(9)	1.395(16)
C(9)-C(10)	1.400(16) [1.472(16)]	1.444(12)	1.430(18)	1.451(15)	1.426(9)	1.424(15)
C(10)-C(1)	1.442(17) [1.458(15)]	1.425(12)	1.43(2)	1.434(16)	1.446(8)	1.447(15)
C(4)-C(11)	1.548(16) [1.516(17)]	1.520(12)	1.50(2)	1.522(15)	1.499(9)	1.538(15)
C(11)-C(12)	1.536(17) [1.572(16)]	1.570(12)	1.545(18)	1.568(17)	1.557(9)	1.547(17)
C(12)-C(6)	1.529(16) [1.470(16)]	1.509(12)	1.55(2)	1.525(15)	1.505(9)	1.506(15)
Peri-region bond angles						
E(1)-C(1)-C(10)	124.3(9) [126.3(8)]	123.4(6)	123.9(10)	120.7(7)	121.5(4)	123.1(7)
C(1)-C(10)-C(9)	131.8(11) [129.2(10)]	131.1(8)	131.7(16)	131.3(9)	131.2(5)	130.6(9)
E(2)-C(9)-C(10)	125.1(9) [124.3(9)]	122.0(6)	122.8(12)	120.7(8)	122.3(5)	123.4(8)
$\Sigma$ of bay angles	381.2(24) [379.8(22)]	376.5(13)	378.4(24)	372.7(16)	375.0(10)	377.1(16)
Splay angle <sup><i>b</i></sup>	21.2 (19.8)	16.5	18.4	12.7	15.0	17.1
C(4)-C(5)-C(6)	111.4(10) [111.8(10)]	110.5(7)	113.3(14)	113.4(9)	111.1(5)	112.2(9)
E(2)-E(1)-C(13)	75.14(1) [73.91(1)]	171.28(1)	174.0(4)	167.95(1)	167.86(1)	172.82(1)
E(1)-E(2)-C(19)	85.76(1) [84.55(1)]	91.48(1)	94.1(4)	86.58(1)	91.70(1)	94.13(1)
Out-of-plane displacement						
E(1)	-0.132(1) [0.194(1)]	0.176(1)	0.310(1)	0.274(1)	0.176(1)	-0.403(1)
E(2)	0.161(1) [-0.128(1)]	-0.193(1)	-0.404(1)	-0.416(1)	-0.063(1)	0.032(1)
Central naphthalene ring to	rsion angles					
C:(6)-(5)-(10)-(1)	-176.50(1) [177.94(1)]	178.40(1)	176.19(1)	175.14(1)	179.22(1)	179.41(1)
C:(4)-(5)-(10)-(9)	178.82(1) [-176.17(1)]	179.23(1)	179.03(1)	174.65(1)	176.81(1)	176.12(1)
van der Waals radii used for ca	alculations: r <sub>vdw</sub> (S) 1.80Å, r <sub>vdv</sub>	v(Se) 1.90Å, r <sub>vdW</sub>	(Te) 2.06Å; <sup>21 b</sup> Splay	angle: $\Sigma$ of the th	ree bay region ang	les – 360.

Table S7 Selected interatomic distances [Å] and angles [°] for Acenap[X][X`] (X/X` = I/I, I/Br, ClCl) compounds A14-A16 [compared to known structures Acenap[X][X`] (X/X` = Br/Br, Br/Cl) A13 and A17]

Compound	A14	A15	A16	A13 <sup>22</sup>	A17 <sup>23</sup>
X X`	ΙI	I Br	Cl Cl	Br Br	Br Cl

X(1)…X(2)	3.5505(12)	3.4145(18)	3.1336(11)	3.308(1)	3.287(1)
$\Sigma \mathbf{r}_{\mathrm{vdW}}$ - X···X <sup>·</sup> <i>a</i>	0.4095	0.4155	0.3664	0.392	0.313
$\% \Sigma r_{vdW}^{a}$	90	89	90	89	91
X(1)-C(1)	2.108(6)	2.041(10)	1.750(2)	1.905(1)	1.893(1)
X(2)-C(9)	2.107(6)	2.032(9)	1.754(2)	1.906(1)	1.938(1)
Acenaphthene bond lengths					
C(1)-C(2)	1.371(10)	1.340(19)	1.376(4)	1.381(1)	1.384(1)
C(2)-C(3)	1.389(10)	1.413(18)	1.411(3)	1.407(1)	1.450(1)
C(3)-C(4)	1.386(11)	1.352(17)	1.367(4)	1.364(1)	1.370(1)
C(4)-C(5)	1.405(10)	1.426(14)	1.412(4)	1.415(1)	1.447(1)
C(5)-C(10)	1.427(8)	1.389(12)	1.418(3)	1.425(1)	1.508(1)
C(5)-C(6)	1.419(10)	1.398(16)	1.417(3)	1.415(1)	1.462(1)
C(6)-C(7)	1.359(11)	1.402(16)	1.364(4)	1.366(1)	1.415(1)
C(7)-C(8)	1.411(10)	1.438(14)	1.408(3)	1.411(1)	1.420(1)
C(8)-C(9)	1.397(11)	1.373(16)	1.370(3)	1.377(1)	1.419(1)
C(9)-C(10)	1.420(10)	1.414(15)	1.424(4)	1.432(1)	1.420(1)
C(10)-C(1)	1.443(10)	1.492(15)	1.430(3)	1.427(1)	1.446(1)
C(4)-C(11)	1.508(10)	1.562(16)	1.507(3)	1.511(1)	1.435(1)
C(11)-C(12)	1.559(12)	1.530(19)	1.554(4)	1.549(1)	1.563(1)
C(12)-C(6)	1.503(10)	1.500(14)	1.512(4)	1.509(1)	1.421(1)
Peri-region bond angles					
X(1)-C(1)-C(10)	125.6(5)	124.4(7)	122.5(2)	123.94(1)	119.77(1)
C(1)-C(10)-C(9)	132.7(6)	131.2(8)	132.1(2)	132.65(1)	138.25(1)
X(2)-C(9)-C(10)	126.9(5)	125.9(7)	122.51(17)	124.20(1)	120.10(1)
$\Sigma$ of bay angles	385.2(13)	381.5(18)	377.1(5)	380.79(1)	378.12(3)
Splay angle <sup>b</sup>	25.2	21.5	17.1	20.8	18.1
C(4)-C(5)-C(6)	110.2(6)	110.1(8)	111.3(2)	110.89(1)	109.40(1)
Out-of-plane displacement					
X(1)	0.098(1)	-0.067(1)	-0.067(1)	-0.027(1)	-0.117(1)
X(2)	-0.146(1)	0.098(1)	0.114(1)	0.074(1)	0.053(1)
Central naphthalene ring torsid	on angles				
C:(6)-(5)-(10)-(1)	179.13(1)	179.12(1)	-178.19(1)	-179.29(1)	179.35(1)
C:(4)-(5)-(10)-(9)	179.25(1)	-179.66(1)	-179.12(1)	-178.71(1)	-172.53(1)

#### Table S8: Crystallographic data for compounds A1-A4.

	A1	A2	A3	A4
Empirical Formula	$C_{18}H_{13}BrS$	C <sub>18</sub> H <sub>13</sub> BrSe	C <sub>18</sub> H <sub>13</sub> BrTe	$C_{18}H_{13}IS$
Formula Weight	341.26	388.16	436.80	388.27
Temperature (°C)	-148(1)	-148(1)	-148(1)	-148(1)
Crystal Colour, Habit	colourless, platelet	colourless, prism	yellow, platelet	colourless, platelet
Crystal Dimensions (mm <sup>3</sup> )	0.15 X 0.09 X 0.02	0.20 X 0.20 X 0.20	0.12 X 0.03 X 0.03	0.15 X 0.15 X 0.02
Crystal System	triclinic	triclinic	orthorhombic	monoclinic
Lattice Parameters	a = 7.400(5) Å	a = 5.645(3) Å	a = 5.8486(17) Å	a = 5.109(4)Å
	b = 9.253(6) Å	b = 11.096(5) Å	b = 21.983(7) Å	b = 15.850(12)Å
	c = 10.744(6)  Å	c = 11.508(5)  Å	c = 22.637(6) Å	c = 17.810(14)  Å
	$\alpha = 79.67(4)^{\circ}$	$\alpha = 97.849(11)^{\circ}$	-	-
	$\beta = 77.28(4)^{\circ}$	$\beta = 95.044(11)^{\circ}$	-	$\beta = 93.721(19)^{\circ}$
	$\gamma = 87.54(5)^{\circ}$	$\gamma = 97.960(13)^{\circ}$	-	-
Volume (Å <sup>3</sup> )	V = 705.9(8)	V = 703.0(6)	V = 2910.4(14)	V = 1439.0(19)
Space Group	P-1	P-1	Pbca	P2 <sub>1</sub> /c

Z value	2	2	8	4
Dcalc (g/cm <sup>3</sup> )	1.605	1.834	1.994	1.792
F000	344	380	1664	760
$\mu$ (MoK $\alpha$ ) (cm <sup>-1</sup> )	30.531	55.071	47.845	23.571
No. of Reflections Measured	5704	6071	21021	11043
Rint	0.0574	0.0298	0.0744	0.1431
Min and Max Transmissions	0.464 - 0.941	0.220 - 0.332	0.525 - 0.866	0.192 - 0.954
Independ. Reflection (No. Variables )	2790(181)	3040(181)	2908(181)	3262(181)
Reflection/Parameter Ratio	15.41	16.80	16.07	18.02
Residuals: $R_1$ (I>2.00 $\sigma$ (I))	0.0716	0.0456	0.0703	0.1043
Residuals: R (All reflect ons)	0.0903	0.0579	0.1047	0.1361
Residuals: wR <sub>2</sub> (All reflections)	0.2237	0.1876	0.2862	0.3414
Goodness of Fit Indicator	1.181	1.197	1.284	1.141
Flack Parameter	-	-	-	-
Maximum peak in Final Diff. Map	0.58 e-/Å <sup>3</sup>	1.46 e-/Å <sup>3</sup>	2.42 e-/Å <sup>3</sup>	1.88 e-/Å <sup>3</sup>
Minimum peak in Final Diff. Map	$-0.82 \text{ e}-/\text{Å}^3$	-1.53 e-/Å <sup>3</sup>	-3.79 e-/Å <sup>3</sup>	-3.17 e-/Å <sup>3</sup>

#### Table S9: Crystallographic data for compounds A5-A6.

	A5	A6	A7	A8
Empirical Formula	C <sub>18</sub> H <sub>13</sub> ISe	C <sub>18</sub> H <sub>13</sub> ITe	$C_{24}H_{18}S_2$	$C_{24}H_{18}Se_2$
Formula Weight	435.17	483.81	370.53	464.33
Temperature (°C)	-148(1)	-148(1)	-148(1)	-180(1)
Crystal Colour, Habit	colourless, prism	yellow, plate	colourless, platelet	colourless, platelet
Crystal Dimensions (mm <sup>3</sup> )	0.09 X 0.09 X 0.03	0.27 X 0.03 X 0.02	0.18 X 0.09 X 0.03	0.20 X 0.10 X 0.01
Crystal System	monoclinic	monoclinic	monoclinic	orthorhombic
Lattice Parameters	a = 7.3312(19) Å	a = 5.9391(18) Å	a = 11.794(6) Å	a = 22.1470(15) Å
	b = 10.845(3) Å	b = 22.301(6) Å	b = 10.806(6) Å	b = 5.221(4)  Å
	c = 18.220(5)  Å	c = 22.556(6) Å	c = 28.856(17)  Å	c = 15.829(6) Å
	-	-	-	-
	$\beta = 95.673(8)^{\circ}$	$\beta = 90.069(9)^{\circ}$	$\beta = 93.435(15)^{\circ}$	-
	-	-	-	-
Volume (Å <sup>3</sup> )	V = 1441.5(6)	V = 2987.5(15)	V = 3671(4)	V = 1830.3(15)
Space Group	$P2_1/c$	P21/c	Cc	Pca2 <sub>1</sub>
Z value	4	8	8	4
Dcalc (g/cm <sup>3</sup> )	2.005	2.151	1.341	1.685
F000	832	1808	1552	920
$\mu(MoK\alpha)$ (cm <sup>-1</sup> )	47.340	40.460	2.943	40.454
No. of Reflections Measured	11619	25496	13181	10530
Rint	0.0665	0.0779	0.1058	0.0591
Min and Max Transmissions	0.518 - 0.868	0.509 - 0.922	0.364 - 0.991	0.656 - 0.960
Independ. Reflection (No. Vari Deles )	2926(181)	6960(361)	6408(469)	3200(235)
Reflection/Parameter Ratio	16.17	19.28	13.66	13.62
Residuals: $R_1$ (I>2.00 $\sigma$ (I))	0.0511	0.0953	0.1115	0.0406
Residuals: R (All reflections)	0.0694	0.1276	0.1540	0.0943
Residuals: wR <sub>2</sub> (All reflections)	0.1582	0.3271	0.3075	0.1333
Goodness of Fit Indicator	1.264	1.260	1.063	0.765
Flack Parameter	-	-	-	0.011(18)
Maximum peak in Final Diff. Map	1.70 e-/Å <sup>3</sup>	2.79 e-/Å <sup>3</sup>	0.80 e-/Å <sup>3</sup>	0.59 e-/Å <sup>3</sup>
Minimum peak in Final Diff. Map	-2.00 e-/Å <sup>3</sup>	-3.55 e-/Å <sup>3</sup>	-0.74 e-/Å <sup>3</sup>	-0.67 e-/Å <sup>3</sup>

5

#### Table S10: Crystallographic data for compounds A9-A12.

	A9	A10	A11	A12
Empirical Formula	$C_{24}H_{18}Te_2$	$C_{24}H_{18}SSe$	$C_{24}H_{18}STe$	$C_{24}H_{18}$ SeTe
Formula Weight	561.61	417.43	466.07	512.97

Temperature (°C)	-148(1)	-148(1)	-180(1)	-148(1)
Crystal Colour, Habit	yellow, platelet	colourless, platelet	colourless, prism	colourless, prism
Crystal Dimensions (mm <sup>3</sup> )	0.12 X 0.09 X 0.03	0.24 X 0.02 X 0.03	0.12 X 0.03 X 0.03	0.20 X 0.20 X 0.20
Crystal System	monoclinic	monoclinic	orthorhombic	monoclinic
Lattice Parameters	a = 10.050(6)Å	a = 5.606(4)  Å	a = 22.326(3) Å	a = 5.641(2) Å
	b = 21.750(11) Å	b = 16.173(9) Å	b = 5.272(10) Å	b = 22.249(10) Å
	c = 9.843(5)  Å	c = 20.919(13) Å	c = 15.640(14)  Å	c = 15.272(7)  Å
	-	-	-	-
	$\beta = 116.242(12)^{\circ}$	$\beta = 97.126(14)^{\circ}$	-	$\beta = 100.477(18)^{\circ}$
	-	-	-	-
Volume (Å <sup>3</sup> )	V = 1929.8(19)	V = 1881.8(19)	V = 1841(4)	V = 1884.7(14)
Space Group	$P2_1/c$	$P2_1/n$	Pca2 <sub>1</sub>	$P2_1/n$
Z value	4	4	4	4
Dcalc (g/cm <sup>3</sup> )	1.933	1.473	1.681	1.808
F000	1064	848	920	992
$\mu(MoK\alpha)$ (cm <sup>-1</sup> )	30.281	21.108	17.338	35.146
No. of Reflections Measured	15461	16208	17293	16286
Rint	0.1036	0.1434	0.0682	0.1052
Min and Max Transmissions	0.298 - 0.913	0.484 - 0.959	0.558 - 0.949	0.299 - 0.495
Independ.Reflection (No. Variables )	4448(235)	4381 (235)	3322 (235)	4318(235)
Reflection/Parameter Ratio	18.93	18.64	14.14	18.37
Residuals: $R_1$ (I>2.00 $\sigma$ (I))	0.0948	0.1324	0.0409	0.0835
Residuals: R (All reflections)	0.1393	0.1948	0.0430	0.1251
Residuals: wR <sub>2</sub> (All reflections)	0.3803	0.3364	0.0985	0.2644
Goodness of Fit Indicator	1.421	1.218	1.084	1.148
Flack Parameter	-	-	-0.01(4)	-
Maximum peak in Final Diff. Map	2.90 e-/Å <sup>3</sup>	0.73 e-/Å <sup>3</sup>	1.58 e-/Å <sup>3</sup>	1.38 e-/Å <sup>3</sup>
Minimum peak in Final Diff. Map	-4.99 e-/Å <sup>3</sup>	-1.16 e-/Å <sup>3</sup>	-1.39 e-/Å <sup>3</sup>	-2.35e-/Å <sup>3</sup>

#### Table S11: Crystallographic data for compounds A14-A16.

	A14	A15	A16
Empirical Formula	$C_{12}H_8I_2$	$C_{12}H_8BrI$	$C_{12}H_8Cl_2$
Formula Weight	406.00	359.00	223.10
Temperature (°C)	-148(1)	-148(1)	-148(1)
Crystal Colour, Habit	colourless, prism	colourless, platelet	colourless, prism
Crystal Dimensions (mm <sup>3</sup> )	0.20 X 0.20 X 0.20	0.15 X 0.12 X 0.03	0.37 X 0.12 X 0.10
Crystal System	monoclinic	monoclinic	triclinic
Lattice Parameters	a = 8.062(3)Å	a = 7.912(3) Å	a = 7.6529(7)  Å
	b = 11.733(4) Å	b = 11.715(5) Å	b = 8.2912(8) Å
	c = 11.966(4)  Å	c = 11.771(5)  Å	c = 8.8382(8)  Å
	-	-	$\alpha = 64.174(2)^{\circ}$
	$\beta = 107.480(8)^{\circ}$	$\beta = 107.423(9)^{\circ}$	$\beta = 78.781(2)^{\circ}$
	-	-	$\gamma = 69.709(2)^{\circ}$
Volume (Å <sup>3</sup> )	V = 1079.6(6)	V = 1041.0(8)	V = 472.83(8)
Space Group	$P2_1/n$	$P2_1/n$	P-1
Z value	4	4	2
Dcalc (g/cm <sup>3</sup> )	2.498	2.290	1.567
F000	744	672	228

$\mu(MoK\alpha)$ (cm <sup>-1</sup> )	57.831	68.804	6.330
No. of Reflections Measured	8943	8650	4081
Rint	0.0337	0.0382	0.0256
Min and Max Transmissions	0.235 - 0.315	0.492 - 0.813	0.821 - 0.939
Independ.Reflection (No. Variables )	2471(127)	2412(127)	1663(127)
Reflection/Parameter Ratio	19.46	18.99	13.09
Residuals: $R_1$ (I>2.00 $\sigma$ (I))	0.0421	0.1031	0.0384
Residuals: R (All reflections)	0.0549	0.1121	0.0506
Residuals: wR <sub>2</sub> (All reflections)	0.1786	0.3718	0.0853
Goodness of Fit Indicator	1.201	1.735	1.168
Flack Parameter	-	-	-
Maximum peak in Final Diff. Map	3.04 e-/Å <sup>3</sup>	8.24 e-/Å <sup>3</sup>	0.27 e-/Å <sup>3</sup>
Minimum peak in Final Diff. Map	-2.94 e-/Å <sup>3</sup>	-3.49 e-/Å <sup>3</sup>	-0.27 e-/Å <sup>3</sup>



5 Fig. S9 The relationship between heteroatom size (sum of van der Waals radii) versus *peri*-distance in compounds adopting type A/AA conformations and bis-halides.



Fig. S10 The relationship between heteroatom size (sum of van der Waals radii) versus peri-distance in compounds adopting type B/AB conformations.

#### 4. References

- <sup>1</sup> N. Tanaka and T. Kasai, Bull. Chem. Soc. Jpn., 1981, 54, 3020-3025.
- <sup>2</sup> Y. Aso, K. Yui, T. Miyoshi and T. Otsubo, *Bull. Chem. Soc. Jpn.*, 1988, **61**, 2013-2018.
- <sup>3</sup> W. D. Neudorff, D. Lentz, M. Anibarro and A. D. Schlüter, Chem. Eur. J., 2003, 9, 2745-2757.
- <sup>4</sup> A. D. Becke, J. Chem. Phys., 1993, 98, 5648-5652; C. Lee, W. Yang and R. G. Parr, Phys. Rev. B, 1988, 37, 785-789.
- <sup>5</sup> R. C. Binning and L. A. Curtiss, J. Comput. Chem., 1990, 11, 1206-1216.
- <sup>6</sup> P. Schwerdtfeger, M. Dolg, W. H. E. Schwarz, G. A. Bowmaker and P. D. W. Boyd, *J. Chem. Phys.*, 1989, **91**, 1762-1774; A. Bergner, M. Dolg, W. Kuechle, H. Stoll and H. Preuss, *Mol. Phys.*, 1993, **80**, 1431-1441; for I the valence basis by Kaupp et al. (including diffuse functions) was employed: M. Kaupp, P. v. R. Schleyer, H. Stoll and H. Preuss, *J. Am. Chem. Soc.*, 1991, **113**, 6012-6020.
- <sup>7</sup> S. Huzinaga, J. Anzelm, M. Klobukowski, E. Radzio-Andzelm, Y. Sakai and H. Tatewaki, in: *Gaussian Basis Sets for Molecular Calculations*, Elsevier, Amsterdam, 1984.
- <sup>8</sup> K. B. Wiberg, *Tetrahedron*, 1968, **24**, 1083-1096.
- <sup>9</sup> A. E. Reed, F. Curtiss and L. A. F. Weinhold, Chem. Rev., 1988, 88, 899-926.
- <sup>10</sup> M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J.Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez and J. A. Pople, Gaussian 03, Revision E.01, Gaussian, Inc., Wallingford CT, 2004.
- <sup>11</sup> J. Autschbach, T. Ziegler, J. Chem. Phys. 2000, 113, 936; J. Autschbach, J. Chem. Phys. 2008, 129, 094105.
- <sup>12</sup> A. D. Becke, *Phys. Rev. A* 1988, 38, 3098-3100; J. P. Perdew, *Phys. Rev. B* 1986, 33, 8822; J. P. Perdew, *Phys. Rev. B* 1986, 34, 7406.
- <sup>13</sup> E. van Lenthe, E. J. Baerends, J. G. Snijders, J. Chem. Phys. **1994**, 101, 9783; E. van Lenthe, R. van Leeuwen, E. J. Baerends, J. G. Snijders, Int. J.
- Quantum Chem. 1996, 57, 281-293; E. van Lenthe, E. J. Baerends, J. G. Snijders, J. Chem. Phys. 1993, 99, 4597.
- <sup>14</sup> E. van Lenthe, J. G. Snijders, E. J. Baerends, J. Chem. Phys. **1996**, 105, 6505.
- <sup>15</sup> E. J. Baerends, D. E. Ellis, P. Ros, Chem. Phys. 1973, 2, 41; G. te Velde, E. J. Baerends, J. Comput. Phys. 1992, 99, 84; G. te Velde, F. M. Bickelhaupt, E. J.Baerends, C. Fonseca Guerra, S. J. A. van Gisbergen, J. G. Snijders, T. Ziegler, J. Comput. Chem. 2001, 22, 931
- <sup>16</sup> E.J. Baerends, T. Ziegler, J. Autschbach, D. Bashford, A. Bérces, F.M. Bickelhaupt, C. Bo, P.M. Boerrigter, L. Cavallo, D.P. Chong, L. Deng, R.M. Dickson, D.E. Ellis, M. van Faassen, L. Fan, T.H. Fischer, C. Fonseca Guerra, A. Ghysels, A. Giammona, S.J.A. van Gisbergen, A.W. Götz, J.A. Groeneveld, O.V. Gritsenko, M. Grüning, S. Gusarov, F.E. Harris, P. van den Hoek, C.R. Jacob, H. Jacobsen, L. Jensen, J.W. Kaminski, G. van Kessel, F. Kootstra, A. Kovalenko, M.V. Krykunov, E. van Lenthe, D.A. McCormack, A. Michalak, M. Mitoraj, J. Neugebauer, V.P. Nicu, L. Noodleman, V.P. Osinga, S. Patchkovskii, P.H.T. Philipsen, D. Post, C.C. Pye, W. Ravenek, J.I. Rodríguez, P. Ros, P.R.T. Schipper, G. Schreckenbach, J.S. Seldenthuis, M. Seth, J.G. Snijders, M. Solà, M. Swart, D. Swerhone, G. te Velde, P. Vernooijs, L. Versluis, L. Visscher, O. Visser, F. Wang, T.A. Wesolowski, E.M. van Wezenbeek, G. Wiesenekker, S.K. Wolff, T.K. Woo, A.L. Yakovlev, ADF2010.02, SCM, Theoretical Chemistry, Vrije Universiteit, Amsterdam, The Netherlands, http://www.scm.com.

- <sup>17</sup> A. L Fuller, L. A. S. Scott-Hayward, Y. Li, M. Bühl, A. M. Z. Slawin and J. D. Woollins, *J. Am. Chem. Soc.*, 2010, 132, 5799–5802.
   <sup>18</sup> CrystalClear 2.0: Rigaku Corporation, 2010. CrystalClear Software User's Guide, Molecular Structure Corporation, © 2000. J. W. P. Flugrath, *Acta Crystallogr., Sect. D*, 1999, D55, 1718-1725.
- <sup>19</sup> CrystalStructure 4.0: Crystal Structure Analysis Package, Rigaku and Rigaku/MSC (2009). 9009 New Trails Dr., The Woodlands, TX 77381, USA.
   <sup>20</sup> SHELX97: G. M. Sheldrick, *Acta Crystallogr., Sect. A*, 2008, 64, 112.
- <sup>21</sup> A. Bondi, J. Phys. Chem., 1964, **68**, 441-451.
- <sup>22</sup> N. Jaiboon, *J. Sci. Soc. Thailand*, 2002, **28**, 271-275.
   <sup>23</sup> R. L. Avoyan, and Yu. T. Struchkov, *Zh. Strukt. Khim.*, 1964, **5**, 407-419.