

Supporting Information for Preparation and Physical Properties of Early-Late Heterobimetallic Compounds Featuring Ir–M Bonds (M = Ti, Zr, Hf)

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Contents:

Experimental Details	S3
General Considerations	S3
General Remarks on the Determination of Molecular Structure by X-ray Diffraction	S3
Preparation of Cp*Ir(μ-N ^t Bu)TiCp ₂ (2-Ti)	S3
UV-vis data for Cp*Ir(μ-N ^t Bu)ZrCp ₂ (2-Zr)	S3
Preparation of Cp*Ir(μ-N ^t Bu)HfCp ₂ (2-Hf)	S3
Preparation of [Cp ₂ Fe][OTf]	S4
Preparation of 2,6-lutidinium triflate, [Lut-H][OTf]	S4
Preparation of Cp*Ir(μ-N ^t Bu)(μ-H)ZrCp ₂ OTf (3-Zr)	S4
Alternative Preparation of Cp*Ir(μ-N ^t Bu)(μ-H)ZrCp ₂ OTf (3-Zr) from Schwartz's Reagent	S4
Preparation of Cp*Ir(μ-N ^t Bu)(μ-H)HfCp ₂ OTf (3-Hf)	S4
General procedure for the preparation of Cp*Ir(μ-N ^t Bu)(μ-N ₃ C ₆ H ₄ R)HfCp ₂	S4
Preparation of Cp*Ir(μ-N ^t Bu)(μ-NPh)HfCp ₂ (5-Hf)	S5
Formation of Cp*Ir(μ-N ^t Bu)(μ-NC ₆ H ₄ R)HfCp ₂	S5
Kinetics Measurements for 4-Hf → 5-Hf	S5
Preparation of Cp*Ir(μ-N ^t Bu)(μ-S)HfCp ₂ (6-Hf)	S6
General Remarks Regarding Computational Methods	S6
DFT Vibrational Frequency Analysis for 2-M	S6
TDDFT Calculations for 2-M	S6
DFT Calculated ¹ H NMR shifts for 3-Zr	S6
DFT Vibrational Frequency Analysis for 3-Zr	S6
Figure S1. ATR-IR Spectra of Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S7
Table S1. ATR-IR Spectral Data for Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S8
Figure S2. Solution-IR Spectra of Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S9
Table S2. Solution-IR Spectral Data for Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S10
Figure S3. Raman Spectra of Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S11
Figure S4. Cyclic Voltammetry Data for Cp*Ir(μ-N ^t Bu)MCp ₂ (2-M)	S12
Figure S5. ATR-IR Spectra of Cp*Ir(μ-N ^t Bu)(μ-H)MCp ₂ OTf (3-M)	S13
Table S3. ATR-IR Spectral Data for Cp*Ir(μ-N ^t Bu)(μ-H)MCp ₂ OTf (3-M)	S14
Figure S6. ¹ H NMR Integral Intensities for 4-Hf → 5-Hf	S15
Figure S7. Determination of <i>k</i> from <i>a</i> for 4-Hf → 5-Hf	S15
Figure S8. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)TiCp ₂ (2-Ti)	S16
Figure S9. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)HfCp ₂ (2-Hf)	S16
Figure S10. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)(μ-H)ZrCp ₂ OTf (3-Zr)	S17
Figure S11. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)(μ-H)ZrCp ₂ OTf (3-Hf)	S18
Figure S12. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)(μ-N ₃ Ph)HfCp ₂ (4-Hf)	S19
Figure S13. ¹ H NMR Spectrum of Cp*Ir(μ-N ^t Bu)(μ-NPh)HfCp ₂ (5-Hf)	S19

Figure S14. ^1H NMR Spectrum of $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-S})\text{HfCp}_2$ (6-Hf)	S20
Figure S15. Orbital Contour Plots for 2-Ti , 2-Zr , and 2-Hf	S21
Figure S16. Frontier Orbital Energies for 2-Ti , 2-Zr , and 2-Hf	S22
Table S4. Optimized Geometry for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Ti)	S23
Table S5. Optimized Geometry for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{ZrCp}_2$ (2-Zr)	S24
Table S6. Optimized Geometry for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{HfCp}_2$ (2-Hf)	S25
Table S7. Optimized Geometry for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-H})\text{ZrCp}_2\text{OTf}$ (3-Zr)	S26
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Ti) using B3LYP	S28
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Ti) using PW91PW91	S35
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Zr) using B3LYP	S43
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Zr) using PW91PW91	S48
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Hf) using B3LYP	S51
Selected TDDFT results for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{TiCp}_2$ (2-Hf) using PW91PW91	S54
References	S57

Experimental Details

General Considerations

All manipulations were carried out under an atmosphere of purified nitrogen in a Vacuum Atmospheres Nexus glovebox, or by standard Schlenk techniques.^{1,2} Inside the glovebox the ambient temperature ranged from 18–22 °C. All glassware was oven-dried at a temperature above 100 °C for at least 3 hours and allowed to cool under dynamic vacuum prior to use. Celite, alumina, and 4 Å sieves were activated by heating to a temperature greater than 180 °C under a dynamic vacuum for 2 d (celite) or 5 d (alumina and 4 Å sieves). Et₂O, *n*-hexane, *n*-pentane, CH₂Cl₂, THF, benzene, and toluene were bubble degassed with nitrogen and forced, under positive pressure, through a column of activated alumina.³ All solvents were stored over 4 Å sieves. C₆D₆ was degassed under a dynamic vacuum prior to use. CD₂Cl₂ was refluxed over CaH₂ for 24 h then distilled. *d*₈-THF was distilled from dark purple solutions of sodium benzophenone. ¹H and ¹³C NMR shifts are referenced against residual solvent resonances (for C₆D₆, 7.16 and 128.4; for *d*₈-THF, 3.58 and 67.6; for CD₂Cl₂, 5.32 and 54.0 ppm). UV-vis spectra were obtained on a Cary spectrophotometer in 1-cm quartz cells. IR spectroscopy was carried out using a Thermo Scientific iS10 FT-IR spectrometer equipped with either a Smart iTR diamond, ATR accessory, or a solution cell with KBr windows. Raman spectra were acquired using a LabRamHR epillumination confocal Raman microscope, manufactured by Horiba Jobin Yvon, in conjunction with a 633 nm, He-Ne laser (1-2 mW) and a CCD detector manufactured by Andor. NMR data were acquired using 600, 500, 400, or 300 MHz spectrometers manufactured by Bruker. Positive-ion ESI-MS were acquired by the QB3 mass spectrometry facility at the University of California, Berkeley using an LTQ FT mass spectrometer, manufactured by Thermo Fisher Scientific, equipped with an electrospray ionization source and a 7 T magnet. Cyclic voltammetry was performed using an EC Epsilon potentiostat manufactured by BASi in conjunction with a glassy carbon working electrode, Pt-wire counter electrode, and Ag-wire pseudo-reference electrode. Combustion analysis was performed by the College of Chemistry Microanalytical Facility at the University of California, Berkeley, or Midwest Microlabs, LLC, Indianapolis, IN. Unless otherwise noted, Teflon-coated magnetic stir bars and magnetic stir plates were used in all reactions. Literature procedures were used for the preparation of Cp*IrN≡Bu,^{4,5} 2-Zr,^{6,7} Cp₂Hf(^{*n*}Bu)₂,⁸ Cp₂Ti[η²-C₂(SiMe₃)₂],⁹ and aromatic azides.¹⁰

General Remarks on the Determination of Molecular Structure by X-ray Diffraction

X-ray diffraction data were collected using either Bruker AXS three-circle or Bruker AXS Microstar kappa-geometry diffractometers coupled to a CCD detector with graphite-monochromated Mo K α radiation ($\lambda = 0.71073$ Å). The structures were solved by direct methods using SHELXS and refined against F^2 on all data by full-matrix least squares with SHELXL-97.¹¹ All non-hydrogen atoms were refined

anisotropically; all hydrogen atoms were included into the model at their geometrically calculated positions and refined using a riding model. Additional details concerning the data collection, data work-up, and crystallographic solutions for compounds 2-Ti, 2-Hf, 3-Zr, 5-Hf are contained in the cif files that are available as electronic supplementary material.

A disordered cyclopentadienyl ligand in the structure of 3-Zr was refined with the help of similarity restraints on 1-2 and 1-3 distances and displacement parameters as well as rigid bond restraints for anisotropic displacement parameters. The ratios were refined freely, while constraining the total occupancy of all components to unity. For the purpose of calculating bond distances and angles between atoms involved in this disorder, only the major component of the disorder was considered.

Preparation of Cp*Ir(μ -N^{*n*}Bu)TiCp₂ (2-Ti)

In the glovebox, a solution of Cp₂Ti(η²-C₂(SiMe₃)₂) (262 mg, 0.752 mmol, 1.2 equiv) in 10 mL of *n*-hexane was added to solid 1 (250 mg, 0.627 mmol, 1.0 equiv) at 20 °C. The resulting mixture was stirred at 20 °C for 14 h before the solvent was removed by application of a dynamic vacuum. To completely remove the C₂(SiMe₃)₂ byproduct, it was necessary to resuspend the mixture in 10 mL of *n*-hexane and subsequently remove solvent from the mixture using a dynamic vacuum, for a second time. The resulting brown powder was washed with *n*-pentane (3 mL \times 4) that had been cooled to -35 °C to remove the excess Cp₂Ti(η²-C₂(SiMe₃)₂). This afforded Cp*Ir(μ -N^{*n*}Bu)TiCp₂ (180 mg, 0.312 mmol, 49.8%) as a bright orange powder. ¹H NMR (400 MHz, C₆D₆, 20 °C): $\delta = 6.42$ (s, 10 H, Cp), 1.75 (s, 15 H, Cp*), 0.78 (s, 9 H, Cp) ppm. ¹³C NMR (100 MHz, C₆D₆, 20 °C): $\delta = 114.2$ (Cp), 82.4 (C₅-Cp*), 73.5 (CMe₃), 32.1 (C(CH₃)₃), 11.0 (Me₅-Cp*) ppm. UV-vis (Et₂O, 20 °C): λ (ϵ) = 272 (25400), 295 (23200), 373 (4760) sh nm (M⁻¹cm⁻¹). EA calcd for C₂₄H₃₄IrNTi: C, 49.99; H, 5.94; N, 2.43; Found: C, 50.08; H, 5.98; N, 2.29%. Crystals suitable for single-crystal X-ray diffraction were grown at -35 °C from an *n*-pentane/Et₂O solution.

UV-vis data for Cp*Ir(μ -N^{*n*}Bu)ZrCp₂ (2-Zr)

UV-vis (Et₂O, 20 °C): λ (ϵ) = 296 (11000), 395 (4200) nm (M⁻¹cm⁻¹). The preparation and characterization of this compound has been previously reported.^{6,7}

Preparation of Cp*Ir(μ -N^{*n*}Bu)HfCp₂ (2-Hf)

As powders, Cp₂Hf(^{*n*}Bu)₂ (530 mg, 1.25 mmol, 1.0 equiv) and 1 (500 mg, 1.25 mmol, 1.0 equiv) were combined in a 50 mL tube. To the combined solids was added 5 mL of C₆H₆, and the resulting mixture was freeze-pump-thaw degassed three times. This mixture was then heated to 105 °C for 3.5 h without stirring. Then, the mixture was frozen, and the solvent removed by pumping on the frozen solution at 20 °C. This procedure gives Cp*Ir(μ -N^{*n*}Bu)HfCp₂ (730 mg, 1.03 mmol, 82.6%) in greater than 90% purity. Recrystallization of the mixture from Et₂O/*n*-pentane mixtures at -35 °C was inefficient and reduced the yield to (220 mg, 0.310 mmol, 25%). ¹H NMR (500 MHz, C₆D₆, 20 °C): 6.08 (s, 10 H, Cp), 1.91 (s, 15 H, Cp*), 1.19 (s, 9 H, ^{*n*}Bu) ppm. ¹³C NMR (100 MHz, C₆D₆, 20 °C): $\delta = 110.0$ (Cp), 87.8 (C₅-Cp*), 73.9 (CMe₃), 33.7 (C(CH₃)₃), 11.47 (Me₅-Cp*) ppm. UV-vis

(Et₂O, 20 °C): λ (ϵ) = 293 (8500) sh, 373(4500) nm (M⁻¹cm⁻¹). EA calcd for C₂₄H₃₄HfIrN: C, 40.76; H, 4.85; N, 1.98; Found: C, 40.60; H, 4.57; N, 2.21%. Crystals suitable for single-crystal X-ray diffraction were grown at -35 °C from a Et₂O/THF solution.

Preparation of [Cp₂Fe][OTf]¹²⁻¹⁴

AgOTf (3 g, 11.6 mmol, 1 equiv) and Cp₂Fe (2.3 g, 12.4 mmol, 1.07 equiv) were combined as solids and then cooled to -35 °C. Then 40 mL of cold (-35 °C) THF was added to the combined solids forming a brown mixture. The mixture was stirred for 25 min, during which time it became dark blue in color. The mixture was filtered through a Celite pad, and the pad was washed with THF (ca. 100 mL) until the washings were colorless. The solvent was then removed using a dynamic vacuum to produce blue powder. This powder was washed with *n*-pentane (50 mL × 2) and Et₂O (20 mL × 3) to remove the excess Cp₂Fe. The washed solids were stored under a dynamic vacuum to remove traces of solvent. The resulting blue powder was collected (3.64 g, 10.8 mmol, 94%). ¹H NMR (500 MHz, CD₂Cl₂, 20 °C): δ = 33.27 (br s, $\Delta\nu_{1/2}$ = 720 Hz, 10 H, Cp) ppm. ¹⁹F NMR (376 MHz, CD₂Cl₂, 20 °C): δ = -89.72 ppm. EA calcd for C₁₁H₁₀F₃FeO₃S: C, 39.43; H, 3.01; Found: C, 39.24; H, 2.93%. ESI-MS(+): *m/z* = 186.0125 (186.0126, Cp₂Fe⁺).

Preparation of 2,6-lutidinium triflate, [Lut-H][OTf]^{15,16}

At 20 °C, to a solution of 2,6-lutidine (2.71 g, 25 mmol, 1 equiv) in 25 mL of *n*-pentane was added triflic acid (3.84 g, 25 mmol, 1 equiv) dropwise. Following the addition, the mixture was allowed to stir for 10 min before the white precipitate was collected by filtration. The so obtained solids were washed with Et₂O (10 mL × 3), and stripped of the remaining solvent under a dynamic vacuum to yield the product (4.5 g, 17.5 mmol, 70%). ¹H NMR (500 MHz, CD₂Cl₂, 20 °C): δ = 14.45 (s, 1 H, *NH*), 8.24 (t, *J*_{HH} = 8 Hz, 1 H, para-H), 7.57 (d, *J*_{HH} = 8 Hz, 9 H, meta-H), -8.09 (s, 1 H, μ -H) ppm. ¹³C NMR (125 MHz, CD₂Cl₂, 20 °C): δ = 154.4 (2,6-C), 146.6 (4-C), 125.5 (3,5-C), 121.0 (q, *J*_{CF} = 320 Hz, CF₃) 19.9 (Me) ppm. ¹⁹F NMR (376 MHz, CD₂Cl₂, 20 °C): δ = -79.4 ppm. EA calcd for C₈H₁₀F₃NO₃S: C, 37.35; H, 3.92; N, 5.45; Found: C, 37.35; H, 3.91; N, 5.45%. ESI-MS(+): *m/z* = 108.08080 (108.08078, C₇H₁₀N⁺).

Preparation of Cp*Ir(μ -N^tBu)(μ -H)ZrCp₂OTf (3-Zr)

At -35 °C, a solution of 2,6-lutidinium triflate (48 mg, 0.187 mmol, 0.95 equiv) in 5 mL THF was added dropwise to a solution of Cp*Ir(μ -N^tBu)ZrCp₂ (122 mg, 0.196 mmol, 1 equiv) in 10 mL of THF. During the addition a change in the color of the solution from red to orange was observed. The mixture was stirred for 15 min before the solvent was removed by the application of a dynamic vacuum. The mixture was then stored under a dynamic vacuum for 14 h to evaporate 2,6-lutidine formed during the reaction. The remaining material was washed with *n*-pentane (5 mL × 3). The product was then obtained via crystallization from THF layered with a mixture of 1:1 Et₂O/*n*-pentane at -35 °C as two crops of orange crystals (121 mg, 0.157 mmol, 84%). ¹H NMR (400 MHz, *d*₈-THF, 20 °C): 6.08 (s, 10 H, Cp), 1.99 (s, 15 H, Cp*), 1.36 (s, 9 H, ^tBu), -8.08 (s, 1 H, μ -H) ppm. ¹³C NMR (100 MHz, *d*₈-THF, 20 °C): δ = 111.2 (Cp), 90.0

(C₅-Cp*), 80.3 (CMe₃), 31.9 (C(CH₃)₃), 10.51 (Me₅-Cp*) ppm. ¹⁹F NMR (376 MHz, *d*₈-THF, 20 °C): δ = -79.2 ppm. ¹H NMR (400 MHz, CD₂Cl₂, 20 °C): δ = 6.06 (s, 10 H, Cp), 1.95 (s, 15 H, Cp*), 1.32 (s, 9 H, ^tBu), -8.09 (s, 1 H, μ -H) ppm. ¹⁹F NMR (376 MHz, CD₂Cl₂, 20 °C): δ = -78.7 ppm. EA calcd for C₂₅H₃₄F₃IrNO₃SZr: C, 39.04; H, 4.46; N, 1.82; Found: C, 38.32; H, 4.25; N, 1.84%. ESI-MS(+): *m/z* = 620.1442 (620.1440, M⁺-OTf).

Alternative Preparation of Cp*Ir(μ -N^tBu)(μ -H)ZrCp₂OTf (3-Zr) from Schwartz's Reagent

A solution of Cp*IrN^tBu (100 mg, 0.251 mmol, 1equiv) in 10 mL of CH₂Cl₂ and added to solid Cp₂Zr(H)Cl (75 mg, 0.256 mmol, 1equiv) at 20 °C. This produced a cloudy suspension that was stirred for 5 min before a solution of Me₃SiOTf (56 mg, 0.252 mmol, 1 equiv) in 3 mL CH₂Cl₂ was added. Upon addition of the final reagent, the mixture immediately clarified to form an orange solution. This solution was stirred an additional 10 min before it was filtered through celite and concentrated to ca. 2 mL. The solution was layered with ca. 10 mL of *n*-pentane and stored at -35 °C to precipitate an orange powder which was collected by filtration. The so-obtained solids were washed with *n*-pentane (5 mL × 3) at 20 °C to afford the product. 124 mg, 0.161, 64%

Preparation of Cp*Ir(μ -N^tBu)(μ -H)HfCp₂OTf (3-Hf)

At -35 °C, a solution of 2,6-lutidinium triflate (14 mg, 0.054 mmol, 0.98 equiv) in 2 mL of THF was added dropwise to a solution of Cp*Ir(μ -N^tBu)HfCp₂ (39 mg, 0.055 mmol, 1 equiv) in 5 mL of THF. Upon addition of 2,6-lutidinium triflate, the initially red solution became yellow in color. The reaction mixture was stirred for 15 min before the solvent was removed by application of a dynamic vacuum. The so-obtained solids were stored under a dynamic vacuum for 6 h to evaporate the 2,6-lutidine formed in the reaction. The remaining material was washed with *n*-pentane (10 mL × 3) and dissolved in a minimum of THF (ca. 2 mL). The THF solution was layered with 10 mL of 1:1 Et₂O/*n*-pentane and stored at -35 °C to afford the product as yellow crystals (34mg, 0.039 mmol, 72%). ¹H NMR (400 MHz, *d*₈-THF, 20 °C): 6.05 (s, 10 H, Cp), 1.98 (s, 15 H, Cp*), 1.38 (s, 9 H, ^tBu), -7.17 (s, 1 H, μ -H) ppm. ¹³C NMR (100 MHz, *d*₈-THF, 20 °C): δ = 109.2 (Cp), 93.6 (C₅-Cp*), 80.2 (CMe₃), 27.9 (C(CH₃)₃), 9.6 (Me₅-Cp*) ppm. ¹⁹F NMR (376 MHz, *d*₈-THF, 20 °C): δ = -79.1 ppm. ¹H NMR (400 MHz, CD₂Cl₂, 20 °C): 6.06 (s, 10 H, Cp), 1.95 (s, 15 H, Cp*), 1.32 (s, 9 H, ^tBu), -8.09 (s, 1 H, μ -H) ppm. ¹⁹F NMR (376 MHz, CD₂Cl₂, 20 °C): δ = -78.9 ppm. EA calcd for C₂₅H₃₅F₃HfIrNO₃S: C, 35.02; H, 4.12; N, 1.63; Found: C, 35.17; H, 4.36; N, 1.70%.

General procedure for the preparation of Cp*Ir(μ -N^tBu)(μ -N₃C₆H₄R)HfCp₂, R = Me₂N, Me, H, Br, CF₃

A ca 0.01 M solution of the appropriate aryl azide in *n*-pentane was cooled to -35 °C. This solution was then transferred onto solid 2-Hf, immediately forming a green solution. The solution was stirred for 2 h before the solvent was removed under dynamic vacuum. Storing the green solids under dynamic vacuum for ca. 6 h removed the excess aryl azide, leaving only the azide complex, Cp*Ir(μ -N^tBu)(μ -N₃C₆H₄R)HfCp₂. The

following lists the amounts used for each procedure: for R=NMe₂, 40 mg, 0.057 mmol, 1 equiv Hf and 9.2 mg, 0.057 mmol, 1.0 equiv azide gave 46 mg, 0.053 mmol, 93%; for 4-Me, 71 mg, 0.10 mmol, 1 equiv Hf and 12 mg, 0.105 mmol, 1.05 equiv azide gave 80 mg, 0.095 mmol, 94%; for 4-H, 80 mg, 0.113 mmol, 1 equiv Hf and 12 mg, 0.117 mmol, 1.05 equiv azide gave 91 mg, 0.110 mmol, 97%; for 4-Br, 40 mg, 0.057 mmol, 1 equiv Hf and 12 mg, 0.060 mmol, 1.05 equiv azide gave 48 mg, 0.052 mmol, 92%; and for 4-CF₃, 40 mg, 0.057 mmol, 1 equiv Hf and 11 mg, 0.058 mmol, 1.05 equiv azide gave 47 mg, 0.051 mmol, 92%.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-N₃C₆H₄NMe₂)HfCp₂

¹H NMR (500 MHz, C₆D₆, 20 °C): 8.10 (d, *J*_{HH} = 7 Hz, 2 H, *o*-Ar), 6.91 (d, *J*_{HH} = 7 Hz, 2 H, *m*-Ar), 5.95 (s, 10 H, Cp), 2.73 (s, 6 H, NMe₂), 1.56 (s, 9 H, ^tBu), 1.46 (s, 15 H, Cp*) ppm. ¹³C NMR (125 MHz, C₆D₆, 20 °C): δ = 148.2 (*ipso*-Ar), 147.5 (Ar), 122.2 (Ar), 114.8 (Ar), 108.4 (Cp), 88.8 (C₅-Cp*), 69.9 (CMe₃), 41.6 (N(CH₃)₂), 36.5 (C(CH₃)₃), 10.9 (*Me*₅-Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-N₃C₆H₄Me)HfCp₂

¹H NMR (500 MHz, C₆D₆, 20 °C): 8.03 (d, *J*_{HH} = 8 Hz, 2 H, *o*-tol), 7.35 (d, *J*_{HH} = 8 Hz, 2 H, *m*-tol), 5.88 (s, 10 H, Cp), 2.34 (s, 3 H, tol-Me), 1.58 (s, 9 H, ^tBu), 1.44 (s, 15 H, Cp*) ppm. ¹³C NMR (125 MHz, C₆D₆, 20 °C): δ = 153.9 (*ipso*-tol), 132.0 (*ipso*-tol), 130.3 (tol), 121.3 (tol), 108.4 (Cp), 89.0 (C₅-Cp*), 70.2 (CMe₃), 36.4 (C(CH₃)₃), 21.6 (tol-Me), 10.7 (*Me*₅-Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-N₃Ph)HfCp₂ (5-Hf)

¹H NMR (400 MHz, C₆D₆, 20 °C): 8.09 (d, *J*_{HH} = 7 Hz, 2 H, *o*-Ph), 7.55 (t, *J*_{HH} = 7 Hz, 2 H, *p*-Ph), 7.17 (t, obscured by residual solvent, 2 H, *m*-Ph), 5.88 (s, 10 H, Cp), 1.57 (s, 9 H, ^tBu), 1.42 (s, 15 H, Cp*) ppm. ¹³C NMR (100 MHz, C₆D₆, 20 °C): δ = 156.0 (*ipso*-Ph), 129.7 (Ph), 123.0 (Ph), 121.3 (Ph), 108.5 (Cp), 89.1 (C₅-Cp*), 70.4 (CMe₃), 36.5 (C(CH₃)₃), 10.8 (*Me*₅-Cp*) ppm. EA calcd for C₃₀H₃₉HfIrN₄: C, 43.60; H, 4.75; N, 6.77; Found: C, 44.0; H, 5.06; N, 5.90%. ESI-MS(+): *m/z* = 829.2376 (829.2348, MH⁺) amu.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-N₃C₆H₄Br)HfCp₂

¹H NMR (600 MHz, C₆D₆, 20 °C): 7.74 (d, *J*_{HH} = 9 Hz, 2 H, *o*-Ar), 7.58 (d, *J*_{HH} = 9 Hz, 2 H, *m*-Ar), 5.80 (s, 10 H, Cp), 1.51 (s, 9 H, ^tBu), 1.35 (s, 15 H, Cp*) ppm. ¹³C NMR (150 MHz, C₆D₆, 20 °C): δ = 155.0 (*ipso*-Ar), 133.2 (Ar), 118.2 (Ar), 115.1 (Ar), 108.6 (Cp), 89.2 (C₅-Cp*), 70.6 (CMe₃), 36.3 (C(CH₃)₃), 10.7 (*Me*₅-Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-N₃C₆H₄CF₃)HfCp₂

¹H NMR (600 MHz, C₆D₆, 20 °C): 7.89 (d, *J*_{HH} = 8 Hz, 2 H, *o*-Ar), 7.74 (d, *J*_{HH} = 8 Hz, 2 H, *m*-Ar), 5.82 (s, 10 H, Cp), 1.53 (s, 9 H, ^tBu), 1.37 (s, 15 H, Cp*) ppm. ¹³C NMR (150 MHz, C₆D₆, 20 °C): δ = 158.8 (*ipso*-Ar), 126.92 (q, *J* = Hz, Ar), 125.8 (Ar), 123.3 (q, *J*_{CF} = 320 Hz, CF₃), 108.7 (Cp), 89.5 (C₅-Cp*), 70.9 (CMe₃), 36.3 (C(CH₃)₃), 10.7 (*Me*₅-Cp*) ppm. One ¹³C resonance was not located, presumably because it is obscured by the resonance for C₆D₆. ¹⁹F NMR (376 MHz, C₆D₆, 20 °C): δ = -60.8 ppm.

Preparation of Cp*Ir(μ-N^tBu)(μ-NPh)HfCp₂ (5-Hf)

A 0.6 mL C₆D₆ solution of 45 mg(0.054 mmol) Cp*Ir(μ-N^tBu)(μ-N₃Ph)HfCp₂ was heated to 120 °C for 6 h. After this time the mixture was analyzed by ¹H NMR spectroscopy which indicated that the starting material had been consumed and conversion to the desired product proceeded in >90%. The material was crystallized from *n*-pentane/Et₂O mixtures at -35 °C in two crops. 36 mg, 0.045, 83% ¹H NMR (600 MHz, C₆D₆, 20 °C): δ = 7.36 (d, *J*_{HH} = 8 Hz, 2 H, *o*-Ph), 6.95 (t, *J*_{HH} = 8 Hz, 1 H, *p*-Ph), 6.91 (t, *J*_{HH} = 8 Hz, 2 H, *m*-Ph), 5.81 (s, 10 H, Cp), 1.71 (s, 9 H, ^tBu), 1.20 (s, 15 H, Cp*) ppm. ¹³C NMR (100 MHz, C₆D₆, 20 °C): δ = 168.2 (*ipso*-Ph), 128.3 (Ph), 125.9 (Ph), 120.1 (Ph), 109.3 (Cp), 87.1 (C₅-Cp*), 68.4 (CMe₃), 36.7 (C(CH₃)₃), 11.0 (*Me*₅-Cp*) ppm.

Formation of Cp*Ir(μ-N^tBu)(μ-NC₆H₄R)HfCp₂ by N₂ Extrusion from Cp*Ir(μ-N^tBu)(μ-N₃C₆H₄R)HfCp₂ (R = NMe₂, Me, Br, CF₃)

The procedure described for the preparation of 5-Hf was followed for each compound in this series.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-NC₆H₄NMe₂)HfCp₂

¹H NMR (400 MHz, C₆D₆, 20 °C): δ = 6.94 (d, *J*_{HH} = 8 Hz, 2 H, *o*-Ar), 6.85 (d, *J*_{HH} = 8 Hz, 2 H, *m*-Ar), 5.86 (s, 10 H, Cp), 3.31 (s, 6 H, NMe₂), 2.76 (s, 9 H, ^tBu), 1.76 (s, 15 H, Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-NC₆H₄Me)HfCp₂

¹H NMR (500 MHz, C₆D₆, 20 °C): δ = 6.96 (d, *J*_{HH} = 8 Hz, 2 H, *o*-tol), 6.87 (d, *J*_{HH} = 8 Hz, 2 H, *m*-tol), 5.83 (s, 10 H, Cp), 2.44 (s, 3 H, tol-Me), 1.73 (s, 9 H, ^tBu), 1.22 (s, 15 H, Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-NC₆H₄Br)HfCp₂

¹H NMR (300 MHz, C₆D₆, 20 °C): δ = 7.62 (d, *J*_{HH} = 9 Hz, 2 H, *o*-Ar), 6.61 (d, *J*_{HH} = 9 Hz, 2 H, *m*-Ar), 5.74 (s, 10 H, Cp), 1.65 (s, 9 H, ^tBu), 1.13 (s, 15 H, Cp*) ppm.

Characterization Data for Cp*Ir(μ-N^tBu)(μ-NC₆H₄CF₃)HfCp₂

¹H NMR (300 MHz, C₆D₆, 20 °C): δ = 7.56 (d, *J*_{HH} = 8 Hz, 2 H, *o*-Ar), 6.70 (d, *J*_{HH} = 8 Hz, 2 H, *m*-Ar), 5.73 (s, 10 H, Cp), 1.62 (s, 9 H, ^tBu), 1.13 (s, 15 H, Cp*) ppm. ¹⁹F NMR (376 MHz, C₆D₆, 20 °C): δ = -60.4 ppm.

Kinetics Measurements for 4-Hf → 5-Hf

A stock solution of 2-Hf (2.5 mmol) in *d*₁₂-mesitylene was prepared and *ca.* 1 equiv of 1,3,5-trimethoxybenzene was added as an internal standard. An aliquot of the solution was transferred to a J. Young tube which was sealed with a Teflon screw-cap under *ca.* 1 atm of N₂. The sample was transported to a 500 MHz NMR spectrometer at 20 °C and inserted into a pre-warmed probe. Temperatures from 105 to 135 °C were studied, and the probe temperature was determined by use of a neat ethylene glycol sample. The sample was given 5 min to ensure that the temperature had reached equilibrium before kinetics measurements began. To obtain kinetic information, single-scan ¹H NMR spectra were obtained in either 5 or 10 min intervals, and the relative concentration of 4-Hf was determined by integration of the Cp resonance, which decreases in intensity with time. The resonances corresponding to 5-Hf increased with time. However, following the concentration of 5-Hf as a function of time did not

produce useful kinetic information due to the fact that **5-Hf** precipitates from the reaction mixture. The integrals for the added 1,3,5-trimethoxybenzene were invariant with respect to time at all temperatures studied. For complexes $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-N}_3\text{C}_6\text{H}_4\text{R})\text{HfCp}_2$ ($\text{R} = \text{Me}_2\text{N}, \text{Me}, \text{Br}, \text{CF}_3$), the kinetic measurements were carried out in an identical fashion at 130 °C. The reaction kinetics were monitored for no fewer than 3 half-lives, except for when the total length of the required run made this inconvenient. Specifically, 1 and 2.6 half-lives of data were obtained at 105 and 115 °C, respectively, and 2.4 half-lives for $\text{Cp}^*\text{Ir}(\mu\text{-}^t\text{Bu})(\mu\text{-N}_3\text{C}_6\text{H}_4\text{NMe}_2)\text{HfCp}_2$ at 130 °C. This experiment was repeated using a solution containing 12.5 mmol of **5-Hf** at 130 °C, and identical kinetic behavior was observed.

Preparation of $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-S})\text{HfCp}_2$ (**6-Hf**)

At 20 °C, an 8 mL solution of **2-Hf** (45 mg, 0.070 mmol, 1 equiv) in C_6H_6 was added to a 2 mL C_6H_6 slurry of S_8 (2.25 mg, 0.0088 mmol, 0.126 equiv). The resulting mixture was stirred for 30 min before it is filtered twice through a glass fiber and the solution frozen. C_6H_6 was removed from the frozen solution to yield a green solid (43.5 mg, 0.064 mmol, 92%). ^1H NMR (500 MHz, C_6D_6 , 20 °C): 5.73 (s, 10 H, Cp), 1.73 (s, 9 H, ^tBu), 1.50 (s, 15 H, Cp*) ppm. ^{13}C NMR (125 MHz, C_6D_6 , 20 °C): $\delta = 109.4$ (Cp), 88.7 ($\text{C}_5\text{-Cp}^*$), 72.2 (CMe_3), 36.2 ($\text{C}(\text{CH}_3)_3$), 11.36 ($\text{Me}_5\text{-Cp}^*$) ppm. ESI-MS: $m/z = 742.1610$ (742.1579 , MH^+) amu.

General Remarks Regarding Computational Methods

Density functional theory (DFT) calculations were performed using the Gaussian09 software package.¹⁷ The basis set SDD was used for Hf, Ir, Ti, and Zr with effective core potentials,¹⁸⁻²⁰ whereas 6-31G* was used for all other atoms.^{21,22} Both hybrid functional B3LYP²³⁻²⁶ and pure DFT functional PW91PW91²⁷⁻³¹ were used for calculations involving 1-M ($\text{M} = \text{Ti}, \text{Zr}, \text{Hf}$). While only B3LYP was used for **2-Zr**. Optimized geometries for these compounds are given in Tables S4–S7. All geometry optimization procedures were initiated from atomic coordinates of single-crystal X-ray structures, and the resulting optimized geometries were subjected to vibrational frequency calculations to ensure that these correspond to a minimum on the potential energy surface.

DFT Vibrational Frequency Analysis for **2-M**

Optimized geometries for **2-M** were subjected to vibrational frequency analysis that also predicted Raman intensity using the Gaussian09 package.¹⁷ Predominantly metal-metal stretching modes were located by inspection of the atomic displacements for

each mode, and were visualized using the Gaussian09 package. A comparison of the energies between the DFT-predicted and experimentally assigned vibrations are shown in Table 3. These energies are not corrected and no scaling factor was used.

Table 3. DFT Predicted Ir–M Stretching Modes for **2-M**.

2-M	experimental	B3LYP	PW91PW91
Ti	238	279.02	280.95
Zr	153	189.18	185.57
Hf	—	164.26	168.99

TDDFT Calculations for **2-M**

The first 50 singlet-to-singlet electronic excited states were calculated for each **2-M** using the TDDFT³²⁻³⁶ implementation provided with the Gaussian09 software package.¹⁷ Both the pure DFT functional PW91PW91²⁷⁻³¹ and the hybrid functional B3LYP²³⁻²⁶ were used. Molecular orbital contour plots were generated from the Gaussian checkpoint file that was written by the calculation using the B3LYP functional. The first 50 electronic excited states for **2-Ti** and the first 20 electronic excited states for **2-Zr** and **2-Hf** are listed below.

DFT Calculated ^1H NMR shifts for **3-Zr**

The NMR shifts were predicted using the gauge-independent atomic orbital method³⁷⁻⁴¹ as it is implemented by the NMR prediction utility⁴²⁻⁴⁴ that is packaged with Gaussian09.¹⁷ This method provided chemical shielding tensors for the hydrogen atoms contained in **3-Zr**. The calculated chemical shielding tensors of individual hydrogen atoms were averaged for each the Cp, Cp*, and ^tBu ligand. The chemical shift tensors for the Cp* ligand was used as the reference signal, and the chemical shift set to 1.99 ppm, the value found in $d_8\text{-THF}$. This led to the predicted shifts of 6.06, 1.24, and -4.39 ppm for the Cp, ^tBu , and hydride ligands, respectively.

DFT Vibrational Frequency Analysis for **3-Zr**

A vibrational frequency analysis was performed by the Gaussian09 computational package.¹⁷ A single vibration was found that only displaced the hydride ligand, while the atomic positions of other atoms were virtually unchanged. This vibrational mode corresponds to an energy of 2047.5 cm^{-1} .

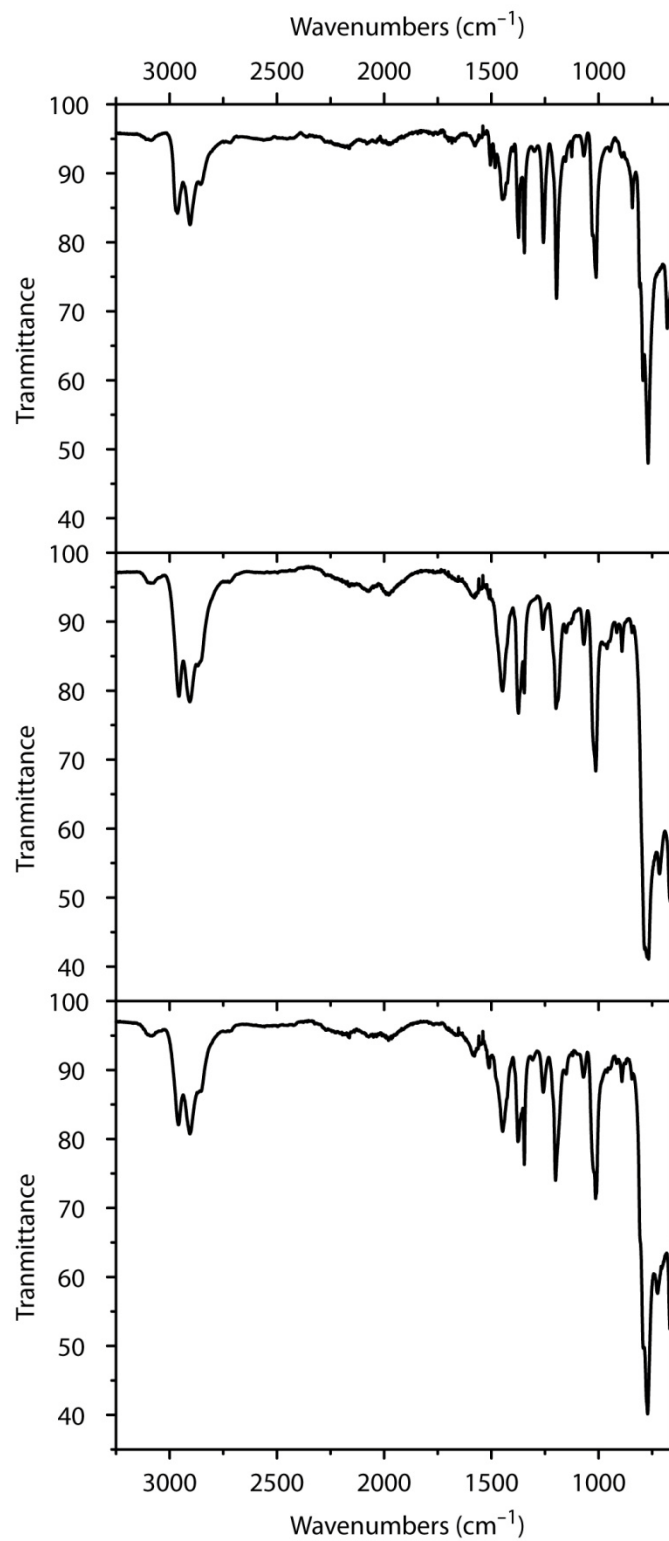


Figure S1. ATR-IR spectroscopy data for Cp*Ir(μ-N^tBu)MCp₂ (2-M).

Table S1. Solid-state (ATR) IR Data for Cp*Ir(μ -N^tBu)MCp₂ (**2-M**).

Ti		Zr		Hf	
Position (cm ⁻¹)	Transmittance	Position (cm ⁻¹)	Transmittance	Position (cm ⁻¹)	Transmittance
3088	94.77	3088	95.58	3088	94.95
2964	84.21	2957	79.19	2959	82.07
2905	82.55	2906	78.39	2906	80.75
2855	88.40	2869	83.59	2869	86.70
1505	91.18	1509	93.38	1511	90.32
1447	86.21	1447	79.95	1447	81.10
1373	80.70	1374	76.74	1376	79.62
1346	78.47	1346	79.68	1346	76.30
1257	79.97	1259	88.90	1258	86.78
1196	71.87	1199	77.41	1201	74.01
1069	92.53	1069	86.71	1071	88.96
1012	74.92	1013	68.34	1014	71.36
948	93.20	915	88.40	915	90.95
891	92.36	891	85.71	891	88.34
843	85.00	844	88.42	845	88.69
808	73.48	785	41.36	808	65.90
793	59.97	774	42.48	792	49.69
769	47.97	767	41.08	771	40.16
—	—	716	53.44	725	57.62
680	67.52	666	49.43	668	52.48

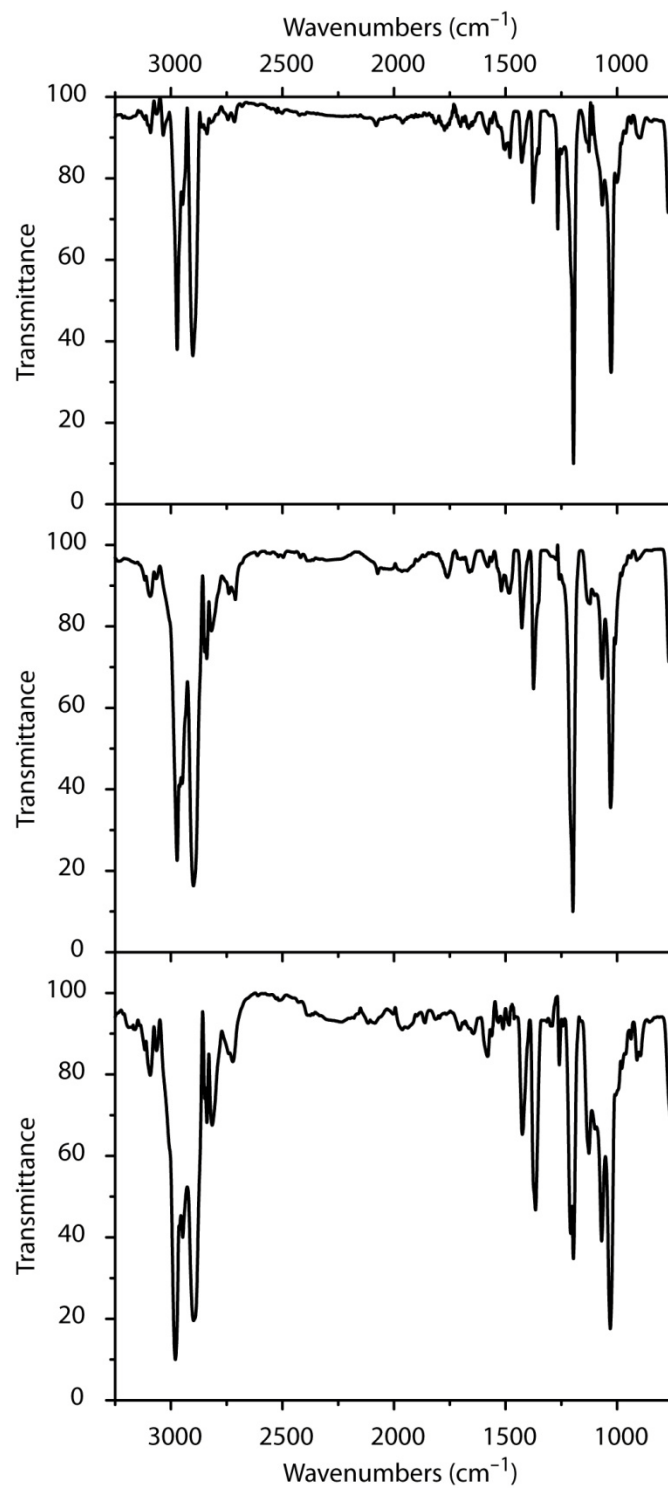


Figure S2. Solution infrared spectroscopy data for Cp*Ir(μ -N^tBu)MCp₂ (2-M) in C₆D₆ using a KBr cell.

Table S2. Solution (C₆D₆) IR Data for Cp*Ir(μ-N^tBu)MCp₂ (**2-M**).

Ti		Zr		Hf	
Position (cm ⁻¹)		Position (cm ⁻¹)		Position (cm ⁻¹)	
3092	(w)	3093	(w)	3093	(w)
2972	(s)	2972	(s)	2980	(s)
2947	(m)	2949	(m)	2947	(m)
2900	(s)	2899	(s)	2898	(s)
1427	(w)	1427	(w)	1425	(m)
1376	(m)	1374	(m)	1366	(m)
1266	(m)	1211	(m)	1209	(m)
1195	(s)	1198	(s)	1196	(s)
1066	(m)	1067	(m)	1069	(m)
1027	(s)	1029	(s)	1030	(s)

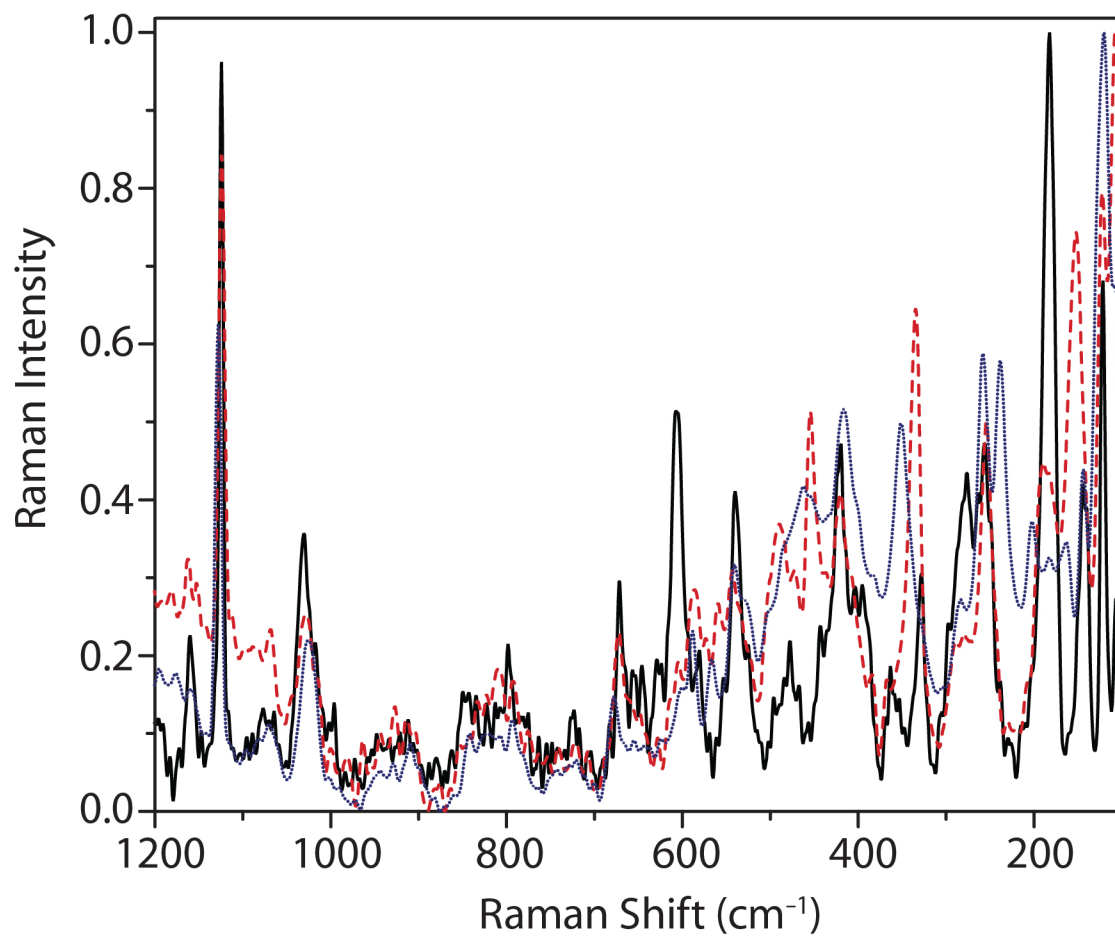


Figure S3. Raman spectra ($\lambda_{\text{ex}} = 633\text{nm}$) of **2-Ti** (.....), **2-Zr** (----), and **2-Hf** (—) as powders at 20 °C are shown from 1200–100 cm^{-1} .

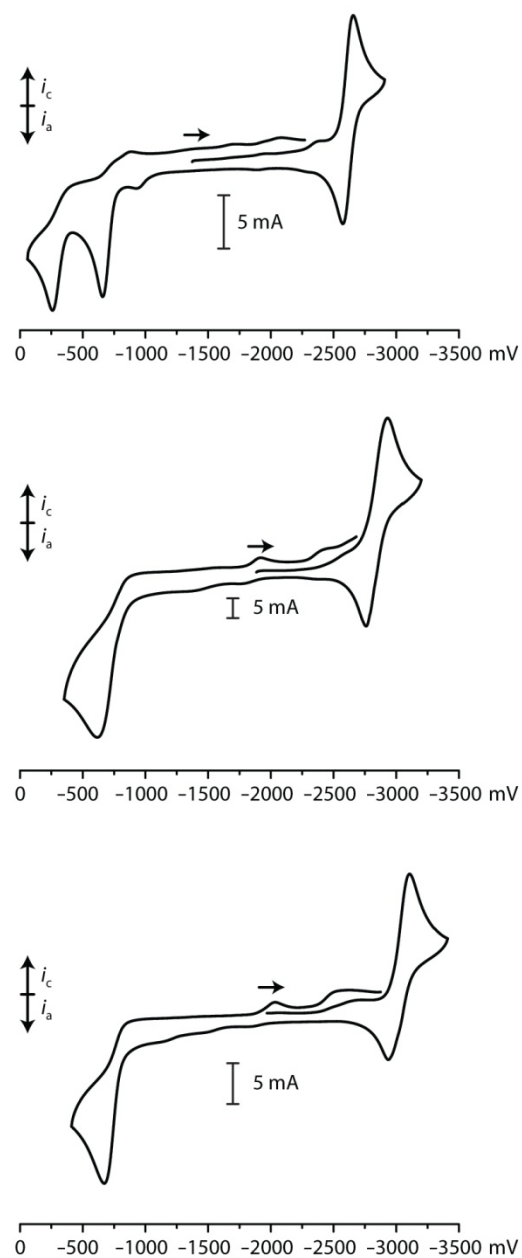


Figure S4. Cyclic voltammetry data for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{MCp}_2$ (**2-M**) are shown from Ti (top) to Zr (middle) to Hf (bottom). Data acquired in 0.1 M $[\text{N}^t\text{Bu}_4][\text{PF}_6]/\text{THF}$ solution at 20 °C. Potentials are shown in reference to $\text{Cp}_2\text{Fe}^{0/+}$.

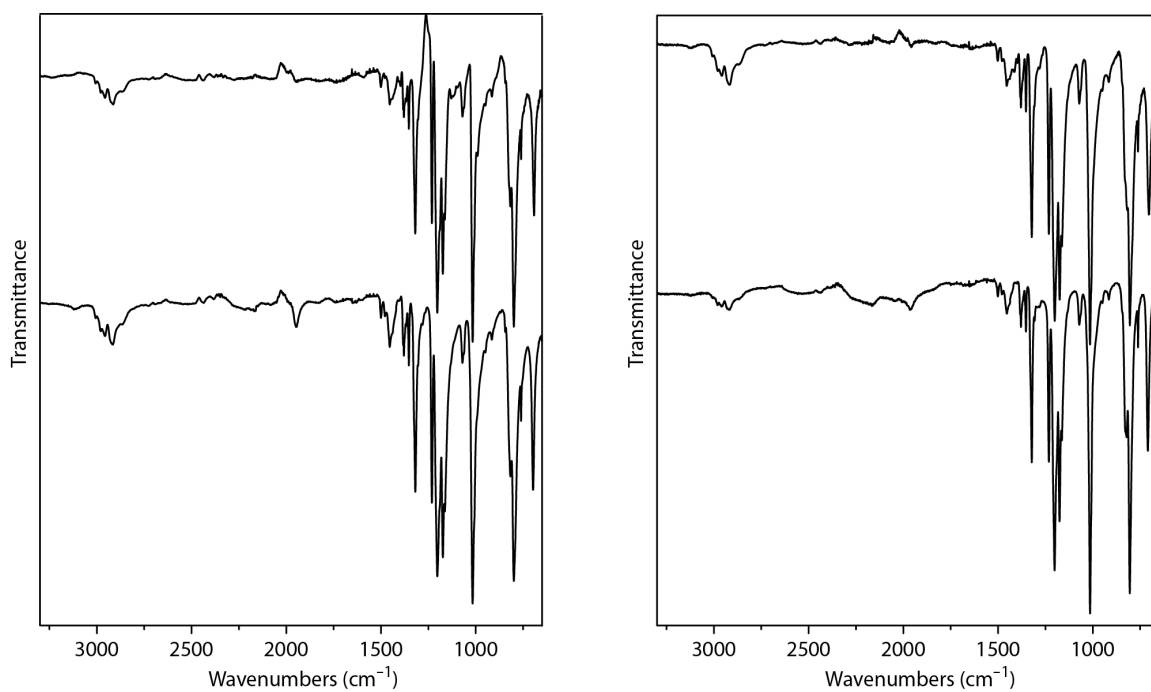


Figure S5. Infrared spectra (ATR) for $\text{Cp}^*\text{Ir}(\mu\text{-N}'\text{Bu})(\mu\text{-D})\text{ZrCp}_2\text{OTf}$ (above) and $\text{Cp}^*\text{Ir}(\mu\text{-N}'\text{Bu})(\mu\text{-H})\text{ZrCp}_2\text{OTf}$ (**3-Zr**, below) are shown on the left, while those for $\text{Cp}^*\text{Ir}(\mu\text{-N}'\text{Bu})(\mu\text{-D})\text{HfCp}_2\text{OTf}$ (above) and $\text{Cp}^*\text{Ir}(\mu\text{-N}'\text{Bu})(\mu\text{-H})\text{HfCp}_2\text{OTf}$ (**3-Hf**, below) are shown on the right.

Table S3. Infrared spectral data for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-H})\text{MCp}_2\text{OTf}$ and $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-D})\text{MCp}_2\text{OTf}$ (M = Zr, Hf).

$\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})$ $(\mu\text{-H})\text{ZrCp}_2\text{OTf}$		$\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})$ $(\mu\text{-D})\text{ZrCp}_2\text{OTf}$		$\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})$ $(\mu\text{-H})\text{HfCp}_2\text{OTf}$		$\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})$ $(\mu\text{-D})\text{HfCp}_2\text{OTf}$	
Position (cm^{-1})	Transmittance	Position (cm^{-1})	Transmittance	Position (cm^{-1})	Transmittance	Position (cm^{-1})	Transmittance
2915	87	2913	94	2919	95	2919	93
1948	91	—	—	1966	95	—	—
1454	86	1454	94	1454	94	1454	93
1379	84	1379	91	1379	92	1379	90
1353	82	1353	88	1353	91	1353	89
1319	52	1319	63	1322	70	1322	71
1231	49	1231	66	1232	70	1232	71
1203	32	1203	45	1202	53	1201	58
1173	37	1173	54	1176	61	1175	61
1070	82	1070	91	1071	92	1071	90
1016	26	1016	38	822	74	1015	55
				805	49	804	57
				761	89	761	83
				709	72	705	74

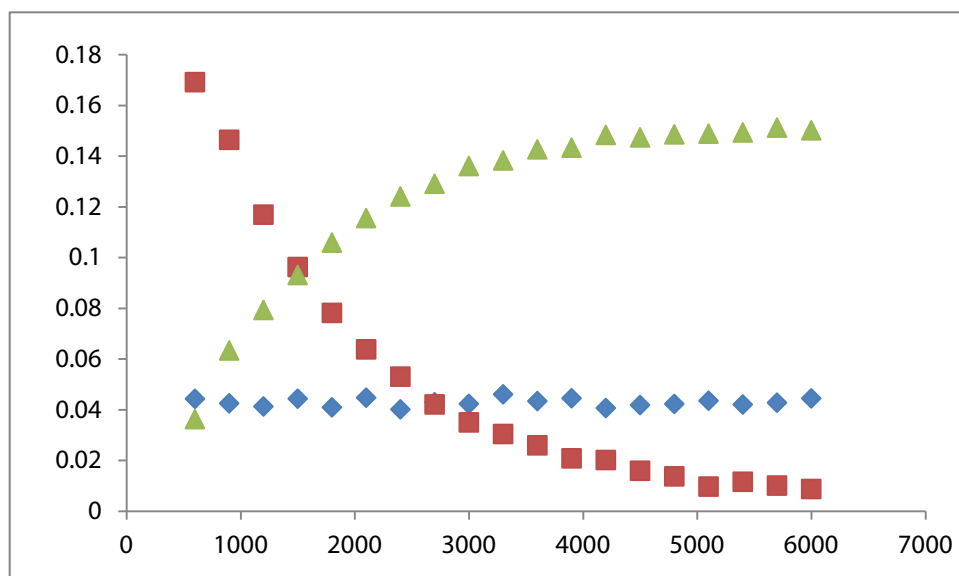


Figure S6. Integral intensities for the ^1H NMR signals for Cp resonances of **4-Hf** (squares) and **6-Hf** (triangles), as well as the methyl resonance of 1,3,5-trimethoxybenzene (diamonds) at 130 °C as a function of time.

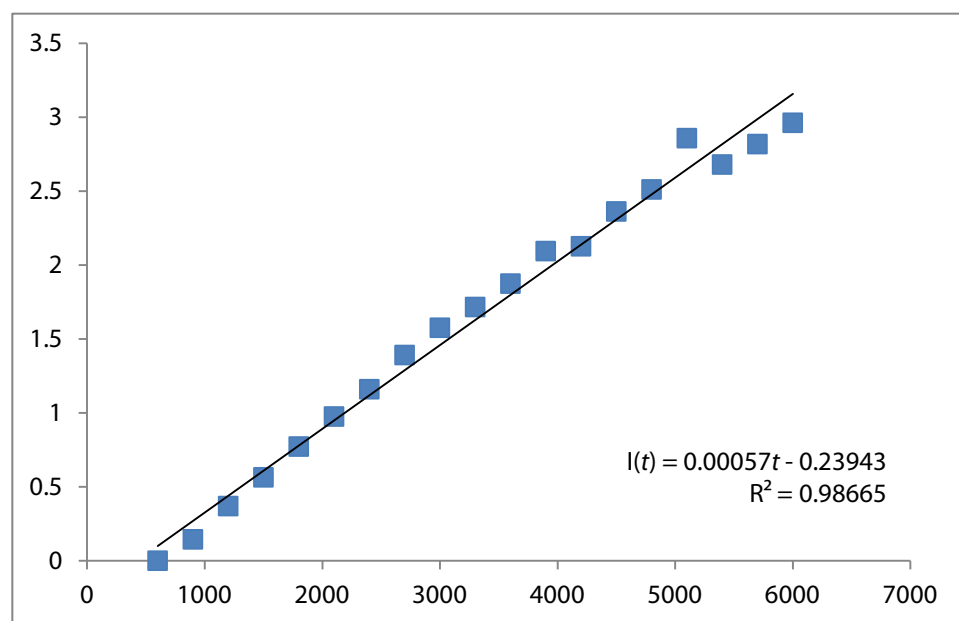


Figure S7. A fit to the natural log of the ^1H NMR integral intensity for **4-Hf** as a function of time. The fit that is shown was used to determine the rate constant at 130 °C.

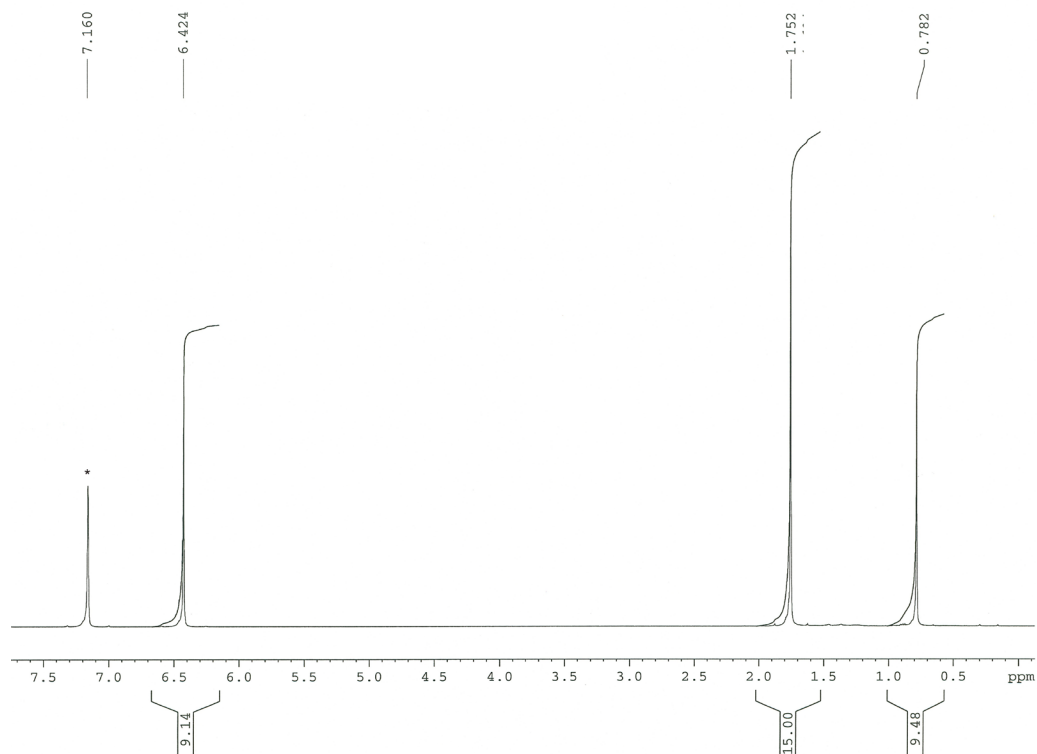


Figure S8. ^1H NMR (400 MHz, C_6D_6 , 20 °C) spectrum of $\text{Cp}^*\text{Ir}(\mu\text{-N}^i\text{Bu})\text{TiCp}_2$ (**2-Ti**).

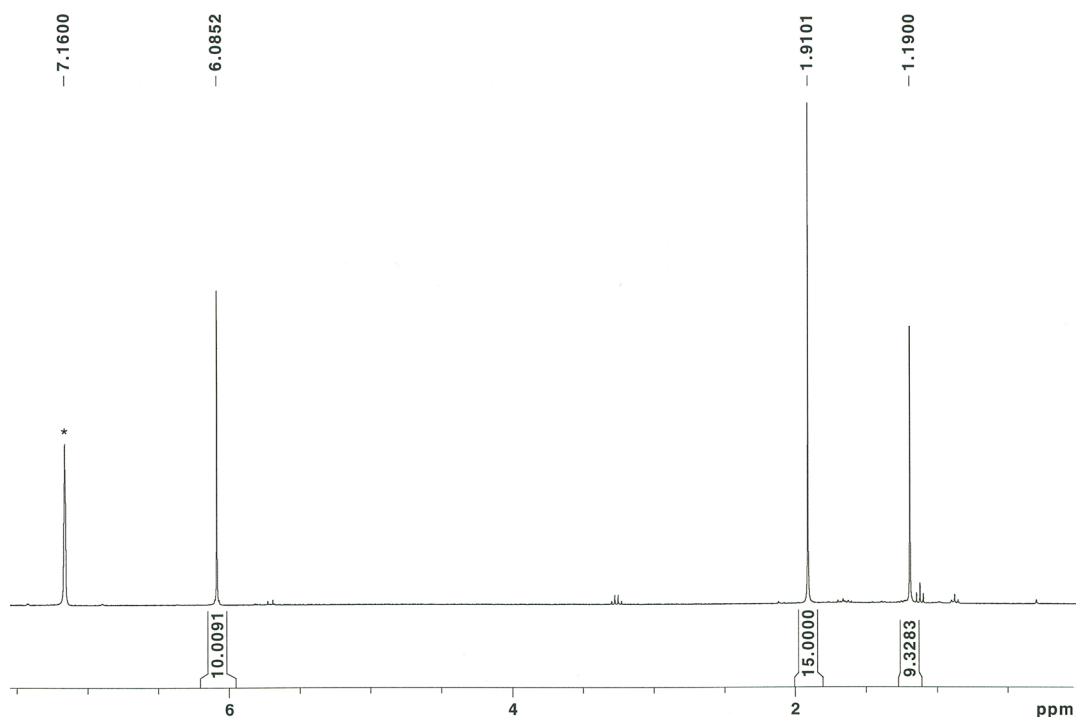


Figure S9. ^1H NMR (400 MHz, C_6D_6 , 20 °C) spectrum of $\text{Cp}^*\text{Ir}(\mu\text{-N}^i\text{Bu})\text{HfCp}_2$ (**2-Hf**).

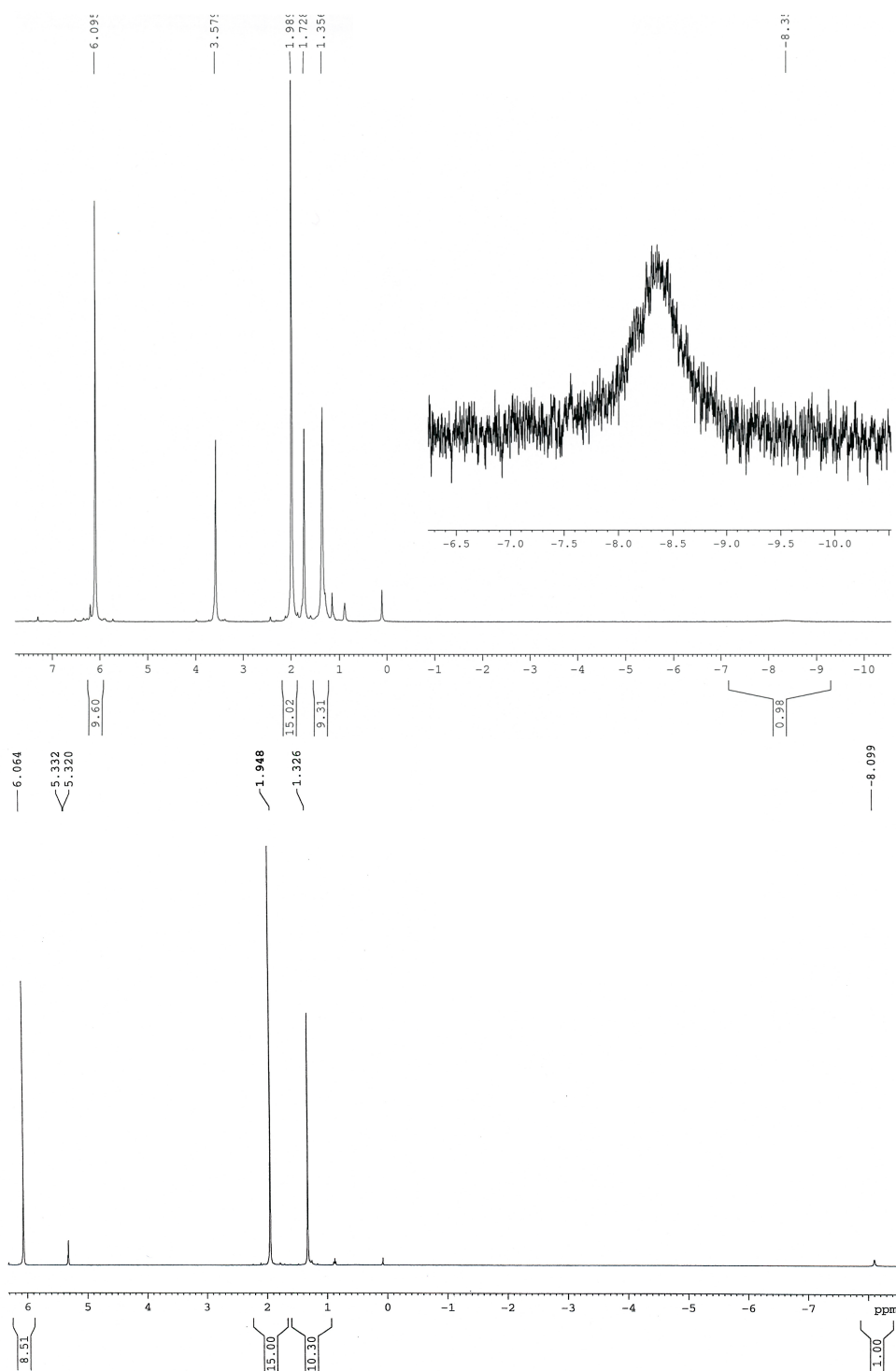


Figure S10. ¹H NMR (500 MHz, 20 °C) spectrum of Cp*Ir(μ-N^tBu)(μ-H)ZrCp₂OTf (3-Zr) in *d*₈-THF (top) and CD₂Cl₂. (bottom).

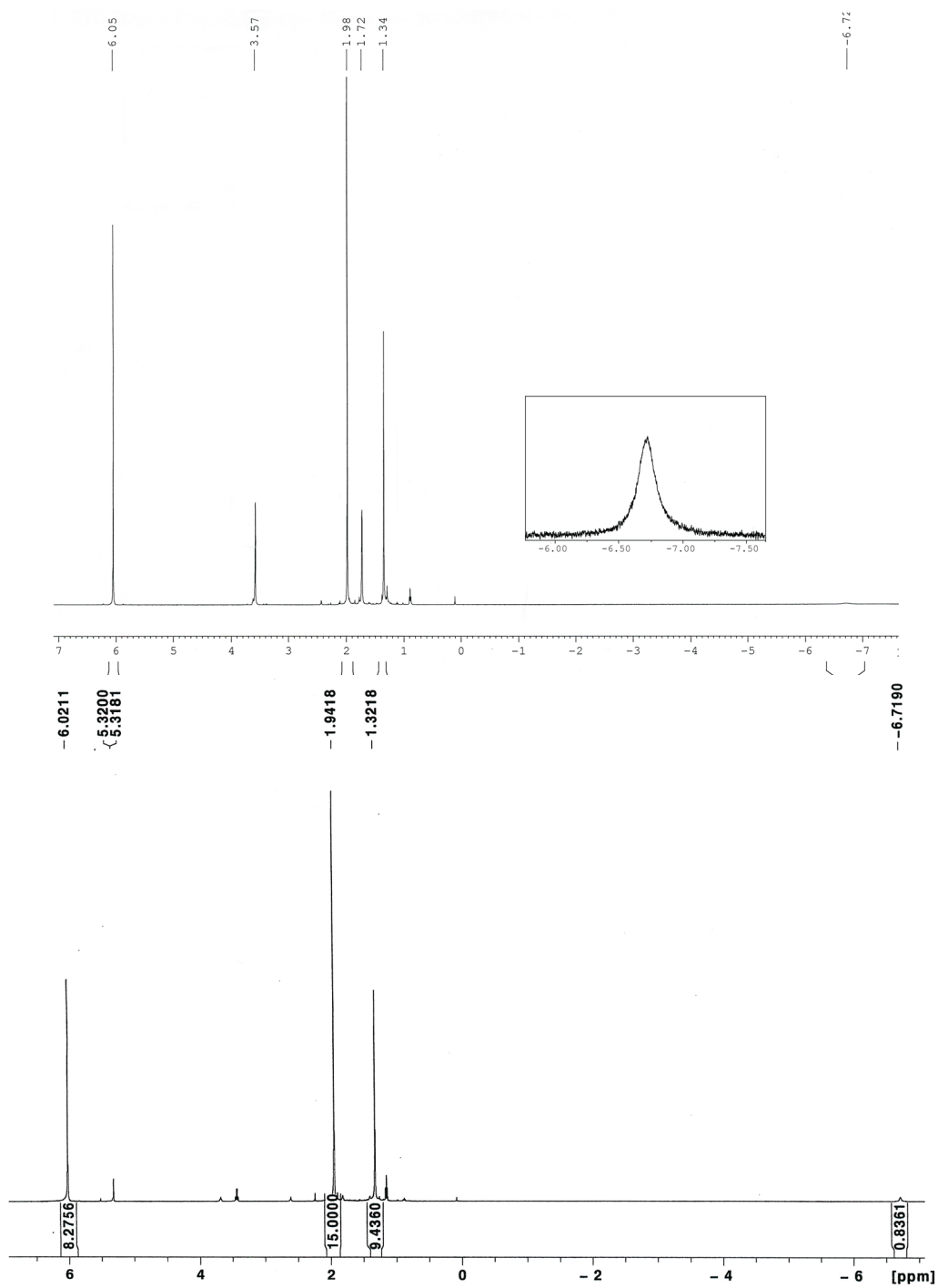


Figure S11. ¹H NMR (500 MHz, 20 °C) spectrum of Cp*Ir(μ-N^tBu)(μ-H)HfCp₂OTf (3-Hf) in *d*₈-THF (top) and CD₂Cl₂ (bottom).

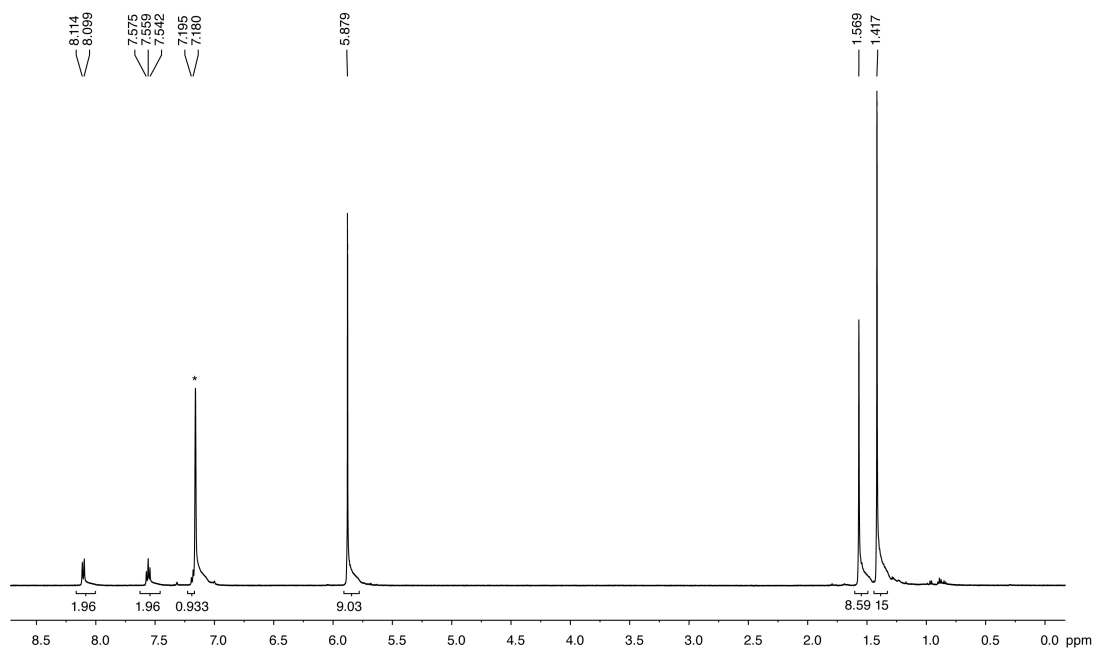


Figure S12. ¹H NMR (300 MHz, C₆D₆, 20 °C) spectrum of Cp*Ir(μ-NⁱBu)(μ-N₃Ph)HfCp₂ (**4-Hf**).

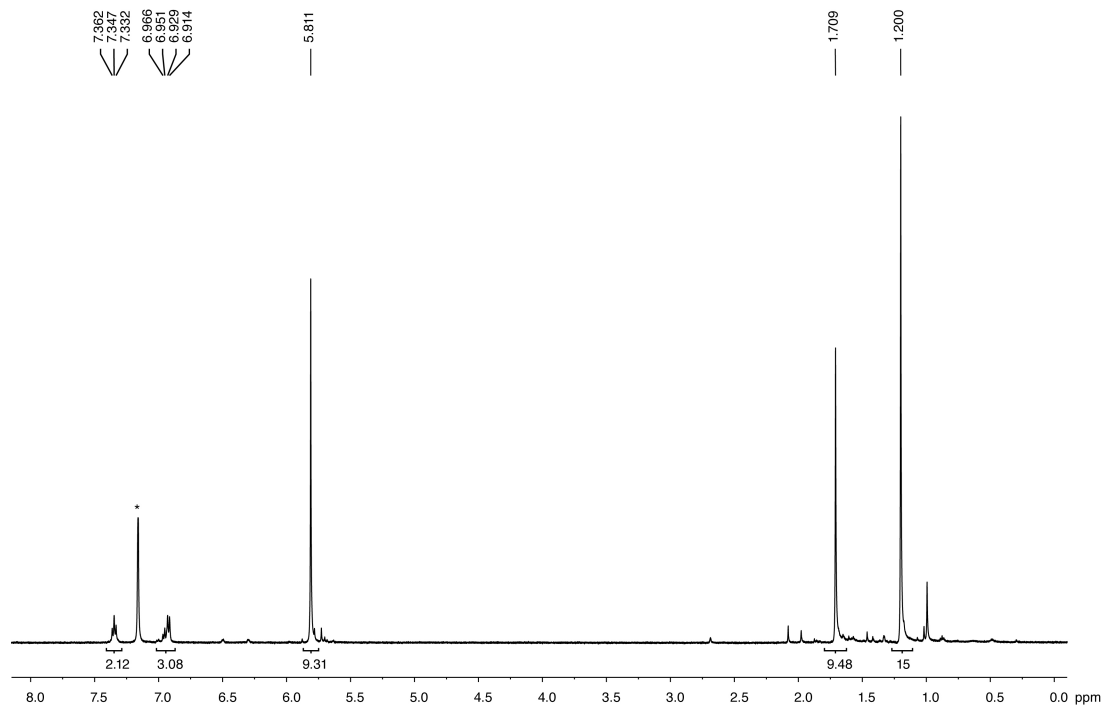


Figure S13. ¹H NMR (500 MHz, C₆D₆, 20 °C) spectrum of Cp*Ir(μ-NⁱBu)(μ-NPh)HfCp₂ (**5-Hf**).

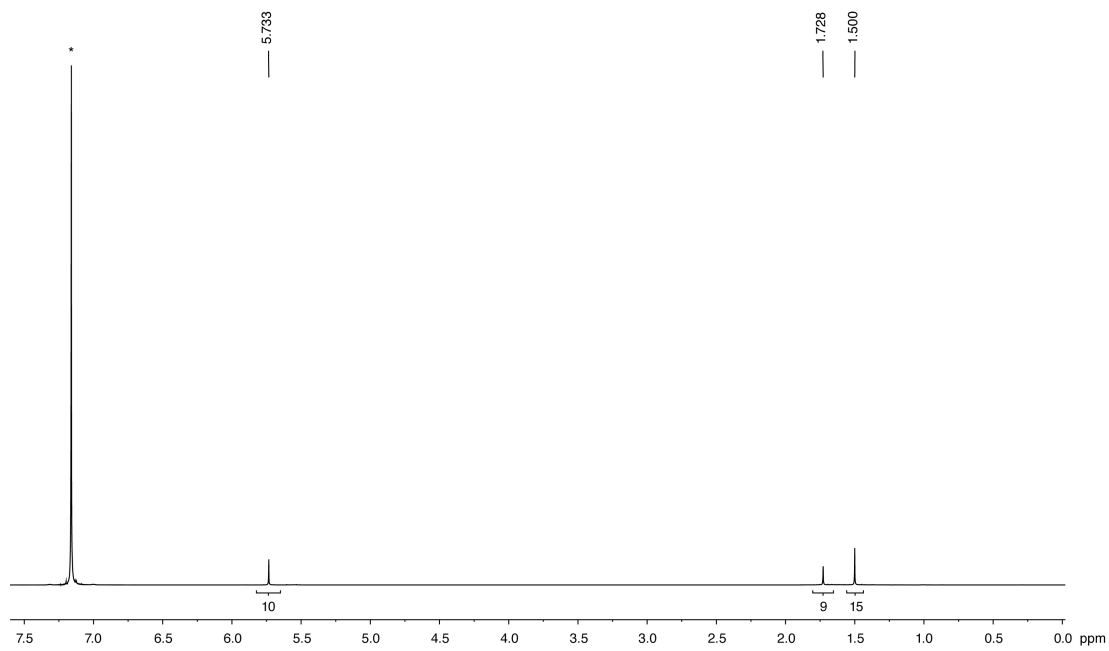


Figure S14. ^1H NMR (500 MHz, C_6D_6 , 20 °C) spectrum of $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})(\mu\text{-S})\text{HfCp}_2$ (**6-Hf**).

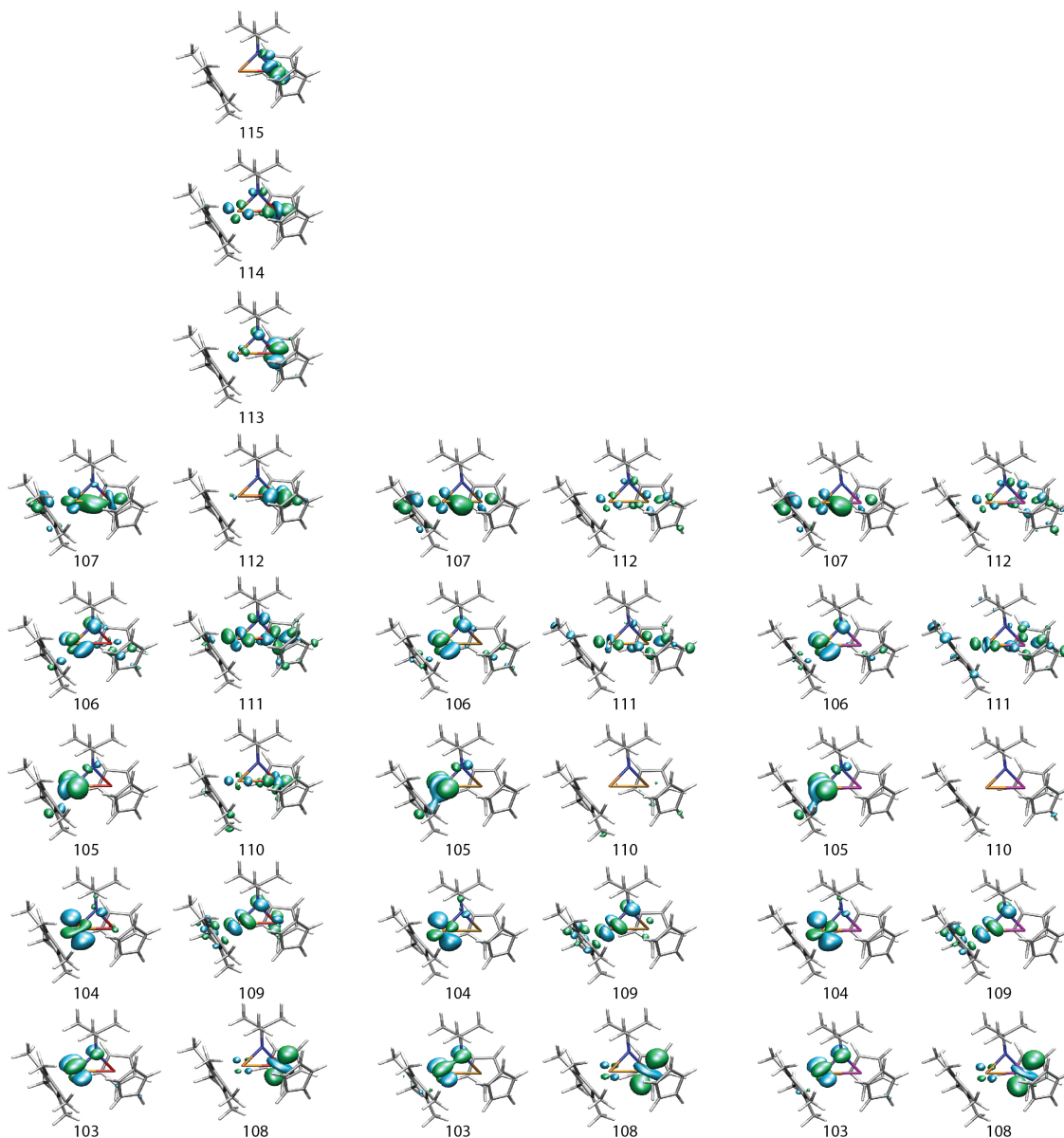


Figure S15. Orbital contour plots for **2-Ti** (left), **2-Zr** (center), and **2-Hf** (right), calculated using the B3LYP functional. In each case, orbital 107 is the HOMO and orbital 108 is the LUMO. These orbitals were rendered at an isosurface value of 0.04, except for orbitals 110–112 for **2-Zr** and **2-Hf**. Because these orbitals are delocalized they were rendered at a value of 0.08. In particular, for both **2-Zr** and **2-Hf** orbital 110 is highly delocalized and contains density in the C-H bonds of the Cp and Cp* ligands.

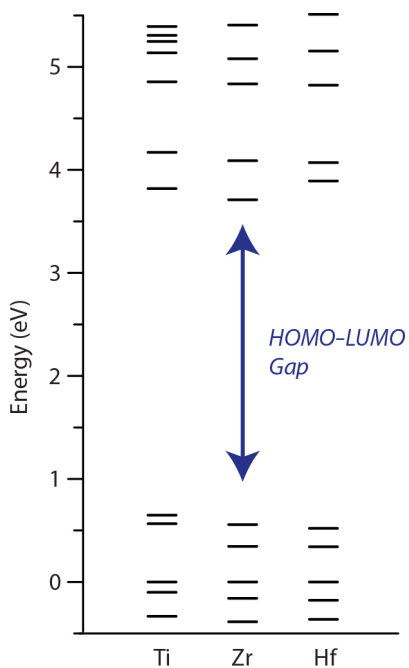


Figure 16. Frontier orbital energies for $\text{Cp}^*\text{Ir}(\mu\text{-N}^t\text{Bu})\text{MCp}_2$ (**2-M**). For each, the energy of orbital 105 is arbitrarily set to 0 eV because this has little contribution from the group 4 metal.

Table S4. Optimized geometries for Cp*Ir(μ -N^tBu)TiCp₂ (**2**-Ti) in Cartesian coordinates (Å).

Atom	B3LYP			PW91PW91		
	x	y	z	x	y	z
Ir	-0.7028	-0.0032	-0.0314	-0.6926	-0.0132	-0.0296
Ti	1.8416	-0.2914	-0.0018	1.8308	-0.2979	-0.0013
N	0.6851	1.2873	-0.0494	0.6811	1.2967	-0.0576
C	0.7075	2.7582	-0.0256	0.6785	2.7673	-0.0388
C	0.1881	3.2476	1.3427	0.1713	3.2437	1.3397
C	2.1449	3.2756	-0.2437	2.1051	3.3033	-0.2837
C	-0.1993	3.3132	-1.1421	-0.2630	3.2981	-1.1392
C	-2.3679	-1.0113	-1.1463	-2.3364	-1.0258	-1.1454
C	-2.3460	-1.4418	-2.5847	-2.3077	-1.4589	-2.5816
C	-2.9887	0.2234	-0.6677	-2.9729	0.2097	-0.6685
C	-3.6002	1.2686	-1.5535	-3.5928	1.2453	-1.5573
C	-2.9382	0.2118	0.7399	-2.9179	0.2045	0.7464
C	-3.4749	1.2489	1.6822	-3.4517	1.2440	1.6857
C	-2.2894	-1.0328	1.1562	-2.2544	-1.0385	1.1693
C	-2.1703	-1.4895	2.5816	-2.1249	-1.4839	2.5959
C	-2.0468	-1.8304	-0.0128	-2.0064	-1.8445	-0.0033
C	-1.5927	-3.2618	-0.0460	-1.5271	-3.2665	-0.0352
C	2.1619	0.1985	-2.3711	2.1535	0.2278	-2.3398
C	3.4615	0.0684	-1.8177	3.4543	0.0559	-1.7852
C	3.6378	-1.2813	-1.4262	3.5957	-1.3078	-1.4100
C	2.4285	-1.9684	-1.6748	2.3614	-1.9614	-1.6660
C	1.5143	-1.0542	-2.2759	1.4693	-1.0138	-2.2608
C	1.5126	-0.3115	2.4834	1.4942	-0.2677	2.4572
C	2.6844	0.4171	2.1903	2.6700	0.4599	2.1440
C	3.6268	-0.4763	1.6032	3.6150	-0.4473	1.5691
C	3.0269	-1.7577	1.5436	3.0128	-1.7362	1.5364
C	1.7083	-1.6467	2.0571	1.6913	-1.6155	2.0557
H	0.1769	4.3429	1.3875	0.1296	4.3443	1.3843
H	-0.8272	2.8778	1.5102	-0.8353	2.8387	1.5222
H	0.8214	2.8784	2.1549	0.8329	2.8895	2.1444
H	2.8199	2.9008	0.5321	2.8004	2.9378	0.4880
H	2.5321	2.9574	-1.2163	2.4825	2.9786	-1.2652
H	2.1724	4.3706	-0.2126	2.1192	4.4048	-0.2609
H	-0.2145	4.4093	-1.1214	-0.3176	4.3988	-1.1104
H	0.1581	2.9949	-2.1268	0.0921	2.9957	-2.1367
H	-1.2204	2.9447	-1.0175	-1.2721	2.8841	-0.9974
H	-3.3307	-1.8169	-2.8961	-3.2964	-1.8366	-2.8993
H	-2.0865	-0.6118	-3.2482	-2.0437	-0.6248	-3.2490
H	-1.6244	-2.2450	-2.7544	-1.5819	-2.2682	-2.7470
H	-2.9808	1.4619	-2.4343	-2.9656	1.4510	-2.4382
H	-3.7343	2.2177	-1.0279	-3.7473	2.1970	-1.0283
H	-4.5880	0.9496	-1.9134	-4.5779	0.9107	-1.9291
H	-3.6211	2.2112	1.1848	-3.6109	2.2070	1.1794
H	-4.4443	0.9409	2.0973	-4.4188	0.9325	2.1194
H	-2.7979	1.4111	2.5261	-2.7603	1.4181	2.5241
H	-1.8881	-0.6668	3.2450	-1.8371	-0.6501	3.2538
H	-1.4191	-2.2754	2.6899	-1.3681	-2.2733	2.7054
H	-3.1250	-1.8930	2.9463	-3.0818	-1.8873	2.9736
H	-1.0023	-3.4760	-0.9410	-0.9234	-3.4678	-0.9327
H	-0.9736	-3.5084	0.8207	-0.9034	-3.5026	0.8393
H	-2.4505	-3.9489	-0.0435	-2.3750	-3.9754	-0.0381
H	1.7179	1.1088	-2.7453	1.7291	1.1605	-2.7011
H	4.1977	0.8571	-1.7245	4.2125	0.8302	-1.6739
H	4.5210	-1.6940	-0.9593	4.4710	-1.7513	-0.9411
H	2.2326	-3.0116	-1.4598	2.1336	-3.0053	-1.4528
H	0.5172	-1.2815	-2.6164	0.4572	-1.2079	-2.5987
H	0.5932	0.0962	2.8752	0.5657	0.1501	2.8358
H	2.8427	1.4698	2.3793	2.8280	1.5233	2.3102
H	4.6310	-0.2240	1.2897	4.6218	-0.1982	1.2393
H	3.4779	-2.6575	1.1492	3.4631	-2.6479	1.1505
H	0.9833	-2.4462	2.1223	0.9603	-2.4175	2.1320

Table S5. Optimized geometries for Cp*Ir(μ -N^tBu)ZrCp₂ (**2-Zr**) in Cartesian coordinates (Å).

Atom	B3LYP			PW91PW91		
	x	y	z	x	y	z
Ir	-0.8453	-0.0284	-0.0261	-0.8465	-0.0264	-0.0357
Zr	1.7910	-0.1901	-0.0043	1.7700	-0.2156	-0.0055
N	0.4416	1.4060	0.0062	0.4275	1.4073	-0.0889
C	0.2800	2.8654	0.0557	0.2821	2.8690	-0.0854
C	1.6592	3.5419	-0.0841	1.6391	3.5149	-0.4357
C	-0.6337	3.3338	-1.0939	-0.7798	3.2957	-1.1176
C	-0.3495	3.2601	1.4077	-0.1577	3.3174	1.3253
C	-2.4062	-1.1862	-1.1403	-2.3833	-1.2781	-1.0572
C	-2.2905	-1.6851	-2.5522	-2.2693	-1.8746	-2.4296
C	-3.2120	-0.0220	-0.7457	-3.1883	-0.0845	-0.7397
C	-3.8825	0.9043	-1.7163	-3.8701	0.7654	-1.7686
C	-3.2233	0.0431	0.6497	-3.1863	0.0845	0.6575
C	-3.9106	1.0546	1.5190	-3.8523	1.1662	1.4533
C	-2.4159	-1.0758	1.1554	-2.3696	-0.9981	1.2377
C	-2.3016	-1.4397	2.6076	-2.2231	-1.2494	2.7098
C	-2.0546	-1.9147	0.0446	-2.0120	-1.9193	0.1801
C	-1.4989	-3.3085	0.1164	-1.4218	-3.2901	0.3436
C	2.6291	0.4719	2.3606	2.6655	0.6576	2.2432
C	1.5690	-0.4364	2.5948	1.5714	-0.1819	2.5941
C	1.9692	-1.7102	2.1147	1.9102	-1.5168	2.2371
C	3.2953	-1.5989	1.6142	3.2306	-1.5127	1.6922
C	3.6996	-0.2473	1.7494	3.6951	-0.1667	1.6851
C	2.7272	-1.7425	-1.8606	2.6261	-1.8613	-1.7722
C	3.7971	-0.8591	-1.5775	3.7369	-1.0064	-1.5270
C	3.4007	0.4475	-1.9705	3.3879	0.3030	-1.9708
C	2.1024	0.3608	-2.5398	2.0796	0.2443	-2.5351
C	1.6769	-0.9864	-2.4655	1.6005	-1.0876	-2.4092
H	2.3277	3.2285	0.7254	2.4105	3.2080	0.2900
H	2.1254	3.2747	-1.0384	1.9693	3.2045	-1.4393
H	1.5710	4.6335	-0.0445	1.5738	4.6148	-0.4211
H	-0.1898	3.0890	-2.0645	-0.4752	2.9941	-2.1320
H	-0.7932	4.4178	-1.0525	-0.9280	4.3879	-1.1060
H	-1.6020	2.8307	-1.0293	-1.7363	2.8011	-0.8911
H	-0.4765	4.3466	1.4816	-0.2902	4.4107	1.3688
H	-1.3279	2.7838	1.5171	-1.1089	2.8309	1.5890
H	0.2831	2.9298	2.2382	0.5930	3.0277	2.0769
H	-2.0847	-0.8686	-3.2506	-2.0814	-1.1027	-3.1912
H	-3.2216	-2.1704	-2.8773	-3.1988	-2.4009	-2.7134
H	-1.4874	-2.4199	-2.6507	-1.4498	-2.6060	-2.4807
H	-3.2187	1.1635	-2.5471	-3.2089	0.9701	-2.6247
H	-4.1971	1.8359	-1.2387	-4.1910	1.7320	-1.3547
H	-4.7774	0.4392	-2.1514	-4.7684	0.2611	-2.1672
H	-4.1952	1.9472	0.9555	-4.1468	2.0149	0.8192
H	-4.8263	0.6405	1.9626	-4.7635	0.7957	1.9558
H	-3.2707	1.3762	2.3470	-3.1852	1.5546	2.2387
H	-2.1558	-0.5524	3.2306	-2.1403	-0.3058	3.2694
H	-3.2116	-1.9445	2.9612	-3.0969	-1.7957	3.1092
H	-1.4597	-2.1133	2.7848	-1.3268	-1.8474	2.9278
H	-0.8767	-3.4493	1.0044	-0.7742	-3.3453	1.2306
H	-0.8846	-3.5430	-0.7569	-0.8129	-3.5731	-0.5278
H	-2.3079	-4.0518	0.1606	-2.2125	-4.0538	0.4605
H	2.6349	1.5223	2.6198	2.7185	1.7348	2.3910
H	0.6020	-0.1917	3.0080	0.6178	0.1466	2.9985
H	1.3730	-2.6138	2.1409	1.2739	-2.3921	2.3627
H	3.8830	-2.4009	1.1871	3.7774	-2.3827	1.3325
H	4.6618	0.1616	1.4678	4.6688	0.1744	1.3366
H	2.7190	-2.8107	-1.6800	2.5744	-2.9253	-1.5412
H	4.7379	-1.1239	-1.1140	4.6707	-1.2897	-1.0453
H	3.9986	1.3466	-1.8811	4.0195	1.1891	-1.9082
H	1.5184	1.1874	-2.9191	1.5185	1.0832	-2.9401
H	0.7318	-1.3728	-2.8170	0.6323	-1.4501	-2.7424

Table S6. Optimized geometries for Cp*Ir(μ -N^tBu)HfCp₂ (**2**-Hf) in Cartesian coordinates (Å).

Atom	B3LYP			PW91PW91		
	x	y	z	x	y	z
Ir	-1.01582	-0.03158	-0.01986	-1.0143	-0.0364	-0.0272
Hf	1.610589	-0.14913	-0.00317	1.5997	-0.1666	-0.0033
N	0.265094	1.429252	-0.05194	0.2496	1.4252	-0.0965
C	0.067979	2.884653	-0.05477	0.0641	2.8820	-0.1133
C	1.429749	3.589843	-0.22424	1.4061	3.5606	-0.4616
C	-0.85933	3.291143	-1.2172	-0.9997	3.2668	-1.1598
C	-0.5674	3.314273	1.284038	-0.3992	3.3392	1.2872
C	-2.57098	-1.22758	-1.09014	-2.5436	-1.2811	-1.0524
C	-2.45287	-1.76969	-2.48617	-2.4180	-1.8784	-2.4236
C	-3.41183	-0.07513	-0.72826	-3.3707	-0.0993	-0.7397
C	-4.10335	0.802824	-1.72836	-4.0534	0.7413	-1.7753
C	-3.41971	0.033279	0.6614	-3.3768	0.0714	0.6549
C	-4.12127	1.057663	1.503385	-4.0565	1.1469	1.4473
C	-2.57555	-1.04542	1.199556	-2.5444	-0.9981	1.2409
C	-2.44301	-1.35571	2.662761	-2.4018	-1.2438	2.7144
C	-2.203	-1.9141	0.11529	-2.1760	-1.9205	0.1879
C	-1.6272	-3.29756	0.22422	-1.5930	-3.2940	0.3557
C	2.60424	0.657641	2.260631	2.5170	0.7537	2.2150
C	1.506634	-0.16075	2.621265	1.4454	-0.1024	2.5921
C	1.803698	-1.49091	2.228599	1.8038	-1.4353	2.2427
C	3.1066	-1.50355	1.657589	3.1169	-1.4121	1.6818
C	3.598219	-0.17429	1.664462	3.5540	-0.0570	1.6501
C	2.390112	-1.79489	-1.8635	2.4867	-1.8332	-1.7392
C	3.566455	-1.07819	-1.53485	3.5933	-0.9735	-1.4948
C	3.363816	0.281599	-1.88707	3.2425	0.3329	-1.9454
C	2.078709	0.392833	-2.48897	1.9384	0.2662	-2.5213
C	1.47167	-0.8856	-2.46907	1.4624	-1.0671	-2.3879
H	2.107487	3.32533	0.594681	2.1791	3.2881	0.2760
H	1.901595	3.297054	-1.16829	1.7547	3.2426	-1.4566
H	1.314714	4.679621	-0.22745	1.3102	4.6583	-0.4667
H	-0.41804	3.009291	-2.17897	-0.6840	2.9458	-2.1648
H	-1.03251	4.373802	-1.22181	-1.1671	4.3562	-1.1749
H	-1.82113	2.779934	-1.12467	-1.9487	2.7609	-0.9265
H	-0.72479	4.398859	1.316877	-0.5601	4.4293	1.3143
H	-1.53133	2.814428	1.416018	-1.3401	2.8319	1.5492
H	0.077142	3.034102	2.123681	0.3517	3.0795	2.0496
H	-2.27255	-0.9717	-3.21288	-2.2326	-1.1055	-3.1850
H	-3.37417	-2.28749	-2.78828	-3.3415	-2.4129	-2.7118
H	-1.63278	-2.48773	-2.56672	-1.5919	-2.6026	-2.4694
H	-3.43687	1.070079	-2.5547	-3.3809	0.9676	-2.6173
H	-4.46228	1.731308	-1.27688	-4.4053	1.6972	-1.3615
H	-4.97126	0.292832	-2.16768	-4.9313	0.2187	-2.1950
H	-4.46242	1.908266	0.907449	-4.3697	1.9863	0.8095
H	-5.00274	0.629447	1.999467	-4.9581	0.7646	1.9581
H	-3.46787	1.446052	2.291542	-3.3916	1.5515	2.2264
H	-2.33414	-0.44223	3.254766	-2.3204	-0.2977	3.2701
H	-3.32807	-1.88886	3.037567	-3.2761	-1.7891	3.1144
H	-1.56944	-1.98215	2.856898	-1.5050	-1.8397	2.9355
H	-1.02352	-3.41388	1.1285	-0.9682	-3.3599	1.2582
H	-0.99025	-3.53685	-0.63174	-0.9656	-3.5744	-0.5033
H	-2.42619	-4.05192	0.263745	-2.3904	-4.0542	0.4483
H	2.686703	1.722561	2.433909	2.5502	1.8334	2.3485
H	0.578286	0.177517	3.057868	0.4917	0.2126	3.0068
H	1.159295	-2.35011	2.360187	1.1858	-2.3198	2.3874
H	3.622965	-2.37352	1.274289	3.6741	-2.2752	1.3220
H	4.565721	0.1501	1.303818	4.5145	0.2989	1.2809
H	2.226973	-2.8554	-1.71296	2.4358	-2.8951	-1.4995
H	4.446975	-1.48382	-1.05576	4.5236	-1.2494	-1.0026
H	4.080149	1.084305	-1.75987	3.8725	1.2201	-1.8859
H	1.629756	1.299594	-2.86907	1.3796	1.0994	-2.9406
H	0.490084	-1.12909	-2.84651	0.4977	-1.4359	-2.7243

Table S7. Optimized geometry for Cp*Ir(μ -N^tBu)(μ -H)ZrCp₂OTf (**3-Zr**) in Cartesian coordinates (Å), computed using the B3LYP functional.

Atom	x	y	z
Ir	1.9617	0.1618	0.0281
Zr	-0.8644	-1.0166	0.0836
H	1.1926	-1.2587	0.2340
N	0.2098	0.9115	-0.0331
O	-2.6946	0.2462	-0.1808
O	-4.4084	0.2426	-1.9855
O	-4.7531	-1.1485	0.0928
S	-4.1642	0.0297	-0.5564
C	-4.9084	1.4976	0.3119
F	-4.6391	1.4409	1.6285
F	-4.4034	2.6364	-0.1759
F	-6.2316	1.4985	0.1435
C	-0.2118	2.3261	-0.1791
C	0.9777	3.2961	-0.1642
C	-0.9494	2.4997	-1.5268
C	-1.1297	2.7012	1.0065
C	3.8982	0.0046	1.1630
C	3.9422	-0.9671	0.1100
C	3.8327	-0.2667	-1.1358
C	3.8599	1.1662	-0.8542
C	3.9019	1.3323	0.5565
C	4.0471	-0.2649	2.6321
C	4.0744	-2.4536	0.2824
C	3.8990	-0.8723	-2.5073
C	4.0043	2.2277	-1.9050
C	4.1089	2.6065	1.3208
C	-2.1115	-2.4742	1.8172
C	-2.2630	-1.1520	2.2944
C	-0.9867	-0.6600	2.6680
C	-0.0437	-1.6983	2.4563
C	-0.7314	-2.8156	1.9141
C	-1.1268	-3.2934	-1.1825
C	-2.1719	-2.4657	-1.6499
C	-1.5853	-1.4051	-2.4058
C	-0.1881	-1.5621	-2.3708
C	0.1059	-2.7215	-1.5894
H	1.5344	3.2137	0.7715
H	0.6153	4.3258	-0.2560
H	1.6534	3.0994	-0.9980
H	-0.2779	2.2483	-2.3554
H	-1.2620	3.5427	-1.6505
H	-1.8342	1.8686	-1.5884
H	-1.9972	2.0478	1.0680
H	-1.4837	3.7320	0.8942
H	-0.5716	2.6380	1.9474
H	3.6860	-1.2617	2.8969
H	3.4899	0.4615	3.2293
H	5.1013	-0.2051	2.9346
H	5.1314	-2.7484	0.2889
H	3.5870	-2.9994	-0.5294
H	3.6321	-2.7915	1.2229
H	4.9389	-0.9401	-2.8543
H	3.3478	-0.2715	-3.2353
H	3.4792	-1.8808	-2.5219
H	3.8329	3.2276	-1.5030
H	3.3089	2.0748	-2.7351
H	5.0202	2.2075	-2.3208
H	5.1608	2.6931	1.6240
H	3.5047	2.6364	2.2318
H	3.8625	3.4870	0.7256
H	-2.9103	-3.0988	1.4423
H	-3.1897	-0.5994	2.3159
H	-0.7691	0.3284	3.0465

H	1.0157	-1.6403	2.6572
H	-0.2869	-3.7694	1.6631
H	-1.2452	-4.1928	-0.5942
H	-3.2293	-2.6164	-1.4830
H	-2.1349	-0.6035	-2.8792
H	0.5402	-0.9018	-2.8185
H	1.0902	-3.1248	-1.3954

First 50 Excited States for 2-Ti as Predicted by TDDFT Using PW91PW91

Excited State	1:	Singlet-A	1.5373 eV	806.52 nm	f=0.0002	<S**2>=0.000
	106 ->108	-0.14151				
	107 ->108	0.69184				
Excited State	2:	Singlet-A	1.6050 eV	772.50 nm	f=0.0003	<S**2>=0.000
	106 ->108	0.69172				
	107 ->108	0.14184				
Excited State	3:	Singlet-A	2.0539 eV	603.67 nm	f=0.0007	<S**2>=0.000
	105 ->108	0.70520				
Excited State	4:	Singlet-A	2.2691 eV	546.40 nm	f=0.0006	<S**2>=0.000
	106 ->109	-0.28864				
	107 ->109	0.63201				
Excited State	5:	Singlet-A	2.3619 eV	524.92 nm	f=0.0020	<S**2>=0.000
	103 ->108	0.17116				
	104 ->108	0.46737				
	106 ->109	0.43745				
	107 ->109	0.21153				
Excited State	6:	Singlet-A	2.3914 eV	518.45 nm	f=0.0107	<S**2>=0.000
	103 ->108	0.11809				
	104 ->108	0.45828				
	106 ->109	-0.45436				
	107 ->109	-0.19820				
Excited State	7:	Singlet-A	2.4352 eV	509.13 nm	f=0.0018	<S**2>=0.000
	103 ->108	0.66766				
	104 ->108	-0.19293				
Excited State	8:	Singlet-A	2.6835 eV	462.02 nm	f=0.0030	<S**2>=0.000
	104 ->109	-0.11175				
	105 ->109	0.69646				
Excited State	9:	Singlet-A	2.8928 eV	428.60 nm	f=0.0004	<S**2>=0.000
	104 ->109	0.68365				
	105 ->109	0.10696				
Excited State	10:	Singlet-A	3.1211 eV	397.25 nm	f=0.0036	<S**2>=0.000
	103 ->109	-0.14699				

106	->110	-0.17887					
106	->111	0.14343					
107	->110	0.54712					
107	->111	0.31603					
107	->113	0.10073					
Excited State	11:	Singlet-A	3.1346 eV	395.54 nm	f=0.0012	<S**2>=0.000	
106	->110	0.30179					
107	->110	-0.21250					
107	->111	0.57806					
107	->112	0.12675					
Excited State	12:	Singlet-A	3.1759 eV	390.39 nm	f=0.0042	<S**2>=0.000	
103	->109	0.13159					
106	->110	0.48937					
106	->111	0.42238					
107	->110	0.17317					
107	->111	-0.15485					
Excited State	13:	Singlet-A	3.2214 eV	384.88 nm	f=0.0066	<S**2>=0.000	
103	->109	0.49583					
106	->110	-0.20079					
106	->112	0.12326					
107	->112	0.40023					
Excited State	14:	Singlet-A	3.2485 eV	381.67 nm	f=0.0041	<S**2>=0.000	
101	->108	0.19464					
102	->108	0.12309					
103	->109	-0.21270					
106	->110	0.17074					
106	->111	-0.29042					
106	->112	-0.16868					
107	->110	0.18502					
107	->111	-0.16578					
107	->112	0.41592					
Excited State	15:	Singlet-A	3.2644 eV	379.81 nm	f=0.0277	<S**2>=0.000	
101	->108	0.11441					
103	->109	-0.27777					
106	->110	-0.22423					
106	->111	0.39593					
106	->113	0.16009					
107	->110	-0.17980					

107 ->112	0.25210						
107 ->113	-0.22776						
Excited State 16:	Singlet-A	3.3028 eV	375.40 nm	f=0.0073	<S**2>=0.000		
101 ->108	-0.15738						
102 ->108	0.50442						
103 ->109	-0.11571						
106 ->112	0.42445						
Excited State 17:	Singlet-A	3.3331 eV	371.98 nm	f=0.0002	<S**2>=0.000		
101 ->108	0.10925						
102 ->108	0.45641						
103 ->109	0.14654						
106 ->112	-0.43166						
107 ->112	-0.19621						
Excited State 18:	Singlet-A	3.4348 eV	360.97 nm	f=0.0002	<S**2>=0.000		
101 ->108	0.62336						
106 ->112	0.24183						
107 ->112	-0.17727						
Excited State 19:	Singlet-A	3.4550 eV	358.85 nm	f=0.0057	<S**2>=0.000		
100 ->108	0.10911						
106 ->113	0.38659						
107 ->113	0.53079						
107 ->114	-0.14300						
Excited State 20:	Singlet-A	3.5339 eV	350.84 nm	f=0.0089	<S**2>=0.000		
100 ->108	0.51568						
102 ->108	0.10796						
106 ->113	-0.31460						
106 ->114	0.22178						
107 ->114	-0.24185						
Excited State 21:	Singlet-A	3.5666 eV	347.63 nm	f=0.0118	<S**2>=0.000		
100 ->108	0.34475						
105 ->110	0.43485						
106 ->113	0.19815						
106 ->114	-0.10793						
107 ->113	-0.14433						
107 ->114	0.27394						
107 ->115	0.13346						

Excited State 22: Singlet-A 3.5822 eV 346.11 nm f=0.0039 <S**2>=0.000
100 ->108 -0.25380
105 ->110 0.53496
105 ->111 -0.14537
106 ->113 -0.19460
106 ->114 0.14983
107 ->114 -0.19185

Excited State 23: Singlet-A 3.6170 eV 342.78 nm f=0.0019 <S**2>=0.000
105 ->110 0.10829
105 ->111 0.67776
107 ->114 -0.10848

Excited State 24: Singlet-A 3.6371 eV 340.89 nm f=0.0006 <S**2>=0.000
99 ->108 0.67005
106 ->114 0.10884

Excited State 25: Singlet-A 3.6742 eV 337.45 nm f=0.0077 <S**2>=0.000
99 ->108 -0.12753
104 ->112 -0.10869
106 ->114 0.49265
106 ->115 0.10260
107 ->113 0.10686
107 ->114 0.40657

Excited State 26: Singlet-A 3.6932 eV 335.71 nm f=0.0007 <S**2>=0.000
105 ->112 0.69613

Excited State 27: Singlet-A 3.7290 eV 332.49 nm f=0.0027 <S**2>=0.000
104 ->110 0.22994
104 ->111 0.10327
106 ->114 0.11711
107 ->114 -0.16500
107 ->115 0.60624

Excited State 28: Singlet-A 3.7383 eV 331.66 nm f=0.0002 <S**2>=0.000
104 ->110 0.64133
107 ->115 -0.21646

Excited State 29: Singlet-A 3.7735 eV 328.56 nm f=0.0005 <S**2>=0.000
104 ->111 0.65803
106 ->115 -0.18392
107 ->115 -0.12392

Excited State	30:	Singlet-A	3.7901 eV	327.13 nm	f=0.0049	<S**2>=0.000
	104 ->111	0.20700				
	106 ->114	-0.18801				
	106 ->115	0.62250				
Excited State	31:	Singlet-A	3.8409 eV	322.80 nm	f=0.0002	<S**2>=0.000
	98 ->108	-0.42066				
	107 ->116	0.55825				
Excited State	32:	Singlet-A	3.8464 eV	322.34 nm	f=0.0021	<S**2>=0.000
	98 ->108	0.53054				
	107 ->116	0.42955				
Excited State	33:	Singlet-A	3.8636 eV	320.90 nm	f=0.0006	<S**2>=0.000
	98 ->108	0.10819				
	102 ->109	-0.17275				
	104 ->112	0.65199				
Excited State	34:	Singlet-A	3.9070 eV	317.34 nm	f=0.0001	<S**2>=0.000
	103 ->110	-0.11012				
	105 ->113	0.44290				
	106 ->116	0.52529				
Excited State	35:	Singlet-A	3.9107 eV	317.04 nm	f=0.0024	<S**2>=0.000
	103 ->110	-0.10720				
	105 ->113	0.48977				
	105 ->114	-0.11906				
	106 ->116	-0.46731				
Excited State	36:	Singlet-A	3.9437 eV	314.38 nm	f=0.0039	<S**2>=0.000
	102 ->109	0.66467				
	104 ->112	0.15231				
Excited State	37:	Singlet-A	3.9778 eV	311.69 nm	f=0.0092	<S**2>=0.000
	101 ->109	0.42841				
	103 ->110	0.32762				
	103 ->111	0.20567				
	105 ->113	0.10310				
	106 ->113	0.13609				
	106 ->114	0.13704				
	107 ->113	-0.12510				
	107 ->114	-0.12607				

107 ->117	-0.10967						
Excited State 38:	Singlet-A	3.9803 eV	311.49 nm	f=0.0144	<S**2>=0.000		
101 ->109	-0.26090						
103 ->110	0.47441						
103 ->111	0.30876						
105 ->114	-0.10734						
106 ->113	-0.10625						
107 ->117	0.11517						
Excited State 39:	Singlet-A	4.0182 eV	308.56 nm	f=0.0036	<S**2>=0.000		
103 ->110	-0.27443						
103 ->111	0.52258						
103 ->112	0.10073						
104 ->113	0.15400						
107 ->117	-0.29351						
Excited State 40:	Singlet-A	4.0375 eV	307.08 nm	f=0.0020	<S**2>=0.000		
103 ->111	0.16923						
103 ->112	0.19463						
104 ->113	0.24219						
107 ->117	0.58756						
Excited State 41:	Singlet-A	4.0568 eV	305.62 nm	f=0.0047	<S**2>=0.000		
104 ->113	-0.47381						
105 ->114	0.48609						
Excited State 42:	Singlet-A	4.0856 eV	303.47 nm	f=0.0127	<S**2>=0.000		
103 ->112	0.20148						
104 ->113	0.21061						
105 ->114	0.29798						
106 ->117	0.49285						
107 ->117	-0.12939						
Excited State 43:	Singlet-A	4.1075 eV	301.85 nm	f=0.0005	<S**2>=0.000		
99 ->109	-0.14363						
103 ->112	0.43721						
104 ->113	-0.32017						
105 ->113	-0.10449						
105 ->114	-0.32019						
106 ->117	0.18815						
Excited State 44:	Singlet-A	4.1179 eV	301.09 nm	f=0.0225	<S**2>=0.000		

99 ->109	0.22404						
100 ->109	0.41467						
103 ->111	0.10165						
103 ->112	-0.25808						
106 ->117	0.35225						
Excited State 45:	Singlet-A	4.1351 eV	299.84 nm	f=0.0222	<S**2>=0.000		
99 ->109	-0.16166						
100 ->109	0.53556						
103 ->110	0.10294						
103 ->112	0.17949						
106 ->117	-0.27394						
Excited State 46:	Singlet-A	4.2030 eV	294.99 nm	f=0.0137	<S**2>=0.000		
105 ->115	0.67510						
Excited State 47:	Singlet-A	4.2142 eV	294.20 nm	f=0.0621	<S**2>=0.000		
99 ->109	0.22158						
101 ->109	0.22873						
104 ->114	0.56424						
105 ->115	-0.13497						
Excited State 48:	Singlet-A	4.3125 eV	287.50 nm	f=0.0483	<S**2>=0.000		
99 ->109	0.20603						
103 ->113	0.52348						
104 ->114	-0.21062						
106 ->118	0.10520						
107 ->119	0.14897						
Excited State 49:	Singlet-A	4.3280 eV	286.47 nm	f=0.0673	<S**2>=0.000		
99 ->109	-0.19136						
103 ->113	0.40011						
104 ->114	0.21784						
105 ->116	0.17521						
107 ->118	0.32187						
107 ->119	-0.15806						
Excited State 50:	Singlet-A	4.3353 eV	285.99 nm	f=0.0029	<S**2>=0.000		
99 ->109	0.10718						
105 ->116	0.67890						
107 ->118	-0.10301						

First 50 Excited States for 2-Ti as Predicted by TDDFT Using B3LYP

Excited State	1:	Singlet-A	1.6084 eV	770.86 nm	f=0.0013	<S**2>=0.000
	100 ->108	0.12298				
	106 ->108	-0.23154				
	107 ->108	0.63955				
Excited State	2:	Singlet-A	2.1848 eV	567.50 nm	f=0.0001	<S**2>=0.000
	106 ->108	0.65013				
	107 ->108	0.24778				
	107 ->109	-0.10102				
Excited State	3:	Singlet-A	2.3783 eV	521.31 nm	f=0.0003	<S**2>=0.000
	106 ->108	0.12071				
	106 ->109	-0.10839				
	107 ->109	0.67280				
Excited State	4:	Singlet-A	2.5023 eV	495.48 nm	f=0.0045	<S**2>=0.000
	103 ->109	0.26679				
	106 ->109	0.63247				
Excited State	5:	Singlet-A	2.6536 eV	467.23 nm	f=0.0027	<S**2>=0.000
	104 ->109	-0.22785				
	105 ->108	-0.19044				
	105 ->109	0.61149				
	105 ->113	0.10733				
Excited State	6:	Singlet-A	2.8174 eV	440.07 nm	f=0.0007	<S**2>=0.000
	104 ->108	-0.39068				
	104 ->109	0.50207				
	105 ->108	-0.14027				
	105 ->109	0.19049				
Excited State	7:	Singlet-A	2.8455 eV	435.72 nm	f=0.0006	<S**2>=0.000
	104 ->108	-0.20717				
	105 ->108	0.63361				
	105 ->109	0.19685				
Excited State	8:	Singlet-A	2.9071 eV	426.49 nm	f=0.0022	<S**2>=0.000
	104 ->108	0.50555				
	104 ->109	0.39535				
	105 ->108	0.15936				
	105 ->109	0.13145				

Excited State	9:	Singlet-A	3.0347 eV	408.56 nm	f=0.0003	<S**2>=0.000
	103 ->108	0.12436				
	106 ->112	-0.22029				
	107 ->112	0.59144				
	107 ->113	0.14493				
	107 ->114	-0.10261				
Excited State	10:	Singlet-A	3.1220 eV	397.13 nm	f=0.0005	<S**2>=0.000
	103 ->108	0.67977				
	107 ->112	-0.11635				
Excited State	11:	Singlet-A	3.2880 eV	377.08 nm	f=0.0002	<S**2>=0.000
	106 ->113	-0.17501				
	107 ->112	-0.15160				
	107 ->113	0.61198				
Excited State	12:	Singlet-A	3.3668 eV	368.25 nm	f=0.0066	<S**2>=0.000
	103 ->109	0.51164				
	106 ->109	-0.21307				
	106 ->113	0.30166				
	107 ->110	0.16001				
	107 ->114	0.12862				
Excited State	13:	Singlet-A	3.5166 eV	352.57 nm	f=0.0075	<S**2>=0.000
	106 ->110	-0.14860				
	106 ->112	-0.14466				
	106 ->113	-0.15742				
	106 ->115	0.13769				
	107 ->110	0.40370				
	107 ->114	0.10067				
	107 ->115	-0.36135				
	107 ->116	0.20406				
Excited State	14:	Singlet-A	3.5729 eV	347.02 nm	f=0.0082	<S**2>=0.000
	106 ->110	0.43617				
	106 ->111	-0.31990				
	106 ->114	0.32727				
	107 ->110	0.20467				
	107 ->111	-0.11982				
Excited State	15:	Singlet-A	3.6814 eV	336.79 nm	f=0.0026	<S**2>=0.000
	106 ->111	-0.11049				
	106 ->112	0.52723				

107 ->111	0.22534
107 ->112	0.20741
107 ->114	-0.14793
107 ->115	-0.19548
107 ->116	0.12229

Excited State 16: Singlet-A 3.7972 eV 326.51 nm f=0.0029 <S**2>=0.000

99 ->108	-0.10569
102 ->108	-0.10555
106 ->110	0.13369
106 ->111	0.23685
106 ->114	-0.12772
106 ->115	-0.15828
106 ->116	0.10062
107 ->110	0.45284
107 ->114	-0.26134
107 ->115	0.20875

Excited State 17: Singlet-A 3.8220 eV 324.40 nm f=0.0086 <S**2>=0.000

99 ->108	-0.17984
100 ->108	-0.15890
102 ->108	-0.25432
106 ->110	0.41380
106 ->111	0.18910
106 ->112	-0.12043
106 ->114	-0.15077
107 ->110	-0.19577
107 ->111	0.11724
107 ->114	0.11062
107 ->115	-0.19762
107 ->116	0.10048

Excited State 18: Singlet-A 3.8473 eV 322.27 nm f=0.0102 <S**2>=0.000

100 ->108	-0.11388
102 ->108	-0.10109
103 ->109	-0.27319
106 ->110	-0.11809
106 ->112	-0.11822
106 ->113	0.51769
107 ->113	0.21612

Excited State 19: Singlet-A 3.8549 eV 321.63 nm f=0.0024 <S**2>=0.000

99 ->108	0.30843
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100 ->108	0.10497						
102 ->108	0.46134						
106 ->110	0.24344						
106 ->111	0.15765						
106 ->113	0.16705						
106 ->114	-0.14835						
107 ->111	0.11453						
Excited State 20:	Singlet-A	3.9769 eV	311.76 nm	f=0.0075	<S**2>=0.000		
100 ->108	0.40043						
101 ->108	0.38561						
102 ->108	-0.14062						
104 ->110	0.16704						
104 ->111	-0.20788						
104 ->114	0.23212						
Excited State 21:	Singlet-A	4.0036 eV	309.68 nm	f=0.0012	<S**2>=0.000		
99 ->108	0.45223						
100 ->108	0.20952						
101 ->108	-0.17226						
102 ->108	-0.37208						
104 ->110	-0.10401						
104 ->111	0.12337						
104 ->114	-0.12207						
Excited State 22:	Singlet-A	4.0236 eV	308.14 nm	f=0.0174	<S**2>=0.000		
99 ->108	-0.34703						
100 ->108	0.42728						
101 ->108	-0.18313						
104 ->110	-0.14813						
104 ->111	0.15251						
104 ->114	-0.18486						
107 ->108	-0.10367						
107 ->111	0.10629						
Excited State 23:	Singlet-A	4.0352 eV	307.26 nm	f=0.0071	<S**2>=0.000		
101 ->108	0.51213						
104 ->110	-0.22645						
104 ->111	0.24842						
104 ->114	-0.27334						
Excited State 24:	Singlet-A	4.0826 eV	303.69 nm	f=0.0437	<S**2>=0.000		
100 ->108	0.12461						

105 ->110	0.37487					
105 ->111	-0.30914					
105 ->114	0.28512					
106 ->112	0.11765					
107 ->111	-0.27445					
Excited State 25:	Singlet-A	4.1260 eV	300.50 nm	f=0.0118	<S**2>=0.000	
105 ->110	0.19116					
105 ->111	-0.13731					
105 ->114	0.10749					
106 ->114	0.19980					
106 ->115	-0.18178					
107 ->111	0.45112					
107 ->114	0.26462					
107 ->115	0.11247					
Excited State 26:	Singlet-A	4.1466 eV	299.00 nm	f=0.0189	<S**2>=0.000	
106 ->111	0.33461					
106 ->112	0.14320					
106 ->114	0.33701					
106 ->115	-0.28330					
106 ->116	0.13886					
107 ->111	-0.19825					
107 ->115	-0.20348					
Excited State 27:	Singlet-A	4.2198 eV	293.81 nm	f=0.0035	<S**2>=0.000	
99 ->109	0.10364					
100 ->109	-0.28221					
101 ->109	0.32052					
104 ->112	0.16169					
104 ->113	0.21412					
105 ->112	0.32968					
105 ->113	0.20079					
106 ->111	-0.12458					
106 ->115	-0.10584					
Excited State 28:	Singlet-A	4.2614 eV	290.94 nm	f=0.0004	<S**2>=0.000	
98 ->108	0.10179					
104 ->112	0.57938					
104 ->113	0.11625					
104 ->114	-0.13523					
105 ->112	-0.10597					
106 ->111	0.15049					

106	->114	0.17086						
106	->115	0.12787						
Excited State	29:	Singlet-A	4.2663 eV	290.61 nm	f=0.0191	<S**2>=0.000		
98	->108	0.47636						
101	->109	-0.13001						
104	->113	-0.10991						
106	->111	-0.19640						
106	->115	-0.24529						
106	->116	0.14312						
107	->114	-0.17997						
107	->115	-0.11970						
Excited State	30:	Singlet-A	4.2864 eV	289.25 nm	f=0.0008	<S**2>=0.000		
98	->108	0.36645						
101	->109	0.11653						
104	->112	-0.28405						
104	->113	0.38666						
106	->111	0.12411						
106	->114	0.13658						
106	->115	0.17291						
Excited State	31:	Singlet-A	4.2935 eV	288.78 nm	f=0.0095	<S**2>=0.000		
98	->108	0.18765						
104	->113	-0.28393						
105	->112	0.43770						
106	->111	0.15955						
106	->114	0.11261						
106	->115	0.25946						
106	->116	-0.14065						
Excited State	32:	Singlet-A	4.3233 eV	286.78 nm	f=0.0011	<S**2>=0.000		
100	->109	0.19634						
101	->109	-0.27310						
104	->113	0.22673						
105	->113	0.53854						
Excited State	33:	Singlet-A	4.3313 eV	286.25 nm	f=0.0013	<S**2>=0.000		
98	->108	-0.11524						
100	->109	0.20505						
101	->109	-0.22723						
104	->113	0.28514						
105	->112	0.36821						

105 ->113	-0.32257						
106 ->115	-0.11358						
Excited State 34:	Singlet-A	4.3674 eV	283.88 nm	f=0.0966	<S**2>=0.000		
98 ->108	-0.20825						
102 ->109	0.39958						
103 ->112	0.10129						
105 ->114	0.15359						
106 ->112	-0.10810						
106 ->114	0.20000						
107 ->114	-0.31665						
Excited State 35:	Singlet-A	4.4089 eV	281.21 nm	f=0.0170	<S**2>=0.000		
105 ->110	0.52122						
105 ->111	0.25786						
105 ->114	-0.29298						
107 ->114	-0.12569						
Excited State 36:	Singlet-A	4.4366 eV	279.46 nm	f=0.0022	<S**2>=0.000		
103 ->110	0.44383						
103 ->111	-0.34763						
103 ->112	0.10101						
103 ->114	0.32213						
104 ->113	0.14596						
Excited State 37:	Singlet-A	4.4688 eV	277.45 nm	f=0.0032	<S**2>=0.000		
102 ->109	-0.10476						
104 ->110	0.53189						
104 ->111	0.21161						
104 ->114	-0.23706						
104 ->115	-0.10223						
107 ->115	-0.13292						
107 ->116	-0.21676						
Excited State 38:	Singlet-A	4.5070 eV	275.09 nm	f=0.0002	<S**2>=0.000		
104 ->110	0.21200						
107 ->115	0.29509						
107 ->116	0.57228						
Excited State 39:	Singlet-A	4.5493 eV	272.53 nm	f=0.0660	<S**2>=0.000		
102 ->109	-0.30457						
103 ->112	0.55442						
103 ->114	-0.15246						

Excited State 40:	Singlet-A	4.5739 eV	271.07 nm	f=0.0005	<S**2>=0.000
106 ->115	0.32370				
106 ->116	0.60718				
Excited State 41:	Singlet-A	4.6536 eV	266.42 nm	f=0.1929	<S**2>=0.000
102 ->109	0.31599				
103 ->109	-0.13367				
103 ->112	0.17673				
103 ->113	0.36296				
105 ->111	0.20628				
105 ->115	-0.11318				
107 ->114	0.15720				
107 ->117	0.18613				
Excited State 42:	Singlet-A	4.6673 eV	265.65 nm	f=0.0097	<S**2>=0.000
105 ->111	-0.10411				
107 ->117	0.65570				
Excited State 43:	Singlet-A	4.7003 eV	263.78 nm	f=0.0455	<S**2>=0.000
102 ->109	-0.13365				
103 ->113	-0.22710				
105 ->111	0.31117				
105 ->114	0.36263				
105 ->115	-0.31665				
105 ->116	0.16755				
106 ->117	-0.15272				
Excited State 44:	Singlet-A	4.7284 eV	262.21 nm	f=0.0131	<S**2>=0.000
100 ->109	0.24487				
101 ->109	0.23260				
103 ->110	0.39429				
103 ->111	0.26068				
103 ->113	-0.15935				
103 ->114	-0.21070				
103 ->115	-0.12044				
104 ->115	-0.11808				
Excited State 45:	Singlet-A	4.7317 eV	262.03 nm	f=0.0337	<S**2>=0.000
100 ->109	0.14514				
101 ->109	0.15033				
102 ->109	-0.11790				
103 ->112	-0.13521				

103 ->113	0.22514
104 ->111	-0.21960
104 ->114	-0.21149
104 ->115	0.29793
104 ->116	-0.16144
106 ->117	0.25672

Excited State 46: Singlet-A 4.7668 eV 260.10 nm f=0.0221 <S**2>=0.000

99 ->109	0.12131
100 ->109	0.32282
101 ->109	0.21732
103 ->110	-0.24941
103 ->111	-0.15587
103 ->112	0.10200
103 ->113	-0.31929
103 ->114	0.19455
106 ->117	0.22624

Excited State 47: Singlet-A 4.7784 eV 259.47 nm f=0.0912 <S**2>=0.000

99 ->109	-0.19318
100 ->109	0.22187
101 ->109	0.24755
102 ->109	-0.12287
103 ->110	-0.16081
103 ->112	-0.16385
103 ->113	0.25366
103 ->114	0.10117
104 ->111	0.11483
104 ->114	0.16613
104 ->115	-0.15838
106 ->117	-0.20584
107 ->114	-0.11010

Excited State 48: Singlet-A 4.7804 eV 259.36 nm f=0.0742 <S**2>=0.000

100 ->109	-0.11254
102 ->109	-0.10314
104 ->111	0.18341
104 ->114	0.20454
104 ->115	-0.18543
104 ->116	0.10397
106 ->117	0.52767
107 ->114	-0.10147

Excited State 49:	Singlet-A	4.8718 eV	254.49 nm	f=0.0075	<S**2>=0.000
99 ->109	0.43233				
105 ->111	-0.23795				
105 ->114	-0.14505				
105 ->115	-0.35296				
105 ->116	0.19050				

Excited State 50:	Singlet-A	4.9028 eV	252.88 nm	f=0.0112	<S**2>=0.000
98 ->109	0.11993				
99 ->109	0.37548				
102 ->112	0.11299				
103 ->112	-0.12137				
105 ->111	0.22907				
105 ->114	0.19138				
105 ->115	0.31891				
105 ->116	-0.16944				
106 ->117	-0.10594				

First 20 Excited States for 2-Zr as Predicted by TDDFT Using PW91PW91

Excited State	1:	Singlet-A	1.8029 eV	687.71 nm	f=0.0001	<S**2>=0.000
107 ->108		0.70615				
Excited State	2:	Singlet-A	1.9644 eV	631.16 nm	f=0.0001	<S**2>=0.000
106 ->108		0.70256				
Excited State	3:	Singlet-A	2.2538 eV	550.11 nm	f=0.0025	<S**2>=0.000
104 ->108		-0.10781				
105 ->108		0.69737				
Excited State	4:	Singlet-A	2.3286 eV	532.44 nm	f=0.0011	<S**2>=0.000
107 ->109		0.69612				
Excited State	5:	Singlet-A	2.5212 eV	491.76 nm	f=0.0054	<S**2>=0.000
104 ->108		-0.11420				
106 ->109		0.68668				
Excited State	6:	Singlet-A	2.6771 eV	463.13 nm	f=0.0089	<S**2>=0.000
104 ->108		0.64380				
105 ->109		0.15389				
106 ->109		0.11820				
107 ->111		0.11493				
Excited State	7:	Singlet-A	2.6964 eV	459.82 nm	f=0.0037	<S**2>=0.000
103 ->108		0.13411				
104 ->108		-0.13650				
104 ->109		-0.13469				
105 ->109		0.65938				
Excited State	8:	Singlet-A	2.7083 eV	457.80 nm	f=0.0000	<S**2>=0.000
103 ->108		0.68943				
105 ->109		-0.12431				
Excited State	9:	Singlet-A	2.9301 eV	423.14 nm	f=0.0003	<S**2>=0.000
104 ->109		0.68648				
105 ->109		0.13589				
Excited State	10:	Singlet-A	3.1867 eV	389.07 nm	f=0.0028	<S**2>=0.000
107 ->110		0.68879				
107 ->111		-0.12987				

Excited State	11:	Singlet-A	3.3291 eV	372.43 nm	f=0.0007	<S**2>=0.000
	106 ->110	0.70432				
Excited State	12:	Singlet-A	3.3620 eV	368.78 nm	f=0.0220	<S**2>=0.000
	101 ->109	0.10838				
	102 ->108	-0.41093				
	103 ->109	0.48441				
	107 ->111	0.26446				
Excited State	13:	Singlet-A	3.3969 eV	364.99 nm	f=0.0007	<S**2>=0.000
	102 ->108	0.56555				
	103 ->109	0.36973				
	107 ->111	0.15243				
Excited State	14:	Singlet-A	3.5553 eV	348.73 nm	f=0.0433	<S**2>=0.000
	103 ->109	-0.19545				
	106 ->111	0.15004				
	107 ->111	0.47040				
	107 ->112	-0.37427				
	107 ->113	0.17595				
Excited State	15:	Singlet-A	3.5916 eV	345.20 nm	f=0.0015	<S**2>=0.000
	101 ->108	-0.12859				
	106 ->111	0.66250				
Excited State	16:	Singlet-A	3.6093 eV	343.51 nm	f=0.0014	<S**2>=0.000
	105 ->110	0.69785				
Excited State	17:	Singlet-A	3.6430 eV	340.34 nm	f=0.0017	<S**2>=0.000
	101 ->108	0.68033				
	106 ->111	0.11103				
Excited State	18:	Singlet-A	3.6969 eV	335.38 nm	f=0.0054	<S**2>=0.000
	107 ->111	0.12852				
	107 ->112	0.46579				
	107 ->113	0.49221				
Excited State	19:	Singlet-A	3.7578 eV	329.94 nm	f=0.0045	<S**2>=0.000
	102 ->109	0.65793				
	107 ->113	0.11774				
	107 ->114	-0.14731				
Excited State	20:	Singlet-A	3.7850 eV	327.57 nm	f=0.0102	<S**2>=0.000

100 ->108	0.57681
102 ->109	-0.14242
107 ->112	-0.13222
107 ->113	0.26386

First 20 Excited States for 2-Zr as Predicted by TDDFT Using B3LYP

Excited State	1:	Singlet-A	2.0331 eV	609.83 nm	f=0.0012	<S**2>=0.000
	107 ->108	0.69299				
Excited State	2:	Singlet-A	2.3438 eV	528.98 nm	f=0.0002	<S**2>=0.000
	106 ->108	0.64818				
	107 ->109	-0.24432				
Excited State	3:	Singlet-A	2.4725 eV	501.46 nm	f=0.0009	<S**2>=0.000
	106 ->108	0.21829				
	106 ->109	-0.18077				
	107 ->109	0.63592				
Excited State	4:	Singlet-A	2.5332 eV	489.43 nm	f=0.0028	<S**2>=0.000
	103 ->109	0.26388				
	106 ->108	0.15867				
	106 ->109	0.60053				
	107 ->109	0.15641				
Excited State	5:	Singlet-A	2.5997 eV	476.92 nm	f=0.0031	<S**2>=0.000
	104 ->109	-0.26762				
	105 ->108	-0.22573				
	105 ->109	0.58638				
Excited State	6:	Singlet-A	2.7907 eV	444.28 nm	f=0.0028	<S**2>=0.000
	104 ->108	-0.13392				
	104 ->109	0.53707				
	105 ->108	-0.40466				
	105 ->109	0.10485				
Excited State	7:	Singlet-A	2.8073 eV	441.65 nm	f=0.0011	<S**2>=0.000
	104 ->108	-0.25474				
	104 ->109	0.24269				
	105 ->108	0.49991				
	105 ->109	0.33618				
Excited State	8:	Singlet-A	2.9927 eV	414.29 nm	f=0.0058	<S**2>=0.000
	104 ->108	0.61311				
	104 ->109	0.25202				
	105 ->108	0.13258				
	107 ->112	0.10580				

Excited State	9:	Singlet-A	3.1539 eV	393.12 nm	f=0.0007	<S**2>=0.000
	103 ->108	0.69798				
Excited State	10:	Singlet-A	3.6362 eV	340.97 nm	f=0.0307	<S**2>=0.000
	103 ->109	0.58154				
	106 ->109	-0.26920				
	107 ->110	0.13754				
	107 ->111	-0.15150				
	107 ->112	0.11368				
Excited State	11:	Singlet-A	3.7702 eV	328.86 nm	f=0.0111	<S**2>=0.000
	103 ->109	-0.12042				
	107 ->110	0.66918				
	107 ->111	-0.10610				
Excited State	12:	Singlet-A	3.8989 eV	318.00 nm	f=0.0108	<S**2>=0.000
	102 ->108	0.60067				
	102 ->109	-0.11388				
	106 ->110	0.23147				
	106 ->111	-0.16017				
	106 ->112	0.11319				
Excited State	13:	Singlet-A	3.9162 eV	316.59 nm	f=0.0036	<S**2>=0.000
	102 ->108	-0.31993				
	106 ->110	0.49452				
	106 ->111	-0.28697				
	106 ->112	0.19151				
	106 ->113	-0.10654				
Excited State	14:	Singlet-A	3.9898 eV	310.75 nm	f=0.0008	<S**2>=0.000
	101 ->108	0.63700				
	102 ->109	0.13276				
	106 ->110	0.10991				
	107 ->115	0.19599				
Excited State	15:	Singlet-A	4.0385 eV	307.00 nm	f=0.0016	<S**2>=0.000
	100 ->109	-0.10087				
	101 ->108	-0.16335				
	102 ->109	0.24665				
	106 ->110	0.39458				
	106 ->111	0.30783				
	106 ->112	-0.26195				
	106 ->113	0.14635				

107 ->115	0.12658						
Excited State 16:	Singlet-A	4.0546 eV	305.78 nm	f=0.0498	<S**2>=0.000		
106 ->115	0.14837						
107 ->110	0.13883						
107 ->111	0.62989						
Excited State 17:	Singlet-A	4.0828 eV	303.68 nm	f=0.0030	<S**2>=0.000		
100 ->108	0.17895						
100 ->109	-0.21038						
101 ->108	-0.12682						
102 ->109	0.51769						
106 ->110	-0.13938						
106 ->111	-0.23096						
106 ->112	0.16976						
Excited State 18:	Singlet-A	4.1090 eV	301.74 nm	f=0.0001	<S**2>=0.000		
99 ->108	-0.16710						
101 ->108	-0.11845						
102 ->109	-0.13173						
106 ->111	-0.12455						
107 ->112	0.12756						
107 ->114	0.29516						
107 ->115	0.50562						
Excited State 19:	Singlet-A	4.1720 eV	297.18 nm	f=0.0044	<S**2>=0.000		
100 ->108	0.64752						
100 ->109	0.10336						
102 ->109	-0.13426						
Excited State 20:	Singlet-A	4.2329 eV	292.90 nm	f=0.0168	<S**2>=0.000		
105 ->110	0.35518						
105 ->111	-0.25016						
105 ->112	0.14153						
106 ->114	-0.10387						
106 ->115	-0.13552						
107 ->111	0.10370						
107 ->112	0.36268						
107 ->113	-0.13230						
107 ->115	-0.19735						

First 20 Excited States for 2-Hf as Predicted by TDDFT Using PW91PW91

Excited State	1:	Singlet-A	2.0081 eV	617.43 nm	f=0.0005	<S**2>=0.000
	107 ->108	0.70256				
Excited State	2:	Singlet-A	2.1266 eV	583.02 nm	f=0.0001	<S**2>=0.000
	106 ->108	0.68780				
	107 ->109	-0.12422				
Excited State	3:	Singlet-A	2.3531 eV	526.91 nm	f=0.0009	<S**2>=0.000
	106 ->108	0.13352				
	107 ->109	0.68150				
Excited State	4:	Singlet-A	2.4323 eV	509.75 nm	f=0.0027	<S**2>=0.000
	104 ->108	-0.10122				
	105 ->108	0.69466				
Excited State	5:	Singlet-A	2.4863 eV	498.67 nm	f=0.0051	<S**2>=0.000
	106 ->109	0.69039				
Excited State	6:	Singlet-A	2.6729 eV	463.86 nm	f=0.0028	<S**2>=0.000
	104 ->109	-0.13966				
	105 ->109	0.68577				
Excited State	7:	Singlet-A	2.8636 eV	432.97 nm	f=0.0087	<S**2>=0.000
	104 ->108	0.63846				
	104 ->109	-0.19259				
	107 ->111	-0.12640				
Excited State	8:	Singlet-A	2.8835 eV	429.98 nm	f=0.0009	<S**2>=0.000
	103 ->108	0.68417				
	104 ->109	-0.11810				
Excited State	9:	Singlet-A	2.9370 eV	422.14 nm	f=0.0013	<S**2>=0.000
	103 ->108	0.13524				
	104 ->108	0.17186				
	104 ->109	0.64890				
	105 ->109	0.11999				
Excited State	10:	Singlet-A	3.2032 eV	387.06 nm	f=0.0019	<S**2>=0.000
	107 ->110	0.69553				
Excited State	11:	Singlet-A	3.2989 eV	375.84 nm	f=0.0004	<S**2>=0.000

106 ->110	0.70582						
Excited State 12:	Singlet-A	3.3605 eV	368.94 nm	f=0.0251	<S**2>=0.000		
101 ->109	0.15150						
103 ->109	0.61957						
107 ->111	-0.25212						
Excited State 13:	Singlet-A	3.5712 eV	347.18 nm	f=0.0044	<S**2>=0.000		
102 ->108	0.62944						
105 ->110	-0.12836						
107 ->111	0.20701						
107 ->112	0.12846						
Excited State 14:	Singlet-A	3.5859 eV	345.76 nm	f=0.0217	<S**2>=0.000		
102 ->108	-0.19154						
105 ->110	-0.22307						
106 ->111	0.46149						
107 ->111	0.33666						
107 ->112	0.24345						
Excited State 15:	Singlet-A	3.6002 eV	344.38 nm	f=0.0014	<S**2>=0.000		
102 ->108	0.11179						
105 ->110	0.63246						
106 ->111	0.26930						
Excited State 16:	Singlet-A	3.6160 eV	342.88 nm	f=0.0191	<S**2>=0.000		
102 ->108	0.19535						
105 ->110	-0.16945						
106 ->111	0.42713						
107 ->111	-0.31051						
107 ->112	-0.34527						
Excited State 17:	Singlet-A	3.7536 eV	330.31 nm	f=0.0044	<S**2>=0.000		
102 ->109	0.67665						
107 ->112	-0.11488						
Excited State 18:	Singlet-A	3.8064 eV	325.73 nm	f=0.0108	<S**2>=0.000		
101 ->108	0.55000						
107 ->111	-0.17445						
107 ->112	0.31802						
107 ->113	0.16235						
Excited State 19:	Singlet-A	3.8231 eV	324.30 nm	f=0.0134	<S**2>=0.000		

101 ->108	-0.24266
101 ->109	-0.13761
102 ->109	0.14400
106 ->112	0.47988
107 ->111	-0.17059
107 ->112	0.28753
107 ->113	0.16209

Excited State 20: Singlet-A 3.8371 eV 323.12 nm f=0.0092 <S**2>=0.000

101 ->108	0.33472
101 ->109	0.12707
105 ->111	-0.11813
106 ->111	-0.10895
106 ->112	0.48011
107 ->111	0.13416
107 ->112	-0.20298
107 ->113	-0.17970

First 20 Excited States for 2-Hf as Predicted by TDDFT Using B3LYP

Excited State	1:	Singlet-A	2.3167 eV	535.18 nm	f=0.0030	<S**2>=0.000
107 ->108		0.69177				
Excited State	2:	Singlet-A	2.4497 eV	506.12 nm	f=0.0005	<S**2>=0.000
103 ->109		0.10480				
106 ->108		-0.33998				
106 ->109		0.14689				
107 ->109		0.57584				
Excited State	3:	Singlet-A	2.4894 eV	498.05 nm	f=0.0025	<S**2>=0.000
103 ->109		0.26086				
106 ->109		0.60195				
107 ->109		-0.22745				
Excited State	4:	Singlet-A	2.5740 eV	481.68 nm	f=0.0011	<S**2>=0.000
104 ->109		-0.21416				
105 ->109		0.51781				
106 ->108		0.31918				
106 ->109		0.10704				
107 ->109		0.21680				
Excited State	5:	Singlet-A	2.5994 eV	476.97 nm	f=0.0025	<S**2>=0.000
105 ->109		-0.38333				
106 ->108		0.51637				
107 ->109		0.24227				
Excited State	6:	Singlet-A	2.8220 eV	439.34 nm	f=0.0001	<S**2>=0.000
104 ->108		-0.12250				
104 ->109		0.64292				
105 ->109		0.21798				
Excited State	7:	Singlet-A	2.9693 eV	417.55 nm	f=0.0032	<S**2>=0.000
104 ->108		-0.14536				
105 ->108		0.66998				
105 ->109		0.12301				
Excited State	8:	Singlet-A	3.2085 eV	386.42 nm	f=0.0085	<S**2>=0.000
104 ->108		0.64916				
104 ->109		0.12499				
105 ->108		0.13260				
107 ->112		0.12889				

Excited State	9:	Singlet-A	3.3210 eV	373.34 nm	f=0.0004	<S**2>=0.000
	103 ->108	0.69828				
Excited State	10:	Singlet-A	3.6559 eV	339.13 nm	f=0.0458	<S**2>=0.000
	103 ->109	0.59616				
	106 ->109	-0.27563				
	107 ->111	-0.13409				
	107 ->112	0.12655				
Excited State	11:	Singlet-A	3.8097 eV	325.44 nm	f=0.0072	<S**2>=0.000
	107 ->110	0.67847				
	107 ->111	0.13836				
Excited State	12:	Singlet-A	3.9540 eV	313.56 nm	f=0.0007	<S**2>=0.000
	102 ->109	0.11172				
	106 ->110	0.65188				
	106 ->111	0.21113				
Excited State	13:	Singlet-A	4.0254 eV	308.01 nm	f=0.0017	<S**2>=0.000
	100 ->109	-0.19499				
	102 ->108	-0.20443				
	102 ->109	0.59630				
	106 ->110	-0.16566				
Excited State	14:	Singlet-A	4.1251 eV	300.56 nm	f=0.0072	<S**2>=0.000
	102 ->108	0.38419				
	106 ->110	-0.13497				
	106 ->111	0.34572				
	106 ->112	-0.26456				
	107 ->111	0.31638				
Excited State	15:	Singlet-A	4.1347 eV	299.86 nm	f=0.0046	<S**2>=0.000
	102 ->108	0.46001				
	102 ->109	0.23936				
	106 ->110	0.13185				
	106 ->111	-0.32955				
	106 ->112	0.25488				
Excited State	16:	Singlet-A	4.1562 eV	298.31 nm	f=0.0474	<S**2>=0.000
	101 ->108	-0.11305				
	102 ->108	-0.24976				
	106 ->111	-0.16524				

106 ->112	0.11842						
107 ->110	-0.12442						
107 ->111	0.56365						
Excited State 17:	Singlet-A	4.1933 eV	295.67 nm	f=0.0019	<S**2>=0.000		
101 ->108	0.67730						
Excited State 18:	Singlet-A	4.2971 eV	288.53 nm	f=0.0063	<S**2>=0.000		
101 ->109	0.10301						
105 ->110	0.61149						
105 ->111	0.28187						
107 ->112	-0.13130						
Excited State 19:	Singlet-A	4.3955 eV	282.07 nm	f=0.0126	<S**2>=0.000		
100 ->108	0.26634						
101 ->109	-0.25536						
105 ->110	0.13995						
106 ->115	-0.23070						
107 ->112	0.48813						
Excited State 20:	Singlet-A	4.4277 eV	280.02 nm	f=0.0065	<S**2>=0.000		
99 ->108	0.21746						
100 ->108	0.45435						
104 ->110	-0.13203						
106 ->111	0.23066						
106 ->112	0.25075						
107 ->112	-0.11107						
107 ->115	-0.21860						

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