Mimics of Small Ribozymes Utilizing a Supramolecular Scaffold

Supporting Information

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Contents

Synthetic details for compunds 5, 6, 7, 2h, 2i, 2n, 2o and 8	<u>S4</u>
¹ H NMR spectrum of N -[2-(N^4 , N^7 , N^{10} -Tri- <i>tert</i> -butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-4-nitrobenzenesulfonamide (5)	<u>S8</u>
13 C NMR spectrum of <i>N</i> -[2-(N^4 , N^7 , N^{10} -Tri- <i>tert</i> -butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-4-nitrobenzenesulfonamide (5)	<u>S12</u>
¹ H NMR spectrum of 1,4,7-tris(<i>tert</i> -butoxycarbonyl)-10-(2-aminoethyl)cyclen (6)	<u>S13</u>
¹³ C NMR spectrum of 1,4,7-tris(<i>tert</i> -butoxycarbonyl)-10-(2-aminoethyl)cyclen (6)	<u>S16</u>
¹ <u>H NMR spectrum of 6-Chloro-N^2, N^4-bis-[2-(N^4, N^7, $N^{10'}$-Tri-<i>tert</i>-butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- 1,3,5-triazine-2,4-diamine (7)</u>	<u>S20</u>
$\frac{{}^{13}\text{C NMR spectrum of 6-Chloro-N^2, N^4-bis-[2-(N^4, N^7, N^{10'}-\text{Tri-tert-butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-}{1,3,5-triazine-2,4-diamine (7)}$	<u>S23</u>
¹ H NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (2h)	S27
13 C NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (2h)	<u>S31</u>
High resolution mass spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (2h)	<u>S33</u>
¹ H NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -[2-(1H-imidazol-5-yl)ethyl]melamine (2i)	<u>S34</u>
13 C NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -[2-(1H-imidazol-5-yl)ethyl]melamine (2i)	<u>S37</u>
High resolution mass spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -[2-(1H-imidazol-5-yl)ethyl]melamine (2i)	S39

¹ H NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -methylmelamine (2n)	<u>S40</u>
13 C NMR spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -methylmelamine (2n)	<u>S42</u>
High resolution mass spectrum of N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -methylmelamine (2n)	S44
¹ H NMR spectrum of 6-Methoxy-N ² ,N ⁴ -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (20)	<u>S45</u>
13 C NMR spectrum of 6-Methoxy- N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (20)	S47
<u>High resolution mass spectrum of 6-Methoxy-N^2, N^4-bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (20)</u>	S49
¹ H NMR spectrum of 2'-O-Methyluridylyl-3',5'-uridine (8)	<u> </u>
¹³ C NMR spectrum of 2'-O-Methyluridylyl-3',5'-uridine (8)	<u> </u>
³¹ P NMR spectrum of 2'-O-Methyluridylyl-3',5'-uridine (8)	<u> </u>
HPLC chromatograms of compounds 2h, 2i, 2n and 2o	<u> </u>
An example of fitting the potentiometric data	S56

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 $N-[2-(N^4, N^7, N^{10}-\text{Tri-tert-butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-4-nitrobenzenesulfonamide (5). 4-Nitrobenzenesulfonyl$ chloride (6,45 g, 29.1 mmol) was dissolved in a mixture of dry CH₂Cl₂ (6 mL) and pyridine (3 mL) and ethanolamine (0.586 mL, 9.71 mmol) was added dropwise at 0 °C (reacted vigorously). The reaction mixture was allowed to warm up to room temperature and stirred for 2,5 h, after which CH₂Cl₂ (50 mL) was added and the resulting solution washed with 2 M aq. HCl (50 mL). The organic phase was filtered and the precipitated light brown solid suspended in CH_2Cl_2 (100 mL) The mixture was washed with 2 M aq. KOH (2 × 100 mL), the organic phase dried with Na₂SO₄ and evaporated to a light brown powder. The crude N-(4-nitrobenzenesulfonyl)aziridine thus obtained (4, 1.00 g, 4.38 mmol) and 1,4,7-tris(tertbutoxycarbonyl)cyclen (3, 2.05 g, 4.33 mmol) were dissolved in MeCN (24 mL). The reaction mixture was stirred at room temperature for 168 h, after which it was evaporated to dryness. The residue was dissolved in CH₂Cl₂ and washed with saturated aq. NaHCO₃. The organic phase was dried with Na₂SO₄ and evaporated to dryness. The residue was purified on a silica gel column eluting with a mixture of EtOAc and petroleum ether (70:30, v/v). Yield 2.14 g (70%) of pale yellow foam. TLC: R_f (MeOH:CH₂Cl₂, 1:9, v/v) = 0.78; R_f (EtOAc:hexane, 7:3, v/v) = 0.62. ¹H NMR ($\delta_{\rm H}$)(500 MHz, CDCl₃): 8.36 (d, 2H, J = 8.8 Hz), 8.10 (d, 2H, J = 8.5 Hz), 3.47 (m, 12H), 3.09 (dd, 2H, J = 4.8 Hz), 2.59 (br, 6H), 1.48 (s, 27H). HRMS (ESI⁺): m/zcalcd 701.3544 obsd 701.3588 [M+H]⁺.

1,4,7-tris(*tert*-butoxycarbonyl)-10-(2-aminoethyl)cyclen (6). Compound 5 (2.00 g, 2.85 mmol) was dissolved in DMF (10.0 mL). Thioglycolic acid (0.456 mL) and DBU (4.0 mL) were added and the resulting mixture was stirred at room temperature for 2,5 h. Ethyl acetate (300mL) was added and the mixture was washed with saturated aq. NaHCO₃ (5 × 40 mL). The organic phase was dried with Na₂SO₄ and evaporated to yield 1.44 g (98%) of 6 as a yellowish glass. ¹H NMR ($\delta_{\rm H}$)(500 MHz, CDCl₃): 3.42 (m, 12H), 2.87 (t, 2H, *J* = 7.1 Hz), 2.67 (m, 6H), 1.47 (s, 9H), 1.45 (s, 18H). HRMS (ESI⁺): *m/z* calcd 516.3756 obsd 516.3722 [M+H]⁺.

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6-Chloro- N^2 , N^4 -bis-[2-(N^4 ', N^7 ', $N^{10'}$ -Tri-*tert*-butoxycarbonyl-1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (7). Compound 6 (1.44 g, 2.79 mmol), cyanuric chloride (0.210 g, 1.139 mmol) and DIPEA (2.5 mL, 14.35 mmol) were dissolved in MeCN (25 mL). The mixture was stirred at room temperature for 4 h, after which it was evaporated to dryness, the residue was dissolved in CH₂Cl₂ (50 mL) and washed with saturated aq. NaHCO₃ (2 × 50 mL). The organic phase was dried with Na₂SO₄ and evaporated to dryness and the residue was purified on a silica gel column eluting with a mixture of Et₃N, MeOH and CH₂Cl₂ (1:5:94, ν/ν). Yield 1.23 g (95%). ¹H NMR ($\delta_{\rm H}$)(500 MHz, CDCl₃): 3.70 – 3.20 (m, 28H), 2.90 – 2.50 (m, 12H), 1.47 (s, 36H), 1.45 (s, 18H). HRMS (ESI⁺): m/z calcd 1142.7063 obsd 1142.6998 [M+H]⁺.

 N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (2h). Compound 6 (0.220 g, 0.427 mmol), 2,4-dichloro-1,3,5-triazine (0.0276 g, 0.175 mmol) and DIPEA (0.4 mL, 2,3 mmol) were dissolved in MeCN (4.0 mL). The mixture was stirred at 50 °C for 216 h after which it was evaporated to dryness. The residue was dissolved in CH₂Cl₂ (20 mL) and washed with saturated aq. NaHCO₃ (2 × 20 mL). The organic phase was dried with Na₂SO₄ and evaporated to dryness and the residue was purified twice on a silica gel column eluting with a mixture of Et₃N, MeOH and CH₂Cl₂ (1:10:89 and 1:5:94, ν/ν , respectively). The product (0.0673 g, 0.0607 mmol) was dissolved in a mixture of TFA (2.0 mL), CH₂Cl₂ (1.6 mL) and MeOH (0.4 mL) and the resulting solution was stirred at room temperature for 4 h, after which it was evaporated to dryness. The residue was passed through Dowex 1x2 (HO⁻) resin to yield 21.6 mg (24%) of the desired product as a free amine. ¹H NMR (δ_{11})(500 MHz, D₂O): 8.35 (br, 1H), 3.13 – 3.00 (m, 12H), 2.99 – 2.85 (m, 28H). ¹³C NMR (δ_C)(125 MHz, D₂O): 172.1, 164.7, 53.6, 51.7, 51.5, 45.9, 45.1, 44.4. HRMS (ESI⁺): m/z calcd 508.4307 obsd 508.4310 [M+H]⁺.

 N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -[2-(1H-imidazol-5-yl)ethyl]melamine (2i). Compound 7 (0.33 g, 0.289 mmol), histamine (0.32 g, 2.88 mmol) and DIPEA (2 mL) were dissolved in MeCN (8 mL). The resulting mixture was stirred at 55 °C for 96 h, after which it was evaporated to dryness. The residue was purified on a silica gel column (300 mL SiO₂) eluting with a mixture of MeOH and CH₂Cl₂ (15:85, v/v). The product (0.255 g, 0.2094 mmol) was dissolved in a mixture of TFA (5.0 mL), MeOH (1.0 mL) and CH₂Cl₂ (4.0 mL) and the resulting solution

was stirred at room temperature for 4 h, after which it was evaporated to dryness. The residue was passed through Dowex 1x2 (HO⁻) resin to yield 86.3 mg (67%) of the desired product as a free amine. ¹H NMR (δ_{H})(500 MHz, D₂O): 7.52 (s, 1H), 6.77 (s, 1H), 3.44 (br, 2H), 3.20 (br, 4H), 2.70 (m, 2H), 2.55 – 2.25 (m, 36H). ¹³C NMR (δ_{C})(125 MHz, D₂O): 177.2, 165.2, 135.6, 134.8, 116.9, 53.4, 51.1, 45.0, 44.4, 43.5, 40.0, 38.5, 26.5. HRMS (ESI⁺): m/z calcd 309.2510 obsd 309.2520 [M+2H]²⁺.

 N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]- N^6 -methylmelamine (2n). Compound 7 (0.100 g, 0.087 mmol) was dissolved in 30% solution of methylamine in ethanol (1.0 mL). After being stirred at room temperature for 16 h the reaction mixture was evaporated to dryness and the residue was purified on a silica gel column eluting with a mixture of Et₃N, MeOH and CH₂Cl₂ (1:5:94, ν/ν). The product (0.090 g, 0.079 mmol) was dissolved in a mixture of TFA (2.5 mL), CH₂Cl₂ (2.0 mL) and MeOH (0.5 mL) and the resulting solution was stirred at room temperature for 4,5 h, after which it was evaporated to dryness. The residue was passed through Dowex 1x2 (HO⁻) resin to yield 40.4 mg (86%) of the desired product as a free amine. ¹H NMR (δ_H)(500 MHz, D₂O): 3.28 (br, 4H), 2.75 (br, 3H), 2.48 (m, 36H). ¹³C NMR (δ_C)(125 MHz, D₂O): 165.4, 53.5, 51.2, 44.9, 44.3, 43.4, 38.6, 26.9. HRMS (ESI⁺): m/z calcd 537.4572 obsd 537.4613 [M+H]⁺.

6-Methoxy- N^2 , N^4 -bis-[2-(1,4,7,10-tetraazacyclododecan-1-yl)ethyl]-1,3,5-triazine-2,4-diamine (20). Compound 7 (0.167 g, 0.146 mmol) was dissolved in 0.1 M methanolic sodium methoxide (10 mL). After being stirred at 50 °C for 16 h the reaction mixture was neutralized with 1.0 M acetic acid in methanol and evaporated to dryness. The residue was dissolved in CH₂Cl₂ (50 mL) and washed with saturated aq. NaHCO₃ (80 mL). The organic phase was dried with Na₂SO₄ and evaporated to dryness and the residue was purified twice on a silica gel column eluting with a mixture of Et₃N, MeOH and CH₂Cl₂ (1:10:89 and 1:5:94, ν/ν , respectively). The product (0.068 g, 0.0597 mmol) was dissolved in a mixture of TFA (2.0 mL), CH₂Cl₂ (1.6 mL) and MeOH (0.4 mL) and the resulting solution was stirred at room temperature for 4 h, after which it was evaporated to dryness. The residue was passed through Dowex 1x2 (HO⁻) resin to yield 26.0 mg (33%) of the desired product as a free amine. ¹H NMR ($\delta_{\rm H}$)(500

MHz, D₂O): 3.80 (br, 3H), 3.50 – 3.15 (m, 6H), 2.85 – 2.35 (m, 34H). ¹³C NMR (δ_C)(125 MHz, D₂O): 170.7, 165.9, 54.2, 50.4, 44.6, 43.3, 42.8, 38.6, 38.4. HRMS (ESI⁺): *m/z* calcd 538.4412 obsd 538.4404 [M+H]⁺.

2'-O-Methyluridylyl-3',5'-uridine (8). 5'-O-(4,4'-Dimethoxytrityl)-2'-O-methyluridine-3'-(2-cyanoethyl)-N,N-diisopropylphosphoramidite (1.00 g, 1.314 mmol) and 2',3'-O-isopropylideneuridine (0.448 g, 1.576 mmol) were dissolved in anhydrous MeCN (10 mL). 0.45 mol L⁻¹ 1H-tetrazole in MeCN (3.5 mL, 1.575 mmol) was added and the resulting mixture was stirred at room temperature for 2 h. A solution of I₂ (0.43 g, 1.69 mmol) in a mixture of THF (8 mL), water (4 mL) and 2,6-lutidine (2 mL) was added and the stirring was continued for 1 h, after which the reaction mixture was concentrated under reduced pressure, the residue was dissolved in CH₂Cl₂ (100 mL) and washed with 5% aq. NaHSO₃ (100 mL). The organic phase was dried with Na₂SO₄ and evaporated to dryness. The residue was purified on a silica gel column (550 mL SiO₂) eluting with a mixture of Et₃N, MeOH and CH_2Cl_2 (1:9:90, v/v). The product (1.0302 g, 1.073 mmol) was then dissolved in saturated methanolic ammonia (10 mL) and the mixture was stirred at room temperature for 1 h, after which it was evaporated to dryness. The residue was dissolved in water (10 mL), 37% aq. HCl (0.5 mL) was added and the resulting mixture stirred at room temperature for 6 h, washed with CH_2Cl_2 (6 × 20 mL) and evaporated to dryness. The residue was passed through Dowex 50Wx8 (Na⁺) resin and evaporated to dryness, yielding 0.5939 (94%) of 8 as the sodium salt. ¹H NMR ($\delta_{\rm H}$)(500 MHz, $D_{2}O$: 7.88 (d, 1H, J = 8.2 Hz), 7.86 (d, 1H, J = 8.1 Hz), 5.88 (d, 1H, J = 3.9 Hz), 5.88 (d, 1H, J = 3.3 Hz), 5.80 (d, 1H, J = 8.2 Hz), 5.79 (d, 1H, J = 8.2 Hz), 5.80 (d, 1H, J = 88.1 Hz), 4.58 (ddd, 1H, $J_1 = 6.7$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz, $J_3 = 2.7$ Hz), 4.26 (d, 1H, J = 3.6 Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz, $J_2 = 5.6$ Hz), 4.26 (m, 1H), 4.21 (m, 2H), 4.15 (ddd, 1H, $J_1 = 11.8$ Hz), 4.26 (m, 2H), 4.26 (m, 2H), 4.15 (m, 2H), 4. 4.1 Hz, $J_3 = 2.7$ Hz), 4.08 (m, 2H), 3.87 (dd, 1H, $J_1 = 13.1$ Hz, $J_2 = 2.5$ Hz), 3.76 (dd, 1H, $J_1 = 13.1$ Hz, $J_2 = 3.8$ Hz), 3.47 (s, 3H). ¹³C NMR (δ_C)(125) MHz, D₂O): 166.0, 165.9, 151.6, 151.3, 141.4, 141.3, 102.4, 102.2, 89.0, 87.3, 83.3, 82.6, 81.5, 73.8, 71.4, 69.3, 64.5, 59.9, 57.9. ³¹P NMR (δ_P)(202 MHz, D₂O): -0.89 (s, 1P). HRMS (ESI): *m*/*z* calcd 563.1032 obsd 563.1146 [M-H]⁻.



file: ...ents\NMR Data\TL139\TL139_1H\1\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 182.790 ppm/cm: 0.36548





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number of scans: 16



time domain size: 65536 points

width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt number of scans: 1654

processed size: 32768 complex points LB: 1.000 GF: 0.0000 Hz/cm: 1201.201 ppm/cm: 9.55075

SpinWorks 3:



number of scans: 1654



number of scans: 1654



transmitter treq.: 125.7/0364 MHz time domain size: 65536 points width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt number of scans: 1654 processed size: 32768 complex points LB: 1.000 GF: 0.0000 Hz/cm: 86.035 ppm/cm: 0.68406



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file: ...uments\NMR Data\TL141\TL141_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.129809 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 38.194 ppm/cm: 0.07637

SpinWorks 3:



number of scans: 16

width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt

number of scans: 1471



Hz/cm: 964.039 ppm/cm: 7.66507



width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt

number of scans: 1471



Hz/cm: 253.456 ppm/cm: 2.01523



number of scans: 1471





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Figure S1. The ¹H NMR signal of the sole aromatic proton of **2h** measured at different temperatures, illustrating the fact that at room temperature equilibrium between the various conformers is slow on NMR time scale.









file: ...uments\NMR Data\TL160\TL160_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 141.063 ppm/cm: 0.28205





time domain size: 65536 points

width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt

number of scans: 16

rreq. or 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 23.041 ppm/cm: 0.04607



file: ...uments\NMR Data\TL160\TL160_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 50.135 ppm/cm: 0.10024



number of scans: 727









file: ...uments\NMR Data\TL153\TL153_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 170.022 ppm/cm: 0.33995





file: ...uments\NMR Data\TL153\TL153_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 30.366 ppm/cm: 0.06072



Hz/cm: 886.191 ppm/cm: 7.04611

width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt

width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt

number of scans: 0



file: ...\My Documents\NMR Data\TL165\1\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 168.777 ppm/cm: 0.33746

file: ...\My Documents\NMR Data\TL165\1\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 46.678 ppm/cm: 0.09333

number of scans: 15000

time domain size: 65536 points

width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt number of scans: 15000

processed size: 32768 complex points LB: 1.000 GF: 0.0000 Hz/cm: 204.961 ppm/cm: 1.62965

file: ...uments\NMR Data\TL159_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 207.390 ppm/cm: 0.41467

file: ...uments\NMR Data\TL159\TL159_1H\fid expt: <zg30: transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16 freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 53.137 ppm/cm: 0.10625

file: ...uments\NMR Data\TL159\TL159_1H\fid expt: <zg30> transmitter freq.: 500.133089 MHz time domain size: 65536 points width: 10330.58 Hz = 20.6557 ppm = 0.157632 Hz/pt number of scans: 16

freq. of 0 ppm: 500.130000 MHz processed size: 32768 complex points LB: 0.300 GF: 0.0000 Hz/cm: 25.373 ppm/cm: 0.05073

time domain size: 65536 points width: 30030.03 Hz = 238.7687 ppm = 0.458222 Hz/pt number of scans: 685

Hz/cm: 671.659 ppm/cm: 5.34036

number of scans: 19

Figure S2. HPLC chromatograms of compounds 2h, 2i, 2n and 2o [Thermo Hypersil HyPurity C18 column ($150 \times 4.6 \text{ mm}$, 5 µm); flow rate = 1 mL min⁻¹; 0.1% aq. TFA and a linear gradient of 0 \rightarrow 30% of 0.1% TFA in MeCN during 15 min, after which 30% of 0.1% TFA in MeCN for 5 min, λ = 220 nm]. The sharp peaks at the end of each chromatogram are artifacts that appear at random retention times even when no sample is injected. The most probable explanation for these peaks is air from the MeCN line.

Figure S3. Examples illustrating the excellence of fit of the potentiometric data to the proposed model, as well as the fact that simpler models fail to accurately describe the protonation equilibria prevalent in the system. (A) An example of experimental titration data (\Box) and the fit (solid line) of the binary system using the model described in the text $([Zn^{II}]_{TOT} = 2.91 \times 10^{-3}$ M, $[L]_{TOT} = 1.57 \times 10^{-3}$ M, $V_0 = 4.55$ cm³). Insert shows the $\Delta pH = pH(observed) - pH(calculated)$ differences of the best fit of the accepted model (\Box) and when the species $Zn_2HL(\bullet)$, $Zn_2L(\bullet)$, $Zn_2H_{-1}L$ (*) or Zn_2H_2L (\blacktriangle) is excluded from the model. (B) An example of experimental titration data (\Box) and the fit (solid black line) of the ternary system using the model described in the text ($[Zn^{II}]_{TOT} = 2.30 \times 10^{-3} \text{ M}$, $[L]_{TOT} = 1.29 \times 10^{-3} \text{ M}, [mUpU]_{TOT} = 1.22 \times 10^{-3} \text{ M}, V_0$ $= 5 \text{ cm}^3$). Red line shows the best fit when the ternary $Zn_2L(mUpU)$ species is omitted from the model.