

Electronic supplementary information

Synthesis, Crystal Structures and Magnetic Properties of bis(μ -dialkoxo)-bridged Linear Trinuclear Copper(II) Complexes with Aminoalcohol Ligands: A Theoretical/Experimental Magneto-Structural Study.

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Table S1 Main hydrogen bond parameters (Å, °) for **1–4**

Complex	D–H···A	d(D–H)	d(H···A)	d(D···A)	∠(D–H···A)
1^a	N1–H1NB···O11 ⁱ	0.85(4)	2.35(4)	3.130(4)	152(3)
	N1–H1NA···O6A ⁱⁱ	0.84(4)	2.12(4)	2.959(8)	171(4)
	N1–H1NA···O6B ⁱⁱ	0.84(4)	2.34(5)	3.170(11)	171(4)
	N2–H2NA···O8B ⁱⁱ	0.79(4)	2.12(5)	2.870(9)	158(4)
	N2–H2NA···O8A ⁱⁱ	0.79(4)	2.43(4)	3.099(5)	142(4)
	N2–H2NB···O12 ⁱ	0.84(4)	2.28(4)	3.020(4)	149(4)
	N3–H3NA···O10 ⁱⁱⁱ	0.88(4)	2.30(4)	3.062(4)	145(3)
	N3–H3NB···O2 ^{iv}	0.86(4)	2.10(4)	2.964(3)	177(3)
	N4–H4NB···O1 ^{iv}	0.77(4)	2.55(4)	3.193(3)	143(3)
	N4–H4NA···O11 ⁱⁱⁱ	0.87(4)	2.16(5)	3.023(4)	171(4)
2^b	O13–H13O···O5	0.79(4)	2.11(4)	2.850(3)	155(4)
	N1–H1NA···O8 ⁱ	0.86(4)	2.23(4)	3.059(3)	162(4)
	N1–H1NA···O10 ⁱ	0.86(4)	2.50(4)	3.244(4)	144(3)
	N2–H2NA···O5 ⁱ	0.85(4)	2.20(4)	2.926(3)	143(3)
	N2–H2NB···O9 ⁱⁱ	0.84(4)	2.31(4)	3.048(3)	147(3)
	N2–H2NB···O8 ⁱ	0.84(4)	2.59(4)	3.074(3)	118(3)
	N3–H3NA···O6 ⁱⁱⁱ	0.91(4)	2.37(4)	3.118(3)	139(3)
	N3–H3NB···O10 ^{iv}	0.84(4)	2.40(4)	3.121(4)	145(3)
	N4–H4NA···O9	0.83(4)	2.34(4)	3.025(3)	140(3)
	N4–H4NB···O1 ^v	0.91(4)	2.29(4)	3.131(3)	154(3)
3^c	N4–H4NB···O7 ⁱⁱⁱ	0.91(4)	2.54(4)	3.061(3)	117(3)
	N1–H1NA···O2 ⁱ	0.77(5)	2.53(5)	3.118(4)	134(4)
	N1–H1NA···Br1 ⁱ	0.77(5)	2.94(5)	3.514(4)	133(5)
	N1–H1NB···Br1 ⁱⁱ	0.79(6)	2.65(6)	3.434(4)	172(4)
	N2–H2NA···Br1 ⁱⁱ	0.84(5)	2.62(5)	3.457(4)	172(4)
	N2–H2NB···O1 ⁱ	0.82(5)	2.47(5)	3.148(4)	141(4)
4^d	N2–H2NB···Br1 ⁱ	0.82(5)	3.01(5)	3.528(4)	123(4)
	N1–H1NA···O5 ⁱ	0.90(4)	2.15(4)	2.992(3)	156(3)
	N1–H1NB···O2 ⁱⁱ	0.79(4)	2.21(4)	2.971(3)	162(3)
	N2–H2NA···O4	0.87(4)	2.47(4)	2.970(3)	117(3)
	N2–H2NB···O5 ⁱⁱⁱ	0.78(3)	2.35(3)	3.111(3)	168(3)
	N2–H2NB···O3 ⁱⁱⁱ	0.78(3)	2.32(3)	2.923(3)	135(3)

^a Symmetry codes for **1**: (i) $x, y, z - 1$; (ii) $-x + 1, -y + 2, -z$; (iii) $-x, -y + 1, -z + 2$; (iv) $-x, -y + 1, -z + 1$.

^b Symmetry codes for **2**: (i) $-x + 3/2, y - 1/2, -z + 1/2$; (ii) $x + 1/2, -y + 1/2, z + 1/2$; (iii) $x - 1/2, -y + 1/2, z - 1/2$; (iv) $-x + 1, -y + 1, -z$; (v) $-x + 1, -y, -z$.

^c Symmetry codes for **3**: (i) $-x, y + 1/2, -z + 1/2$; (ii) $x, y + 1, z$.

^d Symmetry codes for **4**: (i) $-x + 1/2, y + 1/2, -z + 1/2$; (ii) $-x + 1/2, y - 1/2, -z + 1/2$; (iii) $x + 1/2, -y - 1/2, z + 1/2$.

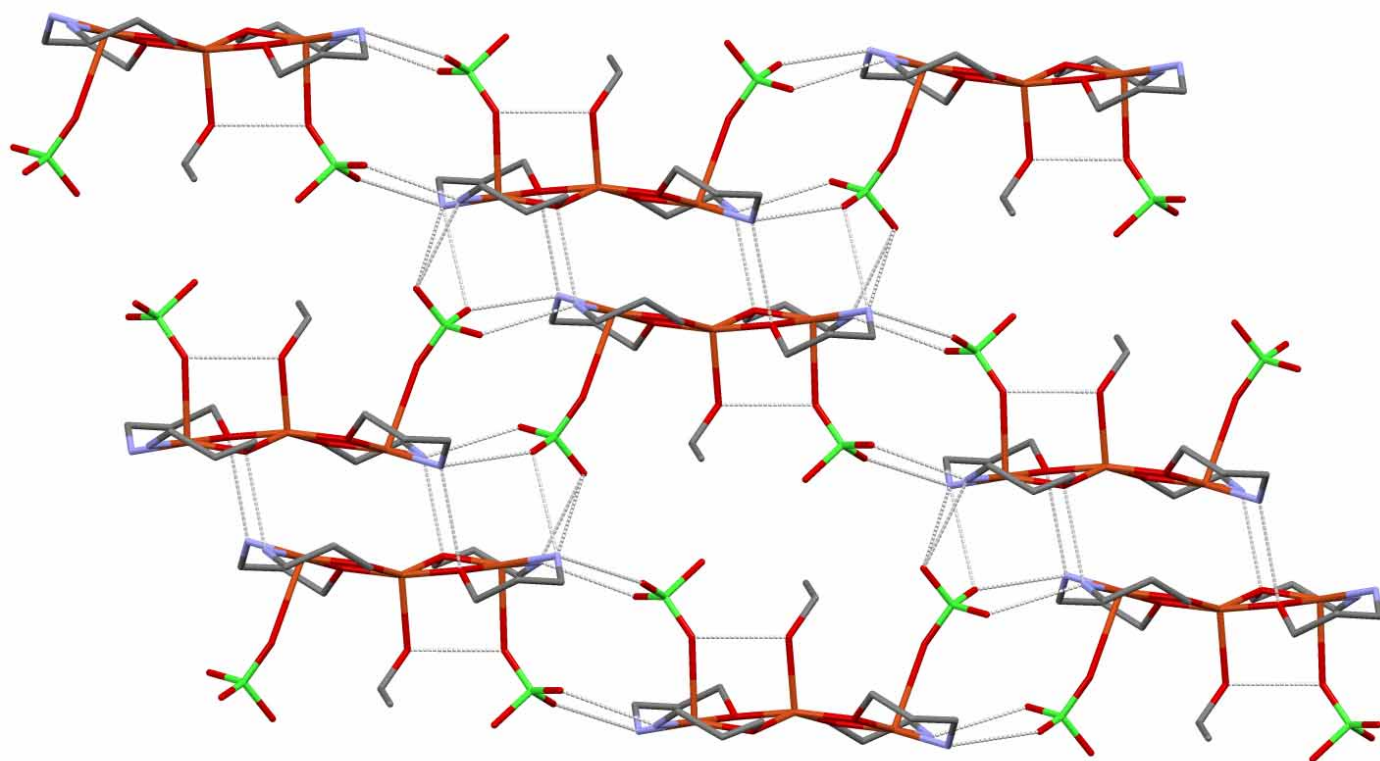


Fig. S1 The partial crystal packing diagram of **1** showing the 2D hydrogen bonding network (grey dashed lines). Hydrogen atoms and minor-occupancy oxygen atoms (O6B, O7B and O8B) of the disordered perchlorate anion are omitted for clarity.

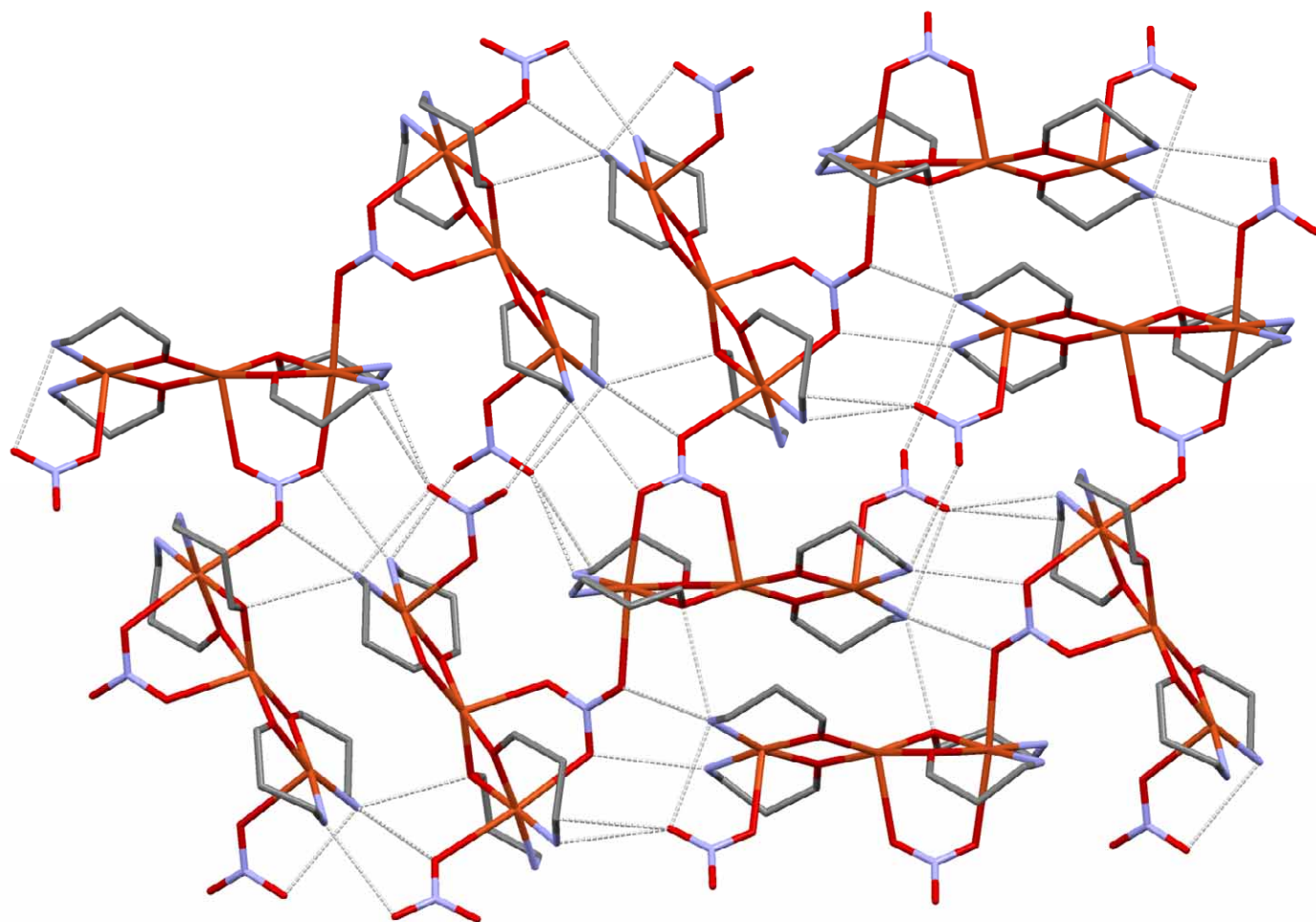


Fig. S2 The partial crystal packing diagram of **2** showing 1D polymeric chains and the extended 2D hydrogen bonding network (grey dashed lines). Hydrogen atoms are omitted for clarity.

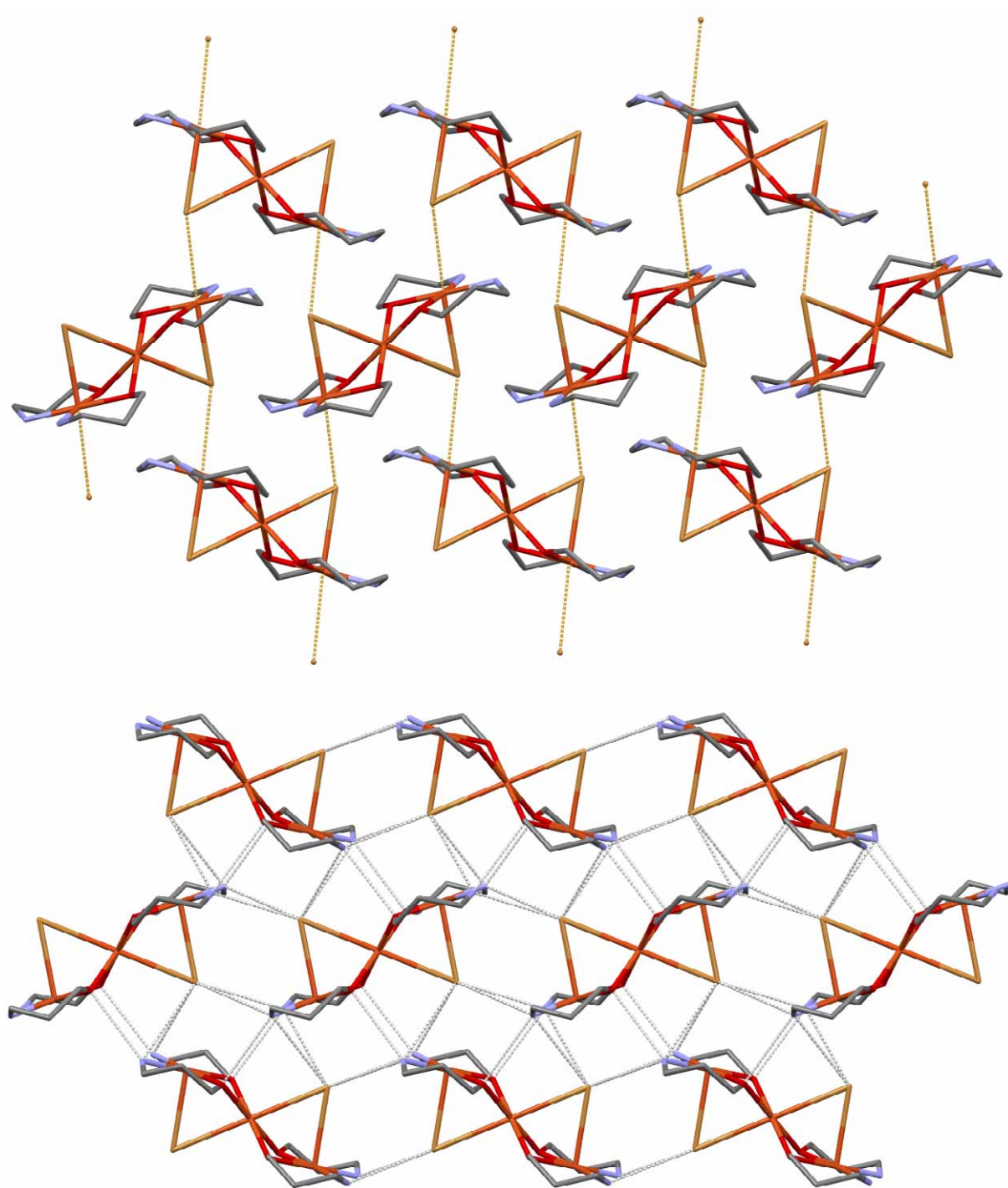


Fig. S3 The partial crystal packing diagram of **3** showing weak Cu...Br interactions (yellow dashed lines) forming the 2D layer (top) and the 2D hydrogen bonding network (bottom, grey dashed lines). Hydrogen atoms are omitted for clarity.

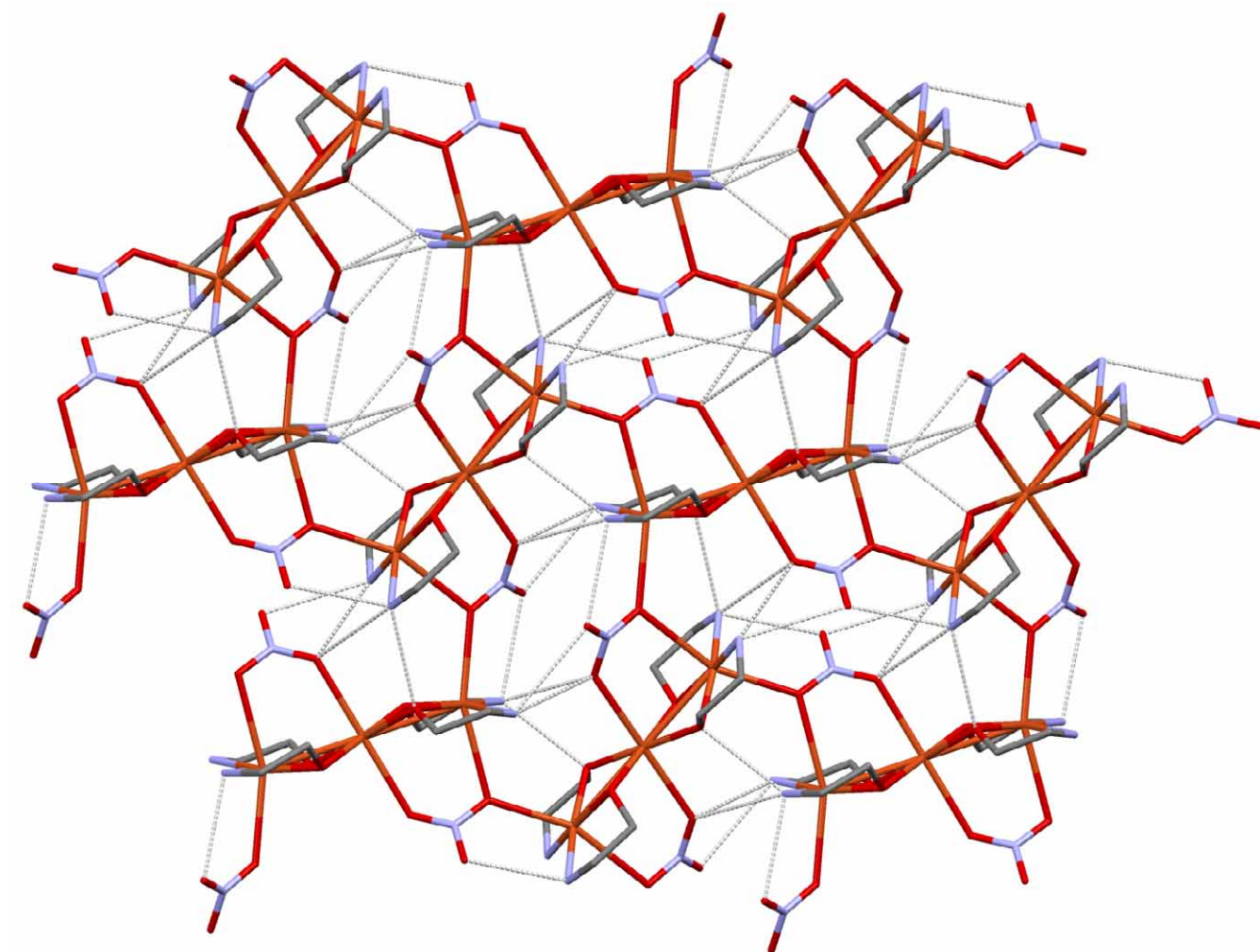


Fig. S4 The partial crystal packing diagram of **4** showing the 2D layer and the 2D hydrogen bonding network (grey dashed lines). Hydrogen atoms are omitted for clarity.