Supplementary Information

Li-Na Ternary Amidoborane for Hydrogen Storage: Experimental and First-principles Study

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Fig. S1 Temperature-programmed desorption (TPD) spectra of LiNH₂BH₃ (LiAB), NaNH₂BH₃ (NaAB) and (Li-Na)NH₂BH₃.



Fig. S2 H₂ evolution from LiNH₂BH₃ (LiAB), NaNH₂BH₃ (NaAB) and Li-Na)NH₂BH₃ during dehydrogenation process. *ca.* 2 *equiv.* H₂ (~ 8.9 wt. % H₂) are detached via a two-step process at temperature below 200 °C, which is similar to the H-desorption processes of LiAB and NaAB. Quantitative measurement of hydrogen desorption at elevated temperatures was carried out on a commercial gas reactor controller provided by the Advanced Materials Corporation. Sample was ramp-heated in argon at 1 K min⁻¹ each time.

No	Li ₄ /Na ₄ arrangements							
	А	В	С	D	Е	F	G	Н
01	Li	Na	Li	Li	Na	Li	Na	Na
02	Li	Li	Na	Li	Na	Li	Na	Na
03	Li	Na	Li	Li	Na	Na	Na	Li
04	Li	Li	Na	Li	Na	Na	Na	Li
05	Li	Na	Na	Li	Li	Na	Li	Na
06	Na	Li	Li	Na	Li	Na	Na	Li
07	Na	Li	Li	Na	Na	Li	Li	Na
08	Na	Na	Li	Li	Na	Na	Li	Li
09	Li	Na	Li	Na	Li	Na	Li	Na
10	Li	Na	Li	Na	Na	Li	Na	Li
11	Na	Na	Li	Li	Na	Li	Li	Na
12	Na	Li	Na	Li	Na	Na	Li	Li
13	Na	Na	Li	Li	Li	Li	Na	Na
14	Li	Li	Li	Li	Na	Na	Na	Na

Table S1 14 possible symmetrically non-equivalent Li₄/Na₄ arrangements on 8-cation sites (A, B, C...H are corresponding sites in Fig. S3) in Li₄Na₄AB₈.



Fig. S3 Labeled 8-cation sites in unit cell of $\text{Li}_{1-x}\text{Na}_x\text{AB}$. The possibilities of Li_4/Na_4 occupation on 8-cation sites are $C_8^4 = 70$, we calculated the pair correlation functions of all possible configurations, and categorized them into 14 symmetrically non-equivalent Li/Na arrangements (Table S1), as the symmetrically equivalent arrangements possess the equal pair correlation functions. The optimized structures are displayed in Fig. S4.



Fig. S4 The optimized structures of 14 possible symmetrically non-equivalent configurations of $Li_4Na_4AB_8$.



Fig. S5 The partial and total electronic density of states of (a) LiNH₂BH₃ (LiAB) and (b) NaNH₂BH₃ (NaAB). The Fermi level is set at zero energy; *s*-electron contributions are depicted in gray.



Fig. S6 The calculated partial and total phonon density of states of (a) LiNH₂BH₃ (LiAB) and (b) NaNH₂BH₃ (NaAB).