

Supplementary Information for:

Metal Induced Folding: Synthesis and Conformational Analysis of the Lanthanide Complexes of two 44-membered Hydrazone Macrocycles†

Jörg Klein,^a Jack K. Clegg,^a Vittorio Saggiomo,^b Lisa Reck,^b Ulrich Lüning^{b*} and Jeremy K. M. Sanders^{*a}

^aUniversity Chemical Laboratory, University of Cambridge, Lensfield Road, Cambridge CB2 1EW, United Kingdom

Fax: (+)44 (0)1223 336017

Email: jkms@cam.ac.uk

^b Otto-Diels-Institut für Organische Chemie, Olshausenstr. 40, D-24098 Kiel, Germany.

Fax: +49-431-880-1558;

Tel: +49-431-880-2450

E-mail: luening@oc.uni-kiel.de

1. X-ray Crystallography.....	2
2. HRMS (High Resolution Mass Spectrometry).....	4
3. UV-Vis Spectroscopy	5
3. NMR	7

1. X-ray Crystallography

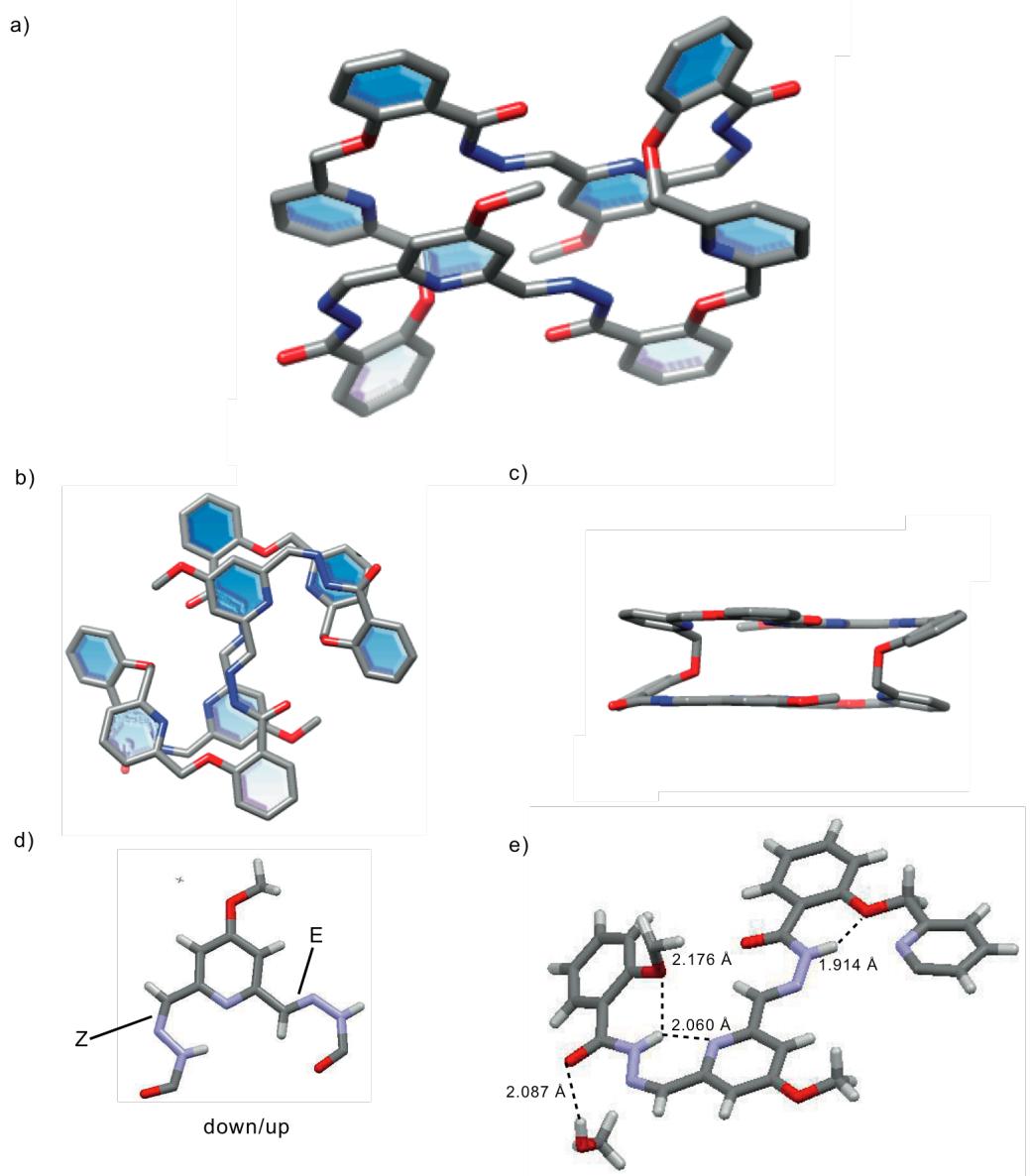


Fig. 1 Solid state structure of unbound ligand **1**. a) tilted view; b) top view; c) side view; d) orientation of hydrazone bonds; e) important hydrogen bonds.

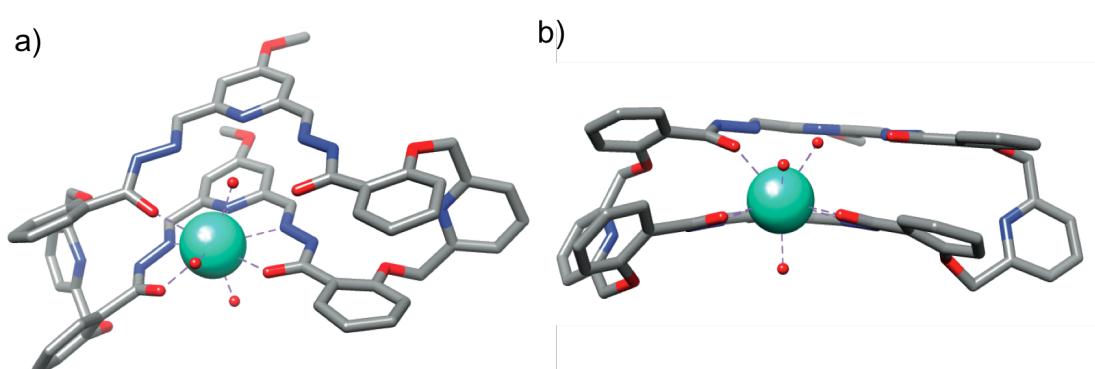


Fig. 2 Solid state structure of Eu³⁺ complex with CN nine. a) tilted view; b) side view.

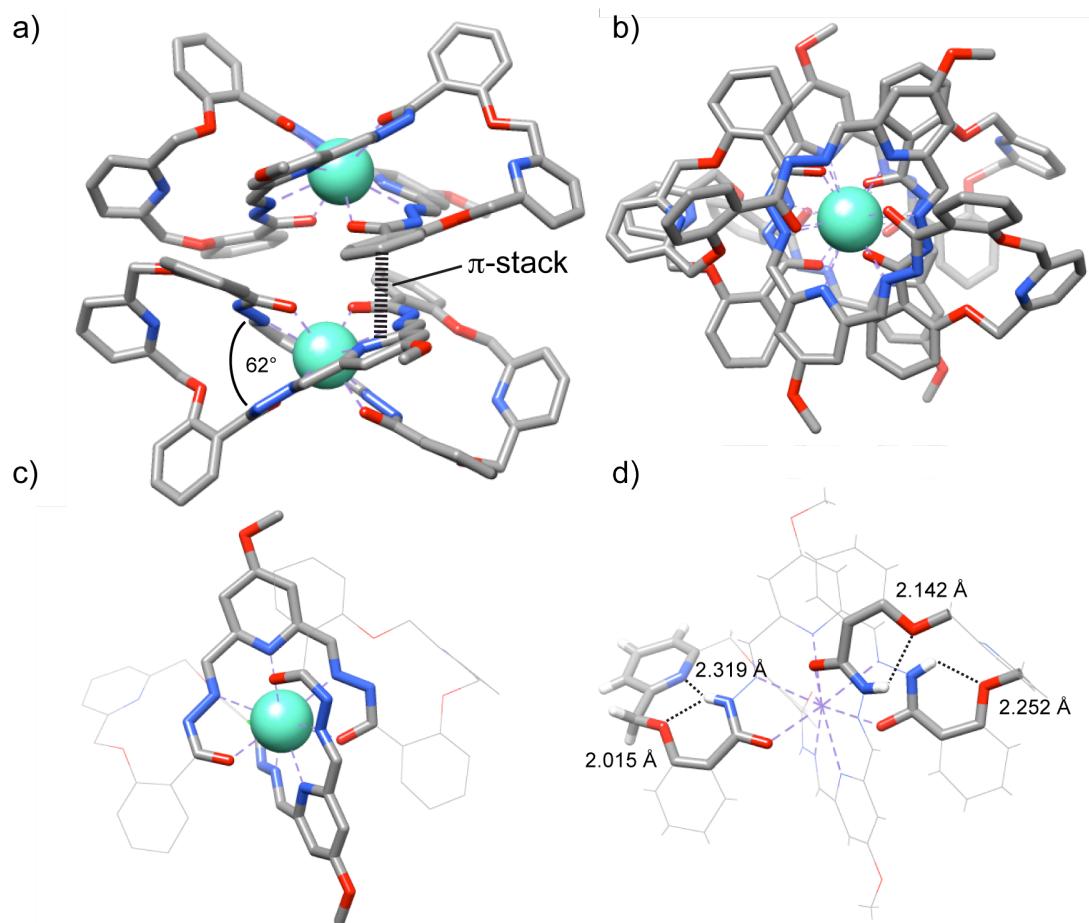


Fig. 3 Solid state structure of Europium complex with CN ten. a) side view; b) top view; c) arrangement of N_3O_2 binding motif; d) important hydrogen bonds.

2. HRMS (High Resolution Mass Spectrometry)

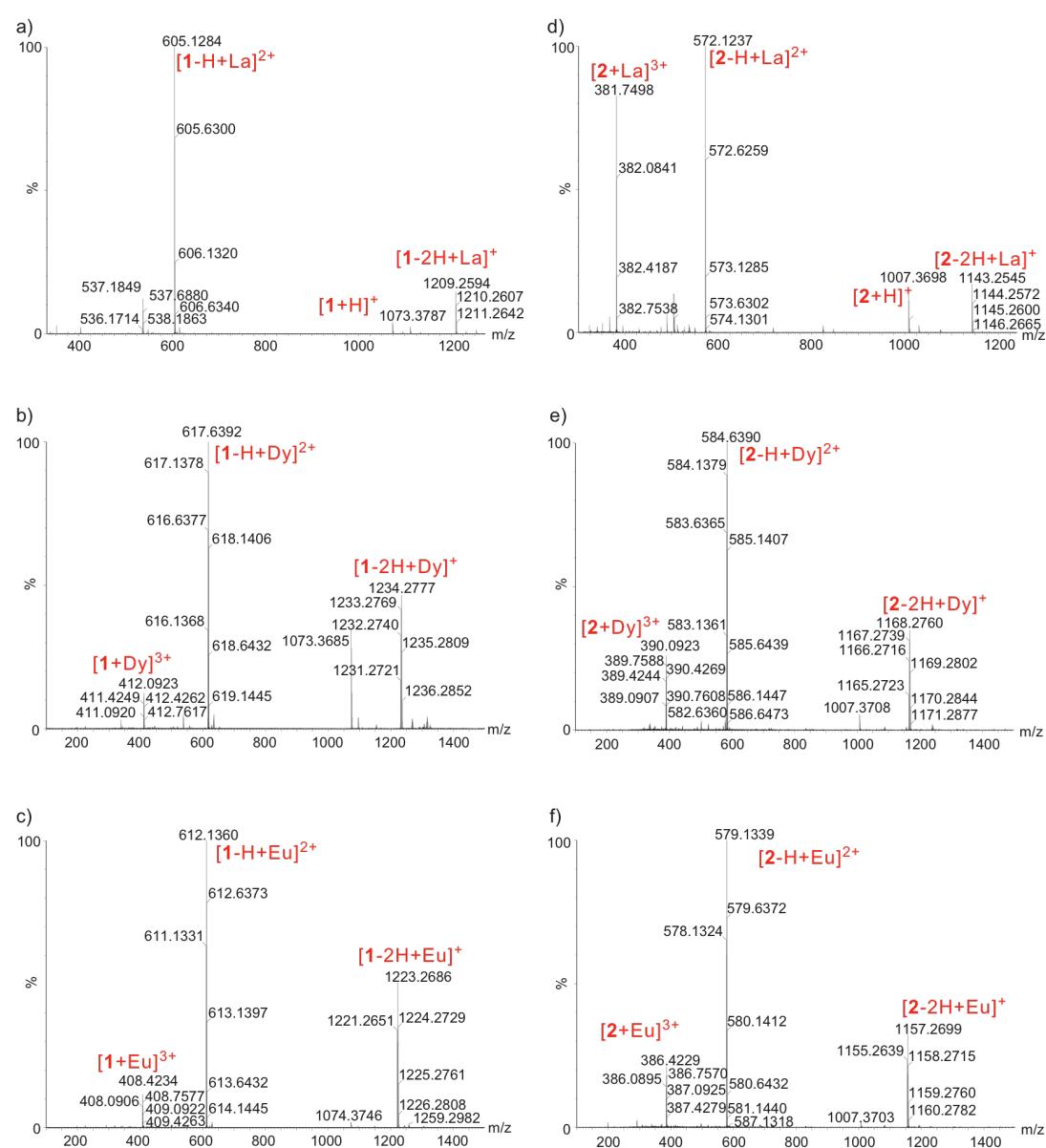


Fig. 4 (-)ESI-HRMS of a) $[1][\text{La}^{3+}]$; b) $[1][\text{Dy}^{3+}]$; c) $[1][\text{Eu}^{3+}]$; d) $[2][\text{La}^{3+}]$; e) $[2][\text{Dy}^{3+}]$; f) $[2][\text{Eu}^{3+}]$. All spectra were recorded in $\text{CDCl}_3/\text{MeOD}$ (1:1).

3. UV-Vis Spectroscopy

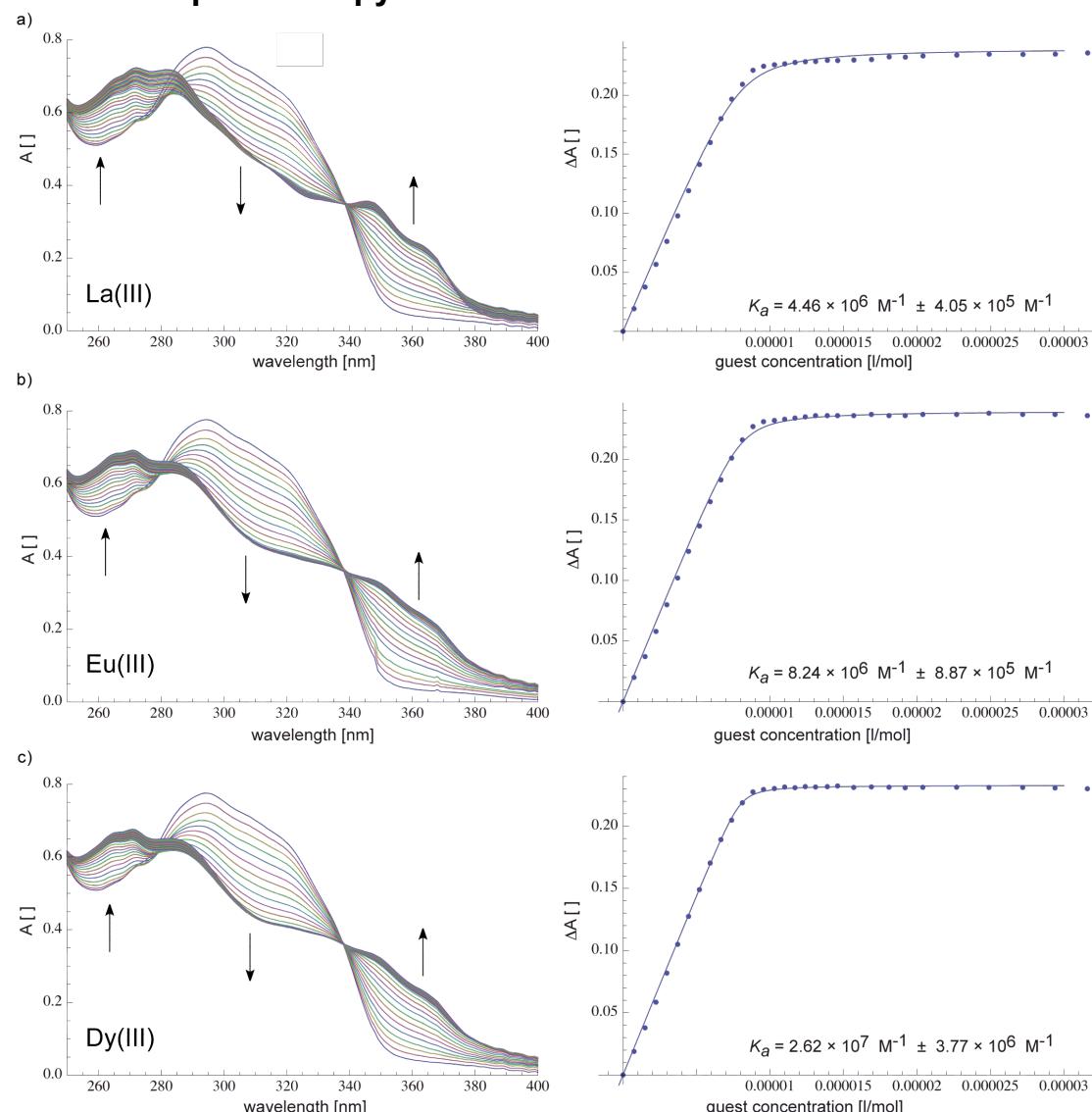


Fig. 5 UV-Vis spectra (left) and binding isotherms (right, $\Delta A_{320\text{nm}}$) of titrations of **1** with a) La^{3+} , b) Eu^{3+} , c) Dy^{3+} . Increasing and decreasing bands are indicated by arrows. Binding isotherms are obtained by plotting the change in absorption at 320 nm against the guest concentration (dots) and fitting it with a model (line).

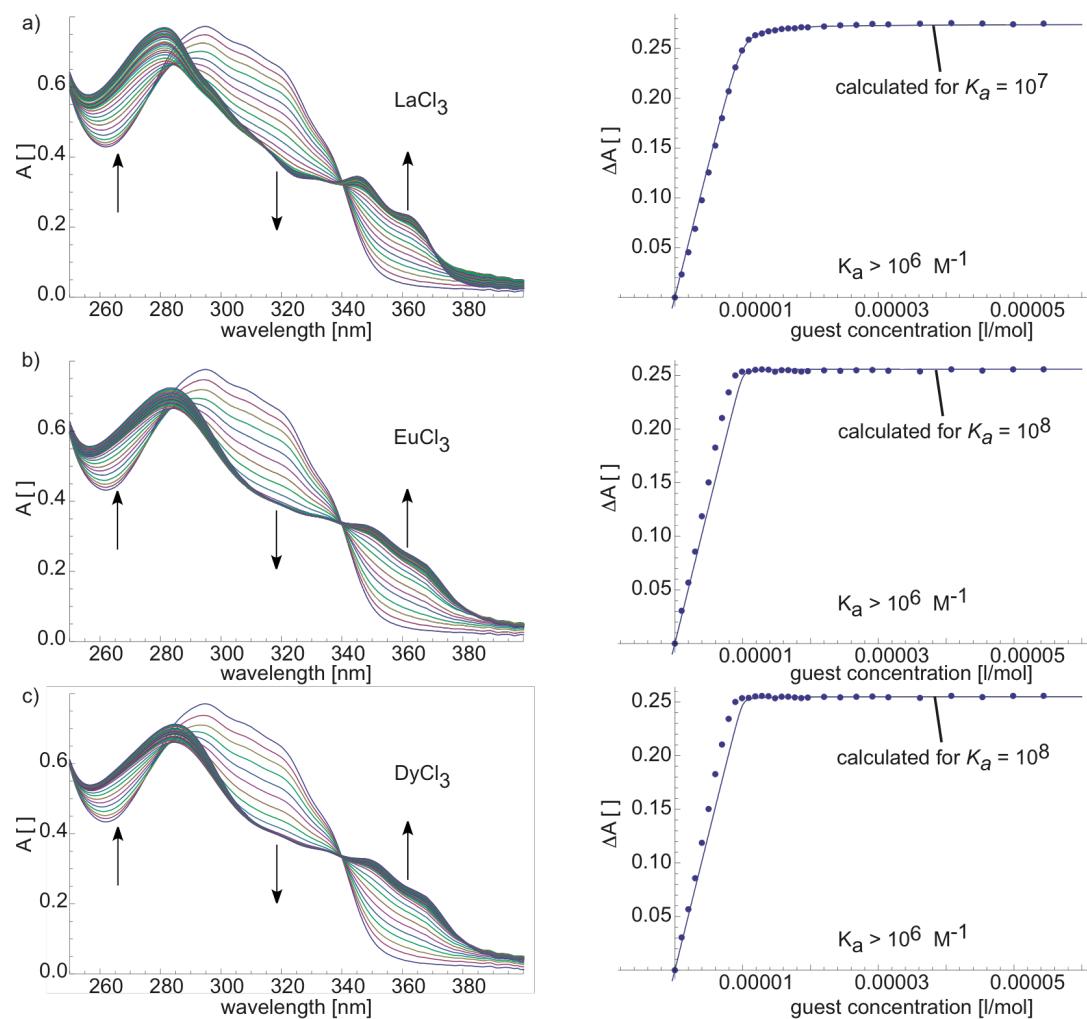


Fig. 6 UV-Vis spectra (left) and binding isotherms (right, $\Delta A_{320\text{nm}}$) of titrations of **2** with a) La^{3+} , b) Eu^{3+} , c) Dy^{3+} . Increasing and decreasing bands are indicated by arrows. Binding isotherms are obtained by plotting the change in absorption at 320 nm against the guest concentration (dots) and fitting it with a model (line).

3. NMR

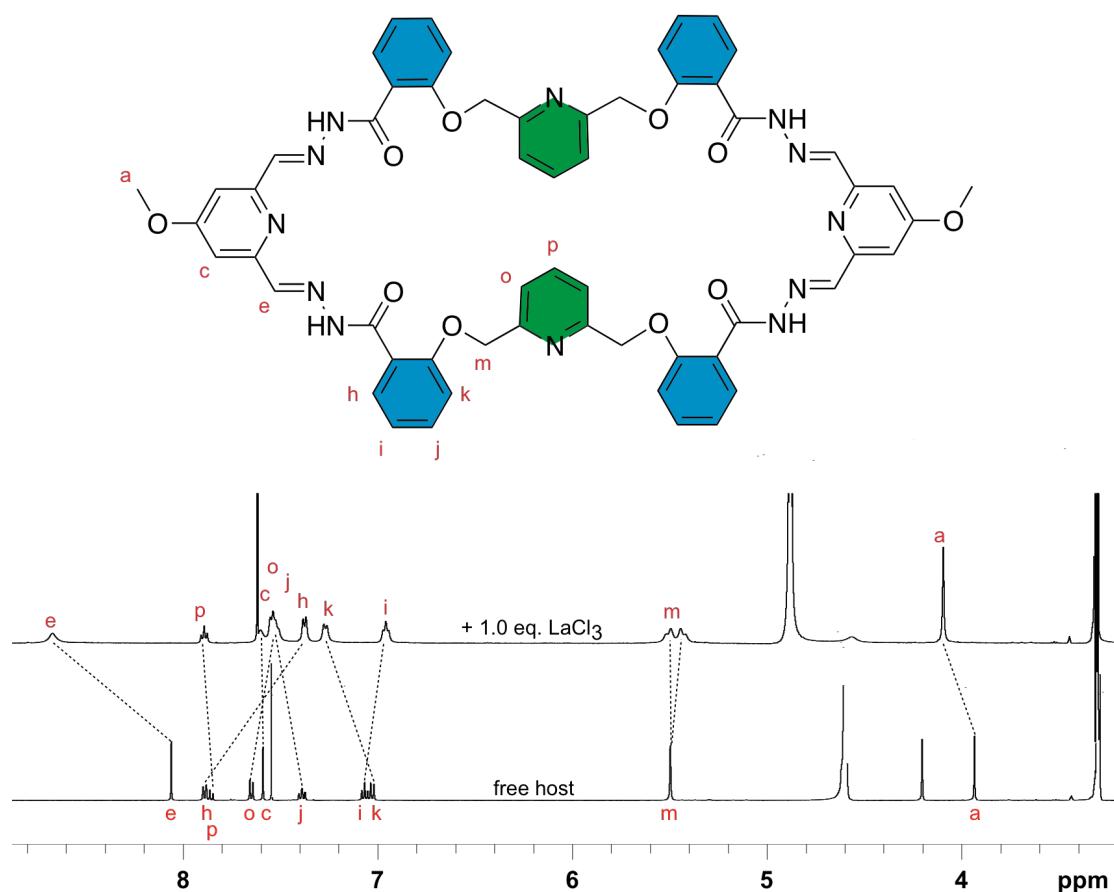


Fig. 7 NMR spectra (500 MHz, 295 K, CDCl₃/MeOD, 1:1) of a **1** and its La³⁺ complex (1 mM).

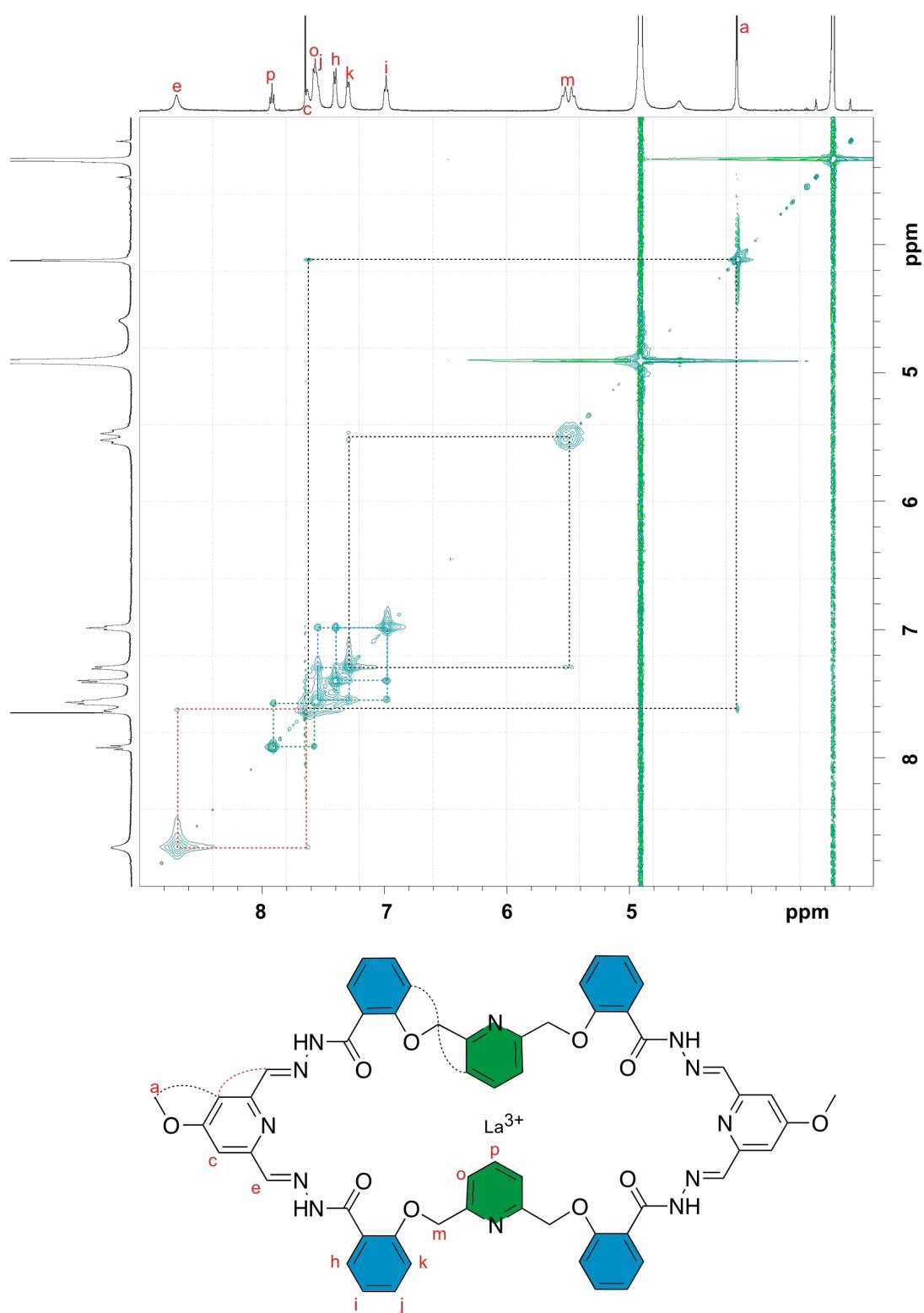


Fig.8: NOESY spectra of **1**[La³⁺] (1 mM in CDCl₃/MeOD, 1:1, 298 K, 500 MHz, mixing time = 800 ms). The cross-peaks are indicated by dotted lines with corresponding colours in the NOESY spectrum and the structure below.

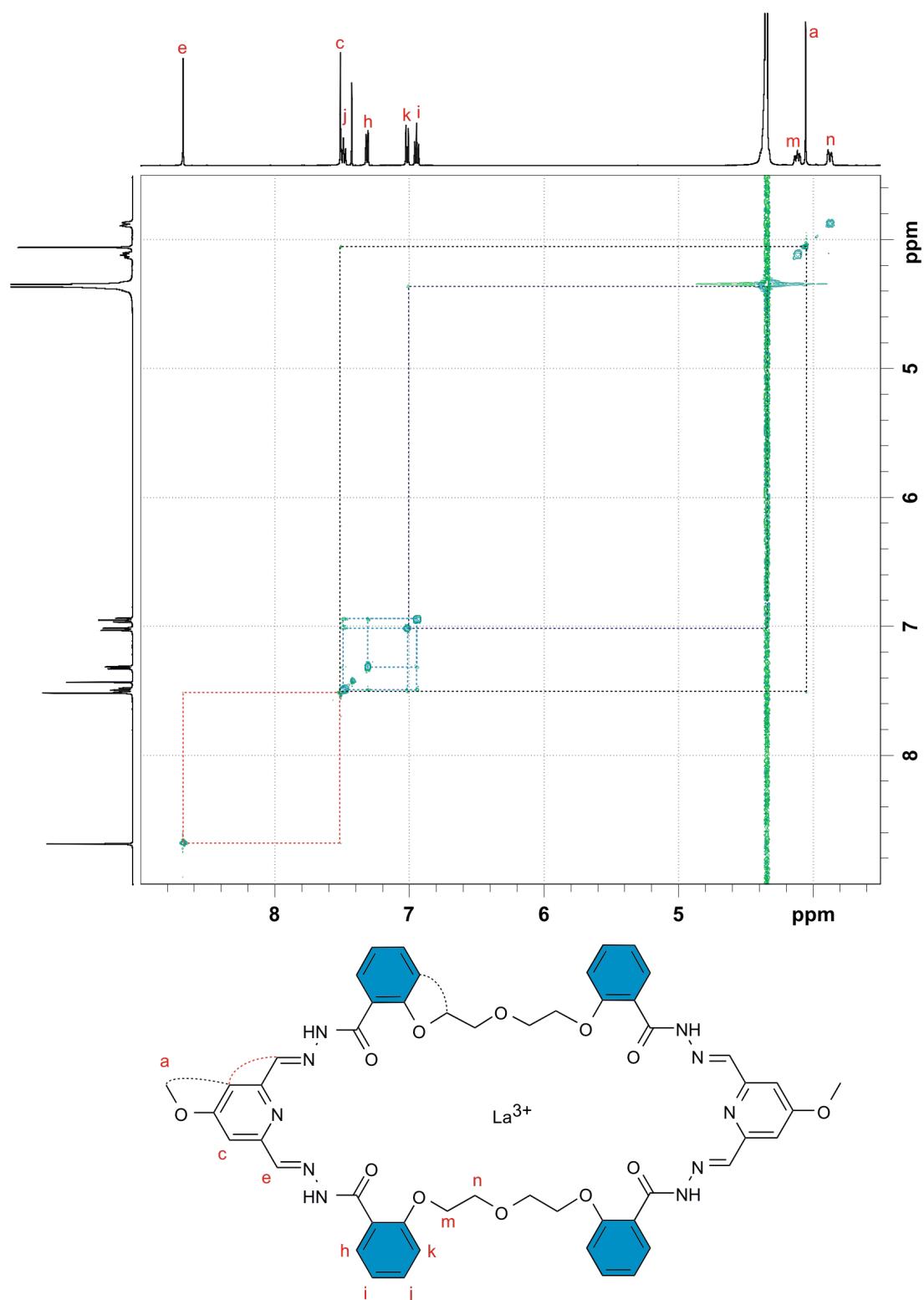


Fig. 9: NOESY spectra of **2**[La³⁺] (1 mM in CDCl₃/MeOD, 1:1, 298 K, 500 MHz, mixing time = 800 ms). The cross-peaks are indicated by dotted lines with corresponding colours in the NOESY spectrum and the structure below.