

Systematic investigations on copper(II) coordination polymers based on organic ligands with mixed carboxylic and nitrogen-based moieties

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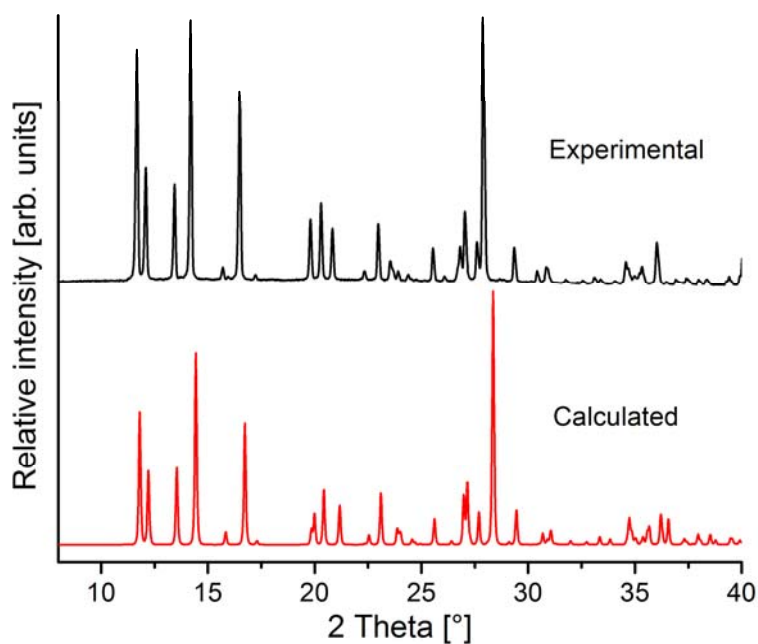


Fig. S1. Experimental XRPD pattern (top) and XRPD pattern calculated from single crystal data of compound **1** (bottom).

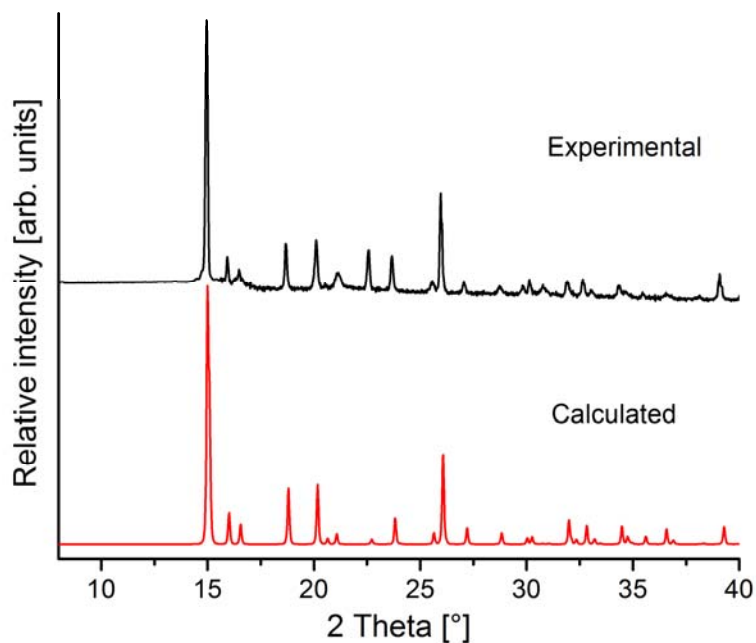


Fig. S2. Experimental XRPD pattern (top) and XRPD pattern calculated from single crystal data of compound **2** (bottom).

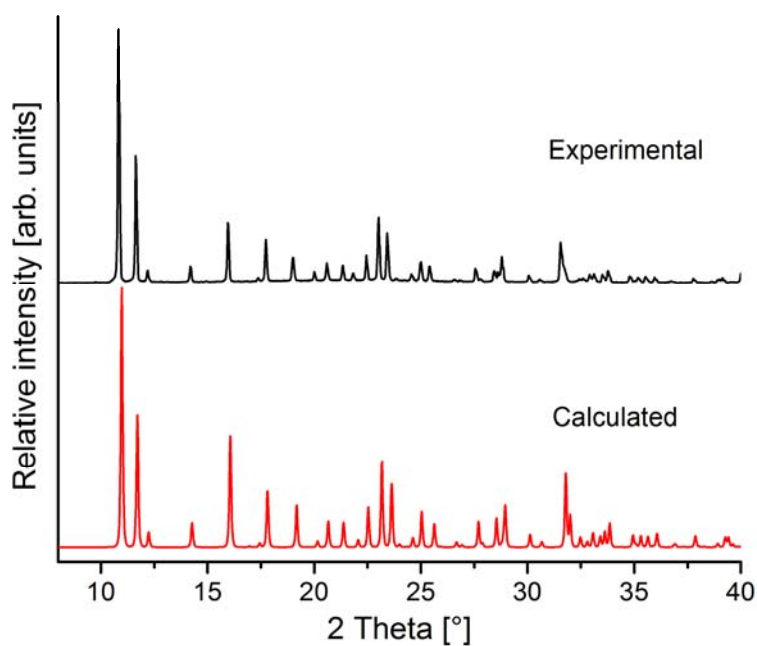


Fig. S3. Experimental XRPD pattern (top) and XRPD pattern calculated from single crystal data of compound **3** (bottom).

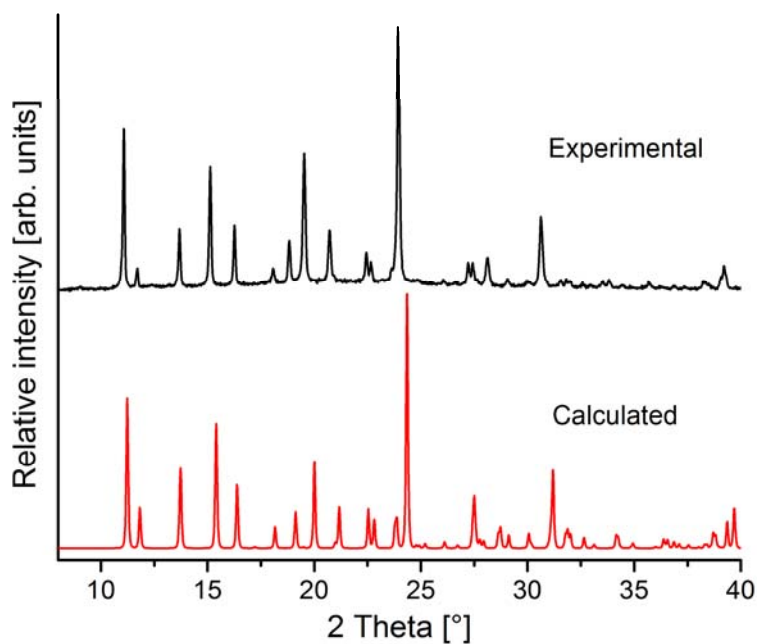


Fig. S4. Experimental XRPD pattern (top) and XRPD pattern calculated from single crystal data of compound **4** (bottom).

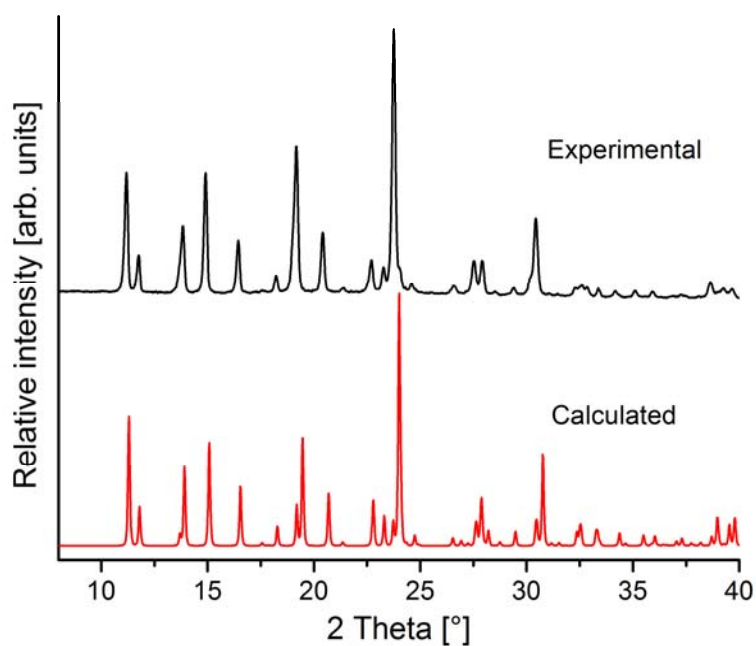


Fig. S5. Experimental XRPD pattern (top) and XRPD pattern calculated from single crystal data of compound **5** (bottom).

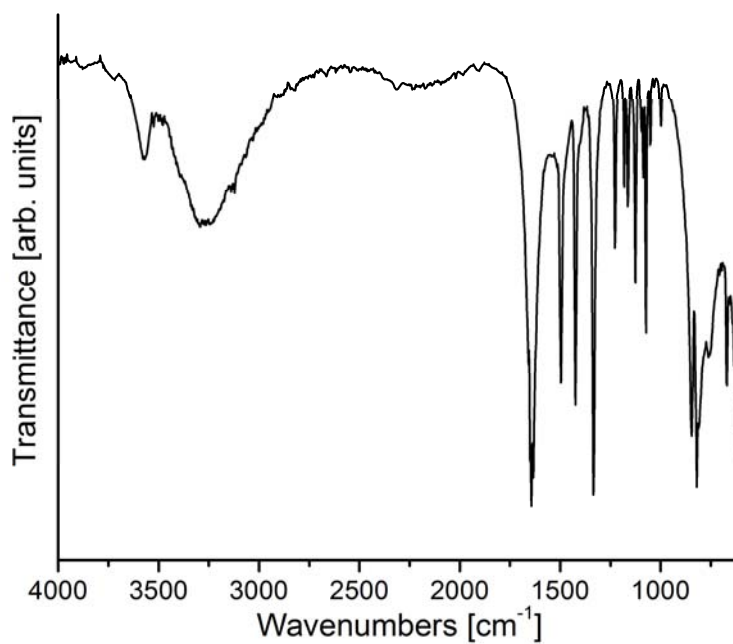


Fig. S6. IR spectroscopic data of compound **1**.

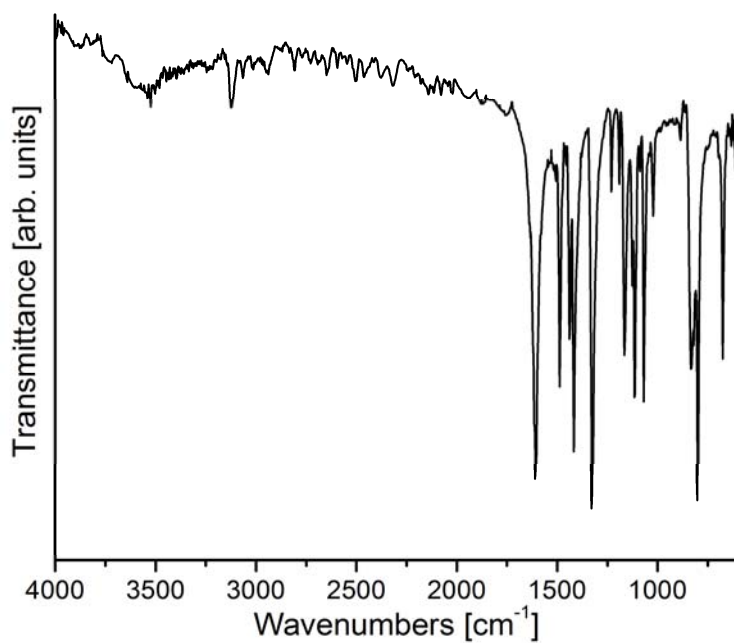


Fig. S7. IR spectroscopic data of compound 2.

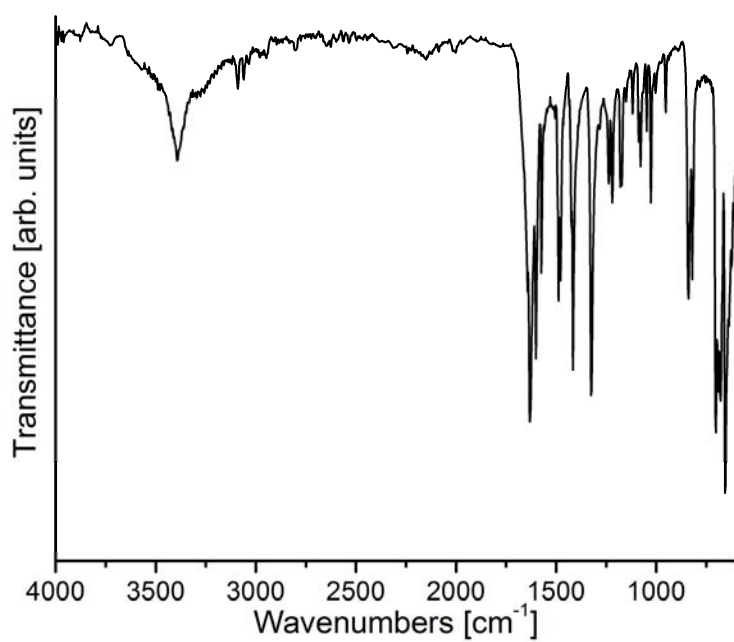


Fig. S8. IR spectroscopic data of compound 3.

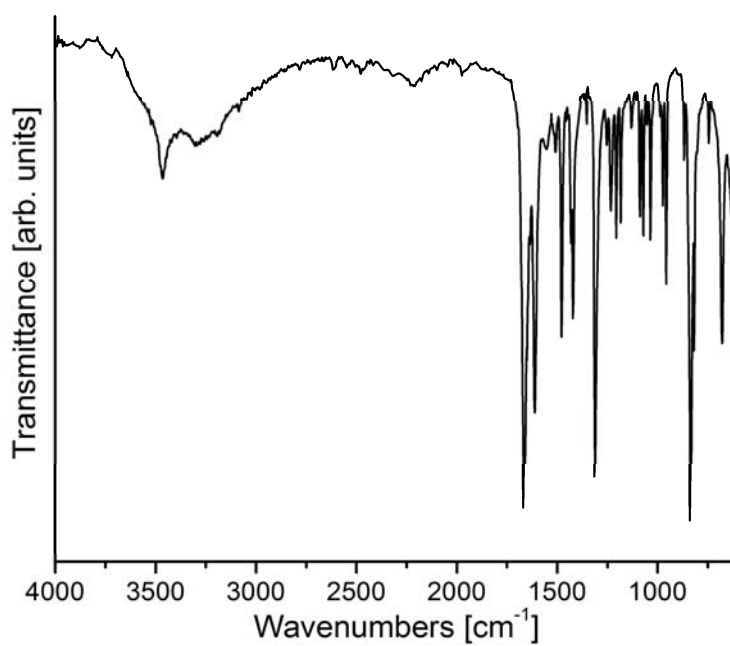


Fig. S9. IR spectroscopic data of compound **4**.

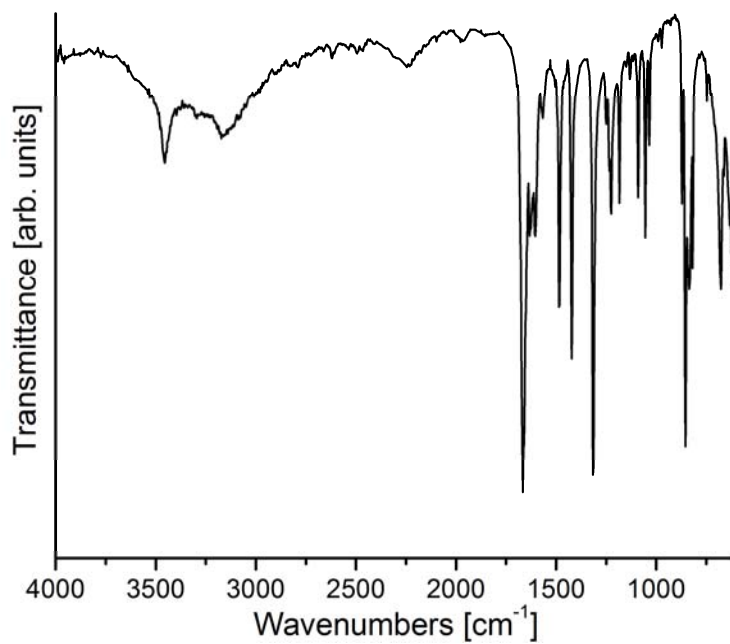


Fig. S10. IR spectroscopic data of compound **5**.

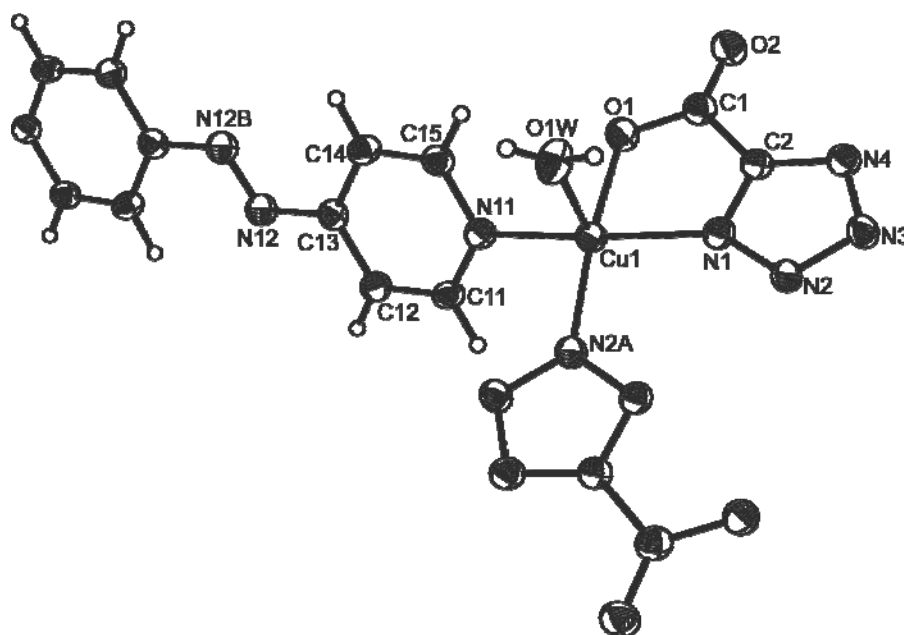


Fig. S11. Crystal structure of $[\text{Cu}(\text{tzc})(\text{azpy})_{0.5}(\text{H}_2\text{O})]_n$ (**5**) with view of the coordination sphere of the copper(II) cation with displacement ellipsoids drawn at the 50% probability level. Selected atoms are labeled. Symmetry codes: A = $-x, -y + 1, -z + 1$; B = $-x + 2, -y + 1, -z + 2$.

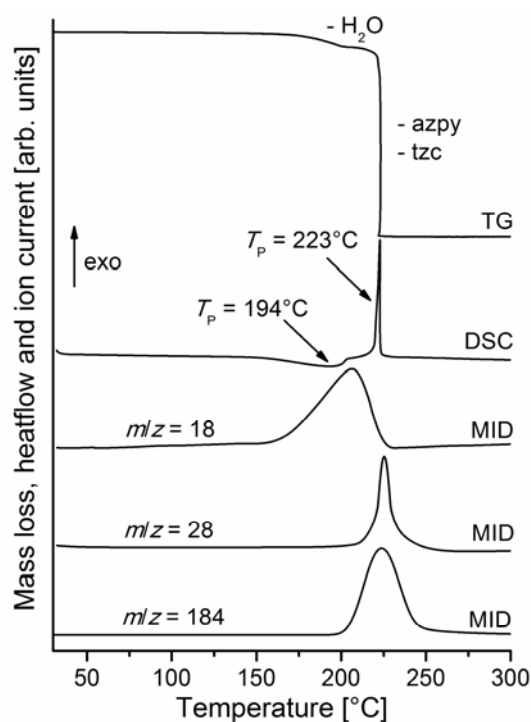


Fig. S12. TG, DSC and MS trend scan curves for compounds **5**. Heating rate = $3\text{ }^\circ\text{C}\cdot\text{min}^{-1}$; $m/z = 18$ (water), 28 (N_2 , decomposition product of tzc), 184 (azpy); given are the peak temperatures T_p [$^\circ\text{C}$].

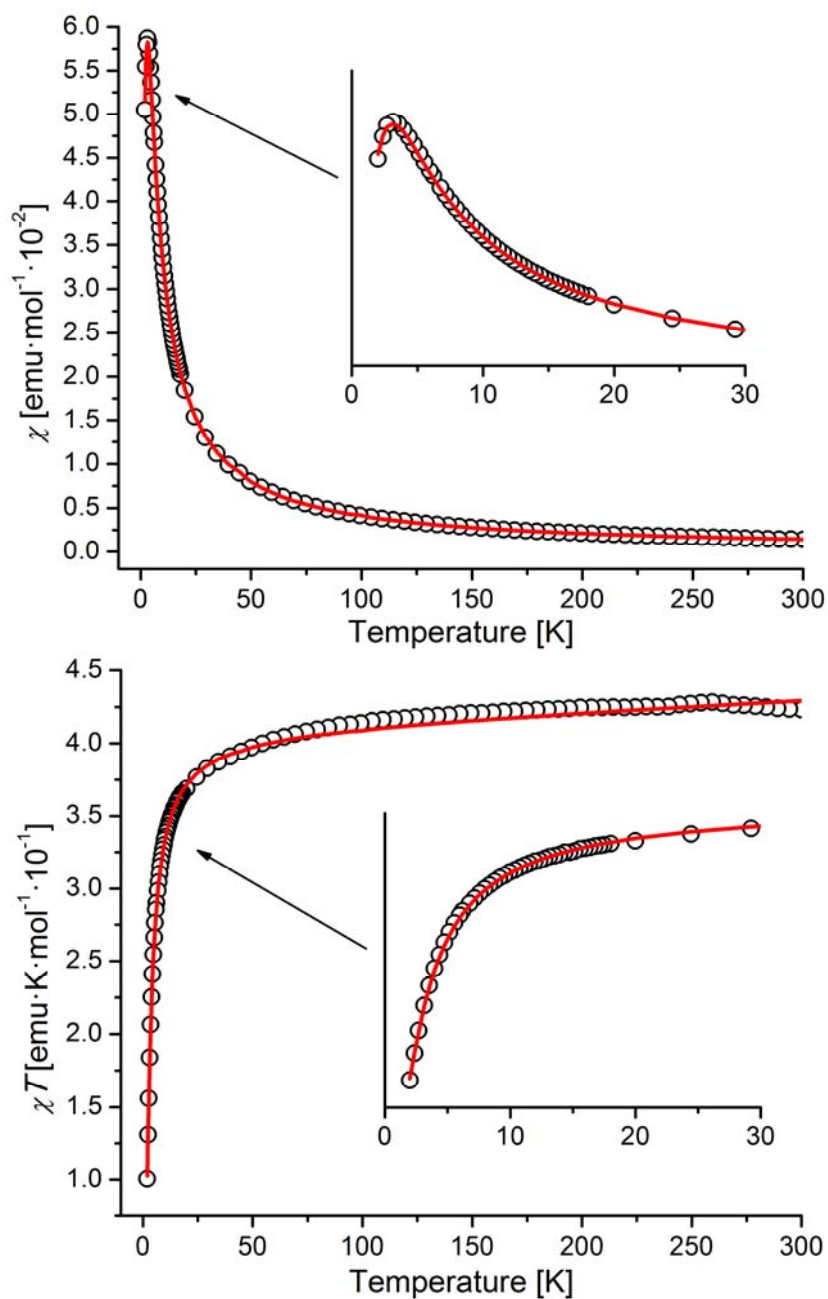


Fig. S13. Temperature dependence (from 300 to 1.8 K) of the magnetic susceptibility (top) and the product of the magnetic susceptibility and temperature (bottom) for compound **4**. The red lines correspond to the best fit of a magnetic dimer model.

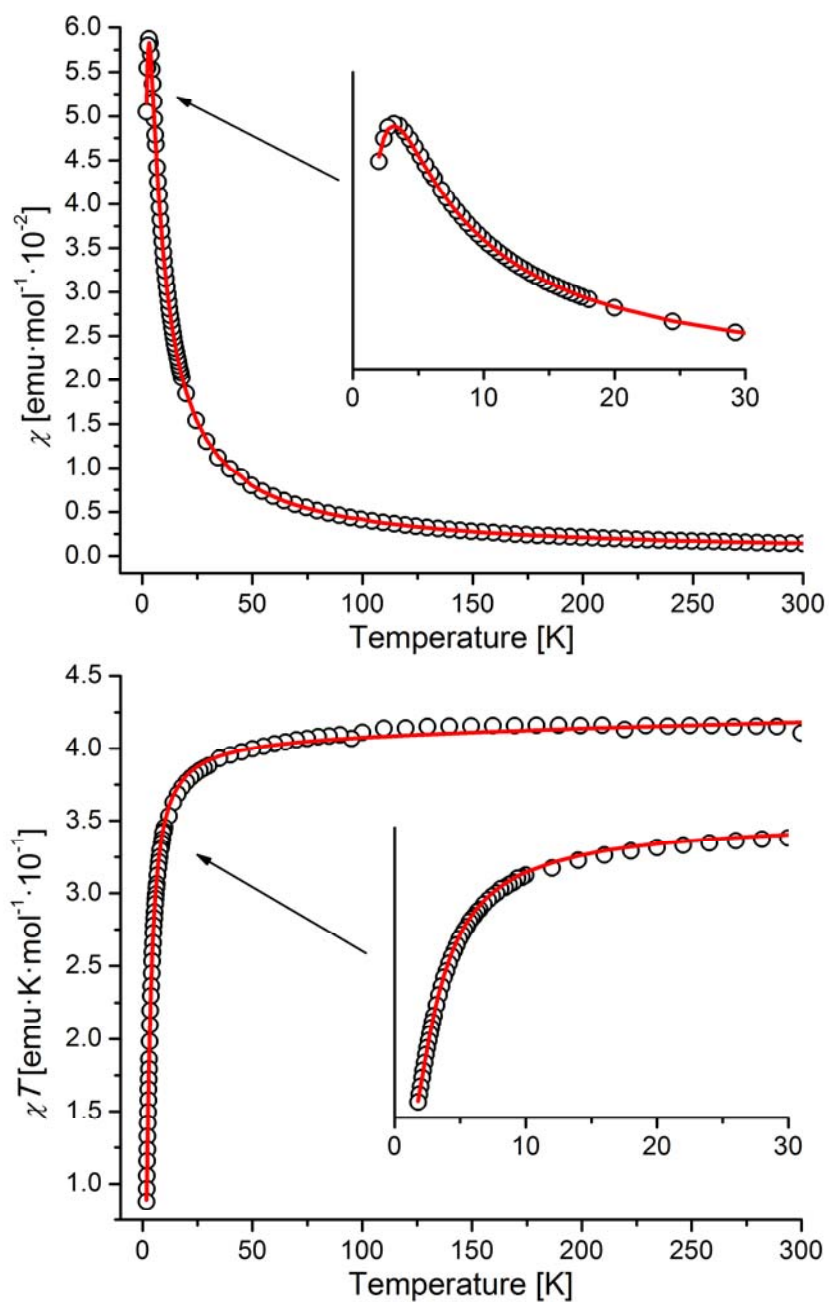


Fig. S14. Temperature dependence (from 300 to 1.8 K) of the magnetic susceptibility (top) and the product of the magnetic susceptibility and temperature (bottom) for compound **5**. The red lines correspond to the best fit of a magnetic dimer model.

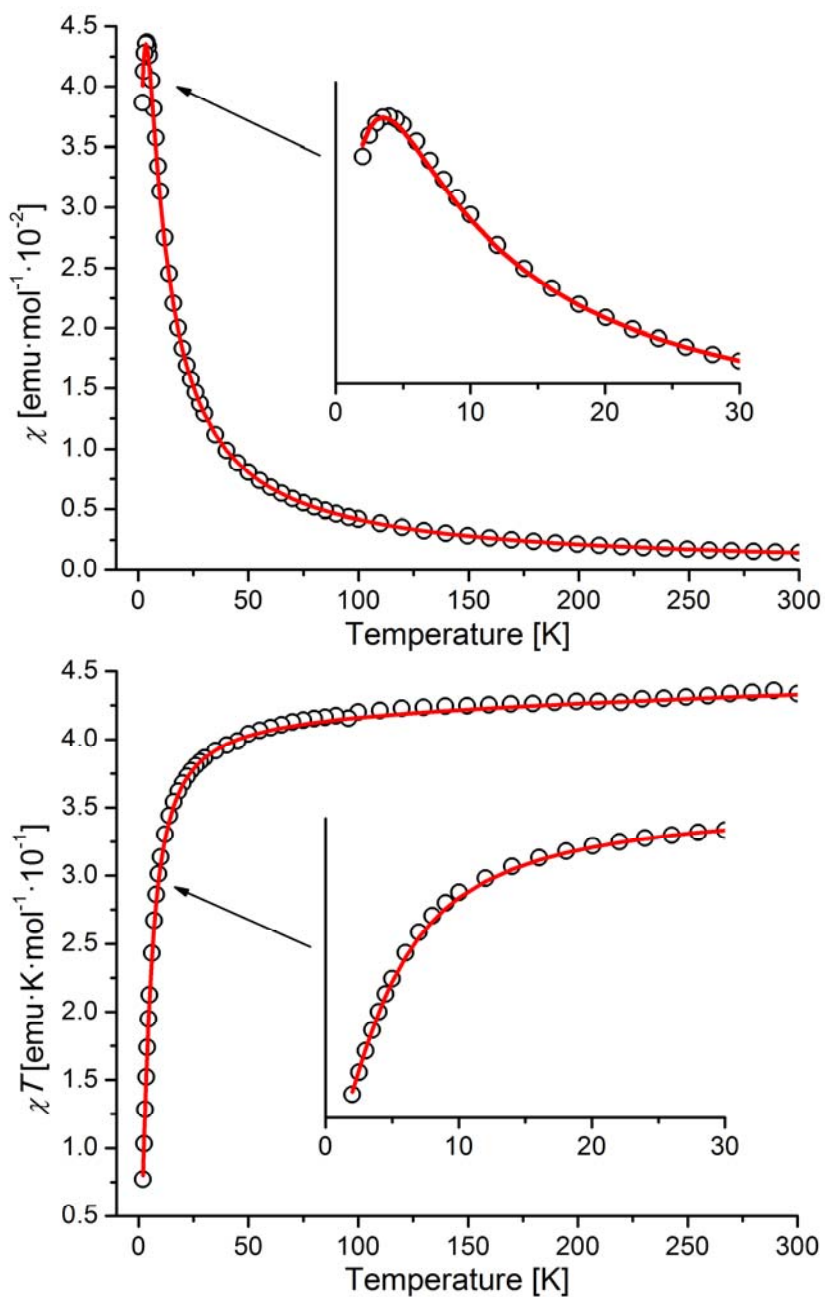


Fig. S15. Temperature dependence (from 300 to 1.8 K) of the magnetic susceptibility (top) and the product of the magnetic susceptibility and temperature (bottom) for compound **2**. The red lines correspond to the best fit of a magnetic chain model.

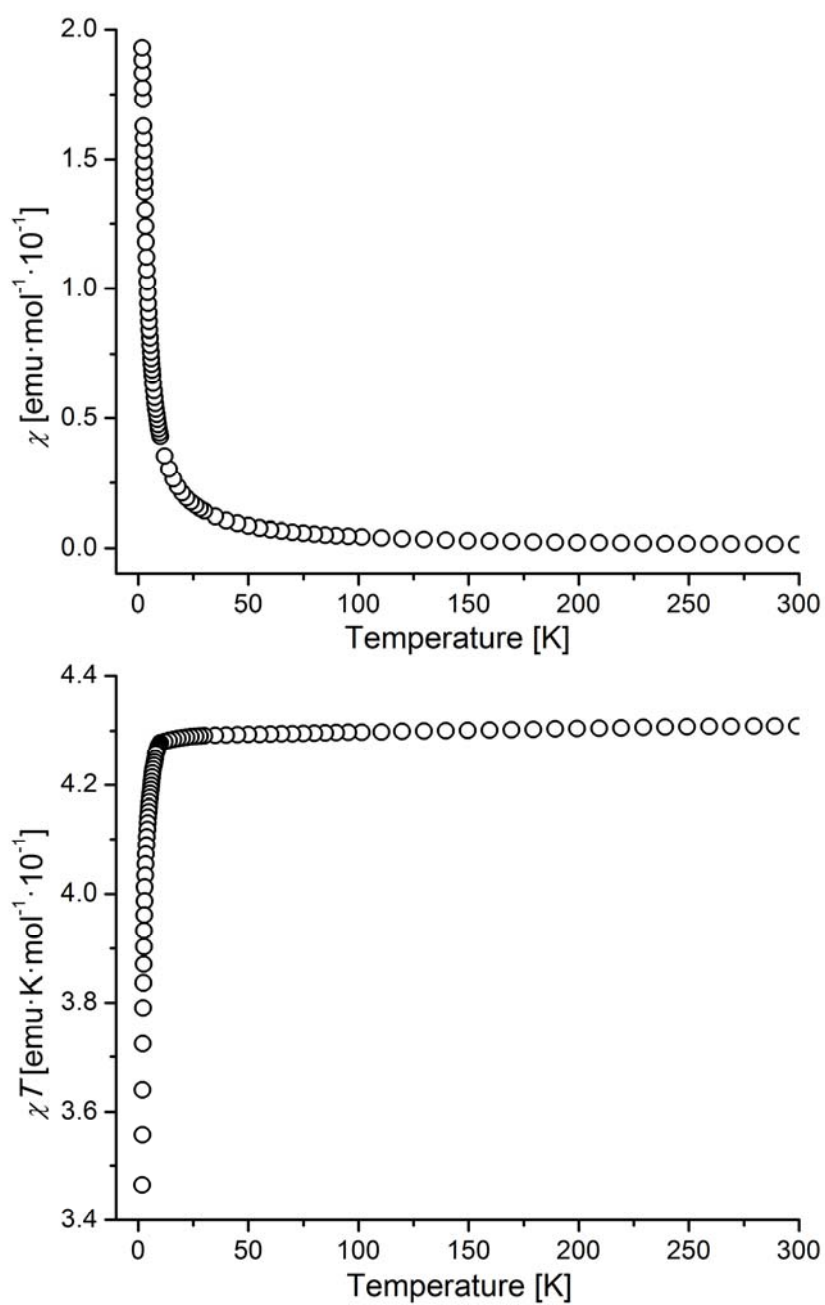


Fig. S16. Temperature dependence (from 300 to 1.8 K) of the magnetic susceptibility (top) and the product of the magnetic susceptibility and temperature (bottom) for compound **3**.

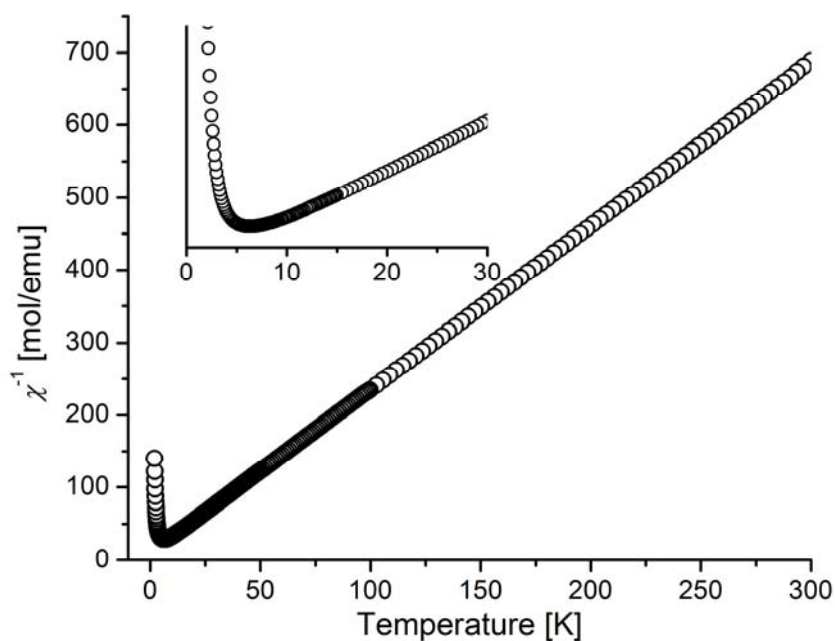


Fig. S17. Temperature dependence (from 300 to 1.8 K) of the reciprocal magnetic susceptibility for compound 1.

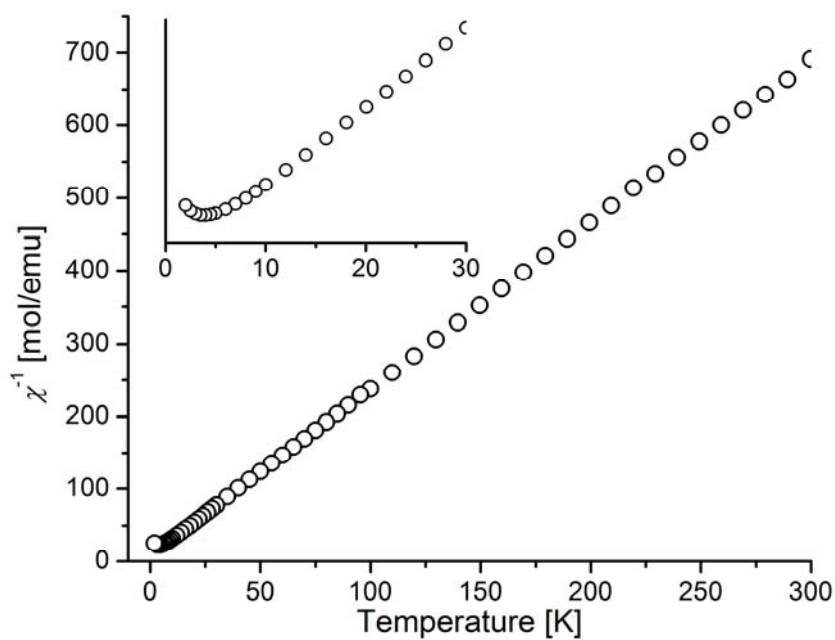


Fig. S18. Temperature dependence (from 300 to 1.8 K) of the reciprocal magnetic susceptibility for compound 2.

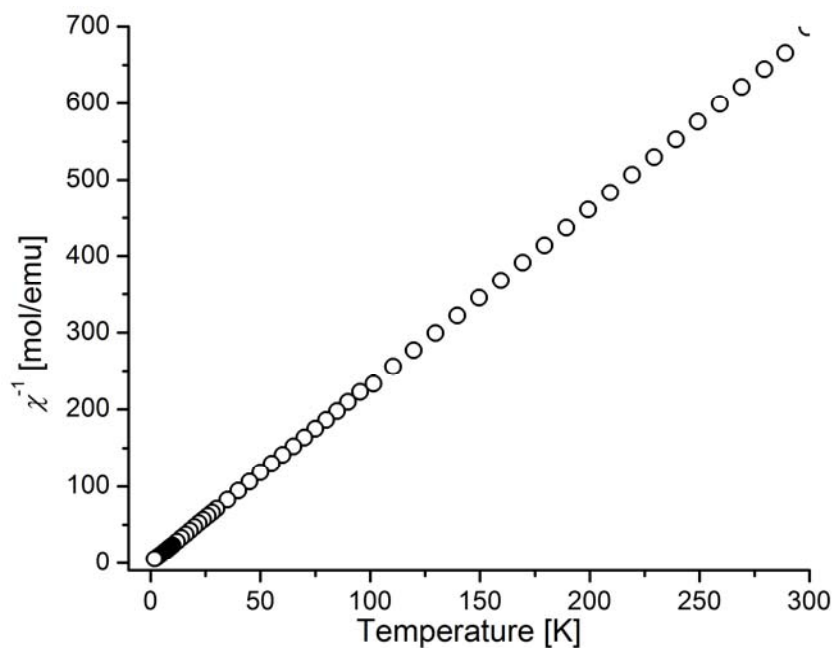


Fig. S19. Temperature dependence (from 300 to 1.8 K) of the reciprocal magnetic susceptibility for compound **3**.

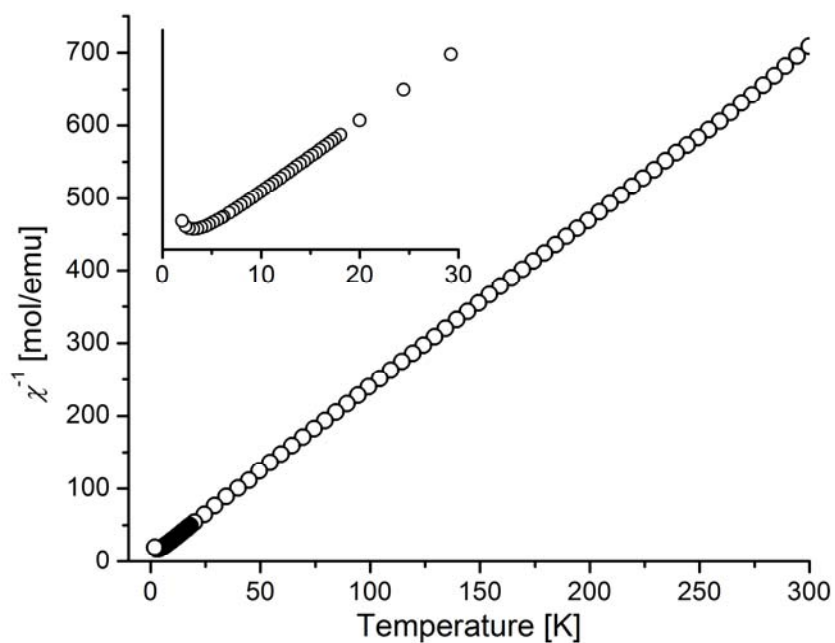


Fig. S20. Temperature dependence (from 300 to 1.8 K) of the reciprocal magnetic susceptibility for compound **4**.

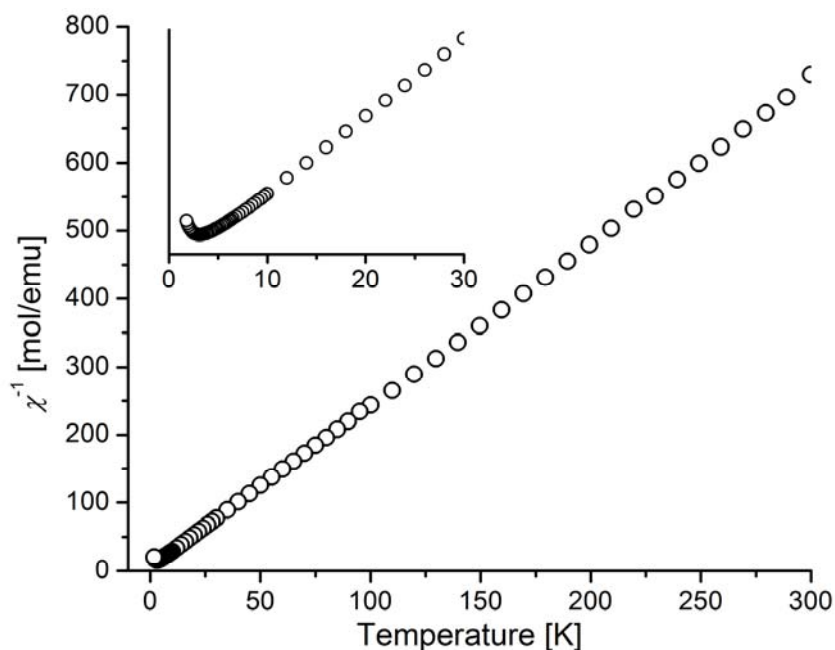


Fig. S21. Temperature dependence (from 300 to 1.8 K) of the reciprocal magnetic susceptibility for compound **5**.

Table S1. Selected bond lengths [Å] and angles [°] for compound **1**.

Cu(1)-O(2A)	1.980(2)
Cu(1)-N(2)	1.994(2)
Cu(1)-N(1A)	2.025(2)
Cu(1)-N(11)	2.032(2)
Cu(1)-O(2W)	2.387(2)
Cu(1)-O(1W)	2.421(2)
O(2A)-Cu(1)-N(2)	174.52(7)
O(2A)-Cu(1)-N(1A)	81.39(7)
N(2)-Cu(1)-N(1A)	93.33(7)
O(2A)-Cu(1)-N(11)	86.37(7)
N(2)-Cu(1)-N(11)	98.99(8)
N(1A)-Cu(1)-N(11)	167.40(7)
O(2A)-Cu(1)-O(2W)	89.00(6)
N(2)-Cu(1)-O(2W)	89.87(7)
N(1A)-Cu(1)-O(2W)	93.45(7)
N(11)-Cu(1)-O(2W)	89.27(7)
O(2A)-Cu(1)-O(1W)	91.71(6)
N(2)-Cu(1)-O(1W)	89.80(7)
N(1A)-Cu(1)-O(1W)	90.78(7)
N(11)-Cu(1)-O(1W)	86.61(7)
O(2W)-Cu(1)-O(1W)	175.77(5)

Symmetry codes: A = -x + 1, -y + 1, -z + 1; B = -x + 2, -y + 2, -z.

Table S2. Hydrogen bonding in compound **1**.

D-H	d(D-H)	d(H..A)	<DHA	d(D..A)	A
O1W-H1A	0.840	1.921	173.04	2.757	O2 [x, y, z - 1]
O1W-H1B	0.840	1.987	164.20	2.804	O3W [-x + 1, -y + 1, -z + 1]
O1W-H1C	0.840	2.004	171.11	2.837	O1W [-x + 1, -y + 2, -z]
O2W-H2A	0.840	1.915	163.50	2.731	O2 [x + 1, y, z - 1]
O2W-H2B	0.840	1.950	170.00	2.781	O2W [-x + 2, -y + 1, -z + 1]
O2W-H2C	0.840	1.972	150.63	2.735	O3W
O3W-H3A	0.840	2.157	152.90	2.930	N3 [x, y - 1, z]
O3W-H3A	0.840	2.581	122.07	3.110	N4 [x, y - 1, z]
O3W-H3B	0.840	2.041	150.84	2.804	O1W [-x + 1, -y + 1, -z + 1]
O3W-H3C	0.840	2.015	143.34	2.735	O2W

Table S3. Selected bond lengths [\AA] and angles [$^\circ$] for compound **2**.

Cu(1)-N(4A)	1.977(4)
Cu(1)-O(1)	1.983(4)
Cu(1)-O(2A)	2.485(3)
Cu(1)-N(11)	2.034(3)
Cu(1)-N(1)	2.364(4)
N(4A)-Cu(1)-O(1)	173.68(15)
N(4A)-Cu(1)-N(11)	90.59(8)
O(1)-Cu(1)-N(11)	89.28(8)
N(11)-Cu(1)-N(11B)	177.38(17)
N(4A)-Cu(1)-N(1)	104.59(16)
O(1)-Cu(1)-N(1)	81.73(15)
N(11)-Cu(1)-N(1)	90.98(8)

Symmetry codes: A = x + 1, y, z; B = x, -y + 1/2, z; C = x, -y + 3/2, z.

Table S4. Selected bond lengths [\AA] and angles [$^\circ$] for compound **3**.

Cu(1)-O(1)	2.004(2)
Cu(1)-N(2A)	2.006(2)
Cu(1)-N(1)	2.022(2)
Cu(1)-N(12B)	2.035(2)
Cu(1)-N(11)	2.377(2)
Cu(1)-O(1W)	2.408(2)
O(1)-Cu(1)-N(2A)	175.14(6)
O(1)-Cu(1)-N(1)	81.70(6)
N(2A)-Cu(1)-N(1)	93.46(7)
O(1)-Cu(1)-N(12B)	87.32(6)
N(2A)-Cu(1)-N(12B)	97.47(7)
N(1)-Cu(1)-N(12B)	168.19(6)
O(1)-Cu(1)-N(11)	87.00(6)
N(2A)-Cu(1)-N(11)	93.76(6)
N(1)-Cu(1)-N(11)	92.90(6)
N(12B)-Cu(1)-N(11)	90.84(6)

O(1)-Cu(1)-O(1W)	88.73(5)
N(2A)-Cu(1)-O(1W)	90.24(5)
N(1)-Cu(1)-O(1W)	83.39(5)
N(12B)-Cu(1)-O(1W)	92.07(5)
N(11)-Cu(1)-O(1W)	174.71(5)

Symmetry codes: A = -x + 1, -y + 1, -z + 1; B = -x + 1/2, y - 1/2, -z + 1/2.

Table S5. Hydrogen bonding in compound **3**.

D-H	d(D-H)	d(H..A)	<DHA	d(D..A)	A
O1W-H1W	0.840	1.983	171.98	2.817	O2 [-x + 3/2, y - 1/2, -z + 1/2]
O1W-H2W	0.840	1.977	169.71	2.808	O2 [x, y - 1, z]

Table S6. Selected bond lengths [Å] and angles [°] for compound **4** and **5**.

	Compound 4	Compound 5
Cu(1)-N(11)	1.979(2)	1.996(3)
Cu(1)-O(1)	1.980(2)	1.985(2)
Cu(1)-N(2A)	1.983(2)	1.992(3)
Cu(1)-N(1)	2.000(2)	2.003(3)
Cu(1)-O(1W)	2.218(2)	2.202(3)
N(11)-Cu(1)-O(1)	87.50(6)	87.87(10)
N(11)-Cu(1)-N(2A)	96.04(7)	95.89(11)
O(1)-Cu(1)-N(2A)	172.63(7)	172.81(11)
N(11)-Cu(1)-N(1)	165.50(7)	165.70(11)
O(1)-Cu(1)-N(1)	81.31(6)	81.54(10)
N(2A)-Cu(1)-N(1)	94.09(7)	93.68(11)
N(11)-Cu(1)-O(1W)	100.61(6)	98.72(11)
O(1)-Cu(1)-O(1W)	92.57(6)	93.73(11)
N(2A)-Cu(1)-O(1W)	93.13(7)	91.75(12)
N(1)-Cu(1)-O(1W)	89.15(6)	91.58(11)

Symmetry codes: A = -x, -y + 1, -z + 1; B = -x + 2, -y + 1, -z + 2.

Table S7. Hydrogen bonding in compound **4**.

D-H	d(D-H)	d(H..A)	<DHA	d(D..A)	A
O1W-H1W	0.840	2.134	152.45	2.905	N3 [x + 1, y, z]
O1W-H2W	0.840	1.981	163.06	2.795	O2 [x, -y + 3/2, z - 1/2]
O1W-H2W	0.840	2.592	138.18	3.267	O1 [x, -y + 3/2, z - 1/2]

Table S8. Hydrogen bonding in compound **5**.

D-H	d(D-H)	d(H..A)	<DHA	d(D..A)	A
O1W-H1W	0.840	2.099	154.18	2.879	N3 [x + 1, y, z]
O1W-H2W	0.840	1.932	163.54	2.748	O2 [x, -y + 3/2, z - 1/2]
O1W-H2W	0.840	2.570	137.47	3.239	O1 [x, -y + 3/2, z - 1/2]