

Table S1. Hydrogen bonds for $[\text{Cu}^{\text{II}}_2(\text{Alc})_4(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}$ (\AA and $^\circ$)

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
N(5)-H(5A)...O(57)#2	0.91	2.19	2.942(3)	139.4
N(5)-H(5A)...O(84)#2	0.91	2.51	3.200(3)	132.5
N(5)-H(5B)...O(1)#2	0.91	1.95	2.843(4)	168.4
N(5)-H(5C)...O(4)#3	0.91	2.11	2.963(3)	156.0
N(66)-H(66A)...O(55)#4	0.91	2.22	3.103(4)	164.0
N(66)-H(66A)...O(56)#4	0.91	2.35	3.024(4)	130.4
N(66)-H(66B)...O(5)#5	0.91	2.19	2.936(4)	138.3
N(66)-H(66B)...O(2)#6	0.91	2.64	3.288(4)	128.5
N(66)-H(66C)...O(9)#1	0.91	2.32	3.069(4)	139.2
N(66)-H(66C)...O(21)	0.91	2.30	3.058(4)	140.7
O(57)-H(57)...O(55)#4	0.84	1.77	2.541(3)	152.4
O(2)-H(2)...O(1)#3	0.84	1.85	2.510(3)	135.0
O(5)-H(5)...O(6)#7	0.84	1.87	2.539(3)	135.1
O(56)-H(56)...O(58)#1	0.84	1.77	2.535(3)	150.0
O(99)-H(1W)...O(84)#8	0.825(18)	2.36(2)	3.174(4)	168(4)
O(99)-H(2W)...O(6)#7	0.809(18)	2.00(2)	2.798(3)	170(4)
O(3)-H(3)...O(58)	0.84	1.98	2.813(3)	170.5

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+2,-z #2 -x+1,-y+2,-z+1 #3 -x+1, y+1/2, -z+1/2

#4 x,-y+3/2,z+1/2 #5 x-1,y,z #6 -x,y-1/2,-z+1/2

#7 -x+1,-y+2,-z #8 x,-y+3/2,z-1/2

Table S2. Hydrogen bonds for $[\text{Co}^{\text{II}}(\text{Pam})_2(\text{H}_2\text{O})_2]$ (\AA and $^\circ$)

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
N(1)-H(1C)...O(1)#2	0.89	2.35	3.038(7)	134.0
N(1)-H(1C)...O(4)#3	0.89	2.32	3.107(7)	147.0
N(1)-H(1A)...O(2)#4	0.89	2.14	2.977(7)	155.4
N(1)-H(1A)...O(7)#5	0.89	2.60	3.125(7)	118.9
N(1)-H(1B)...O(6)#6	0.89	2.30	3.085(6)	147.1
N(1)-H(1B)...O(3)#7	0.89	2.61	3.239(7)	128.2
O(7)-H(71)...O(6)#3	0.91(8)	1.89(8)	2.743(6)	156(7)
O(7)-H(72)...O(8)#4	0.84(9)	2.13(9)	2.889(6)	150(8)
O(7)-H(72)...O(5)#1	0.84(9)	2.51(8)	2.911(7)	111(7)
O(8)-H(8)...O(1)#8	0.82	2.14	2.850(6)	145.3
O(8)-H(8)...O(5)	0.82	2.68	3.124(6)	115.9
O(2)-H(2)...O(6)#9	0.82	2.11	2.509(6)	109.8
O(5)-H(5)...O(3)#8	0.82	1.86	2.473(6)	130.3

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+1,-z+1 #2 x+1,y,z #3 -x+1,-y+1,-z+1
#4 -x+1/2,y-1/2,-z+3/2 #5 x+1/2,-y+1/2,z+1/2
#6 -x+3/2,y-1/2,-z+3/2 #7 -x+1,-y+1,-z+2 #8 -x+1/2,y+1/2,-z+3/2
#9 x-1/2,-y+3/2,z+1/2

Table S3. Hydrogen bonds for [Ni(Pam)₂(H₂O)₂] (Å and °).

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(9)-H(14)...O(7)#2	0.98(7)	1.55(7)	2.522(3)	173(6)
O(6)-H(13)...O(8)	0.86(6)	2.75(6)	3.105(3)	106(4)
O(6)-H(13)...O(2)#3	0.86(6)	2.04(6)	2.840(3)	155(5)
O(3)-H(12)...O(7)#4	0.97(5)	1.81(5)	2.741(3)	160(4)
N(1)-H(1C)...O(1)#4	0.89	2.36	3.122(4)	143.4
N(1)-H(1C)...O(2)#5	0.89	2.30	3.032(4)	138.8
N(1)-H(1A)...O(7)#6	0.89	2.30	3.081(4)	147.2
N(1)-H(1B)...O(3)#7	0.89	2.65	3.148(4)	116.5
N(1)-H(1B)...O(9)#8	0.89	2.12	2.971(4)	159.4
O(3)-H(3)...O(8)#1	0.73(5)	2.51(5)	2.921(3)	117(5)
O(3)-H(3)...O(6)#8	0.73(5)	2.25(5)	2.887(3)	145(5)
O(8)-H15...O(5)#2	1.20(4)	1.27(4)	2.464(3)	171(5)

Symmetry transformations used to generate equivalent atoms:

#1 -x,-y+2,-z #2 x-1/2,-y+3/2,z+1/2 #3 -x+1/2,y-1/2,-z+1/2
#4 -x+1,-y+2,-z #5 x+1,y,z #6 -x+3/2,y+1/2,-z+1/2
#7 x+1/2,-y+5/2,z+1/2 #8 -x+1/2,y+1/2,-z+1/2

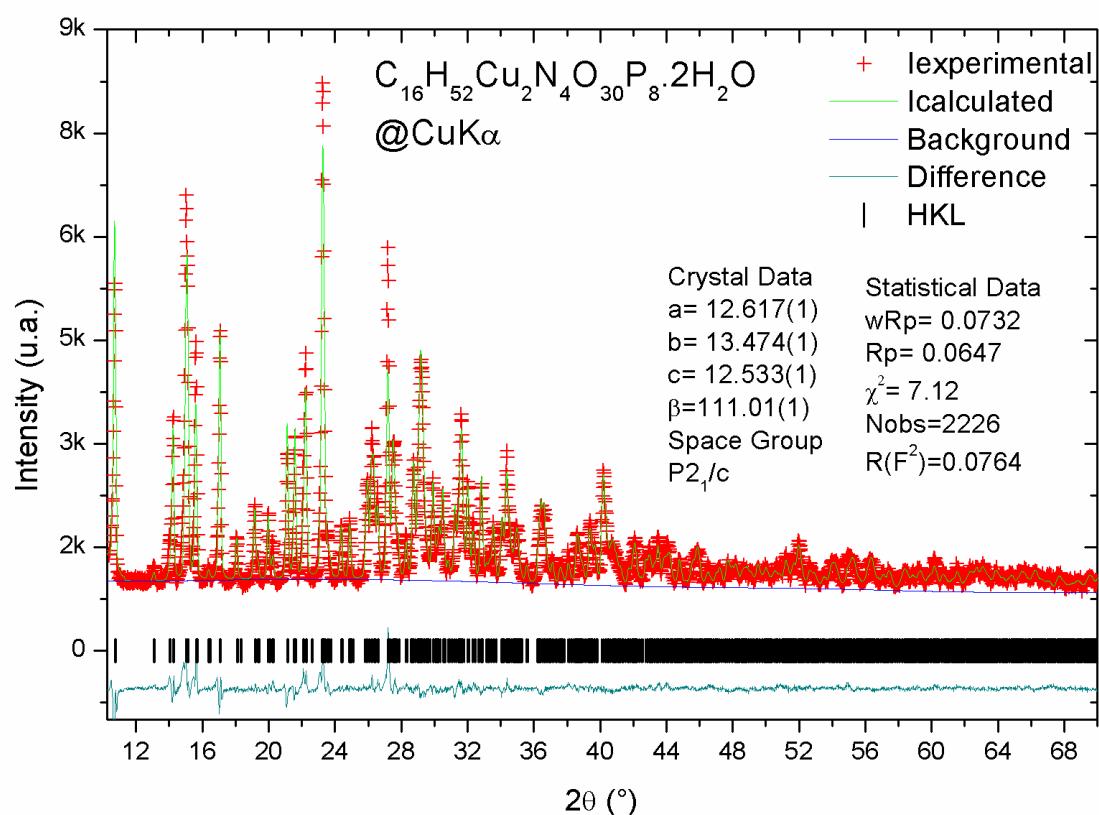


Figure S1- Rietveld refinement for the $\text{C}_{16}\text{H}_{52}\text{Cu}_2\text{N}_4\text{O}_{30}\text{P}_8 \cdot 2\text{H}_2\text{O}$ compound, indicating the experimental data, the calculated intensities, the background function, the difference curve and the Bragg positions.

Table S4. Level of BSA binding of the complexes after 48 h incubation

Compound	Interaction with BSA(%)	Compound	Interaction with BSA(%)
$[(\text{Cu}^{\text{II}}(\text{Pam})]\cdot\text{H}_2\text{O})_n$	98	$[\text{Cu}^{\text{II}}_2(\text{Ale})_4(\text{H}_2\text{O})_2]$	94
$[\text{Co}^{\text{II}}(\text{Pam})_2(\text{H}_2\text{O})_2]$	76	$[\text{Co}^{\text{II}}_2(\text{Ale})_4(\text{H}_2\text{O})_2]$	54
$[\text{Ni}^{\text{II}}(\text{Pam})_2(\text{H}_2\text{O})_2]$	91	$[\text{Ni}^{\text{II}}_2(\text{Ale})_4(\text{H}_2\text{O})_2]$	88
$[\text{Mn}^{\text{II}}(\text{Pam})_2(\text{H}_2\text{O})_2]$	0	$[\text{Mn}^{\text{II}}_2(\text{Ale})_4(\text{H}_2\text{O})_2]$	52