

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/materials

Supporting Information

Single Crystal X-ray Structure Study of the $\text{Li}_{2-x}\text{Na}_x\text{Ni}[\text{PO}_4]\text{F}$ System

Hamdi Ben Yahia,[‡] Masahiro Shikano,* Shinji Koike, Kuniaki Tatsumi and Hironori Kobayashi

Received (in XXX, XXX) Xth XXXXXXXXXX 2011, Accepted Xth XXXXXXXXXX 20XX

DOI: 10.1039/b000000x

Electronic Supplementary Information (ESI)

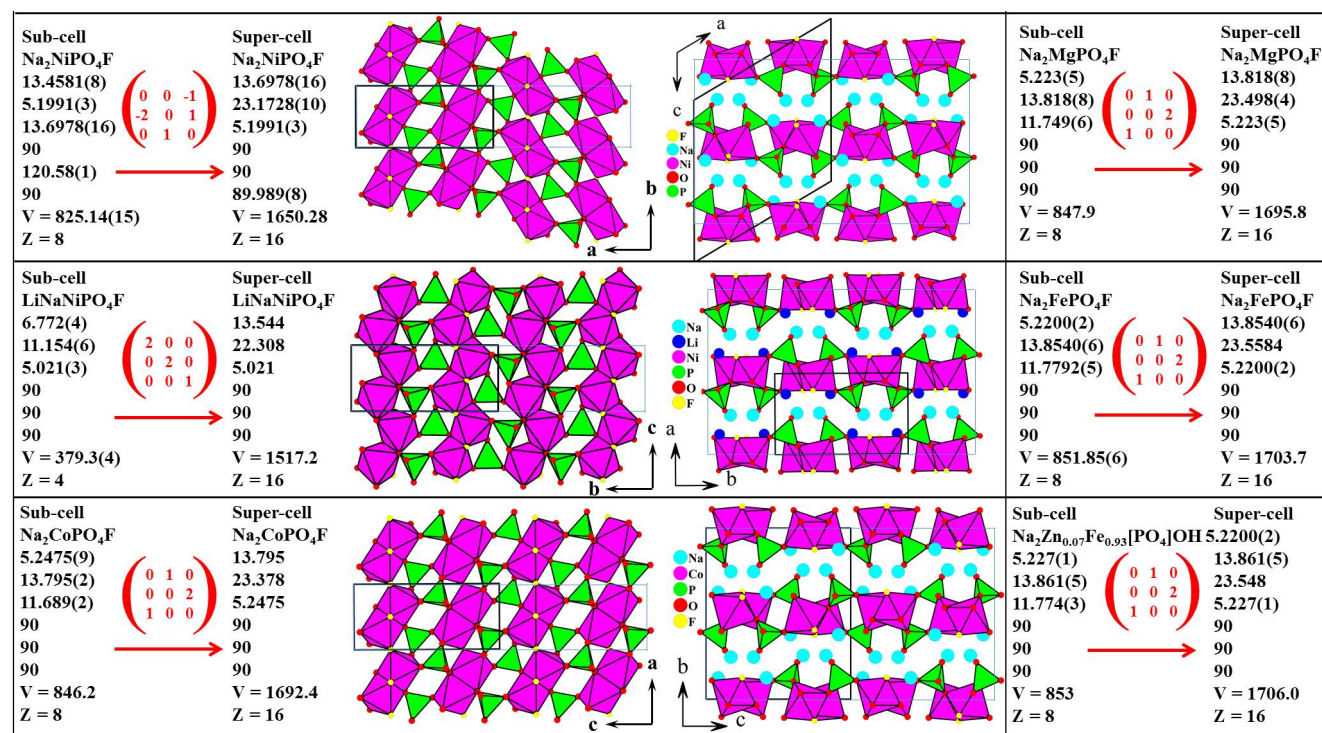


Fig. S1 Subcell-supercell transformations of the layered $A_2M[\text{PO}_4]\text{X}$ compound structures ($A = \text{Li}$, and Na ; $M = \text{Mg}$, Fe , Co , and Ni ; $X = \text{F}$, and OH). Black and blue lines indicate the sub-cell and the super-cell, respectively.

(a)				Li_{1.3}Na_{0.7}Ni[PO₄]F				(b)			
Bond valence for :		O1	2.088 (5)	Bond valence for :		O1	2.088 (5)			O1	2.088 (5)
		O2	2.115 (9)			O2	2.115 (9)			O2	2.115 (9)
		O3	2.005 (8)			O3	2.005 (8)			O3	2.005 (8)
		O4	2.032 (6)			O4	2.032 (6)			O4	2.032 (6)
		O5	1.882 (6)			O5	1.882 (6)			O5	1.882 (6)
		O6	2.002 (5)			O6	2.002 (5)			O6	2.002 (5)
		O7	1.286 (3)			F1	1.018 (2)			F1	1.018 (2)
		O8	1.192 (5)			F2	0.963 (4)			F2	0.963 (4)

LiNaNi[PO₄]F			
Bond valence for :		O1	1.964 (14)
		O2	2.012 (13)
		O3	1.908 (13)
		O4	2.096 (14)
		O5	1.356 (8)

Na₂Ni[PO₄]F			
Bond valence for :		O1	1.96 (2)
		O2	1.93 (2)
		O3	2.06 (3)
		O4	2.01 (4)
		O5	1.98 (4)
		O6	2.08 (4)
		O7	2.01 (4)
		O8	2.07 (2)
		O9	1.526 (13)
		O10	1.424 (12)

Fig. S2. The bond valence sum of the anions at the early stage of the crystal structure refinement (a), and in the final crystal structure (b). The arrows indicate the anions that have been attributed to be oxygen atoms, but in reality, in the final crystal structure, they correspond to Fluorine atoms.

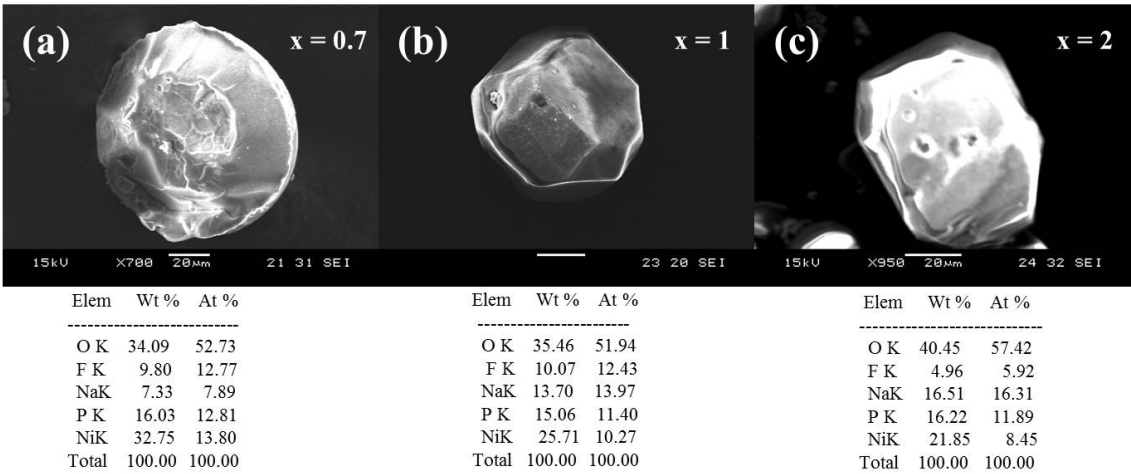


Fig. S3. SEM images of the $\text{Li}_{2-x}\text{Na}_x\text{Ni}[\text{PO}_4]\text{F}$ ($x = 0.7, 1$, and 2) single crystals used for the XR data collection. The results of the EDX analyses are also provided.

Table S1. Anisotropic displacement parameters ($\text{\AA}^2$) for $\text{Li}_{2-x}\text{Na}_x\text{Ni}[\text{PO}_4]\text{F}$ compounds ($x = 0.7, 1, \text{ and } 2$). The anisotropic displacement factor exponent takes the form: $-2\pi^2[(ha^*)^2U_{11}+...+2hka^*b^*U_{12}]$.

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
$\text{Li}_{1.3}\text{Na}_{0.7}\text{Ni}[\text{PO}_4]\text{F}$						
Na1/Li1	0.0221(7)	0.0180(8)	0.0169(7)	0.0057(5)	0.0048(5)	0.0031(5)
Na2/Li2	0.015(2)	0.011(2)	0.026(2)	0	0.0035(15)	0
Na3/Li3	0.021(3)	0.013(3)	0.0128(19)	0	-0.002(2)	0
Ni1	0.0066(2)	0.0066(2)	0.0082(2)	-0.00018(12)	-0.00071(10)	0.00028(12)
Ni2	0.0085(2)	0.0060(2)	0.0074(2)	0.00042(12)	-0.00061(11)	-0.00006(12)
P1	0.0111(4)	0.0066(3)	0.0056(3)	0	0.0009(2)	0
P2	0.0057(3)	0.0072(3)	0.0100(3)	0	0.0001(2)	0
O1	0.0081(6)	0.0083(6)	0.0259(7)	0.0001(5)	-0.0042(6)	-0.0001(6)
O2	0.0340(13)	0.0236(13)	0.0118(9)	0	0.0007(8)	0
O3	0.0129(11)	0.0221(12)	0.0152(9)	0	0.0040(7)	0
O4	0.0074(9)	0.0079(10)	0.0251(9)	0	0.0032(8)	0
O5	0.0143(9)	0.0087(10)	0.0054(8)	0	-0.0009(6)	0
O6	0.0232(8)	0.0077(7)	0.0075(6)	0.0009(5)	-0.0006(5)	0.0004(6)
F1	0.0091(7)	0.0079(8)	0.0157(7)	0	-0.0003(6)	0
F2	0.0188(8)	0.0104(8)	0.0100(6)	0	-0.0021(6)	0
$\text{LiNaNi}[\text{PO}_4]\text{F}$						
Li	0.009(3)	0.006(2)	0.004(3)	0.0029(17)	-0.005(4)	-0.0030(18)
Na	0.0140(8)	0.0140(7)	0.0166(9)	0.0026(5)	-0.0015(13)	0.0005(6)
Ni	0.0059(2)	0.0049(2)	0.0060(2)	-0.00009(13)	0.0005(2)	-0.00021(16)
P	0.0050(4)	0.0041(4)	0.0056(4)	-0.0003(2)	-0.0004(5)	0.0000(3)
O1	0.0079(11)	0.0123(10)	0.0152(12)	-0.0029(8)	-0.0004(13)	-0.0013(10)
O2	0.0141(14)	0.0081(10)	0.0078(12)	-0.0058(8)	-0.0018(10)	0.0019(8)
O3	0.0085(13)	0.0102(11)	0.0072(11)	0.0017(9)	0.0006(10)	0.0005(8)
O4	0.0098(12)	0.0054(10)	0.0103(11)	-0.0016(9)	0.0014(10)	-0.0008(9)
F	0.0068(10)	0.0121(9)	0.0116(9)	-0.0023(7)	-0.0004(8)	0.0019(8)
$\text{Na}_2\text{Ni}[\text{PO}_4]\text{F}$						
Na1	0.009(2)	0.0206(18)	0.019(3)	-0.0002(10)	0.007(2)	0.0032(11)
Na2	0.014(2)	0.0151(14)	0.015(3)	-0.0015(13)	0.006(2)	-0.0010(11)
Na3	0.022(3)	0.0193(17)	0.016(3)	0.0010(10)	0.013(2)	0.0008(10)
Na4	0.013(2)	0.0203(15)	0.009(3)	0.0028(12)	0.006(2)	0.0016(10)
Ni1	0.0093(5)	0.0089(3)	0.0074(5)	-0.0003(3)	0.0040(4)	-0.0002(3)
Ni2	0.0056(5)	0.0106(3)	0.0072(5)	0.0001(3)	0.0028(4)	0.0003(4)
P1	0.0088(10)	0.0083(7)	0.0090(10)	0.0003(8)	0.0066(8)	0.0004(9)
P2	0.0045(9)	0.0103(6)	0.0078(10)	-0.0005(8)	0.0041(8)	0.0008(9)
O1	0.020(2)	0.0102(17)	0.016(2)	-0.0020(16)	0.0128(19)	-0.0015(17)
O2	0.019(2)	0.0079(18)	0.013(2)	0.0034(16)	0.0091(19)	0.0010(17)
O3	0.015(2)	0.0090(19)	0.016(2)	0.0008(14)	0.0120(18)	-0.0033(16)
O4	0.010(4)	0.016(3)	0.003(3)	-0.0001(15)	-0.001(3)	-0.0001(16)
O5	0.013(3)	0.019(2)	0.010(3)	-0.0007(19)	0.005(3)	-0.006(2)
O6	0.015(3)	0.020(2)	0.004(3)	-0.0001(19)	0.003(3)	0.001(2)
O7	0.003(3)	0.017(3)	0.006(4)	0.0014(14)	-0.003(3)	0.0019(15)
O8	0.012(2)	0.0101(18)	0.014(2)	0.0014(15)	0.0057(18)	0.0023(16)
F1	0.0124(14)	0.0053(12)	0.017(2)	-0.0006(10)	0.007(2)	-0.0003(18)
F2	0.0125(14)	0.0068(11)	0.014(2)	0.0012(9)	0.0033(19)	0.0002(16)