

Electronic Supplementary Information (ESI)

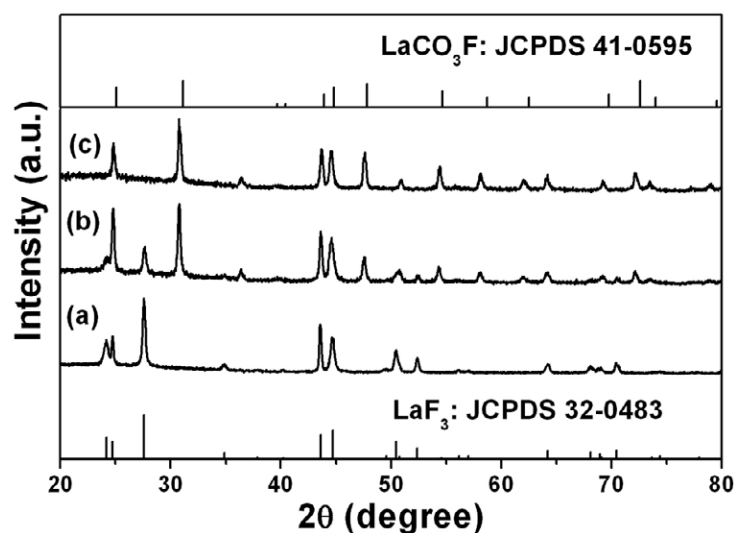


Figure S1. XRD patterns of the as-prepared products using NaBF₄ as the F⁻ source at different amount of Na₂CO₃: (a) 8 mmol, (b) 12 mmol, (c) 14 mmol. The standard data of hexagonal LaF₃ (JCPDS No. 32-0483) and hexagonal LaCO₃F (JCPDS No. 41-0595) are used as references.

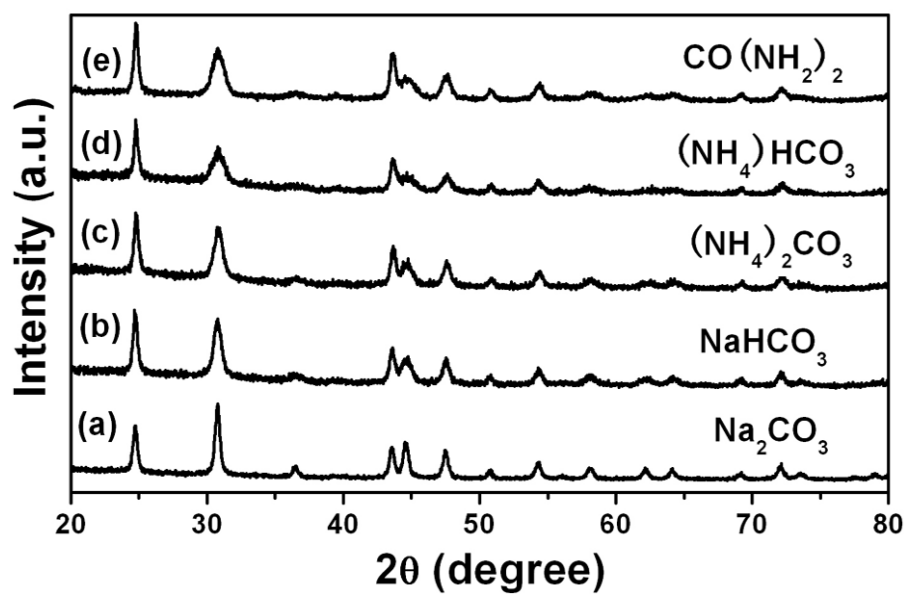


Figure S2. The XRD patterns of as-prepared LaCO_3F products at pH = 9 using different carbon sources: (a) Na_2CO_3 , (b) NaHCO_3 , (c) $(\text{NH}_4)_2\text{CO}_3$ (d) NH_4HCO_3 , (e) $\text{CO}(\text{NH}_2)_2$.

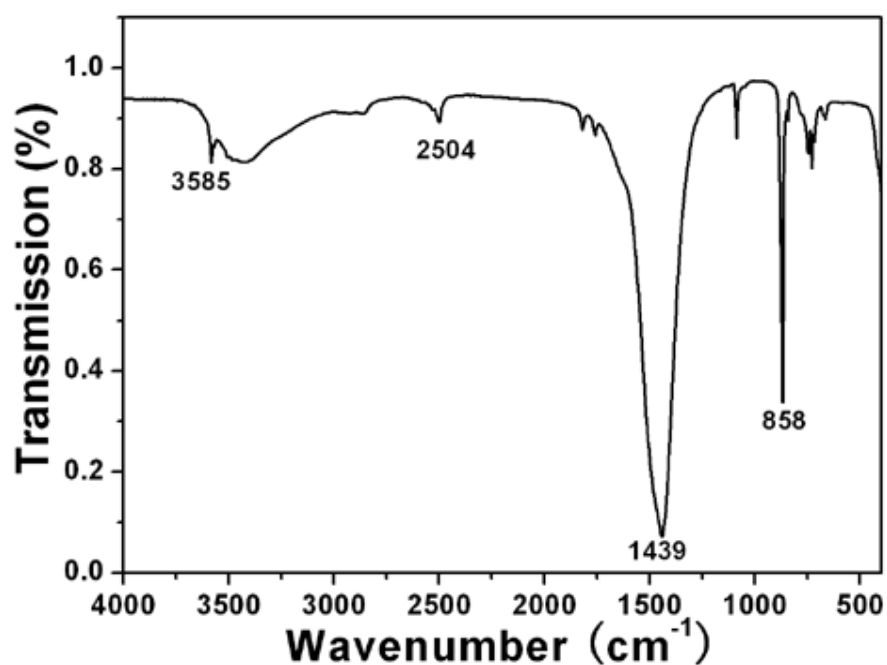


Figure S3. FT-IR spectrum of the as-synthesized LaCO_3F product prepared at $\text{pH} = 9$ using Na_2CO_3 as carbon source.

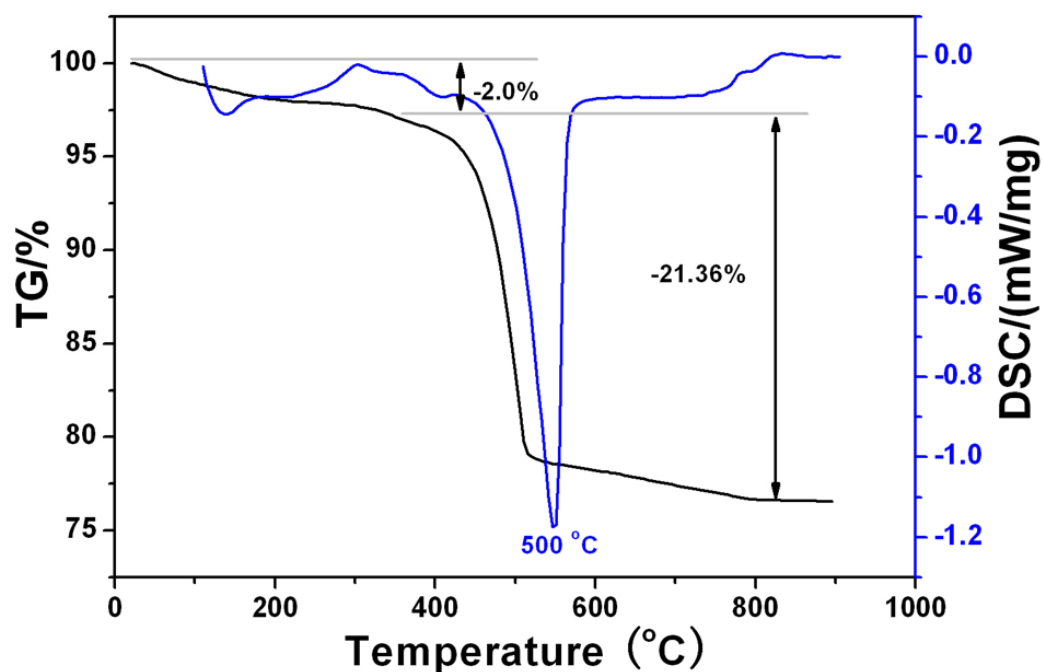
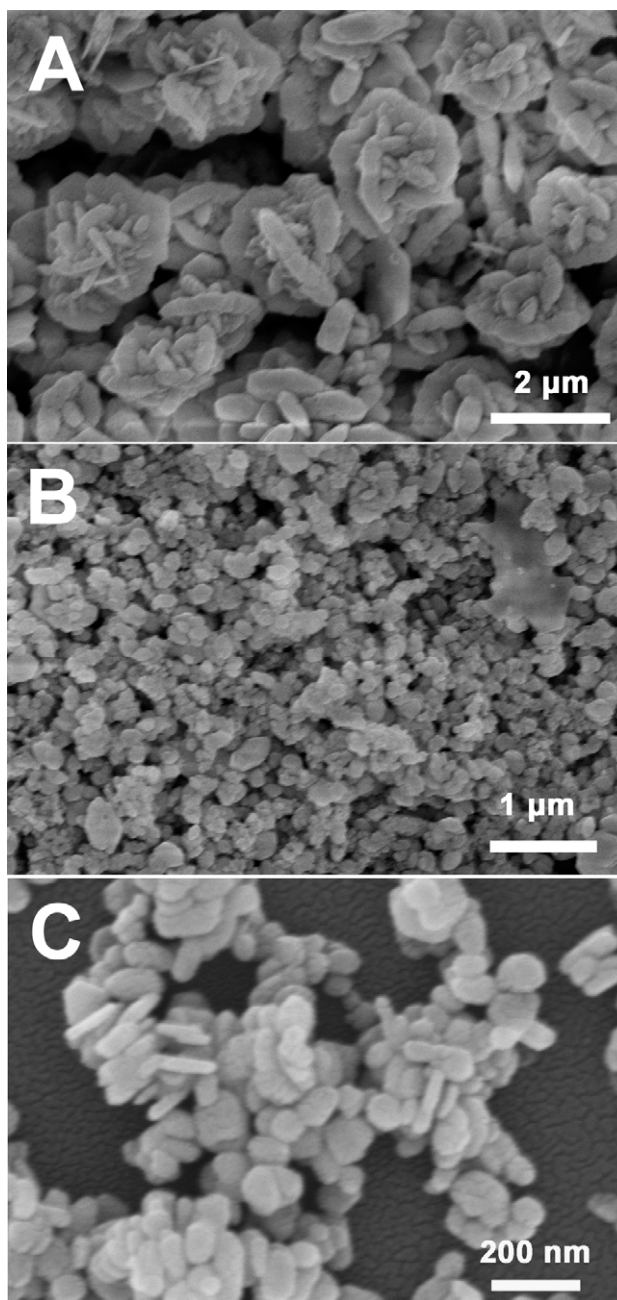


Figure S4. TG-DSC curves of as-prepared LaCO_3F precursor at pH = 9 using Na_2CO_3 as carbon source.



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Figure S5. SEM images of the as-prepared products using NaBF_4 as the F^- source at different amount of Na_2CO_3 : (A) 14 mmol, LaCO_3F ; (B) 12 mmol, $\text{LaF}_3 + \text{LaCO}_3\text{F}$; (C) 8 mmol, LaF_3 .

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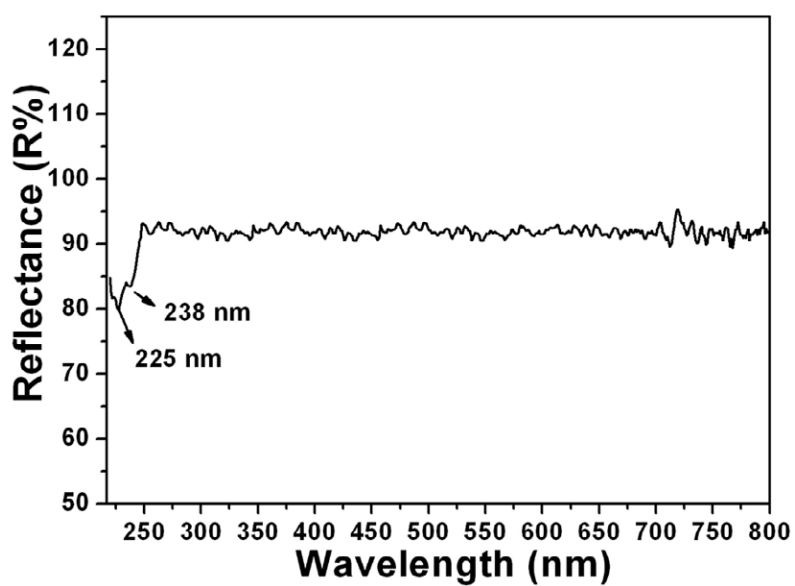


Figure S6. Diffuse reflection spectrum of LaOF product obtained by calcining LaCO_3F at 600 °C for 4 h.