Electronic Supplementary Information

A Manganese Oxido Complex Bearing Facially Coordinating Trispyridyl Ligands – Is Coordination Geometry Crucial for Water Oxidation Catalysis?

Hans-Martin Berends,^a Anne-Marie Manke,^a Christian Näther,^a Felix Tuczek^{*a} and Philipp Kurz^{*a}



Fig. SI 1 Resonance Raman spectra recorded with different excitation wavelengths.



Fig. SI 2 Plots of inverse magnetic susceptibility and effective magnetic moment per dinuclear complex versus temperature at H = 0.1 T for a polycrystalline sample of complex 1.



Fig. SI 3 Plot of χT vs. T at H = 0.1 T for a polycrystalline sample of complex **1**. The solid line shows a least square fit to the Bleaney-Bowers equation for a $S_1 = S_2 = 2$ dimer resulting in weak antiferromagnetically coupling (J = -0.25 cm⁻¹, H = -2JS_1S_2).¹



Fig. SI 4 Cyclic voltammogram of the ligand tpdm measured in acetonitrile.



Fig. SI 5 Cyclic voltammogram of 1 measured with different scan rates.



Fig. SI 6 Changes observed for absorption spectra of 1 (2.5 mM in acetonitrile / 0.1 M TBAClO₄) in an OTTLE cell (1mm path length) recorded during the electrochemical oxidation of the complex at a potential of +1 200 mV vs. Ag/Ag^+ and following re-reduction at 0 mV.



Fig. SI 7 UV/Vis spectra of 1 recorded after oxidation with *t*BuOOH.



Fig. SI 8 Changes observed for absorption spectra of 1 (2.5 mM in acetonitrile / 0.1 M TBAClO₄) in an OTTLE cell (1mm path length) recorded during the electrochemical reduction of the complex at a potential of -500 mV vs. Ag/Ag^+ and following re-oxidation at +500 mV.



Fig. SI 9 EPR spectrum of **1** recorded after electrochemical oxidation at a potential of 1450 mV vs. Ag/Ag⁺.



Fig. SI 10 Experimental (solid line) and simulated (dotted line) X-band EPR spectra for **1** oxidized chemically with five equivalents of *t*BuOOH.



Fig. SI 11 X-band EPR spectra for 1 reduced electrochemically at -400 mV vs. Ag/Ag⁺.

Experiment Title	G4b 5K 2mW DM*1000 esp
Operator	xuser
Spin Hamiltonian =	(Mn : beta B.g.S + S.A.I[Mn] + S.A.I[Mn1]).;
Experiment Type	Continuous Wave
Spectrum Type	Randomly Oriented Spectrum
Method	Matrix Diagonalisation
Superhyperfine Method	Matrix Diagonalisation
Spin System	
Number of Paramagnetic Centres	1
Atomic Element for Electron Spin	Mn
Electron Spin	0.5
Number of Nuclei	2
Atomic Element for Nuclear Spin	Mn
Number of Equivalent Nuclei	1
Number of Isotopes	1
Nuclear Spin - Mn-55	2.5
Nuclear g value - Mn-55	1.3874800
Natural Abundance - Mn-55	100.0000%
Atomic Element for Nuclear Spin	Mn1
Number of Equivalent Nuclei	1
Number of Isotopes	1
Nuclear Spin - Mn1-55	2.5
Nuclear g value - Mn1-55	1.38/4800
Natural Abundance - Mn1-55	100.0000%
Instrument Parameters	
Abscissa - 1	Magnetic Field
Center Field [G]	3500.000200
Sweep Width [G]	2000.000000
Number of Points	5000
Abscissa - 2	None
Temperature [K]	0.000000
Microwave Frequency [GHz]	9.644088
Harmonic	First
Phase	Normal
Units	Gauss
Cavity Mode	Perpendicular
Magnetic Field - 1 Theta [deg]	0.000000
Magnetic Field - 1 Phi [deg]	90.000000
Magnetic Field - 2 Theta [deg]	90.000000
Magnetic Field - 2 Phi [deg]	90.000000
File/Execution Information	
Sophe Input File Version	1.0.1
Sophe Output File Version	1.0.1
Rewrite Input File [y/n]	n
Input Spectrum Directory	/home/xuser//echem
Input Spectrum Filename	G4b_5K_2mW.par
Output Spectrum 10	Xepr
Output Spectrum Eilenome	/nome/xuser//ecnem
Input Format	echelliu Bruker ESD
Output Format	Bruker FSP
Sulput Pormai	DIUNUI LOI

Table SI 1 X-Sophe parameters for the $Mn_2^{III,IV}$ form of **1** obtained by electrochemical
oxidation.

Hostna Execu Batch	ame tion Mod File Nan	le ne			ESP Single Int File	eractive				
Sophe Numb Numb Interp Symm	Paramet er of The er of Fiel olation S netry	ers eta Orienta d Segmen tep [G]	tions ts		25 10 1.000000 Orthorhor	nbic				
Line S Line S Electric Line S Line S Line w	Shape Par Shape Mo on Spin / Shape Shape Cut ridth Unit	ameters del Nucleus toff [No. S	td. Dev.]		Angular E Mn Mn Gaussian 6.000000 Gauss	Dependence	of g			
Transi Transi Numb	tion Prob tion Thre er of Tra	babilities eshold nsitions			0.010000 0					
Optim Metho Error Error Norma Total Outpu Outpu Outpu	isation P od Function alisation Number of t Results t Parame t Spectra	arameters Method of Iteratior Every N I ters To To	ns terations		Simplex Raw Data 6.0000000 Peak Extr 1000 25 Disk Disk	e-03 ema				
Matriz Tenso	r	Values		Devi	ation		Paramet Half W	er Space /idth		Means of Varying Par.
Mn-g	1.99811 0.00000 0.00000	0.00000 2.00369 0.00000	0.00000 0.00000 1.98482	0.500000 0.00000 0.00000	0.00000 0.500000 0.00000	0.00000 0.00000 0.500000	0.500000 0.00000 0.00000	0.00000 0.500000 0.00000	0.00000 0.00000 0.500000	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
Mn-A	-149.992 0.00000 0.00000	0.00000 -168.994 0.00000	0.00000 0.00000 -107.989	2.00000 0.00000 0.00000	0.00000 2.00000 0.00000	0.00000 0.00000 2.00000	20.0000 0.00000 0.00000	0.00000 20.0000 0.00000	0.00000 0.00000 20.0000	$\begin{array}{cccc} 4 & 0 & 0 \\ 0 & 5 & 0 \\ 0 & 0 & 6 \end{array}$
Mn1-A	82.9967 0.00000 0.00000	0.00000 72.0060 0.00000	0.00000 0.00000 79.9954	$\begin{array}{c} 1.00000\\ 0.00000\\ 0.00000\end{array}$	0.00000 1.00000 0.00000	0.00000 0.00000 1.00000	20.0000 0.00000 0.00000	0.00000 20.0000 0.00000	0.00000 0.00000 20.0000	7 0 0 0 8 0 0 0 9
Linewic Linewic Linewic	lth x lth y lth z	15.50 15.50 15.50)31)31)31	1.000 1.000 1.000	00 00 00	20.0000 20.0000 20.0000) 10) 10) 10			

Table SI 2 EPR parameters for the $Mn_2^{III,IV}$ form of **1** obtained by chemical oxidation as determined by a simulation of the signal using the X Sophe software suite. Hyperfine coupling constants A are given in gauss.

	g	A ^{III}	A ^{IV}
x	2.005	-168	72
у	1.996	-149	83
Z.	1.985	-107	79

1. C. J. O'Connor, Prog. Inorg. Chem., 1982, 29, 203-283.