Electronic Supporting Information (ESI) for:

# Isolation and characterization of unusual multinuclear Schiff base complexes: Rearrangements reactions and octanuclear cluster formation

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# **Copies of NMR and MS spectra for compounds 1-11:**<sup>1</sup>

#### <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for 1:



\*Please note the peak located around  $\delta = 1.8$  ppm indicating the presence of OAc.

<sup>&</sup>lt;sup>1</sup> Please note that in the case of the  ${}^{13}C{}^{1}H$  NMR experiments DEPT was used. In some cases, the  ${}^{13}C$  NMR traces are of mediocre quality as a result of the rather low solubility of the respective complexes.



#### MALDI(+) MS (dctb) for 1:





Calculated pattern for  $Zn_8$  complex 1 ( $C_{38}H_{40}N_4O_{12}Zn_4 \cdot dctb \cdot Na$ ):

Note the data for dctb below:



#### <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **2**:



MALDI(+) MS (dctb) for 2:





Calculated pattern for  $Zn_8$  complex 2 ( $C_{82}H_{62}N_6O_{16}Zn_8$ ):

# <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **3**:







Note that complex 3 was not very soluble under the MS conditions used.



### Calculated pattern for $Zn_8$ complex **3** ( $C_{34}H_{32}N_4O_{12}Zn_4$ ):

# <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **4**:



### MS analysis of 4:



Calculated pattern for  $Zn_8$  complex 4 ( $C_{42}H_{36}N_4O_{12}Zn_4 \cdot dctb \cdot Zn$ ):



Note the data for dctb below:



# <sup>1</sup>H NMR spectrum (Acetone- $d_6$ ) for **5**:



# <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **6**:









### MALDI(+) MS (dctb) for 6 (enlargement):



Calculated pattern for  $Zn_8$  complex 6 ( $C_{88}H_{64}N_8O_{16}Zn_8$ ):



### <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for 8:





# <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **8**, enlargement of aromatic region:

#### MALDI(+) MS (dctb) for 8:





### MALDI(+) MS (dctb) for 8 (enlargement):



### X-ray diffraction analysis for 8:



Displacement ellipsoids shown at the 50% probability level; note that the co-crystallized solvent molecules and H-atoms are omitted for clarity. Only a partial numbering scheme (for the Zn centres) is provided here.

# <sup>1</sup>H NMR spectrum (DMSO- $d_6$ ) for **9**:



### <sup>1</sup>H NMR (EXTENDED region) spectrum (DMSO- $d_6$ ) for **9**:



# <sup>13</sup>C{<sup>1</sup>H} NMR spectrum (DMSO- $d_6$ ) for **9**:



### POV-RAY presentation of the X-ray structure determined for 10:



Note the presence of four 3,5-dimethylpyridines coordinating to some of the Zn centres that are highlighted in green. Co-crystallized solvent molecules and H-atoms are omitted for clarity.

#### <sup>1</sup>H NMR (DMSO- $d_6$ ) of the crystalline material obtained for the conversion of **1** $\rightarrow$ **10**:

Comparison between pure complex **6** and the crystalline material (**10**) obtained after treatment of **1** with pyridine:



On top in **RED**: aromatic region for the pure complex 6. Below, in **BLACK** the same aromatic region for the crystalline material of the conversion  $1\rightarrow 10$ .





MALDI(+) MS (dctb) for **10**, showing the cluster around m/z = 1445:



### MALDI(+) MS (dctb) for **10**, showing the cluster around m/z = 2011:



Please note the resemblance with the isotopic cluster for complex **6**.

X-ray structure (ORTEP) determined for 11:



#### <sup>1</sup>H NMR (DMSO- $d_6$ ) of the crystalline material obtained for the conversion of $4\rightarrow 11$ :

Comparison between pure complex **8** and the crystalline material (**11**) obtained after treatment of **4** with pyridine:



On top in **RED**: aromatic region for the pure complex **8**.

Below, in **BLACK** the same aromatic region for the crystalline material of the conversion  $4\rightarrow 11$ .

### MALDI(+) MS (dctb) for 11:





### MALDI(+) MS (dctb) for **11**, showing the cluster around m/z = 1512:





Please note the resemblance with the isotopic cluster for complex 8.



NMR conversion of 1 into 6 in DMSO- $d_6$  at 130°C.

The salicylaldehyde used was 2,3-dihydroxy-benzaldehyde; the reaction was carried out in an NMR tube using DMSO- $d_6$  as solvent. Amounts: **1** (7.5 mg, 0.00746 mmol), 2,3-dihydroxy-benzaldehyde (2.0 mg, 0.0145 mmol) and DMSO- $d_6$  (0.5 mL).

Comparison of <sup>1</sup>H NMR traces for the conversion of  $1 \rightarrow 2$ .

Note that this experiment was repeated twice with virtually the same result. Below only the aromatic region is shown (DMSO- $d_6$ ):



On top, in **RED**, the pure crystalline complex 2 in DMSO- $d_6$ . Below in **BLACK** the complex isolated after treatment of 1 with 2,3dihydroxybenzaldehyde in pyridine. Some residual pyridine (reaction solvent) is also present.

Comparison of <sup>1</sup>H NMR traces for the conversion of  $3\rightarrow 7$ .



On top, in **RED**, the pure complex **7** in DMSO- $d_6$ <sup>2</sup>. Below in **BLACK** the complex isolated after treatment of **3** with 2,3-dihydroxybenzaldehyde in pyridine.

<sup>&</sup>lt;sup>2</sup> For the synthesis and further analysis of this complex refer to: R. M. Haak, A. Decortes, E. C. Escudero-Adán, M. Martínez Belmonte, E. Martin, J. Benet-Buchholz and A. W. Kleij, *Inorg. Chem.*, 2011, **50**, 7934.

Comparison of <sup>1</sup>H NMR traces for the conversion of  $4\rightarrow 8$ .



On top, in **RED**, the independently prepared complex **8** in DMSO- $d_6$ . Below in **BLACK** the complex isolated after treatment of **4** with 2,3-dihydroxybenzaldehyde in pyridine. Traces of residual pyridine (reaction medium) are also present.