## Supporting Information to "Encapsulating [FeFe]-hydrogenase model compounds in peptide hydrogels dramatically modifies stability and photochemistry"

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## Full reference 48:

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Figure S1. Comparison between a fresh (2 hours) and 13 day-old gel sample containing ( $\mu$ -S(CH<sub>2</sub>)<sub>3</sub>S)Fe<sub>2</sub>(CO)<sub>4</sub>(PMe<sub>3</sub>)<sub>2</sub>



**Figure S2.** Comparison of FTIR spectra for  $(\mu$ -S(CH<sub>2</sub>)<sub>3</sub>S)Fe<sub>2</sub>(CO)<sub>4</sub>(PMe<sub>3</sub>)<sub>2</sub> in the gel phase, in the pregel solution (without Fmoc-LL gelator) and the gel without the metal carbonyl. Dramatic line broadening is observed for the carbonyl stretch region in the pre-gel solution. However, this solution is unstable and precipitates out completely within 20 minutes of preparation by sonication.



**Figure S3.** Peak heights in the amide I (~1600-1700 cm<sup>-1</sup>) and CO stretch region (~1850-2050 cm<sup>-1</sup>) of the IR spectrum for ( $\mu$ -S(CH<sub>2</sub>)<sub>3</sub>S)Fe<sub>2</sub>(CO)<sub>4</sub>(PMe<sub>3</sub>)<sub>2</sub> in the gel phase. The decrease in the CO region occurs at a different temperature from the decrease in the amide I region. The peak at 1590 cm<sup>-1</sup> is attributed to COO<sup>-</sup> absorption from the deprotonated C-termini of Fmoc-LL.



**Figure S4.** Fluorescence emission spectra of an Fmoc-LL hydrogel at various temperatures. The dashed line represents the emission in methanol at 20 °C. The inset shows the intensity at 386 nm as function of temperature.

