

Supplementary Information for

Multiphase enantioselective Kharasch–Sosnovsky allylic oxidation based on neoteric solvents and ditopic copper complexes

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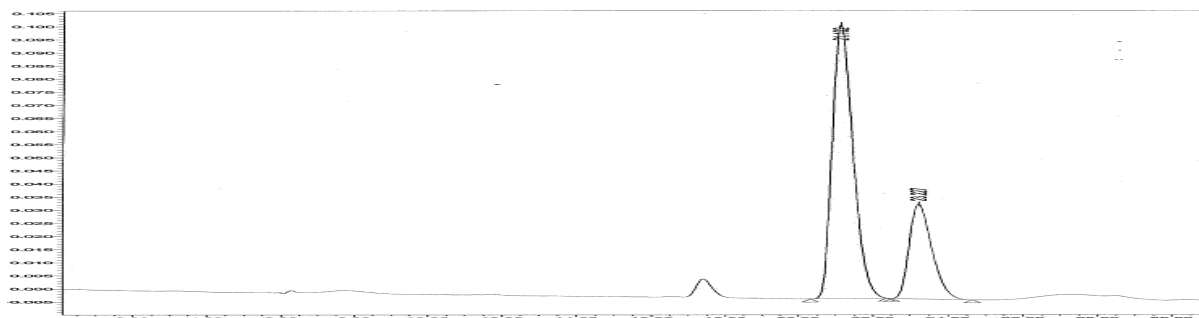
Chromatographic analyses of the reaction of cyclohexene and cycloheptene with *tert*-butyl perbenzoate, catalyzed by copper complexes

Reactions were monitored by gas chromatography. FID from Hewlett-Packard 5890II; cross-linked methyl silicone column: 25 m × 0.2 mm × 0.33 μm; helium as carrier gas. 20 psi; injector temperature: 230 °C; detector temperature: 250 °C; oven program: 65 °C (3 min), 25 °C min⁻¹ to 250 °C (3 min); retention times:

- *n*-decane 6.1 min.
- *tert*-butyl perbenzoate 8.1 min.
- 2-cyclohexenyl benzoate 11.2 min.
- 2-cycloheptenyl benzoate (**2c**) 11.2 min.

Enantioselectivities were determined by HPLC on a Waters Alliance Chromatograph with a PDA detector (λ = 220 nm). (*S*)-enantiomer was the major product in all cases.

2-Cyclohexenyl 1-benzoate: CHIRALCEL OJ (DAICEL) column, 0.46 cm x 25 cm, *n*-hexane; 0.80 mL min⁻¹; retention times: (*S*)-isomer 21.2 min, (*R*)-isomer 23.2 min.



2-Cycloheptenyl 1-benzoate: CHIRALCEL OJ (DAICEL) column, 0.46 cm x 25 cm, *n*-hexane; 0.35 mL min⁻¹; retention times: (*R*)-isomer 27.8 min, (*S*)-isomer 30.0 min.



Table S1 Results obtained from the allylic oxidation of cyclohexene in BTFEP, using the *i*Pr-click-DAX-CuPF₆ complex as catalyst. Deactivation experiments. Before the fourth run, the catalytic phase is divided into two equal aliquots.

Run	Yield (%)	%ee
1	65	83
2	67	74
3	63	65
4	64	45
4*	69	68
5*	68	65
6*	64	68

*After addition of 15% fresh ligand.

Table S2 Results obtained from the allylic oxidation of cyclohexene in [BMIM][PF₆], using the *i*Pr-AzaBox- and *i*Pr-PyBox-CuCl complexes as catalyst.

Ligand	Run	Time (days)	Yield (%)	%ee
<i>i</i> PrAzaBox	1	2	5	8
	2	2	13	11
	3	2	22	42
	4 ^a	2	77	45
	5	2	3	7
<i>i</i> PrPyBox	1	2	73	57
	2	2	47	48
	3	2	21	9
<i>i</i> PrAzaBox ^b	1	2	26	33
	2	4	82	40
	3	5	56	33

^a After addition of phenylhydrazine. ^b Phenylhydrazine is added in each reaction.