

Supporting Information

Photophysical and Electrochemical Properties of Platinum(II) Complexes Bearing a Chromophore-Acceptor Dyad and their Photocatalytic Hydrogen Evolution

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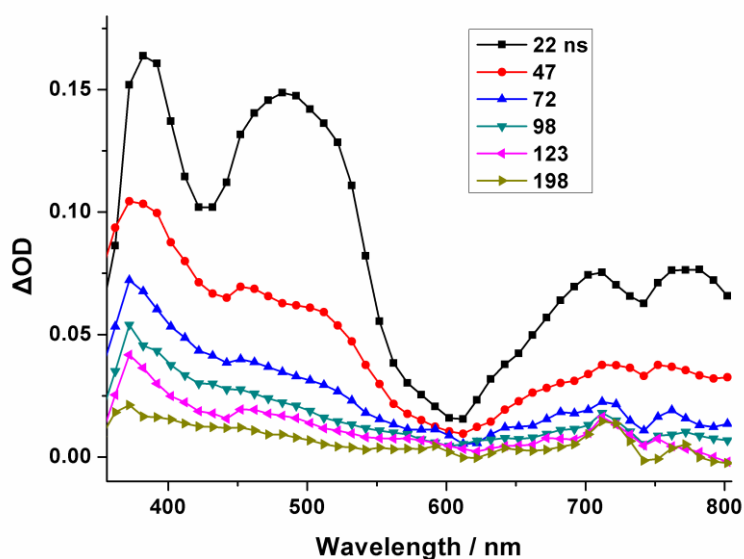


Fig. S1. Transient absorption spectra of complex **2** in CH₃CN ($\lambda_{\text{exc}} = 355$ nm) at room temperature.

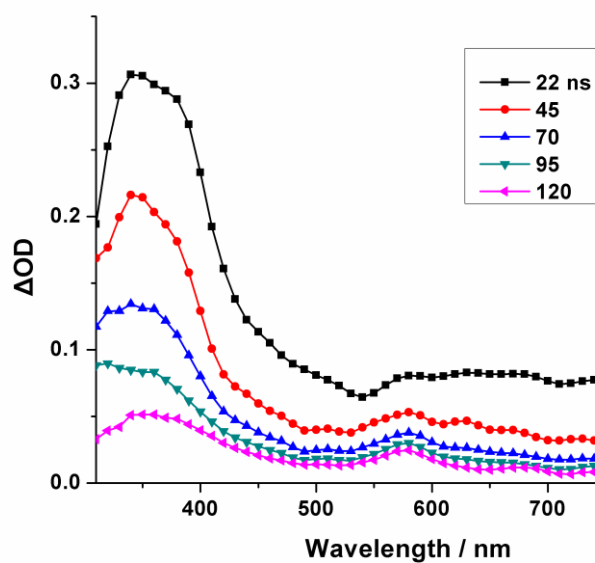


Fig. S2. Transient absorption spectra of complex **3** in CH₃CN ($\lambda_{\text{exc}} = 355$ nm) at room temperature.

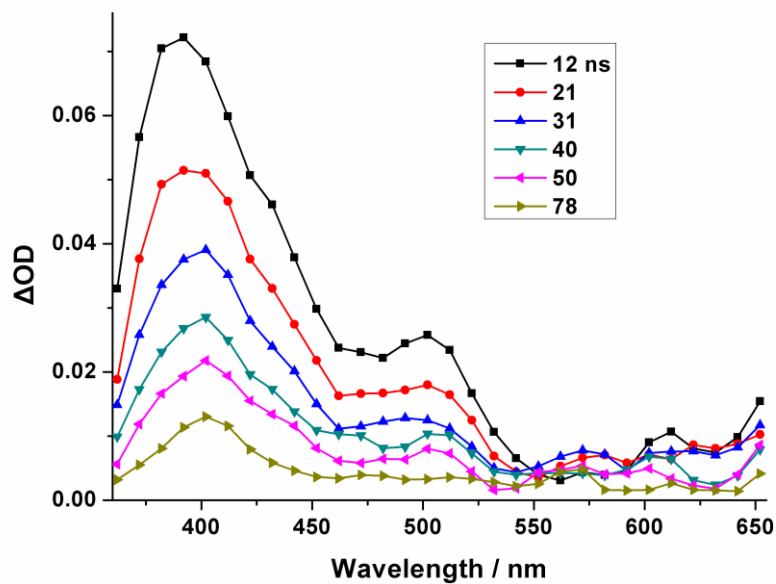


Fig. S3. Transient absorption spectra of complex **4** in CH₃CN ($\lambda_{\text{exc}} = 355$ nm) at room temperature.

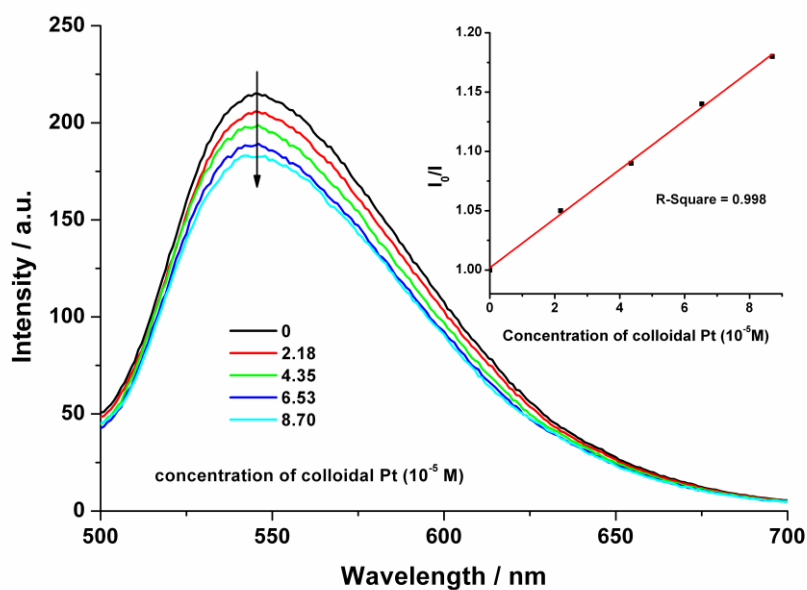


Fig. S4. Emission spectral changes of **1** (1.3×10^{-5} M) in degassed $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (v/v, 1:1) at room temperature as a function of the colloidal Pt concentration. Inset: Stern-Volmer plot for emission quenching of **1**.

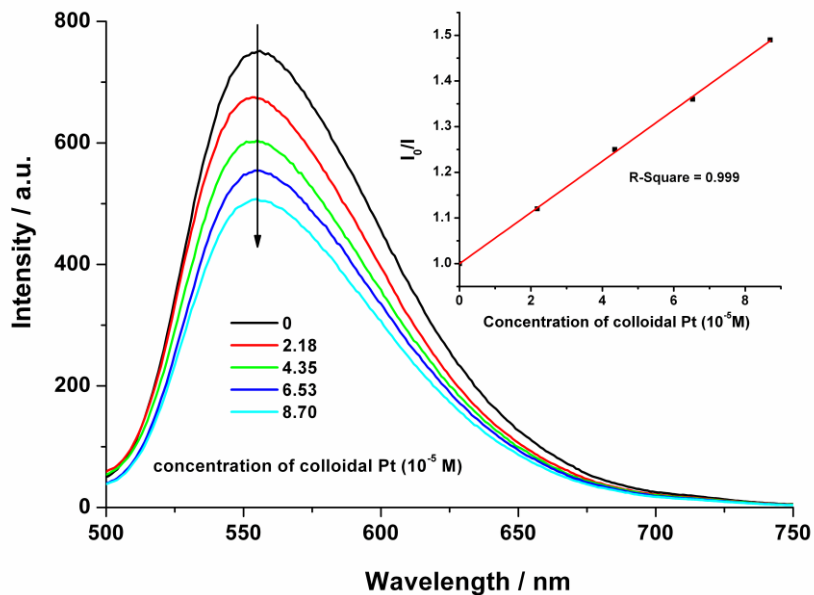


Fig. S5. Emission spectral changes of **9** (2.0×10^{-5} M) in degassed $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (v/v, 1:1) at room temperature as a function of the colloidal Pt concentration. Inset: Stern-Volmer plot for emission quenching of **9**.

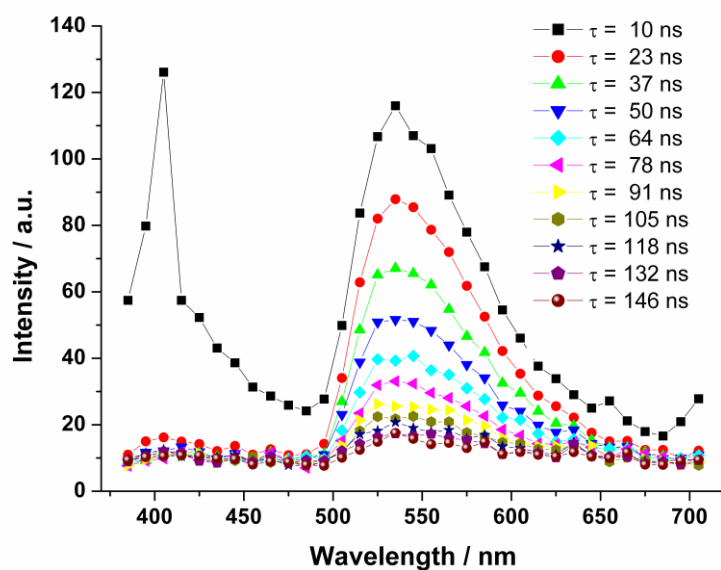


Fig. S6. Time-resolved emission spectra of complex **3** (2.5×10^{-5} M) with excitation at 355 nm in the mixed $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (v/v, 1:1) in the absence of Pt nanoparticles (4.3×10^{-5} M).

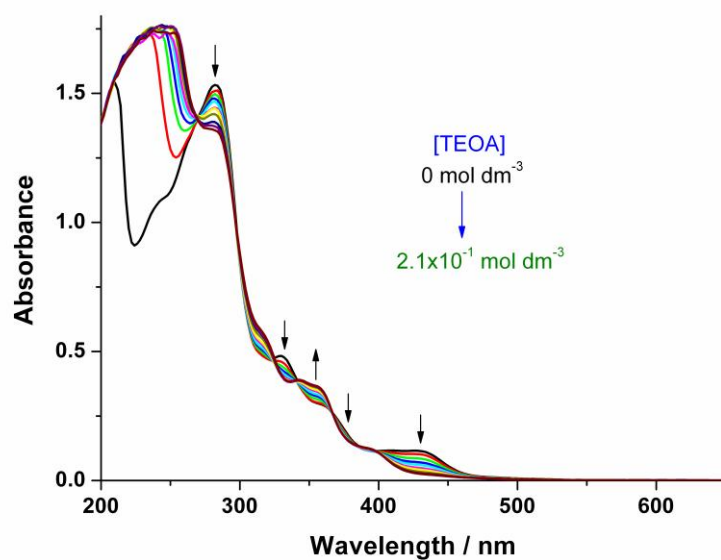


Fig. S7. Changes in the absorption spectra of **5** (3.3×10^{-5} M) in acetonitrile upon addition of various concentrations of TEOA (0– 2.1×10^{-1} M).

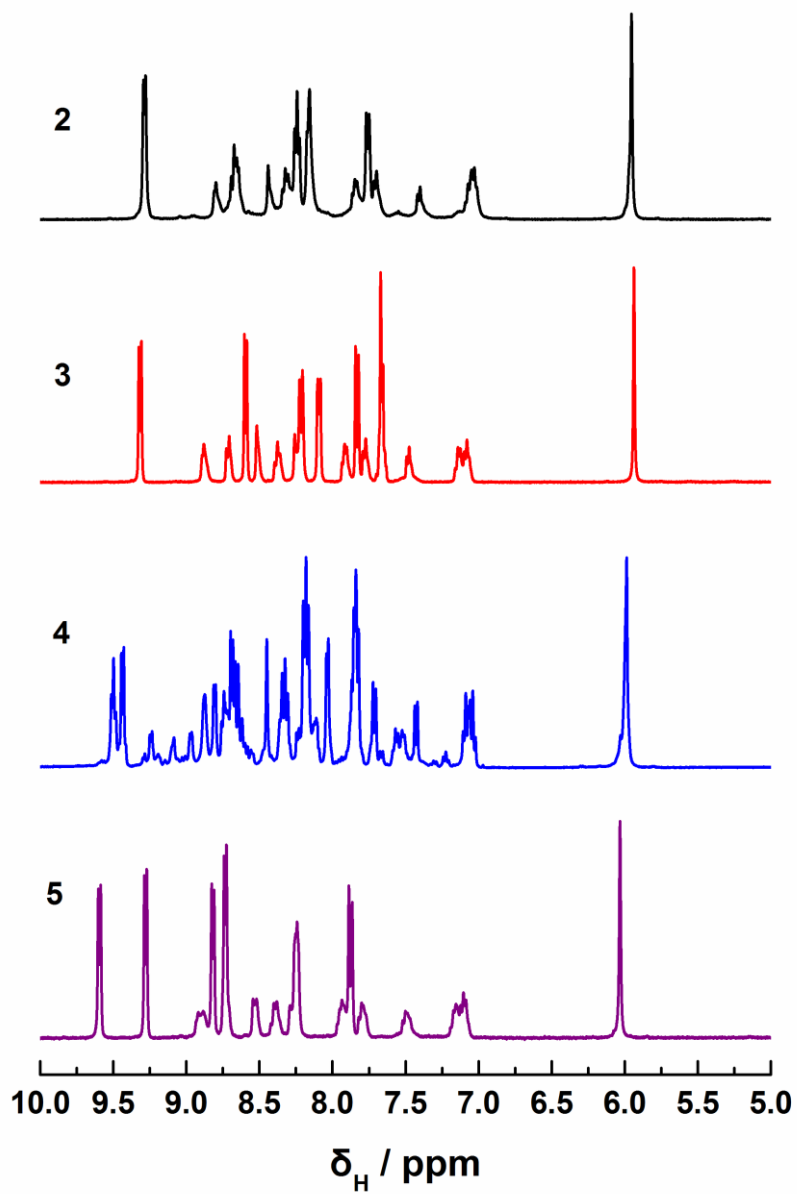


Fig. S8. ¹H NMR spectra of complexes 2–5 in DMSO-d₆.

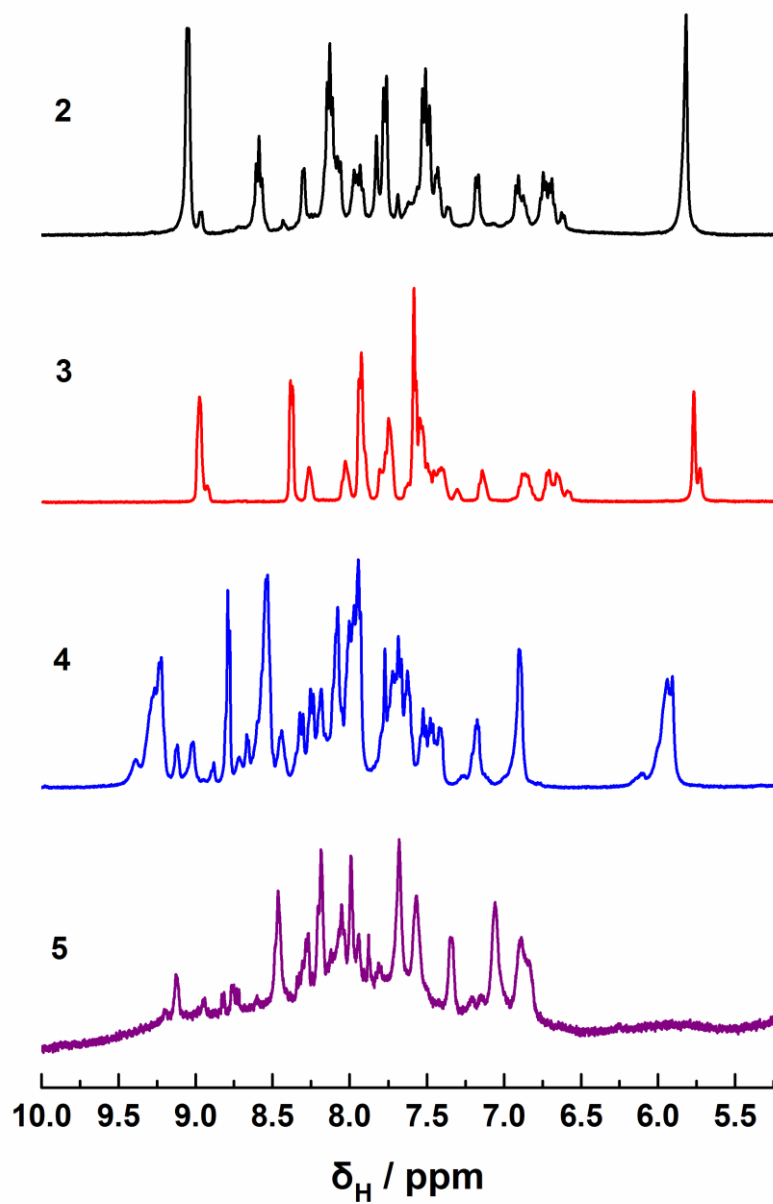


Fig. S9. ¹H NMR spectra of complexes 2–5 in DMSO-d₆/D₂O (v/v, 2:1), in the presence of excessive amounts of TEOA.

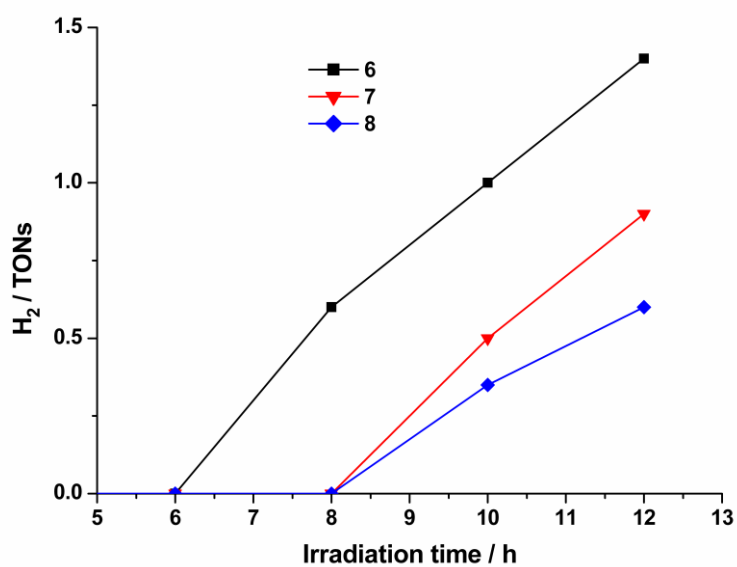


Fig. S10. Hydrogen production from system containing TEOA (2.2×10^{-2} M), complexes **6–8** (4.4×10^{-5} M) and colloidal Pt (8.7×10^{-5} M) in CH₃CN/H₂O (v/v, 1:1) upon irradiation ($\lambda > 390$ nm) at pH 7.0. TONs are calculated based on the Pt(II) chromophore.