

Supporting Information

Manuscript: Photophysical Properties of MM Quadruply Bonded Complexes (M = Mo, W) Supported by Carboxylate Ligands: Charge Delocalization and Dynamics in S₁ and T₁ States

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Compound	1
Chemical Formula	C ₈₄ H ₉₂ Mo ₂ O ₁₀ S ₂
Formula Weight	1517.58
Temperature (K)	180(2)
Space Group	Monoclinic, P2 ₁ /n
<i>a</i> (Å)	9.7536(1)
<i>b</i> (Å)	14.6021(1)
<i>c</i> (Å)	26.2150(2)
α (°)	
β (°)	94.0908(2)
γ (°)	
<i>V</i> (Å ³)	3724.11(5)
<i>Z</i>	2
<i>D</i> _{calcd} (Mg/m ³)	1.353
Crystal Size (mm)	0.35 X 0.15 X 0.12
Theta range for data collection	2.59 to 26.37°
μ , (Mo, K α) (mm ⁻¹)	0.452
Reflections collected	14852
Unique reflections	7605 [R(int)= 0.0192]
Completeness to theta	(26.37) 99.9%
Data/restraints/parameters	7605 / 0 / 451
R1 ^a (%) (all data)	3.05 (4.25)
wR2 ^b (%)(all data)	7.04 (7.51)
Goodness-of-fit on F ²	1.024
Largest diff. peak and hole (e Å ⁻³)	0.707 and -0.578

$$^a R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$$

$$^b wR2 = \{ \sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2] \}^{1/2}$$

Table S1. Select crystallographic information from structure **1**.

Compound 1

Mo(1)-O(4)	2.1022(15)	Mo(1)-O(3)	2.5007(18)
Mo(1)-O(1)#1	2.1160(14)	O(1)-C(1)	1.270(2)
Mo(1)-Mo(1)#1	2.1164(6)	O(1)-Mo(1)#1	2.1160(14)
Mo(1)-O(2)	2.1184(14)	O(2)-C(1)	1.273(2)
Mo(1)-O(3)#1	2.1243(15)	O(3)-C(18)	1.272(2)
O(3)-Mo(1)#1	2.1243(15)	C(2)-C(3)	1.193(3)
O(4)-Mo(1)-O(1)#1	90.68(5)	O(1)#1-Mo(1)-O(2)	176.69(5)
O(4)-Mo(1)-Mo(1)#1	91.44(4)	Mo(1)#1-Mo(1)-O(2)	90.41(4)
O(1)#1-Mo(1)-Mo(1)#1	92.88(4)	O(4)-Mo(1)-O(3)	90.36(7)
O(4)-Mo(1)-O(2)	88.84(5)	C(2)-C(3)-C(4)	174.7(2)
C(18)-O(4)-Mo(1)	118.15(12)	O(1)-C(1)-O(2)	123.31(18)

Table S2. Select bond lengths (Å) and bond angles (°) for compounds **1**

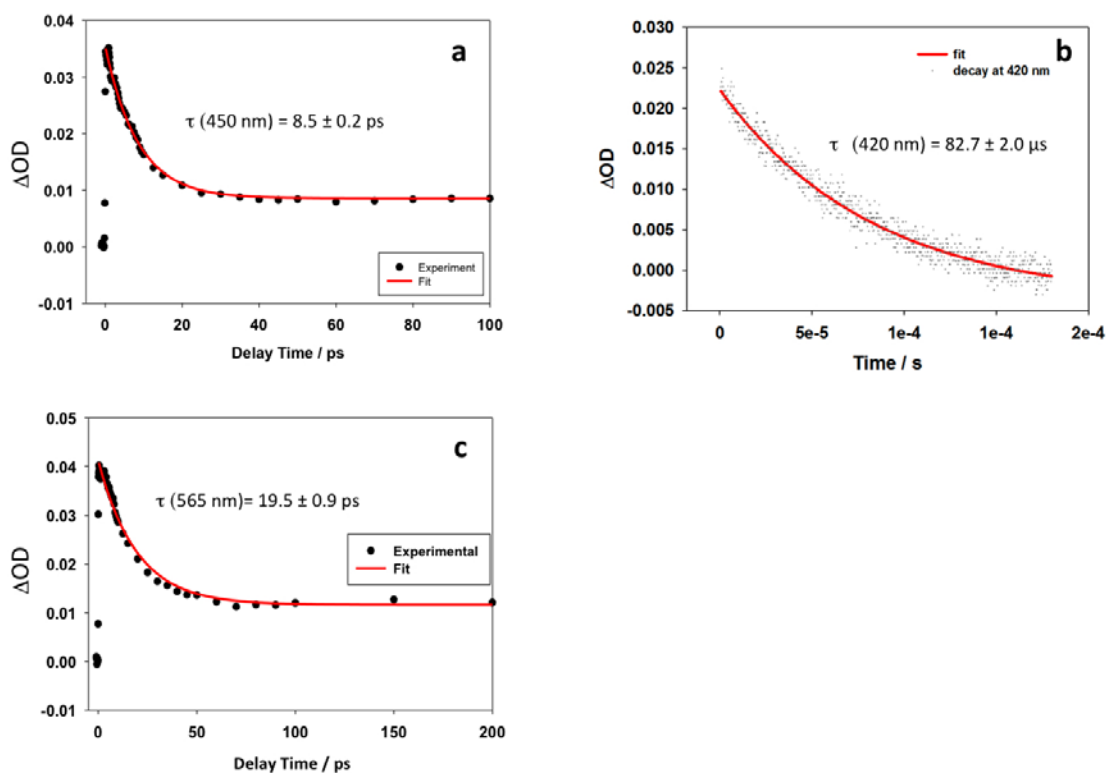


Fig. S1 (a) Kinetic trace from fs TA of **1** monitored at 450 nm, $\lambda_{\text{ex}} = 514 \text{ nm}$; (b) Kinetic trace from ns TA of **1** monitored at 420 nm, $\lambda_{\text{ex}} = 532 \text{ nm}$; (c) Kinetic trace from fs TA of **2** monitored at 565 nm, $\lambda_{\text{ex}} = 675 \text{ nm}$.

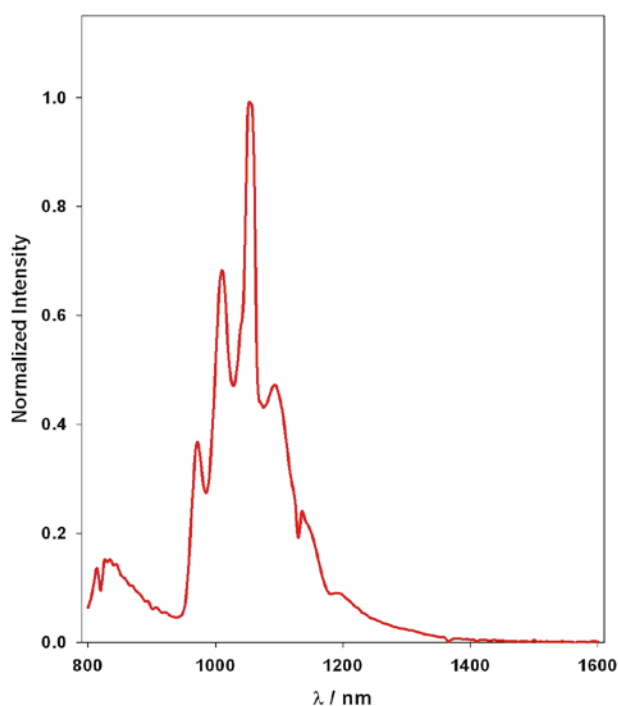


Fig. S2 Near-IR emission of **1** recorded in 2-methyl THF at 77 K.

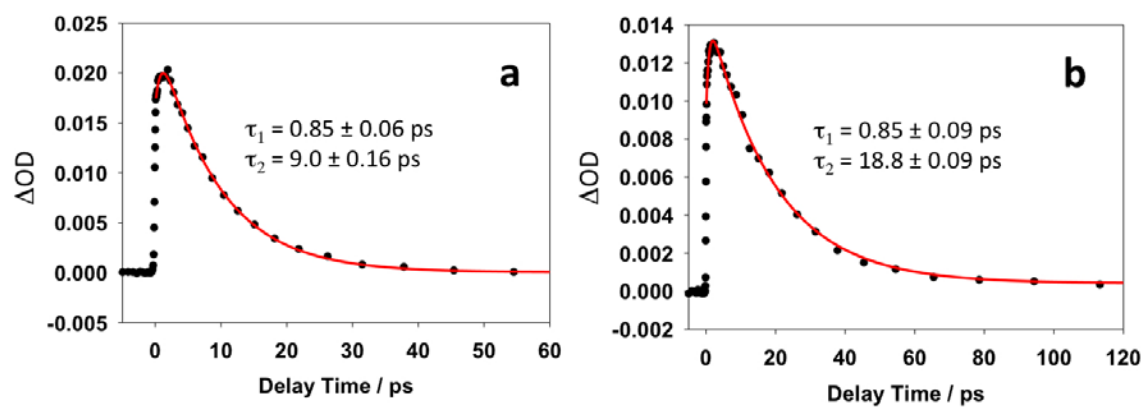


Fig. S3 (a) Kinetic trace from fs TRIR of **1** monitored at 2110 cm^{-1} , $\lambda_{\text{ex}} = 514\text{ nm}$; (b) Kinetic trace from fs TRIR of **2** monitored at 2150 cm^{-1} , $\lambda_{\text{ex}} = 675\text{ nm}$.