

**Syntheses, characterizations and properties of  
[Mo<sub>2</sub>O<sub>2</sub>S<sub>2</sub>]-based oxothiomolybdenum wheels incorporating  
bisphosphonate ligands.**

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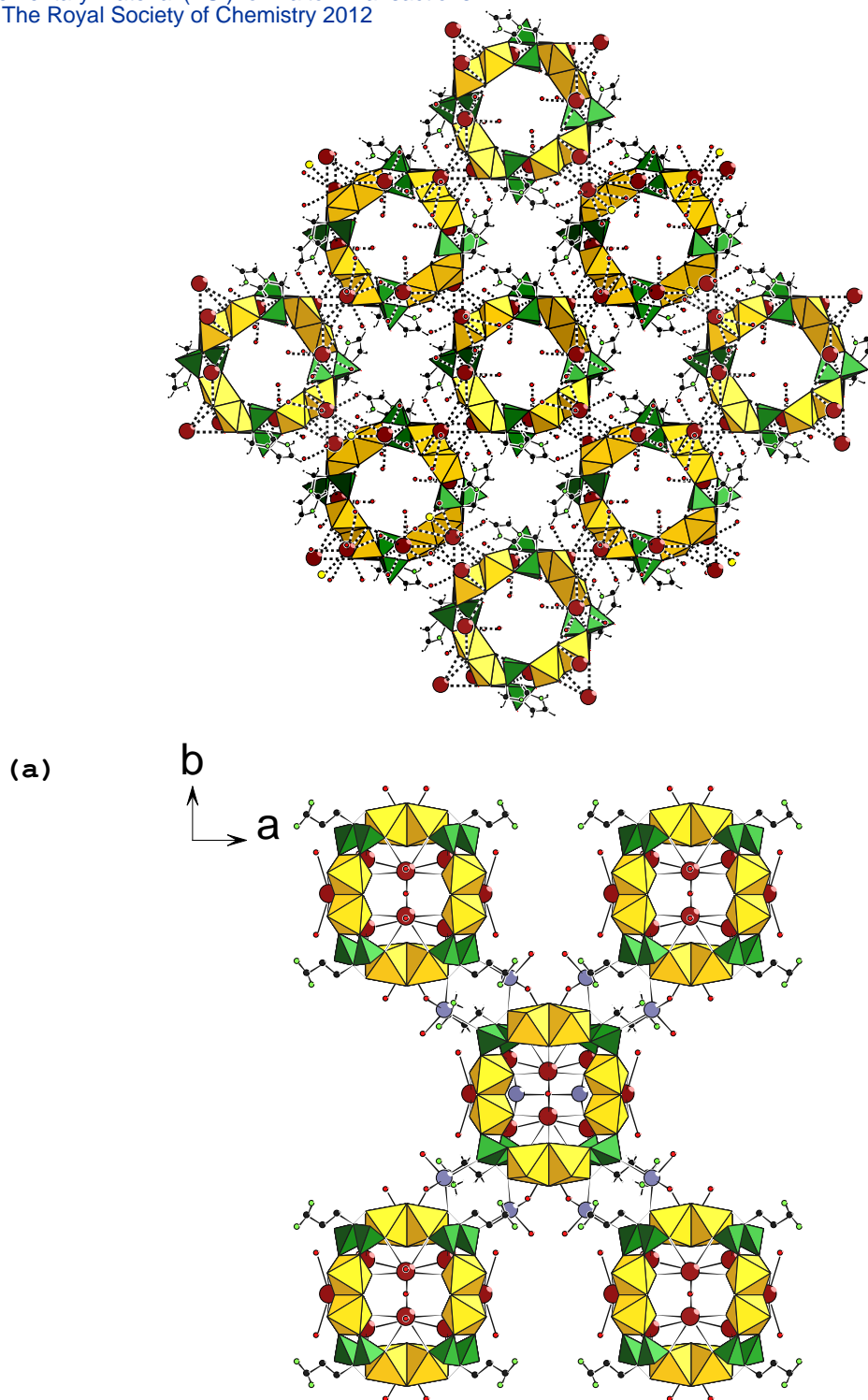
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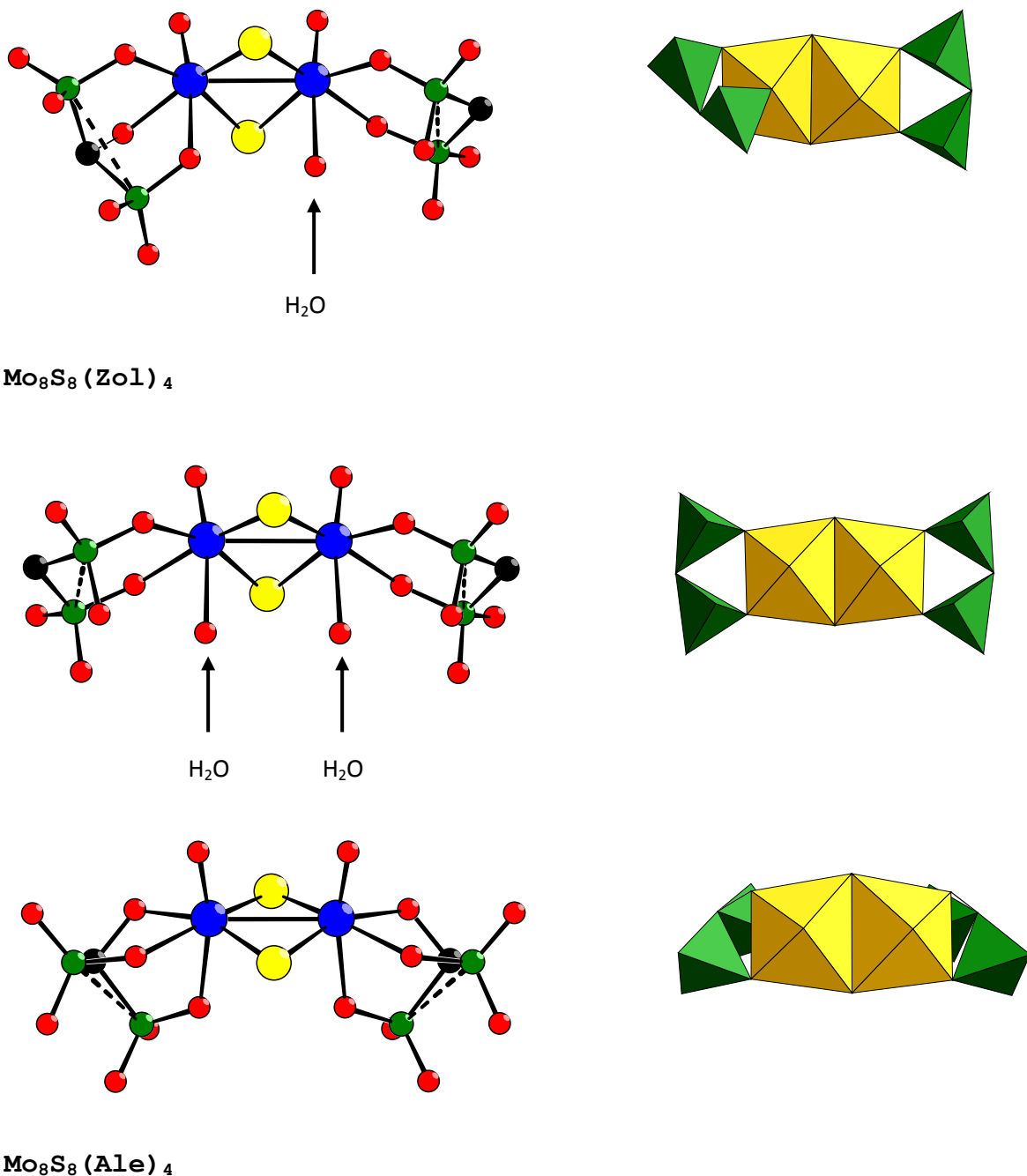
**Supporting Information**

**Table SI1.** Comparison of selected bond distances in **Mo<sub>8</sub>S<sub>8</sub>(Ale)<sub>4</sub>**, **Mo<sub>8</sub>S<sub>8</sub>(Zol)<sub>4</sub>** and **Mo<sub>8</sub>O<sub>8</sub>(Ale)<sub>4</sub>**.

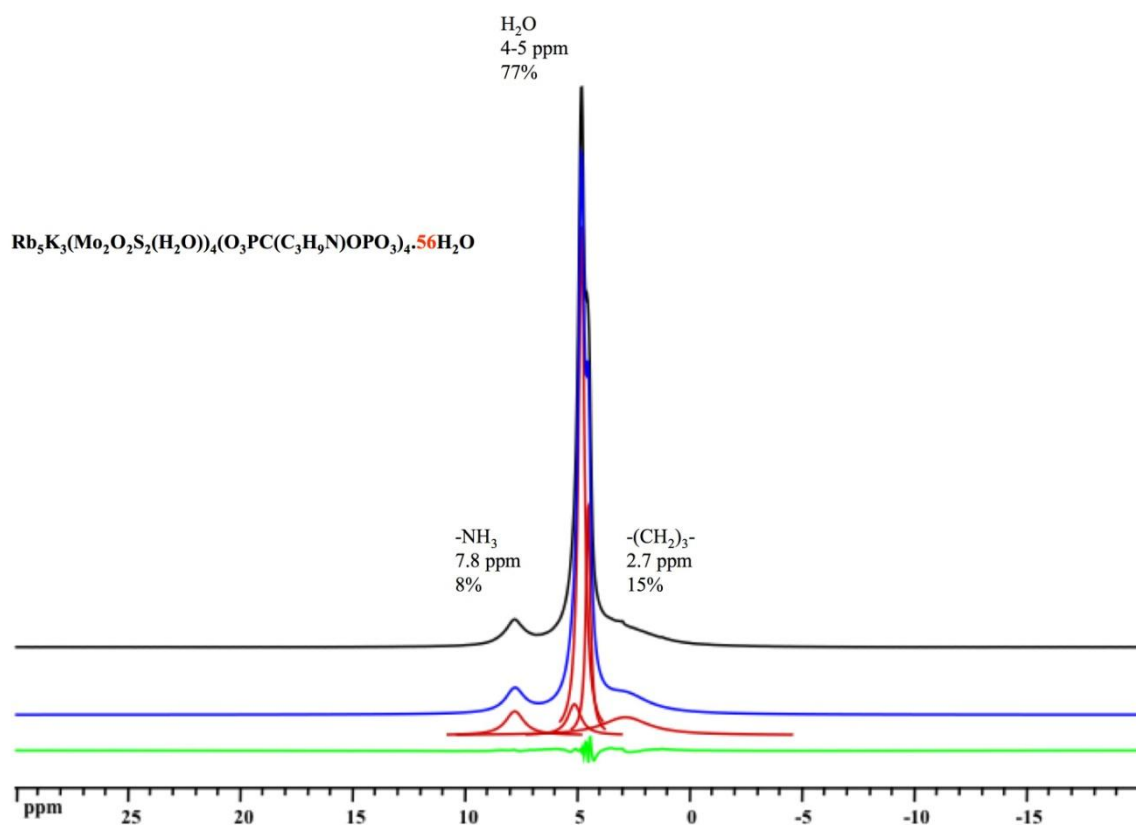
<b>Mo<sub>8</sub>S<sub>8</sub>(Ale)<sub>4</sub></b>		<b>Mo<sub>8</sub>S<sub>8</sub>(Zol)<sub>4</sub></b>		<b>Mo<sub>8</sub>O<sub>8</sub>(Ale)<sub>4</sub></b>			
Mo (1)-O (1)	1.6906 (11)	Mo (1)-O (1)	1.677 (3)	Mo (1)-O (1)	1.681 (4)	Mo (5)-O (25)	1.704 (4)
Mo (1)-O (2)	2.0698 (11)	Mo (1)-O (2)	2.130 (4)	Mo (1)-O (2)	1.932 (4)	Mo (5)-O (9)	1.944 (4)
Mo (1)-O (3)	2.1058 (11)	Mo (1)-O (3)	2.200 (3)	Mo (1)-O (3)	1.946 (4)	Mo (5)-O (8)	1.948 (4)
Mo (1)-S (1)	2.3219 (4)	Mo (1)-O (4)	2.204 (3)	Mo (1)-O (4)	2.074 (4)	Mo (5)-O (26)	2.057 (4)
Mo (1)-S (2)	2.3226 (4)	Mo (1)-S (2)	2.3263 (14)	Mo (1)-O (5)	2.079 (4)	Mo (5)-O (27)	2.110 (4)
Mo (1)-O (4)	2.3924 (10)	Mo (1)-S (1)	2.3295 (15)	Mo (1)-O (6)	2.408 (4)	Mo (5)-O (28)	2.287 (4)
Mo (1)-Mo (2)	2.8387 (2)	Mo (1)-Mo (1)	2.8525 (8)	Mo (1)-Mo (6)	2.5834 (7)		
						Mo (6)-O (29)	1.705 (5)
Mo (2)-O (5)	1.6832 (11)	Mo (2)-O (5)	1.679 (4)	Mo (2)-O (7)	1.681 (4)	Mo (6)-O (2)	1.934 (4)
Mo (2)-O (6)	2.1032 (10)	Mo (2)-O (7)	2.109 (3)	Mo (2)-O (8)	1.938 (4)	Mo (6)-O (3)	1.941 (4)
Mo (2)-O (7)	2.1074 (10)	Mo (2)-O (6)	2.111 (4)	Mo (2)-O (9)	1.942 (4)	Mo (6)-O (30)	2.045 (4)
Mo (2)-S (1)	2.3243 (4)	Mo (2)-S (3)	2.3078 (17)	Mo (2)-O (10)	2.072 (4)	Mo (6)-O (31)	2.116 (4)
Mo (2)-S (2)	2.3261 (4)	Mo (2)-S (4)	2.3275 (16)	Mo (2)-O (11)	2.092 (4)	Mo (6)-O (32)	2.311 (4)
Mo (2)-O (8)	2.3965 (11)	Mo (2)-O (8)	2.402 (3)	Mo (2)-O (12)	2.416 (4)		
		Mo (2)-Mo (2)	2.8583 (8)	Mo (2)-Mo (5)	2.5968 (7)	Mo (7)-O (33)	1.695 (4)
						Mo (7)-O (20)	1.929 (4)
				Mo (3)-O (13)	1.703 (4)	Mo (7)-O (21)	1.949 (4)
				Mo (3)-O (15)	1.951 (4)	Mo (7)-O (34)	2.100 (4)
				Mo (3)-O (14)	1.952 (4)	Mo (7)-O (35)	2.111 (4)
				Mo (3)-O (16)	2.036 (4)	Mo (7)-O (36)	2.358 (4)
				Mo (3)-O (17)	2.129 (4)		
				Mo (3)-O (18)	2.310 (4)	Mo (8)-O (37)	1.692 (4)
				Mo (3)-Mo (8)	2.5921 (7)	Mo (8)-O (14)	1.942 (4)
						Mo (8)-O (15)	1.950 (4)
				Mo (4)-O (19)	1.702 (4)	Mo (8)-O (38)	2.071 (4)
				Mo (4)-O (20)	1.937 (4)	Mo (8)-O (39)	2.076 (4)
				Mo (4)-O (21)	1.951 (4)	Mo (8)-O (40)	2.413 (4)
				Mo (4)-O (23)	2.056 (4)		
				Mo (4)-O (22)	2.090 (4)		
				Mo (4)-O (24)	2.357 (4)		
				Mo (4)-Mo (7)	2.5758 (7)		



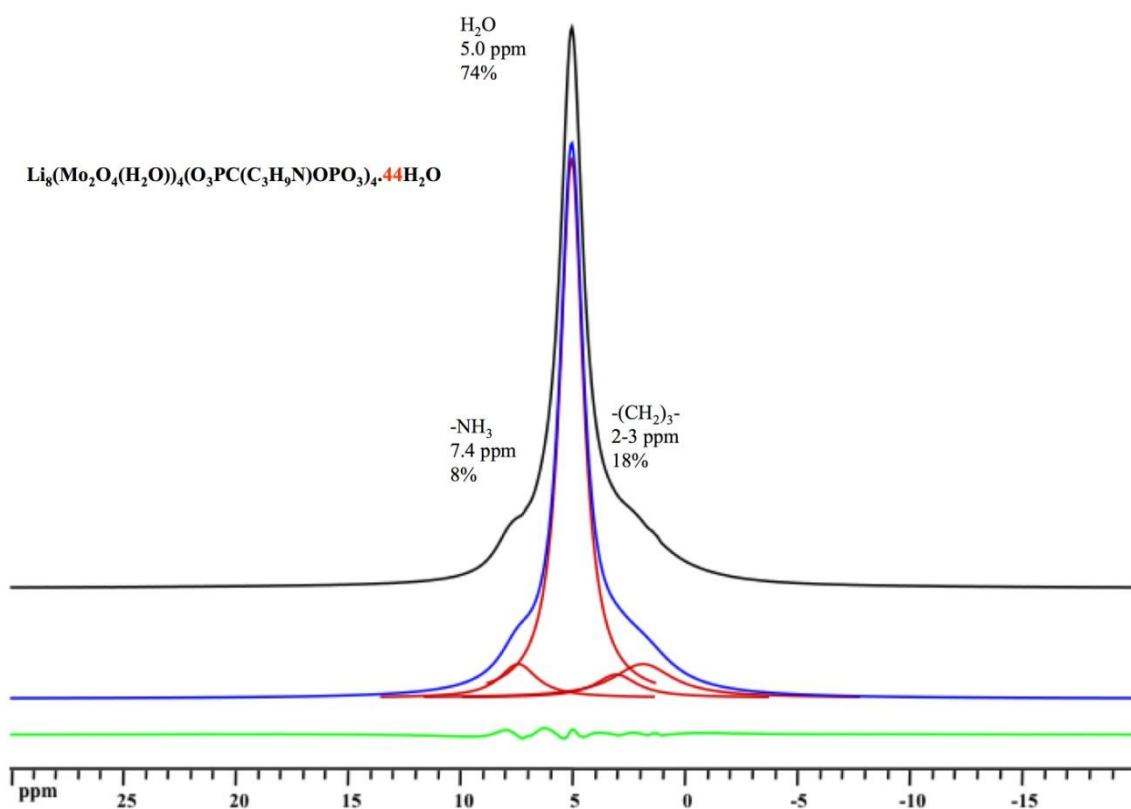
**Figure S11:** View along the *c* axis of the 3D structure of (a)  $\text{Rb}_8\text{Mo}_8\text{S}_8(\text{Zol})_{4.32}\text{H}_2\text{O}$  and (b)  $\text{Rb}_{4.75}\text{K}_{3.25}\text{Mo}_8\text{S}_8(\text{Ale})_{4.25}\text{H}_2\text{O}$ ; brown spheres = Rb, blue spheres = K, red spheres = O, green spheres = N, black spheres = C, yellow octahedra = MoO<sub>2</sub>S<sub>2</sub>, green tetrahedra = PCO<sub>3</sub>.



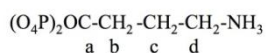
**Figure SI2:** View of (a) a  $\{\text{Mo}^{\text{V}}_2\text{S}_2\text{O}_2\}^{12+}$  dimer with BP ligands connected symmetrically in **Mo<sub>8</sub>S<sub>8</sub>(Zol)<sub>4</sub>** and (b) the two kinds of  $\text{Mo}^{\text{V}}$  dimers with BP ligands connected dissymmetrically in **Mo<sub>8</sub>S<sub>8</sub>(Ale)<sub>4</sub>**; red spheres = O, blue spheres = Mo, green spheres = P, black spheres = C, yellow octahedra =  $\text{MoO}_2\text{S}_2$ , green tetrahedra =  $\text{PCO}_3$ .



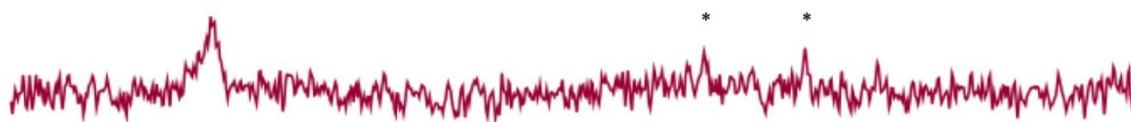
**Figure SI3:** Experimental  $^1\text{H}$  MAS NMR spectrum (black line) and simulated spectrum (blue line) with the decomposition (red lines) and difference spectrum (green line) of  **$\text{Mo}_8\text{S}_8(\text{Ale})_4$** .



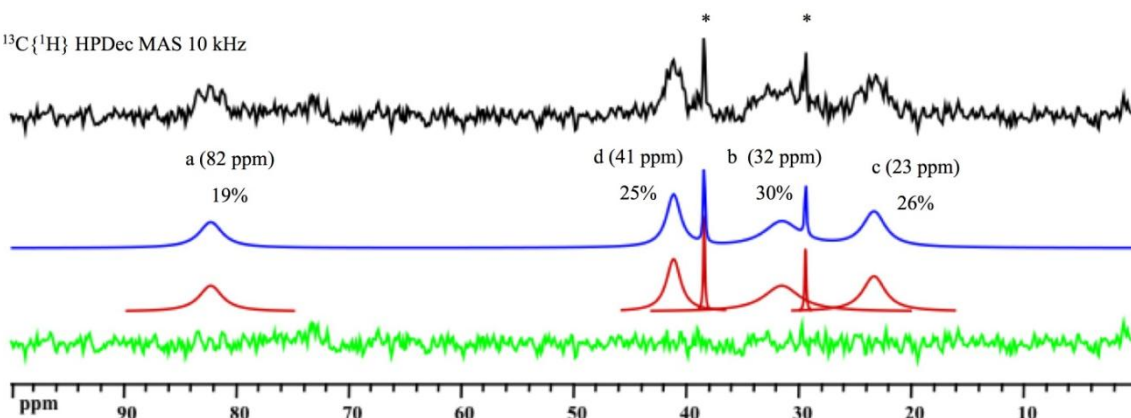
**Figure SI4:** Experimental  $^1\text{H}$  MAS NMR spectrum (black line) and simulated spectrum (blue line) with the decomposition (red lines) and difference spectrum (green line) of  **$\text{Mo}_8\text{O}_8(\text{Ale})_4$** .



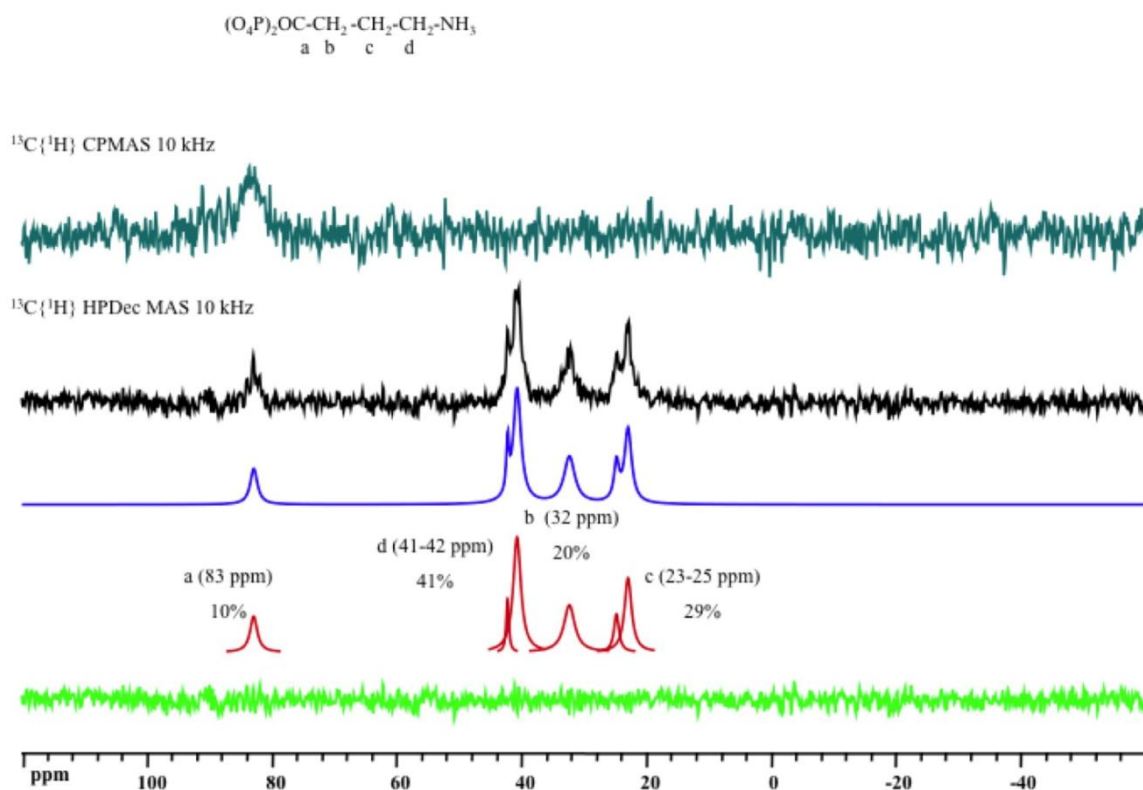
$^{13}\text{C}\{^1\text{H}\}$  CPMAS 10 kHz



$^{13}\text{C}\{^1\text{H}\}$  HPDec MAS 10 kHz

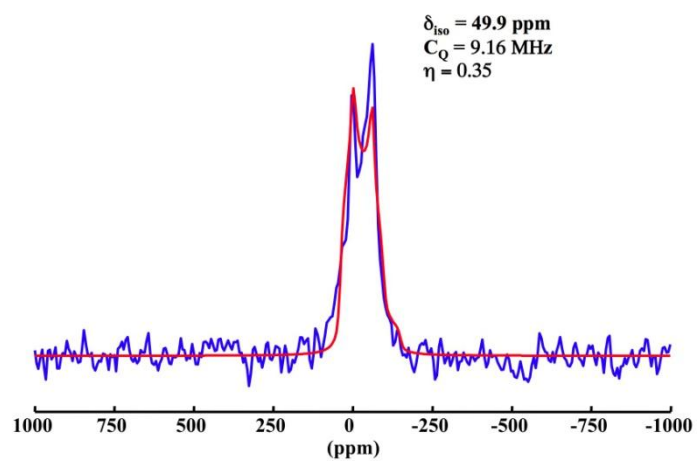


**Figure SI5:** Experimental  $^{13}\text{C}\{^1\text{H}\}$  CPMAS (top) and MAS (bottom black line) NMR spectra and simulated spectrum (blue line) with the decomposition (red lines) and difference spectrum (green line) of **Mo<sub>8</sub>S<sub>8</sub>(Ale)<sub>4</sub>** acquired under heteronuclear TPPM decoupling. Asterisks denote the signals of adamantane due to sample contamination in the NMR rotor.

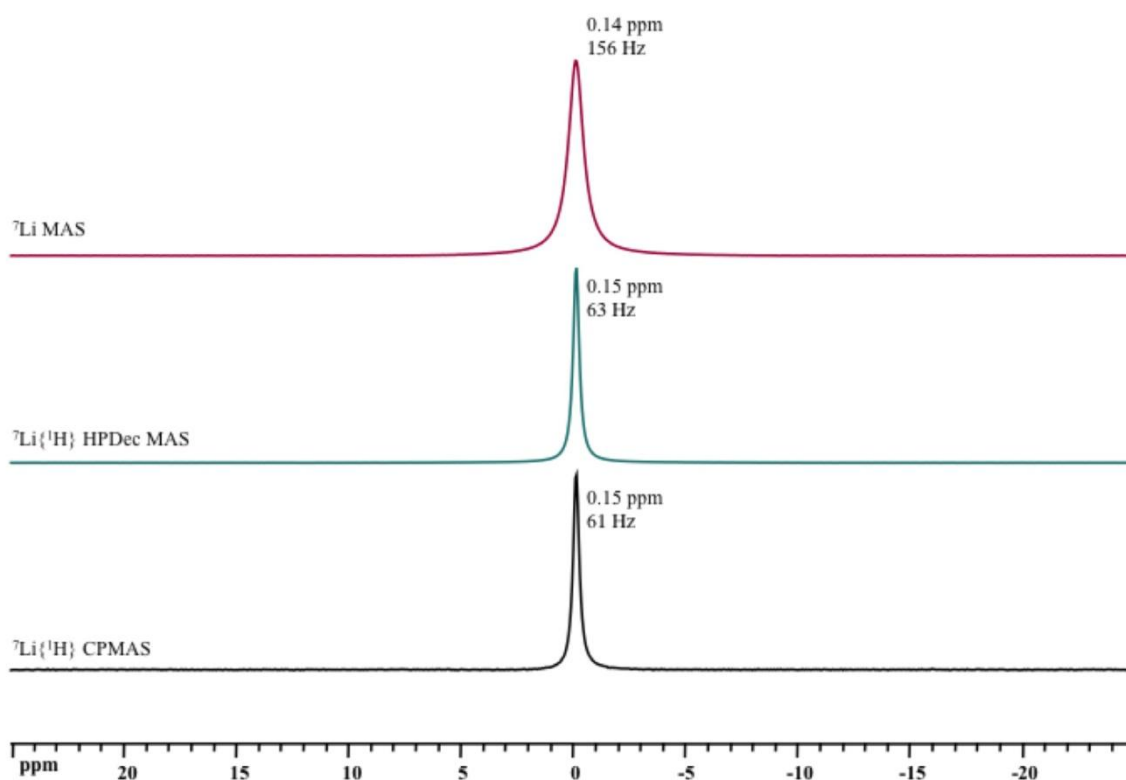


**Figure SI6:** Experimental  $^{13}\text{C}\{^1\text{H}\}$  CPMAS (top) and MAS (bottom, black line) NMR spectra and simulated spectrum (blue line) with the decomposition (red lines) and difference spectrum (green line) of **Mo<sub>8</sub>O<sub>8</sub>(Ale)<sub>4</sub>** acquired under heteronuclear TPPM decoupling.

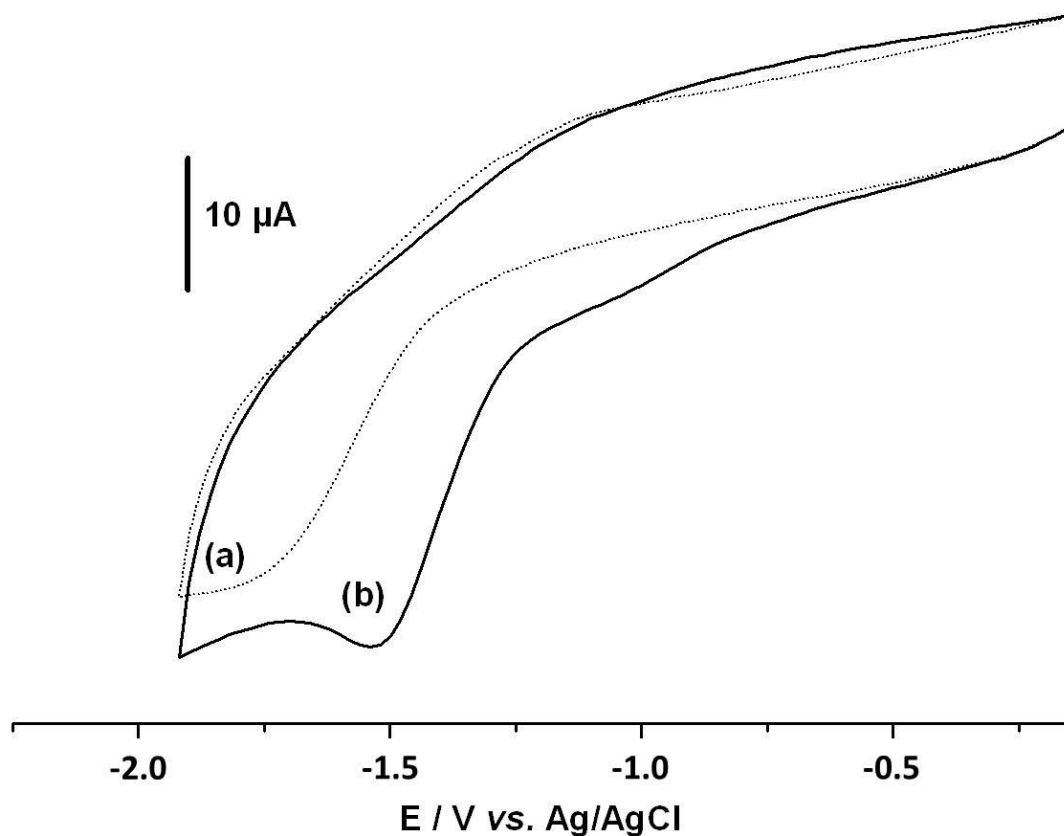




**Figure SI7:** Experimental  $^{87}\text{Rb}$  MAS NMR spectrum (blue line) and simulated spectrum (red line) of  **$\text{Mo}_8\text{S}_8(\text{Ale})_4$** .



**Figure SI8:**  $^7\text{Li}$  MAS NMR spectra of  $\text{Mo}_8\text{O}_8(\text{Ale})_4$  acquired (top) without CP and heteronuclear decoupling, (medium) without CP but under heteronuclear TPPM decoupling, and (bottom) with CP and heteronuclear TPPM decoupling.



**Figure SI9:** Cyclic voltammograms of  $Mo_8S_8(Ale)_4$  deposited onto a glassy carbon electrode (b) compared to the bare glassy carbon electrode (a) in  $CH_3CN + 0.1 M$  tetrahexylammonium perchlorate as electrolyte in the presence of trifluoroacetic acid  $53 \mu M$ . Scan rate  $0.1 V s^{-1}$ ; potentials are given vs  $AgCl/Ag$  electrode.