Encapsulation Controlled Single Molecule Magnetism in

Tetrathiafulvalene-Capped Cyanide-Bridged Cubes

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Supporting Information



Figure S1. Interactions between TTF moieties.

TTF1 interacts with TTF 1(inversion symmetry), TTF2 and TTF3. TTF2 & 3 interact only with TTF1. TTF4 has no direct interaction with another TTF group.

The overlap between TTF1 is largest although there is no S-S contact between them, probably due to face-to-face π - π interaction.

Overlap integrals $S (\times 10^{-3} \text{ eV})$ $S_{1-2} = 1.284, S_{1-3} = -1.400, S_{1-1} = 11.42$



Figure S2. ESI mass spectra of **2** (top) and **3** (bottom). For **2** $[M-4(PF_6)]^{3+}$ seen at 1547.6 *m/z* (calcd: 1547.5) and $[M-4(PF_6)]^{4+}$ at 1160.6 (1160.7) *m/z* (M = $[Fe_4Ni_4(CN)_{12}(pztp)_4(L2)_4](PF_6)_4$.



Figure S3. ESI-MS spectra of **3**. Isotope pattern of the $[M-3(BF_4)+H^+]^{4+}$ (top), ($[M-3(BF_4)]^{3+}$ (centre) and ($[M-3(BF_4)]^{2+}$ (bottom) peaks at [found (calcd.)] 1100.6 (1100.5), 1467.1 (1467.1) and 2200.6 (2200.5) *m/z* respectively. M = Na[Fe₄Ni₄(CN)₁₂(tp)₄L2₄](BF₄)₃



Figure S4. IR spectra of 1, 2, and 3.



Figure S5. Cyclic voltammograms of L1 (top) and L2 (bottom) at a carbon electrode in 1.0 mM solution of MeCN using n-Bu₄NPF₆ as an electrolyte at a scan rate of 0.10 Vs⁻¹.



Figure S6. Bulk electrolysis with coulometry of 2 at 1.2 V versus SCE.



Figure S7. Bulk electrolysis with coulometry of **3** at 1.2 V versus SCE.



Figure S8. Bulk electrolysis with coulometry of **3** at -0.4 V versus SCE.



Figure S9. Uv-vis-nir spectra of 2 (dotted line) and 3 (solid line).



Figure S10. Electronic absorption spectra of **3** upon electrochemical reduction from 0.36 to -0.40 V (potential vs. SCE, 1.0 mM, 0.1 M n-Bu₄NPF₆ in acetonitrile, with Pt mesh working electrode, and Pt wire counter electrode).



Figure S11. Magnetic field induced magnetization for 1, 2, and 3 at 1.8 K.