

Supporting Information for:

Evaluation of quinoline as a remote sensitiser of visible and near-infrared lanthanide luminescence in solution and solid state

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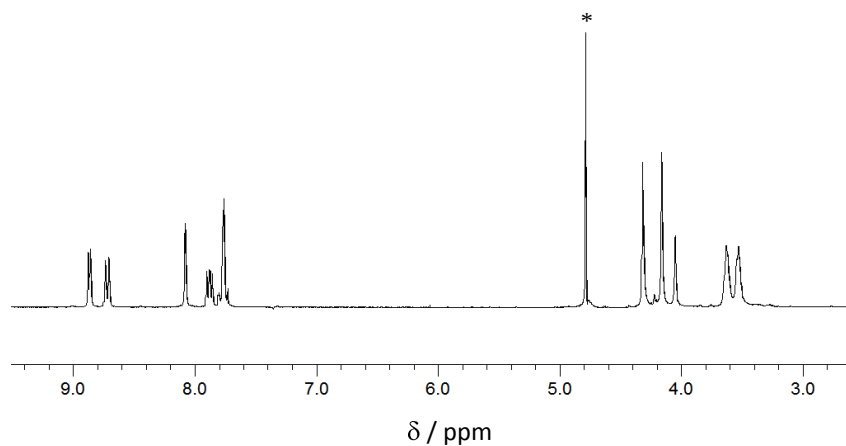


Figure S1: 300 MHz ^1H NMR spectrum of $\text{H}_3\text{L}^{\text{a}}$ in D_2O (* = solvent).

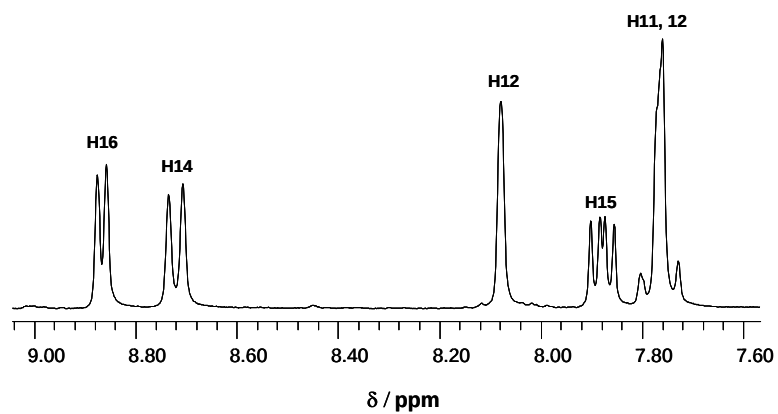


Figure S2 : Enlargement of the aromatic region of the 300 MHz ^1H NMR spectrum of H_3L in D_2O .

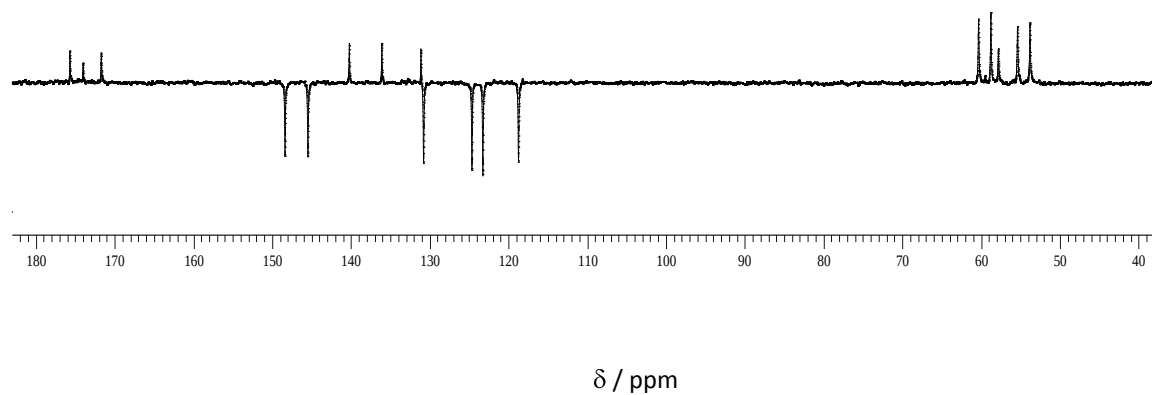


Figure S3: 75 MHz ^{13}C { ^1H } PENDANT NMR spectrum of H_3L in D_2O .

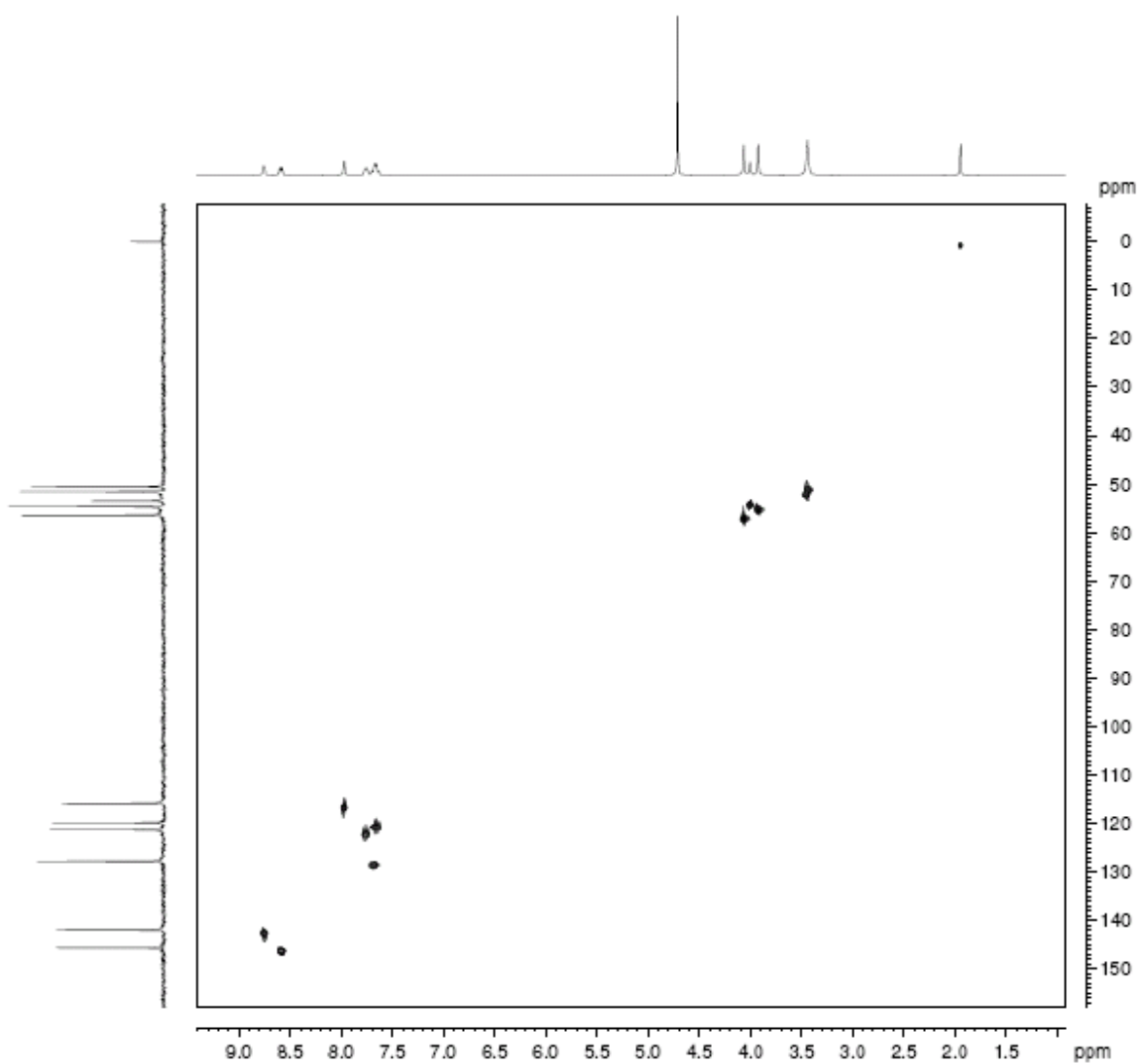


Figure S4: ^1H - ^{13}C HSQC NMR spectrum of **H₃L** in D_2O .

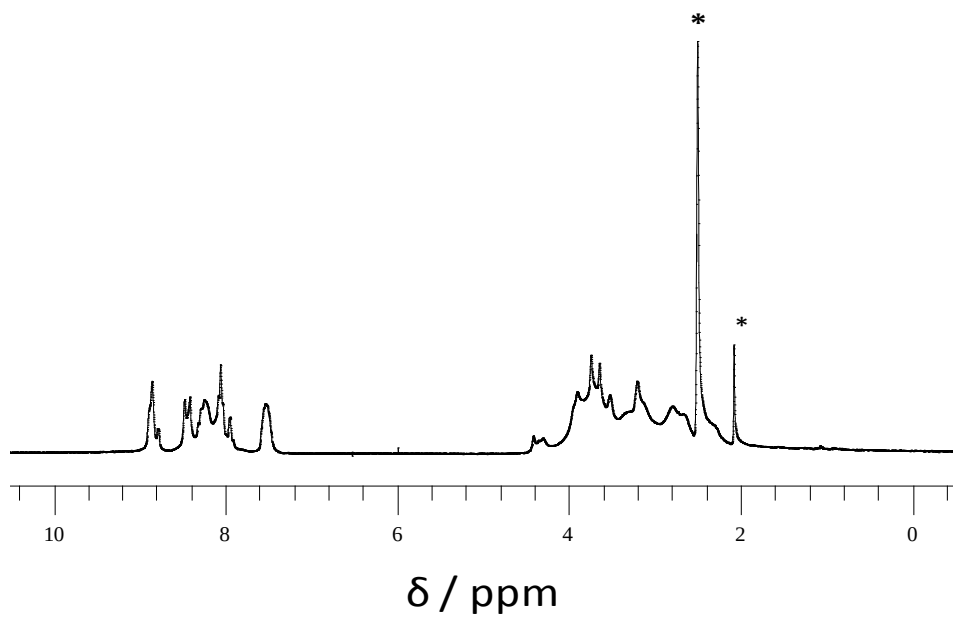


Figure S5 : 300 MHz ^1H NMR spectrum of **YL** in d_6 -DMSO (* = solvent).

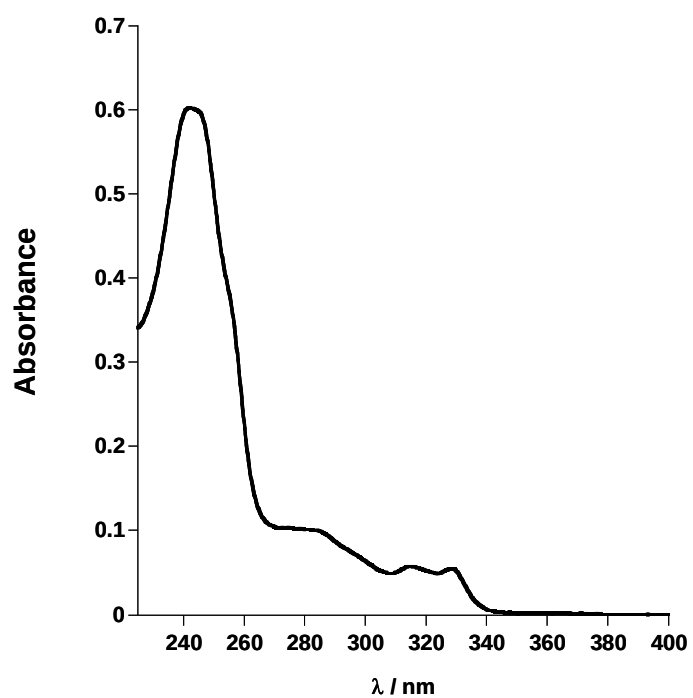


Figure S6: UV-Vis absorption spectrum of $1.20 \times 10^{-5} \text{ mol dm}^{-3} \text{ H}_3\text{L}$ in water.

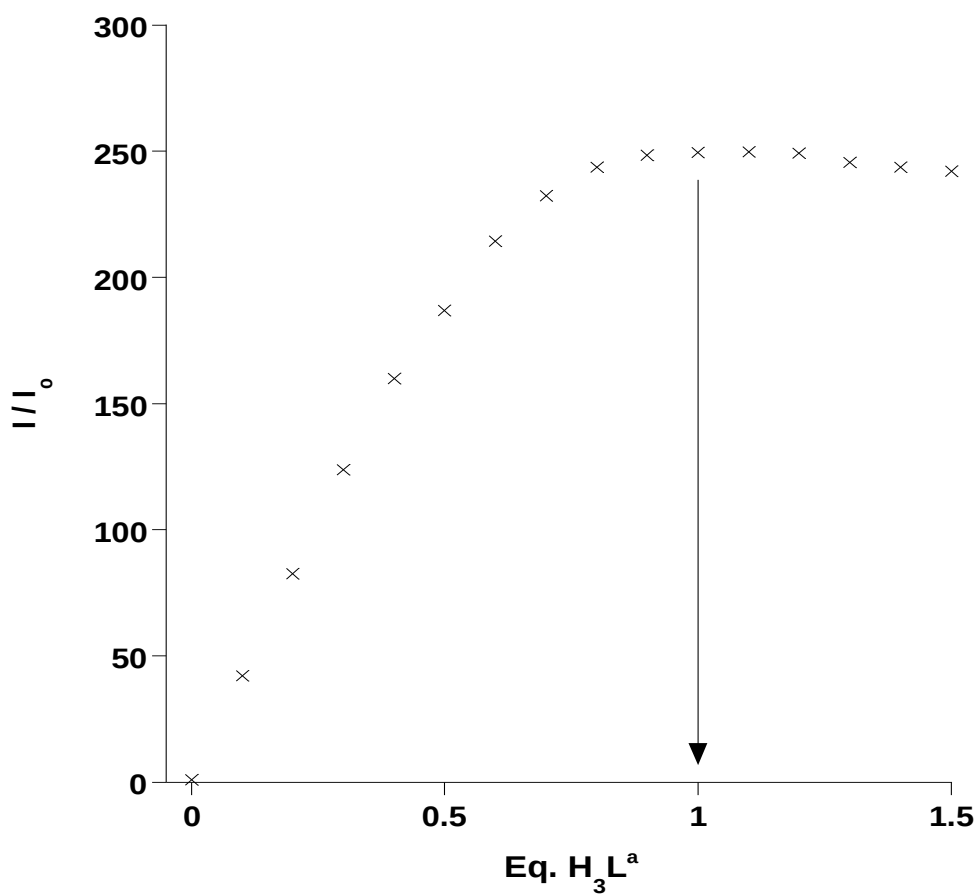


Figure S7: Plot of relative growth in emission intensity of the 613 nm band (I/I_0) vs. equivalents of H_3L $\lambda_{exc} = 330$ nm.

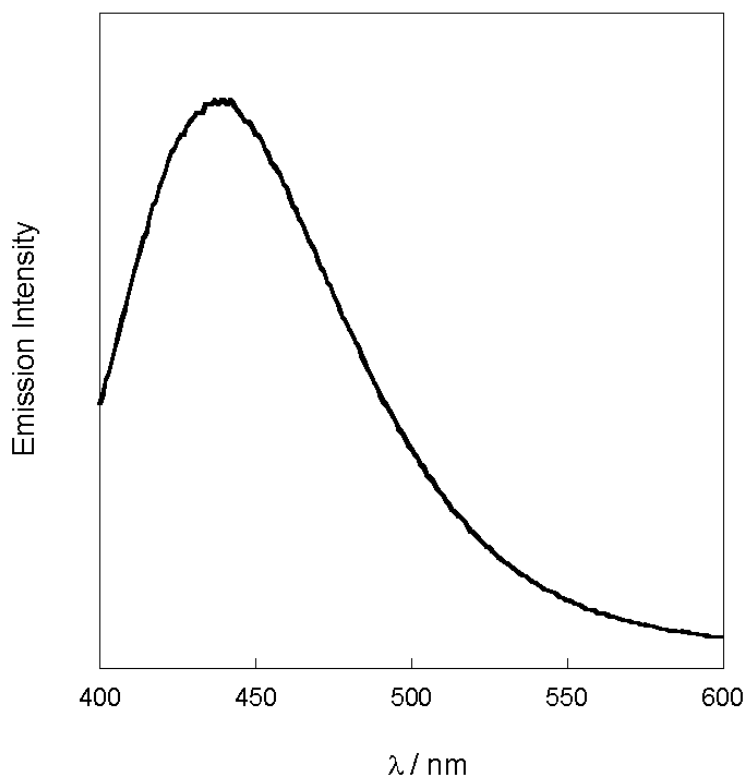


Figure S8: Emission spectrum of **GdL** in methanol. $\lambda_{\text{exc}} = 330 \text{ nm}$.

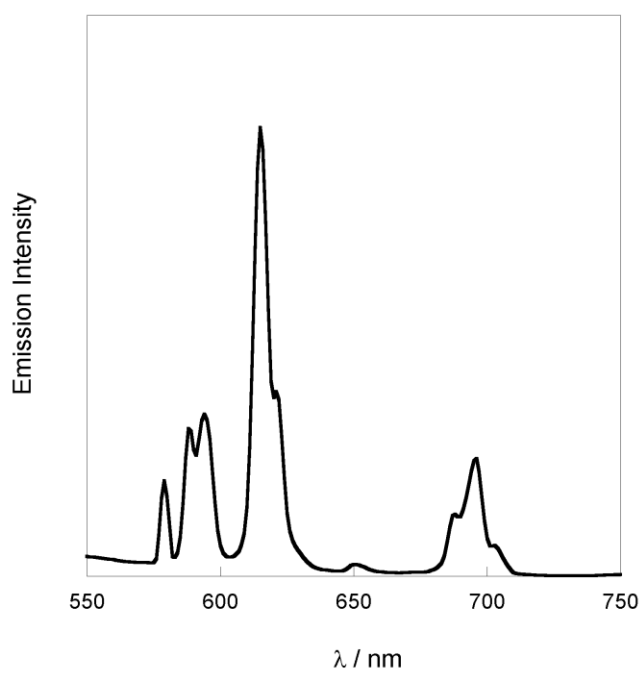


Figure S9: Emission spectrum of **EuL** in methanol. $\lambda_{\text{exc}} = 375 \text{ nm}$.

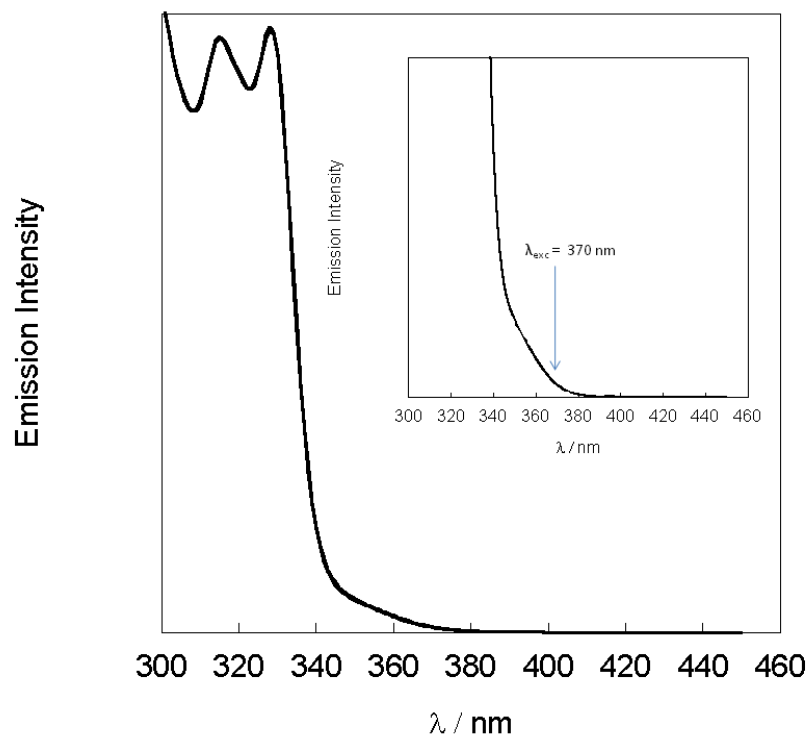


Figure S10: Excitation spectrum associated with emission observed in Figure S9. **EuL** in methanol. $\lambda_{\text{em}} = 614 \text{ nm}$. Inset: enlargement of red-edge, quinoline excitation band.

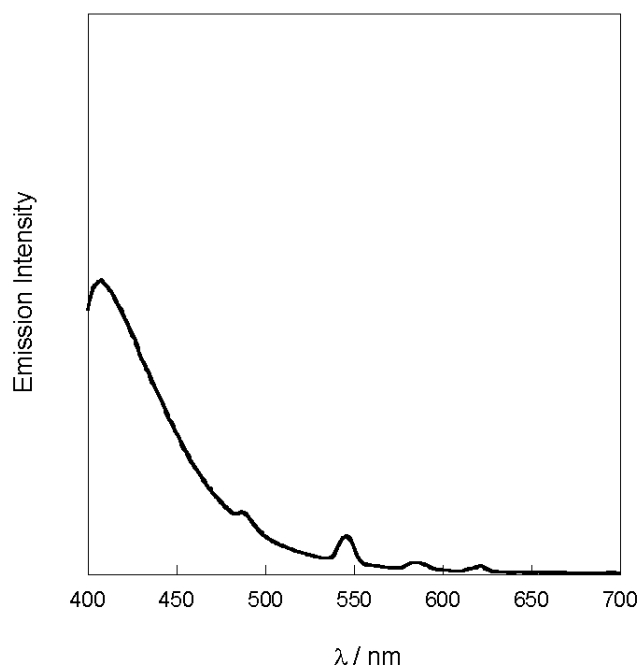


Figure S11: Emission spectrum of **TbL** in methanol. $\lambda_{\text{exc}} = 330 \text{ nm}$.

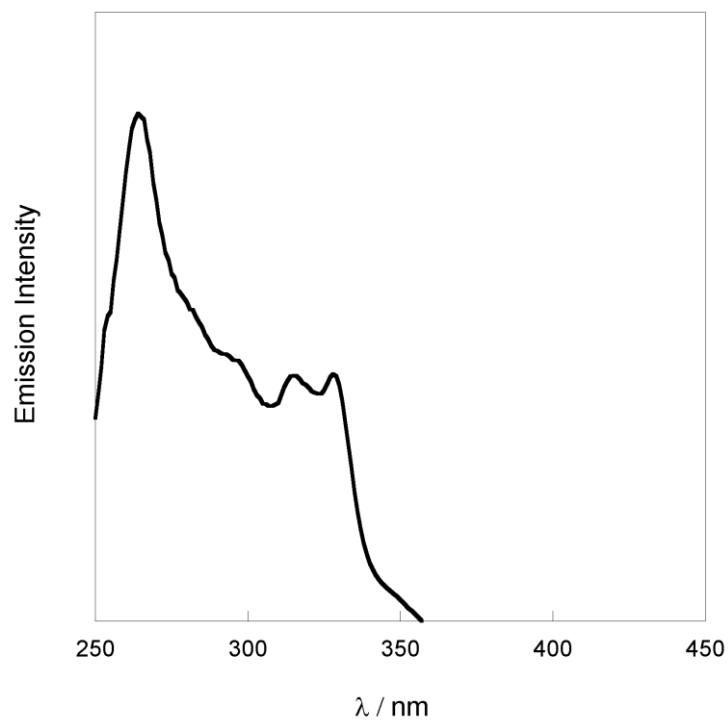


Figure S12: Corrected excitation spectrum of **SmL** in methanol. $\lambda_{\text{em}} = 596$ nm.

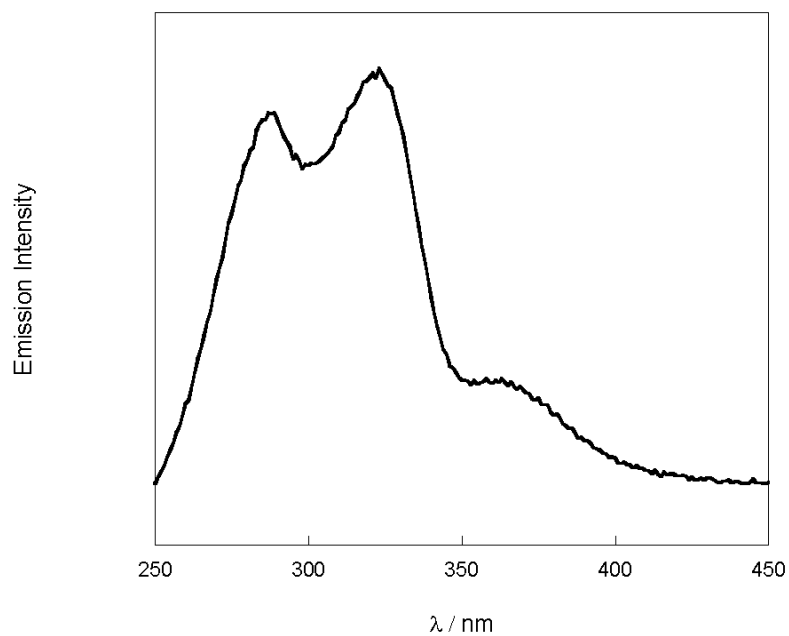


Figure S13: Corrected Excitation spectrum of **NdL** powder. $\lambda_{\text{em}} = 1064$ nm.

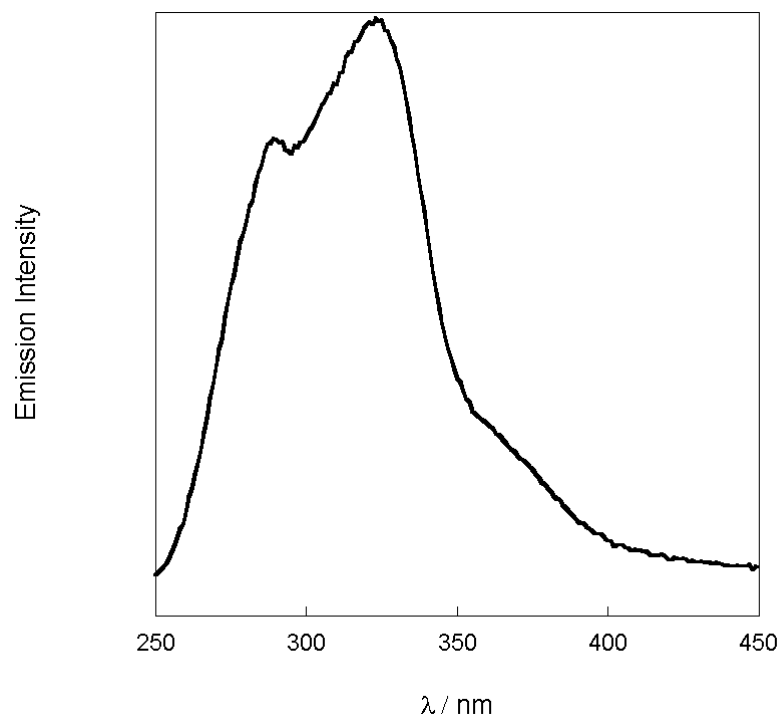


Figure S14: Corrected excitation spectrum of **YbL** powder. $\lambda_{\text{em}} = 980$ nm.

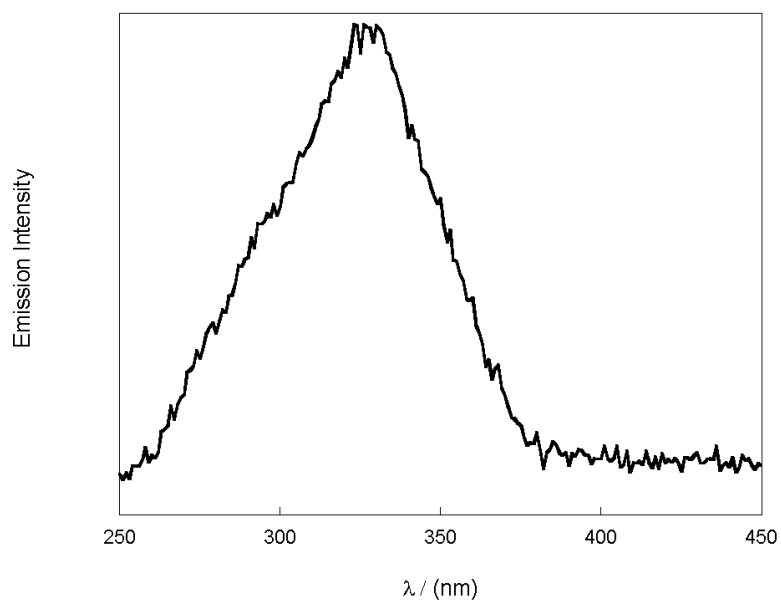


Figure S15: Uncorrected excitation spectrum of **ErL** powder. $\lambda_{\text{em}} = 1520$ nm.

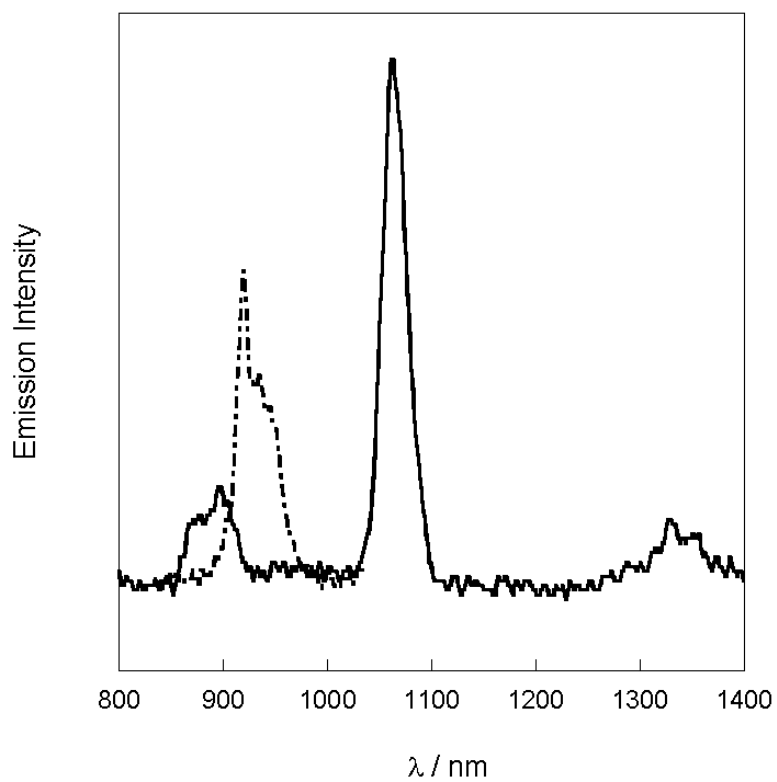


Figure S16: Emission spectra of **NdL** (—) and **YbL** (---) in anhydrous methanol. $\lambda_{\text{exc}} = 330$ nm.